

Stereoselective Synthesis of 2,3-Dihydroxy-4-dimethylamino-5-methoxypentanoic Acid, A Fragment of Calyculins ----- Determination of the Absolute Configuration of Calyculins¹

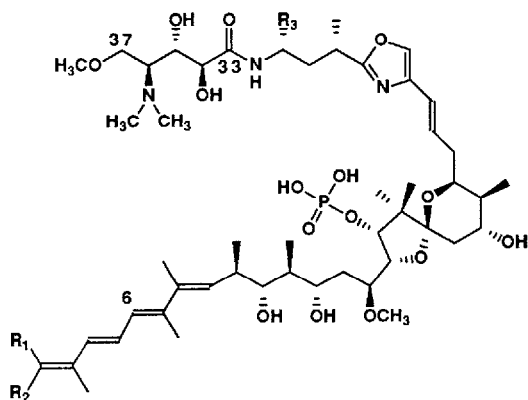
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Abstract 2,3-Dihydroxy-4-dimethylamino-5-methoxypentanoic acid (**2**) with (2R,3R,4R)-configuration has been stereoselectively prepared from (S)-pyroglutaminol (**3**) and revealed to be the enantiomer of the compound derived from calyculins (**1**), which provides the conclusive evidence on the absolute configuration of calyculins

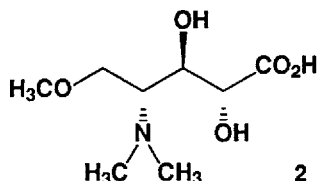
Calyculins, isolated from the marine sponge *Discodermia calyx*,² have unique structures shown in **1** as well as intriguing biological activities, e.g., cytotoxicity, antitumor activity,² inhibition of protein phosphatases,³ etc. Attempts to synthesize calyculins have been actively undertaken by several groups.⁴

The structures of calyculins have been established by X-ray diffraction and spectral studies except their absolute configuration.² Very recently, Matsunaga and Fusetani have disclosed⁵ that hydrolysis of a mixture of calyculins A, B, E, and F afforded 2,3-dihydroxy-4-dimethylamino-5-methoxypentanoic acid, the C₃₃-C₃₇ portion of calyculins, which has been revealed to have (2S,3S,4S)-configuration by the CD spectral studies, and hence the absolute configuration of calyculins should be depicted as the structures **1**.



Calyculins (1)

	R ₁	R ₂	R ₃	
A	CN	H	H	E 6Z isomer of A
B	H	CN	H	F 6Z isomer of B
C	CN	H	CH ₃	G 6Z isomer of C
D	H	CN	CH ₃	H 6Z isomer of D

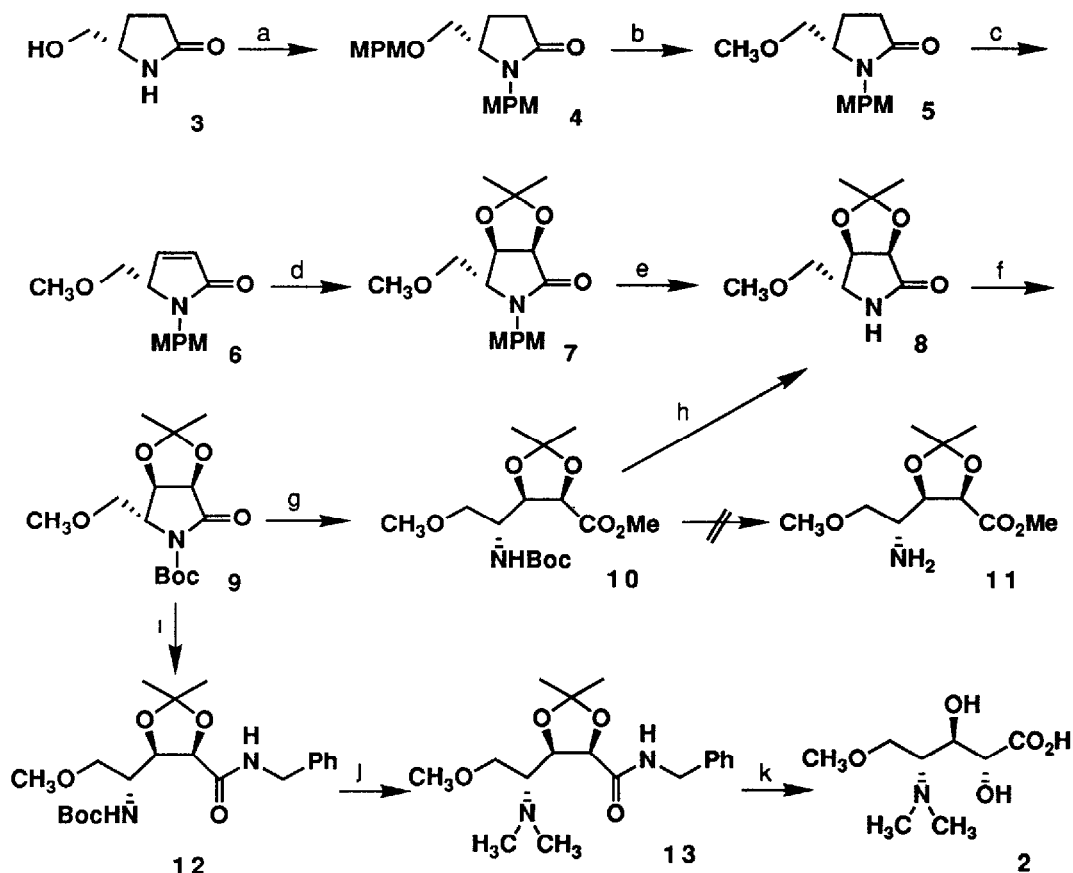


We now wish to report a stereoselective synthesis of 2,3-dihydroxy-4-dimethylamino-5-methoxypentanoic acid (**2**) with (2R,3R,4R)-configuration, which unambiguously determines the absolute configuration of calyculins (**1**)

The synthesis commenced with (S)-pyroglutaminol (**3**),⁶ which was treated with p-methoxybenzyl chloride (MPMCl) under basic conditions to give the bis-MPM-pyrrolidinone **4**, mp 58.5-59.5 °C, $[\alpha]_D^{24} +32.2^\circ$ (c 1.25, CHCl₃), as shown in Scheme 1. Selective removal of the O-MPM function with DDQ⁷ followed by methylation afforded the methoxy derivative **5** as a colorless oil, $[\alpha]_D^{24} +85.5^\circ$ (c 1.2, CHCl₃). Successive treatment of **5** with lithium diisopropylamide (LDA), PhSeCl, and hydrogen peroxide⁸ yielded the α , β -unsaturated lactam **6** as a colorless oil, $[\alpha]_D^{24} -10.4^\circ$ (c 2, CHCl₃). *cis*-Dihydroxylation via catalytic osmylation followed by acetonide formation furnished the acetonide **7**, mp 74-75 °C, $[\alpha]_D^{26} +40.6^\circ$ (c 0.5, CHCl₃), as a single diastereoisomer. Apparently, attack of osmium tetroxide has mainly occurred from the opposite side to the methoxymethyl group.^{8,9} Removal of the N-MPM moiety of **7** was accomplished by use of ceric ammonium nitrate (CAN),¹⁰ while trifluoroacetic acid¹¹ as well as the electrochemical method¹² was not effective at all. The deprotected product **8** was treated with di-*tert*-butyl dicarbonate (Boc₂O) to give the required Boc derivative **9** as a yellow oil, $[\alpha]_D^{24} -92.3^\circ$ (c 1, CHCl₃). Hydrolytic ring-cleavage with lithium hydroxide followed by methyl esterification with trimethylsilyldiazomethane (TMSCHN₂)¹³ quantitatively afforded the ester **10**. Although selective removal of the Boc function easily proceeded with trimethylsilyl triflate (TMSOTf), neutralization after the reaction furnished the cyclized product **8** instead of the required amino acid methyl ester **11**. However, after hydrolytic ring-cleavage of **9**, treatment with benzylamine in the presence of diphenyl phosphorazidate (DPPA, (PhO)₂P(O)N₃)¹⁴ and triethylamine afforded the benzylamide **12**, mp 122-123 °C, $[\alpha]_D^{23} +26^\circ$ (c 0.56, CHCl₃), in 78% yield. The benzylamide **12** smoothly underwent (1) removal of the Boc group with TMSOTf, (2) catalytic, reductive N,N-dimethylation, and (3) reprotection of the diol as the acetonide, giving the N,N-dimethyl benzylamide **13** as a colorless oil, $[\alpha]_D^{24} +50.5^\circ$ (c 0.52, CHCl₃), in 83% yield. Final acid hydrolysis afforded (2R,3R,4R)-2,3-dihydroxy-4-dimethylamino-5-pentanoic acid (**2**), mp 95-99 °C, $[\alpha]_D^{22} +36^\circ$ (c 0.09, EtOH-H₂O/1:1). Its ¹H-NMR spectrum was completely identical with that of the sample from calyculins,⁵ but the sign of the specific rotation¹⁵ and the Cotton effect curve of the CD spectrum were reversed.

This synthesis provides not only the conclusive evidence on the absolute configuration of calyculins (**1**) but also an efficient entry into the total synthesis of the biologically interesting calyculins (**1**).

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Scheme 1

a) MPMCl, NaH, DMSO, rt, 24h, 73% b) DDQ, CH₂Cl₂-H₂O (19 1), rt, 4h, CH₃I, NaH, THF, rt, 1h, 75% c) LDA, THF, -78°C, 30min, PhSeCl, -78°C, 30min, 30% aq H₂O₂, EtOAc, rt, 30min, 49% d) OsO₄, N-methylmorpholine N-oxide, toluene-acetone-water, rt, 12 5h, 2,2-dimethoxypropane, p-TsOH H₂O, rt, 5h, 67% e) CAN, CH₃CN-H₂O (7 3), 0°C, 20min then rt, 1 5h f) Boc₂O, 4-(N,N-dimethylamino)pyridine, CH₃CN, rt, 10h, 45% from 7 g) LiOH, THF-H₂O, 0°C, 30min, TMSCHN₂, benzene-methanol, quant h) TMSOTf, CH₂Cl₂, 0°C, 1h, aq NaHCO₃, 83% i) LiOH, THF-H₂O, 0°C, 30min, PhCH₂NH₂, DPPA, Et₃N, DMF, 0°C, 1h then rt, 41h, 78% j) TMSOTf, CH₂Cl₂, 0°C, 1h, 37% aq HCHO, AcOH, 5% Pd-C, H₂, MeOH, rt, 60h, 2,2-dimethoxypropane, p-TsOH H₂O, rt, 3h, 83% k) 6N aq HCl, MeOH, 130°C, 1 5h, 58%

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- 15 The compound **2** of natural origin shows $[\alpha]_D^{23} -21^\circ$ (c 0.07, EtOH-H₂O/1:1) ⁵

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