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An asymmetric C8/C8'-tripyrrole-linked sequence-selective pyrrolo[2,1-c][1,4]benzodiazepine (PBD) dimer DNA interstrand cross-linking agent spanning 11 DNA base pairs

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Abstract—A novel sequence-selective extended PBD dimer 4 has been synthesized that binds with high affinity to an interstrand cross-linking site spanning 11 DNA base pairs. Despite its molecular weight (984.07) and length, the molecule has significant DNA interstrand cross-linking potency (\sim 100-fold greater than the clinically used agent melphalan) and sub-micromolar cytotoxicity in a number of tumour cell lines, suggesting that it readily penetrates cellular and nuclear membranes to reach its DNA target. © 2008 Elsevier Ltd. All rights reserved.

There is growing interest in small molecules that recognize DNA sequence,¹ and examples are known that interact with DNA non-covalently (e.g., the hairpin polyamides²) or covalently through monoalkylation (e.g., the pyrrolobenzodiazepines³) or cross-linking (e.g., bizelesin⁴) mechanisms. SJG-136 (1, Fig. 1) is a potent, sequence-selective interstrand DNA cross-linking agent containing pyrrolobenzodiazepine units that is presently in Phase I evaluation in the clinic.^{5–7} The propensity of SJG-136 to preferentially target 5'-Pu-GATC-Py DNA sequences is thought to contribute to its antitumour activity.8 In 2001, Bando and co-workers reported the structure of an efficient CPI-based crosslinking agent (2), the polypyrrole-imidazole core of which allows sequence-specific cross-linking.⁹ More recently, syntheses of the first examples of heterocyclic-linked PBD dimers have been independently reported; Kumar and Lown have described molecules of type 3, although DNA cross-linking and sequence-selectivity data were

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not reported,¹⁰ and similar extended PBD dimers have been reported in the patent literature by our group.¹¹

In order to systematically study the effect of length and composition of the polyheterocyclic linker on DNA sequence-selectivity, binding affinity and cross-linking efficiency of C8/C8'-linked PBD dimers of type **3**, we have synthesized different families of dimers containing a variety of heterocyclic linkers and with different regio-chemical arrangements of constituent components. These heterocyclic units are known from the work of Dervan and co-workers to recognize sequences in the minor groove.²

We report here the synthesis and evaluation of the extended asymmetrically-linked tripyrrole-containing PBD dimer 4 (AT-235) which spans 11 DNA base pairs with a sequence-selectivity that can be rationalized based on both its covalent and non-covalent interactions with DNA base pairs in the minor groove.

The synthesis of **4** was achieved through sequential amide coupling between the three main components **5**, **6** (to give **7**) and **8** as shown in Scheme 1.

Keywords: DNA-binding; Sequence-selective; Interstrand; Cross-linking; Anticancer agent; Pyrrolobenzodiazepine; PBD; PBD dimer.



Figure 1. Structures of the DNA interstrand cross-linking agents 1 (SJG-136), the CPI-based dimer 2, the previously reported PBD dimer 3 and the new asymmetric PBD dimer 4 (AT-235).



Scheme 1. Reagents and condition: (a) EDCI, HOBt, DMF, 71%; (b) 4 N HCl in dioxane, quant; (c) oxalyl chloride, THF, DMF cat, then 7 and DIPEA, 44%; (d) Pd(PPh_3)₄, pyrrolidine, CHCl₃, 85%.

The known Boc-tripyrrole acid **5** was prepared by an improvement of a method of Boger and co-workers¹² that required no complex work-up or purification procedures (Scheme 2).

The novel chirally pure C8-aminopropyl-N10-alloc-protected PBD 6 was synthesized in 7 steps from the previreported 4-hydroxy-5-methoxy-2-nitroously benzaldehyde starting material $(14)^{13}$ (Scheme 3). The aliphatic Boc amino side chain was introduced by Mitsunobu phenol etherification and the aldehyde oxidized with potassium permanganate to yield the nitro acid 15. Coupling to S-pyrrolidinemethanol gave the nitro alcohol 16 which was hydrogenated followed by alloc protection of the resulting amine to give alcohol 17. Exposure to TEMPO/BAIB¹⁴ led to smooth oxidative ring closure to give 18. The best conditions for the final Boc deprotection were found to be TFA/DCM/water 47/ 47/6 which afforded the capping unit **6** in an overall yield of 17% over 7 steps. This intermediate required only one chromatographic purification and was chirally pure. Further protection of the C11-hydroxy functionality before the next coupling step was found to be unnecessary due to the superior nucleophilicity of the amino functionality. The C11-methyl ether (8) of the PBD acid capping unit described by Wells and co-workers¹⁵ was employed in order to improve coupling efficiency and avoid side-reactions. Finally, cascade removal of all protective groups of 9 under Deziel conditions¹⁶ yielded the extended PBD dimer 4^{17} in 85% yield.

PBD dimer 4 was shown to effectively interstrand crosslink linear plasmid pUC18 DNA with an XL₅₀ value of 0.23 μ M (Fig. 2 and Table 1),¹⁸ a potency 4-fold less than SJG-136 (0.06 μ M) but approximately 100-fold greater than the clinically used nitrogen mustard melphalan (20.0 μ M). As the molecule is designed to span



Figure 2. Cross-linking gel¹⁸ for extended PBD dimer **4** (AT-235) using linearized pUC18 DNA. Controls: DS, double stranded DNA; SS, single stranded DNA. The XL₅₀ was determined to be 0.23 μ M.

Table 1. Average cytotoxicity values from the NCI 60-cell panel andresults from a K562 line, and DNA cross-linking data for 1 (SJG-136)and extended dimer 4

Compound	NCI (µM)			K562 GI_{50}^{b}	${\rm XL}_{50}{}^{\rm a}$
	GI_{50}^{b}	TGI ^c	LC_{50}^{d}	(µM)	(µM)
1 4	0.007 0.014	0.087 0.49	0.562 22.9	0.008 0.037	0.060 0.23

 $^a XL_{50}{:}$ Dose providing 50% cross-linking of DS pUC18 plasmid DNA. 18

^bGI₅₀: Dose inhibiting 50% cell growth.

°TGI: Dose inhibiting 100% cell growth.

^d LC₅₀: Dose killing 50% of cells.

11 DNA base pairs with a degree of selectivity within this span, the probability of the perfectly matched site(s) appearing within the pUC18 sequence is lower than that for SJG-136 which spans 6 bp, and this could be one possible explanation for the lower observed potency of 4 in this assay compared to 1.

PBD dimer 4 was also shown to have significant cytotoxicity in both the NCI 60-cell line panel and in the K562 leukaemia cell line (Table 1).



Scheme 2. Reagents and conditions: (a) 10% Pd/C, H_2 , DMF, quant; (b) 11, DMF, 50%; (c) 10% Pd/C, H_2 , DMF, quant; (d) 13, DMF, 94%; (e) NaOH, MeOH, water, quant.



Scheme 3. Reagents and conditions: (a) PPh₃, DEAD, (3-hydroxy-propyl)carbamic acid *tert*-butyl ester, THF; (b) KMnO₄, water, acetone, 32% over 2 steps; (c) *S*-pyrrolidinemethanol, EDCI, HOBt, DMF, 93%; (d) 10% Pd/C, H₂, EtOAc, quant; (e) allyl chloroformate, pyridine, DCM, 86%; (f) TEMPO, BAIB, DCM, 72%; (g) TFA, DCM, water (47/47/6), quant.

The sub-micromolar GI_{50} and TGI cytotoxicity values suggest that, despite its molecular weight (984.07) and length (spanning 11 bp), **4** is still capable of traversing cellular and nuclear membranes to reach the genome of these cells.

DNA footprinting⁸ of **4** on a 543 bp fragment of BCL-2 DNA¹⁹ showed that it binds with high affinity (i.e., 10 nM) to a 5'-GCTTATAATGG-3' sequence (Fig. 3). This may be rationalized through simulation studies (Table 2) involving 4 bound to a range of potential binding sites which predict a preference for interstrand rather than intrastrand cross-linking with the covalently modified guanines on opposite strands separated by 7 bp (i.e., sequence B). The predicted binding energy at the main footprint site ($\Delta \Delta E_{\text{bind}} = 42.6$. kcal mol⁻¹) was higher than those values calculated for the hypothetical cross-linking motifs shown in Table 2. However, as sequences A. B and C are not present in the footprinted BCL-2 fragment, it is possible that higher-affinity binding sites may exist in longer DNA sequences. Figure 4 shows a molecular model in which 4 is bound to the footprinted 5'-GCTTATAATGG-3' sequence. The two PBD moieties are bound to the 3'-CGA (opposite 5'-GCT) and 5'-TGG triplet sequences (covalent binding



Figure 3. DNA footprinting gel of 4 bound to a 543 bp fragment of BCL-2 DNA¹⁹ showing a well-defined footprint at a 5'-AGCTTA TAATGG-3' site at a concentration of 10 nM or above after 18 h incubation.

Table 2. Energy calculations using duplex DNA sequences A-F to predict preferences for interstrand (A-C) versus intrastrand (D-F) cross-linking and the optimal number of base pairs (6–8) separating covalently modified guarines

	Simulated binding sequence $(5' \rightarrow 3')$	$\Delta\Delta E_{\rm bind}^{\rm a}$ /kcal mol ⁻¹
A	CGCAGAAAATTTCTGCG	13.6 ± 0.6
В	CGCAGAAAATTTCTGCG	$0 74 \pm 0.6$
D	CGCAGAAAATTTGTGCG	10.1 ± 0.6
Е	CGCAGAAAATTTGTGCG	17.6 ± 0.6
F	CGCAGAAAATTTTGTGCG	23.5 ± 0.6

The results predict that sequence **B** (5'-A<u>G</u>AAAATTT<u>C</u>T-3') has the lowest binding energy, suggesting that **4** is more likely to form an interstrand rather than an intrastrand cross-link with a spacing of 7 bp between the covalently-linked guanine residues. ^a $\Delta\Delta E_{\text{bind}}$ is the MM-PBSA^{22,23} binding energy relative to the lowest

 $\Delta\Delta E_{\text{bind}}$ is the MM-PBSA^{22,23} binding energy relative to the lowest energy binding sequence which was **B** (thus appearing as 0 kcal mol⁻¹). The error estimate is the standard error in the mean over all included trajectory snapshots.



Figure 4. Representative snapshot from a molecular dynamics simulation (Sander program in Amber8²⁰) showing 4 bound to a 5'- $G\underline{C}TTATAAT\underline{G}G$ -3' sequence with the two PBD units covalently bound to guanines on opposite strands (positions underlined).

base pair positions underlined) with the heterocyclic linker interacting with the central 5'-TATAA base pairs as predicted by the literature.^{2,15} According to this model, the dimer takes up the curvature of the DNA minor groove with little distortion of the helix.

In Figure 3 an additional weaker footprint was observed at a 5'-TCACTATCTCCCGGTTA-3' sequence at a concentration of approximately 1 μ M. This sequence does not contain a predicted cross-linking site for 4, but does contain potential sites of monoalkylation for one PBD unit such as 3'-GGG, a preferred monomeric PBD binding sequence.²¹ This could explain the lower preference (by 2 orders of magnitude) of 4 for this site.

Through a greater understanding of SAR for extended PBD dimers of this type, we hope to provide the basis of a novel approach to DNA sequence recognition through an interstrand cross-linking mechanism. This is being further explored by the synthesis and evaluation of analogues of **4** with differently-structured linkers which will be reported in due course.

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References and notes

- 1. Thurston, D. E. *Chemistry and Pharmacology of Anticancer Drugs*; CRC Press (Taylor & Francis): Boca Raton, Florida, USA, 2006.
- Kielkopf, C. L.; White, S.; Szewczyk, J. W.; Turner, J. M.; Baird, E. E.; Dervan, P. B.; Rees, D. C. *Science* 1998, 282, 111.
- 3. Thurston, D. E. Br. J. Cancer 1999, 80, 65.
- Schwartz, G. H.; Patnaik, A.; Hammond, L. A.; Rizzo, J.; Berg, K.; Von Hoff, D. D.; Rowinsky, E. K. Ann. Oncol. 2003, 14, 775.
- Alley, M. C.; Hollingshead, M. G.; Pacula-Cox, C. M.; Waud, W. R.; Hartley, J. A.; Howard, P. W.; Gregson, S. J.; Thurston, D. E.; Sausville, E. A. *Cancer Res.* 2004, 64, 6700.
- Gregson, S. J.; Howard, P. W.; Hartley, J. A.; Brooks, N. A.; Adams, L. J.; Jenkins, T. C.; Kelland, L. R.; Thurston, D. E. J. Med. Chem. 2001, 44, 737.
- Hartley, J. A.; Spanswick, V. J.; Brooks, N.; Clingen, P. H.; McHugh, P. J.; Hochhauser, D.; Pedley, R. B.; Kelland, L. R.; Alley, M. C.; Schultz, R.; Hollingshead,

M. G.; Schweikart, K. M.; Tomaszewski, J. E.; Sausville, E. A.; Gregson, S. J.; Howard, P. W.; Thurston, D. E. *Cancer Res.* **2004**, *64*, 6693.

- Martin, C.; Ellis, T.; McGurk, C. J.; Jenkins, T. C.; Hartley, J. A.; Waring, M. J.; Thurston, D. E. *Biochemistry* 2005, 44, 4135.
- Bando, T.; Iida, H.; Saito, I.; Sugiyama, H. J. Am. Chem. Soc. 2001, 123, 5158.
- 10. Kumar, R.; Lown, J. W. Eur. J. Med. Chem. 2005, 40, 641.
- Howard, P. W.; Gregson, S. J.; Tiberghien, A.C. WO 2005/085250 A1 Sep 15th, 2005.
- 12. Boger, D. L.; Fink, B. E.; Hedrick, M. P. J. Am. Chem. Soc. 2000, 122, 6382.
- Fukuyama, T.; Lin, S. C.; Li, L. P. J. Am. Chem. Soc. 1990, 112, 7050.
- DeMico, A.; Margarita, R.; Parlanti, L.; Vescovi, A.; Piancatelli, G. J. Org. Chem. 1997, 62, 6974.
- Wells, G.; Martin, C. R. H.; Howard, P. W.; Sands, Z. A.; Laughton, C. A.; Tiberghien, A.; Woo, C. K.; Masterson, L. A.; Stephenson, M. J.; Hartley, J. A.; Jenkins, T. C.; Shnyder, S. D.; Loadman, P. M.; Waring, M. J.; Thurston, D. E. J. Med. Chem. 2006, 49, 5442.
- 16. Deziel, R. Tetrahedron Lett. 1987, 28, 4371.
- 17. Data for **4**. ¹H NMR (400 MHz, DMSO-*d*₆): δ 9.90 (s, 3H), 8.10 (m, 1H), 7.79 (d, 2H, *J* = 4.36 Hz), 7.35 (s, 2H), 7.25 (s, 1H), 7.20 (2s, 2H), 7.05 (s, 1H), 6.90 (s, 2H), 6.84 (s, 2H), 4.21–4.01 (m, 4H), 3.85–3.81 (m, 15H), 3.61 (m, 2H), 3.46–3.35 (m, 6H), 2.45 (m, 2H), 2.38–2.17 (m, 4H), 2.13–1.89 (m, 8H); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 168.7, 164.2, 158.3, 150.2, 150.1, 146.8, 140.5, 122.7, 122.0, 121.9, 119.7, 118.4, 118.4, 117.8, 114.8, 111.2, 110.0, 104.6, 104.2, 103.9, 67.7, 66.4, 55.8, 55.5, 53.6, 53.3, 48.5, 46.3, 36.0, 35.8, 35.5, 30.2, 28.9, 28.9, 28.7, 24.7, 23.6, 22.3; IR (cm⁻¹): *v* 3306, 2949, 1638, 1597, 1555, 1508, 1465, 1435, 1405, 1262, 1216, 1090, 1064. MS (ES⁺) *m/z* (relative intensity): 984.4 ([M+H]⁺, 100); HRMS (TOF MS ES⁺): Calcd for C₅₁H₅₇N₁₁O₁₀ ([M+H]⁺): 984.4363. Found: 984.4359.
- Hartley, J. A.; Berardini, M. D.; Souhami, R. L. Anal. Biochem. 1991, 193, 131.
- Accession No. NC_000018.8 (Chromosome 18; DNA Bases 59136813–59136823).
- Amber Molecular Dynamics Package: http:// amber.scripps.edu.
- Thurston, D. E.. In *Molecular Aspects of Anticancer Drug–DNA Interactions*; Neidle, S., Waring, M. J., Eds.; The Macmillan Press Ltd.: London, UK, 1993; Vol. 1, p 54.
- Kollman, P. A.; Massova, I.; Reyes, C.; Kuhn, B.; Huo, S. H.; Chong, L.; Lee, M.; Lee, T.; Duan, Y.; Wang, W.; Donini, O.; Cieplak, P.; Srinivasan, J.; Case, D. A.; Cheatham, T. E. *Acc. Chem. Res.* **2000**, *33*, 889.
- 23. Srinivasan, J.; Cheatham, T. E.; Cieplak, P.; Kollman, P. A.; Case, D. A. J. Am. Chem. Soc. **1998**, *120*, 9401.