

A Ring-Closing Yne-Carbonyl Metathesis of Ynamides

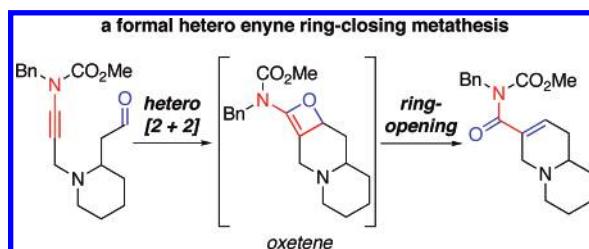
Kimberly C. M. Kurtz, Richard P. Hsung,*† and Yanshi Zhang

Department of Chemistry, University of Minnesota, Minneapolis, Minnesota 55455

rphsung@pharmacy.wisc.edu

Received October 13, 2005

ABSTRACT



An acid-catalyzed ring-closing ynamide-carbonyl metathesis is described here. This hetero RCM methodology is applicable to the construction of carbocycles as well as heterocycles such as chromenes, quinolizidines, indolizidines, and pyrrolizidines.

Ring-closing metathesis (RCM) is widely recognized as one of the most powerful tools in organic synthesis.¹ Considerable investigation has taken place in the areas of alkene-alkyne (or enyne) ring-closing metathesis, and many inventive works appear in the literature.^{2–4} However, less common are those metatheses where one of the reacting partners is a carbonyl-containing functional group; such transformations are formal

* Current address: Division of Pharmaceutical Sciences and Department of Chemistry, University of Wisconsin, Madison, WI 53705-2222.

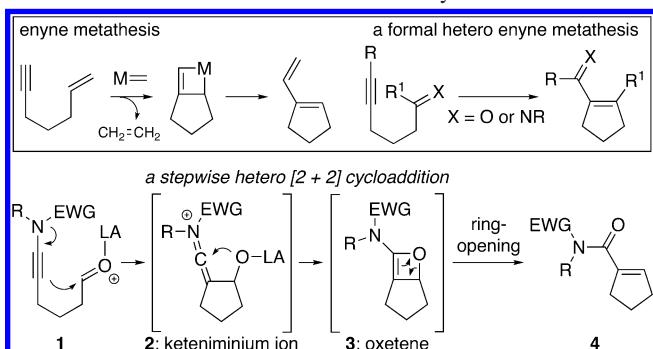
(1) For reviews on RCM, see: (a) Trnka, T. M.; Grubbs, R. H. *Acc. Chem. Res.* **2001**, *34*, 18. (b) Schrock, R. R.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2003**, *42*, 4592. (c) Walters, M. A. *Prog. Heterocycl. Chem.* **2003**, *15*, 1. (c) Nakamura, I.; Yamamoto, Y. *Chem. Rev.* **2004**, *104*, 2127. (d) Deiters, A.; Martin, S. F. *Chem. Rev.* **2004**, *104*, 2199. (e) McReynolds, M. D.; Dougherty, J. M.; Hanson, P. R. *Chem. Rev.* **2004**, *104*, 2239. (f) Wallace, D. J. *Angew. Chem., Int. Ed.* **2005**, *44*, 1912.

(2) For reviews on enyne metathesis, see: (a) Diver, S. T.; Giessert, A. *J. Chem. Rev.* **2004**, *104*, 1317. (b) Mori, M. *J. Mol. Catal. A* **2004**, *213*, 73. (c) Poulsen, C. S.; Maden, R. *Synthesis* **2003**, 1. (d) Mori, M. In *Handbook of Metathesis*; Grubbs, R. H., Ed.; Wiley-VCH: Weinheim, 2003. (e) Fürstner, A. *Angew. Chem., Int. Ed.* **2000**, *39*, 3012.

(3) For some recent examples of enyne metathesis, see: (a) Hansen, E. C.; Lee, D. *J. Am. Chem. Soc.* **2004**, *126*, 15074. (b) Castarlenas, R.; Eckert, M.; Dixneuf, P. H. *Angew. Chem., Int. Ed.* **2005**, *44*, 2576. (c) Galan, B. R.; Giessert, A. J.; Keister, J. B.; Diver, S. T. *J. Am. Chem. Soc.* **2005**, *127*, 5762. (d) Kim, M.; Lee, D. *Org. Lett.* **2005**, *7*, 1865.

(4) For enyne metathesis using heteroatom-substituted alkynes, see: (a) Saito, N.; Sato, Y.; Mori, M. *Org. Lett.* **2002**, *4*, 803. (b) Huang, J.; Xiong, H.; Hsung, R. P.; Rameshkumar, C.; Mulder, J. A.; Grebe, T. P. *Org. Lett.* **2002**, *4*, 2417. (c) Schramm, M. P.; Reddy, D. S.; Kozmin, S. A. *Angew. Chem., Int. Ed.* **2001**, *40*, 4274. (d) Timmer, M. S. M.; Ovaa, H.; Filippov, D. V.; Van der Marel, G. A.; Van Boom, J. H. *Tetrahedron Lett.* **2001**, *42*, 8231. (e) Clark, J. S.; Hamelin, O. *Angew. Chem., Int. Ed.* **2000**, *39*, 372.

Scheme 1. Formal Hetero Enyne RCM



hetero enyne metatheses (Scheme 1).⁵ Most recently, Krische reported an elegant account describing a catalytic intramolecular metathesis with acetylenic ketones and aldehydes.⁶ Our interest in developing useful synthetic methodologies

(5) For leading examples, see: (a) Curini, M.; Epifano, F.; Maltese, F.; Rosati, O. *Synlett* **2003**, 552. (b) Viswanathan, G. S.; Li, C.-J. *Tetrahedron Lett.* **2002**, *43*, 1613. (c) Wempe, M. F.; Grunwell, J. R. *Tetrahedron Lett.* **2000**, *41*, 6709. (d) Balog, A.; Geib, S. J.; Curran, D. P. *J. Org. Chem.* **1995**, *60*, 345. For some earlier studies, see: (e) Harding, C. E.; Hanack, M. *Tetrahedron Lett.* **1971**, *12*, 1253. (f) Balf, R. J.; Rao, B.; Weiler, L. *Can. J. Chem.* **1971**, *49*, 3135.

(6) For a leading reference, see: Rhee, J. U.; Krische, M. J. *Org. Lett.* **2005**, *7*, 2493.

using ynamides^{7–10} led us to explore the possibility of ring-closing ynamide-carbonyl metathesis possibly promoted by a Lewis acid (**1 → 4**), which is not known.¹¹ Mechanistically, this reaction would proceed through ring opening of an amide-substituted oxetene intermediate **3** formed through a stepwise hetero [2 + 2] cycloaddition pathway.^{12,13} We communicate here our preliminary success in ring-closing yne-carbonyl metathesis employing ynamides.

To achieve such a hetero ring-closing metathesis, we commenced our work by screening a range of Lewis acids as summarized in Table 1. Specifically, oxazolidinone,

Table 1. Screening of Lewis Acids and Catalysts

entry	aldehyde ^a	Lewis acid	mol %	time [h]	temp [°C]	product	yield [%] ^b
1		BF ₃ -OEt ₂	5	0.5	rt		51
2		BF ₃ -OEt ₂	10	0.5	rt		74
3		BF ₃ -OEt ₂	25	0.5	rt		82
4	5	BF ₃ -OEt ₂	50	0.5	rt	6	74
5		BF ₃ -OEt ₂	75	0.5	rt		71
6		BF ₃ -OEt ₂	100	0.5	rt		74
7		InCl ₃	25	20	-78 °C to rt		7
8	5	Mg(OTf) ₂	25	5	rt	6	85
9		Sn(OTf) ₂	25	5	rt		92
10		Ti(O-i-Pr) ₃ Cl	25	20	-78 °C to rt		nd ^c
11		TiCl ₄	25	20	-78 °C to rt		nd
12		Zn(OTf) ₂	25	5	rt		81
13		BF ₃ -OEt ₂	25	0.5	rt		51
14		Mg(OTf) ₂	25	5	rt		45
15		Sn(OTf) ₂	25	5	rt		42
16		Sn(OTf) ₂	25	5	rt		43
17	7	Zn(OTf) ₂	25	5	rt	8	56
18		BF ₃ -OEt ₂	25	20	-78 °C to rt		46
19		Mg(OTf) ₂	25	5	rt		14
20		Zn(OTf) ₂	25	5	rt		48
21		Sn(OTf) ₂	25	5	rt		48
22	9	In(OTf) ₃	25	20	rt	10	4

^a All reactions were carried out in CH₂Cl₂ at a concentration of 0.01 M. ^b Isolated yields. ^c nd: not determined.

camphor lactam, and sulfonamide based ynamides **5**, **7**, and **9** were used, as they represent three major classes of ynamides. Entries 1–6 establish that BF₃-OEt₂ is an excellent catalyst with a loading of 25 mol % (entry 3), giving the highest yield of the oxazolidinone-based metathesis product **6**.¹⁴ However, other Lewis acids also worked comparably well at room temperature (entries 8, 9, and 12).

In general, metal triflate catalyzed reactions appear to be slower than those catalyzed by BF₃-OEt₂ and required reaction times of 5 h or greater. The exceptions are InCl₃ (entry 7), TiCl₄, and Ti(O-i-Pr)₃Cl (entries 10 and 11),¹⁵ which failed to give the desired product. Hetero-RCM of less stable ynamides such as **7** (aza-camphor substituted) and **9** (sulfonyl

(7) For reviews on ynamides, see: (a) Zifcsak, C. A.; Mulder, J. A.; Hsung, R. P.; Rameshkumar, C.; Wei, L.-L. *Tetrahedron* **2001**, *57*, 7575. (b) Mulder, J. A.; Kurtz, K. C. M.; Hsung, R. P. *Synlett* **2003**, 1379. (c) Katritzky, A. R.; Jiang, R.; Singh, S. K. *Heterocycles* **2004**, *63*, 1455.

substituted) were also feasible, although yields of the corresponding metathesis products were lower than those of **5**. We attribute these lower yields to the relative stability of ynamides **7** and **9**.

The generality of this hetero RCM is shown in Table 2. Specifically, six- and seven-membered rings are also acces-

Table 2. Scope of the Yne-Carbonyl Ring-Closing Metathesis

substrate ^a	product ^b	substrate ^a	product ^b
11: 25 mol % ^c	12: 75%	17: 25 mol %	18: 88%
13: 25 mol %	14: 35%	19: 125 mol % ^e	20: 65%
15: 25 mol % ^d	16: 33%	21: 125 mol % ^e	22: 75%

^a All reactions were carried out with BF₃-OEt₂ in CH₂Cl₂ at a concentration of 0.01 M. ^b Isolated yields. ^c Entries for substrates **11**, **13**, **19**, and **21** were done at room temperature. ^d Reaction temperature for **15** and **17** went from -78 °C to room temperature. ^e 1.25 equiv of BF₃-OEt₂ was used at room temperature with a concentration of 0.04 M.

sible via this method (**11 → 12** and **13 → 14**), though the seven-membered ring was formed in lower yield. In addition, ynamides **15** and **17** also underwent the hetero ring-closing

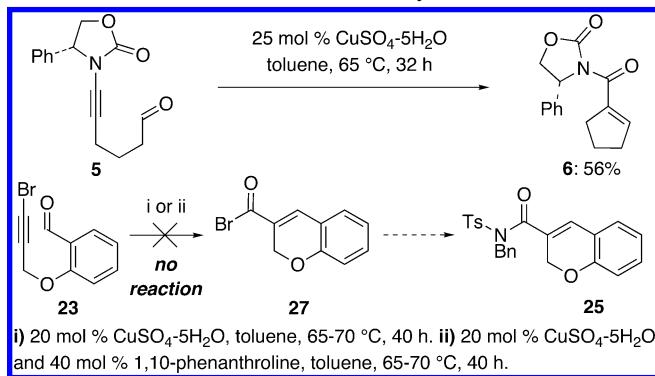
(8) For recent reports on ynamides, see: (a) Dunetz, J. R.; Danheiser, R. L. *J. Am. Chem. Soc.* **2005**, *127*, 5776. (b) Riddell, N.; Villeneuve, K.; Tam, W. *Org. Lett.* **2005**, *7*, 3681. (c) Zhang, Y. *Tetrahedron Lett.* **2005**, *46*, 6483. (d) Martinez-Esperon, M. F.; Rodriguez, D.; Castedo, L.; Saá, C. *Org. Lett.* **2005**, *7*, 2213. (e) Bendikov, M.; Duong, H. M.; Bolanos, E.; Wudl, F. *Org. Lett.* **2005**, *7*, 783. (f) Marion, F.; Coulomb, J.; Courillon, C.; Fensterbank, L.; Malacria, M. *Org. Lett.* **2004**, *6*, 1509. (g) Rosillo, M.; Domínguez, G.; Casarrubios, L.; Amador, U.; Pérez-Castells, J. *J. Org. Chem.* **2004**, *69*, 2084. (h) Couty, S.; Liégault, B.; Meyer, C.; Cossy, J. *Org. Lett.* **2004**, *6*, 2511. (i) Rodríguez, D.; Castedo, L.; Saá, C. *Synlett* **2004**, 783. (j) Rodríguez, D.; Castedo, L.; Saá, C. *Synlett* **2004**, 377. (k) Hirano, S.; Tanaka, R.; Urabe, H.; Sato, F. *Org. Lett.* **2004**, *6*, 727. (l) Klein, M.; König, B. *Tetrahedron* **2004**, *60*, 1087. (m) Marion, F.; Courillon, C.; Malacria, M. *Org. Lett.* **2003**, *5*, 5095. (n) Witulski, B.; Alayrac, C.; Tevzaadze-Saeftel, L. *Angew. Chem., Int. Ed.* **2003**, *42*, 4257. (o) Tanaka, R.; Hirano, S.; Urabe, H.; Sato, F. *Org. Lett.* **2003**, *5*, 67. (p) Witulski, B.; Lumtscher, J.; Bergsträber, U. *Synlett* **2003**, 708. (q) Naud, S.; Cintrat, J.-C. *Synthesis* **2003**, 1391. (r) Witulski, B.; Alayrac, C. *Angew. Chem., Int. Ed.* **2002**, *41*, 3281. (s) Timbart, J.-C.; Cintrat, J.-C. *Chem. Eur. J.* **2002**, *8*, 1637.

metathesis with ketones to give methyl cyclopentenes **16** and **18** in 33% and 89% yields, respectively.

Intriguingly, ynamides **19** and **21** tethered with imides also underwent hetero RCM to give indolizidine motifs **20** and **22**, respectively, in good yields. In these cases, however, 1.25 equiv of $\text{BF}_3\text{-OEt}_2$ had to be used and triflate salts were entirely ineffective. These two reactions suggest that the initial 1,2-addition of ynamides could likely proceed through either the *N*-acyl imidinium intermediate **A** (most likely) or **B** (Table 2) obtained via $\text{BF}_3\text{-OEt}_2$ activation of one of the two imide carbonyl groups.¹⁶

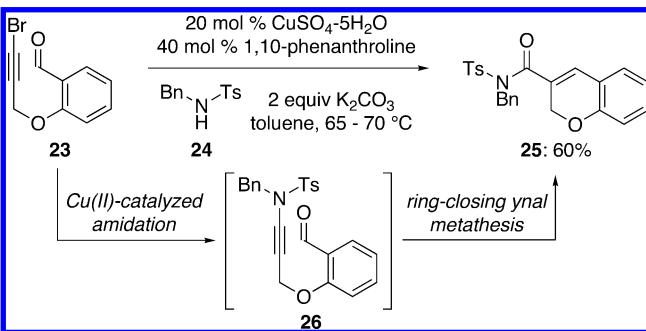
Another interesting example of this ynamide-carbonyl RCM is given in Scheme 2. During the synthesis of ynamide

Scheme 3. $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$ -Catalyzed Metathesis



i) 20 mol % $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$, toluene, 65–70 °C, 40 h. ii) 20 mol % $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$ and 40 mol % 1,10-phenanthroline, toluene, 65–70 °C, 40 h.

Scheme 2. Tandem Amidation–Metathesis



26 under our amidation conditions,^{10b} we discovered that the ynal ring-closing metathesis had occurred concomitantly to give chromene **25** with an overall yield of 60%, constituting a tandem amidation–metathesis.

(9) For our recent work on ynamides, see: (a) Zhang, Y.; Hsung, R. P.; Zhang, X.; Huang, J.; Slafer, B. W.; Davis, A. *Org. Lett.* **2005**, *7*, 1047 (b) Tracey, M. R.; Zhang, Y.; Frederick, M. O.; Mulder, J. A.; Hsung, R. P. *Org. Lett.* **2004**, *6*, 2209. (c) Shen, L.; Hsung, R. P. *Tetrahedron Lett.* **2003**, *44*, 9353. (d) Frederick, M. O.; Hsung, R. P.; Lambeth, R. H.; Mulder, J. A. Tracey, M. R. *Org. Lett.* **2003**, *5*, 2663. (e) Mulder, J. A.; Kurtz, K. C. M.; Hsung, R. P.; Coverdale, H. A.; Frederick, M. O.; Shen, L.; Zifcsak, C. A. *Org. Lett.* **2003**, *5*, 1547. (f) Mulder, J. A.; Hsung, R. P.; Frederick, M. O.; Tracey, M. R.; Zifcsak, C. A. *Org. Lett.* **2002**, *4*, 1383.

(10) For the synthesis of ynamides, see: (a) Frederick, M. O.; Mulder, J. A.; Tracey, M. R.; Hsung, R. P.; Huang, J.; Kurtz, K. C. M.; Shen, L.; Douglas, C. J. *J. Am. Chem. Soc.* **2003**, *125*, 2368. (b) Zhang, Y.; Hsung R. P.; Tracey, M. R.; Kurtz, K. C. M.; Vera, E. L. *Org. Lett.* **2004**, *6*, 1151. Also see: (c) Dunetz, J. R.; Danheiser, R. L. *Org. Lett.* **2003**, *5*, 4011. (d) Couty, S.; Barbazanges, M.; Meyer, C.; Cossy, J. *Synlett* **2005**, 906. (e) Wei, L.-L.; Mulder, J. A.; Xiong, H.; Zifcsak, C. A.; Douglas, C. J.; Hsung, R. P. *Tetrahedron* **2001**, *57*, 459.

(11) For a preliminary disclosure of this work, see: Kurtz, K. C. M.; Hsung, R. P.; Zhang, Y. 229th ACS National Meeting, San Diego, CA, March, 2005; Abstract ORGN-620.

(12) (a) Fuks, R.; Viehe, H. G. *Chem. Ber.* **1970**, *103*, 564. (b) Hsung, R. P.; Zifcsak, C. A.; Wei, L.-L.; Douglas, C. J.; Xiong, H.; Mulder, J. A. *Org. Lett.* **1999**, *1*, 1237.

(13) (a) Mori, S.; Shindo, M. *Org. Lett.* **2004**, *6*, 3945. (b) Shindo, M.; Sato, Y.; Yoshikawa, T.; Koretsune, R.; Shishido, K. *J. Org. Chem.* **2004**, *69*, 3912. (c) Shindo, M.; Matsumoto, K.; Mori, S.; Shishido, K. *J. Am. Chem. Soc.* **2002**, *124*, 6840. (d) For a review, see: Shindo, M. *Synthesis* **2003**, 2275.

(14) Procedures and characterizations for all new compounds can be found in Supporting Information.

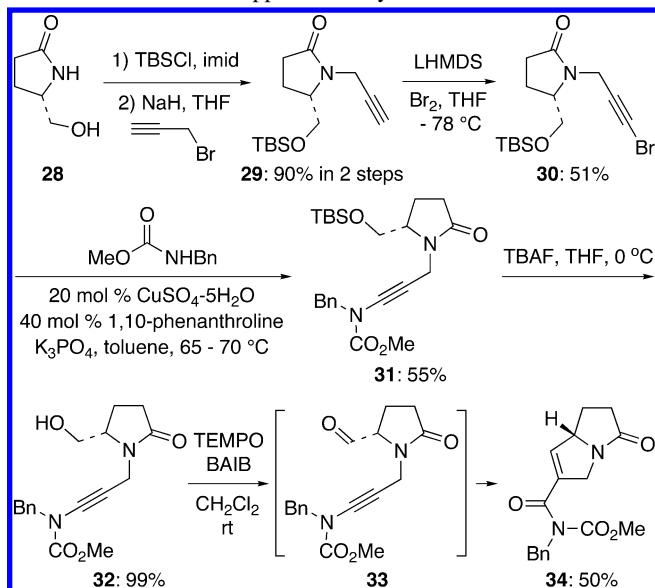
(15) In some of these cases, trace amounts of the α -chloroenamide were observed.

(16) For some examples of related additions to imide carbonyls, see: (a) Lee, J.; Ha, J. D.; Cha, J. K. *J. Am. Chem. Soc.* **1997**, *119*, 8127. Also see recently: (b) Dvornikovs, V.; Hoye, T. R. 229th ACS National Meeting, San Diego, CA, March, 2005; Abstract ORGN-665.

To clarify that this is likely a tandem amidation–metathesis and not tandem metathesis–amidation, we carried out two control studies. As shown in Scheme 3, hetero RCM of ynamide **5** could be achieved using 25 mol % of $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$, although heating at 65 °C was necessary to drive the reaction to completion over 32 h, leading to **6** in 56%. On the other hand, hetero RCM of alkynyl bromide **23** did not occur while heating with 20 mol % of $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$ or 40 mol % of 1,10-phenanthroline in addition to $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$ at 65 °C.¹⁷ This effectively rules out acyl bromide **27** as an intermediate leading to chromene **25**. Because we were intrigued with potential reactivities of alkynyl halides¹⁷ such as **23**, we further attempted the ring-closing metathesis by employing 10 mol % of AgBF_4 at 90 °C in $(\text{CH}_2\text{Cl})_2$ for 14 h, conditions described in Krische's report.⁶ However, we only observed a trace amount of the potential hetero RCM product.

Having established the feasibility of an ynamide–carbonyl ring-closing metathesis, we were able to successfully apply this hetero RCM to the synthesis of an optically enriched

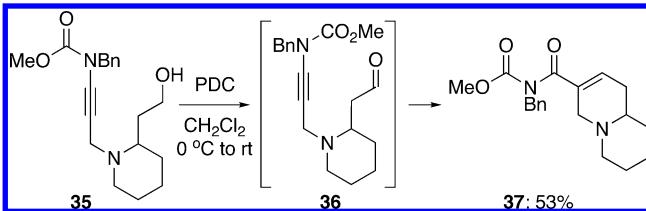
Scheme 4. An Approach to Pyrrolizidine Alkaloids



pyrrolizidine motif. As shown in Scheme 4, starting from (*S*)-5-hydroxymethyl-pyrrolidin-2-one **28**, TBS protection and propargylation of the amide nitrogen gave **29** in good yield over 2 steps. Bromination was carried out with LHMDS and bromine, and subsequent Cu(II)-catalyzed cross-coupling^{10b} yielded ynamide **31**. Removal of the TBS protecting group in **31** was followed by oxidation using TEMPO and BAIB (bisacetoxy-iodobenzene).¹⁸ However, under these oxidative conditions, hetero RCM occurred *in situ* to give pyrrolizidinone **34** in 50% yield.

While this success provides a rather facile approach toward pyrrolizidine alkaloids, we were intrigued by this tandem oxidation–metathesis, especially after finding that PDC oxidation of ynamide **35** also led to quinolizidine **37** in 53% overall yield via the same tandem oxidation–metathesis process (Scheme 5).

Scheme 5. Tandem Oxidation–Metathesis



We suspected that HOAc or a related Brønsted acidic species¹⁹ was promoting the hetero RCM during the oxidation of **32** or **35**. As shown in Scheme 6, when in excess, HOAc could promote the RCM of ynamide **5**, although the reaction is much slower than using $\text{BF}_3\text{-OEt}_2$ (entry 1 versus entries 2–4). A stronger Brønsted acid such as PNBSA^{9d,f} (*p*-nitrobenzenesulfonic acid) was an improvement, albeit still inferior to $\text{BF}_3\text{-OEt}_2$ in terms of the required loading and reaction time (entries 5 and 6). It turned out that HNTf₂ was again a “magic” Brønsted acid^{9a,20,21} with an efficiency comparable to that of $\text{BF}_3\text{-OEt}_2$. By using 25 mol %, the

(17) For a recent account on methods using alkynyl halides, see: Yoo, W.-J.; Allen, A.; Villeneuve, K.; Tam, W. *Org. Lett.* **2005**, 7, ASAP.

(18) De Mico, A.; Margarita, R.; Parlanti, L.; Vescovi, A.; Piancatelli, G. *J. Org. Chem.* **1997**, 62, 6974.

(19) Actually, ynamide **5** (neat) slowly underwent hetero RCM upon standing in the refrigerator, and over a period of 4.5 months, it gave a mixture of **5** and the metathesis product **6** in 2.7:1 ratio.

Scheme 6. Comparison with Brønsted Acids

Detailed description: Scheme 6 compares the conversion of ynamide 5 to product 6 using various Brønsted acids. The table shows the reaction conditions and isolated yields for each entry.

entry	acid	equiv	time [h]	yield [%] ^a
1	$\text{BF}_3\text{-OEt}_2$	0.25	0.25	74
2	HOAc ^b	0.25	15	trace
3	HOAc ^b	1.25	30	40 ^c
4	HOAc ^b	3.25	48	37 ^d
5	PNBSA ^e	0.25	15	trace
6	PNBSA	1.25	0.5	73
7	HNTf ₂	0.25	0.25	64

^a Isolated yields. ^bEntries 2–4 represent the same continuous reaction in which HOAc was added successively. ^cConversion of **5**. ^dIsolated yield at 50% conversion. ^ePNBSA = *p*-nitrobenzenesulfonic acid.

RCM product **6** was isolated in 64% yield after only 15 min at room temperature (entry 7). However, overall none of these reactions was as clean as those promoted by $\text{BF}_3\text{-OEt}_2$. Nevertheless, the feasibility of using Brønsted acids to promote this hetero ring-closing metathesis provides another useful dimension to this methodology, especially when the RCM occurs in a tandem manner.

In summary, an acid-catalyzed ring-closing ynamide–carbonyl metathesis has been developed. This method is applicable to the construction of carbo- as well as oxygen- and nitrogen-containing heterocycles.

Acknowledgment. Authors thank NIH-NIGMS (GM066055) and NSF (CHE-0094005) for support.

Supporting Information Available: Experimental and ¹H NMR spectral and characterizations for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL052487S

(20) For some examples of Brønsted acid catalysis, see: (a) Sun, J.; Kozmin, S. A.; *J. Am. Chem. Soc.* **2005**, 127, 13512. (b) Williams, A. L.; Johnston, J. N. *J. Am. Chem. Soc.* **2004**, 126, 1612. (c) Zhang, L.; Kozmin, S. A. *J. Am. Chem. Soc.* **2004**, 126, 10204. (d) Cossy, J.; Lutz, F.; Alaize, V.; Meyer, C. *Synlett* **2002**, 45. (e) Ishihara, K.; Hiraiwa, Y.; Yamamoto, H. *Synlett* **2001**, 1851.

(21) For a study on the acidity of HNTf₂, see: Thomazeau, C.; Olivier-Bourbigou, H.; Magna, L.; Luts, S.; Gilbert, B. *J. Am. Chem. Soc.* **2003**, 125, 5264.