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## Synthesis of Pyrrolidine Derivatives by the Palladium-Catalyzed Cyclization of N-(2-Cyclohexenyl)bromoacetamide Derivatives

Shyh-Chyun Yang\* (楊世群), Fu-Hsiung Tseng (曾富雄) and Chung-Wei Hung (黃宗煒) Graduate Institute of Pharmaceutical Sciences, Kaohsiung Medical College, Kaohsiung 80708, Taiwan, R.O.C.

*N*-(2-Cyclohexenyl)bromoacetamide derivatives underwent intramolecular cyclization to give pyrrolidine derivatives in the presence of a base and a catalytic amount of palladium catalyst.

#### INTRODUCTION

Pyrrolidines bearing bicyclic skeletons are components of various alkaloids<sup>1</sup> or amino acid derivatives with potent physiological activities.<sup>2</sup> The efficient construction of the pyrrolidine ring is important to their syntheses. Most approaches to pyrrolidine involve acyl-nitrogen bond formation,<sup>3</sup> but the apparently equally useful mode of carboncarbon bond formation has received little attention. Recently, it has become very popular to use organometallic reagents in the synthesis of heterocyclic ring systems;<sup>4</sup> the extensive studies of palladium catalysis in particular have uncovered many new processes for organic synthesis.<sup>5</sup> In the alkylation of olefins via  $\sigma$ -alkylpalladium(II) complexes, reactions have been limited to methyl, allyl, and benzyl halides.6 The report that  $\alpha$ -halocarbonylated compounds afford fairly stable σ-alkylmetal complexes7 with low-valent metal complexes8 prompted us to extend this reaction by intramolecular cyclization of N-(2-cyclohexenyl)bromoacetamide derivatives via alkylpalladium intermediate.

#### **RESULTS AND DISCUSSION**

The required N-(2-cyclohexenyl)bromoacetamide derivatives 5 were prepared by bromoacetylation of allylic amines. Thus, 3-bromocyclohexene (1) was reacted with amines 2 in the presence of  $K_2CO_3$  to give 2-cyclohexenylamines 3, which were treated with bromoacetyl bromides 4 to produce N-(2-cyclohexenyl)bromoacetamide derivatives 5 (Scheme I).

When N-benzyl-N-(2-cyclohexenyl)bromoacetamide (5a) (1 mmol) was allowed to react with 10 mol% 1:2 Pd(OAc)<sub>2</sub>-PPh<sub>3</sub> in refluxing MeCN in the presence of NEt<sub>3</sub> for 2 h, the direct introduction of the olefin group into the bromoacetamide system led to cyclization to afford the product of 6a in 21% yield (entry I in Table 1). In order to increase the yield of the cyclized product and to investigate

#### Scheme I



the reaction mechanism, this reaction was examined under various reaction conditions. As indicated in Table 1, the bidentate ligand 1,4-bis(diphenylphosphino)butane (dppb) increased the yield of the cyclized product significantly (entries 9 and 10). One possible explanation for this observation is that suitable phosphine ligands help to maximize the concentration of the active catalyst by preventing the loss of palladium metal. It is well-known that, in catalytic processes using a palladium compound as catalyst precursor, loss of the palladium catalyst due to deposition of palladium metal is a distinct possibility.9 To regenerate the palladium(0) catalyst, an amine base such as NEt<sub>3</sub>, K<sub>2</sub>CO<sub>3</sub> or proton sponge [1,8-bis(dimethylamino)naphthalene] was required. Three useful solvents are MeCN, HMPA or DMF; MeCN gave the best results (entries 1-3). Among palladium catalysts including Pd(OCOCF<sub>3</sub>)<sub>2</sub> (entrics 7 and 10), Pd(PPh<sub>3</sub>)<sub>4</sub> (entry 8), PdCl<sub>2</sub> (entry 11), and PdCl<sub>2</sub>-PPh<sub>3</sub> (entry 6),  $Pd(OAc)_2$  was found to be superior (entry 9).

Table 2 summarizes the results for the reaction of 5b-c with a palladium catalyst. Treatment of 5b with  $Pd(OAc)_2$ -dppb gave the cyclized products 6b and 7a in the yield of 40% and 8%, respectively (entry 1 in Table 2). An examination of the <sup>1</sup>H NMR spectrum of 6b thus obtained indicated the presence of a mixture of two diastereomers in a ratio of

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Table 1. Pd-catalyzed Cyclization of N-Benzyl-N-(2-cyclo-

Entry	Catalyst	(mol eq.)	Base	Solvent <sup>a</sup>	Yield of 6a <sup>b</sup>
1	Pd(OAc) <sub>2</sub>	(0.10)	NEt3	MeCN	21%
	-PPh <sub>3</sub>	(0.20)			
2	Pd(OAc) <sub>2</sub>	(0.10)	NEt <sub>3</sub>	HMPA	17
	-PPh3	(0.20)			
3	Pd(OAc) <sub>2</sub>	(0.10)	NEt <sub>3</sub>	DMF	12
	-PPh3	(0.20)			
4	Pd(OAc) <sub>2</sub>	(0.10)	Proton	MeCN	14
	-PPh3	(0.20)	sponge		
5	Pd(OAc) <sub>2</sub>	(0.10)	K <sub>2</sub> CO <sub>3</sub>	MeCN	10
	-PPh3	(0.20)			
6	PdCl <sub>2</sub>	(0.10)	NEt <sub>3</sub>	MeCN	13
	-PPh3	(0.20)			
7	Pd(OCOCF3)2	(0.10)	NEt <sub>3</sub>	MeCN	13
	-PPh3	(0.20)			
8	Pd(PPh3)4	(0.10)	NEt <sub>3</sub>	MeCN	10
9	Pd(OAc) <sub>2</sub>	(0.10)	NEt <sub>3</sub>	MeCN	55
	-dppb	(0.20)			
10	Pd(OCOCF <sub>3</sub> ) <sub>2</sub>	(0.10)	NEt <sub>3</sub>	MeCN	39
	-dppb	(0.20)			
11	PdCl <sub>2</sub>	(0.10)	NEt <sub>3</sub>	MeCN	17

<sup>a</sup> in MeCN was refluxed for 2 h; in HMPA or DMF was stirred at  $65 \degree$ C for 6 h. <sup>b</sup> Isolated yield.

1:1. The structure of 7a was determined by its 2D-NMR spectra (C, H-COSY) and NOE experiments. This palladium-catalyzed cyclization presumably proceeds by oxidative addition of the bromoacetyl group to palladium(0) species, followed by insertion of the double bond into the  $\sigma$ -alkylpalladium bond and the loss of a palladium hydride species by  $\beta$ -elimination to afford compound 6b. The palladium hydride species is attacked by the base to regenerate the palladium(0) catalyst. Compound 7a was considered to be a reductive elimination product from 9 and no  $\beta$ -elimination of the intermediate 8 to compound 10 was observed. Steric repulsion between a methyl group and a large palladium bromide group may selectively afford 7a and contribute to the formation of the dehydrohalogenation product 6b (Scheme II). By changing the catalyst to Pd(OCOCF<sub>3</sub>)<sub>2</sub>dppb, the reaction afforded only compound 6b (entry 2). In the presence of PdCl<sub>2</sub>-dppb, the products 6b and 7a were obtained in the yield of 16% and 5%, respectively (entry 3). Compound 5c was treated with Pd(OAc)<sub>2</sub>-dpph to give the sole cyclized product 6c in the yield of 43% (entry 4). Treatment of 5d with Pd(OAc)2-dppb gave the cyclized Table 2. Pd-catalyzed Cyclization of 5b-d



<sup>a</sup> in the presence of NEt<sub>3</sub> and MeCN, and was refluxed for 2 h. <sup>b</sup> Isolated yield.

#### Scheme II



products 6d as a mixture of two diastereomers in a ratio of 1:1 and 7b in fairly good yields (entry 5).

These results suggest that  $\sigma$ -alkylpalladium complexes, formed from  $\alpha$ -bromoacetamide derivatives, may be useful intermediates in organic synthesis.

#### EXPERIMENTAL SECTION

IR absorption spectra were recorded on a Shimadzu IR-27G spectrophotometer. Proton NMR was measured with a Varian Gemini-200 spectrometer. Chemical shifts ( $\delta$ )

and coupling constants (Hz) were measured with respect to TMS. MS and high resolution mass spectra (HRMS) were taken on a Hewlett Packard 5989A or JEOL JMS D-100 instrument, with a direct inlet system.

#### *N*-Benzyl-*N*-(2-cyclohexenyl)amine (3a)

To a solution of benzylamine (2a) (3.19 g, 30 mmol) and K<sub>2</sub>CO<sub>3</sub> (10.28 g, 74.3 mmol) in MeCN (20 mL) was added a solution of 3-bromocyclohexene (1) (4 g, 24.8 mmol) in MeCN (10 mL) under ice-cooling. The reaction mixture was then stirred at room temperature overnight. Solvent was removed under reduced pressure and column chromatography (4:1 *n*-hexane/ethyl acetate) of the residue afforded 3a (3.85 g, 83%). IR (CHCl<sub>3</sub>) v 3325 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.33 (bs, 1H, NH), 1.47-2.01 (m, 6H), 3.13-3.26 (m, 1H, NCH), 3.82 (bs, 2H, NCH<sub>2</sub>), 5.70-5.78 (m, 2H, vinyl), 7.21-7.36 (m, 5H, ArH); EI-MS *m/z* 187 (M<sup>+</sup>). HRMS Calcd for C<sub>13</sub>H<sub>17</sub>N 187.1361. Found 187.1360.

#### N-Phenyl-N-(2-cyclohexenyl)amine (3b)

As described for 3a, reaction of aniline (2b) (2.77 g, 30 mmol) with 3-bromocyclohexene (1) (4 g, 24.8 mmol) in the presence of K<sub>2</sub>CO<sub>3</sub> (10.28 g, 74.3 mmol) afforded 3b (3.43 g, 80%). IR (CHCl<sub>3</sub>) v 3340 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.57-2.06 (m, 6H), 3.49 (bs, 1H, NH), 3.98 (bs, 1H, NCH), 5.70-5.90 (m, 2H, vinyl), 6.58-6.72 (m, 3H, ArH), 7.11-7.23 (m, 2H, ArH); EI-MS *m*/z 173 (M<sup>+</sup>). HRMS Calcd for C<sub>12</sub>H<sub>15</sub>N 173.1204. Found 173.1205.

## N-Benzyl-N-(2-cyclohexenyl)bromoacetamide (5a)

A solution of *N*-benzyl-*N*-(2-cyclohexenyl)amine (**3a**) (2 g, 10.7 mmol) and K<sub>2</sub>CO<sub>3</sub> (2.95 g, 21.4 mmol) in acetone (20 mL) was added a solution of bromoacetyl bromide (**4a**) (2.59 g, 12.8 mmol) in ether (10 mL) under ice-cooling. The reaction mixture was then stirred at room temperature for 5 h. Solvent was removed under reduced pressure and column chromatography (4:1 *n*-hexane/ethyl acetate) of the residue afforded **5a** (1.81 g, 55%). IR (CHCl<sub>3</sub>) v 1640 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.25-2.23 (m, 6H), 3.65, 3.99 (s and s, 2H, COCH<sub>2</sub>), 4.30-4.72 (m, 3H, NCH<sub>2</sub>Ph and NCH), 5.43-5.53 (m, 1H, vinyl), 5.85-5.96 (m, 1H, vinyl), 7.18-7.40 (m, 5H, ArH); EI-MS *m*/z 309 (M<sup>+</sup>+2), 307 (M<sup>+</sup>), 228 (M<sup>+</sup>-Br). HRMS Calcd for C<sub>15</sub>H<sub>18</sub><sup>79</sup>BrNO 307.0572. Found 307.0571.

#### N-Benzyl-N-(2-cyclohexenyi)-2-bromopropionamide (5b)

As described for 5a, reaction of N-benzyl-N-(2-cyclohexenyl)amine (3a) (2 g, 10.7 mmol) with 2-bromopropionyl bromide (4b) (2.77 g, 12.8 mmol) in the presence of  $K_2CO_3$  (2.95 g, 21.4 mmol) afforded 5b (1.55 g, 45%). IR (CHCl<sub>3</sub>) v 1645 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22-2.05 (m, 6H), 1.69, 1.71 (d and d, J = 6.5 Hz, 3H, CH<sub>3</sub>), 4.11-4.25 (m, 1H, CHBr), 4.40-4.88 (m, 3H, NCH<sub>2</sub>Ph and NCH), 5.43-5.94 (m, 2H, vinyl), 7.14-7.41 (m, 5H, ArH); EI-MS *m/z* 323 (M<sup>+</sup>+2), 321 (M<sup>+</sup>), 242 (M<sup>+</sup>-Br). HRMS Calcd for C<sub>16</sub>H<sub>20</sub><sup>79</sup>BrNO 321.0728. Found 321.0728.

#### N-(2-Cyclohexenyl)-N-phenylbromoacetamide (5c)

As described for **5a**, reaction of *N*-phenyl-*N*-(2-cyclohexenyl)amine (**3b**) (2 g, 11.5 mmol) with bromoacetyl bromide (**4a**) (2.8 g, 13.9 mmol) in the presence of  $K_2CO_3$  (3.19 g, 23.1 mmol) afforded **5c** (2.04 g, 60%). IR (CHCl<sub>3</sub>) v 1640 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.26-1.96 (m, 6H), 3.57 (s, 2H, CH<sub>2</sub>Br), 5.35-5.43 (m, 1H, NCH), 5.59-5.64 (m, 1H, vinyl), 5.74-5.80 (m, 1H, vinyl), 7.19-7.45 (m, 5H, ArH); EI-MS *m*/z 295 (M<sup>+</sup>+2), 293 (M<sup>+</sup>), 214 (M<sup>+</sup>-Br). HRMS Calcd for C<sub>14</sub>H<sub>16</sub><sup>79</sup>BrNO 293.0415. Found 293.0416.

#### N-(2-Cyclohexenyl)-N-phenyl-2-bromopropionamide (5d)

As described for **5a**, reaction of *N*-phenyl-*N*-(2-cyclohexenyl)amine (**3b**) (2 g, 11.5 mmol) with 2-bromopropionyl bromide (**4b**) (2.99 g, 13.9 mmol) in the presence of K<sub>2</sub>CO<sub>3</sub> (3.19 g, 23.1 mmol) afforded **5d** (1.92 g, 54%). IR (CHCl<sub>3</sub>) v 1630 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.26-1.91 (m, 6H), 1.72 (d, *J* = 6.8 Hz, 3H, CH<sub>3</sub>), 4.10 (q, *J* = 6.8 Hz, 1H, CHCH<sub>3</sub>), 5.36-5.41 (m, 1H, NCH), 5.65-5.69 (m, 1H, vinyl), 5.74-5.79 (m, 1H, vinyl), 7.07-7.42 (m, 5H, ArH); EI-MS *m*/z 309 (M\*+2), 307 (M\*), 228 (M\*-Br). HRMS Calcd for C<sub>13</sub>H<sub>18</sub><sup>79</sup>BrNO 307.0572. Found 307.0573.

## Cyclization of *N*-benzyl-*N*-(2-cyclohexenyl)bromoacetamide (5a) with Pd(OAc)<sub>2</sub>

A mixture of **5a** (308 mg, 1 mmol), Pd(OAc)<sub>2</sub> (22 mg, 0.1 mmol), dppb (85 mg, 0.2 mmol), and NEt<sub>3</sub> (202 mg, 2 mmol) in MeCN (10 mL) was refluxed for 2 h. Solvent was removed under reduced pressure and column chromatography (2:1 *n*-hexane/ethyl acetate) of the residue afforded *N*-benzyl-4-indolin-2-one (**6a**) (125 mg, 55%). IR (CHCl<sub>3</sub>) v 1660 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.50-2.29 (m, 4H), 2.57-2.82 (m, 3H), 3.52-3.62 (m, 1H, NCH), 3.98 (d, *J* = 15.0 Hz, 1H, NCHPh), 5.02 (d, *J* = 15.0 Hz, 1H, NCHPh), 5.28-5.95 (m, 2H, vinyl), 7.24-7.32 (m, 5H, ArH), EI-MS *m*/z 227 (M<sup>\*</sup>). HRMS Calcd for C<sub>15</sub>H<sub>17</sub>NO 227.1310. Found 227.1310.

## Cyclization of N-benzyl-N-(2-cyclohexenyl)-2-bromopropionamide (5b) with Pd(OAc)<sub>2</sub>

As described for 5a, reaction of 5b (322 mg, 1 mmol), Pd(OAc)<sub>2</sub> (22 mg, 0.1 mmol), dppb (85 mg, 0.2 mmol), and NEt<sub>3</sub> (202 mg, 2 mmol) in MeCN (10 mL) afforded *N*-benzyl-3-methyl-4-indolin-2-one (6b) (96 mg, 40%) as an isomeric mixture (1:1, determined by <sup>1</sup>H NMR spectroscopy) and N-benzyl-4-bromo-3-methyl-2-oxo-perhydroindole (7a) (26 mg, 8%). 6b: IR (CHCl<sub>3</sub>) v 1660 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(CDCI_3) \delta 1.19$  and 1.28 (d, both J = 6.8 Hz, total 3H, CH<sub>3</sub>), 1.35-2.10 (m, 4H), 2.27-2.70 (m, 2H, CHCHCH<sub>3</sub>), 3.36-3.65 (m, 1H, NCH), 3.96 and 4.05 (d, both J = 15.2 Hz, total 1H, NC<u>H</u>Ph), 4.96 and 4.98 (d, both J = 15.0 Hz, total 1H, NCHPh), 5.71-5.76 (m, 2H, vinyl), 7.18-7.37 (m, 5H, ArH); EI-MS m/z 241 (M<sup>+</sup>). HRMS Calcd for C<sub>16</sub>H<sub>19</sub>NO 241.1467. Found 241.1466. 7a: IR (CHCl<sub>3</sub>) v 1660 cm<sup>-4</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.25 (d, J = 7.1 Hz, 3H, CH<sub>3</sub>), 1.33-2.04 (m, 6H), 2.24-2.33 (m, 1H, CHCHCH<sub>3</sub>), 2.49 (quin, J = 7.2Hz, 1H, CHCH<sub>3</sub>), 3.56-3.65 (m, 1H, NCH), 4.01 (d, J = 15.0Hz, 1H, NCHPh), 4.25-4.33 (m, 1H, CHBr), 4.90 (d, J =15.0 Hz, 1H, NCHPh), 7.23-7.34 (m, 5H, ArH); EI-MS m/z 323 (M\*+2), 321 (M\*), 242 (M\*-Br). HRMS Calcd for C<sub>16</sub>H<sub>20</sub><sup>79</sup>BrNO 321.0728. Found 321.0727.

## Cyclization of N-(2-cyclohexenyl)-N-phenylbromoacetamide (5c) with Pd(OAc)<sub>2</sub>

As described for **5a**, reaction of **5c** (294 mg, 1 mmol), Pd(OAc)<sub>2</sub> (22 mg, 0.1 mmol), dppb (85 mg, 0.2 mmol), and NEt<sub>3</sub> (202 mg, 2 mmol) in MeCN (10 mL) afforded *N*phenyl-4-indolin-2-one (**6c**) (92 mg, 43%). IR (CHCl<sub>3</sub>) v 1680 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.72-2.15 (m, 4H), 2.67-2.75 (m, 2H, COCH<sub>2</sub>), 2.96-3.08 (m, 1H, C<u>H</u>CH=CH), 4.33-4.42 (m, 1H, NCH), 5.59-5.92 (m, 2H, vinyl), 7.14-7.41 (m, 5H, ArH); EI-MS *m*/z 213 (M<sup>+</sup>). HRMS Calcd for C<sub>14</sub>H<sub>15</sub>NO 213.1154. Found 213.1153.

## Cyclization of N-(2-cyclohexenyl)-N-phenyl-2-bromopropionamide (5d) with Pd(OAc)<sub>2</sub>

As described for 5a, reaction of 5d (308 mg, 1 mmol), Pd(OAc)<sub>2</sub> (22 mg, 0.1 mmol), dppb (85 mg, 0.2 mmol), and NEt<sub>3</sub> (202 mg, 2 mmol) in MeCN (10 mL) afforded Nphenyl-3-methyl-4-indolin-2-one (6d) (30 mg, 13%) as an isomeric mixture (1:1, determined by <sup>1</sup>H NMR spectroscopy) and N-phenyl-4-bromo-3-methyl-2-oxo-perhydroindole (7b) (199 mg, 64%). 6d: mp 124-126 °C. IR (CHCl<sub>3</sub>) v 1660 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.23 and 1.32 (d, both J =6.8 Hz, total 3H, CH<sub>3</sub>), 1.53-2.11 (m, 4H), 2.45-2.60 (m, 2H, CHCHCH<sub>3</sub>), 4.01-4.29 (m, 1H, NCH), 5.74-5.87 (m, 2H, vinyl), 7.26-7.58 (m, 5H, ArH); EI-MS m/z 227 (M\*). HRMS Calcd for C<sub>15</sub>H<sub>17</sub>NO 227.1310. Found 227.1309. 7b: IR (CHCl<sub>3</sub>) v 1640 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.28 (d, J = 6.6 Hz, 3H, CH<sub>3</sub>), 1.49-1.97 (m, 6H), 2.51-2.59 (m, 2H, CHCHCH<sub>3</sub>), 4.31-4.41 (m, 1H, NCH), 4.46-4.52 (m, 1H, CHBr), 7.16-7.50 (m, 5H, ArH); EI-MS m/z 309 (M<sup>+</sup>+2),

307 (M<sup>+</sup>), 228 (M<sup>+</sup>-Br). HRMS Calcd for  $C_{15}H_{18}^{79}BrNO$  307.0572. Found 307.0572.

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#### **Key Words**

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#### REFERENCES

- ApSimon, J. The Total Syntheses of Natural Products; Wiley: New York, 1977, Vol. 3.
- 2. (a) Teetz, V.; Geiger, R.; Gaul, H. Tetrahedron Lett.
  1984, 25, 4479. (b) Teetz, V.; Gaul, H. Tetrahedron Lett. 1984, 25, 4483 and references cited therein.
- Frank, R. L.; Schmitz, W. R.; Zeidman, B. Org. Synth. Coll. 1955, 3, 328.
- 4. (a) Davies, S. G. Organotransition Metal Chemistry: Applications to Organic Synthesis; Pergamon: Oxford, 1982. (b) Abel, E. W.; Stone, F. G. A.; Wilkinson, G. Comprehensive Organometallic Chemistry II; Pergamon: Oxford, 1995.
- (a) Tsuji, J. Organic Synthesis with Palladium Compounds; Springer-Verlag: Heidelberg, 1980. (b) Heck, R. F. Palladium Reagents in Organic Syntheses; Academic Press: London, 1985. (c) Tsuji, J. Palladium Reagents and Catalysts; John-Wiley and Sons Ltd.: Chichester, 1995.
- (a) Fitton, P.; Johnson, M. P.; McKepn, J. E. J. Chem. Soc., Chem. Commun. 1968, 6. (b) Pearson, G.; Figdore, P. E. J. Am. Chem. Soc. 1980, 102, 1541.
- Cambillau, C.; Mirize, M. C. J. Chem. Soc., Chem. Commun. 1982, 211.
- Yoshisato, E.; Tsutsumi, S. J. Chem. Soc., Chem. Commun. 1968, 33.
- (a) Hosokawa, T.; Murahashi, S.-I. Acc. Chem. Res. 1990, 23, 49. (b) James, D. E.; Stille, J. K. J. Am. Chem. Soc. 1976, 98, 1810.