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Catalytic Enantioselective Dehydrogenative Si–O Coupling to Access Chiroptical Silicon-Stereogenic Siloxanes and Alkoxysilanes

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ABSTRACT: A rhodium-catalyzed enantioselective construction of triorgano-substituted silicon-stereogenic siloxanes and alkoxysilanes is developed. This process undergoes a direct intermolecular dehydrogenative Si–O coupling between dihydrosilanes with silanols or alocohols, giving access to a variety of highly functionalized chiral siloxanes and alkoxysilanes in decent yields with excellent *stereo*control, that significantly expand the chemical space of the silicon-centered chiral molecules. Further utility of this process was illustrated by the construction of CPL-active (circularly polarized luminescence) silicon-stereogenic alkoxysilane small organic molecules. Optically pure bis-alkoxysilane containing two silicon-stereogenic centers and three pyrene groups displayed a remarkable g_{lum} value with a high fluorescence quantum efficiency ($g_{lum} = 0.011$, $\Phi_F = 0.55$), which could have great potential application prospects in chiral organic optoelectronic materials.

S ilicon-containing molecules are of great academic and industrial importance with widespread applications in many areas. In the past few decades, the development of new methods for the preparation of novel types of organosilicon compounds has been intensively studied, which has led to their broad use in a diverse range of organic, organometallic, and polymeric chemistry.¹⁻³ In particular, siloxanes and alkoxysilanes (silyl ethers) are fundamentally important skeletons, which are widely used as privileged monomers in silicon-based materials, and valuable protecting groups, reagents, intermediates in organic synthesis.⁴⁻¹⁰ Despite vast methods for the synthesis of siloxanes and alkoxysilanes,¹⁻¹⁰ the access of these organosilicon compounds bearing silicon-stereogenic centers in enantioenriched forms has been significantly less explored, which severely hampers their applications in the design and development of new chiral silicon-based materials.

Historically, the synthesis of silicon-stereogenic alkoxysilanes relied on the optical resolution and kinetic resolution with chiral alcohol auxiliaries.¹¹⁻¹⁵ Despite the success of these resolution methods, they are limited in substrate scope along with low efficiency and poor atom-economy. Thus, the development of novel synthetic methods toward siliconstereogenic alkoxysilanes based on asymmetric synthesis is more attractive but was proved to be very challenging. In the 1970s, the Corriu group developed a seminal work for the asymmetric alcoholysis of dihydrosilanes in the presence of rhodium catalyst.^{16,17} When a chiral alcohol, such as (-)-menthol, was used in the Si-O coupling reaction, a moderate enantioselectivity of the silicon-stereogenic center was achieved (48% ee, determined after stereospecific nucleophilic displacement, Scheme 1a). Later, Leighton and co-workers realized a chiral alcohol substrate- and chiral copper catalyst-induced variant, which produced the desired silicon-stereogenic alkoxysilanes with high diastereoselectivity (90:10 dr, Scheme 1a).¹⁸ In addition, a number of asymmetric substitutions of one of two amino or alkoxy groups or chlorine

atoms at the silicon atom has also been demonstrated for the generation of silicon-stereogenic alkoxysilanes via the control of stoichiometric chiral reagents or chiral auxiliaries.¹⁹⁻²⁵ Apart from the stoichiometric chiral substrates-induced approach, asymmetric catalysis for the construction of silicon-stereogenic silanes is undoubtably more fascinating.²⁶⁻³⁴ To the best of our knowledge, the first example for the catalytic enantioselective alcoholysis of dihydrosilanes with achiral alcohols was also reported by Corriu and co-workers in the 1970s using a chiral Rh/(S,S)-DIOP catalyst (19% ee obtained).^{16,17} Later, the Xu group made several efforts to access silicon-stereogenic alkoxysilanes via enantioselective alcoholysis of dihydrosilanes^{35,36} (Scheme 1b). Besides the alcoholysis of dihydrosilanes, a few examples of transition-metal-catalyzed enantiose-lective hydrosilylation of ketones $^{37-39}$ and desymmetrization of tetraorganosilanes via cleavage of a Si-C bond along with formation of a Si-O bond^{40,41} could also produce the siliconstereogenic alkoxysilanes. It is noteworthy that the catalytic asymmetric synthesis of enantioenriched silicon-stereogenic siloxanes has remained unknown up to date. Given the increasing demand for the synthesis of novel functionalmaterials-oriented chiral organosilicon compounds, the exploration and development of new general strategies for the facile access of silicon-stereogenic siloxanes and alkoxysilanes with high efficiency and enantioselectivity are highly desirable.

Recently, we have reported an enantioselective intramolecular C–H silylation strategy for the synthesis of enantioenriched silicon-stereogenic silanes.^{42–45} With the

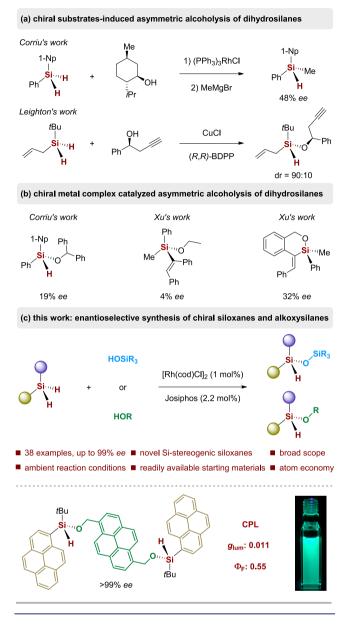
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Scheme 1. Asymmetric Synthesis of Si-Stereogenic Alkoxysilanes



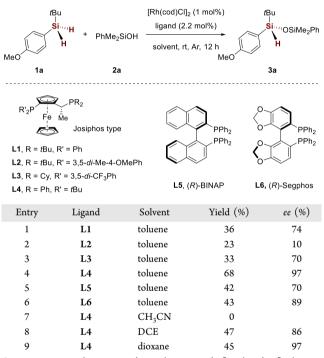
continued interest in the construction of novel siliconstereogenic organosilicon compounds, we questioned whether we could expand the toolbox to achieve a general intermolecular dehydrogenative Si-H/O-H coupling, enabling the streamlined synthesis of silicon-stereogenic siloxanes and alkoxysilanes. Herein, we report the development of a rhodium-catalyzed enantioselective intermolecular dehydrogenative Si-O coupling reaction between dihydrosilanes with silanols or alocohols, which gives access to a wide range of highly functionalized chiral siloxanes and alkoxysilanes in good to excellent yields and enatioselectivities (Scheme 1c). Further utility of this enantioselective dehydrogenative Si-O coupling was illustrated by the construction of CPL-active (circularly polarized luminescence) silicon-stereogenic alkoxysilane small organic molecules. Optically pure bis-alkoxysilane containing two silicon-stereogenic centers and three pyrene groups displayed a remarkable g_{lum} value with a high fluorescence quantum efficiency ($g_{lum} = 0.011$, $\Phi_F = 0.55$), which could have

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great potential applications in chiral organic optoelectronic materials.

At the outset of our research, we mainly focused on the catalytic asymmetric synthesis of enantioenriched siliconstereogenic siloxanes. We commenced our studies by examining the intermolecular dehydrogenative Si–O coupling reaction of *tert*-butyl(4-methoxyphenyl)silane **1a** with dimethyl(phenyl)silanol **2a** in the presence of Rh catalyst. At room temperature in the solvent of toluene, several chiral diphosphine ligands were tested. Treatment of **1a** with **2a** by using $[Rh(cod)Cl]_2$ (1 mol %) as the catalyst and Josiphos L**1** (2.2 mol %) as the chiral ligand successfully delivered the desired Si–O coupling silicon-stereogenic siloxane product **3a** in 36% yield with good enantiocontrol (74% *ee*) (Table 1,

Table 1. Development of Reaction Conditions^a

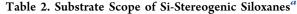


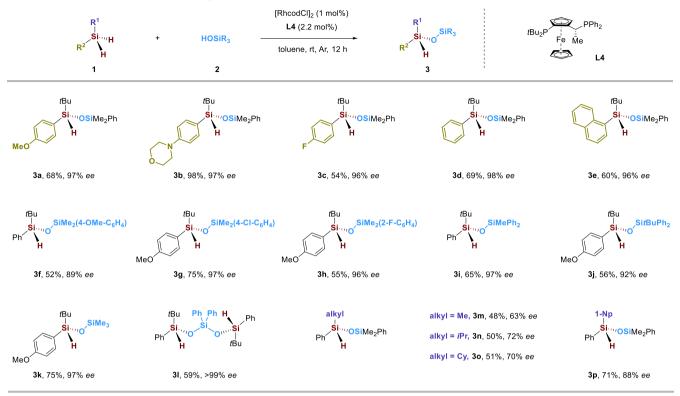
^{*a*}Conditions: **1a** (0.24 mmol), **2a** (0.2 mmol), $[Rh(cod)Cl]_2$ (1 mol %), ligand (2.2 mol %), in 2.0 mL solvent, under argon atmosphere, 12 h. The yield was determined by ¹H NMR using CH_2Br_2 as internal standard. The *ee* values were determined by chiral HPLC.

entry 1). Other Josiphos type ligands with different electronic properties (L2 and L3) did not increase the yields and enantioselectivities (Table 1, entries 2 and 3). Interestingly, when exchanging the *t*Bu-substituent and the Ph-substituent on the two phosphine atoms of the Josiphos ligand (L4), both the yield and enantioselectivity were sharply increased affording 68% yield with 97% *ee* (Table 1, entry 4). Further examination of the chiral ligands found that BINAP L5 and Segphos L6 were also effective for this reaction, albeit in a relatively lower yields and *ee* (Table 1, entries 5–6). Investigation of other common solvents in the presence of L4 disclosed that DCE (1,2-dichloroethane) and 1,4-dioxane could also afford moderate yields of the desired siloxane 3a with good enantioselectivities, while MeCN failed to produce the corresponding product (Table 1, entries 7–9).

Having identified the optimized conditions for the enantioselective dehydrogenative Si-O coupling reaction, we

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"Conditions: 1 (0.24 mmol), 2 (0.2 mmol), [Rh(cod)Cl]₂ (1 mol %), L4 (2.2 mol %), in 2.0 mL toluene, under argon atmosphere, 12 h. Isolated yields. The *ee* values were determined by chiral HPLC.

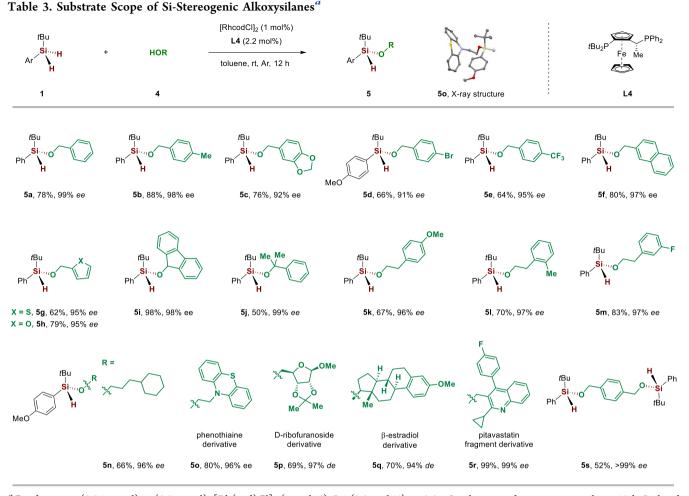
next assessed the scope of this transformation to establish the general methodology for the construction of enantioenriched silicon-stereogenic siloxanes (Table 2). First, the aromatic rings (yellow part) of dihydrosilane substrates bearing different functional groups, including an electron-donating methoxy group (1a), amino group (1b), electron-withdrawing fluoro group (1c), and phenyl, naphthyl groups (1d, 1e), all reacted smoothly with dimethyl(phenyl)silanol 2a to afford the desired silicon-stereogenic siloxane products 3a-3e in moderate to good yields (54-98%) with excellent enantioselectivities (96-98% ee). Second, the scope of silanols (blue part) was investigated under the standard conditions. Methoxy, chloro, and fluoro groups (2f-2h) on the aromatic ring of the dimethylphenylsilanols were well tolerated in the reaction, producing the corresponding siloxanes 3f-3h in good yields with 89-97% ee. Moreover, methyldiphenylsilanol (1i), tertbutyldiphenylsilanol (1j), and trimethylsilanol (1k) were also found to be competent substrates in the reaction, giving the desired product 3i-3k in 56-75% yields without the loss of enantioselectivities (92-97% ee). It is noteworthy that the enantiopure bis-silicon-stereogenic siloxane 3l (>99% ee) could easily be constructed via this strategy by the double coupling of silanediol with dihydrosilane, which could be highly attractive to organic material chemists due to their wide applications as privileged monomers in silicon-based materials. 46-48 Third, replacing the bulky *t*Bu group into a methyl (1m), isopropyl (1n), cyclohexyl (1o), or 1-naphthyl (1p) group on the dihydrosilane gave the corresponding siloxane products 3m-3p in moderate to good yields with 63-88% ee (purple part).

Since this intermolecular dehydrogenative Si-O coupling reaction enables the facile access to silicon-stereogenic siloxane

derivatives with high enantioselectivity, we questioned whether we could expand this process to the coupling of dihydrosilane with alcohol, challenging the historic problem of the catalytic enantioselective alcoholysis of dihydrosilane for the construction of architecturally complex and functionally diverse enantioenriched silicon-stereogenic alkoxysilanes with high efficiency and enantioselectivity. On the basis of the established reaction conditions (for a detailed account of the optimization study, see Supporting Information Table S1), a wide range of alcohols displaying a variety of substituents were found to be suitable substrates in the Rh-catalyzed enantioselective dehydrogenative Si-O coupling reaction (Table 3). Benzylic alcohol substrates bearing various electron-donating or withdrawing substituents, such as methyl, methoxy, bromo, and trifluoromethyl, were all well accommodated in this transformation to deliver the corresponding silicon-stereogenic alkoxysilanes in 64-88% yields with 91-99% ee (5a-5e). This process was also effective with naphthalene (5f) and heterocycles such as furan (5g) and thiophene (5h). Sterically hindered secondary and tertiary alcohols 5i and 5j also proceeded smoothly in the reaction without any difficulty affording the desired products with excellent enantioselectivities (98-99% ee). Besides benzylic alcohol, aliphatic alcohols with a number of different substituents were also suitable substrates undergoing this dehydrogenative Si-O coupling giving the corresponding products 5k-5n in 66-83% yields with 96-97% ee. To further illustrate the utility of this methodology, we then examined the reaction employing the core structures of several bioactive molecules, pharmaceuticals, or materials building blocks, such as phenothiaine (40), D-ribofuranoside (4p), β -estradiol (4q),

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^{*a*}Conditions: 1 (0.24 mmol), 4 (0.2 mmol), $[Rh(cod)Cl]_2$ (1 mol %), L4 (2.2 mol %), in 2.0 mL toluene, under argon atmosphere, 12 h. Isolated yields. The *ee* values were determined by chiral HPLC. X-ray crystallographic analysis of **50** allowed determination of the absolute configuration; and configurations of the products **5** were assigned by analogy.

and pitavastatin fragment (4r). We were delighted to find that the corresponding silicon-stereogenic alkoxysilane products 5o-5r could be obtained in good yields with excellent stereoselectivities, irrespective of existing diverse functional groups and complex molecular structures. Finally, the interesting bis-silicon-stereogenic alkoxysilane 5s was also obtained by the double coupling between 1,4-phenylenedimethanol with dihydrosilane in 52% yield with excellent enantioselectivity (>99% ee).

In recent years, there has been growing interest in the chiroptical molecules that display circularly polarized luminescence (CPL), which has great potential applications in 3D displays, communication of spin information, information storage and processing, CPL lasers, and biological probes.⁴⁹⁻⁵³ Among various chiral luminescent systems, chromophores with silicon-stereogenic centers exhibited potential intense CPL emission properties due to their $\sigma^* - \pi^*$ conjugation.^{54–59} To demonstrate the utility of these interesting chiral organosilicon compounds, we finally synthesized three enantioenriched silicon-stereogenic alkoxysilanes 6a-6c containing pyrene groups⁶⁰ under the standard conditions (Figure 1a) and examined their photophysical properties. Under UV light irradiation (365 nm), the solution of 6a and 6b in CH₂Cl₂ exhibited blue luminescence, while 6c displays bright green luminescence (Figure 1b). These compounds show similar

UV-vis spectra; the absorption maxima are at around 317, 332, and 349 nm, respectively (Figure 1c). The emission maxima vary from 396 to 508 nm. Both the monomer (around 400 nm) and excimer (around 500 nm) emission peaks were observed for 6a and 6b. However, only a red-shifted, broadened excimer emission peak (around 500 nm) was seen for 6c, which may originate from the multiple pyrene units within the molecule in proximity (Figure 1d).6 It is worthy of mention that 6c has a larger Stokes shifts (159 nm) and a higher absolute fluorescence quantum yield (0.55)compared with 6a and 6b (see Supporting Information Table S2 for details), which is suitable for the potential applications in chiroptical materials. Then, we further investigated the chiroptical properties of these π -conjugated silicon-stereogenic alkoxysilanes via circular dichroism (CD) and circular polarized luminescence (CPL) spectra. The CD spectra are mirror images of each other, in which 6a, 6b, and their enantiomers display clear Cotton effects at about 330 and 348 nm, while (R, R)-6c and (S, S)-6c exhibit stronger opposite signals at around 330 and 355 nm, respectively (Figure 1e). Finally, the CPL experiment which reflected the chirality of emitting excited states was conducted. To our delight, 6a-6c are all CPL-active, displaying clear mirror images of each pair of the enantiomers (Figure 1f). It is noteworthy that (R, R)-6c and (S, S)-6c containing two silicon-stereogenic centers and

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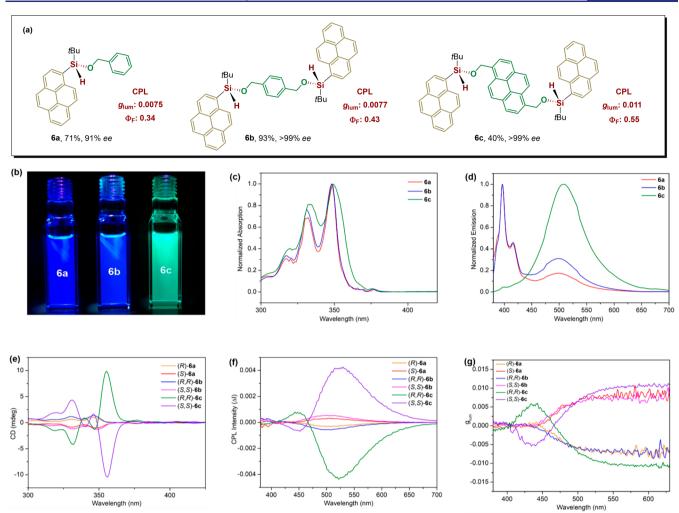


Figure 1. Photophysical properties investigations. (a) Enantioselective synthesis of effective CPL-active silicon-stereogenic alkoxysilanes. (b) Fluorescence images of π -conjugated silicon-stereogenic alkoxysilanes ($\lambda_{ex} = 365 \text{ nm}$). (c) Absorption spectra of π -conjugated silicon-stereogenic alkoxysilanes in CH₂Cl₂ (10⁻⁵ M). (d) Emission spectra of π -conjugated silicon-stereogenic alkoxysilanes in CH₂Cl₂ (10⁻⁵ M). (d) Emission spectra of π -conjugated silicon-stereogenic alkoxysilanes in CH₂Cl₂ (10⁻⁵ M). (e) CD spectra of **6a–6c** and their enantiomers in CH₂Cl₂ (2.0 × 10⁻⁵ M) at room temperature. (f) CPL spectra of **6a–6c** and their enantiomers in CH₂Cl₂ (1.0 × 10⁻³ M) at room temperature, excited at 349 nm. (g) g_{lum} values–wavelength curve for **6a–6c** and their enantiomers.

three pyrene groups show intense CPL signals that range from 470 to 675 nm. The luminescence dissymmetry factors (g_{lum}) were measured as -1.1×10^{-2} for (R, R)-6c and $+1.1 \times 10^{-2}$ for (S, S)-6c at 573 nm (Figure 1g). To the best of our knowledge, it is fairly rare that small organic compounds with a stereogenic center exhibit such a high g_{lum} value with a high fluorescence quantum efficiency ($g_{lum} = 0.011$, $\Phi_F = 0.55$).⁶² To further demonstrate the importance of silicon-stereogenic centers in these chiroptical alkoxysilanes, the chiral-at-carbon linked pyrene analogue of 6a was synthesized and numbered as 6a'. The photophysical properties showed that the absorption and CD spectra of 6a' were similar to 6a. However, the excimer emission peak of 6a' became structureless. In addition, both the CPL intensity and $\Phi_{\rm F}$ of **6a'** ($g_{\rm lum}$ = 0.0030, $\Phi_{\rm F}$ = 0.07) were lower than 6a (see Supporting Information Table S2 and Figure S1 for details). This comparison indicates that chromophores linked with silicon-stereogenic centers could be quite attractive. We believe that these easily accessed siliconstereogenic alkoxysilanes with intense CPL properties could have great potential application in the development of interesting chiroptical devices in the near future.

In summary, we have developed a rhodium-catalyzed enantioselective intermolecular dehydrogenative Si–O coupling reaction between dihydrosilanes with silanols or alocohols, which gives access to a wide range of highly functionalized chiral siloxanes and alkoxysilanes in good to excellent yields and enatioselectivities. Further utility of this process was illustrated by the construction of CPL-active silicon-stereogenic alkoxysilane small organic molecules displaying large $g_{\rm lum}$ values with high emission efficiency, which could have great potential applications in chiral organic optoelectronic materials.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.1c01106.

Materials and methods, experimental procedures, optimization studies, characterization data, photophysical properties studies, ¹H, ¹³C, and ¹⁹F NMR spectra, HPLC spectra and mass spectrometry data of new compounds (PDF)

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Accession Codes

CCDC 2053379 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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