Svnlett

Y. Matsushima et al.

Letter

Rhodium(I)-Catalyzed Cycloisomerization of 1,6-Enynes

Yuji Matsushima^a Eric M. Phillips^a Robert G. Bergman^b Jonathan A. Ellman^{*a}

^a Department of Chemistry, Yale University, New Haven, CT 06520, USA jonathan.ellman@yale.edu

^b Lawrence Berkeley National Laboratory and Department of Chemistry, University of California, Berkeley, CA 94720, USA

This manuscript is dedicated to Peter Vollhardt, who launched SYNLETT 25 years ago and since that time has continuously served in a number of important editorial capacities.



Received: 21.01.2015 Accepted after revision: 23.02.2015 Published online: 27.02.2015 DOI: 10.1055/s-0034-1380359; Art ID: st-2015-b0044-I

Abstract A new and unexpected rhodium(I)-catalyzed cycloisomerization of 1,6-enynes is reported. Several different alkyne substitution patterns were evaluated under the reaction conditions, including a deuterated derivative that provides some insight into the reaction mechanism.

Key words rhodium, homogeneous catalysis, ring closure, isomerization, enones

Previously we have published on rhodium(1)-catalyzed C–H alkenylation and electrocyclization cascades for the convergent assembly of 1,2-dihydropyridines **4** from α , β -unsaturated imines **1** and alkynes **2** (Scheme 1).^{1,2} The 1,2-di-hydropyridine products have further proven to be versatile intermediates for the synthesis of a variety of heterocyclic structures,¹ including pyridines,^{1,3,4} piperidines,⁵ isoquinuclidines,⁶ and tropanes.⁷



1,2-dihydropyridines

To access complex, multicyclic heterocycles **6** with high levels of regiocontrol we have explored the intramolecular alkenylation of substrates **5** with the alkyne tethered to the α , β -unsaturated imine via the nitrogen substituent (Scheme 2).⁸ Subsequent electrocylization provides **6** with bridgehead double bonds.



Scheme 2 Intramolecular C–H alkenylation–electrocyclization cascade of substrates 5 with alkynes tethered to the nitrogen

In the present study we explored cyclization of 1,6-enyne substrates **7** in which the alkynyl group is tethered to the α , β -unsaturated imine functionality with a different connectivity than that used for **5** (Scheme 3). However, to our surprise, the rhodium(I)-catalyzed reaction of 1,6-enynes **7** did not provide any of the expected bicyclic products **8**, but rather resulted in the cycloisomerization products **9**, which upon hydrolysis provided exocyclic enals **10**⁹ with good levels of *Z*-selectivity.

We began our investigations by exploring the rhodium(1)-catalyzed transformation of α , β -unsaturated imine **7a** (Scheme 4). Using conditions previously determined to be optimal for α , β -unsaturated imine C–H bond functionalization, which employed Rh[Cl(coe)₂]₂ as the precatalyst and the commercially available electron-rich phosphine 4-Me₂NPhPEt₂, exocyclic enal **10a** was obtained in 48% yield, predominantly as the less stable *Z*-isomer. The stereo-

1534

Y. Matsushima et al.



Scheme 3 Unexpected reaction pathway for substrate 7 with alkynes tethered to the β -carbon of the α , β -unsaturated imine

chemistry of **10a** was further rigorously confirmed by complete isomerization to the more stable *E*-isomer **11a**⁹ under acidic conditions.



Scheme 4 Rhodium-catalyzed cycloisomerization of **7a** and confirmation of stereochemistry of hydrolysis product **10a** by equilibration to **11a**

An intriguing aspect of this cycloisomerization reaction is the formal *trans*-C–H bond addition across the alkynyl group.¹⁰ We hypothesized that **7a** might first isomerize to a terminal allene or alkyne prior to cyclization and therefore evaluated methyl-deuterated substrate **7b** (Scheme 5). Product **10b** was isolated in the same yield and stereoisomeric purity as **10a** with the methyl group remaining fully deuterated without any deuterium transfer to other sites in the structure. This result argues against the cycloisomerization first proceeding by π -bond isomerization.

Two additional substrates were evaluated to demonstrate that the reaction is applicable to substitution patterns beyond methyl alkyne derivatives. As shown in Scheme 6, ethyl 1,6-enyne **7c** and benzyl 1,6-enyne **7d** provided cyclic products **10c** and **10d**, respectively, in comparable yields and with very high selectivity for the *Z*-alkene isomer. We believe that the higher selectivity for these



Letter

Downloaded by: University of Florida. Copyrighted material.

Scheme 5 No deuterium exchange occurs upon rhodium-catalyzed cyclisomerization of deuterated substrate **7b**

more sterically hindered products is due to reduced isomerization during imine hydrolysis upon filtration through alumina.¹¹



Scheme 6 C–H alkenylation–electrocyclization cascade to provide 1,2-dihydropyridines

Cycloisomerizations of 1,6-envnes to give cyclohexenebased products have previously been reported using ruthenium and molybdenum metathesis catalysts,¹² cationic gold catalysts,¹³ and even rhodium(I) catalysts.^{14,15} However, almost all of the previous reports, including all of the rhodium(I)-catalyzed transformations, employ 1,6-enyne substrates that incorporate a terminal alkyne. Transformations of 1,6-envnes with internal alkynes to give cyclohexenyl products are limited to ruthenium and molybdenum catalysts proceeding by endo-selective envne ring-closing metathesis pathways. In fact, the previously reported rhodium(I)-catalyzed cycloisomerizations of 1,6-enynes all contained terminal alkynes and monosubstituted alkenes, and for this class of 1,6-enynes, mechanistic studies support a reaction pathway that proceeds via a rhodium-vinylidene intermediate. It is notable that rhodium-vinylidene intermediates are not accessible for the 1,6-envnes 7 reported here that incorporate internal alkynes.

In conclusion, we have identified a novel rhodium(I)catalyzed cycloisomerization of 1,6-enynes incorporating internal alkyne moieties that gives functionalized sixmembered carbocyclic systems. Further mechanistic inquiry will be necessary to elucidate the reaction mechanism.

Acknowledgment

This work was supported by the NIH under Grant No. GM069559 (J.A.E.). R.G.B. was supported by the Director, Office of Science, Office of Basic Energy Sciences, and by the Division of Chemical Sciences,

Y. Matsushima et al.

Geosciences, and Biosciences of the U.S. Department of Energy at LBNL under Contract No. DE-AC02-05CH11231. E.M.P. also acknowledges support from an NRSA postdoctoral fellowship (F32GM090661).

Supporting Information

Supporting information for this article is available online at http://dx.doi.org/10.1055/s-0034-1380359. Included are the synthesis procedures and analytical data for the cycloisomerization substrates and intermediates in their synthesis. In addition, spectra are provided for intermediates and cycloisomerization substrates **7a-d** and products **10a-d** and **11a**.

References and Notes

- (a) Colby, D. A.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2008, 130, 3645. (b) For a general review on the synthesis and further elaboration of 1,2-dihydropyridines prepared by C–H bond functionalization–electrocyclization cascades, see: Mesganaw, T.; Ellman, J. A. Org. Process Res. Dev. 2014, 18, 1097.
- (2) For reviews on the synthesis and applications of 1,2-dihydro-pyridines, see: (a) Tanaka, K.; Fukase, K.; Katsumura, S. Synlett **2011**, 2115. (b) Bull, J. A.; Mousseau, J. J.; Pelletier, G.; Charette, A. B. Chem. Rev. **2012**, *112*, 2642. (c) Silva, E. M. P.; Varandas, P. A. M. M.; Silva, A. M. S. Synthesis **2013**, *45*, 3053.
- (3) Martin, R. M.; Bergman, R. G.; Ellman, J. A. J. Org. Chem. 2012, 77, 2501.
- (4) For leading references on alternative syntheses of pyridines via C-H activation, see: (a) Parthasarathy, K.; Jeganmohan, M.; Cheng, C.-H. Org. Lett. 2008, 10, 325. (b) Hyster, T. K.; Rovis, T. Chem. Commun. 2011, 47, 11846. (c) Too, P. C.; Noji, T.; Lim, Y. J.; Li, X.; Chiba, S. Synlett 2011, 2789. (d) Neely, J. M.; Rovis, T. J. Am. Chem. Soc. 2013, 135, 66. (e) Zhang, Q.-R.; Huang, J.-R.; Zhang, W.; Dong, L. Org. Lett. 2014, 16, 1684.
- (5) (a) Duttwyler, S.; Lu, C.; Rheingold, A. L.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2012, 134, 4064. (b) Duttwyler, S.; Chen, S.; Takase, M. K.; Wiberg, K. B.; Bergman, R. G.; Ellman, J. A. Science 2013, 339, 678. (c) Ischay, M. A.; Takase, M. K.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2013, 135, 2478. (d) Duttwyler, S.; Chen, S.; Lu, C.; Mercado, B. Q.; Bergman, R. G.; Ellman, J. A. Angew. Chem. Int. Ed. 2014, 53, 3877. (e) Mesganaw, T.; Ellman, J. A. Org. Process Res. Dev. 2014, 18, 1105.
- (6) Martin, R. M.; Bergman, R. G.; Ellman, J. A. Org. Lett. 2013, 15, 444.
- (7) Ischay, M. A.; Takase, M. K.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2013, 135, 2478.
- (8) Yotphan, S.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2008, 130, 2452.
- (9) Experimental Procedures for the Generation of Imines and Rhodium-Catalyzed Cycloisomerization (Z)-2-(2-Methylcyclohex-2-en-1-ylidene)acetaldehyde (10a) In an inert atmosphere box, to the solution of 7a (86 mg, 0.63 mmol) in toluene (3 mL) was added benzylamine (68 mg, 0.63 mmol) and 3 Å MS (800 mg). The flask was removed from the box, and the mixture was stirred at r.t. for 3 h. The 3 Å MS were removed via filtration over Celite, which was washed with toluene (8 mL). The filtrate was degassed and brought into an inert atmosphere box. To the solution was added a solution of [RhCl(coe)₂]₂ (23 mg, 0.031 mmol) and Me₂NPhPEt₂ (13 mg,

0.62 mmol) in toluene (2 mL), and the mixture was stirred at 75 °C for 1 h. After removal of the solvent, the residual oil was purified by column chromatography on grade III Al₂O₃ (hexanes–EtOAc, 100:0 to 99:1,) to afford **10a** (*Z*/*E* = 6.7:1) as colorless oil (41 mg, 0.30 mmol, 48% yield). *R*_f = 0.70 (hexanes–EtOAc. 4:1). ¹H NMR (400 MHz, CDCl₃): δ = 10.23 (d, *J* = 8.4 Hz, 1 H), 6.11 (ddd, *J* = 5.4, 2.7, 1.3 Hz, 1 H), 5.81 (d, *J* = 8.4 Hz, 1 H), 2.44 (ddd, *J* = 6.5, 4.3, 1.2 Hz, 2 H), 2.30–2.20 (m, 2 H), 2.17 (dd, *J* = 3.3, 1.8 Hz, 3 H), 1.84–1.74 (m, 2 H). ¹³C NMR (101 MHz, CDCl₃): δ = 192.28, 156.42, 138.87, 131.95, 127.42, 35.59, 26.55, 26.04, 22.34. IR (thin film): 2927, 2867, 2834, 1653, 1615, 1580, 1448, 1406, 1223, 1178, 1149, 1130, 1101, 1025, 948 cm⁻¹. MS (EI): *m*/*z* [M]* calcd for C₉H₁₂O*: 136.09; found: 136.10.

(*Z*)-2-[2-(Methyl-*d*₃)cyclohex-2-en-1-ylidene]acetaldehyde (10b)

Compound **10b** was synthesized according to the procedure used for compound **10a**. From 80 mg (0.57 mmol) of **7b** was obtained 39 mg (0.28 mmol, 49% yield) of **10b** (*Z*/*E* = 6.7:1) as colorless oil after purification by column chromatography on grade III Al₂O₃ eluting with hexanes–EtOAc (100:0 to 99:1). R_f = 0.70 (hexanes–EtOAc. 4:1). ¹H NMR (400 MHz, CDCl₃): δ = 10.21 (d, *J* = 8.5 Hz, 1 H), 6.11 (td, *J* = 4.2, 1.3 Hz, 1 H), 5.81 (d, *J* = 8.5 Hz, 1 H), 2.48–2.40 (m, 2 H), 2.29–2.21 (m, 2 H), 1.83–1.74 (m, 2 H). ¹³C NMR (126 MHz, CDCl₃): δ = 192.28, 156.44, 138.87, 131.89, 127.46, 35.60, 26.57, 22.38. IR (thin film): 3009, 2926, 1651, 1611, 1580, 1406, 1230, 1156, 1137, 1096, 1051, 974 cm⁻¹. MS (EI): *m*/*z* [M]⁺ calcd for C₉H₉D₃O⁺: 139.11; found: 139.15. (*Z*)-**2-(2-Ethylcyclohex-2-en-1-ylidene)acetaldehyde (10c)**

Compound **10c** was synthesized according to the procedure used for compound **10a**. From 40 mg (0.27 mmol) of **7c** was obtained 18 mg (0.12 mmol, 45% yield) of **10c** (*Z*/*E* = 20:1) as colorless oil after purification by column chromatography on grade III Al₂O₃ eluting with hexanes–EtOAc (100:0 to 99:1). R_f = 0.70 (hexanes–EtOAc. 4:1). ¹H NMR (400 MHz, CDCl₃): δ = 10.11 (d, *J* = 8.5 Hz, 1 H), 6.08 (ddd, *J* = 5.4, 2.8, 1.3 Hz, 1 H), 5.76 (d, *J* = 8.5 Hz, 1 H), 2.53–2.44 (m, 2 H), 2.44–2.37 (m, 2 H), 2.30– 2.22 (m, 2 H), 1.84–1.75 (m, 2 H), 1.11 (t, *J* = 7.4 Hz, 3 H). ¹³C NMR (126 MHz, CDCl₃): δ = 192.40, 156.54, 138.12, 135.90, 126.41, 36.28, 30.99, 26.52, 22.88, 13.16. IR (thin film): 2966, 2928, 2876, 1660, 1617, 1583, 1453, 1409, 1222, 1174, 1150, 1127, 1107, 1081cm⁻¹. MS (EI): *m*/*z* [M]⁺ calcd for C₁₀H₁₄O⁺: 150.10; found: 150.10.

(Z)-2-(2-Benzylcyclohex-2-en-1-ylidene)acetaldehyde (10d)

Compound **10d** was synthesized according to the procedure used for compound **10a**. From 63 mg (0.30 mmol) of **7d** was obtained 32 mg (0.15 mmol, 51% yield) of **10d** (*Z*/*E* = 15.5:1) as colorless oil after purification by column chromatography on grade III Al₂O₃ eluting with hexanes–EtOAc (100:0 to 99:1). *R*_f = 0.70 (hexanes–EtOAc. 4:1). ¹H NMR (500 MHz, CDCl₃): δ = 10.04 (d, *J* = 8.4 Hz, 1 H), 7.29 (t, *J* = 7.5 Hz, 2 H), 7.21 (t, *J* = 7.4 Hz, 1 H), 7.13 (d, *J* = 7.3 Hz, 2 H), 6.01 (td, *J* = 4.0, 1.0 Hz, 1 H), 5.73 (d, *J* = 8.4 Hz, 1 H), 3.83 (s, 2 H), 2.50–2.43 (m, 2 H), 2.32 (ddd, *J* = 6.0, 5.1, 2.0 Hz, 2 H), 1.90–1.82 (m, 2 H). ¹³C NMR (126 MHz, CDCl₃): δ = 192.10, 155.61, 140.14, 138.71, 135.05, 128.58, 128.55, 126.66, 126.41, 44.05, 36.22, 26.72, 22.65. IR (thin film): 3025, 2925, 2862, 1688, 1653, 1616, 1582, 1495, 1453, 1405, 1223, 1147, 1124, 978 cm⁻¹. MS (EI): *m*/*z* [M]⁺ calcd for C₁₅H₁₆O⁺: 212.12; found: 212.10.

(*E*)-2-(2-Methylcyclohex-2-en-1-ylidene)acetaldehyde (11a) To a solution of **10a** (40 mg, 0.29 mmol) in THF (2 mL) was added a 1 M aqueous solution of HCl (1 mL, 1 mmol), and the mixture was stirred at r.t. for 18 h. After being neutralized with aq Na₂CO₃, the aqueous layer was extracted with EtOAc (3×5 ۸

Y. Matsushima et al.

mL). The combined organic layers were washed with H₂O and brine and dried over with MgSO₄. After filtration, the solvent was removed under reduced pressure. The residual oil was purified by column chromatography on silica gel (33:1, pentane–Et₂O) to afford **11a** as a colorless oil (36 mg, 0.26 mmol, 89% yield). $R_f = 0.70$ (hexanes–EtOAc. 4:1). ¹H NMR (400 MHz, CDCl₃): $\delta = 10.15$ (d, J = 8.1 Hz, 1 H), 6.18 (t, J = 4.3 Hz, 1 H), 5.93 (d, J = 8.1 Hz, 1 H), 2.95–2.86 (m, 2 H), 2.26 (dtd, J = 7.8, 4.1, 1.9 Hz, 2 H), 1.85 (dd, J = 3.1, 1.7 Hz, 3 H), 1.80 (dt, J = 12.5, 6.2 Hz, 2 H). ¹³C NMR (126 MHz, CDCl₃): $\delta = 191.49$, 157.28, 138.03, 132.99, 122.62, 26.32, 25.82, 22.26, 19.65. IR (thin film): 2924, 2860, 1663, 1622, 1591, 1455, 1435, 1401, 1386, 1370, 1177, 1145, 1087, 1048 cm⁻¹. MS (EI): m/z [M]⁺ calcd for C₉H₁₂O⁺: 136.09; found: 136.00.

- (10) For an unusual rhodium-catalyzed intramolecular *trans* hydroacylation of an alkyne, see: Tanaka, K.; Fu, G. C. *J. Am. Chem. Soc.* **2001**, *123*, 11492.
- (11) In previous studies on rhodium-catalyzed β -alkylation of α , β unsaturated imines with alkenes, we observed >95:5 *Z*/*E* selectivity for rhodium-catalyzed alkylation, but this *Z*/*E* ratio degraded by as much as 5–15% during alumina-mediated imine

hydrolysis. This loss in stereochemical purity was dependent upon the structure of the imine. Colby, D. A.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. **2006**, *128*, 5604.

- (12) (a) Trillo, B.; Gulias, M.; Lopez, F.; Castedo, L.; Mascarenas, J. L. *J. Organomet. Chem.* **2005**, 690, 5609. (b) Sashuk, V.; Grela, K. *J. Mol. Catal. A: Chem.* **2006**, 257, 59. (c) Singh, R.; Schrock, R. R.; Müller, P.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2007**, 129, 12654. (d) Lee, Y.-J.; Schrock, R. R.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2009**, 131, 10652.
- (13) Nieto-Oberhuber, C.; Munoz, M. P.; Lopez, S.; Jiminez-Nunez, E.; Nevado, C.; Herrero-Gomez, E.; Raducan, M.; Echavarren, A. M. *Chem. Eur. J.* **2006**, *12*, 1677.
- (14) (a) Grigg, R.; Stevenson, P.; Worakun, T. *Tetrahedron* **1988**, 44, 4967. (b) Kim, H.; Lee, C. J. Am. Chem. Soc. **2005**, 127, 10180. (c) Joo, J. M.; Yuan, Y.; Lee, C. J. Am. Chem. Soc. **2006**, 128, 14818. (d) Joo, J. M.; Ramoncito, A. D.; Lee, C. Org. Lett. **2010**, 12, 5704.
- (15) For leading references on rhodium-catalyzed cycloisomerizations that proceed by ene-type pathways, see: (a) Okazaki, E.; Okamoto, R.; Shibata, Y.; Noguchi, K.; Tanaka, K. Angew. Chem. Int. Ed. 2012, 51, 6722. (b) Okamoto, R.; Okazaki, E.; Noguchi, K.; Tanaka, K. Org. Lett. 2011, 13, 4894.