

Annulative π -Extension (APEX) of Indoles to Pyrido[1,2-*a*]indoles Using 4-Oxo Peroxides as C4 Units

Xin Wang, Chenhao Lou, Leiyang Lv,* and Zhiping Li*



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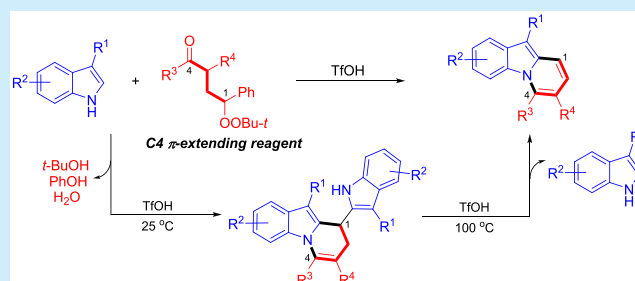


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ABSTRACT: Annulative π -extension (APEX) of 3-substituted indoles to pyrido[1,2-*a*]indoles is developed by using 4-oxo peroxides as π -extending reagents, which are employed as versatile C4 building blocks. This transformation is initiated by Brønsted acid-mediated Hock rearrangement of the peroxy group. Notably, the pyrido[1,2-*a*]indole products are obtained by elimination of the indole moiety from the corresponding dihydropyrido[1,2-*a*]indoles, which could be selectively formed at room temperature.



Nitrogen-fused heterocycles are important structural units prevalently found in the field of medicinal chemistry and material chemistry.¹ Pyrido[1,2-*a*]indoles are known for official candidates that possess a wide spectrum of pharmaceutical activities² and as building blocks to construct various organic materials.³ Owing to their significant applications, various strategies have been developed to synthesize such pyrido[1,2-*a*]indole motifs (Scheme 1). The established methods have been mainly focused on the construction of heteroatomic skeleton, that is, pyrrole ring B⁴ or pyridine ring C⁵ (Scheme 1a). As is well-known, annulative π -extension (APEX)⁶ enables rapid access to fused aromatic molecules from simple aromatic compounds in one step (Scheme 1b), state-of-the-art APEX transformation of indoles represents the most appealing approach for the synthesis of pyrido[1,2-*a*]indoles from the viewpoint of step-economy and availability of starting materials (Scheme 1c). In this context, several π -extending reagents such as diazoenal,⁷ *ortho*-haloarylalkyne,⁸ alkyne,⁹ etc. have been developed to build such valuable scaffolds in the presence of metal catalysts. However, the associated metal residues limit their application in the pharmaceutical field. Therefore, further improvements in terms of reaction sustainability, diversity, and especially developing new π -extending reagents are highly desirable.

A series of cyclocondensations of heteroarenes with 1,4-dicarbonyl compounds have been established for the annulative π -extension.¹⁰ However, this protocol failed to gain popularity due to the low selectivity and efficiency. Recently, we have reported that 4-oxo *tert*-butyl peroxides¹¹ acted as versatile C4 building blocks for the selective synthesis of furans,¹² pyrroles, as well as indoles via Brønsted acid-catalyzed Hock rearrangement of the peroxy group.¹³ Continuing with our interest in the manipulations of organoperoxides, we envisioned that APEX reactions of 3-

substituted indoles with 4-oxo peroxides may enable the synthesis of analogous π -extended pyrido[1,2-*a*]indoles in one step. Herein, we report our efforts in the APEX of indoles to prepare pyrido[1,2-*a*]indoles (Scheme 1d).

We initiated the study by investigating the reaction of 3-methyl-1*H*-indole **1a** and 4-oxo peroxide **2a** to optimize the reaction conditions (Table 1). Initially, the desired pyrido[1,2-*a*]indole product **3a** was not observed at 25 °C; instead, an unexpected dihydropyrido[1,2-*a*]indole **4a** was obtained in 16% yield in the presence of 0.5 equiv of TfOH (entry 1). The structure of **4a** was established by NMR spectroscopy and confirmed by single-crystal X-ray analysis (Figure 1). The yields of **4a** were increased upon raising the loading of TfOH (entries 2 and 3). The product **4a** was obtained in 62% yield at 50 °C with 8% of **3a** formed (entry 4). Further elevating the reaction temperature to 85 °C resulted in the formation of **4a** and **3a** in 35% and 45% yields, respectively (entry 5). To ensure higher chemoselectivity, preparation of **4a** was preferred at 25 °C, although in moderate yield (48%). To our delight, increasing the amount of TfOH to 2.0 equiv and the reaction time to 3 h afforded the target product **3a** in 74% yield, with no **4a** detected (entry 6). Further increasing the loading of TfOH to 3.0 equiv did not improve the reaction efficiency (entry 7). It is worth noting that **3a** was obtained in 76% yield with 3.0 equiv of **1a** and 2.0 or 3.0 equiv of TfOH under 100 °C for 5 h (entries 8 and 9). Lewis acid such as BF₃·OEt₂ afforded **3a** in 49% yield (entry 10), while the reaction was

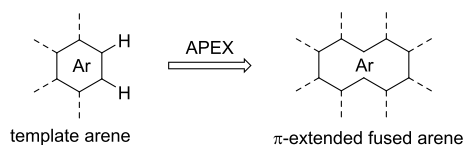
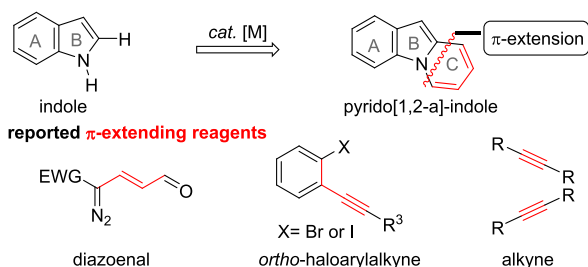
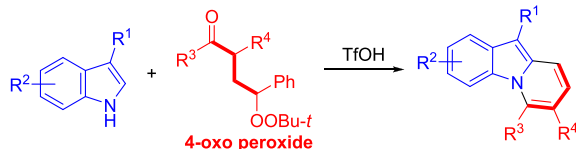
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Scheme 1. Synthetic Strategies for the Pyrido[1,2-*a*]indoles

(a) One ring construction

(b) Concept of Annulative π -Extension (APEX)(c) APEX of indole to pyrido[1,2-*a*]indole(d) Indole APEX with 4-oxo peroxide as π -extending reagent (this work)Table 1. Optimization of the Reaction Conditions^a

entry	acid	T (°C)	3a (%) ^b	4a (%) ^b
1	TfOH (0.5 equiv)	25	N.D.	16
2	TfOH (1.0 equiv)	25	N.D.	30
3	TfOH (1.5 equiv)	25	N.D.	48
4	TfOH (1.5 equiv)	50	8	62
5	TfOH (1.5 equiv)	85	45	35
6 ^c	TfOH (2.0 equiv)	85	74	N.D.
7 ^c	TfOH (3.0 equiv)	85	74	N.D.
8 ^d	TfOH (2.0 equiv)	100	76	N.D.
9 ^d	TfOH (3.0 equiv)	100	76	N.D.
10	BF ₃ ·OEt ₂ (2.0 equiv)	100	49	N.D.
11	FeCl ₃ (2.0 equiv)	100	N.D.	N.D.

^aReaction conditions: **1a** (0.5 mmol), **2a** (0.1 mmol), MeCN (2.0 mL), 24 h, under air, unless otherwise noted. ^bReported yields were based on **2a** and determined by ¹H NMR using CH₂Br₂ as internal standard. ^c**1a** (0.3 mmol), 3 h. ^d**1a** (0.3 mmol), 5 h.

messy and neither **3a** nor **4a** was detected in the case of FeCl₃ (entry 11).

With the optimal reaction conditions in hand, we then evaluated the reaction scope (Schemes 2 and 3). First, the preparation of diverse pyrido[1,2-*a*]indole derivatives was investigated (Scheme 2). A variety of 3-substituted indoles **1**

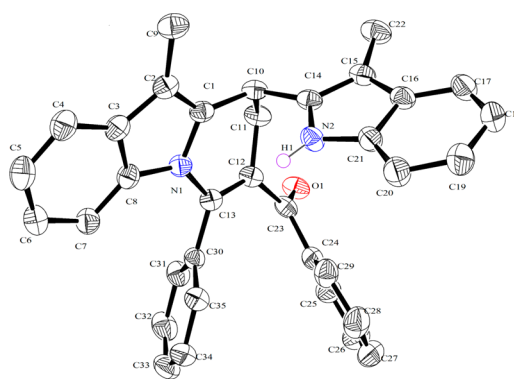
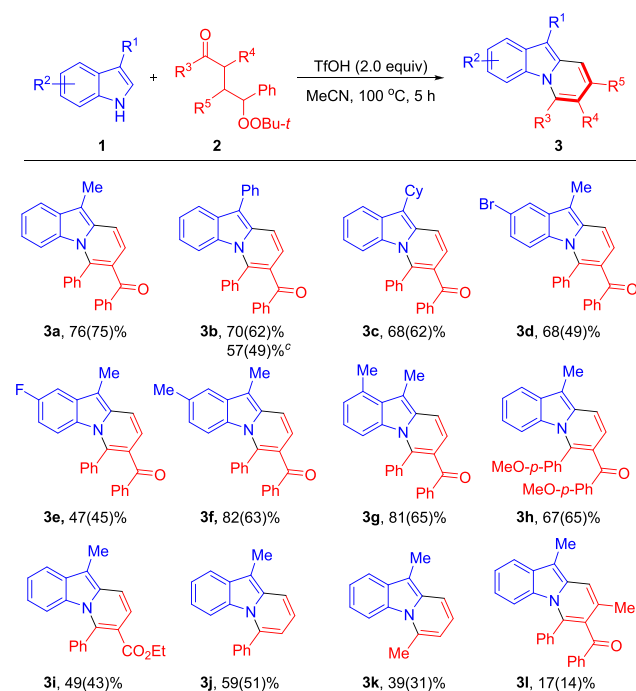


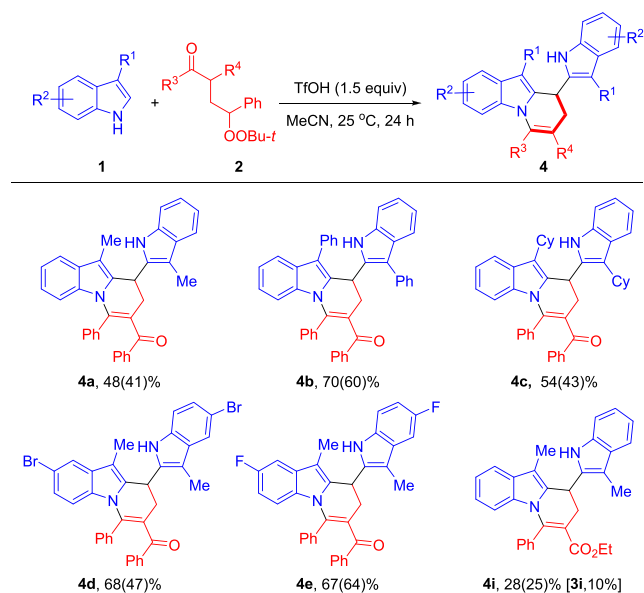
Figure 1. X-ray diffraction of **4a**. Thermal ellipsoids are at the 50% probability level.

Scheme 2. Synthesis of Pyrido[1,2-*a*]indoles **3**^{a,b}

^aReaction conditions: **1** (0.3 mmol), **2** (0.1 mmol), TfOH (0.2 mmol), MeCN (2.0 mL), 100 °C, 5 h, under air. ^bReported yields were based on **2** and determined by ¹H NMR using an internal standard; isolated ones were given in parentheses. ^cReactions were carried out in 1.0 mmol.

reacted smoothly with **2a** to give the pyrido[1,2-*a*]indole products **3a–3g** in 47–82% yields, indicating that the 4-oxo peroxide **2a** plays as an efficient C4 π -extending reagent. When the reaction was carried out in 1.0 mmol, **3b** was obtained in 57% yield. It should be noted that electron-rich indoles are more favorable in the current APEX reaction (**3f** and **3g**). Peroxide derived from β -keto ester also reacted with **1a** to give the corresponding product **3i**. Importantly, monosubstituted (**3j** and **3k**) and trisubstituted (**3l**) annulation products were synthesized by application of the corresponding 4-oxo peroxide **2**, albeit in relatively lower yields. We hypothesized that the two former ones (**3j** and **3k**) are due to the electronic effect, yet the exact reason for the latter (**3l**) is unclear at the current stage.

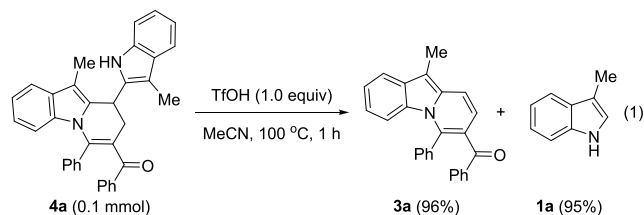
Next, we explored the chemodivergent formation of the dihydropyrido[1,2-*a*]indole products **4** in the presence of **1.5**

Scheme 3. Synthesis of Dihydropyrido[1,2-*a*]indole 4^{a,b}

^aReaction conditions: 1 (0.5 mmol), 2 (0.1 mmol), TfOH (0.15 mmol), solvent (2.0 mL), 25 °C, 24 h, under air. ^bReported yields were based on 2 and determined by ¹H NMR using an internal standard; isolated ones were given in parentheses.

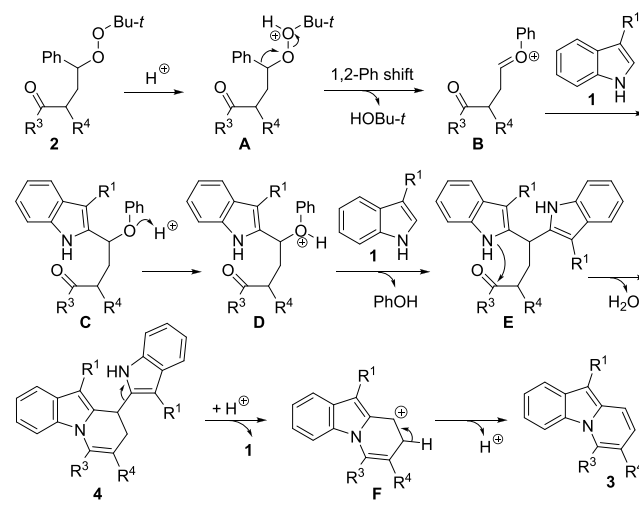
equiv of TfOH at 25 °C (Scheme 3). To our delight, 4a–4e were obtained selectively under the standard conditions in moderate to good yields. In contrast, 4i was obtained in 28% yield along with 10% of 3i when peroxide that derived from β -keto ester was tested with 1a under the same reaction conditions.

A control experiment was carried out to gain the relationship between dihydropyrido[1,2-*a*]indole 4 and pyrido[1,2-*a*]indole 3. For example, when 4a was treated with TfOH (1.0 equiv) in 100 °C for 1 h, both 3a and 3-methyl-1*H*-indole 1a were obtained almost quantitatively (eq 1). This result clearly demonstrates that the annulative π -extension product 3 is formed via decomposition of the corresponding dihydropyrido[1,2-*a*]indole 4.



On the basis of the above experiment and literature reports,^{11–13} a tentative mechanism was proposed (Scheme 4). Initially, the protonation of the peroxide 2 gives the intermediate A, which undergoes the Hock rearrangement (1,2-migration of the phenyl group) to generate the key oxonium intermediate B. Then addition of electron-rich 1a toward B generates the intermediate C, which is protonated and substituted by another molecule of indole to form intermediate E. Subsequent intramolecular cyclocondensation of E delivers the dihydropyrido[1,2-*a*]indole product 4. Finally, high temperature (100 °C) and acid¹⁴ facilitate 4 to rearomatize to afford the pyrido[1,2-*a*]indole product 3 through intermediate F by release of 1.

Scheme 4. A Tentative Reaction Mechanism



In conclusion, we have demonstrated 4-oxo *tert*-butyl peroxides as new C4 π -extending reagents in the acid mediated step-economy annulation of indoles to afford *N*-fused conjugated products. This chemodivergent transformation proceeded in a stepwise manner; that is, the dihydropyrido[1,2-*a*]indoles could be isolated at 25 °C, which readily rearomatized to form the pyrido[1,2-*a*]indoles promoted via acid at high temperature by releasing the indole moiety.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c02062>.

Detailed experimental procedures and compound characterization data, single-crystal X-ray structure of 4a, and NMR spectra of all new compounds (PDF)

Accession Codes

CCDC 2083477 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Authors

Leiyang Lv – Department of Chemistry, Renmin University of China, Beijing 100872, China; orcid.org/0000-0002-2850-8952; Email: lvleiyang2008@ruc.edu.cn

Zhiping Li – Department of Chemistry, Renmin University of China, Beijing 100872, China; orcid.org/0000-0001-7987-279X; Email: zhipingli@ruc.edu.cn

Authors

Xin Wang – Department of Chemistry, Renmin University of China, Beijing 100872, China

Chenhao Lou – Department of Chemistry, Renmin University of China, Beijing 100872, China

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.orglett.1c02062>

Notes

The authors declare no competing financial interest.

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