Article

Subscriber access provided by Iowa State University | Library

Electrocatalytic Tandem Synthesis of 1,3-Disubstituted Imidazo[1,5-a]quinolines via Sequential Dual Oxidative C(sp3)-H Amination in Aqueous Medium

Peng Qian, Zicong Yan, Zhenghong Zhou, Kangfei Hu, Jiawei Wang, Zhibin Li, Zhenggen Zha, and Zhiyong Wang

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.8b03014 • Publication Date (Web): 16 Jan 2019 Downloaded from http://pubs.acs.org on January 17, 2019

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Electrocatalytic Tandem Synthesis of 1,3-Disubstituted Imidazo[1,5-*a*]quinolines via Sequential Dual Oxidative C(sp3)–H Amination in Aqueous Medium

Peng Qian,[†] Zicong Yan,[†] Zhenghong Zhou, Kangfei Hu, Jiawei Wang, Zhibin Li, Zhenggen Zha* and Zhiyong Wang*

Hefei National Laboratory for Physical Sciences at Microscale, CAS Key Laboratory of Soft Matter Chemistry & Center for Excellence in Molecular Synthesis of Chinese Academy of Sciences, Collaborative Innovation Center of Suzhou Nano Science and Technology & School of Chemistry and Materials Science, University of Science and Technology of China, Hefei, Anhui 230026, P. R. China.

[†]These authors contributed equally to this work

E-mail:zwang3@ustc.edu.cn

zgzha@ustc.edu.cn



ABSTRACT: An NH₄I-mediated tandem electrosynthesis of 1,3-disubstituted imidazo[1,5-a]quinolines was developed from readily available starting materials in aqueous medium, affording a variety of 1,3-disubstituted imidazo[1,5-a]quinolines with good to excellent yields.

Introduction

N-containing compounds are widely existed in pharmaceutical, agrochemical and material industries.¹ Methods development for direct C(sp3)–H/N–H oxidative coupling to construct C–N bonds in a more efficient and sustainable manner has been pursued by organic chemists.^{2, 3} The rapid growth in direct amination of C(sp3)–H bonds catalysed by transition-metals has been witnessed in recent years.^{2d-e} However, the residue of transition metal catalysts on drugs synthesis leads to concern about human healthy. Afterwards, some metal-free amination reactions³ were reported in the presence of stoichiometric chemical oxidants. Nevertheless, the usage of stoichiometric chemical oxidants inevitably leads to waste, undesired reactions and low atom economy. Therefore, it remains to be desirable and challenging for organic chemists to construct C–N bonds under metal catalysts and chemical oxidants free conditions.

Organic electrochemistry provides the possibility for C–N formation because it employs electrons as reagents.^{4, 5} In general, the electrochemical reaction can be performed at room temperature without the use of chemical oxidants, which makes it safety and low cost for large-scale industrial application. Recently, organic electrochemistry has undergone a renaissance. Major elegant contributions have been made by Moeller,⁶ Xu,⁷ Yoshida⁸ Lei,⁹ Mei,¹⁰ and Ackermannm¹¹ for C(sp2)–H amination reactions. However, the direct intermolecular C(sp3)–H amination cyclization of reactions have not been explored yet.

1,3-Disubstituted imidazo[1,5-*a*]quinolines as one of the important scaffolds in N-containing compounds, are widely present in pharmaceutical agent molecules and optoelectronic materials.¹² However, the reported methods¹³ usually suffered from the metal catalysts, chemical oxidants and high temperature. To overcome these issues and in continuation of our recent work on electrochemistry in the presence of iodine,¹⁴ we herein report a tandem electrosynthesis of 1,3-disubstituted imidazo[1,5-*a*]quinolines from readily available amines and amino acids under mediation of NH₄I in aqueous medium at room temperature in the absence of metal and external chemical oxidants. To the best of our knowledge, this work should represent one of the most simple, green and efficient approach for the construction of 1,3-disubstituted imidazo[1,5-*a*]quinolines under mild conditions.

Results and Discussion

Initially, the model reaction of ethyl 2-(quinolin-2-yl)acetate **1a** with benzylamine **2a** was electrolyzed in an undivided cell at a constant current density of 15 mA/cm², while KI was used as the supporting salt in the co-solvent of CH₃CN/H₂O (5 : 1). Gratifyingly, the desired product was obtained with a yield of 66% (Table 1, entry 1). Inspired by this result, the supporting salts were firstly investigated. All iodide salts could conduct this reaction smoothly while either NH₄Br or LiClO₄ hardly afforded the desired products (entries 5 and 6). Among these tested iodide salts,

Table 1. Optimization of the Reaction Conditions^a

		+ Ph [^] NH ₂ Et	electrolysis rt		OEt
	1a	2a		3aa	
entry	electrolyte	solvent	anode/cathode	j (mA/cm ²)	yield $(\%)^b$
1	KI	CH ₃ CN/H ₂ O	Pt/Pt	15	66
2	Me ₄ NI	CH ₃ CN/H ₂ O	Pt/Pt	15	65
3	<i>n</i> -Bu ₄ NI	CH ₃ CN/H ₂ O	Pt/Pt	15	82
4	NH ₄ I	CH ₃ CN/H ₂ O	Pt/Pt	15	95
5	NH ₄ Br	CH ₃ CN/H ₂ O	Pt/Pt	15	trace
6	LiClO ₄	CH ₃ CN/H ₂ O	Pt/Pt	15	n. d.
7 ^c	NH ₄ I	CH ₃ CN/H ₂ O	Pt/Pt	15	65
8	NH ₄ I	CH ₃ CN/H ₂ O	Pt/Pt	20	83
9	NH ₄ I	CH ₃ CN/H ₂ O	Pt/Pt	10	92
10	NH ₄ I	CH ₃ CN/H ₂ O	C/Pt	15	65
11	NH ₄ I	CH ₃ CN/H ₂ O	Pt/C	15	71
12	NH ₄ I	EtOH	Pt/Pt	15	88
13	NH ₄ I	DMF	Pt/Pt	15	89

14	NH ₄ I	CH ₃ CN	Pt/Pt	15	91
15 ^d	NH4I	CH ₃ CN/H ₂ O	Pt/Pt	15	93
16 ^e	NH4I	CH ₃ CN/H ₂ O	Pt/Pt	15	91
17 ^f	NH ₄ I	CH ₃ CN/H ₂ O	Pt/Pt	15	95

^{*a*}Reaction conditions: **1a** (0.3 mmol), **2a** (1.2 mmol), electrolyte (0.3 mmol), $CH_3CN/H_2O = 5:1$ (3 mL); the electrolysis was conducted in an undivided cell at room temperature. ^{*b*}Yields of the isolated products. ^{*c*}The loading of NH₄I was 20 mol %. ^{*d*}The ration of CH₃CN/H₂O was 4:1. ^{*e*}Under O₂. ^{*f*}Under N₂. n. d. = not detected.

NH₄I gave the best result (entries 1–4). When we decreased the loading of NH₄I to 20 mol %, the reaction was still carried smoothly to afford the desired product with a moderate yield (entry 7). A sacrificial yield was observed when the current density was increased or decreased (entries 8 and 9). The experimental results showed that the Pt electrode was the optimal electrode materials both for anode and cathode (entries 10 and 11). Moreover, changing the solvent to EtOH, DMF, CH₃CN or the ration of CH₃CN/H₂O to 4 : 1 had little effect on the yield (entries 12–15). When the reaction was performed under an oxygen atmosphere or nitrogen atmosphere, the desired product can be obtained with yields of 91% and 95%, respectively (entries 16 and 17). These results indicated that the reaction atmosphere also had little influence on the yields. After investigation in detail, the optimal electrolytic conditions were described as entry 4 of table 1.

With the optimized electrolytic conditions established, we investigated the scope of aromatic benzylamines first under the standard conditions. As shown in Table 2, all of the substrates could tolerate this reaction well, providing the desired products **3aa–3av** with 78–96% yields. The electronic effect of the substituents on the phenyl ring of the benzylamines had little influence on the reaction. Both electron-donating and electron-withdrawing substitutions afforded the desired products with excellent yields (**3ab–3ag** vs **3ap** and **3aq**). Halide substituents, such as F, Cl, Br, were also proceeded smoothly to afford the desired products with 78–93% yields (**3ah–3ao**). When the ring-fused and the multi-substituted benzylamines were employed as substrates, the corresponding products can be obtained with 89% and 82% yields (**3ar** and **3as**), respectively. When the phenyl ring of the benzylamines was replaced by the heterocyclic rings, these heterocyclic rings can be compatible with the reaction conditions, giving the corresponding

product in 84–95% yields (**3at–3av**). However, for the aliphatic amines with long chain, no desired product was observed (**3aw**).

Table 2. The Scope of Amines^a

	NH, + R ¹ ∕NH ₂ <u>CH₃(</u> Et Pt/Pt J 2a-2w	4I (1 equiv) <u>CN/H₂O (5/1)</u> = 15 mA/cm ² rt	N OEt R ¹ 3aa-3aw
entry	R ¹	product	yield $(\%)^b$
1	Ph (2a)	3aa	95
2	4-Me-Ph (2b)	3ab	90
3	3-Me-Ph (2c)	3ac	90
4	2-Me-Ph (2d)	3ad	89
5	4-OMe-Ph (2e)	3ae	91
6	2-OMe-Ph (2f)	3af	96
7	3,4-CH ₂ O-Ph (2g)	3ag	91
8	4-F-Ph (2h)	3ah	87
9	3-F-Ph (2i)	3ai	90
10	2-F-Ph (2j)	3aj	86
11	4-Cl-Ph (2k)	3ak	89
12	3-Cl-Ph (2l)	3al	91
13	2-Cl-Ph (2m)	3am	88
14	4-Br-Ph (2n)	3an	78
15	2-Br-Ph (2o)	3 ao	93
16	4- CF ₃ -Ph (2p)	3ap	96

17	3-CF ₃ -Ph (2q)	3aq	90
18	1-naphthyl (2r)	3ar	89
19	2-Cl-4-F-Ph(2s)	3as	82
20	2-furyl (2t)	3at	87
21	2-thienyl (2u)	3au	84
22	2-pryidyl (2 v)	3av	95
23	Bn (2w)	3aw	n.d.

^{*a*}Reaction conditions: **1a** (0.3 mmol), **2a-2w** (1.2 mmol), NH₄I (0.3 mmol), CH₃CN/H₂O = 5:1 (3 mL); the electrolysis was conducted in an undivided cell at room temperature. ^{*b*}The isolated yields after column chromatography. n. d. = not detected.

Subsequently, a variety of ethyl 2-(quinolin-2-yl)acetate derivatives were also examined. As shown in scheme 1, both electron-donating and electron-withdrawing groups on the quinolinyl ring had influence on the reaction (3ba-3ha). Generally, the withdrawing groups of R³ gave superior results to the donating groups of R³ (3ha vs 3ea-3ga). When R² was varied from COOEt to COOMe or benzoyl group, the reaction also worked well, affording the corresponding products with 85% and 65% yields, respectively (3ia and **3ja**). Furthermore, ethyl 2-(isoquinolin-1-yl)acetate 1k and ethyl 2-(pyridin-2-yl)acetate 1l, in which the quinolinyl ring was changed, were also good substrates in this reaction, providing the desired products with good yields (3ka and 3la).

Scheme 1. The Scope of 1





^{*a*}Reaction conditions: **1b-11** (0.3 mmol), **2a** (1.2 mmol), NH₄I (0.3 mmol), CH₃CN/H₂O = 5:1 (3 mL); the electrolysis was conducted in an undivided cell at room temperature; the isolated yields after column chromatography.

To demonstrate the practicality of this reaction, a gram scale experiment was conducted (scheme 2a). Gratifyingly, the desired product **3aa** can be obtained with good yields. It was noted that an unexpected transfer esterification cyclizaiton product was obtained when the co-solvent was replaced by the methanol (scheme 2b). What is more, this electrocatalytic reaction can be also applied to the synthesis of polysubstituted oxazoles **3ma** in aqueous medium (scheme 2c), which further enhanced the practicality of this reaction.

Scheme 2. Gram-Scale and Transfer Esterification Cyclization Reaction.



Recently, utilization of α -amino acids for the synthesis of heterocyclic compounds are increasing due to its low cost and readily available. Then we further extended the substrate scope by the use of the amino acids in place of amines with a modified condition (scheme 3). As expected, both aromatic and alkyl amino acids could proceed the reaction smoothly to afford the corresponding

products with moderate yields (**3aa–3aB**). The wide scope of amino acids provided a great practicality of this electrocatalytic reaction.

Scheme 3. Electrosynthesis of 1,3-Disubstituted Imidazo[1,5-a]quinolines from Amino Acids.



R⁴ = ^{*n*}Pr, **3ay**, **41%** R⁴ = ^{*i*}Pr, **3az**, **38%**

R⁴ = ^sBu, **3aA**, **39%** R⁴ = ⁱBu, **3aB**, **48%** To gain a better understanding of the reaction mechanism, some control experiments were carried out (scheme 4). First, the reaction was intercepted to investigate the possible reaction intermediate. Actually, both amine 7 and desired product **3aa** could be observed when the model substrates were electrolyzed for 12 h. The imine 7 could be further transformed into 3aa with a prolonged electrolysis time (scheme 4a). The imine intermediate could be further confirmed by the forming of ethyl 2-((4-methoxyphenyl)imino)-2-(quinolin-2-yl)acetate 7' (scheme 4b). These results indicate that the imine 7 was a key intermediate under our conditions. Furthermore, when the secondary amine, such as piperidine, was used to replace benzylamine, an aminated product 5' was obtained with 80% yield (scheme 4c). This result implies that the aminated intermediate may be also involved under our conditions, which could be further oxidized into the imine intermediate 7. On the other hand, the imine intermediate 7 could be also formed via the condensation reaction between 6 and 2a. When the compound 6 was employed under the standard conditions, only 42%yield of the desired product was obtained and the reaction mixture was messy (scheme 4d). This result indicates that the compound 6 was not the major intermediate in the reaction. When molecular iodine was employed as oxidant, the desired product and intermediate 7 were obtained with the yield of 32% and 50%, respectively (scheme 4e). These results suggest that the in situ electrogenerated molecular iodine should be the active specie in the reaction.

Scheme 4. Control Experiments for the Reaction



Based on the above-mentioned results and the previous reports,¹³ a possible reaction mechanism was proposed and depicted in scheme 5. First, anodically *in situ* generated molecular iodine reacts with substrate **1a** to generate the iodinated intermediate **4**. The reaction between intermediate **4** and benzylamine **2a** gives intermediate **5**. Subsequently, the intermediate **5** could be sequentially oxidized in the presence of molecular iodine to yield intermediate **8**, which could be further converted to the desired product **3aa** through the tandem cyclization process. Meanwhile, the proton is reduced on the cathode surface with the liberation of hydrogen gas. In the case of phenylglycine as the coupling partner, it reacts with the iodinated intermediate **4** to afford intermediate **9**, followed by the oxidation of molecular iodine to give intermediate **10**. The intermediate **11** is unstable and easily undergoes decarboxylative/oxidative amination/aromatization process to give the final product **3aa**.

Scheme 5. Proposed Reaction Mechanism for the Formation of 3aa



In summary, we developed a facile and efficient approach for the electrosynthesis of 1,3-disubstituted imidazo[1,5-a]quinolines in aqueous medium under the metal and chemical oxidants free conditions. A wide scope of substrates, readily available starting materials and gram-scale experiments made this reaction practicality.

Experimental Section

General Information: All products were characterized by ¹H NMR and ¹³C{¹H}NMR, using TMS as an internal reference (¹H NMR: 400MHz, ¹³C{¹H}NMR: 100MHz). HRMS (ESI) data were recorded on a Q-TOF Premier. Commercial reagent and solvents were used without purification unless otherwise indicated.

Preparation of Substrates: Substrates 11, 1m and 2 are commercially available. Substrates 1f, 1i and 1j were prepared according to the previously reported procedures.¹⁶⁻¹⁸ Other substrates 1 (1a-1e, 1g, 1h and 1k) were synthesized by using our previous literature procedure.¹⁹

General Procedures for the Synthesis of 1: A mixture of 7.08 g diisopropylamine (9.8 mL, 70 mmol) and 10 mL of dried THF was added to a three-necked flask by syringe under N₂. After cooling down to -78 °C, 28.4 mL of *n*-butyl lithium (2.4 M in hexane, 68 mmol) was slowly added. The mixture was stirred for 30 minutes. Then 3.5 g of 2-methyl quinoline (3.3 mL, 24 mmol) was

slowly added. The colour turned to orange rapidly then dark-brown. After 30 minutes, 10.6 g of diethyl carbonate (11 mL, 90 mmol) was added and the mixture was stirred for 2 hours. The reaction was quenched by 10 mL of water followed by extracted by EtOAc. The combined organic layers were dried over Na₂SO₄ and evaporated in vacuo. The residue was purified by silica gel column chromatography (petroleum ether / ethyl acetate = 20:1) to give **1a** as a light yellow oil: (3.8 g, 74%); ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.14 (d, J = 8.4 Hz, 1H), 8.07 (d, J = 8.7 Hz, 1H), 7.82–7.80 (m, 1H), 7.73–7.68 (m, 1H), 7.55–7.51 (m, 1H), 7.44 (d, J = 8.4 Hz, 1H), 4.20 (q, J = 7.1 Hz, 2H), 4.04 (s, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 170.5, 154.8, 147.8, 136.7, 129.6, 129.1, 127.5, 127.1, 126.4, 121.8, 61.1, 44.8, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₄NO₂ 216.1025; Found 216.1023.

Synthesis of 1b: The title compound was prepared according to the general working procedure for the synthesis of **1** using 6-chloro-2-methylquinoline (10 mmol, 1.77 g) as the starting material in 70% (1.74 g) yield as yellow solid. m.p. 101–103 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.05 (d, J = 8.5 Hz, 1H), 8.00 (d, J = 9.0 Hz, 1H), 7.79–7.78 (m, 1H), 7.65–7.62 (m, 1H), 7.46 (d, J = 8.5 Hz, 1H), 4.20 (q, J = 7.1 Hz, 2H), 4.03 (s, 2H), 1.27 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 170.3, 155.2, 146.1, 135.8, 132.2, 130.7, 130.6, 127.6, 126.2, 122.7, 61.2, 44.7, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₃ClNO₂ 250.0635; Found 250.0634. **Synthesis of 1c:** The title compound was prepared according to the general working procedure for the synthesis of **1** using 6-bromo-2-methylquinoline (10 mmol, 2.21 g) as the starting material in 75% (2.20 g) yield as yellow solid. m.p. 104–106 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.05 (d, J = 8.5 Hz, 1H), 7.97–7.93 (m, 2H), 7.78–7.75 (m, 1H), 7.46 (d, J = 8.5 Hz, 1H), 4.20 (q, J = 7.1 Hz, 2H), 4.03 (s, 2H), 1.27 (t, J = 7.2 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 170.2, 155.3, 146.3, 135.7, 133.1, 130.8, 129.6, 128.2, 122.7, 120.3, 61.2, 44.7, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₃BrNO₂ 294.0130; Found 294.0131.

Synthesis of 1d: The title compound was prepared according to the general working procedure for the synthesis of 1 using 6-fluoro-2-methylquinoline (10 mmol, 1.61 g) as the starting material in 60% (1.40 g) yield as yellow oil. ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.08-8.03$ (m, 2H), 7.48–7.39 (m, 3H), 4.20 (q, J = 7.1 Hz, 2H), 4.01 (s, 2H), 1.26 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 170.4$, 160.4 (d, J = 246.5 Hz), 154.2 (d, J = 2.7 Hz), 144.9, 136.0 (d,

J = 5.2 Hz), 131.6 (d, J = 9.0 Hz), 127.6 (d, J = 9.8 Hz), 122.6, 119.7 (d, J = 25.5 Hz), 110.5 (d, J = 21.7 Hz), 61.2, 44.6, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₃FNO₂ 234.0930; Found 234.0929.

Synthesis of 1e: The title compound was prepared according to the general working procedure for the synthesis of **1** using 6-methoxy-2-methylquinoline (10 mmol, 1.73 g) as the starting material in 65% (1.59 g) yield as yellow solid. m.p. 45–47 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.02$ (d, J = 8.5 Hz, 1H), 7.95 (d, J = 9.2 Hz, 1H), 7.39–7.33 (m, 2H), 7.06–7.05 (m, 1H), 4.19 (q, J = 7.1 Hz, 2H), 3.99 (s, 2H), 3.91 (s, 3H), 1.26 (t, J = 7.2 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 170.7$, 157.7, 152.2, 143.9, 135.5, 130.4, 128.0, 122.3, 122.0, 105.1, 61.1, 55.5, 44.5, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₄H₁₆NO₃ 246.1130; Found 246.1128.

Synthesis of 1g: The title compound was prepared according to the general working procedure for the synthesis of 1 using 6-(*tert*-butyl)-2-methylquinoline (10 mmol, 1.99 g) as the starting material in 62% (1.68 g)yield as yellow oil. ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.09$ (d, J = 8.4 Hz, 1H), 8.00 (d, J = 8.9 Hz, 1H), 7.81–7.78 (m, 1H), 7.71–7.70 (m, 1H), 7.40 (d, J = 8.4 Hz, 1H), 4.18 (q, J = 7.1 Hz, 2H), 4.02 (s, 2H), 1.41 (s, 9H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 170.6$, 154.1, 149.2, 146.4, 136.6, 128.6, 128.6, 126.8, 122.5, 121.7, 61.1, 44.8, 34.9, 31.2, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₇H₂₂NO₂ 272.1651; Found 272.1651.

Synthesis of 1h: The title compound was prepared according to the general working procedure for the synthesis of **1** using 2-methyl-6-(trifluoromethyl)quinoline (10 mmol, 2.11 g) as the starting material in 70% (1.98 g) yield as yellow solid. m.p. 105–107 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.22$ (d, J = 8.5 Hz, 1H), 8.18–8.12 (m, 2H), 7.88–7.85 (m, 1H), 7.54 (d, J = 8.5 Hz, 1H), 4.21 (q, J = 7.1 Hz, 2H), 4.07 (s, 2H), 1.27 (t, J = 7.2 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 170.1$, 157.3, 148.8, 137.3, 130.3, 128.3 (q, J = 32.4 Hz), 126.0, 125.5 (q, J = 4.4 Hz), 125.3 (q, J = 3.1 Hz), 124.0 (q, J = 269.9 Hz), 123.1, 61.3, 44.8, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₄H₁₃F₃NO₂ 284.0898; Found 284.0899.

Synthesis of 1k: The title compound was prepared according to the general working procedure for the synthesis of 1 using 1-methylisoquinoline (10 mmol, 1.43 g) as the starting material in 64% (1.38 g) yield as yellow oil. ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.47$ (d, J = 5.7 Hz, 1H), 8.10–8.07 (m, 1H), 7.84 (d, J = 8.1 Hz, 1H), 7.71–7.67 (m, 1H), 7.64–7.59 (m, 2H), 4.35 (s, 2H),

 4.19 (q, J = 7.2 Hz, 2H), 1.23 (t, J = 7.1 Hz, 3H); ${}^{13}C{}^{1}H{}NMR$ (CDCl₃, 100 MHz, ppm): δ = 170.4, 154.6, 141.9, 136.4, 130.2, 127.6, 127.4, 127.4, 125.2, 120.5, 61.2, 42.2, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₃H₁₄NO₂ 216.1025; Found 216.1025.

Representative Procedures for the Synthesis of 1,3-Disubstituted Imidazo[1,5-*a*]quinolines from Amines (method A): An undivided cell was equipped with a magnet stirrer, platinum-plate $(1.0 \times 1.0 \text{ cm}^2)$ electrode as the working electrode and counter electrode. In the electrolytic cell, a mixture of ethyl 2-(quinolin-2-yl)acetate derivatives 1 (0.3 mmol), amines 2 (1.2 mmol), NH₄I (0.3 mmol, 43.5 mg), CH₃CN/H₂O= 5:1 (3 mL) was allowed to stir and electrolyze at a constant current conditions (15 mA/cm²) under room temperature until the reaction intermediate finished (TLC analysis). Then the solvent was removed with a rotary evaporator and the residue was purified by column chromatography on silica gel to afford the desired product. The product was dried under high vacuum for at least 0.5 h before it was weighed and characterized by NMR spectroscopy.

Representative Procedures for the Synthesis of 1,3-Disubstituted Imidazo[1,5-*a*]quinolines from Amino Acids (method B): An undivided cell was equipped with a magnet stirrer, platinum-plate (1.0 x 1.0 cm²) electrode as the working electrode and counter electrode. In the electrolytic cell, a mixture of ethyl 2-(quinolin-2-yl)acetate 1a, amino acids 2' (1.2 mmol), *n*-Bu₄NI (0.3 mmol, 110.8 mg), CH₃CN/NaHCO₃ (Sat) = 5:1 (3 mL) was allowed to stir and electrolyze at a constant current conditions (10 mA/cm²) under 50 °C (water bath) until the ethyl 2-(quinolin-2-yl)acetate 1a finished (TLC analysis). Then the solvent was removed with a rotary evaporator and the residue was purified by column chromatography on silica gel to afford the desired product. The product was dried under high vacuum for at least 0.5 h before it was weighed and characterized by NMR spectroscopy.

Gram-Scale Synthesis of 3aa

An undivided cell was equipped with a magnet stirrer, platinum-plate (1.5 x 1.5 cm²) electrode as the working electrode and counter electrode. In the electrolytic cell, a mixture of ethyl 2-(quinolin-2-yl)acetate **1a** (4.8 mmol), benzylamine **2a** (19.2 mmol) NH₄I (4.8 mmol, 696 mg), CH₃CN/H₂O= 5:1 (48 mL) was allowed to stir and electrolyze at a constant current conditions (15 mA/cm²) under room temperature until the reaction intermediate finished (about 3 d). Then the solvent was removed with a rotary evaporator and the residue was recrystallized by EtOAc to afford the desired product (1.21 g, 80% yield).

Transfer Esterification Cyclizaiton Reaction

An undivided cell was equipped with a magnet stirrer, platinum-plate $(1.0 \times 1.0 \text{ cm}^2)$ electrode as the working electrode and counter electrode. In the electrolytic cell, a mixture of ethyl 2-(quinolin-2-yl)acetate **1a** (0.3 mmol), benzylamine **2a** (1.2 mmol), *n*-Bu₄NI (0.3 mmol, 110.8 mg), MeOH (3 mL) was allowed to stir and electrolyze at a constant current conditions (15 mA/cm²) under room temperature until the reaction intermediate finished (TLC analysis). Then the solvent was removed with a rotary evaporator and the residue was purified by column chromatography on silica gel to afford the desired product. The product was dried under high vacuum for at least 0.5 h before it was weighed and characterized by NMR spectroscopy.

Ethyl 1-phenylimidazo[1,5-*a*]*quinoline-3-carboxylate* (**3aa**)^{13a}

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 95% yield, (90 mg), m.p. 167–168 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.20$ (d, J = 9.5 Hz, 1H), 7.75–7.73 (m, 1H), 7.66–7.63 (m, 2H), 7.57–7.48 (m, 4H), 7.42–7.38 (m, 2H), 7.28–7.23 (m, 1H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.7$, 142.5, 134.6, 133.0, 132.1, 130.0, 129.9, 129.0, 128.9, 128.4, 126.3, 125.9, 125.4, 123.4, 117.7, 117.6, 60.6, 14.7.

Ethyl 1-(p-tolyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ab**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 90% yield, (89 mg) , m.p. 160–163 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.19$ (d, J = 9.4 Hz, 1H), 7.74–7.72 (m, 1H), 7.55–7.51 (m, 3H), 7.42–7.38 (m, 2H), 7.34–7.32 (m, 2H), 7.29–7.24 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 2.48 (s, 3H),1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.7$, 142.7, 140.0, 134.5, 132.2, 130.0, 129.8, 129.6, 128.9, 128.3, 126.2, 125.8, 125.4, 123.2, 117.7, 117.6, 60.6, 21.6, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₂ 331.1447; Found 331.1445.

Ethyl 1-(m-tolyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ac**)

The title compound was prepared according to the general working procedure A (28 h, 52.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 90% yield, (89 mg) , m.p. 139–141 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.5 Hz, 1H), 7.74–7.72 (m, 1H), 7.55 (d, J = 8.6 Hz, 1H), 7.50 (s, 1H), 7.44–7.36 (m, 5H), 7.29–7.25 (m, 1H) 4.52 (q, J = 7.1 Hz, 2H), 2.44 (s, 3H),1.48 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.6, 142.6, 138.7, 134.4, 132.7, 132.0, 130.5, 130.5, 128.8, 128.6, 128.2, 126.8, 126.2, 125.7, 125.3, 123.2, 117.5, 60.5, 21.3, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₂ 331.1447; Found 331.1447.

Ethyl 1-(o-tolyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ad**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 89% yield, (88 mg), m.p. 142–145 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.22 (d, J = 9.5 Hz, 1H), 7.73 (d, J = 7.8 Hz, 1H), 7.51–7.47 (m, 2H), 7.42–7.35 (m, 4H), 7.28–7.21 (m, 2H), 4.51 (q, J = 7.1 Hz, 2H), 2.02 (s, 3H),1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.7, 141.8, 138.5, 134.1, 133.0, 132.3, 130.6, 130.4, 130.3, 128.9, 126.5, 126.2, 125.8, 125.2, 123.1, 117.7, 116.1, 60.6, 19.6, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₂ 331.1447; Found 331.1446.

Ethyl 1-(4-methoxyphenyl)imidazo[1,5-a]quinoline-3-carboxylate (3ae)

The title compound was prepared according to the general working procedure A (36 h, 67.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 91% yield, (94 mg), m.p. 171–173 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.18$ (d, J = 9.4 Hz, 1H), 7.72 (d, J = 7.7 Hz, 1H), 7.57–7.55 (m, 3H), 7.41–7.37 (m, 2H), 7.29–7.25 (m, 1H), 7.06–7.04 (m, 2H), 4.50 (q, J = 7.1 Hz, 2H), 3.92 (s, 3H),1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.7$, 160.8, 142.5, 134.5, 132.3, 131.3, 128.9, 128.3, 126.2, 125.7, 125.4, 125.2, 123.2, 117.7, 117.5, 114.3, 60.6, 55.4, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₃ 347.1396; Found 347.1396.

Ethyl 1-(2-methoxyphenyl)imidazo[1,5-a]quinoline-3-carboxylate (3af)

The title compound was prepared according to the general working procedure A (36 h, 67.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 96% yield, (100 mg), m.p. 173–175 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.21 (d, J = 9.4 Hz, 1H), 7.72–7.70 (m, 1H), 7.63–7.61 (m, 1H), 7.57–7.52 (m, 1H), 7.49 (d, J = 8.6 Hz, 1H), 7.40–7.36 (m, 2H), 7.27–7.23 (m, 1H), 7.16–7.12 (m, 1H), 7.00–6.98 (m, 1H), 4.54 –4.45 (m, 2H), 3.55 (s, 3H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.8, 158.3, 139.8, 134.5, 132.7, 132.1, 131.7, 128.6, 128.4, 126.2, 125.6, 125.0, 123.1, 122.5, 121.1, 117.7, 116.5, 110.8, 60.5, 55.3, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₃ 347.1396; Found 347.1397.

Ethyl 1-(benzo[d][1,3]dioxol-5-yl)imidazo[1,5-a]quinoline-3-carboxylate (**3ag**)

The title compound was prepared according to the general working procedure A (27 h, 50.4 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 91% yield, (98 mg), m.p. 209–212 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.19$ (d, J = 9.4 Hz, 1H), 7.75–7.73 (m, 1H), 7.62 (d, J = 8.5 Hz, 1H), 7.44–7.39 (m, 2H), 7.35–7.31 (m, 1H), 7.14–7.12 (m, 1H), 7.07–7.06 (m, 1H), 6.96 (d, J = 8.0 Hz, 1H), 6.10 (s, 2H), 4.50 (d, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.6, 149.0, 148.0, 142.0, 134.5, 132.1, 129.0, 128.5, 126.3, 126.3, 125.8, 125.4, 124.2, 123.1, 117.7, 117.6, 110.3, 108.8, 101.6, 60.6, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₇N₂O₄ 361.1188; Found 361.1185.$

Ethyl 1-(4-fluorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ah**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 87% yield, (87 mg), m.p. 173–176 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.19$ (d, J = 9.5 Hz, 1H), 7.76–7.74 (m, 1H), 7.66–7.61 (m, 2H), 7.48–7.40 (m, 3H), 7.32–7.21 (m, 3H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.7$, 163.6 (d, J = 249.1 Hz), 141.4, 134.6, 132.0 (d, J = 8.5 Hz), 129.1, 129.1 (d, J = 3.5 Hz), 128.5, 126.4, 126.0, 125.4, 123.4, 117.7, 117.3, 116.2, 116.0, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₀H₁₅FN₂O₂Na 357.1015; Found 357.1010. *Ethyl 1-(3-fluorophenyl)imidazo[1,5-a]quinoline-3-carboxylate* (**3ai**)

Page 17 of 34

The Journal of Organic Chemistry

The title compound was prepared according to the general working procedure A (15 h, 28.0 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 90% yield, (90 mg), m.p. 139–142 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.19$ (d, J = 9.4 Hz, 1H), 7.76–7.74 (m, 1H), 7.54–7.49 (m, 2H), 7.45–7.38 (m, 4H), 7.33–7.25 (m, 2H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.5$, 162.7 (d, J = 246.9 Hz), 140.9 (d, J = 2.7 Hz), 134.8 (d, J = 8.1 Hz), 134.6, 131.8, 130.6 (d, J = 8.3 Hz), 129.1, 128.6, 126.6, 126.1, 125.8 (d, J = 3.2 Hz), 125.4, 123.5, 117.5 (d, J = 15.2 Hz), 117.3, 117.1, 116.9, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆FN₂O₂ 335.1196; Found 335.1194.

Ethyl 1-(2-fluorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (3aj)

The title compound was prepared according to the general working procedure A (15 h, 28.0 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 86% yield, (86 mg), m.p. 183–185 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.21 (d, J = 9.5 Hz, 1H), 7.76–7.71 (m, 2H), 7.61–7.56 (m, 1H), 7.51 (d, J = 8.6 Hz, 1H), 7.44–7.40 (m, 2H), 7.39–7.35 (m, 1H), 7.33–7.29 (m, 1H), 7.25–7.20 (m, 1H), 4.54–7.48 (m, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.5, 160.9 (d, J = 246.8 Hz), 136.7, 134.7, 132.4 (d, J = 1.6 Hz), 132.3, 132.2, 129.0, 128.9, 126.6, 126.0, 125.2, 124.9 (d, J = 3.6 Hz), 123.7, 121.6 (d, J = 15.4 Hz), 117.5, 116.2, 115.9, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆FN₂O₂ 335.1196; Found 335.1193.

Ethyl 1-(4-chlorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (3ak)

The title compound was prepared according to the general working procedure A (15 h, 28.0 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 89% yield, (93 mg), m.p. 199–202 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.4 Hz, 1H), 7.77–7.75 (m, 1H), 7.62–7.59 (m, 2H), 7.54–7.50 (m, 3H), 7.46–7.41 (m, 2H), 7.34–7.30 (m, 1H), 4.51 (q, J = 7.2 Hz, 2H), 1.47 (t, J = 7.1 Hz, 1H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.5, 141.2, 136.0, 134.7, 131.9, 131.4, 131.3, 129.2, 129.1, 128.5, 126.5, 126.0, 125.4, 123.5, 117.6, 117.4, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₀H₁₅ClN₂O₂Na 373.0720; Found 373.0723.

Ethyl 1-(3-chlorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (3al)

The title compound was prepared according to the general working procedure A (15 h, 28.0 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 91% yield, (95 mg) , m.p. 167–169 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.5 Hz, 1H), 7.77–7.75 (m, 1H), 7.70–7.69 (m, 1H), 7.56–7.51 (m, 3H), 7.49–7.41 (m, 3H), 7.34–7.30 (m, 1H) 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.5, 140.8, 134.9, 134.7, 134.6, 131.8, 130.1, 130.1, 130.1, 129.2, 128.6, 128.1, 126.6, 126.1, 125.4, 123.6, 117.6, 117.5, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆ClN₂O₂ 351.0900; Found 351.0896.

Ethyl 1-(2-chlorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (3am)

The title compound was prepared according to the general working procedure A (22 h, 41.0 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 88% yield, (92 mg), m.p. 189–192 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.23 (d, J = 9.5 Hz, 1H), 7.76–7.74 (m, 1H), 7.70–7.67 (m, 1H), 7.56–7.54 (m, 2H), 7.50–7.40 (m, 3H), 7.32–7.29 (m, 2H) 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.6, 139.2, 135.4, 134.3, 132.7, 132.6, 132.1, 131.6, 129.8, 129.0, 129.0, 127.5, 126.6, 126.0, 125.1, 123.3, 117.6, 116.0, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆ClN₂O₂ 351.0900; Found 351.0904.

Ethyl 1-(4-bromophenyl)imidazo[1,5-a]quinoline-3-carboxylate (3an)

The title compound was prepared according to the general working procedure A (8 h, 14.9 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 78% yield, (92 mg), m.p. 215–218 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.18$ (d, J = 9.4 Hz, 1H), 7.75 (d, J = 7.8 Hz, 1H), 7.68 (d, J = 8.3 Hz, 2H), 7.55–7.50 (m, 3H), 7.45–7.40 (m, 2H), 7.33–7.29 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.5$, 141.2, 134.7, 132.1, 131.9, 131.5, 129.2, 128.5, 126.5, 126.1, 125.4, 124.3, 123.7, 117.7, 117.4, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆BrN₂O₂ 395.0395; Found 395.0397.

Ethyl 1-(2-bromophenyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ao**)

Page 19 of 34

The Journal of Organic Chemistry

The title compound was prepared according to the general working procedure A (12 h, 22.4 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 93% yield, (110 mg), m.p. 184–186 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.23 (d, J = 9.5 Hz, 1H), 7.76–7.73 (m, 2H), 7.67–7.65 (m, 1H), 7.55–7.51 (m, 1H), 7.49–7.40 (m, 3H), 7.30–7.28 (m, 2H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.6, 140.4, 134.8, 134.2, 133.0, 132.7, 132.1, 131.7, 129.0, 129.0, 128.0, 126.6, 126.0, 125.3, 125.1, 123.2, 117.6, 116.2, 60.7, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆BrN₂O₂ 395.0395; Found 395.0392.

Ethyl 1-(4-(trifluoromethyl)phenyl)imidazo[1,5-*a*]quinoline-3-carboxylate (**3ap**)

The title compound was prepared according to the general working procedure A (10 h, 18.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 97% yield, (112 mg), m.p. 215–217 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.22 (d, J = 9.4 Hz, 1H), 7.84–7.78 (m, 5H), 7.49–7.45 (m, 3H), 7.35–7.31 (m, 1H), 4.52 (q, J = 7.1 Hz, 2H), 1.48 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 140.8, 136.5, 134.9, 131.8 (q, J = 32.7 Hz), 131.7, 130.4, 129.3, 128.6, 126.7, 126.2, 125.8 (q, J = 3.7 Hz), 125.5, 123.9, 123.9 (q, J = 270.9 Hz), 117.7, 117.4, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₆F₃N₂O₂ 385.1164; Found 385.1160.

Ethyl 1-(3-(trifluoromethyl)phenyl)imidazo[1,5-a]quinoline-3-carboxylate (3aq)

The title compound was prepared according to the general working procedure A (14 h, 26.1 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 3:1) to give the product as a light yellow solid: 90% yield, (104 mg), m.p. 163–166 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.6 Hz, 1H), 7.98 (s, 1H) 7.86–7.81 (m, 2H), 7.78–7.75 (m, 1H), 7.69–7.65 (m, 1H), 7.46–7.42 (m, 3H), 7.31–7.27 (m, 1H), 4.51 (q, J = 7.2 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 140.7, 134.8, 133.8, 133.2, 131.7, 131.4 (q, J = 32.6 Hz), 129.4, 129.3, 128.5, 127.0 (q, J = 3.7 Hz), 126.6, 126.6 (q, J = 3.7 Hz), 126.2, 125.5, 123.8, 123.7 (q, J = 270.1 Hz), 117.6, 117.3, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₆F₃N₂O₂ 385.1164; Found 385.1162.

Ethyl 1-(naphthalen-1-yl)imidazo[1,5-a]quinoline-3-carboxylate (**3ar**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 89% yield, (98 mg), m.p. 196–198 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.30 (d, J = 9.5 Hz, 1H), 8.10 (d, J = 8.3 Hz, 1H), 7.97 (d, J = 8.2 Hz, 1H), 7.78–7.76 (m, 1H), 7.72–7.71 (m, 1H), 7.67–7.63 (m, 1H), 7.52–7.48 (m, 1H), 7.46 (d, J = 9.5 Hz, 1H), 7.35–7.26 (m, 3H), 7.03–6.97 (m, 2H), 4.53 (q, J = 7.1 Hz, 2H), 1.48 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.8, 140.7, 134.4, 133.6, 132.5, 132.0, 130.6, 130.6, 129.2, 128.8, 128.7, 128.5, 127.3, 126.6, 126.5, 125.7, 125.6, 125.3, 125.2, 123.5, 117.7, 117.1, 60.6, 14.7. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₄H₁₈N₂O₂Na 389.1266; Found 389.1267.

Ethyl 1-(2-chloro-4-fluorophenyl)imidazo[1,5-a]quinoline-3-carboxylate (**3as**)

The title compound was prepared according to the general working procedure A (16 h, 29.9 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 82% yield, (90 mg), m.p. 218–220 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.24$ (d, J = 9.5 Hz, 1H), 7.79–7.77 (m, 1H), 7.71–7.67 (m, 1H), 7.48–7.44 (m, 2H), 7.37–7.30 (m, 3H), 7.25–7.20 (m, 1H), 4.52 (q, J = 7.1 Hz, 2H), 1.48 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.6$ (d, J = 253.0 Hz), 163.4, 138.2, 136.7 (d, J = 10.3 Hz), 134.4, 133.9 (d, J = 9.1 Hz), 132.0, 129.1, 129.1, 129.0 (d, J = 3.7 Hz), 126.6, 126.1, 125.1, 123.4, 117.6, 117.5 (d, J = 25.0 Hz), 115.8, 114.9 (d, J = 21.4 Hz), 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₅FCIN₂O₂ 369.0806; Found 369.0803.

Ethyl 1-(furan-2-yl)imidazo[1,5-a]quinoline-3-carboxylate (**3at**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 87% yield, (80 mg), m.p. 144–146 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.4 Hz, 1H), 7.78–7.75 (m, 1H), 7.68–7.67 (m, 1H), 7.49–7.41 (m, 3H), 7.15–7.13 (m, 1H), 6.97–6.96 (m, 1H), 6.69–6.67 (m, 1H), 4.51 (q, J = 7.1 Hz, 2H), 1.48 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 143.9, 143.6, 134.8, 132.5, 131.8,

129.2, 129.0, 127.0, 126.2, 125.3, 123.7, 117.3, 116.9, 114.0, 112.0, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₈H₁₅N₂O₃ 307.1083; Found 307.1082.

Ethyl 1-(thiophen-2-yl)imidazo[1,5-a]quinoline-3-carboxylate (**3au**)

The title compound was prepared according to the general working procedure A (14 h, 26.1 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 4:1) to give the product as a light yellow solid: 84% yield, (81 mg), m.p. 165–168 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.20 (d, J = 9.4 Hz, 1H), 7.76–7.74 (m, 1H), 7.62–7.60 (m, 1H), 7.56 (d, J = 8.6 Hz, 1H), 7.45–7.40 (m, 3H), 7.36–7.32 (m, 1H), 7.24–7.22 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.5, 135.2, 134.9, 132.8, 132.1, 130.7, 129.0, 128.9, 128.7, 127.5, 126.7, 126.0, 125.4, 123.6, 117.4, 117.2, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₈H₁₅N₂SO₂ 323.0854; Found 323.0850.

Ethyl 1-(pyridin-2-yl)imidazo[1,5-a]quinoline-3-carboxylate (**3av**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 2:1) to give the product as a light yellow solid: 95% yield, (90 mg), m.p. 156–158 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.72–8.70 (m, 1H), 8.17 (d, J = 9.4 Hz, 1H), 7.99–7.97 (m, 1H), 7.94–7.90 (m, 1H), 7.72–7.70 (m, 1H), 7.47–7.38 (m, 4H), 7.32–7.27 (m, 1H), 4.49 (q, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 151.5, 149.2, 141.2, 137.3, 135.0, 131.7, 128.7, 128.2, 126.9, 126.0, 125.7, 125.3, 124.3, 123.5, 118.3, 117.3, 60.6, 14.5. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₁₉H₁₅N₃O₂Na 340.1062; Found 340.1059.

Ethyl 7-*chloro-1-phenylimidazo*[1,5-*a*]*quinoline-3-carboxylate* (**3ba**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 95% yield, (100 mg), m.p. 177–180 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.22 (d, J = 9.5 Hz, 1H), 7.69–7.68 (m, 1H), 7.62–7.51 (m, 5H), 7.41 (d, J = 9.2 Hz, 1H), 7.29 (d, J = 9.5 Hz, 1H), 7.21–7.18 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H), 1.46 (t, J = 7.1 Hz, 3H

3H); ${}^{13}C{}^{1}H}NMR$ (CDCl₃, 100 MHz, ppm): $\delta = 163.4$, 142.5, 134.0, 132.5, 131.4, 130.4, 130.1, 129.8, 129.0, 128.3, 127.9, 126.7, 125.0, 123.8, 118.9, 118.8, 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆ClN₂O₂ 351.0900; Found 351.0900.

Ethyl 7-*bromo-1-phenylimidazo*[1,5-*a*]*quinoline-3-carboxylate* (**3ca**)

The title compound was prepared according to the general working procedure A (16 h, 29.9 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 99% yield, (117 mg), m.p. 197–199 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.21 (d, J = 9.5 Hz, 1H), 7.86 (s, 1H), 7.62–7.59 (m, 2H), 7.57–7.51 (m, 3H), 7.34–7.29 (m, 3H), 4.50 (q, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 142.6, 134.1, 132.5, 131.2, 131.1, 130.9, 130.2, 129.8, 129.1, 127.1, 125.0, 123.8, 119.2, 119.1, 119.0, 60.8, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆BrN₂O₂ 395.0395; Found 395.0391.

Ethyl 7-fluoro-1-phenylimidazo[1,5-a]quinoline-3-carboxylate (3da)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a white solid: 98% yield, (98 mg), m.p. 184–186 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.24$ (d, J = 9.5 Hz, 1H), 7.65–7.62 (m, 2H), 7.58–7.52 (m, 3H), 7.49–7.46 (m, 1H), 7.42–7.39 (m, 1H), 7.34 (d, J = 9.5 Hz, 1H), 7.01–6.96 (m, 1H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.5$, 159.7 (d, J = 245.8 Hz), 142.4, 134.0, 132.6, 130.1, 129.9, 129.0, 128.5(d, J = 2.2 Hz), 127.1 (d, J = 8.6 Hz), 125.4 (d, J = 2.7 Hz), 123.7, 119.4 (d, J = 8.2 Hz), 118.9, 116.1 (d, J = 23.8 Hz), 113.9 (d, J = 22.5 Hz), 60.7, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₆FN₂O₂ 335.1196; Found 335.1198. *Ethyl 7-methoxy-1-phenylimidazo[1,5-a]quinoline-3-carboxylate* (**3ea**)

The title compound was prepared according to the general working procedure A (28 h, 52.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a white solid: 85% yield, (88 mg), m.p. 168–170 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.18$ (d, J = 9.4 Hz, 1H), 7.65–7.62 (m, 2H), 7.55–7.52 (m, 3H), 7.40 (d, J = 9.4 Hz, 1H), 7.34 (d, J = 9.5 Hz, 1H), 7.14–7.13 (m, 1H), 6.85–6.82 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 3.86 (s, 3H),

 1.46 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.7, 157.0, 142.0, 134.2, 133.0, 130.0, 129.9, 128.9, 126.8, 126.4, 126.1, 123.3, 118.9, 118.1, 116.5, 110.6, 60.6, 55.6, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₃ 347.1396; Found 347.1399. *Ethyl 7-methyl-1-phenylimidazo[1,5-a]quinoline-3-carboxylate* (**3fa**)

The title compound was prepared according to the general working procedure A (28 h, 52.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a white solid: 85% yield, (84 mg), m.p. 188–191 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.16$ (d, J = 9.4 Hz, 1H), 7.64–7.62 (m, 2H), 7.55–7.50 (m, 4H), 7.36–7.34 (m, 2H), 7.08–7.06 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 2.42 (s, 3H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.7$, 142.3, 135.7, 134.5, 132.9, 130.1, 130.0, 129.9, 129.6, 128.9, 128.8, 126.4, 125.4, 123.2, 117.5, 117.4, 60.6, 20.9, 14.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₁H₁₉N₂O₂ 331.1447; Found 331.1445.

Ethyl 7-(*tert-butyl*)-1-*phenylimidazo*[1,5-a]quinoline-3-carboxylate (**3ga**)

The title compound was prepared according to the general working procedure A (28 h, 52.2 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 86% yield, (96 mg), m.p. 134–136 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.18 (d, J = 9.4 Hz, 1H), 7.69 (d, J = 2.4 Hz, 1H), 7.66–7.63 (m, 2H), 7.57–7.51 (m, 3H), 7.42–7.40 (m, 2H), 7.33–7.30 (m, 1H), 4.50 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H), 1.35 (s, 9H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.8, 148.9, 142.2, 134.6, 130.1, 130.0, 129.8, 129.0, 128.8, 126.8, 126.3, 125.1, 125.1, 123.1, 117.4, 117.2, 60.6, 34.6, 31.2, 14.7. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₄H₂₄N₂O₂Na 395.1735; Found 395.1731.

Ethyl 1-phenyl-7-(trifluoromethyl)imidazo[1,5-a]quinoline-3-carboxylate (**3ha**)

The title compound was prepared according to the general working procedure A (12 h, 22.4 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a yellow solid: 98% yield, (113 mg), m.p. 170–172 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.28 (d, J = 9.5 Hz, 1H), 8.01 (s, 1H), 7.64–7.53 (m, 6H), 7.50–7.47 (m, 1H), 7.43 (d, J = 9.5 Hz, 1H), 4.51 (q, J = 7.1 Hz, 2H), 1.47 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.3, 142.9, 134.2, 133.8, 132.3, 130.3, 129.8, 129.1, 127.9 (q, J = 33.3 Hz), 126.1 (q,

J = 4.0 Hz), 125.5, 125.3, 124.7 (q, J = 3.5 Hz), 124.0, 123.5 (q, J = 270.6 Hz), 119.2, 118.1, 60.8, 14.6. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₁H₁₅F₃N₂O₂ Na 407.0983; Found 407.0980.

Methyl 1-phenylimidazo[1,5-*a*]*quinoline-3-carboxylate* (**3ia**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 85% yield, (77 mg), m.p. 160–162 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.21 (d, J = 9.4 Hz, 1H), 7.76–7.74 (m, 1H), 7.66–7.64 (m, 2H), 7.57–7.51 (m, 4H), 7.43–7.40 (m, 2H), 7.29–7.25 (m, 1H), 4.01 (s, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 164.0, 142.6, 134.7, 132.9, 132.1, 129.9, 129.9, 129.0, 128.9, 128.5, 126.5, 125.9, 125.4, 123.0, 117.6, 117.5, 51.7. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₉H₁₅N₂O₂ 303.1134; Found 303.1130.

Phenyl(1-phenylimidazo[1,5-a]quinolin-3-yl)methanone (**3ja**)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 20:1) to give the product as a light yellow solid: 65% yield, (68 mg), m.p. 168–170 °C; ¹H NMR (CD₃SOCD₃, 400 MHz, ppm): δ = 8.37 (d, J = 9.4 Hz, 1H), 8.33–8.31 (m, 2H), 7.99 (d, J = 7.8 Hz, 1H), 7.80 (d, J = 9.5 Hz, 1H), 7.74–7.71 (m, 2H), 7.67–7.58 (m, 4H), 7.54–7.50 (m, 3H), 7.44–7.39 (m, 2H); ¹³C{¹H}NMR (CD₃SOCD₃, 100 MHz, ppm): δ = 187.6, 141.9, 138.8, 135.6, 133.4, 132.4, 131.8, 130.6, 130.6, 130.3, 129.8, 129.6, 129.4, 128.8, 128.5, 126.7, 125.8, 118.0, 117.4. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₄H₁₇N₂O 349.1341; Found 349.1340.

Ethyl 3-phenylimidazo[5,1-a]isoquinoline-1-carboxylate (3ka)

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 6:1) to give the product as a light yellow solid: 94% yield, (61 mg), m.p. 152–154 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 9.85 (d, J = 8.1 Hz, 1H), 7.98 (d, J = 7.4 Hz, 1H), 7.78–7.75 (m, 2H), 7.69–7.65 (m, 2H), 7.60–7.51 (m, 4H), 6.97 (d, J = 7.4 Hz, 1H), 4.57 (q, J = 7.1 Hz, 2H), 1.51 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 164.3, 140.9, 132.8, 129.8, 129.4, 129.1, 128.9,

128.9, 128.6, 127.8, 126.8, 125.3, 124.3, 120.3, 115.8, 61.0, 14.6. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₂₀H₁₆N₂O₂ Na 339.1110; Found 339.1107.

Ethyl 3-phenylimidazo[1,5-a]pyridine-1-carboxylate (3la)^{13a}

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a light yellow solid: 75% yield, (60 mg), m.p. 129–130 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.30-8.23$ (m, 2H), 7.80–7.77 (m, 2H), 7.54–7.45 (m, 3H), 7.14–7.10 (m, 1H), 6.79–6.75 (m, 1H) 4.49 (q, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 163.6$, 139.2, 135.4, 129.6, 129.1, 129.0, 128.8, 124.2, 122.5, 121.8, 120.1, 114.4, 60.4, 14.7.

Ethyl 2,5-diphenyloxazole-4-carboxylate (3ma)²⁰

The title compound was prepared according to the general working procedure A (24 h, 44.8 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 30:1) to give the product as a yellow solid: 65% yield, (57 mg), m.p. 85–86 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): $\delta = 8.20-8.10$ (m, 4H), 7.54–7.46 (m, 6H), 4.47 (q, J = 7.1 Hz, 2H), 1.44 (t, J = 7.1 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): $\delta = 162.3$, 159.8, 155.1, 131.1, 130.3, 128.8, 128.6, 128.4, 128.3, 127.1, 126.9, 126.4, 61.5, 14.3.

Ethyl imidazo[1,5-*a*]*quinoline-3-carboxylate* (**3ax**)

The title compound was prepared according to the general working procedure B (6 h, 7.5 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 2:1) to give the product as a pink solid: 50% yield, (36 mg), m.p. 157–159 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.65 (s, 1H), 8.09–8.04 (m, 2H), 7.80–7.78 (m, 1H), 7.69–7.65 (m, 1H), 7.55–7.51 (m, 1H), 7.41 (d, J = 9.5 Hz, 1H), 4.49 (q, J = 7.1 Hz, 2H), 1.48 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.4, 132.8, 130.6, 129.6, 129.2, 128.0, 126.4, 126.0, 124.4, 124.2, 117.5, 114.9, 60.6, 14.6. HRMS (ESI-TOF) m/z: [M + Na]⁺ Calcd for C₁₄H₁₂N₂O₂ Na 263.0797; Found 263.0794.

Ethyl 1-propylimidazo[1,5-*a*]*quinoline-3-carboxylate* (**3ay**)

The title compound was prepared according to the general working procedure B (7 h, 8.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a yellow solid: 41% yield, (35 mg), m.p. 127–129 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.18 (d, J = 8.6 Hz, 1H), 8.11 (d, J = 9.5 Hz, 1H), 7.76–7.74 (m, 1H), 7.65–7.61 (m, 1H), 7.51–7.47 (m, 1H), 7.31 (d, J = 9.4 Hz, 1H), 4.48 (q, J = 7.1 Hz, 2H), 3.43–3.39 (m, 2H), 2.04–1.94 (m, 2H), 1.46 (t, J = 7.1 Hz, 3H), 1.13 (t, J = 7.4 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.6, 144.6, 134.6, 132.7, 129.1, 128.8, 125.6, 125.6, 121.9, 117.9, 116.7, 60.5, 34.3, 20.7, 14.7, 13.9. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₇H₁₉N₂O₂ 283.1447; Found 283.1447.

Ethyl 1-isopropylimidazo[1,5-a]quinoline-3-carboxylate (3az)^{19b}

The title compound was prepared according to the general working procedure B (7 h, 8.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a yellow solid: 38% yield, (32 mg), m.p. 118–119 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.29 (d, J = 8.7 Hz, 1H), 8.13 (d, J = 9.4 Hz, 1H), 7.77–7.75 (m, 1H), 7.65–7.61 (m, 1H), 7.51–7.47 (m, 1H), 7.31 (d, J = 9.5 Hz, 1H), 4.49 (q, J = 7.1 Hz, 2H), 3.92–3.82 (m, 1H), 1.62 (d, J = 6.7 Hz, 6H), 1.47 (t, J = 7.4 Hz, 3H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.8, 150.0, 134.6, 132.8, 129.2, 128.8, 125.8, 125.6, 125.5, 122.0, 118.1, 117.3, 60.5, 30.3, 21.5, 14.6.

Ethyl 1-(sec-butyl)imidazo[1,5-a]quinoline-3-carboxylate (**3aA**)

The title compound was prepared according to the general working procedure B (7 h, 8.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a yellow solid: 39% yield, (35 mg), m.p. 116–118 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.23 (d, J = 8.6 Hz, 1H), 8.12 (d, J = 9.5 Hz, 1H), 7.76–7.73 (m, 1H), 7.64–7.60 (m, 1H), 7.50–7.46 (m, 1H), 7.29 (d, J = 9.4 Hz, 1H), 4.48 (q, J = 7.0 Hz, 2H), 3.64–3.56 (m, 1H), 2.28–2.18 (m, 1H), 1.92–1.81 (m, 1H), 1.59 (d, J = 6.7 Hz, 3H), 1.46 (t, J = 7.1 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.8, 149.4, 134.4, 132.8, 129.2, 128.7, 125.8, 125.5, 125.5, 122.1, 118.1, 117.2, 60.4, 37.0, 28.4, 18.6, 14.6, 12.0. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₈H₂₁N₂O₂ 297.1603; Found 297.1602.

Ethyl 1-isobutylimidazo[1,5-a]quinoline-3-carboxylate (**3aB**)

The title compound was prepared according to the general working procedure B (7 h, 8.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 5:1) to give the product as a yellow solid: 48% yield, (43 mg), m.p. 140–142 °C; ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.19 (d, J = 8.6 Hz, 1H), 8.13 (d, J = 9.4 Hz, 1H), 7.77–7.75 (m, 1H), 7.66–7.61 (m, 1H), 7.51–7.47 (m, 1H), 7.32 (d, J = 9.4 Hz, 1H), 4.50 (q, J = 7.1 Hz, 2H), 3.35 (d, J = 7.2 Hz, 2H), 2.42–2.32 (m, 1H), 1.46 (t, J = 7.1 Hz, 3H), 1.06 (d, J = 6.6 Hz, 6H); ¹³C{¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 163.6, 143.8, 134.6, 132.5, 129.1, 128.7, 125.5, 125.5, 121.9, 117.8, 116.7, 60.4, 40.7, 26.4, 22.3, 14.6. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₈H₂₁N₂O₂ 297.1603; Found 297.1602.

Ethyl 2-((4-methoxyphenyl)imino)-2-(quinolin-2-yl)acetate (7')

The title compound was prepared according to the general working procedure A (7 h, 13.1 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 20:1) to give the product as a light yellow oil: 92% yield, (92 mg); ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.33 (d, J = 8.6 Hz, 1H), 8.25 (d, J = 8.5 Hz, 1H), 8.12 (d, J = 8.5 Hz, 1H), 7.87–7.84 (m, 1H), 7.75–7.71 (m, 1H), 7.62–7.58 (m, 1H), 7.17–7.13 (m, 2H), 6.95–6.91 (m, 2H), 4.38 (q, J = 7.1 Hz, 2H), 3.84 (s, 3H), 1.25 (t, J = 7.1 Hz, 3H); ¹³C {¹H} NMR (CDCl₃, 100 MHz, ppm): δ = 165.7, 159.8, 157.9, 153.3, 147.4, 142.2, 136.6, 130.3, 129.6, 128.7, 127.9, 127.5, 121.9, 118.4, 114.1, 61.4, 55.4, 14.1. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₁₉N₂O₃ 335.1396; Found 335.1393.

Ethyl 2-(piperidin-1-yl)-2-(quinolin-2-yl)acetate (5')

The title compound was prepared according to the general working procedure A (2 h, 3.7 F mol⁻¹) and purified by column chromatography (petroleum ether / ethyl acetate = 15:1) to give the product as a light yellow oil: 82% yield, (73 mg); ¹H NMR (CDCl₃, 400 MHz, ppm): δ = 8.16 (d, J = 8.5 Hz, 1H), 8.10 (d, J = 8.3 Hz, 1H), 7.82–7.76 (m, 2H), 7.72–7.68 (m, 1H), 7.55–7.51 (m, 1H), 4.46 (s, 1H), 4.26–4.15 (m, 2H), 2.57–2.47 (m, 4H), 1.64–1.59 (m, 4H), 1.50–1.43 (m, 2H), 1.23 (t, J = 7.2 Hz, 3H); ¹³C {¹H}NMR (CDCl₃, 100 MHz, ppm): δ = 171.0, 157.4, 147.5, 136.5,

129.5, 129.4, 127.6, 127.4, 126.6, 120.6, 76.9, 60.9, 52.2, 25.9, 24.2, 14.2. HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₈H₂₃N₂O₂ 299.1760; Found 299.1757.

ASSOCIATED CONTENT

Supporting Information

¹H NMR and ¹³C {¹H}NMR spectra for the substrates **1** (**1a-1e, 1g, 1h** and **1k**) and all the products as well as the photographic depiction of the electrolysis setup. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

Fax: 86-551-3631760. E-mail: zwang3@ustc.edu.cn

zgzha@ustc.edu.cn

ORCID

Zhiyong Wang: 0000-0002-3400-2851

Zhenggen Zha: 0000-0002-5388-0115

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS:

We are grateful for the financial support from the National Natural Science Foundation of China (21672200, 21472177, 21432009, 21772185) and the assistance of the product characterization

from the Chemistry Experiment Teaching Center of University of Science and Technology of China. This work was supported by the Strategic Priority Research Program of the Chinese Academy of Sciences, Grant No. XDB20000000.

REFERENCES

- (a) Vitaku, E.; Smith, D. T.; Njardarson, J. T. Analysis of the Structural Diversity, Substitution Patterns, and Frequency of Nitrogen Heterocycles among U.S. FDA Approved Pharmaceuticals. *J. Med. Chem.* 2014, *57*, 10257.
- 2 For selected reviews see: (a) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. Recent Advances in the Transition Metal-catalyzed Twofold Oxidative C–H Bond Activation Strategy for C–C and C–N Bond Formation. *Chem. Soc. Rev.* 2011, 40, 5068. (b) Shin, K.; Kim, H.; Chang, S. Transition-Metal-Catalyzed C–N Bond Forming Reactions Using Organic Azides as the Nitrogen Source: A Journey for the Mild and Versatile C–H Amination. *Acc. Chem. Res.* 2015, 48, 1040. (c) Park, Y.; Kim, Y.; Chang, S. Transition Metal-Catalyzed C–H Amination: Scope, Mechanism, and Applications. *Chem. Rev.* 2017, *117*, 9247. (d) Che, C. M.; Lo, V. K. Y.; Zhou, C. Y.; Huang, J. S. Selective Functionalisation of Saturated C–H Bonds with Metalloporphyrin Catalysts. *Chem. Soc. Rev.* 2011, 40, 1950. (e) Ramirez, T. A.; Zhao, B.; Shi, Y. Recent Advances in Transition Metal-catalyzed sp3 C–H Amination Adjacent to Double Bonds and Carbonyl Groups. *Chem. Soc. Rev.* 2012, 41, 931.
- 3 For selected examples see: (a) Kim, H. J.; Kim, J.; Cho, S. H.; Chang, S. Intermolecular Oxidative C–N Bond Formation under Metal-Free Conditions: Control of Chemoselectivity between Aryl sp2 and Benzylic sp3 C–H Bond Imidation. J. Am. Chem. Soc. 2011, 133, 16382. (b) Souto, J. A.; Zian, D.; Muniz, K.; Iodine(III)-Mediated Intermolecular Allylic Amination under Metal-Free Conditions. J. Am. Chem. Soc. 2012, 134, 7242.
- 4 For selected reviews see: (a) Francke, R.; Little, R. D.; Redox Catalysis in Organic Electrosynthesis: Basic Principles and Recent Developments. *Chem. Soc. Rev.* 2014, 43, 2492. (b) Yan, M. Kawamata, Y. Baran, P. S. Synthetic Organic Electrochemical Methods Since 2000: On the Verge of a Renaissance. *Chem. Rev.*, 2017, 117, 13230.

(c) Jiang, Y.; Xu, K.; Zeng, C. Use of Electrochemistry in the Synthesis of Heterocyclic Structures. *Chem. Rev.*, 2018, *118*, 4485. (d) Tang, S.; Liu, Y. C.; Lei, A. W. Electrochemical Oxidative Cross-coupling with Hydrogen Evolution: A Green and Sustainable Way for Bond Formation. *Chem.* 2018. *4*, 27. (e) Wiebe, A.; Gieshoff, T.; Mçhle, S.; Rodrigo, E. Zirbes, M. Waldvogel, S. R. Electrifying Organic Synthesis. *Angew. Chem. Int. Ed.* 2018, *57*, 5594. (f) Ma, C.; Fang, P. Mei, T.-S. Recent Advances in C–H Functionalization Using Electrochemical Transition Metal Catalysis. *ACS. Catal.* 2018, *8*, 7179. (g) Kärkäs, M. D. Electrochemical Strategies for C–H Functionalization and C–N Bond Formation. *Chem. Soc. Rev.*, 2018, *47*, 5786.

- For selected examples see: (a) Liang, S.; Zeng, C-C.; Luo, X-G.; Ren, F-Z.; Tian, H-Y.; Sun, B-G.; Little, R. D. Electrochemically Catalyzed Amino-oxygenation of Styrenes: *n*-Bu₄NI Induced C–N Followed by a C–O Bond Formation Cascade for the Synthesis of Indulines. *Green Chem.*, 2016, *18*, 2222. (b) Fu, N.; Sauer, G. S.; Saha, A.; Loo, A.; Lin, S. Metal-catalyzed Electrochemical Diazidation of Alkenes. *Science* 2017, *357*, 575. (c) Zhang, S.; Li, L. J.; Xue, M. Y.; Zhang, R. K.; Xu, K.; Zeng, C. C. Electrochemical Formation of *N*-Acyloxy Amidyl Radicals and Their Application: Regioselective Intramolecular Amination of sp2 and sp3 C–H Bonds. *Org. Lett.* 2018, *20*, 3443. (d) Li, J. Huang, W. H.; Chen, J. Z.; He, L. F.; Cheng, X. Li, G. G. Electrochemical Aziridination by Alkene Activation Using a Sulfamate as the Nitrogen Source. *Angew. Chem. Int. Ed.* 2018, *57*, 5695. (e) Herold, S.; Bafaluy, D.; Muñiz, K. Anodic Benzylic C(sp3)–H Amination: Unified Access to Pyrrolidines and Piperidines. *Green Chem.*, 2018, *20*, 3191.
- 6 (a) Xu, H-C.; Moeller. K. D. Intramolecular Anodic Olefin Coupling Reactions: The Use of a Nitrogen. J. Am. Chem. Soc. 2008, 130, 13542. (b) Xu, H-C.; Moeller. K. D. Intramolecular Anodic Olefin Coupling Reactions and the Synthesis of Cyclic Amines. J. Am. Chem. Soc. 2010, 132, 2839. (c) Campbell, J. M.; Xu, H-C.; Moeller. K. D. Investigating the Reactivity of Radical Cations: Experimental and Computational Insights into the Reactions of Radical Cations with Alcohol and *p*-Toluene Sulfonamide Nucleophiles. J. Am. Chem. Soc. 2012, 134, 18338.

- 7 (a) Hou, Z-W.; Mao, Z.-Y.; Zhao, H-B.; Melcamu, Y. Y.; Lu, X.; Song, J. S. Xu. H-C. Electrochemical C-H/N-H Functionalization for the Synthesis of Highly Functionalized (Aza)indoles. *Angew. Chem. Int. Ed.* 2016, *55*, 9168. (b) Zhao, H-B.; Hou, Z-W.; Liu, Z-J.; Zhou, Z-F.; Song, S.; Xu. H-C. Amidinyl Radical Formation through Anodic N-H Bond Cleavage and Its Application in Aromatic C-H Bond Functionalization. *Angew. Chem. Int. Ed.* 2017, *56*, 587.
- 8 (a) Morofuji, T.; Shimizu, A.; Yoshida. J-I.; Electrochemical C-H Amination: Synthesis of Aromatic Primary Amines via N-Arylpyridinium Ions. J. Am. Chem. Soc.
 2013, 135, 5000. (b) Morofuji, T.; Shimizu, A.; Yoshida. J-I. Direct C-N Coupling of Imidazoles with Aromatic and Benzylic Compounds via Electrooxidative C-H Functionalization. J. Am. Chem. Soc. 2014, 136, 4496. (c) Morofuji, T.; Shimizu, A.; Yoshida. J-I. Heterocyclization Approach for Electrooxidative Coupling of Functional Primary Alkylamines with Aromatics. J. Am. Chem. Soc. 2015, 137, 9816.
- 9 (a) Tang, S.; Wang, D.; Liu, Y. C.; Zeng, L.; Lei. A. W. Cobalt-catalyzed Electrooxidative C-H/N-H [4+2] Annulation with Ethylene or Ethyne. *Nature Communications*, 2018, *9*, 798. (b) Gao, X. L.; Wang, P.; Zeng, L.; Tang, S.; Lei. A. W. Cobalt(II)-Catalyzed Electrooxidative C-H Amination of Arenes with Alkylamines. *J. Am.* Chem. Soc. 2018, *140*, 4195.
- 10 Yang, Q.-L.; Wang, X.-Y.; Lu, J.-Y.; Zhang, L.-P.; Fang, P.; Mei. T.-S. Copper-Catalyzed Electrochemical C-H Amination of Arenes with Secondary Amines. J. Am. Chem. Soc. 2018, 140, 11487.
- (a) Mei, R. H.; Sauermann, N.; Oliveira, J. C. A.; Ackermann. L. Electroremovable Traceless Hydrazides for Cobalt-Catalyzed Electrooxidative C-H/N-H Activation with Internal Alkynes. *J. Am. Chem. Soc.* 2018, *140*, 7913. (b) Tian, C.; Massignan, L.; Meyer, T. H.; Ackermann. L. Electrochemical C-H/N-H Activation by Water-Tolerant Cobalt Catalysis at Room Temperature. *Angew. Chem. Int. Ed.* 2018, , 2383.
- 12 (a) Browne, L. J.; Gude, C.; Rodriguez, H.; Steele. R. E. Fadrozole Hydrochloride: A Potent, Selective, Nonsteroidal Inhibitor of Aromatase. *J. Med. Chem.* 1991, *34*, 725.
 (b) Kim, D. Potent 1,3,4-Trisubstituted Pyrrolidine CCR5 Receptor Antagonists:

Effects of Fused Heterocycles on Antiviral Activity and Pharmacokinetic Properties. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 2129. (c) Hranjec, M.; Kralj, M.; Piantanida, I.; Sedić, M.; Šuman, L.; Pavelić, K.; Karminski-Zamola, G. Novel Cyano- and Amidino-Substituted Derivatives of Styryl-2-Benzimidazoles and Benzimidazo[1,2-*a*]quinolines. Synthesis, Photochemical Synthesis, DNA Binding, and Antitumor Evaluation, Part 3. *J. Med. Chem.* **2007**, *50*, 5696.

13 For recent examples for the synthesis of 1,3-disubstituted imidazo[1,5-a]quinolines see: (a) Yan, Y. Z.; Zhang, Y. H.; Zha, Z. G. and Wang, Z. Y. Mild Metal-Free Sequential Dual Oxidative Amination of C(sp3)-H bonds: Efficient Synthesis of Imidazo[1,5-a]pyridines. Org. Lett. 2013, 15, 2274. (b) Wang, H. Q.; Xu, W. T.; Xin, L. L.; Liu, W. M. Wang, Z. Q. and Xu, K. Synthesis of 1,3-Disubstituted Imidazo[1,5-a]pyridines from Amino Acids via Catalytic Decarboxylative Intramolecular Cyclization. J. Org Chem. 2016, 81, 3681. (c) Tan, Z. D.; Zhao, H.; Zhou, C. J.; Jiang, H. F. and Zhang, M. Aerobic Copper-Catalyzed Halocyclization of Methyl N-Heteroaromatics with Aliphatic Amines: Access to Functionalized Imidazo-Fused N-Heterocycles. J. Org. Chem. 2016, 81, 9939. (d) Sheng, J.; Liu, J. D.; Zhao, H.; Zheng. L. Y. and Wei, X. C. Metal-free Synthesis of Imidazo[1,5-a]pyridines via Elemental Sulfur Mediated Sequential Dual Oxidative Csp3-H Amination. Org. Biomol. Chem., 2018, 16, 5570. (e) Sandeep, M.; Dushyant, P. S.; Sravani, B.; Reddy. K. R. Direct Access to Halogenated Fused Imidazo[1,5-a]N-heteroaromatics through Copper-Promoted Double Oxidative C-H Amination and Halogenation. Eur. J. Org. Chem. 2018, 3036. (f) Feng, C. T.; Wei, H-J.; Li, J.; Peng, Y.; Xu, K. Synthesis of Cyanide-Functionalized Imidazo[1,5-*a*]quinolines via Copper-Mediated Aerobic Three-Component Cyclizations. Adv. Synth. Catal. 2018, 360, 1. (g) Qian, P.; Yan, Z. C.; Zhou, Z. H.; Hu, K. F.; Wang, J. W.; Li, Z. B.; Zha, Z. G.; Wang, Z. Y. Electrocatalytic Intermolecular C(sp3)–H/N–H Coupling of Methyl N–Heteroaromatics with Amines and Amino Acids: Access to Imidazo-Fused N-Heterocycles Org. Lett. 2018, 20, 6359.

- (a) Zhang, Z. L.; Su, J. H.; Zha, Z. G.; Wang, Z. Y. A Novel Approach for the One-Pot Preparation of *a*-Ketoamides by Anodic Oxidation. *Chem. Commun.*, 2013, 49, 8982. (b) Zhang, Z. L.; Su, J. H.; Zha, Z. G.; Wang, Z. Y. Electrochemical Synthesis of the Aryl *a*-Ketoesters from Acetophenones Mediated by KI. *Chem.-Eur. J.*, 2013, 19, 17711. (c) Gao, H. H.; Zha, Z. G.; Zhang, Z. L.; Ma, H. Y.; Wang, Z. Y. A Simple and Efficient Approach to Realize Difunctionalization of Arylketones with Malonate Esters via Electrochemical Oxidation. *Chem. Commun.*, 2014, 50, 5034. (d) Xu, K.; Zhang, Z. L.; Qian, P.; Zha, Z. G.; Wang, Z. Y. Electrosynthesis of Enaminones Directly from Methyl Ketones and Amines with Nitromethane as a Carbon Source. *Chem. Commun.*, 2015, 51, 11108. (e) Li, Y. N.; Gao, H. H.; Zhang, Z. L.; Qian, P.; Bi, M. X.; Zha, Z. G.; Wang, Z. Y. Electrochemical Synthesis of *a*-Enaminones from Aryl Ketones. *Chem. Commun.*, 2016, 52, 8600. (f) Qian, P.; Su, J-H.; Wang, Y. K.; Bi, M. X.; Zha, Z. G.; Wang, Z. Y. Electrocatalytic C–H/N–H Coupling of 2'-Aminoacetophenones for the Synthesis of Isatins. *J. Org Chem.* 2017, 82, 6434.
- 15 (a) Wang, Q.; Zhang, S.; Guo. F. F.; Zhang, B. Q.; Hu, P.; Wang, Z. Y. Natural α-Amino Acids Applied in the Synthesis of Imidazo[1,5-*a*]*N*-heterocycles under Mild Conditions. *J. Org. Chem.* 2012, *77*, 11161. (b) Wang, Q. Q.; Xu, K.; Jiang. Y-Y.; Liu, Y-G.; Sun, B-G.; Zeng, C-C.; Electrocatalytic Minisci Acylation Reaction of *N*-Heteroarenes Mediated by NH₄I. *Org. Lett.* 2017, *19*, 5517.
- 16 Yu, Y.; Liu, Y.; Liu, A. X.; Xie, H. X.; Li, H.; Wang, W. Ligand-Free Cu-Catalyzed [3 + 2] Cyclization for the Synthesis of Pyrrolo[1,2-*a*]quinolines with Ambient Air as a Terminal Oxidant. *Org. Biomol. Chem.*, 2016, *14*, 7455.
- 17 Jiang, X. Y.; Boehm, P.; Hartwig, J. F. Stereodivergent Allylation of Azaaryl Acetamides and Acetates by Synergistic Iridium and Copper Catalysis. J. Am. Chem. Soc. 2018, 140, 1239.
- Wang, T-L.; Ouyang, G. H.; He, Y-M.; Fan, Q-H. Asymmetric Tandem Reduction of 2-(Aroylmethyl)quinolines with Phosphine-Free Ru-TsDPEN Catalyst. *Synlett*, 2011, 7, 939.
- 19 (a) Feng, C. T.; Yan, Y. Z.; Zhang, Z. L.; Xu, K.; Wang, Z. Y. Cerium(III)-Catalyzed

Cascade Cyclization: an Efficient Approach to Functionalized Pyrrolo[1,2-*a*]quinolines. *Org. Biomol. Chem.*, **2014**, *12*, 4837. (b) Yan, Z. C.; Wan, C. F.; Yang, Y.; Zha, Z. G.; Wang, Z. Y. The Synthesis of Imidazo[1,5-*a*]quinolines via a Decarboxylative Cyclization under Metal-Free Conditions. *RSC Adv.*, **2018**, *8*, 23058.

20 Wan, C. F.; Zhang, J. T.; Wang, S. J.; Fan, J. M.; Wang, Z. Y. Facile Synthesis of Polysubstituted Oxazoles via A Copper-Catalyzed Tandem Oxidative Cyclization. *Org. Lett.* 2010, *12*, 2338.

ACS Paragon Plus Environment