

Supporting Information

A versatile tripodal amide receptor for the encapsulation of anions or hydrated anions *via* formation of dimeric capsules

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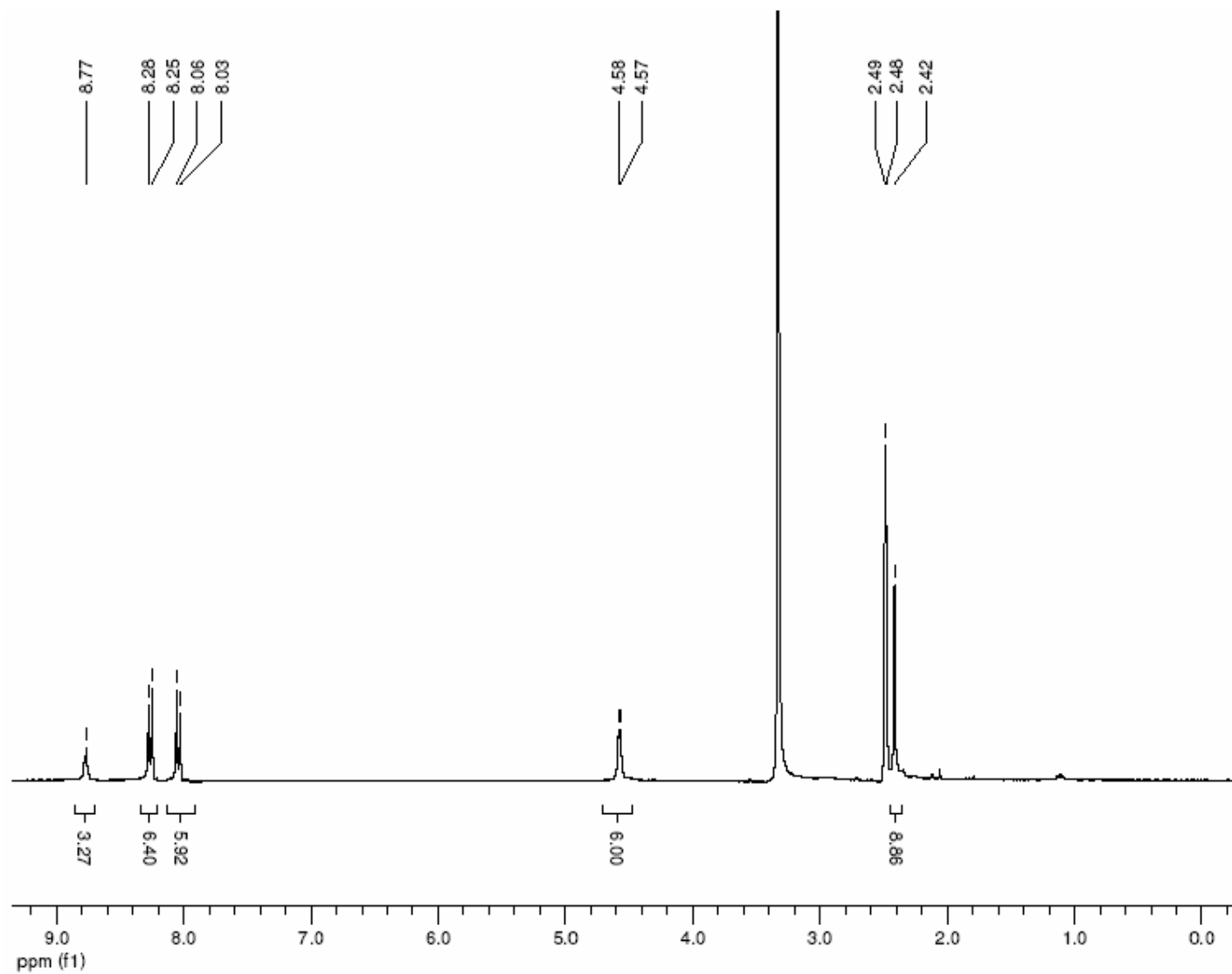


Figure S1. ¹H-NMR spectra of Complex L¹ in DMSO-d₆ at 25°C

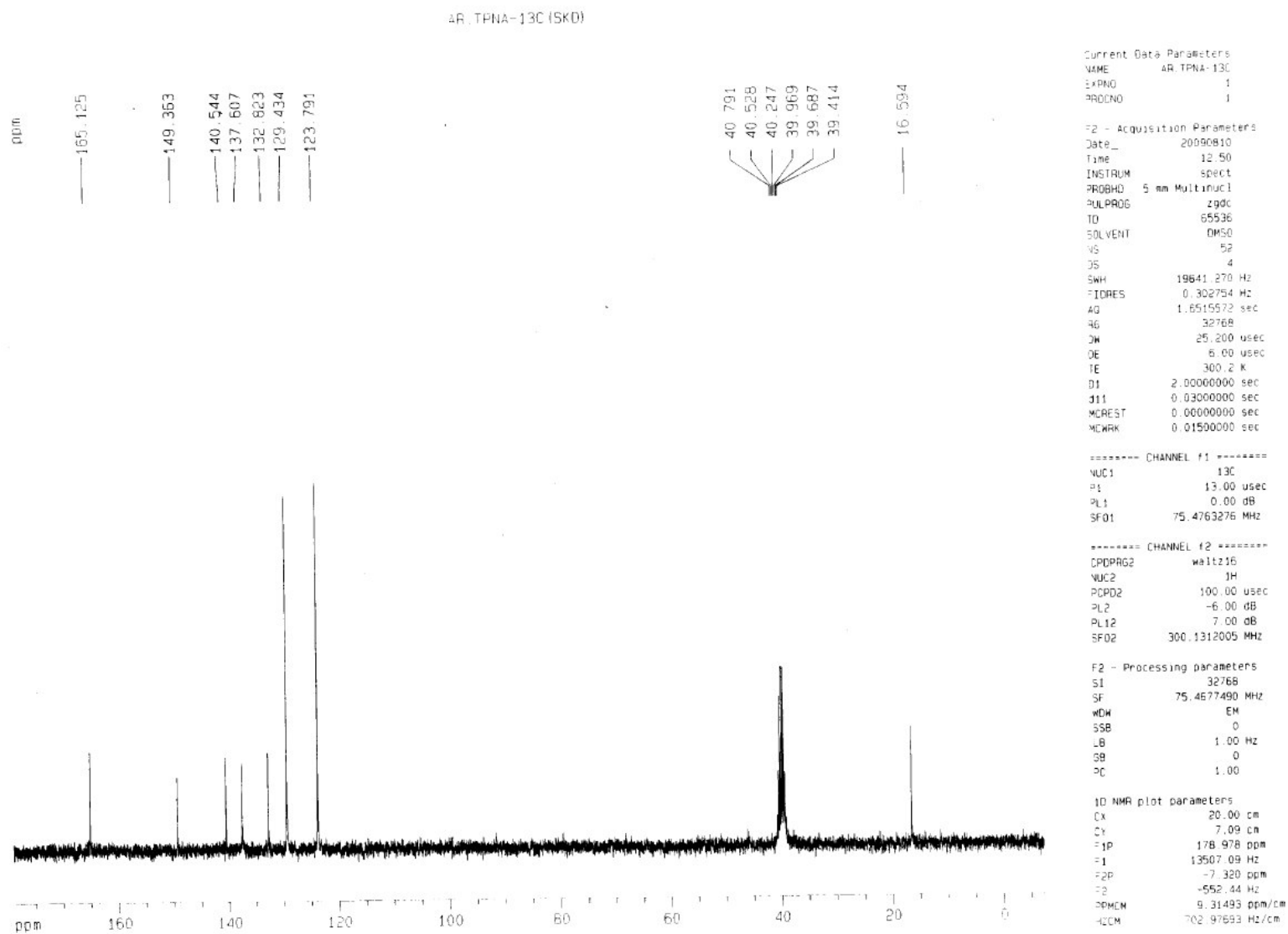


Figure S2. ^{13}C -NMR spectra of L^1 in DMSO-d_6 at 25°C

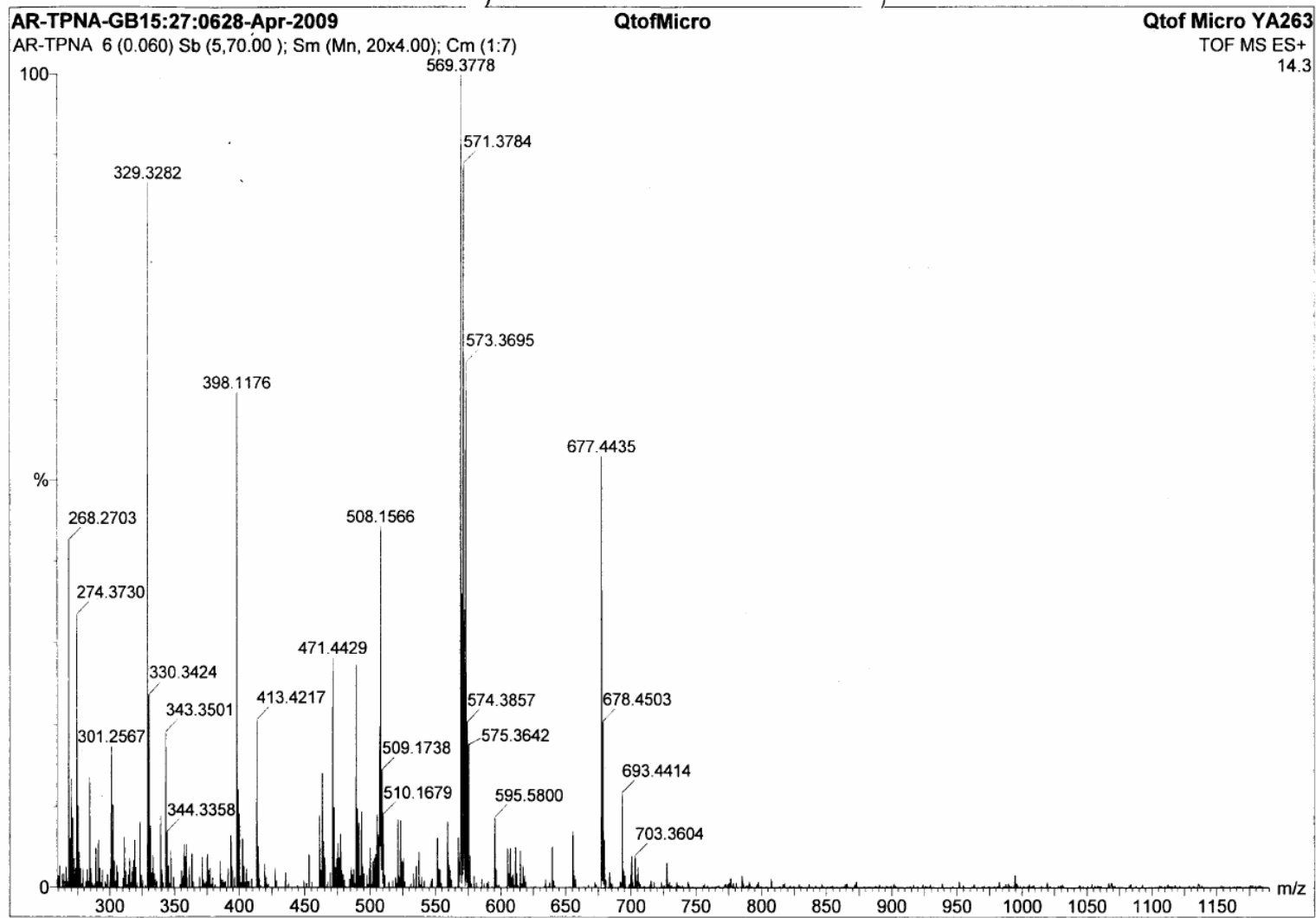


Figure S3. HRMS(ESI) spectrum of L^1 .

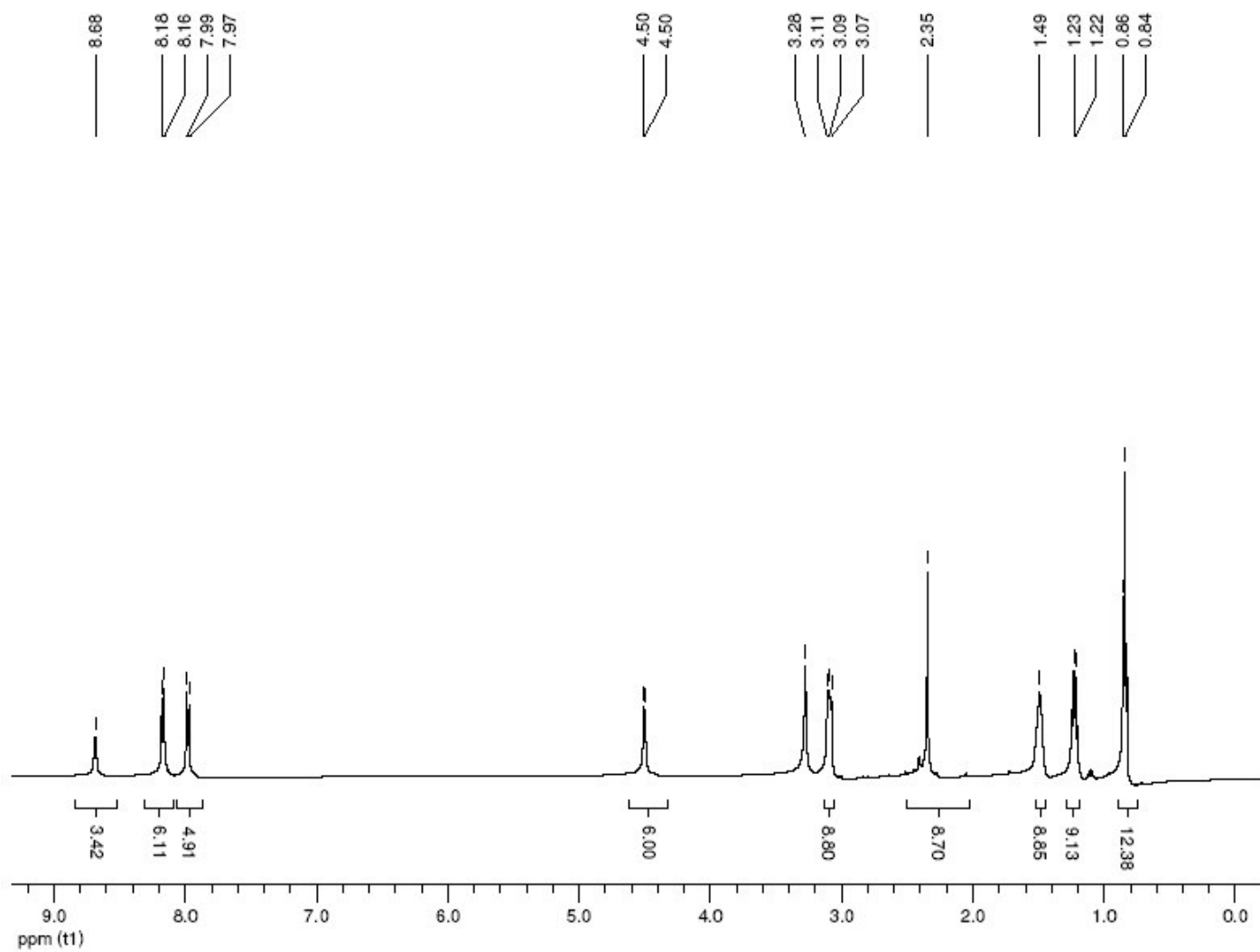


Figure S4. ^1H -NMR spectra of Complex 1 in DMSO-d_6 at 25°C

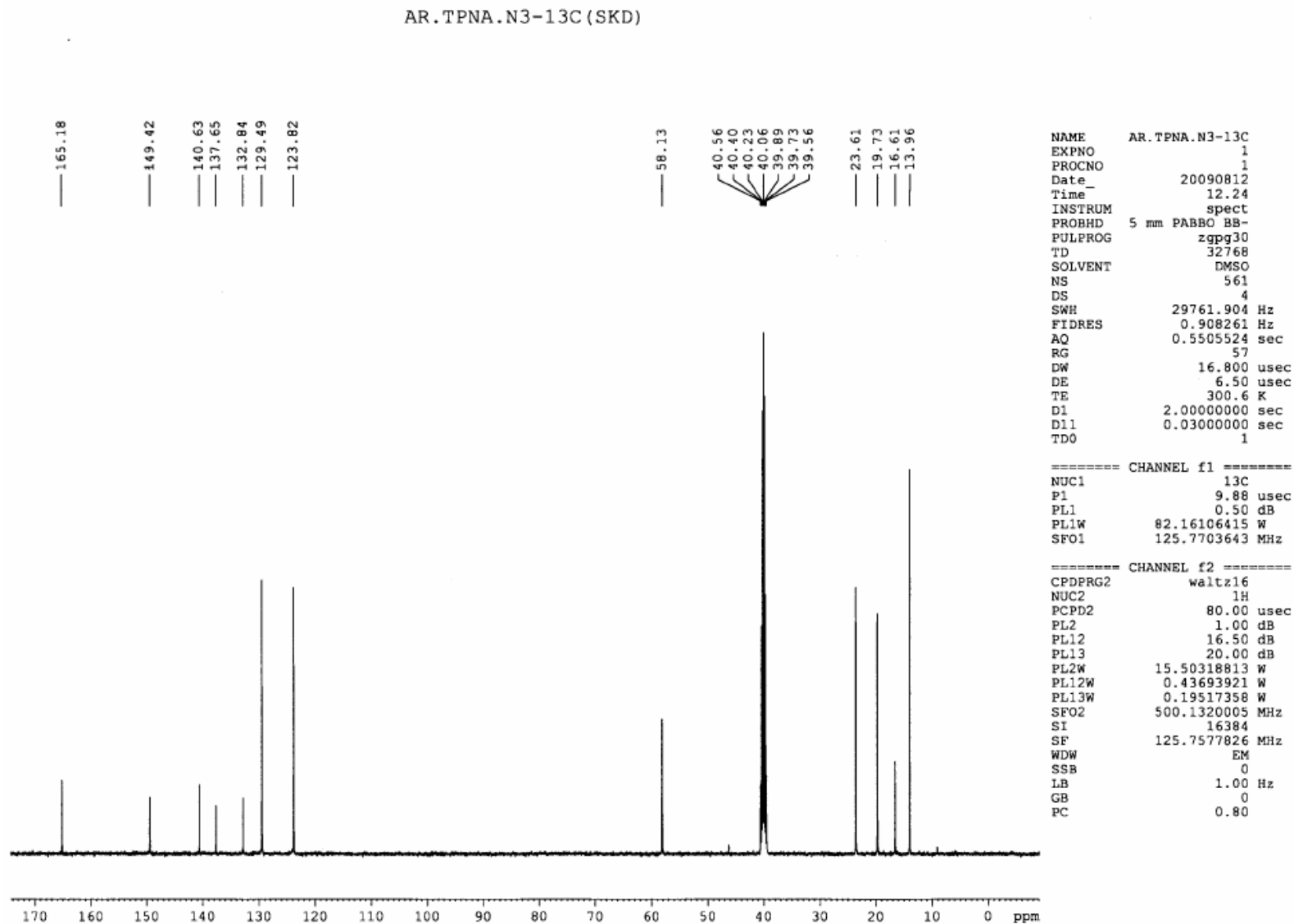


Figure S5. ^{13}C -NMR spectrum of Complex 1 DMSO- d_6 at 25°C.

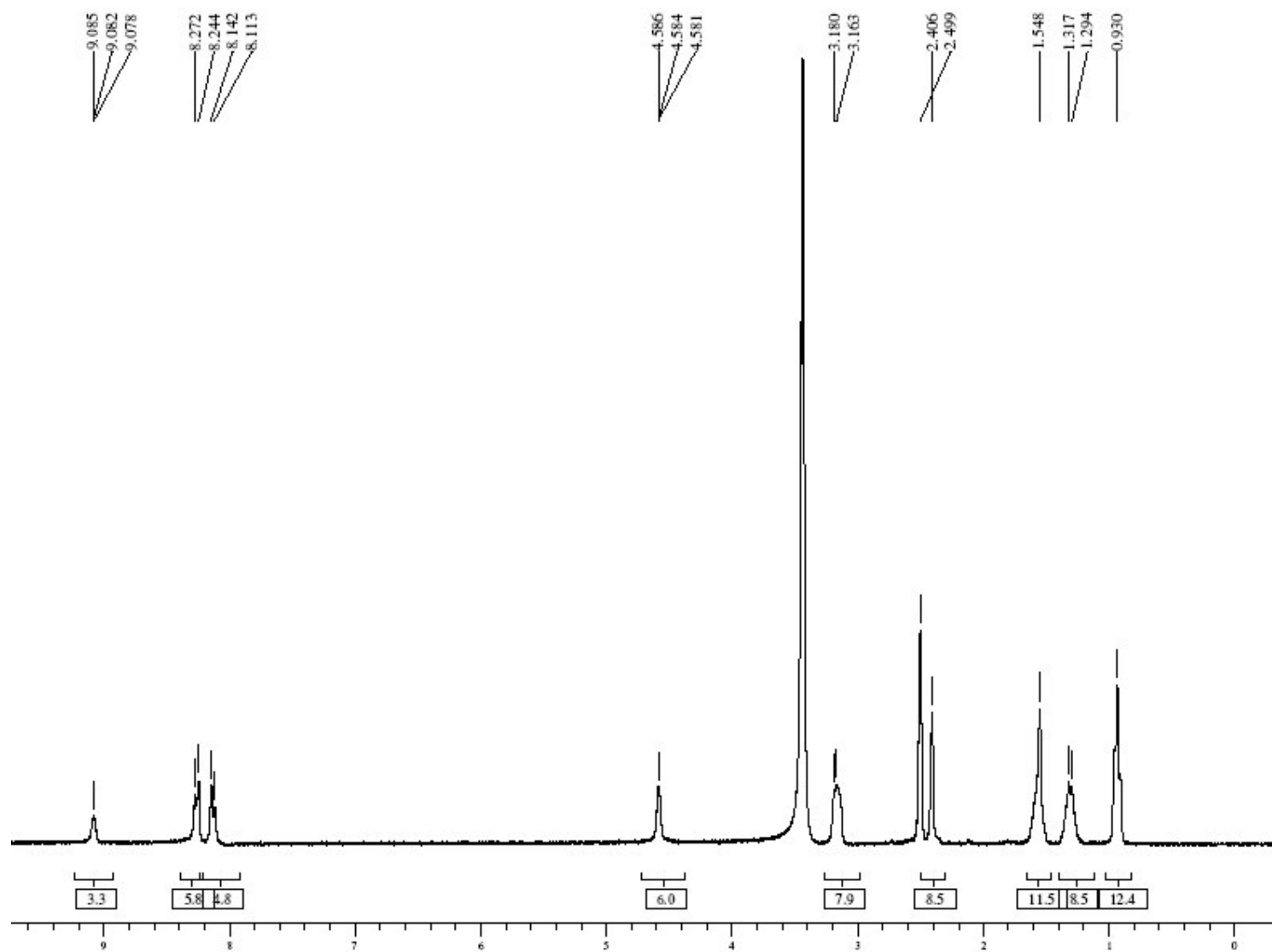


Figure S6. ^1H -NMR spectrum of Complex 2 DMSO-d_6 at 25°C .

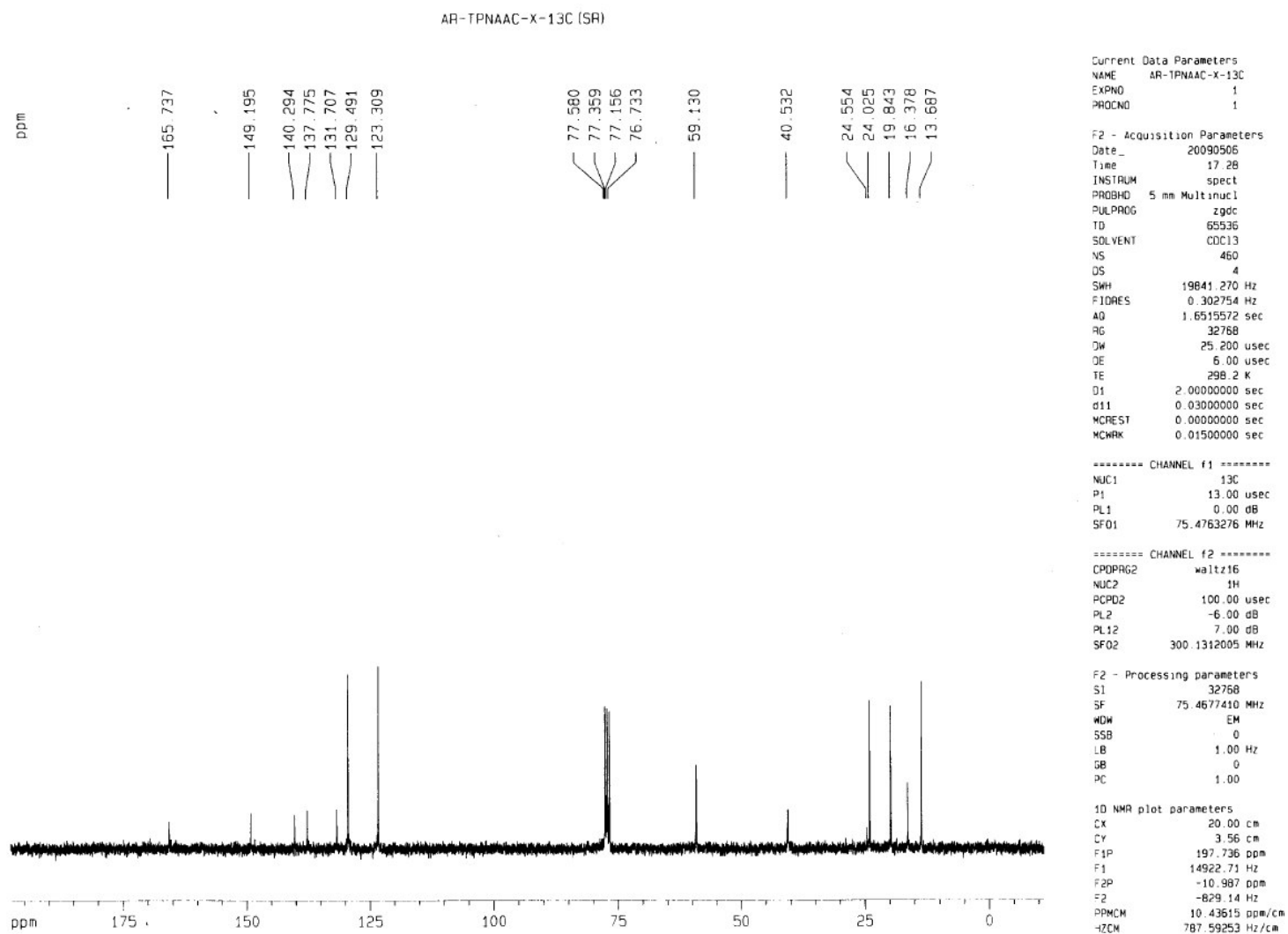


Figure S7. ^{13}C -NMR spectrum of Complex 2 CDCl_3 at 25°C .

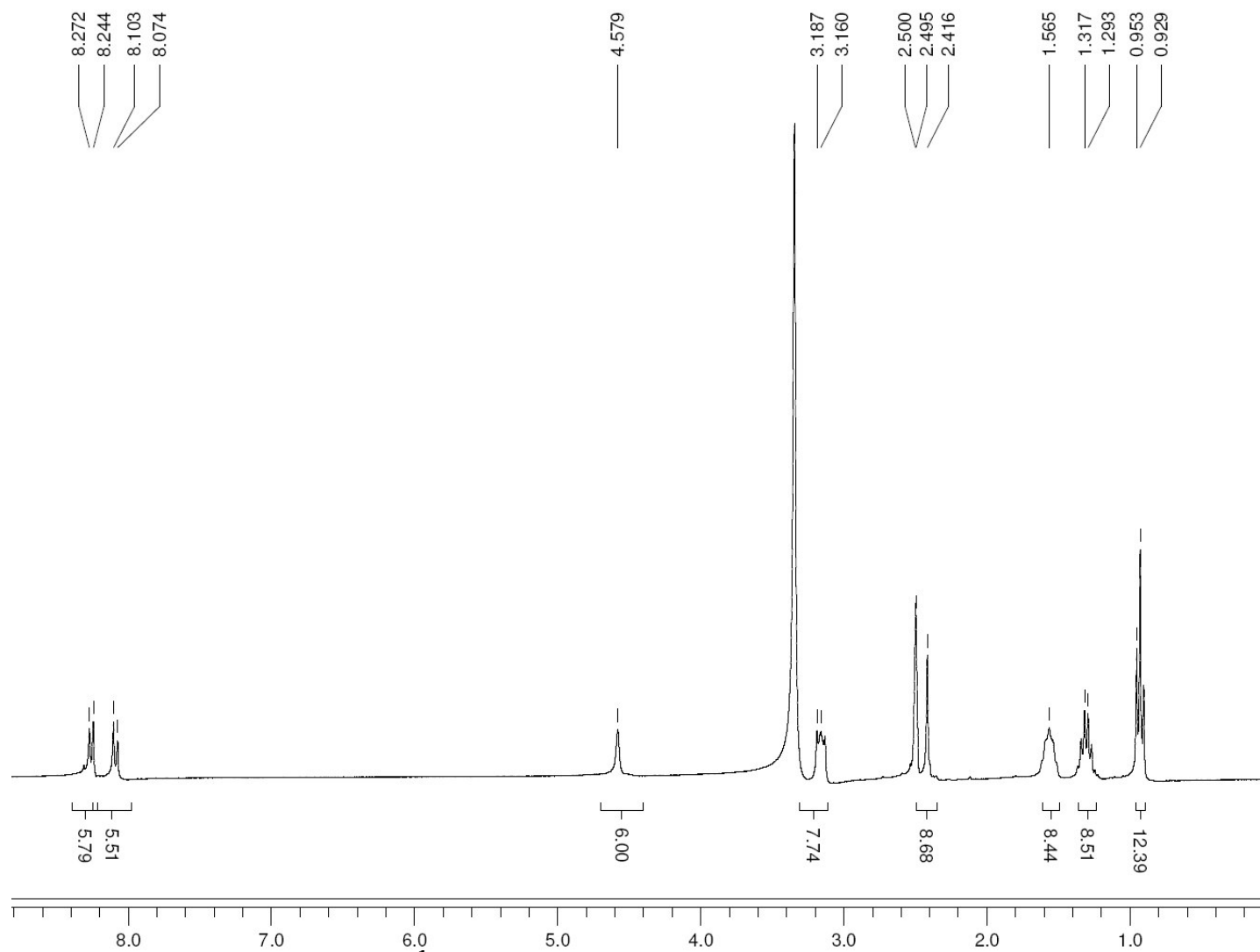


Figure S8. ^1H -NMR spectrum of Complex 3 DMSO-d_6 at 25°C .

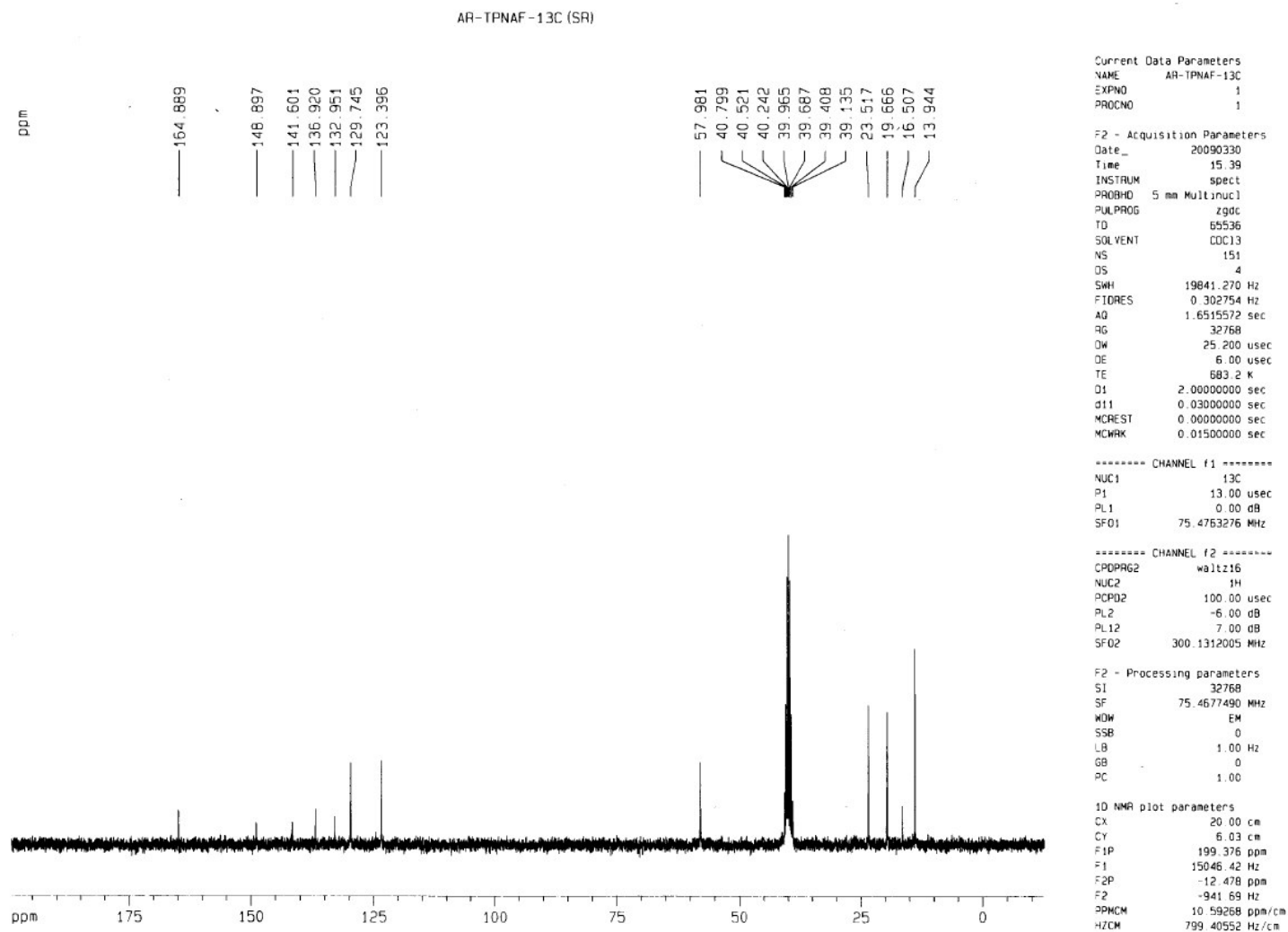


Figure S9. ^{13}C -NMR spectrum of Complex 3 DMSO- d_6 at 25°C .

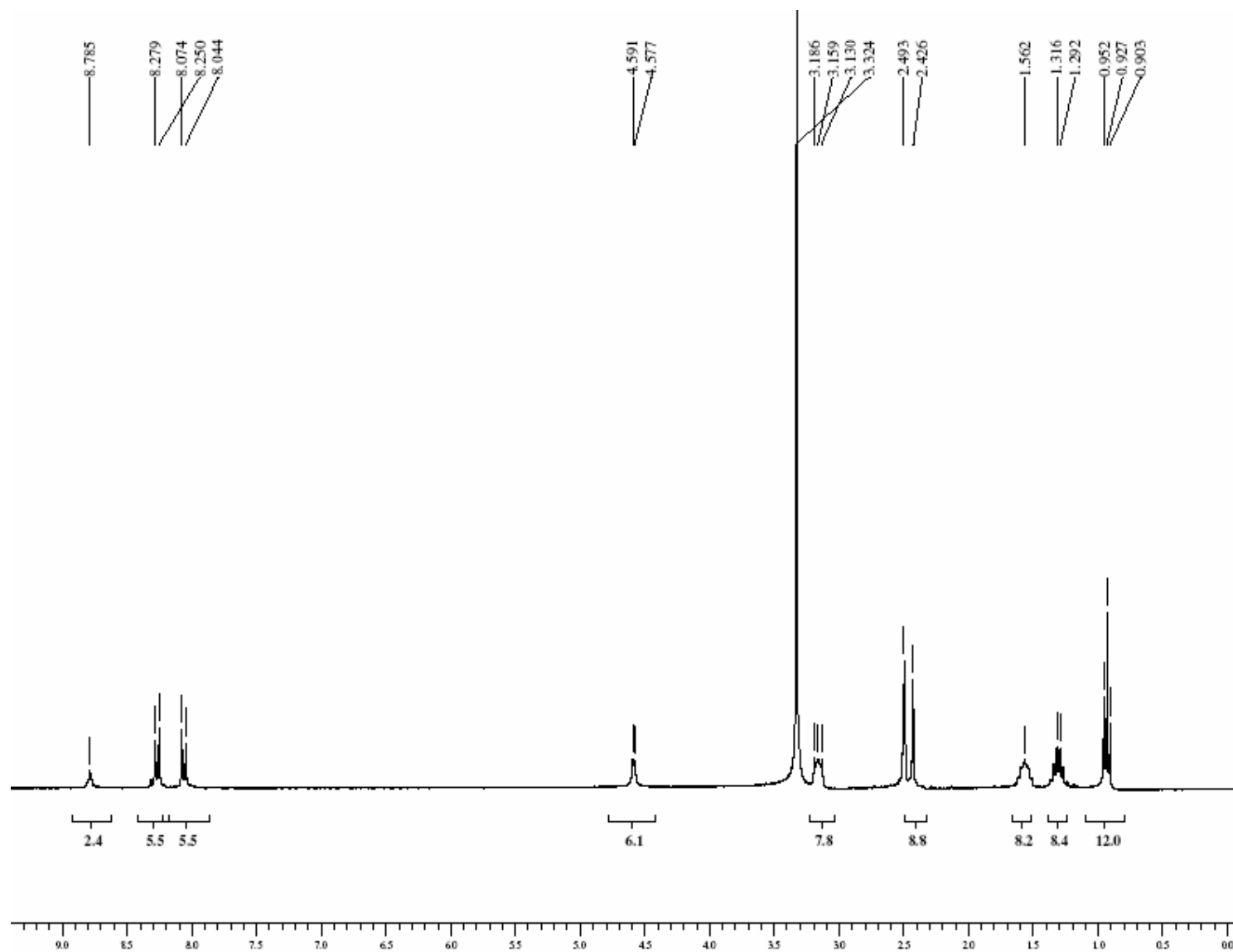


Figure S10. ¹H-NMR spectrum of Complex 4 DMSO-d₆ at 25°C.

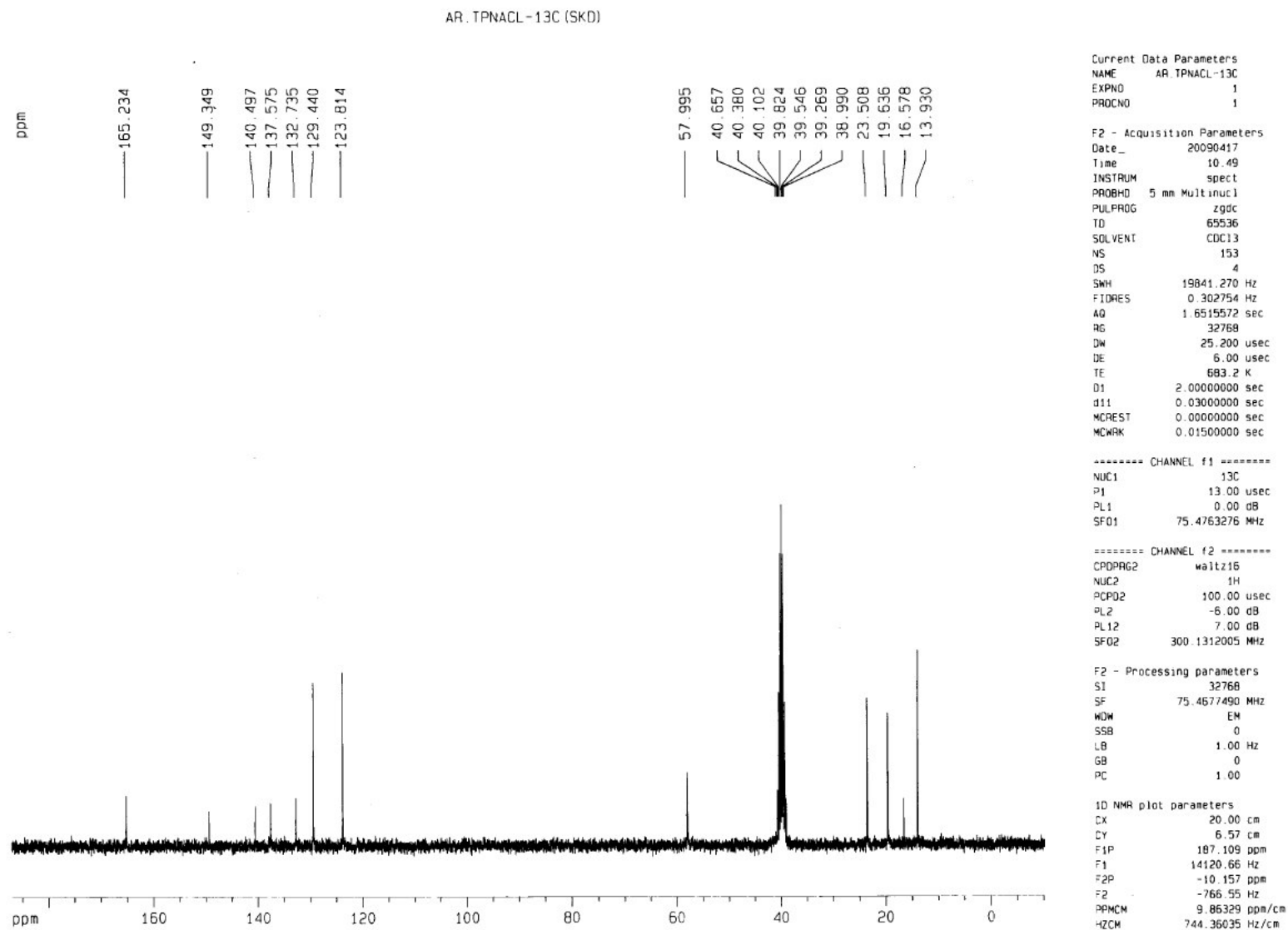


Figure S11. ^{13}C -NMR spectrum of Complex 4 DMSO- d_6 at 25°C.

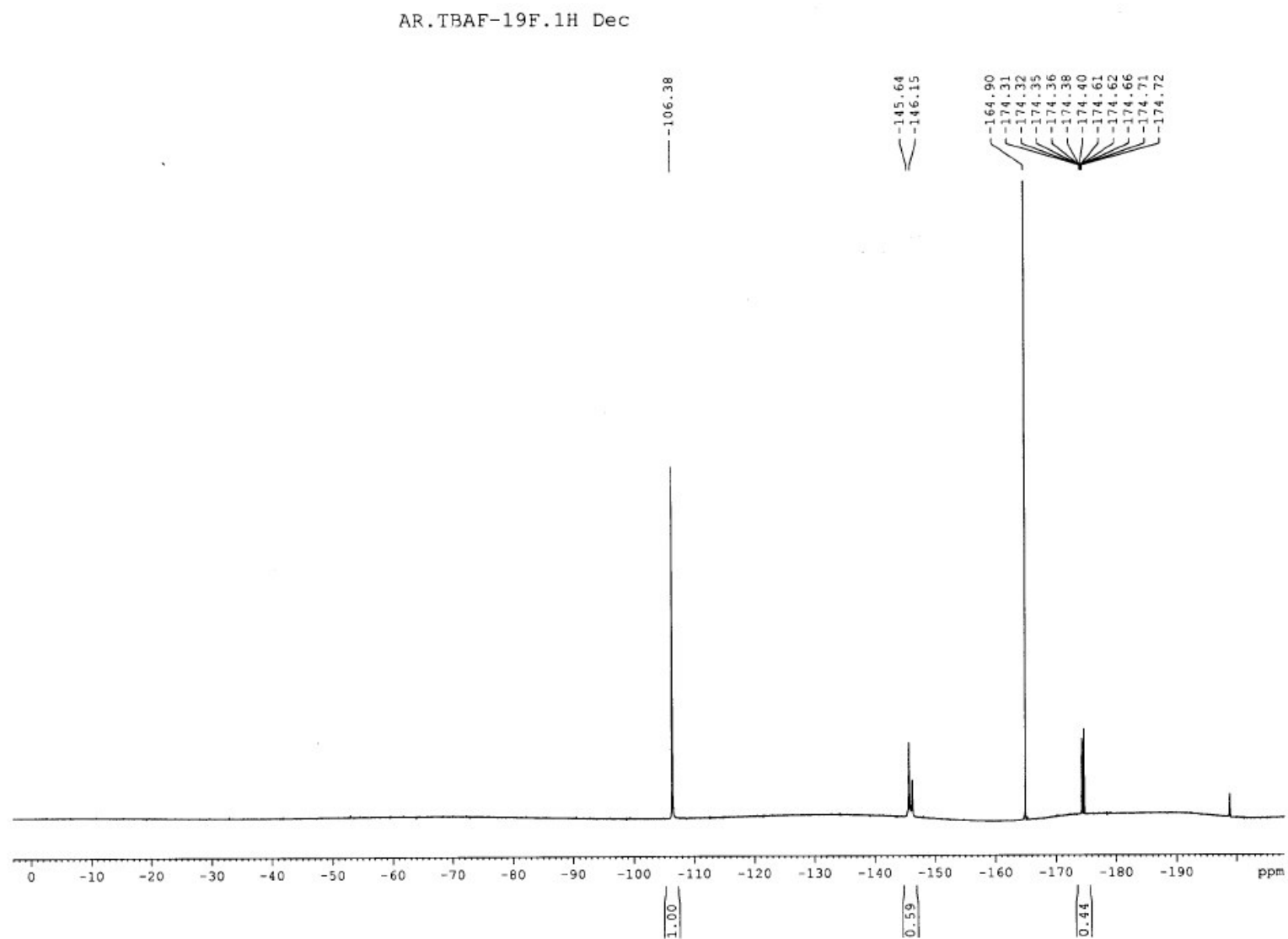


Figure S12. ^{19}F -NMR spectra of TBA-F in $\text{DMSO-}d_6$ at 25°C . Hexafluorobenzene was used as standard.

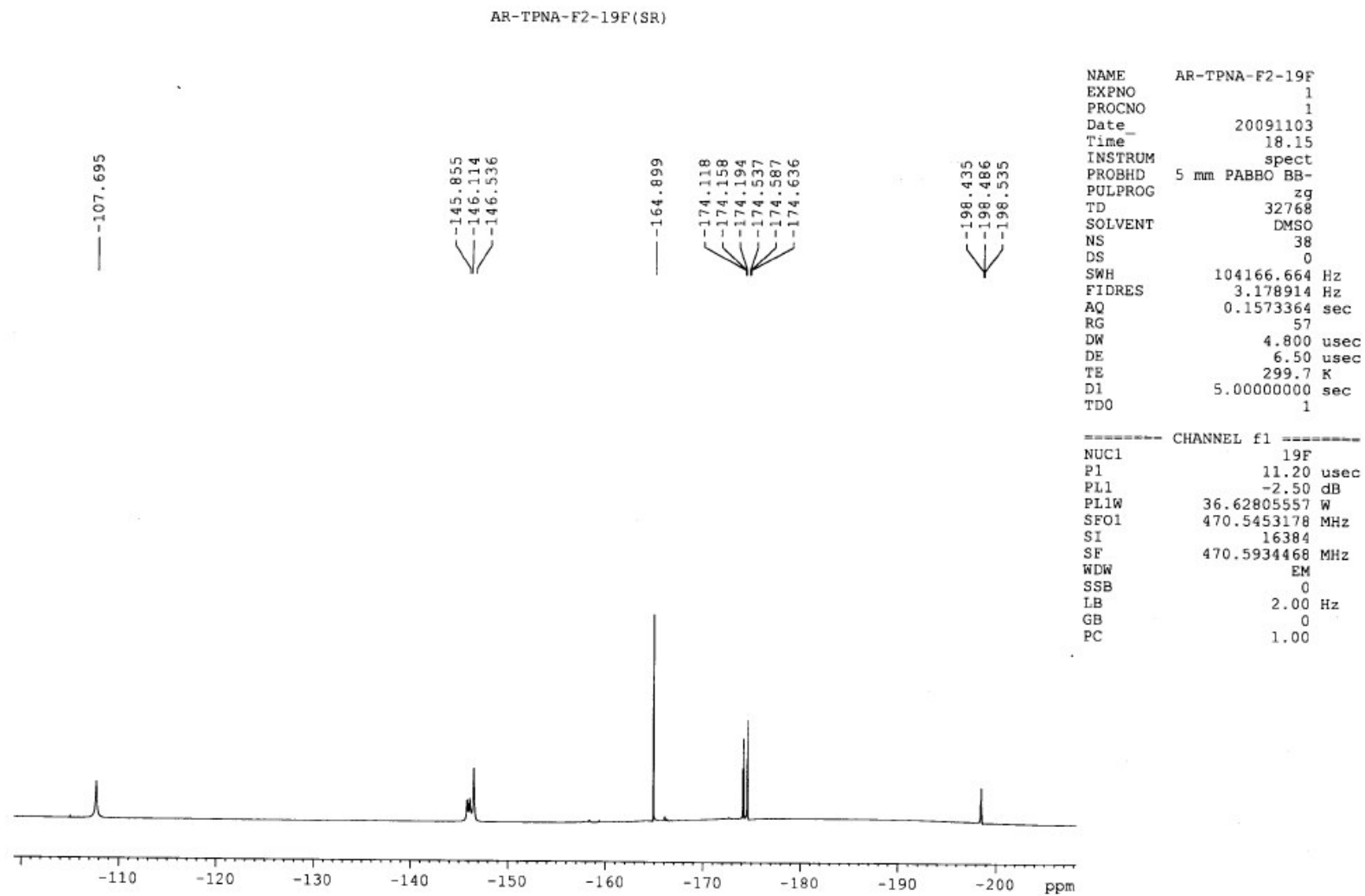


Figure S13. ^{19}F -NMR spectra of mixture of TBA-F and L^1 in $\text{DMSO-}d_6$ at 25°C . Hexafluorobenzene was used as standard.