This article was downloaded by: [University Of Pittsburgh] On: 17 October 2013, At: 03:41 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

A Facile Regioselective 1,6-O-Diacetylation Method of Methyl-Glycopyranosides and Their Dimethyl Phosphonates

Guang-tao Zhang $^{a\ b}$, Zhong-wu Guo a & Yong-zheng Hui a

^a National Laboratory of Bio-organic and Natural Products Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai, 200032, China

^b Department of Chemical Physics, the University of Science and Technology of China, Hefei, 230026, China

Published online: 22 Aug 2006.

To cite this article: Guang-tao Zhang , Zhong-wu Guo & Yong-zheng Hui (1997) A Facile Regioselective 1,6-O-Diacetylation Method of Methyl-Glycopyranosides and Their Dimethyl Phosphonates, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 27:11, 1907-1917, DOI: 10.1080/00397919708006792

To link to this article: http://dx.doi.org/10.1080/00397919708006792

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

A FACILE REGIOSELECTIVE 1,6-*O*-DIACETYLATION METHOD OF METHYL-GLYCOPYRANOSIDES AND THEIR DIMETHYL PHOSPHONATES

Guang-tao Zhang^{1,2}, Zhong-wu Guo^{1*}, and Yong-zheng Hui^{1*}

 National Laboratory of Bio-organic and Natural Products Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai, 200032, China. 2. Department of Chemical Physics, the University of Science and Technology of China, Hefei, 230026, China.

Abstract: A general method of 1,6-di-O-acetylation of methyl- α/β -glycopyranosides and a strategy to synthesize protected dimethyl (2,3,4-tri-O-benzyl- α -D-glycosyl) phosphonate having a free OH at C-6 are reported.

Saccharides play an important role in the development, growth, function and survival of the organism. ¹ Many oligosaccharides have biological activities contain 1-6 units, such as gentiobiose, dextran and etc.. ² A kind of sulfated mannose polysaccharide with anti-HIV activity from the Pacific tunicate *Didemmim malle* discovered recently also consists of 1-6 units. ³ So protected

Copyright © 1997 by Marcel Dekker, Inc.

^{*}To whom correspondence should be addressed.

sugars having acetyl or free hydroxyl groups at 1-C and 6-C are important starting materials for the preparation of oligo- and polysaccharides and glycoconjugates.

Our current interest in glycosyl phosphonate as glycosyltransferase inhibitors directed our attention to the search for a facile method for the synthesis of the required starting materials, namely benzyl protected α or β -hexopyranoses having 1-OH or/and 6-OH or 1-OAc or/and 6-OAc. These compounds can be easily converted into glycosyl donors ⁴ containing -C(=NH)CCl₃, -SEt, -SPh, -SMe, halides and phosphate⁵ and phosphonate⁶ as inhibitors at the anomeric carbon. The most widely used and traditional method of preparation of 1,6diacetylhexoses is a multistep procedure to synthesize the 1,6-anhydropyranoses first, and then to open the ring to afford the desired materials. ⁷ Longer synthetic routes lead, in general, to anomeric mixtures, difficult to separate and with the consequent loss of yield.

We now report a direct and general regio- and stereoselective diacetylation procedure at 1-C and 6-C for methyl tetra-O-benzyl- α/β -D-glycopyranoses **2a-2c** (as shown in Scheme I) by treatment with Ac₂O/H₂SO₄ without solvent, to give in good to excellent yield the corresponding diacetylated derivatives at 1-C and 6-C **3a-3c** (as shown in Scheme II). Selective hydrolysis easily affords the corresponding 1-OH or/and 6-OH groups benzyl protected glycopyranoses. Diacetyl derivatives reacted directly with P(OMe)₃ under Ar in the presence of trimethylsilyl triflate (TMSOTf) to give glycosyl phosphonates (shown in Scheme III), which are potential inhibitors of glycosidases and glycosyltransferases. ⁶ The method has been successfully applied to methyl tetra-O-benzyl- α -Dmannopyranose and corresponding glucopyranose, galactopyranose and other hexopyranoses. Generally, α -pyranoses only give α -isomers, but for β -pyranoses, the products are mixtures of α - and β -isomers. Longer reaction times and higher temperatures give three or four acetyl α - and β -isomers. The relative reactivity of differently located groups was different. For α -D-glucopyranose, the order is: 1-OMe > 6-OBn > 3-OBn > 4-OBn > 2-OBn, while for α -D-mannopyranose it is: 1-OMe > 6-OBn > 4-OBn > 3-OBn > 2-OBn, as observed previously. ⁸





Table I. The Prepar	ation of Benzy	vlated Co	mpounds
---------------------	----------------	-----------	---------

Substrates	Products	Temp.	R. T. (h) 24	Yield* (%) 94
(1a-c)	(2a-c)	(°C)		
Methyl α-D-glucopyranoside	2a	75		
Methyl α -D-galactopyranoside	2b	75	24	89.5
Methyl α -D-mannopyranoside	2c	75	24	97

^{*}note: Isolated yield



Scheme II





Experimental

General methods. Optical rotations were measured with a Perkin-Elmer Model 241 MC Polarimeter at 25 °C. TLCs were performed on precoated plates of Silica Gel HF₂₅₄ (0.5 mm, Qingdao, China) and detected by 10% sulfuric acid in methanol. Flash column chromatography was performed on Silica Gel H (400 mesh). ¹H NMR spectra were recorded at 300 MHz on a Bruker AM-300 spectrometer with tetramethylsilane as the internal standard (unless otherwise specified). Mass spectra were recorded on a VG QUATTRO MS instrument.

General procedure for preparation of methyl 2,3,4,6-tetra-O-benzyl-Dglycopyranosides:

A mixture of methyl D-glycopyranosides (5.0 g), sodium hydride (8.5 g

Substrates	Products	Ac ₂ O:H ₂ SO ₄	R. T.	Yield*
(2a-c)			(min)	(%)
Methyl 2,3,4,6-tetra-O-benzyl	3a	100:1	7	95
α -D-glucopyranoside				(α)
Methyl 2,3,4,6-tetra-O-benzyl	3b	100:1	10	72 (α/β
β -D-galactopyranoside				4.3/1)
Methyl 2,3,4,6-tetra-O-benzyl	3c	100:1	10	90.5
α-D-mannopyranoside				(α)
Methyl 2,3,4,6-tetra-O-benzyl	3a + 4a	100:1	90	82
α-D-glucopyranoside				$(\alpha \text{ and } \beta)$

O-benzyl- α/β -D-glycopyranosides

Table II: The Preparation of 1,6-O-Di-acetyl-2,3,4-tri-

*note: Isolated yield

60%, Fluka) and benzyl chloride (125 mL) was heated at 70-75 °C with stirring for about 3 h, and was monitored by TLC (petroleum ether/ethyl acetate, 3/1). After the reaction was complete, the insoluble material was removed and the filtrate was concentrated *in vacuo*. The crude product was purified by column chromatography eluting with petroleum ether/ethyl acetate (5/1), affording 2 (a-c).

General procedure for the preparation of 1,6-di-*O*-acetyl-2,3,4-tri-*O*-benzyl-D-glycopyranoses:

To a solution of methyl 2,3,4,6-tetra-O-benzyl-D-glycopyranosides (3

mmol) and acetic anhydride (7.2 mL) concentrated sulfuric acid (0.072 mL) was added dropwise at 0 °C. After 7 to 10 min at the same temperature the mixture was quenched by addition of saturated sodium hydrogen carbonate solution and worked up as routine. The crude product was purified by column chromatography eluting with petroleum ether/ethyl acetate ($10/1 \rightarrow 6/1$), affording 3 (a-c) and 4a.

General procedure for the preparation of dimethyl (2,3,4-tri-O-benzyl- α -D-glycosyl) phosphonate (6):

To a solution of 1.00 g (1.873 mmol) of 3a and 0.5 mL (4.25 mmol) of P(OMe)₃ in 5 mL of CH₂Cl₂ under Ar at 0 °C, 0.6 mL (3.3 mmol) of TMSOTF were added. After 12 h, 0.3 mL (2.5 mmol) of P(OMe)₃ and 0.3 mL (1.7 mmol) of TMSOTf were added. After addition 12 h (0-5 °C), the mixture was worked up as usual , and the residue was separated by flash chromatography on silica gel (petroleum ether/ethyl acetate/dichloromethane 1/1/1) to afford 0.561 g (87.3%) of 5 and 0.487 g of reactant. 0.305 g of 5 was hydrolyzed in 10 mL of MeONa/MeOH (0.2 M) overnight. After completion, the mixture was neutralized with Dowex-50w (H⁺), the crude was separated by chromatography (petroleum ether/ethyl acetate/dichloromethane 1/1/1) to give 1.381 g (100%) of 6.

Methyl 2,3,4,6-tetra-*O*-benzyl-α-D-glucopyranoside (2a):

 $[\alpha]_D = +20.9$ ° (c 0.7, CHCl₃), R_f = 0.8 (petroleum ether/ethyl acetate, 3/1); EIMS (m/z): 553 (M-1), 253, 181, 91; ¹H NMR (300 MHz, CDCl₃) δ 3.39 (s, OCH₃), 3.58 (dd, 1H, J_{1,2} = 2.7 Hz, J_{2,3} = 9.5 Hz, H-2), 4.01 (t, 1H, J_{2,4} = 8.9 Hz, H-3), 3.65 (m, 1H, H-5), 3.75 (m, 2H, H-6a, H-6b), 4.66 (d, 1H, H-1), 5.00 and 4.84 (AB, each 1H, J = 10.8 Hz, PhCH₂), 4.85 and 4.48 (AB, each 1H, J = 10.9 Hz,

PhCH₂), 4.82 and 4.48 (AB, each 1H, J = 11.6 Hz, PhCH₂), 4.68 and 4.62 (AB, each 1H, J = 12.2 Hz, PhCH₂), 7.23-7.38 (m, 20H, 4 C₆H₅); ¹³C NMR (CDCl₃) δ 55.17, 57.72, 68.67, 70.18, 73.38, 73.53, 75.00, 77.82, 80.01, 82.18, 98.26 (C-1), 128.4-127.5, 138.9-138.0.

Methyl 2,3,4,6-tetra-O-benzyl-β-D-galactopyranoside (2b):

 $[\alpha]_{D} = -3.94$ ° (*c* 2.03, CHCl₃), R_f = 0.64 (petroleum ether/ethyl acetate, 3/1); EIMS (m/z): 553 (M-1), 253, 181, 91; ¹H NMR (300 MHz, CDCl₃) δ 7.50-7.30 (m, 20H, 4C₆H₅), 4.95 and 4.62 (AB, each 1H, J = 11.7 Hz, PhCH₂), 4.90 and 4.76 (AB, each 1H, J = 11.0 Hz, PhCH₂), 4.73 (d, 2H, PHCH₂), 4.46 and 4.40 (AB, each 1H, J = 11.8 Hz, PhCH₂), 4.28 (d, 1H, J_{1.2} = 7.6 Hz, H-1), 3.90 (bd, 1H, J_{3.4} = 2.8 Hz, H-4), 3.81 (dd, 1H, J_{2.3} = 9.7 Hz, H-2), 3.62-3.57 (m, 3H, H-5, H-6a, H-6b), 3.55 (s, 3H, CH₃O), 3.52 (dd, 1H, H-3); Anal. Calcd. for C₃₅H₃₈O₆ (554.68); C, 75.78; H, 6.90. Found: C, 75.20; H, 6.71.

Methyl 2,3,4,6-tetra-O-benzyl- α -D-mannopyranoside (2c):

 $[\alpha]_D = +31.57$ ° (*c* 1.38, CHCl₃), R_f = 0.71 (petroleum ether/ethyl acetate, 3/1); EIMS (m/z): 553 (M-1), 91; ¹H NMR (300 MHz, CDCl₃) δ 7.46-7.15 (m, 20H, 4C₆H₅), 4.92 and 4.54 (AB, each 1H, J = 10.8 Hz, PhCH₂), 4.81 (d, 1H, J_{1,2} = 1.7 Hz, H-1), 4.77 (s, 2H, PhCH₂), 4.70 and 4.58 (AB, each 1H, J = 12.1 Hz, PhCH₂), 4.63 (s, 2H, PhCH₂), 4.02 (dd, 1H, J_{3,4} = J_{4,5} = 9.2 Hz, H-4), 3.92 (dd, 1H, J_{2,3} = 3.0 Hz, H-3), 3.83-3.74 (m, 4H, H-2, H-5, H-6a, H-6b), 3.35 (s, 3H, CH₃O); Anal. Calcd. for C₃₅H₃₈O₆ (554.68): C, 75.78; H, 6.90. Found: C, 75.35; H, 6.88.

1,6-Di-O-acetyl-2,3,4-tri-O-benzyl-α-D-glucopyranose (3a):

 $[\alpha]_D = +51.3$ ° (c 0.15, CHCl₃), $R_f = 0.57$ (petroleum ether/ethyl acetate, 3/1);

EIMS (m/z): 534 (M), 91(base); ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.24 (m, 15H, 3C₆H₅), 6.36 (d, 1H, J_{1.2} = 3.5 Hz, H-1), 5.04 and 4.88 (AB, each 1H, J = 10.8 Hz, PhCH₂), 4.76 and 4.68 (AB, each 1H, J = 11.5 Hz, PhCH₂), 4.74 and 4.62 (AB, each 1H, J = 10.7 Hz, PhCH₂), 4.40-4.20 (m, 2H, H-6a, H-6b), 4.03 (dd, 1H, J_{2.3} = 9.6 Hz, J_{3.4} = 9.2 Hz,H-3), 3.97 (ddd, 1H, J_{4.5} = 9.9 Hz, J_{5.6a} = 4.0 Hz, J_{5.6b} = 2.2 Hz, H-5), 3.72 (dd, 1H, H-2), 3.62 (dd, 1H, H-4), 2.19 (s, 3H, CH₃CO), 2.06 (s, 3H, CH₃CO); ¹³C NMR (75 MHz, CDCl₃) δ 170.6, 169.3, 138.5, 137.7, 137.6, 130.6-126.7, 89.7 (C-1), 81.7, 80.1, 79.4, 79.0, 78.5, 62.7, 57.1, 21.06, 20.8; Anal. Calcd. for C₃₁H₃₄O₈ (534.61): C, 69.60; H, 6.41. Found: C, 69.33; H, 6.39.

1,6-Di-O-acetyl-2,3,4-tri-O-benzyl- α/β -D-galactopyranose (α/β : 4.3/1) (3b):

 $[\alpha]_D = +41.15$ ° (*c* 0.3, CHCl₃); EIMS (m/z): 533 (M-1), 473, 329, 91(base); ¹H NMR (300 MHz, CDCl₃) δ 7.50-7.30 (m, 15H, 3C₆H₅), 6.44 (d, 1H, J_{1,2} = 3.7 Hz, H-1), 5.05 and 4.66 (AB, each 1H, J = 11.4 Hz, PhCH₂), 4.92 and 4.81 (AB, each 1H, J = 11.6 Hz, PhCH₂), 4.76 and 4.72 (AB, each 1H, J = 11.5 Hz, PhCH₂), 4.23 (dd, 1H, J_{2,3} = 9.8 Hz, H-2), 4.16 (dd, 1H, J_{5,6a} = 5.6 Hz, J_{6a,6b} = 10.8 Hz, H-6a), 4.10 (dd, 1H, J_{5,6b} = 5.6 Hz, H-6b), 4.05 (dd, 1H, H-5), 3.96 (d, 1H, J_{3,4} = 2.7 Hz, H-4), 3.93 (dd, 1H, H-3), 2.15 (s, 3H, CH₃CO), 2.01 (s, 3H, CH₃CO).

1,6-Di-O-acetyl-2,3,4-tri-O-benzyl-α-D-mannopyranose (3c):

 $[\alpha]_D = +29.15$ ° (c 6.21, CHCl₃), R_f = 0.42 (petroleum ether/ethyl acetate, 3/1); EIMS (m/z): 533 (M-1), 443, 191, 91(base); ¹H NMR (300 MHz, CDCl₃) δ 7.40-7.20 (m, 15H, 3C₆H₅), 6.19 (d, 1H, J_{1,2} = 2.0 Hz, H-1), 4.96 and 4.60 (AB, each 1H, J = 10.5 Hz, PhCH₂), 4.78 and 4.72 (AB, each 1H, J = 12.4 Hz, PhCH₂), 4.58 (s, 2H, PhCH₂), 4.33 (m, 2H, H-6a, H-6b), 4.00 (dd, 1H, $J_{3,4} = J_{4,5} = 9.5$ Hz, H-4), 3.90-3.80 (m, 2H, H-3, H-5), 3.74 (dd, 1H, H-2), 2.06 (s, 3H, CH₃CO), 2.00 (s, 3H, CH₃CO); Anal. Calcd. for C₃₁H₃₄O₈ (534.61): C, 69.60; H, 6.41. Found: C, 70.13; H, 6.40.

1,3,6-Tri-O-acetyl-2,4-di-O-benzyl-α/β-D-glucopyranose (α/β: 1.2/1) (4a):

 $R_f = 0.51$ (petroleum ether/ethyl acetate, 3/1); EIMS (m/z): 485 (M-1), 91(base); ¹H NMR (300 MHz, CDCl₃) δ α anomer: 7.37-7.16 (m, 10H, 2C₆H₅), 6.33 (d, 1H, J_{1,2} = 3.4 Hz, H-1), 5.52 (t, 1H, J_{2,3} = J_{3,4} = 9.6 Hz, H-3), 4.75-4.45 (m, 4H, 2 PhCH₂), 4.26 (bd, 2H, H-6a, H-6b), 4.00 (m, 1H, H-5), 3.58 (t, 1H, J_{3,4} = J_{4,5} = 9.5 Hz, H-4), 3.57 (dd, 1H, J_{1,2} = 3.4 Hz, J_{2,3} = 9.6 Hz, H-2), 2.13 (s, 3H, CH₃CO), 2.05 (s, 3H, CH₃CO), 1.99 (s, 3H, CH₃CO); β anomer: 7.37-7.16 (m, 10H, 2C₆H₅), 6.65 (d, 1H, J_{1,2} = 8.1 Hz, H-1), 5.32 (t, 1H, J_{2,3} = J_{3,4} = 9.3 Hz, H-3), 4.71-4.41 (m, 4H, 2 PhCH₂), 3.88 (t, 1H, H-4), 3.71 (m, 1H, H-5), 3.51 (dd, 1H, _{2,3} = 11.3 Hz, H-2), 2.08 (s, 3H, CH₃CO), 2.06 (s, 3H, CH₃CO), 1.90 (s, 3H, CH₃CO).

Dimethyl (6-*O*-acetyl-2,3,4-*O*-tri-benzyl-α-D-glucopyranosyl) phosphonate (5):

 $R_f = 0.28$ (petroleum ether/ethyl acetate/dichloromethane, 1/1/1); EIMS (m/z): 589 (M); HREIMS: 584.2168 (Calcd. for $C_{31}H_{37}O_9P$, 584.2175); ³¹P NMR (121.5 MHz, CDCl₃) δ 24.45; ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.24 (m, 15H, 3C₆H₅), 4.98 and 4.82 (AB, each 1H, J = 10.9 Hz, PhCH₂),), 4.87 and 4.57 (AB, each 1H, J = 8.5 Hz, PhCH₂), 4.80 and 4.68 (AB, each 1H, J = 9.4 Hz, PhCH₂), 4.49 (dd, 1H, $J_{1,2} = 7.0$ Hz, $J_{P,H1} = 11.3$ Hz, H-1), 4.34 (t, 1H, $J_{2,3} = J_{3,4} = 8.6$ Hz, H-3), 4.29 (bs, 1H, H-6a), 4.19 (dd, 1H, $J_{5,6b} = 4.7$ Hz, $J_{6a,6b} = 15.9$ Hz, H-6b), 4.14 (m, 1H, H-5), 3.94(dt, 1H, $J_{P,H2} = 31.2$ Hz, H-2), 3.82 (d, 3H, $J_{P-O-C-H} = 10.6$ Hz, CH₃O), 3.74 (d, 3H, $J_{P-O-C-H} = 10.8$ Hz, CH₃O), 3.45 (t, 1H, $J_{3,4} = J_{4,5} = 8.7$ Hz, H-4), 2.01 (s, 3H, CH₃CO).

Dimethyl (2,3,4-O-tri-benzyl-a-D-glucopyranosyl) phosphonate (6):

 $R_f = 0.33$ (petroleum ether/ethyl acetate: 1/1); EIMS (m/z): 543 (M+1), 451, 329, 181, 91 (base); ³¹P NMR (121.5 MHz, CDCl₃) δ 25.14; ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.26 (m, 15H, 3C₆H₅), 4.93 and 4.81 (AB, each 1H, J = 11.1 Hz, PhCH₂),), 4.83 and 4.63 (AB, each 1H, J = 11.2 Hz, PhCH₂), 4.76 and 4.65 (AB, each 1H, J = 11.3 Hz, PhCH₂), 4.46 (dd, 1H, J_{1.2} = 7.0 Hz, J_{P,H1} = 11.1 Hz, H-1), 4.28 (t, 1H, J_{2.3} = J_{3.4} = 8.5 Hz, H-3), 3.97-3.63 (m, 4H, H-2, H-5, H-6a, H-6b), 3.76 (d, 3H, J_{P-0-C-H} = 10.8 Hz, CH₃O), 3.70 (d, 3H, J_{P-0-C-H} = 10.8 Hz, CH₃O), 3.45 (t, 1H, J_{3.4} = J_{4.5} = 8.7 Hz, H-4).

References:

1 a. Sharon, S. and Lis, H. Scientific Amer. 1993, 268, 82; b. Roslyl, M. B. and Sabine L. F. Chemistry & Biology, 1996, 3, 145.

2 a. Joseph, D. C. and Robert, A. H. Nat. Prod. Rep., **1996**, 31, 151; b. Kurada, M.; Mimaki, Y., and Sashida, Y. Phytochemistry, **1995**, 40, 1071.

3 Riccio, R., Kinnel, R. B., Bifulo G., and Scheuer P. J. Tetrahedron Lett., 1996, 37, 1979.

4 a. Paulsen, H. Angew. Chem., Int. ed. Engl., 1982, 21, 155 and 1990, 29, 823; b. Zuomond, H. M. van der Laan, van der Marel, G. A., van Boon J. H. Carbohydr.

METHYL-GLYCOPYRANOSIDES

Res., 1993, 224, 153; Schmidt, R. R. Angew. Chem., Int. ed. Engl., 1986, 25, 212; d. Toshima, K. and Tatsuta, K., Chem. Ber. 1993, 93, 1053; e. Sinay, P. Pure Appl. Chem., 1991, 63, 519 and 1978, 50, 1437.

5 a. Sim, M. M., Kondo, H., Wong, C. H. J. Am. Chem. Soc., 1993, 115, 2260; b. Ebwein, A. and Schmidt, R. R. Liebigs Ann. Chem., 1988, 675.

6 a. Meuwly, R. and Vasella, A. Helv. Chim. Acta., **1986**, 69, 25; b. Barnes, N. J., Probert, M. A., and Wight, R. H. J. Chem. Soc. Perkin Trans., I, **1996**, 431; c. Vaghefi, M. M., Bernacki, R. J. Dalley, N. K., Wilson, B. E., and Robins, R. J. Med. Chem., **1987**, 30, 1383.

7 a. Zemplem, G., Csuros, Z., and Angyl, Z. Ber. **1937**, *70*, 1848; b. Miyamoto, Y. and Ogawa, S. J. Chem. Soc., Perkin Trans., **1**, **1989**, 1013.

8 a. Ogawa, T. and Sasajima, K. Carbohydr. Res., 1981, 93, 67; b. Paulsen, H. and Lebubn, R. Liebigs Ann. Chem., 1983, 1047; c. Shah, R. N., Baptista, J., Perdomo, G. R., Carver, J. P., and Krepinsky, J. J. J. Carbohydr. Chem., 1987, 6, 645.

(Received in the UK 28th October 1996)