



Directed diversity-oriented synthesis. Ring-fused 5- to 10-membered rings from a common peptidomimetic 2-pyridone precursor

Magnus Sellstedt, G. Krishna Prasad, K. Syam Krishnan, Fredrik Almqvist*

Department of Chemistry, Umeå University, 901 57 Umeå, Sweden
Umeå Centre for Microbial Research, Umeå University, 901 57 Umeå, Sweden

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ABSTRACT

A variety of ring-fused 2-pyridone-based central fragments were prepared using a strategy inspired by diversity-oriented synthesis. The produced compounds are diverse, yet focused, analogs of biologically active peptidomimetic 2-pyridones.

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2-Pyridones are privileged structures in medicinal chemistry,¹ and many biologically active ring-fused 2-pyridones exist, for example, the natural product derived anti-cancer drug, topotecan,² and the acetylcholinesterase inhibitor, huperzine A.³ The ring-fused 2-pyridones **1** and **2** (Fig. 1) belong to a class of compounds designed as peptidomimetics.⁴ This type of thiazolo ring-fused 2-pyridone has various biological effects depending on their substitution pattern, for example, as antibacterials—targeting bacterial virulence,^{5,6} and as modulators of the fibrillation of amyloid proteins, such as Alzheimer β -peptides⁷ and the Parkinson's associated α -synuclein.⁸ Previous synthetic efforts to improve, or to alter the biological activity of these compounds have focused on variation of the substituents⁹ as well as changes of the central fragment itself.^{10,11} Here, we report the synthesis of a series of central fragment analogs of compounds **1** and **2** using a strategy inspired by the concepts of diversity-oriented synthesis (DOS).^{12,13}

The aim with DOS is to prepare compound collections with widely diverse central fragments to find biologically active substances without having a pre-determined target.¹⁴ Considering the diverse biological activities of the thiazolo ring-fused 2-pyridones, we believe that further exploration of the chemical space around this class will generate many new compounds that are active in a variety of biological assays. In efforts toward this goal, we have developed a strategy involving directed diversity-oriented synthesis, where the produced compounds are relatively close to compounds **1** and **2** in the chemical space, but still considerably

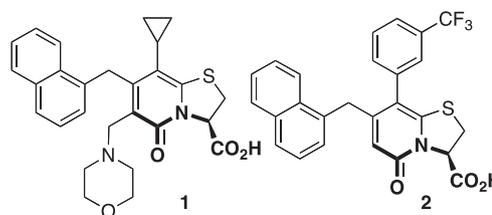


Figure 1. Biologically active thiazolo ring-fused 2-pyridones with peptidomimetic backbones (highlighted).

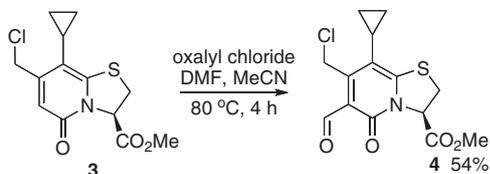
more varied than what is usually achieved by substituent variation. Inspired by the use of a 2-fluorobenzene sulfonyl chloride as a two-sided electrophile in the preparation of a DOS-library,¹⁵ compound **3** was prepared in analogy with previously prepared compounds,^{16,17} and then formylated¹⁸ to give compound **4**, which contains two different electrophilic sites (Scheme 1). This key starting material allowed the synthesis of compounds with large variations of the left-hand side of the peptidomimetic 2-pyridone by orthogonally reacting the two electrophilic groups to form various ring-structures.

To start with, two new heteroaromatic scaffolds were prepared. Heteroaromatics are often easily functionalized and give rigid compounds, which can be beneficial for biological activity.¹⁹ Ring-fused pyrroles have been prepared from other compounds with two electrophilic sites,^{20,21} and reaction of **4** with primary amines under basic conditions gave the pyrroles **5** and **6** (Scheme 2A).

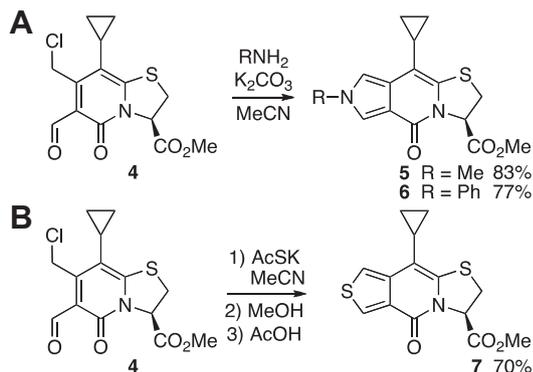
The corresponding ring-fused thiophene **7** was obtained by reaction with potassium thioacetate followed by cleavage of the

* Corresponding author.

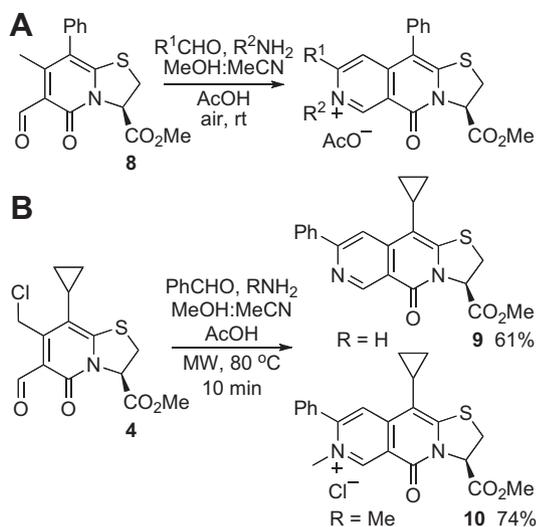
E-mail address: fredrik.almqvist@chem.umu.se (F. Almqvist).



Scheme 1. Preparation of a 2-pyridone with two electrophilic sites.



Scheme 2. Ring-fused five-membered heteroaromatics.



Scheme 3. Improved syntheses of naphthyridones and naphthyridonium salts.

acetyl group and dehydration in a one-pot procedure (Scheme 2B). The use of sodium sulfide instead of potassium thioacetate gave significant amounts of a dimeric compound (according to LC–MS).

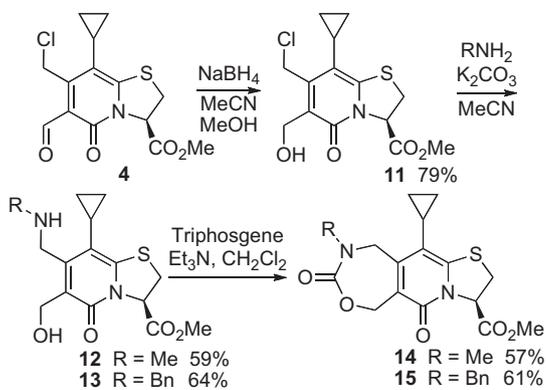
We previously described a three-component reaction that produced dihydronaphthyridones from methyl and formyl substituted 2-pyridones such as **8**. Subsequent oxidation with air, or more generally, with chloranil, gave naphthyridones and naphthyridonium salts (Scheme 3A).²² Compound **4** participated in similar reactions when reacted with aldehydes and amines under mildly acidic conditions (Scheme 3B). The naphthyridone **9** and the naphthyridonium salt **10** were formed without the need of an extra oxidation step, thus providing a more straightforward route to these types of compounds.

Although new heteroaromatic scaffolds are valuable in medicinal chemistry,²³ compounds with a higher proportion of sp^3 -carbons and stereochemical complexity often interact more specifically with proteins.²⁴ A higher proportion of saturated carbons also decreases

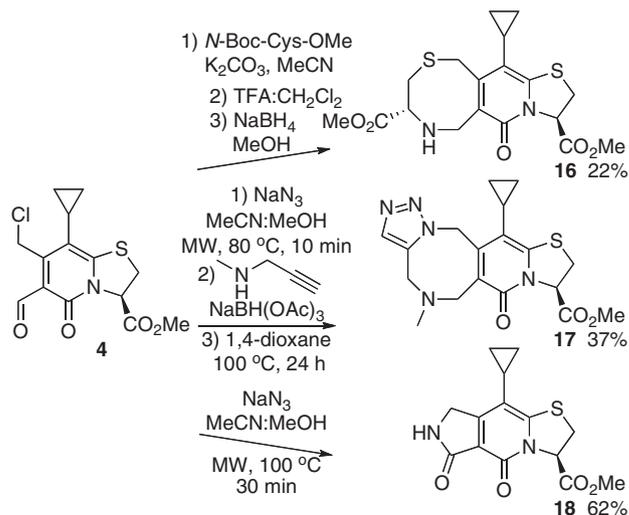
structural planarity, which typically improves aqueous solubility.²⁵ By preparing medium-sized rings with saturated carbons, less planar compounds with restricted flexibility were created. To synthesize the seven-membered carbamates **14** and **15**, compound **4** was reduced to the alcohol **11**, reacted with an amine, and then ring-closed with triphosgene ($\text{Cl}_3\text{COCOCCl}_3$) (Scheme 4).

Two compounds containing eight-membered rings were synthesized by nucleophilic substitution of the chloride followed by reductive amination of the aldehyde (Scheme 5). Thus, Boc-protected cysteine methyl ester was reacted with **4** and then deprotected with trifluoroacetic acid. Ring-closure to compound **16** was accomplished with sodium borohydride in methanol. The use of non-protected cysteine gave several by-products. Triazole **17** was prepared by displacement of the chloride with sodium azide followed by reductive amination with *N*-methyl propargylamine. A thermal intramolecular Huisgen cyclization^{26,27} then gave the ring-closed product **17**. During the first step in the synthesis of **17**, small amounts of the ring-fused pyrrolidone **18** were formed. This type of transformation usually requires acidic conditions and is then regarded an intramolecular Schmidt rearrangement,²⁸ but in our case acid did not promote the formation of **18**. However, it was found that slightly elevated temperatures and longer reaction times increased the amount of this product and allowed its isolation in 62% yield.

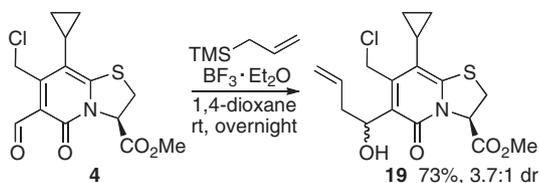
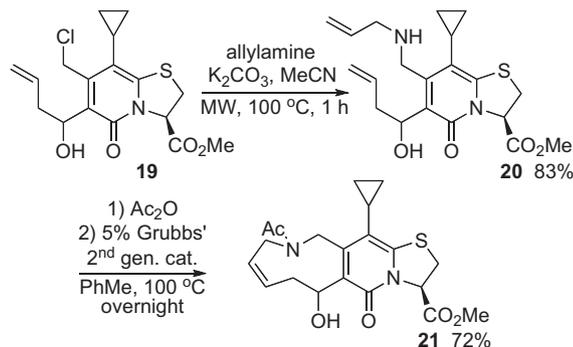
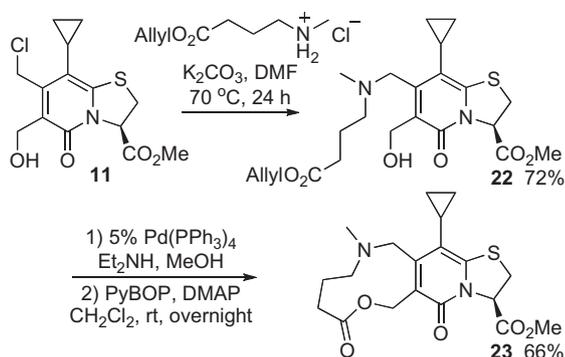
We planned to prepare a nine-membered ring system by ring-closing metathesis.²⁹ To introduce the alkenes for the metathesis,



Scheme 4. Synthesis of seven-membered carbamates.



Scheme 5. Synthesis of eight-membered rings and a pyrrolidone ring-fused compound.

Scheme 6. Hosomi-Sakurai allylation of **4**.Scheme 7. Ring-closing metathesis provided compound **21**.

Scheme 8. Lactonization into a 10-membered ring.

compound **4** was first allylated with allyltrimethylsilane.³⁰ Different conditions for the allylation were evaluated, and the best results were accomplished in 1,4-dioxane with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ as the acid (Scheme 6). After purification, a 3.7:1 diastereomeric mixture of **19** was obtained in 73% yield (2.5:1 dr before purification). A lower diastereomeric ratio (1.8:1) was obtained in dichloromethane, and with SnCl_4 as the Lewis acid the diastereoselectivity inverted to give a 1:2.2 mixture.

Next, compound **19** was reacted with allylamine to introduce the second alkene needed for the metathesis. The resulting amine **20** was then *N*-acetylated, and the ring-closing metathesis was realized by the use of Grubbs' second generation catalyst³¹ in toluene (20 mM) to give **21** in 72% yield (Scheme 7).

For the 10-membered case, we considered a lactone ring-closure. Although medium-sized lactones are often difficult to prepare because of competing formation of dimeric species,³² substrates with rigidifying elements such as *Z*-double bonds can give efficient reactions.³³ The 10-membered lactone **23** was prepared from the alcohol **11**, which was first reacted with the allyl ester of *N*-methyl- γ -aminobutyric acid to give **22**. This product was then subjected to palladium-catalyzed deallylation, followed by lactonization to give **23** using PyBOP as the coupling reagent³⁴ (Scheme 8). The ring-closure proceeded well using modest dilute conditions (50 mM) without the need for slow addition of the substrate.

In conclusion, we have applied a strategy of directed diversity-oriented synthesis to prepare a series of 5- to 10-membered ring-fused structural peptidomimetics starting from a chloromethyl and formylated compound. This type of two-sided electrophile can potentially be used to construct a wide range of ring-fused heterocycles. The compounds produced here are diverse analogs of biologically active 2-pyridones that will be evaluated in a variety of biological assays in the near future.

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Supplementary data

Supplementary data (experimental procedure and spectral data) associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2012.08.100>.

References and notes

- Smith, A. B., III; Atasoylu, O.; Beshore, D. C. *Synlett* **2009**, 2643–2646.
- Kollmannsberger, C.; Mross, K.; Jakob, A.; Kanz, L.; Bokemeyer, C. *Oncology* **1999**, *56*, 1–12.
- Liu, J. S.; Zhu, Y. L.; Yu, C. M.; Zhou, Y. Z.; Han, Y. Y.; Wu, F. W.; Qi, B. F. *Can. J. Chem.* **1986**, *64*, 837–839.
- Svensson, A.; Larsson, A.; Emtenäs, H.; Hedenström, M.; Fex, T.; Hultgren, S. J.; Pinkner, J. S.; Almqvist, F.; Kihlberg, J. *ChemBioChem* **2001**, *2*, 915–918.
- Cegelski, L.; Pinkner, J. S.; Hammer, N. D.; Cusumano, C. K.; Hung, C. S.; Chorell, E.; Åberg, V.; Walker, J. N.; Seed, P. C.; Almqvist, F.; Chapman, M. R.; Hultgren, S. J. *Nat. Chem. Biol.* **2009**, *5*, 913–919.
- Pinkner, J. S.; Remaut, H.; Buelens, F.; Miller, E.; Åberg, V.; Pemberton, N.; Hedenström, M.; Larsson, A.; Seed, P.; Waksman, G.; Hultgren, S. J.; Almqvist, F. *Proc. Natl. Acad. Sci. U.S.A.* **2006**, *103*, 17897–17902.
- Åberg, V.; Norman, F.; Chorell, E.; Westermark, A.; Olofsson, A.; Sauer-Eriksson, A. E.; Almqvist, F. *Org. Biomol. Chem.* **2005**, *3*, 2817–2823.
- Horvath, I.; Weise, C. F.; Andersson, E. K.; Chorell, E.; Sellstedt, M.; Bengtsson, C.; Olofsson, A.; Hultgren, S. J.; Chapman, M. R.; Wolf-Watz, M.; Almqvist, F.; Wittung-Stafshede, P. E. L. *J. Am. Chem. Soc.* **2012**, *134*, 3439–3444.
- Chorell, E.; Pinkner, J. S.; Bengtsson, C.; Banchelin, T. S.-L.; Edvinsson, S.; Linusson, A.; Hultgren, S. J.; Almqvist, F. *Bioorg. Med. Chem.* **2012**, *20*, 3128–3142.
- Sellstedt, M.; Almqvist, F. *Org. Lett.* **2008**, *10*, 4005–4007.
- Sellstedt, M.; Almqvist, F. *Org. Lett.* **2009**, *11*, 5470–5472.
- Schreiber, S. L. *Science* **2000**, *287*, 1964–1969.
- Burke, M. D.; Schreiber, S. L. *Angew. Chem., Int. Ed.* **2004**, *43*, 46–58.
- Morton, D.; Leach, S.; Cordier, C.; Warriner, S.; Nelson, A. *Angew. Chem., Int. Ed.* **2009**, *48*, 104–109.
- Samarakoon, T. B.; Loh, J. K.; Rolfe, A.; Le, L. S.; Yoon, S. Y.; Lushington, G. H.; Hanson, P. R. *Org. Lett.* **2011**, *13*, 5148–5151.
- Chorell, E.; Bengtsson, C.; Sainte-Luce Banchelin, T.; Das, P.; Uvell, H.; Sinha, A. K.; Pinkner, J. S.; Hultgren, S. J.; Almqvist, F. *Eur. J. Med. Chem.* **2011**, *46*, 1103–1116.
- Emtenäs, H.; Alderin, L.; Almqvist, F. *J. Org. Chem.* **2001**, *66*, 6756–6761.
- Pemberton, N.; Pinkner, J. S.; Jones, J. M.; Jakobsson, L.; Hultgren, S. J.; Almqvist, F. *Tetrahedron Lett.* **2007**, *48*, 4543–4546.
- Kubinyi, H. *Perspect. Drug Discovery Des.* **1998**, *9*, 225–252.
- Senda, S.; Hirota, K.; Asao, T.; Yamada, Y. *Synthesis* **1978**, 463–465.
- Stephan, D.; Gorgues, A.; Le, C. A. *Tetrahedron Lett.* **1988**, *29*, 1025–1028.
- Sellstedt, M.; Almqvist, F. *Org. Lett.* **2011**, *13*, 5278–5281.
- Pitt, W. R.; Parry, D. M.; Perry, B. G.; Groom, C. R. *J. Med. Chem.* **2009**, *52*, 2952–2963.
- Clemons, P. A.; Bodycombe, N. E.; Carrinso, H. A.; Wilson, J. A.; Shamji, A. F.; Wagner, B. K.; Koehler, A. N.; Schreiber, S. L. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *107*, 18787–18792.
- Ishikawa, M.; Hashimoto, Y. *J. Med. Chem.* **2011**, *54*, 1539–1554.
- Huisgen, R. *Proc. Chem. Soc.* **1961**, 357–396.
- Hotha, S.; Anegundi, R. I.; Natu, A. A. *Tetrahedron Lett.* **2005**, *46*, 4585–4588.
- Milligan, G. L.; Mossman, C. J.; Aube, J. J. *Am. Chem. Soc.* **1995**, *117*, 10449–10459.
- Fu, G. C.; Grubbs, R. H. *J. Am. Chem. Soc.* **1992**, *114*, 7324–7325.
- Hosomi, A.; Sakurai, H. *Tetrahedron Lett.* **1976**, *17*, 1295–1298.
- Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. *Org. Lett.* **1999**, *1*, 953–956.
- Gérard, R. *Tetrahedron* **1995**, *51*, 2777–2849.
- Guillerm, D.; Linstrumelle, G. *Tetrahedron Lett.* **1985**, *26*, 3811–3812.
- López-Macià, A.; Jiménez, J. C.; Royo, M.; Giralt, E.; Albericio, F. *Tetrahedron Lett.* **2000**, *41*, 9765–9769.