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Facile synthesis and antitumor activity of novel 2-trifluoromethylthieno[2,3-*d*]pyrimidine derivatives

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ABSTRACT

A series of novel 2-trifluoromethylthieno[2,3-*d*]pyrimidine derivatives were synthesized by a facile three-step procedure that afforded advantages of mild reaction conditions, simple protocol and good yields. The structures of the final compounds were confirmed by IR, NMR, EI-MS, elemental analysis, and X-ray diffraction. Preliminary bioassay results showed that some of the analogs exhibit excellent antitumor activity against MCF-7 and HepG2, especially compounds **3a**, **3b**, **3e** and **3h** exhibited higher activity than the positive control gefitinib.

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1. Introduction

A pyrimidine nucleus fused with another heterocycle is widely used in the design and discovery of novel bioactive molecules and drugs [1]. For example, many anticancer agents that act as tyrosine kinase inhibitors typically contain an aminopyrimidine group as a core moiety. Different quinazoline derivatives, such as gefitinib [2], erlotinib [3] and lapatinib [4] (Fig. 1), have gained market approval worldwide. Recently, the thieno[2,3-*d*]pyrimidine core, which is evaluated as a bioisostere of the quinazoline core, was used in the mechanism-based design and synthesis of new antitumor agents [5–8]. Introducing a fluorine-containing substituent, particularly the trifluoromethyl group, led to higher biological activity and lower toxicity compared with their non-fluorinated analogs [9,10]. A number of thieno[2,3-*d*]pyrimidine derivatives with different substituents at the C-2 and C-4 positions were found to exert potential antitumor activity [11–14]. However, trifluoromethyl-substituted thieno[2,3-*d*]pyrimidines at the C-2 position have seldom been reported [10,15]. In this work we prepared new thieno[2,3-*d*]pyrimidines by introducing the trifluoromethyl

group at the C-2 position and different substituents at the C-4 position to explore the potential of thieno[2,3-*d*]pyrimidines as antitumor compounds.

Synthetic protocols for 2,4-disubstituted thieno[2,3-*d*]pyrimidines usually involve a conversion of 2-substituted-thieno[2,3-*d*]pyrimidin-4-ones to 4-chloro-2-substituted-thieno[2,3-*d*]pyrimidines, which commonly undergo multi-step procedures (Scheme 1) [16–20]. These procedures possess disadvantages, such as rigorous conditions, long reaction time, complex handling, and poor total yields. Therefore, developing a facile process to produce 2,4-disubstituted thieno[2,3-*d*]pyrimidines is necessary. We utilized a new convenient and efficient method to synthesize a series of novel *N*⁴-substituted 2-trifluoromethyl-6,7-dihydro-5*H*-clopenta[4,5]thieno[2,3-*d*]pyrimidin-4-amines by reaction of appropriate amines with 4-chloro-2-trifluoromethyl-6,7-dihydro-5*H*-clopenta[4,5]thieno[2,3-*d*]pyrimidine, which started directly from 2-amino-5,6-dihydro-4*H*-cyclopenta[*b*]thiophene-3-carbonitrile and trifluoroacetic acid (TFA) in the presence of phosphorous oxychloride by a one-pot procedure. Our original synthetic strategy is outlined in Scheme 2.

2. Experimental

All the chemicals used in the synthesis were of analytical grade. IR spectra were recorded on a Nicolet NEXUS 470 FT-IR spectrophotometer in the 4000–400 cm^{−1} range. NMR spectra

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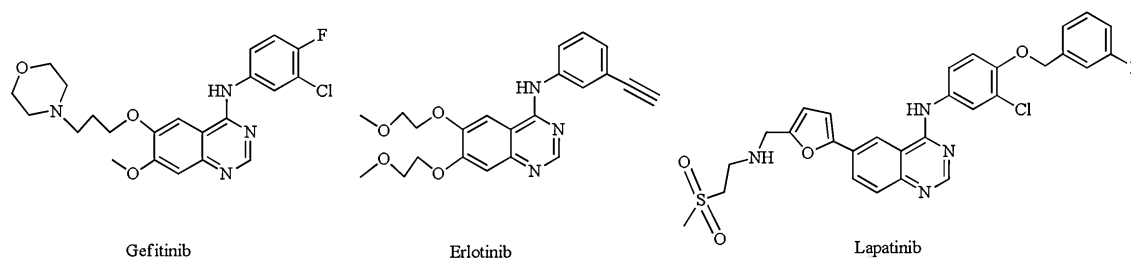
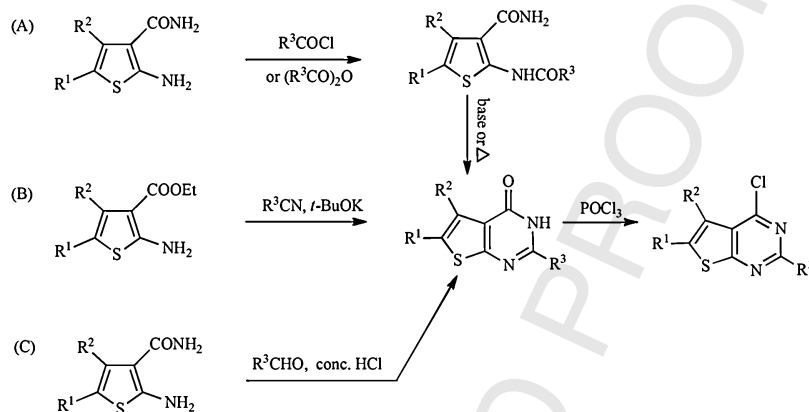


Fig. 1. Representative examples of anticancer quinazoline compounds.



Scheme 1. Preparation of 4-chloro-2-substituted-thieno[2,3-d]pyrimidines based on literature methods.

were obtained on a Varian XL-400 MHz spectrometer with TMS as the internal standard and DMSO- d_6 as the solvent. MS spectra were performed by a Thermo DSQ II mass spectrometer using the electron ionization (EI) method. Elemental analysis was carried out on a Vario EL III CHNSO analyzer, the accepted deviation of experimental values from the calculated ones is 0.3%. X-ray diffraction data were collected on a Bruker Smart APEX-II CCD diffractometer equipped with a graphite-monochromatized Mo K α ($\lambda = 0.71073 \text{ \AA}$) radiation. Melting points were measured with an X-4 digital melting-point apparatus and uncorrected.

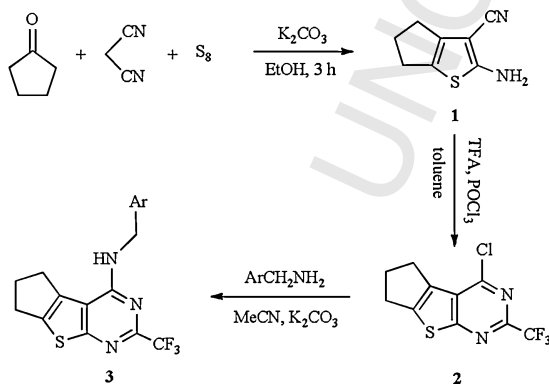
Synthesis of 2-amino-5,6-dihydro-4H-cyclopenta[b]thiophene-3-carbonitrile (**1**): Cyclopentanone (0.84 g, 10 mmol), malononitrile (0.66 g, 10 mmol), elemental sulfur (0.35 g, 11.0 mmol), K₂CO₃ (0.28 g, 2.0 mmol) and 15 mL of dry ethanol were stirred at reflux for 3 h. The insoluble material was filtered off, and the solvent was removed by evaporation under reduced pressure. The crude product was washed with water and recrystallized from ethanol to give yellowish crystals in 81% yield. Mp 152–153 °C (Lit. [21]; 151 °C).

Synthesis of 4-chloro-2-trifluoromethyl-6,7-dihydro-5H-clopenta[4,5]thieno[2,3-d]pyrimidine (**2**): A mixture of compound **1** (0.82 g, 5 mmol), TFA (0.5 mL), toluene (8 mL) and phosphoryl trichloride (1.5 mL) was heated to 80 °C with good stirring. The progress of the reaction was monitored by TLC with petroleum ether–ethyl acetate (3:1, v/v) as a developing solvent. Toluene was removed by vacuum distillation after the completion of the reaction. The residue was poured over crushed ice and neutralized with a saturated sodium bicarbonate solution. The aqueous mixture was extracted with diethyl ether and the organic layer was washed with water followed by saturated aqueous sodium chloride. After evaporation of the solvent, the residue was recrystallized from *n*-hexane to afford the yellowish compound **2** in 65% yield. Mp 142–143 °C. Anal. Calcd. for C₁₀H₆ClF₃N₂S: C 43.10, H 2.17, N 10.05; found: C 42.95, H 2.32, N 9.93.

Synthesis of compounds **3a–3k**: A mixture of compound **2** (1.39 g, 5 mmol), appropriate amine (5 mmol), and K₂CO₃ (1.38 g, 10 mmol) in acetonitrile (10 mL) was heated under reflux for 2–3 h. When the reaction was completed (TLC), acetonitrile was removed by evaporation, the residue was washed with water and purified by normal chromatography to afford the desired products **3a–3k** as white solid.

Compound **3a**: Yield 86%, mp 128–129 °C; ¹H NMR (400 MHz, DMSO- d_6): δ 7.80 (s, 1H, NH), 7.41–7.20 (m, 5H, Ar-H), 4.72 (s, 2H, Ar-CH₂), 3.08–2.88 (m, 4H, 5- and 7-CH₂), 2.43–2.40 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO- d_6): δ –69.02; IR (KBr, cm^{–1}): 3418 (N–H), 1573 (C=N), 1344, 1126 (CF₃); EI-MS (%): *m/z* 349 (M⁺, 75.9), 106 (78.4), 91 (100); Anal. Calcd. for C₁₇H₁₄F₃N₃S: C 58.44, H 4.04, N 12.03; found: C 58.27, H 3.89, N 11.94.

Compound **3b**: Yield 78%, mp 145–146 °C; ¹H NMR (400 MHz, DMSO- d_6): δ 7.71 (s, 1H, NH), 7.33–7.03 (m, 4H, Ar-H), 4.71 (s, 2H, Ar-CH₂), 3.07–2.90 (m, 4H, 5- and 7-CH₂), 2.42–2.39 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO- d_6): δ –69.17, –114.04; IR (KBr, cm^{–1}): 3425 (N–H), 1568 (C=N), 1361, 1137 (CF₃); EI-MS (%): *m/z* 367 (M⁺, 77.6), 124 (100), 109 (65.2); Anal. Calcd. for C₁₇H₁₃F₄N₃S: C 55.58, H 3.57, N 11.44; found: C 55.70, H 3.46, N 11.52.



Scheme 2. The synthetic route of title compounds **3a–3k**.

Compound **3c**: Yield 82%, mp 135–136 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.83 (s, 1H, NH), 7.43–7.09 (m, 4H, Ar-H), 4.67 (s, 2H, Ar-CH₂), 3.09–2.92 (m, 4H, 5- and 7-CH₂), 2.50–2.40 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.00, –116.30; IR (KBr, cm^{–1}): 3469 (N–H), 1590 (C=N), 1339, 1143 (CF₃); EI-MS (%): *m/z* 367 (M⁺, 46.8), 124 (58.8), 109 (100); Anal. Calcd. for C₁₇H₁₃F₃N₃S: C 55.58, H 3.57, N 11.44; found: C 55.43, H 3.72, N 11.28.

Compound **3d**: Yield 79%, mp 153–154 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.74 (s, 1H, NH), 7.43–7.26 (m, 4H, Ar-H), 4.79 (s, 2H, Ar-CH₂), 3.13–2.95 (m, 4H, 5- and 7-CH₂), 2.50–2.45 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.12; IR (KBr, cm^{–1}): 3452 (N–H), 1580 (C=N), 1367, 1121 (CF₃); EI-MS (%): *m/z* 383 (M⁺, 64.6), 140 (61.3), 125 (100); Anal. Calcd. for C₁₇H₁₃ClF₃N₃S: C 53.20, H 3.41, N 10.95; found: C 53.31, H 3.56, N 11.10.

Compound **3e**: Yield 84%, mp 126–127 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.87 (s, 1H, NH), 7.42–7.31 (m, 4H, Ar-H), 4.67 (s, 2H, Ar-CH₂), 3.13–2.89 (m, 4H, 5- and 7-CH₂), 2.50–2.40 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.01; IR (KBr, cm^{–1}): 3462 (N–H), 1568 (C=N), 1333, 1121 (CF₃); EI-MS (%): *m/z* 383 (M⁺, 70.0), 140 (64.4), 125 (100); Anal. Calcd. for C₁₇H₁₃ClF₃N₃S: C 53.20, H 3.41, N 10.95; found: C 53.40, H 3.27, N 10.86.

Compound **3f**: Yield 75%, mp 150–151 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.72 (s, 1H, NH), 7.30–7.11 (m, 4H, Ar-H), 4.68 (s, 2H, Ar-CH₂), 3.13–2.96 (m, 4H, 5- and 7-CH₂), 2.50–2.45 (m, 2H, 6-CH₂), 2.38 (s, 3H, CH₃); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.10; IR (KBr, cm^{–1}): 3468 (N–H), 1585 (C=N), 1339, 1121 (CF₃); EI-MS (%): *m/z* 363 (M⁺, 44.0), 120 (19.9), 105 (100); Anal. Calcd. for C₁₈H₁₆F₃N₃S: C 59.49, H 4.44, N 11.56; found: C 59.64, H 4.29, N 11.46.

Compound **3g**: Yield 80%, mp 124–125 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.67 (s, 1H, NH), 7.15–6.94 (m, 4H, Ar-H), 4.52 (s, 2H, Ar-CH₂), 2.97–2.80 (m, 4H, 5- and 7-CH₂), 2.37–2.27 (m, 2H, 6-CH₂), 2.10 (s, 3H, CH₃); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.06; IR (KBr, cm^{–1}): 3413 (N–H), 1580 (C=N), 1334, 1137 (CF₃); EI-MS (%): *m/z* 363 (M⁺, 40.6), 120 (30.9), 105 (100); Anal. Calcd. for C₁₈H₁₆F₃N₃S: C 59.49, H 4.44, N 11.56; found: C 59.37, H 4.53, N 11.75.

Compound **3h**: Yield 81%, mp 157–158 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.89 (s, 1H, NH), 7.64–7.59 (m, 4H, Ar-H), 4.77 (s, 2H, Ar-CH₂), 3.11–2.94 (m, 4H, 5- and 7-CH₂), 2.49–2.44 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –61.18, –69.09; IR (KBr, cm^{–1}): 3429 (N–H), 1568 (C=N), 1322, 1126 (CF₃); EI-MS (%): *m/z* 417 (M⁺, 73.6), 174 (100), 159 (39.2); Anal. Calcd. for C₁₈H₁₃F₆N₃S: C 51.80, H 3.14, N 10.07; found: C 51.92, H 3.30, N 9.89.

Compound **3i**: Yield 74%, mp 154–155 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.52 (s, 1H, NH), 7.18–6.84 (m, 4H, Ar-H), 4.70 (s, 2H, Ar-CH₂), 3.83 (s, 3H, OCH₃), 3.12–2.96 (m, 4H, 5- and 7-CH₂), 2.49–2.44 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.09; IR (KBr, cm^{–1}): 3468 (N–H), 1585 (C=N), 1361, 1121 (CF₃); EI-MS (%): *m/z* 379 (M⁺, 59.5), 136 (32.1), 121 (100); Anal. Calcd. for C₁₈H₁₆F₃N₃OS: C 56.98, H 4.25, N 11.08; found: C 57.15, H 4.09, N 11.23.

Compound **3j**: Yield 73%, mp 123–124 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.70 (s, 1H, NH), 7.09–6.64 (m, 4H, Ar-H), 4.55 (s, 2H, Ar-CH₂), 3.57 (s, 3H, OCH₃), 2.98–2.81 (m, 4H, 5- and 7-CH₂), 2.37–2.30 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.06; IR (KBr, cm^{–1}): 3432 (N–H), 1579 (C=N), 1339, 1132 (CF₃); EI-MS (%): *m/z* 379 (M⁺, 68.2), 136 (100), 121 (57.5); Anal. Calcd. for C₁₈H₁₆F₃N₃OS: C 56.98, H 4.25, N 11.08; found: C 57.11, H 4.34, N 11.20.

Compound **3k**: Yield 83%, mp 118–119 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.77 (s, 1H, NH), 7.33–6.84 (m, 4H, Ar-H), 4.62 (s, 2H, Ar-CH₂), 3.68 (s, 3H, OCH₃), 3.07–2.91 (m, 4H, 5- and 7-CH₂), 2.49–2.40 (m, 2H, 6-CH₂); ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –69.08; IR (KBr, cm^{–1}): 3474 (N–H), 1596 (C=N), 1367, 1115 (CF₃); EI-MS (%): *m/z* 379 (M⁺, 17.0), 136 (4.9), 121 (100); Anal. Calcd. for C₁₈H₁₆F₃N₃OS: C 56.98, H 4.25, N 11.08; found: C 56.86, H 4.17, N 10.85.

3. Results and discussion

The synthesis was initiated by allowing readily available cyclopentanone to react with malononitrile and sulfur to form thiophene **1** based on the modified Gewald procedure. In this context, we have found that the Gewald reaction efficiently occurs in the presence of potassium carbonate (K₂CO₃) as a heterogeneous base catalyst under reflux in ethanol. To the best of our knowledge, the use of K₂CO₃ in the synthesis of 2-aminothiophenes has not been reported. To show the merits of the present work, we compared results obtained from K₂CO₃ with those previously reported [21–27]. Table 1 reveals that K₂CO₃ is an inexpensive, highly efficient, and green catalyst that can produce thiophene **1** in short time and favorable yield. The key intermediate **2** was efficiently prepared directly from thiophene **1**, TFA, and phosphorous oxychloride using toluene as a solvent via a one-pot procedure, which presents several advantages, such as milder reaction conditions, simpler handling, and better yields, compared with traditional multi-step methods (Routes A–C in Scheme 1). Subsequently, the chloride **2** reacts with appropriate amines to form **3**.

The structures of compounds **3a–3k** were characterized by IR, ¹H NMR, ¹⁹F NMR, EI-MS, and elemental analysis. ¹H NMR spectra show the expected occurrence of signals from the NH (δ 7.90–7.52), aryl protons (δ 7.65–6.60), benzyl CH₂ (δ 4.80–4.50), and three cycloalkyl methylene protons (δ 3.15–2.25). The ¹⁹F signal assigned to the trifluoromethyl (CF₃) group at the C-2 position of the thieno[2,3-*d*]pyrimidine ring appears near δ –69.0. In addition, EI mass spectra gave the anticipated M⁺ peak. The spectroscopic data are in good agreement with the proposed chemical structures of the synthesized compounds.

To further confirm the structures of these compounds and provide a basis for the studies of structure–activity relationships, the crystal structure of compound **3h** was determined by single-crystal X-ray diffraction. A colorless single crystal of compound **3h**

Table 1

Comparison of different methods for synthesizing thiophene **1** via Gewald reaction of cyclopentanone and malononitrile.

Entry	Catalyst	Condition	Time	Yield ^a (%)	Ref.
1	K ₂ CO ₃	Ethanol, reflux	3 h	81	This work
2	Calcined Mg–Al hydrotalcite	Ethanol, 60 °C	12 h	85	[22]
3	L-Proline	DMF, 60 °C	10 h	79	[23]
4	Bovine serum albumin	DMF, 50 °C	4 h	65	[24]
5	KF-alumina	Ethanol, MW ^b	8 min	57	[21]
6	KF-alumina	Ethanol, reflux	5.5 h	55	[21]
7	Nano ZnO	Solvent free, 100 °C	6 h	49	[25]
8	KG-60-piperazine	Ethanol, reflux	4 h	47	[26]
9	Morpholine	Ethanol, reflux	2 h	31	[27]

^a Isolated yield.

^b Microwave heating.

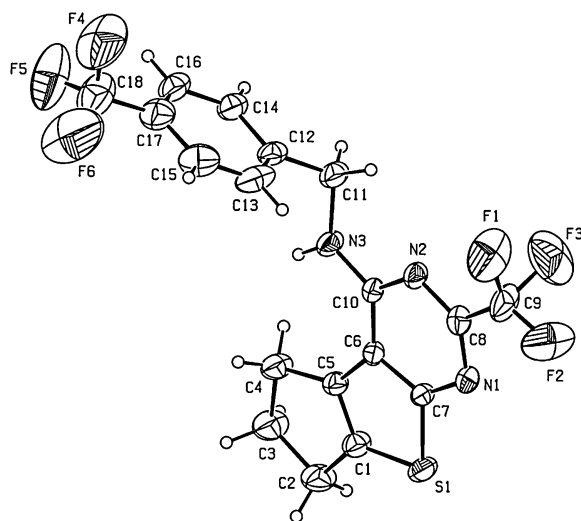


Fig. 2. Molecular structure of compound 3h.

Table 2

The *in vitro* antitumor activity against MCF-7 and HepG2 for title compounds 3a–3k.

Entry	Compd.	Ar	IC ₅₀ ^a (μmol L ^{−1})	
			Breast MCF-7	Liver HepG2
1	3a	C ₆ H ₅	11.79	13.50
2	3b	3-FC ₆ H ₄	5.18	7.31
3	3c	4-FC ₆ H ₄	28.83	20.16
4	3d	2-ClC ₆ H ₄	52.57	25.92
5	3e	4-ClC ₆ H ₄	13.84	8.63
6	3f	2-CH ₃ C ₆ H ₄	83.62	>100
7	3g	4-CH ₃ C ₆ H ₄	>100	>100
8	3h	4-CF ₃ C ₆ H ₄	15.02	16.72
9	3i	2-CH ₃ OC ₆ H ₄	>100	>100
10	3j	3-CH ₃ OC ₆ H ₄	63.89	>100
11	3k	4-CH ₃ OC ₆ H ₄	>100	>100
12	Gefitinib		23.52	18.36

^a IC₅₀: compound concentration required to inhibit tumor cell proliferation by 50%.

The *in vitro* antitumor activity of the newly synthesized compounds 3a–3k against MCF-7 (human breast cancer) and HepG2 (human hepatocellular liver carcinoma) cell lines was evaluated by the standard MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide] assay [29] using gefitinib as a positive control. As described in Table 2, the results of preliminary bioassay reveal that compounds 3a, 3b, 3c, 3d, 3e and 3h exhibit good antitumor activity against MCF-7 and HepG2. Moreover, 3a, 3b, 3e and 3h possessed higher antitumor activity than the positive control gefitinib. The results imply that different substituents at different positions of the benzene ring significantly affect the antitumor activity of the resultant compounds. Incorporation of electron-donating groups, such as methyl (as in 3f and 3g) and methoxy (as in 3i–3k) groups, in the benzene ring led to a decrease of the antitumor activity against both cell lines. Further studies will focus on structural optimization and structure–activity relationships of this class of compounds.

4. Conclusion

A series of novel 2-trifluoromethylthieno[2,3-d]pyrimidine derivatives were synthesized by a facile three-step procedure. The procedure exhibits several advantages, such as mild reaction conditions, simple protocol, and good yields. Their structures were characterized by IR, ¹H NMR, ¹⁹F NMR, EI-MS and elemental analysis, and the structure of 3h was further elucidated by single-crystal X-ray diffraction. The preliminary bioassay results imply that some of the compounds exhibit excellent antitumor activity against MCF-7 and HepG2 cells. These compounds will be further studied in future research.

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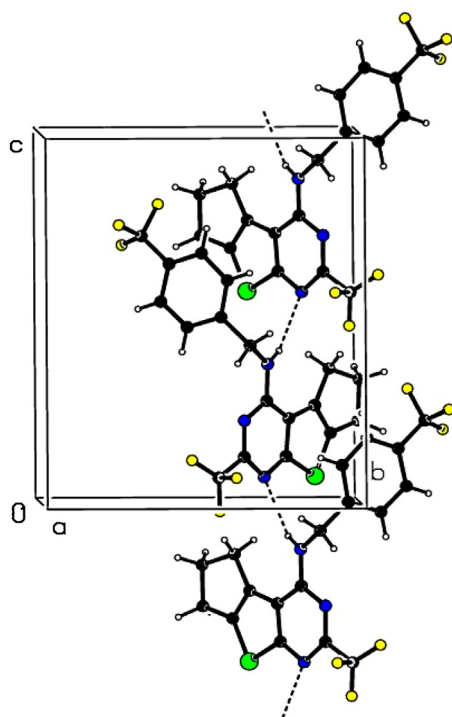


Fig. 3. Packing diagram of compound 3h.

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