Cascade Reaction between Methyl 3-Dehydroshikimate, Arylamines, and 2-Chloroalkyl Esters under Microwave Conditions: A Practical and Biomass-Based Synthesis of N-Aryl-1,4-benzoxazin-3-ones

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Abstract: A biomass-based, metal- and ligand-free assembly of *N*-aryl-1,4-benzoxazin-3-ones from methyl 3-dehydroshikimate (3-MHDS), involving aliphatic to aromatic transformation followed by C–O and C–N forming reactions, is presented. A variety of *N*-aryl-1,4-benzoxazin-3-ones were efficiently synthesized in a one-pot consecutive manner under microwave conditions.

Key words: methyl 3-dehydroshikimate, *N*-aryl-1,4-benzoxazin-3ones, metal-free, cascade reaction, cyclization

N-Substituted 1,4-benzoxazin-3-ones are an intriguing class of privileged scaffold for a variety of biologically interesting molecules, ranging from potential drugs, natural products, and agrochemicals.¹ For example, various bioactive molecules (Figure 1) such as mineralocorticoid receptor modulating agents,² SGLT2 inhibitors,³ thrombin inhibitors,⁴ and the herbicide flumioxazin,⁵ are all characterized by the N-substituted 1,4-benzoxazin-3-one motif. As a result, considerable efforts have been made to synthesize these compounds.⁶ Generally, methods to access this scaffold involve the treatment of 2-halophenols and 2aminophenols with bifunctional reagents such as 2-haloamides.7 2-haloalkanoyl halides,⁸ or excess 2bromoalkanoates9 through palladium- or copper-catalyzed and ligand-involved processes. The reaction of 2-nitrophenols with a bifunctional reagent can also lead to 1,4benzoxazin-3-ones, but a reductive process is further required to convert the nitro group into an amino group.¹⁰ However, the aforementioned methods for the assembly of N-aryl-1,4-benzoxazin-3-ones are indirect, require preformed coupling partners such as 2-(2-halophenoxy)alkanoates or N-aryl-2-haloacetamides, and use transitionmetal catalysts.^{7,9b} In addition, iodo-substituted precursors and long reaction times are required to obtain satisfactory yields.^{7b} Very recently, the Qi group reported an impressive, one-pot, three-component protocol for the construction of N-aryl-1,4-benzoxazin-3-ones through palladium(II) acetate and XPhos-catalyzed reactions between 2-halophenols, ethyl 2-bromoalkanoates, and arylamines under microwave conditions.¹¹ However, this

SYNTHESIS 2014, 46, 1167–1176 Advanced online publication: 17.03.2014 DOI: 10.1055/s-0033-1338606; Art ID: SS-2013-H0833-OP © Georg Thieme Verlag Stuttgart · New York method still suffered from drawbacks such as the requirements for transition-metal catalysts, phosphine ligands, and high temperatures (150 °C). Moreover, from the perspective of sustainable chemistry, the utilization of renewable biomass, the construction of platform compounds, and the development of cost-efficient methods to convert platform compounds into various value-added chemicals, are fundamental tasks for chemists.¹² To the best of our knowledge, no metal- and ligand-free protocol for the synthesis of *N*-aryl-1,4-benzoxazin-3-ones has been reported that is based on renewably sourced materials. Therefore, an alternative method that is facile, efficient, and makes use of biomass-derived starting materials to access N-substituted 1,4-benzoxazin-3-ones is of genuine interest.



Figure 1 Examples of bioactive N-substituted 1,4-benzoxazin-3-ones

Over the past few years, a novel strategy to access to Nsubstituted anilines from nonaromatic cyclohexenones via sequential amination/aromatization reactions has emerged as an effective and alternative synthetic pathway.¹³ Our group has also recently demonstrated the feasibility of a biomass-derived, cost-competitive cyclohexenone, methyl 3-dehydroshikimate (3-MDHS, 1), in the facile synthesis of N-substituted 2-aminophenols via tandem cross-coupling and aromatization reactions.^{14a} In the course of our study, we envisioned that this aliphatic to aromatic methodology for generating N-substituted 2-aminophenols could further be exploited for the syn-

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thesis of N-aryl-1,4-benzoxazin-3-ones 3 through a cascade process by using inexpensive and readily accessible 3-MDHS (1), primary amines, and bifunctional reagents, such as chloroacetyl chloride or 2-chloroalkyl esters, as the reactants (Scheme 1). This approach may be metaland ligand-free, for the pivotal CAr-NH-CAr' subunit is generated by the formation of N-arylated 2-aminophenols 4 in the absence of a metal catalyst and ligand.^{14a} Hence, these N-arylated 2-aminophenols, which are generated in situ and highly functionalized, could serve as platform compounds for further transformations. In continuation of our work on the development of novel and practical methods for biomass conversion and utilization,14 we herein describe a facile, metal- and ligand-free protocol utilizing 3-MDHS (1) for the one-pot synthesis of N-aryl-1,4-benzoxazin-3-ones under microwave irradiation.



Scheme 1 Strategy utilizing 3-MDHS (1) for the construction of *N*-aryl-1,4-benzoxazin-3-ones

Based on an improved method from our previous findings^{14a} for the synthesis of **1** (see the Supporting Information), we started our investigation by using aniline (2a)as a model substrate in the reaction with 1 and 2-halogenated bifunctional reagents (Table 1). As expected, the cascade reaction between 1 and aniline (2a) catalyzed by 4-toluenesulfonic acid under microwave conditions in N,N-dimethylformamide, dimethyl sulfoxide, or N-methylpyrrolidin-2-one gave intermediate 4a (Step 1), followed by subsequent metal-free C-O and C-N forming reaction with chloroacetyl chloride to afford the desired product 3a in good yields (entries 1–3). Although chloroacetyl chloride seemed to be highly reactive and effective in this process, it is a moisture-sensitive, corrosive, and irritating substrate. Alternatively, we then carried out the reaction by using the more benign methyl chloroacetate as the bifunctional reagent and ethanol as the solvent. To our delight, **3a** was obtained in 69% isolated yield (entry 4). Subsequently, various solvents such as dimethyl sulfoxide, N,N-dimethylformamide, N-methylpyrrolidin-2-one, N,N-dimethylacetamide, PEG-200, PEG-400, and 1,4-dioxane were screened. We were pleased to find that the desired product 3a could be obtained in these solvents with yields ranging from 52-93%, among which N,N-dimethylformamide was optimal (entries 4-10). The effect of different bases on step 2 was also investigated. Results showed that both potassium and cesium carbonate gave excellent yields (entries 6 and 11), while other bases such as triethylamine, pyridine, DBU, sodium methoxide, and sodium hydroxide delivered low to moderate yields (entries 14-18). In contrast, the desired product 3a was not obtained and only the intermediate 4a was isolated in the absence of a base (entry 19). Temperature studies showed that 110 °C was favorable for this two-step conversion (entries 6, 20, and 21). In addition, results showed that microwave irradiation was advantageous over conventional heating in terms of yield and reaction time (entry 6 vs. 22). Therefore, the optimized reaction conditions consist of 1 (2.2 mmol) and arylamine 2 (2.0 mmol) in N,N-dimethylformamide (3 mL) using 4-toluenesulfonic acid (0.10 mmol) as the catalyst for step 1, and methyl chloroacetate (2.4 mmol) using potassium carbonate (4.0 mmol) as the base for step 2, both under microwave conditions at 110 °C (entry 6).

With the optimized reaction conditions in hand, a variety of arylamines were examined to explore the scope of the substrates. As summarized in Table 2, this one-pot, twostep transformation showed good compatibility with various functional groups such as Me, OMe, F, Cl, Br, I, NO₂, CF_3 , Ac, and CO_2Me , affording the desired products in moderate to high yields. In general, electron-rich anilines required shorter reaction times and afforded higher yields compared to electron-deficient substrates. For example, anilines possessing an electron-donating group such as methyl or methoxy gave excellent yields of 3b and 3c (entries 2 and 3), whereas substrates bearing electron-withdrawing groups such as acetyl, nitro, methoxycarbonyl, and trifluoromethyl afforded moderate yields of products **3h**,**i**,**q**,**r**,**t** (entries 8, 9, 17, 18, and 20). No reaction took place when 2,4-dinitroaniline was used as the substrate, which might be due to the reduced nucleophilicity of the corresponding substrate (entry 16). It could be concluded that the reactivity of halogen-substituted anilines decreased in the order I > Br > Cl > F by comparing the product yields of 3d-g, respectively (entries 4-7). The steric effect of the substrates was also observed. For instance, the ortho-substituted anilines afforded lower yields of the desired products 3j-m compared to the substrates bearing the same substituents at the *para*-position **3b**,**c**,**e**,**f** (entries 10-13 vs. 2, 3, 5, 6). Moreover, 2,4,6-trimethylaniline exhibited relative low reactivity in this reaction and gave 30 in 68% yield. It was noteworthy that the diamino substrate 4,4'-benzidine reacted smoothly with two equivalents of 3-MDHS (1) and methyl chloroacetate to afford the corresponding symmetrical bi-N-aryl-1,4-benzoxazin-3-one 3v (entry 22). Additionally, benzophenone hydrazone and 1naphthylamine also readily underwent this consecutive transformation and gave the desired products 3w,x, respectively, in high yields (entries 23 and 24). Notably, the workup procedure of the reaction is simple; the desired products 3 could substantially be separated in pure form from the reaction mixture by precipitating with water.

Finally, the scope and generality of our protocol was further explored by using methyl 2-chloropropanoate as the bifunctional reagent (Table 3). Gratifyingly, methyl 2-

tions and afforded the desired product in 82% yield (entry

6). In addition, high yields were obtained when polycyclic

arylamines such as 4'-chlorobiphenyl-4-ylamine and 1-

naphthylamine were used as substrates (entries 7 and 8).

It is worth mentioning that compound 5h existed as dia-

stereoisomers in a 1:1 molar ratio as confirmed by ¹H NMR spectroscopy, probably owing to the steric hin-

 Table 1 Optimization of the Reaction Conditions^a

MeO ₂ C	O + D + P-TsOH MW	→ MeO ₂ C, NH OH	step 2 bifunctional reagent, base MW	MeO ₂ C	
1	2a	4a		3a	
Entry	Solvent	Base	Step 1 ^b	Step 2 ^b	Yield ^c (%)
1	DMF	K ₂ CO ₃	110 °C, 8 min	60 °C, 90 min	90 ^d
2	DMSO	K ₂ CO ₃	110 °C, 8 min	60 °C, 90 min	87 ^d
3	NMP	K ₂ CO ₃	110 °C, 8 min	60 °C, 90 min	89 ^d
4	EtOH	K ₂ CO ₃	78 °C, 25 min	78 °C, 30 min	69
5	DMSO	K ₂ CO ₃	110 °C, 8 min	110 °C, 5 min	80
6	DMF	K ₂ CO ₃	110 °C, 8 min	110 °C, 5 min	93
7	NMP	K ₂ CO ₃	110 °C, 10 min	110 °C, 5 min	90
8	PEG-200	K ₂ CO ₃	110 °C, 10 min	110 °C, 15 min	65
9	PEG-400	K ₂ CO ₃	110 °C, 10 min	110 °C, 15 min	56
10	1,4-dioxane	K ₂ CO ₃	100 °C, 20 min	100 °C, 5 min	52
11	DMF	Cs ₂ CO ₃	110 °C, 10 min	110 °C, 5 min	93
12	DMF	Na ₂ CO ₃	110 °C, 10 min	110 °C, 10 min	85
13	DMF	NaHCO ₃	110 °C, 10 min	110 °C, 10 min	trace
14	DMF	Et ₃ N	110 °C, 10 min	110 °C, 8 min	45
15	DMF	pyridine	110 °C, 10 min	110 °C, 10 min	40
16	DMF	DBU	110 °C, 10 min	110 °C, 10 min	38
17	DMF	NaOMe	110 °C, 10 min	110 °C, 10 min	75
18	DMF	NaOH	110 °C, 10 min	110 °C, 10 min	60
19	DMF	_	110 °C, 10 min	110 °C, 110 min	e
20	DMF	K ₂ CO ₃	90 °C, 16 min	90 °C, 10 min	85
21	DMF	K ₂ CO ₃	130 °C, 8 min	130 °C, 10 min	89
22	DMF	K ₂ CO ₃	110 °C, 240 min	110 °C, 5 h	83 ^f
23	DMF	K ₂ CO ₃	110 °C, 8 min	60 °C, 1.5 h	90 ^d

^a Typical procedure under microwave conditions: step 1: **1** (0.41 g, 2.2 mmol), **2a** (2.0 mmol), *p*-TsOH (0.10 mmol), solvent (3.0 mL); step 2: add methyl chloroacetate (2.4 mmol), K₂CO₃ (4.0 mmol), stir.

^b The ramp time is included as part of the reaction time.

^c Isolated yield.

^d Chloroacetyl chloride was used as the bifunctional reagent for step 2 under conventional heating.

^e Not detected, only intermediate **4a** was obtained.

^f Reaction was carried out in an oil bath.

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drance of the bulky α -naphthyl group that hampers the inversion of the configuration and the rotation of the C–N bond.

In conclusion, a facile and metal-free protocol for the onepot assembly of *N*-aryl-1,4-benzoxazin-3-ones through the sequential reaction between 3-MDHS (1), anilines, and bifunctional 2-chloroalkyl esters has been developed. Various N-aryl-1,4-benzoxazin-3-ones were readily synthesized in moderate to excellent yields. Notable features

Table 2 Microwave-Assisted One-Pot Synthesis of N-Aryl-1,4-benzoxazin-3-ones^a

MeO ₂ C O +	ArNH ₂ + MeO CI —	<i>p</i> -TsOH, then base MeO ₂			
ŌН 1	2	14) 4 4	3 3		
Entry	Ar	T_1^{b} (min)	T_2^{b} (min)	Product	Yield ^c (%)
1	Ph	8	5	3a	93
2	4-MeOC ₆ H ₄	8	5	3b	97
3	$4-MeC_6H_4$	8	5	3c	96
4	$4-IC_6H_4$	8	6	3d	90
5	$4-BrC_6H_4$	8	6	3e	87
6	$4-ClC_6H_4$	10	7	3f	83
7	$4-FC_6H_4$	8	8	3g	78
8	$4-AcC_6H_4$	8	7	3h	85
9	$4-O_2NC_6H_4$	12	6	3i	82
10	$2-MeC_6H_4$	10	6	3ј	89
11	$2-BrC_6H_4$	12	7	3k	82
12	$2-ClC_6H_4$	12	7	31	75
13	2-MeOC ₆ H ₄	8	6	3m	90
14	2,5-Cl ₂ C ₆ H ₃	13	10	3n	72
15	2,4,6-Me ₃ C ₆ H ₂	12	8	30	68
16	2,4-(O ₂ N) ₂ C ₆ H ₃	20	0	3p	d
17	$3-\text{MeO}_2\text{CC}_6\text{H}_4$	8	7	3q	80
18	$3-F_3CC_6H_4$	8	7	3r	71
19	$3-MeC_6H_4$	8	5	3s	92
20	$3-O_2NC_6H_4$	8	7	3t	83
21	$3-ClC_6H_4$	8	6	3u	85
22 ^e	$4-(4-H_2NC_6H_4)C_6H_4^{f}$	12	9	$3v^{g}$	88
23	1-naphthyl	8	5	3w	87
24	N=CPh ₂	8	6	3x	90

^a Reactions conditions: 1 (0.41 g, 2.2 mmol), 2 (2.0 mmol), p-TsOH (0.10 mmol), DMF (3.0 mL), methyl chloroacetate (2.4 mmol), K₂CO₃ (4.0 mmol), microwaves, 110 °C; step 1: T_1 ; step 2: T_2 .

^b The ramp time is included as part of the reaction time.

^d Not detected.

^e Reactions condition: **1** (0.82 g, 4.4 mmol), 4,4'-benzidine (2.0 mmol), *p*-TsOH (0.20 mmol), DMF (8.0 mL), methyl chloroacetate (6.0 mmol), K₂CO₃ (8.0 mmol) under typical procedures.

^f Aniline was 4,4'-benzidine.

^g Product is *N*,*N*'-biphenylbis[6-(methoxycarbonyl)-2*H*-1,4-benzoxazin-3(4*H*)-one].

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^c Isolated yield.

MeO ₂ C 0	ArNH ₂ + MeO CI -	<i>p</i> -TsOH, then base MW	MeO ₂ C		
Entry	Ar	T_1^{b} (min)	T_2^{b} (min)	Product	Yield ^c (%)
1	Ph	8	5	5a	92
2	4-MeC ₆ H ₄	8	5	5b	93
3	$4-FC_6H_4$	8	8	5c	75
4	$4\text{-BrC}_6\text{H}_4$	8	6	5d	83
5	$4-O_2NC_6H_4$	12	7	5e	72
6	$4-AcC_6H_4$	8	8	5f	82
7	$4-(4-ClC_6H_4)C_6H_4$	8	6	5g	87
8	1-naphthyl	8	6	5h	85 ^d

Table 3 Expanding of Substrate Scope Using Methyl 2-chloropropanoate as the Bifunctional Reagent^a

^a Reactions conditions: methyl 2-chloropropanoate was used as bifunctional reagent according to the typical microwave procedures.

^b The ramp time is included as part of the reaction time.

° Isolated yield.

 $^{d 1}$ H NMR spectrum showed that **5h** existed as diastereoisomers in a 1:1 molar ratio, which may be due to the steric hindrance of the bulky α -naphthyl group, hindering the inversion of the configuration and the rotation of the C–N bond (see the Supporting Information for details).

such as the biomass-involved transformation, metal- and ligand-free reaction conditions, step-efficient procedures, and broad substrate scope have made this protocol a practical alternative to the existing transition-metal-catalyzed methods.

(-)-Shikimic acid (chromatography grade) was kindly provided as a natural product by Guangxi Wan Shan Spice Co. Ltd. (-)-Methyl 3-dehydroshikimate was readily prepared from (-)-shikimic acid through an improved method based on our previously reported protocol (see the Supporting Information for details). Petroleum ether (PE) was boiling fraction 60-90 °C. Other reagents and solvents were purchased from commercial sources and used without further purification unless otherwise stated. Reactions were monitored by TLC. Column chromatography was performed with silica gel (200-300 mesh) using a EtOAc-PE system as eluent. Melting points were measured on a Thiele apparatus and are uncorrected. Microwave experiments were carried out with a scientific WBFY microwave reactor in a flask connected with a condenser under atmosphere pressure. This microwave reactor was a monomode device with a tunable power controller. Reaction temperature was detected using an infrared thermometer and the ramp time is included as part of the reaction time. ¹H NMR and ¹³C NMR spectra were measured on a 400 MHz spectrometer (¹H 400 MHz, ¹³C 100 MHz) using CDCl₃ or DMSO- d_6 as the solvent at r.t. calibrated using residual undeuterated solvent as an internal reference. HRMS spectra were recorded on a LC-Q-TOF (ESI) apparatus. MS spectra were measured on a Shimadzu GC-MS QP5050A in EI mode or a Thermo Finnigan LCQ DECA XP ion trap mass spectrometer in ESI mode.

Methyl 4-Aryl-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylates 3a–o,q–x; General Procedure

To a mixture of 3-MDHS (1; 0.41 g, 2.2 mmol) and *p*-TsOH (0.10 mmol) in DMF (3 mL) was added arylamine **2** (2.0 mmol). The flask was then placed in the microwave synthesizer and the mixture was irradiated (240 W, 110 °C) with stirring for the indicated time (T_1). Then, K₂CO₃ (0.55 g, 4 mmol) and methyl chloroacetate (2.4

mmol) were added to the mixture and it was irradiated (240 W, 110 °C) with stirring for the indicated time (T_2). Upon cooling, the mixture was poured into H₂O, the resulting solid was filtered and dried to furnish the desired product in pure form in most cases. If necessary, these products could be further purified by recrystallization (EtOAc–PE) or column chromatography (silica gel, 200–300, EtOAc–PE).

Methyl 3-Oxo-4-phenyl-3,4-dihydro-2*H*-1,4-benzoxazine-6carboxylate (3a)

White needles; yield: 0.53 g (93%); mp 161–163 °C; $R_f = 0.51$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.62 (dd, J = 8.4, 1.6 Hz, 1 H), 7.60–7.53 (m, 3 H), 7.37 (d, J = 7.6 Hz, 2 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.85 (d, J = 1.6 Hz, 1 H), 4.92 (s, 2 H), 3.69 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 165.3, 163.3, 148.2, 135.4, 130.6, 130.1, 129.0, 129.0, 125.2, 123.5, 116.9, 116.6, 67.4, 52.1.

MS (EI): m/z (%) = (283 [M]⁺, 100), 252 ([M – OCH₃]⁺, 23), 224 ([M – COOCH₃]⁺, 5), 198 (11), 178 (10), 77 (18).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₆H₁₃NNaO₄: 306.0737; found: 306.0735.

Methyl 4-(4-Methoxyphenyl)-3-oxo-3,4-dihydro-2*H*-1,4-ben-

zoxazine-6-carboxylate (3b) White solid; yield: 0.61 g (97%); mp 151–153 °C; $R_f = 0.52$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.60 (dd, J = 8.4, 2.0 Hz, 1 H), 7.27 (d, J = 9.2 Hz, 2 H), 7.16 (d, J = 8.4 Hz, 1 H), 7.12 (d, J = 9.2 Hz, 2 H), 6.90 (d, J = 2.0 Hz, 1 H), 4.89 (s, 2 H), 3.83 (s, 3 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.4, 159.2, 148.2, 130.8, 130.1, 127.6, 125.1, 123.5, 116.8, 116.6, 115.2, 67.5, 55.4, 52.1.

MS (EI): *m*/*z* (%) = 313 ([M]⁺, 100), 284 (73), 228 (10), 178 (8), 77 (10).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₅NNaO₅: 336.0842; found: 336.0838.

Methyl 4-(4-Methylphenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3c)

White solid; yield: 0.57 g (96%); mp 174–176 °C; $R_f = 0.43$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.60 (dd, J = 8.4, 2.0 Hz, 1 H), 7.39 (d, J = 8.4 Hz, 2 H), 7.23 (d, J = 8.4 Hz, 2 H), 7.16 (d, J = 8.4 Hz, 1 H), 6.88 (d, J = 2.0 Hz, 1 H), 4.90 (s, 2 H), 3.70 (s, 3 H), 2.40 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 165.3, 163.3, 148.2, 138.5, 132.7, 130.6, 128.6, 125.1, 123.5, 116.9, 116.8, 116.6, 67.4, 52.1, 20.8.

MS (EI): m/z (%) = 297 ([M]⁺, 91), 268 (100), 266 ([M – OCH₃]⁺, 18), 240 (9), 178 (10), 77 (23).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₅NNaO₄: 320.0893; found: 320.0887.

Methyl 4-(4-Iodophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3d)

Off-white solid; yield: 0.74 g (90%); mp 165–167 °C; $R_f = 0.62$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.98 (d, J = 8.8 Hz, 2 H), 7.64 (dd, J = 8.4, 1.6 Hz, 1 H), 7.22 (d, J = 8.4 Hz, 2 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.88 (d, J = 2.0 Hz, 1 H), 4.92 (s, 2 H), 3.74 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.3, 148.3, 139.0, 135.2, 131.3, 130.2, 125.4, 123.6, 117.0, 116.5, 95.4, 67.4, 52.1.

MS (EI): m/z (%) = 409 ([M]⁺, 100), 380 (73), 316 (18), 77 (18).

HRMS (ESI-TOF): $m/z \ [M + Na]^+$ calcd for $C_{16}H_{12}INNaO_4$: 431.9703; found: 431.9696.

Methyl 4-(4-Bromophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3e)

White solid; yield: 0.63 g (87%); mp 166–168 °C; $R_f = 0.47$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.80 (dd, J = 6.8, 2.0 Hz, 2 H), 7.62 (dd, J = 8.4, 2.0 Hz, 1 H), 7.36 (dd, J = 6.8, 2.0 Hz, 2 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.85 (d, J = 2.0 Hz, 1 H), 4.91 (s, 2 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.3, 148.3, 134.7, 133.2, 131.3, 130.3, 125.4, 123.6, 122.2, 117.0, 116.4, 67.4, 52.1.

MS (EI): *m/z* (%) = 363 ([M + 2]⁺, 100), 361 ([M]⁺, 99), 332 ([M – OCH₃]⁺, 73), 178 (18), 77 (8).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}^{-79}BrNNaO_4$: 383.9842; found 383.9832.

Methyl 4-(4-Chlorophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3f)

White solid; yield: 0.53 g (83%); mp 156–158 °C; $R_f = 0.61$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.67 (d, J = 8.8 Hz, 2 H), 7.63 (dd, J = 8.4, 2.0 Hz, 1 H), 7.43 (d, J = 8.8 Hz, 2 H), 7.19 (d, J = 8.4 Hz, 1 H), 6.85 (d, J = 2.0 Hz, 1 H), 4.91 (s, 2 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.4, 148.2, 134.3, 133.6, 131.0, 130.3, 130.2, 125.3, 123.6, 117.0, 116.4, 67.4, 52.1.

MS (EI): m/z (%) = 319 ([M + 2]⁺, 100), 317 ([M]⁺, 33), 286 ([M - OCH₃]⁺, 14), 240 (9), 178 (10).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}^{35}$ ClNNaO₄: 340.0347; found: 340.0340.

Methyl 4-(4-Fluorophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3g)

White solid; yield: 0.47 g (78%); mp 140–142 °C; $R_f = 0.40$ (EtOAc–PE, 1:4).

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¹H NMR (400 MHz, DMSO- d_6): δ = 7.62 (dd, J = 8.4, 1.6 Hz, 1 H), 7.44 (d, J = 6.8 Hz, 4 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.84 (d, J = 1.6 Hz, 1 H), 4.91 (s, 2 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.4, 161.8 (d, ¹*J*_{FC} = 245 Hz, 4′C), 148.2, 131.5 (d, ⁴*J*_{FC} = 2.2 Hz, 1′C), 131.3 (d, ³*J*_{FC} = 8.9 Hz, 2′C), 130.6, 125.3, 123.6, 117.2, 117.0 (d, ²*J*_{FC} = 22 Hz, 3′C), 116.9, 67.4, 52.1.

MS (EI): m/z (%) = 301 ([M]⁺, 100), 272 ([M - OCH₃]⁺, 98), 216 (7), 178 (8).

HRMS (ESI-TOF): $m/z \ [M + Na]^+$ calcd for $C_{16}H_{12}FNNaO_4$: 324.0643; found: 324.0636.

Methyl 4-(4-Acetylphenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3h)

White solid; yield: 0.55 g (85%); mp 160–162 °C; $R_f = 0.32$ (EtOAc–PE, 1:2.5).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.16$ (d, J = 8.4 Hz, 2 H), 7.63 (dd, J = 8.4, 2.00 Hz, 1 H), 7.55 (d, J = 8.4 Hz, 2 H), 7.21 (d, J = 8.4 Hz, 1 H), 6.85 (d, J = 2.0 Hz, 1 H), 4.93 (s, 2 H), 3.70 (s, 3 H), 2.66 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 197.3, 165.2, 163.3, 148.4, 139.5, 137.0, 130.2, 130.0, 129.4. 125.5, 123.6, 117.1, 116.6, 67.5, 52.1, 26.9.

MS (EI): m/z (%) = 325 ([M]⁺, 100), 294 ([M – OCH₃]⁺, 14), 254 (9), 178 (6).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{18}H_{15}NNaO_5$: 348.0842; found: 348.0839.

Methyl 4-(4-Nitrophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3i)

Faint yellow solid; yield: 0.54 g (82%); mp 183–185 °C; $R_f = 0.52$ (EtOAc–PE, 1:2.5).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.47$ (d, J = 8.8 Hz, 2 H), 7.74 (d, J = 8.8 Hz, 2 H), 7.66 (dd, J = 8.4, 2.0 Hz, 1 H), 7.24 (d, J = 8.4 Hz, 1 H), 6.88 (d, J = 2.0 Hz, 1 H), 4.96 (s, 2 H), 3.72 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.2, 163.4, 148.4, 147.4, 141.3, 130.6, 129.9, 125.6, 125.3, 123.7, 117.2, 116.5, 67.5, 52.1.

MS (EI): m/z (%) = 328 ([M]⁺, 100), 297 ([M – OCH₃]⁺, 18), 269 ([M – COOCH₃]⁺, 7), 178 (6), 77 (4).

HRMS (ESI-TOF): $m/z \ [M + Na]^+$ calcd for $C_{16}H_{12}N_2NaO_6$: 351.0588; found: 351.0583.

Methyl 4-(2-Methylphenyl)-3,4-dihydro-2*H*-1,4-benzoxazine-6carboxylate (3j)

White solid; yield: 0.53 g (89%); mp 93–95 °C; $R_f = 0.66$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.63 (dd, J = 8.4, 2.0 Hz, 1 H), 7.49–7.39 (m, 3 H), 7.29 (d, J = 7.6 Hz, 1 H), 7.20 (d, J = 8.4 Hz, 1 H), 6.73 (d, J = 2.0 Hz, 1 H), 4.95 (q, J = 27.60 Hz, 2 H), 3.69 (s, 3 H), 2.02 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 165.3, 162.9, 148.3, 136.3, 134.1, 131.5, 129.7, 129.5, 129.2, 127.8, 125.4, 123.8, 117.0, 116.1, 67.4, 52.1, 16.8.

MS (EI): m/z (%) = 297 ([M]⁺, 100), 268 ([M - OCH₃]⁺, 38), 240 ([M - COOCH₃]⁺, 9), 178 (5), 77 (7).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₅NNaO₄: 320.0893; found: 320.0896.

Methyl 4-(2-Bromophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3k)

Off-white solid; yield: 0.59 g (82%); mp 81–83 °C; $R_f = 0.54$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO-*d*₆): δ = 7.91 (dd, *J* = 8.4, 1.2 Hz, 1 H), 7.67–7.63 (m, 2 H), 7.62–7.58 (m, 1 H), 7.56–7.50 (m, 1 H), 7.23

(d, *J* = 8.4 Hz, 1 H), 6.71 (d, *J* = 2.0 Hz, 1 H), 4.96 (s, 2 H), 3.70 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.2$, 162.9, 148.2, 134.2, 133.9, 131.7, 131.5, 129.7, 129.2, 125.6, 123.8, 122.8, 117.2, 115.9, 67.4, 52.2.

MS (EI): m/z (%) = 363 ([M + 2]⁺, 27), 361 ([M]⁺, 25), 332 ([M - OCH₃]⁺, 9), 282 (100), 75 (18).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}BrNNaO_4$: 383.9842; found: 383.9844.

Methyl 4-(2-Chlorophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (31)

White solid; yield: 0.47 g (75%); mp 106–108 °C; $R_f = 0.63$ (EtOAc–PE, 1:2.5).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.78–7.75 (m, 1 H), 7.65 (dd, J = 8.4, 2.0 Hz, 1 H), 7.61–7.59 (m, 3 H), 7.22 (d, J = 8.4 Hz, 1 H), 6.74 (d, J = 2.0 Hz, 1 H), 4.97 (s, 2 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.2$, 163.0, 148.2, 132.5, 132.3, 131.7, 131.4, 130.7, 129.3, 129.2, 125.7, 123.9, 117.2, 115.8, 67.3, 52.2.

MS (EI): m/z (%) = 319 ([M + 2]⁺, 18), 317 ([M]⁺, 55), 286 ([M - OCH₃]⁺, 9), 282 (100), 240 (5), 178 (7), 75 (14).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}CINNaO_4$: 340.0347; found: 340.0349.

Methyl 4-(2-Methoxyphenyl)-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3m)

White solid; yield: 0.56 g (90%); mp 148–150 °C; $R_f = 0.29$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.60 (dd, J = 8.4, 2.0 Hz, 1 H), 7.53 (m, 1 H), 7.33 (dd, J = 8.0, 1.6 Hz, 1 H), 7.28 (dd, J = 8.0, 0.8 Hz, 1 H), 7.12–7.17 (m, 2 H), 6.83 (d, J = 2.0 Hz, 1 H), 4.91 (s, 2 H), 3.70 (s, 6 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.1, 155.2, 148.2, 130.9, 130.6, 130.3, 125.2, 123.7, 123.1, 121.4, 116.9, 116.2, 113.0, 67.3, 55.8, 52.1.

MS (EI): m/z (%) = 313 ([M]⁺, 100), 282 ([M - OCH₃]⁺, 23), 254 ([M - COOCH₃]⁺, 6), 178 (4), 75 (7).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₅NNaO₅: 336.0842; found: 336.0845.

Methyl 4-(2,5-Dichlorophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benz-oxazine-6-carboxylate (3n)

White solid; yield: 0.51 g (72%); mp 136–138 °C; $R_f = 0.71$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 7.87$ (d, J = 2.4 Hz, 1 H), 7.83 (d, J = 8.8 Hz, 1 H), 7.72 (dd, J = 8.8, 2.4 Hz, 1 H), 7.67 (dd, J = 8.4, 2.0 Hz, 1 H), 7.24 (d, J = 8.4 Hz, 1 H), 6.77 (d, J = 2.0 Hz, 1 H), 5.03–4.93 (dd, J = 15.2 Hz, 2 H), 3.75 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.1 (C=O), 163.0 (C=O), 148.0, 133.7, 132.8, 132.0, 131.8, 131.6, 131.4, 128.9, 125.8, 123.9, 117.3, 115.6, 67.2, 52.2.

MS (EI): m/z (%) = 351 ([M]⁺, 25), 320 ([M - OCH₃]⁺, 7), 316 (100), 178 (7).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₆H₁₁³⁵Cl₂NNaO₄: 373.9957; found: 373.9951.

Methyl 3-Oxo-4-(2,4,6-trimethylphenyl)-3,4-dihydro-2*H*-1,4benzoxazine-6-carboxylate (30)

White solid; yield: 0.44 g (68%); mp 120–122 °C; $R_f = 0.57$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.63 (dd, J = 8.4, 2.0 Hz, 1 H), 7.21 (d, J = 8.4 Hz, 1 H), 7.10 (s, 2 H), 6.71 (d, J = 2.0 Hz, 1 H), 4.98 (s, 2 H), 3.71 (s, 3 H), 2.32 (s, 3 H), 1.95 (s, 6 H). MS (EI): m/z (%) = 325 ([M]⁺, 100), 294 ([M – OCH₃]⁺, 9), 266 ([M – COOCH₃]⁺, 6), 238 (11), 77 (8).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₉H₁₉NNaO₄: 348.1206; found: 348.1215.

Methyl 4-[3-(Methoxycarbonyl)phenyl]-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3q) White solid: 0.54 g (80%); mp 119-121 °C: R = 0.48

White solid; yield: 0.54 g (80%); mp 119–121 °C; $R_f = 0.48$ (EtOAc–PE, 1:2.5).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.11$ (d, J = 8.0 Hz, 1 H), 7.98 (s, 1 H), 7.76 (t, 1 H), 7.69 (d, J = 8.0 Hz, 1 H), 7.62 (dd, J = 8.4, 2.0 Hz, 1 H), 7.19 (d, $J_1 = 8.4$ Hz, 1 H), 6.79 (d, J = 1.6 Hz, 1 H), 4.93 (s, 2 H), 3.87 (s, 3 H), 3.69 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.4$, 165.2, 163.5, 148.3, 135.9, 133.9, 131.6, 130.8, 130.4, 130.0, 129.7, 125.4, 123.6, 117.0, 116.4, 67.5, 52.4, 52.1.

MS (EI): m/z (%) = 341 ([M]⁺, 100), 310 ([M – OCH₃]⁺, 16), 282 ([M – COOCH₃]⁺, 3), 178 (5).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{18}H_{15}NNaO_6$: 364.0792; found: 364.0792.

Methyl 3-Oxo-4-[3-(trifluoromethyl)phenyl]-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3r)

White solid; yield: 0.50 g (71%); mp 108–110 °C; $R_f = 0.40$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO-*d*₆): δ = 7.94–7.73 (m, 4 H), 7.63 (dd, *J* = 8.4, 1.6 Hz, 1 H), 7.20 (d, *J* = 8.4 Hz, 1 H), 6.80 (d, *J* = 1.6 Hz, 1 H), 4.93 (s, 2 H), 3.71 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.2, 163.5, 148.2, 136.3, 133.5, 131.5, 130.9 (q, ${}^{2}J_{CF}$ = 34 Hz, 1 C), 130.2, 126.3, 126.0, 125.9, 125.4, 123.6 (q, ${}^{1}J_{CF}$ = 274 Hz, CF₃), 117.1, 116.3, 67.4, 52.1.

MS (EI): m/z (%) = 351 ([M]⁺, 100), 320 ([M – OCH₃]⁺, 18), 292 ([M – COOCH₃]⁺, 3), 178 (6).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₂F₃NNaO₄: 374.0611; found: 374.0602.

Methyl 4-(3-Methylphenyl)-3,4-dihydro-2*H*-1,4-benzoxazine-6carboxylate (3s)

White solid; yield: 0.55 g (92%); mp 137–139 °C; $R_f = 0.44$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.60 (dd, J = 8.4, 2.0 Hz, 1 H), 7.48 (t, 1 H), 7.34 (d, J = 8.0 Hz, 1 H), 7.15 (t, 3 H), 6.88 (d, J = 2.0 Hz, 1 H), 4.91 (s, 2 H), 3.70 (s, 3 H), 2.37 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.3, 148.3, 139.8, 135.3, 130.6, 129.9, 129.7, 129.3, 125.9, 125.2, 123.6, 116.9, 116.7, 67.5, 52.1, 20.8.

MS (EI): m/z (%) = 297 ([M]⁺, 100), 266 ([M – OCH₃]⁺, 73), 238 ([M – COOCH₃]⁺, 5), 178 (9), 77 (6).

HRMS (ESI-TOF): $m/z \ [M + Na]^+$ calcd for $C_{17}H_{15}NNaO_4$: 320.0893; found: 320.0894.

Methyl 4-(3-Nitrophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3t)

White solid, yield: 0.54 g (83%); mp >200 °C; $R_f = 0.44$ (EtOAc–PE, 1:2.5).

¹H NMR (400 MHz, DMSO-*d*₆): δ = 8.40 (m, 2 H), 7.90 (d, *J* = 4.8 Hz, 2 H), 7.64 (dd, *J* = 8.4, 2.0 Hz, 1 H), 7.21 (d, *J* = 8.4 Hz, 1 H), 6.82 (d, *J* = 1.6 Hz, 1 H), 4.93 (s, 2 H), 3.70 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.6, 148.8, 148.3, 136.5, 136.0, 131.6, 130.2, 125.5, 124.7, 124.1, 123.7, 117.2, 116.4, 67.5, 52.2.

MS (EI): m/z (%) = 328 ([M]⁺, 100), 297 ([M – OCH₃]⁺, 23), 253 (18).

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HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}N_2NaO_6$: 351.0588; found: 351.0585.

Methyl 4-(3-Chlorophenyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3u)

White solid; yield: 0.54 g (85%); mp 146–148 °C; $R_f = 0.62$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 7.64-7.51$ (m, 3 H), 7.58 (s, 1 H), 7.40-7.38 (m, 1 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.84 (d, J = 2.0 Hz, 1 H), 4.91 (s, 2 H), 3.72 (s, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.2, 163.4, 148.2, 136.8, 134.0, 131.7, 130.3, 129.3, 129.3, 128.0, 125.4, 123.6, 117.0, 116.4, 67.4, 52.2.

MS (EI): m/z (%) = 319 ([M + 2]⁺, 33), 317 ([M]⁺, 100), 286 ([M - OCH₃]⁺, 98), 258 ([M - COOCH₃]⁺, 3), 178 (9).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{16}H_{12}CINNaO_4$: 340.0347; found: 340.0348.

N,N'-Biphenylbis[6-(methoxycarbonyl)-2*H*-1,4-benzoxazin-3(4*H*)-one] (3v)

White solid; yield: 0.99 g (88%); mp >200 °C; $R_f = 0.24$ (EtOAc–PE, 1:2).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.03$ (d, J = 8.4 Hz, 4 H), 7.64 (dd, J = 8.4, 2.0 Hz, 2 H), 7.53 (d, J = 8.4 Hz, 4 H), 7.21 (t, J = 8.4 Hz, 2 H), 6.98 (d, J = 2.0 Hz, 2 H), 4.96 (s, 4 H), 3.71 (s, 6 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.3, 163.4, 148.4, 139.5, 135.1, 130.5, 129.7, 128.6, 125.4, 123.6, 117.0, 116.7, 67.5, 52.1.

MS (ESI): $m/z = 565 [M + H]^+$.

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{32}H_{24}N_2NaO_8$: 587.1425; found: 587.1416.

Methyl 4-(1-Naphthyl)-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3w)

Pink crystals; yield: 0.58 g (87%); mp 157–159 °C; $R_f = 0.53$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, CDCl₃): $\delta = 8.02$ (d, J = 8.4 Hz, 1 H), 7.95 (d, J = 8.0 Hz, 1 H), 7.69 (dd, J = 8.4, 2.0 Hz, 1 H), 7.62 (t, J = 8.0 Hz, 1 H), 7.43–7.56 (m, 4 H), 7.12 (d, J = 8.4 Hz, 1 H), 6.91 (d, J = 2.0 Hz, 1 H), 5.00–4.89 (dd, J = 15.2 Hz, 2 H), 3.68 (s, 3 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 165.9, 163.7, 148.3, 134.9, 131.4, 130.4, 130.1, 129.8, 128.8, 127.5, 127.4, 126.7, 126.1, 125.9, 124.9, 121.8, 118.0, 116.9, 68.0, 52.0.

MS (EI): m/z (%) = 333 ([M]⁺, 100), 302 ([M – OCH₃]⁺, 9), 291 (8), 244 (9), 178 (7), 77 (9).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{20}H_{15}NNaO_4$: 356.0893; found: 356.0888.

Methyl 4-[(Diphenylmethylene)amino]-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (3x)

White solid; yield: 0.69 g (90%); mp 146–148 °C; $R_f = 0.68$ (EtOAc–PE, 1:3).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.84 (d, J = 2.0 Hz, 1 H), 7.77– 7.73 (m, 3 H), 7.56–7.52 (m, 1 H), 7.44–7.30 (m, 5 H), 7.20–7.17 (m, 2 H), 6.96 (d, J = 8.4 Hz, 1 H), 4.42 (s, 2 H), 3.86 (s, 3 H).

 13 C NMR (100 MHz, CDCl₃): δ = 179.2, 166.3, 155.9, 147.8, 136.0, 134.7, 132.1, 129.9, 129.8, 129.1, 128.3, 128.1, 127.4, 126.3, 124.9, 116.6, 116.3, 67.5, 52.1.

MS (ESI): m/z (%) = 386 ([M]⁺, 21), 355 ([M – OCH₃]⁺, 3), 180 (100), 77 (53).

HRMS (ESI-TOF): $m/z \ [M + Na]^+$ calcd for $C_{23}H_{18}N_2NaO_4$: 409.1159; found: 409.1152.

Methyl 4-Aryl-2-methyl-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylates 5a-h; General Procedure

To a mixture of 3-MDHS (1; 0.41 g, 2.2 mmol) and *p*-TsOH (0.10 mmol) in DMF (3 mL) was added arylamine **2** (2.0 mmol). The flask was then placed into the microwave synthesizer and the mixture was irradiated (240 W, 110 °C) with stirring for the indicated time (T_1). Then, K₂CO₃ (0.55 g, 4 mmol) and methyl 2-chloropropanoate (2.4 mmol) were added to the mixture and it was irradiated (240 W, 110 °C) with stirring for the indicated time (T_2). Upon cooling, the mixture was poured into H₂O, the resulting solid was filtered and dried to furnish the desired product in pure form.

Methyl 2-Methyl-4-phenyl-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (5a)

White solid; yield: 0.55 g (92%); mp 145–147 °C; $R_f = 0.65$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.62 (dd, J = 8.4, 2.0 Hz, 1 H), 7.58 (t, J = 8.0 Hz, 2 H), 7.53 (t, J = 7.2 Hz, 1 H), 7.35 (d, J = 7.2 Hz, 2 H), 7.19 (d, J = 8.0 Hz, 1 H), 6.85 (d, J = 2.0 Hz, 1 H), 5.03 (q, J = 6.8 Hz, 1 H), 3.69 (s, 3 H), 1.54 (d, J = 6.80 Hz, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.5$, 165.3, 147.8, 135.8, 130.9, 130.1, 129.0, 128.9, 125.2, 123.7, 117.2, 116.6, 73.5, 52.1, 16.2.

MS (EI): m/z (%) = 297 ([M]⁺, 76), 266 ([M - OCH₃]⁺, 9), 254 (100).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{17}H_{15}NNaO_4$: 320.0893; found: 320.0894.

Methyl 2-Methyl-4-(4-methylphenyl)-3-oxo-3,4-dihydro-2H-

1,4-benzoxazine-6-carboxylate (5b) White solid; yield: 0.58 g (93%); mp 141–143 °C; $R_f = 0.49$ (EtOAc–PE, 1:5).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.62 (dd, J = 8.4, 2.0 Hz, 1 H), 7.38 (d, J = 8.4 Hz, 2 H), 7.21 (d, J = 8.0 Hz, 2 H), 7.18 (d, J = 8.4 Hz, 1 H), 6.88 (d, J = 2.0 Hz, 1 H), 5.01 (q, J = 6.8 Hz, 1 H), 3.70 (s, 3 H), 2.40 (s, 3 H), 1.53 (d, J = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.5$, 165.3, 147.8, 138.4, 133.1, 130.9, 130.6, 128.6, 125.2, 123.7, 117.1, 116.7, 73.5, 52.1, 20.7, 16.2.

MS (EI): m/z (%) = 311 ([M]⁺, 71), 296 ([M – CH₃]⁺, 2), 280 ([M – OCH₃]⁺, 5), 268 (100), 240 (6), 182 (4).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{18}H_{17}NNaO_4$: 334.1050; found: 334.1052.

Methyl 4-(4-Fluorophenyl)-2-methyl-3-oxo-3,4-dihydro-2H-

1,4-benzoxazine-6-carboxylate (5c) White solid; yield: 0.47 g (75%); mp 119–121 °C; $R_f = 0.62$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.62 (dd, J = 8.4, 2.0 Hz, 1 H), 7.44 (d, J = 7.2 Hz, 4 H), 7.20 (d, J = 8.4 Hz, 1 H), 6.84 (d, J = 2.0 Hz, 1 H), 5.01 (q, J = 6.8 Hz, 1 H), 3.71 (s, 3 H), 1.54 (d, J = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.7, 165.3, 161.8 (d, ¹*J*_{FC} = 245 Hz, 4′C), 147.8, 132.0 (d, ⁴*J*_{FC} = 2.5 Hz, 1′C), 131.3 (d, ³*J*_{FC} = 8.1 Hz, 2′C), 130.9, 125.3, 123.7, 117.3, 117.2 (d, ²*J*_{FC} = 22.6 Hz, 3′C), 116.6, 73.5, 52.1, 16.3.

MS (EI): m/z (%) = 315 ([M]⁺, 65), 284 ([M - OCH₃]⁺, 7), 272 (100), 254 (2).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for C₁₇H₁₄FNNaO₄: 338.0799; found: 338.0802.

Methyl 4-(4-Bromophenyl)-2-methyl-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (5d)

White solid; yield: 0.62 g (83%); mp 152–154 °C; $R_f = 0.51$ (EtOAc–PE, 1:5).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.79 (d, J = 8.4 Hz, 2 H), 7.63 (dd, J = 8.4, 1.6 Hz, 1 H), 7.35 (d, J = 8.4 Hz, 2 H), 7.20 (d, J = 8.4 Hz, 1 H), 6.85 (d, J = 2.0 Hz, 1 H), 5.01 (q, J = 6.8 Hz, 1 H), 3.72 (s, 3 H), 1.54 (d, J = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 165.5$, 165.3, 147.8, 135.2, 133.1, 131.3, 130.6, 125.4, 123.8, 122.0, 117.3, 116.5, 73.5, 52.1, 16.2.

MS (EI): m/z (%) = 377 ([M + 2]⁺, 69), 375 ([M]⁺, 65), 346 ([M - OCH₃]⁺, 8), 334 (100), 332 (98), 253 (6).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{17}H_{14}BrNNaO_4$: 397.9998; found: 397.9998.

Methyl 4-(4-Nitrophenyl)-2-methyl-3-oxo-3,4-dihydro-2*H*-1,4benzoxazine-6-carboxylate (5e)

Gray solid; yield: 0.49 g (72%); mp 151–153 °C; $R_f = 0.36$ (EtOAc–PE, 1:5).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.43$ (d, J = 8.8 Hz, 2 H), 7.71 (d, J = 8.4 Hz, 2 H), 7.65 (dd, J = 8.4, 2.0 Hz, 1 H), 7.24 (d, J = 8.4 Hz, 1 H), 6.87 (d, J = 1.6 Hz, 1 H), 5.01 (q, J = 6.8 Hz, 1 H), 3.70 (s, 3 H), 1.56 (d, J = 6.8 Hz, 3 H).

MS (EI): m/z (%) = 342 ([M]⁺, 82), 311 ([M - OCH₃]⁺, 9), 299 (100), 253 (41).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{17}H_{14}N_2NaO_6$: 365.0744; found: 365.0746.

Methyl 4-(4-Acetylphenyl)-2-methyl-3-oxo-3,4-dihydro-2*H*-1,4benzoxazine-6-carboxylate (5f)

Yellow solid; yield: 0.56 g (82%); mp 193–195 °C; $R_f = 0.38$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.15$ (d, J = 8.8 Hz, 2 H), 7.64 (dd, $J_1 = 8.4$, 2.0 Hz, 1 H), 7.54 (d, J = 8.0 Hz, 2 H), 7.22 (d, J = 8.4 Hz, 1 H), 6.86 (d, J = 2.0 Hz, 1 H), 5.03 (q, J = 6.8 Hz, 1 H), 3.70 (s, 3 H), 2.65 (s, 3 H), 1.55 (d, J = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 197.3$, 165.4, 165.2, 147.9, 140.0, 136.9, 130.5, 129.9, 129.3, 125.5, 123.8, 117.3, 116.6, 73.5, 52.1, 26.9, 16.1.

MS (EI): m/z (%) = 339 ([M]⁺, 65), 308 ([M - OCH₃]⁺, 8), 296 (100), 253 (8), 77 (9).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{19}H_{17}NNaO_5$: 362.0999; found: 362.1001.

Methyl 4-(4'-Chlorobiphenyl-4-yl)-2-methyl-3-oxo-3,4-dihydro-2*H*-1,4-benzoxazine-6-carboxylate (5g)

White solid; yield: 0.71 g (87%); mp 202–204 °C; $R_f = 0.55$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 7.88 (d, J = 8.4 Hz, 2 H), 7.80 (d, J = 8.4 Hz, 2 H), 7.64 (dd, J = 8.4, 2.0 Hz, 1 H), 7.55 (d, J = 8.4 Hz, 2 H), 7.45 (d, J = 8.0 Hz, 2 H), 7.21 (d, J = 8.4 Hz, 1 H), 6.93 (d, J = 1.6 Hz, 1 H), 5.04 (q, J = 6.8 Hz, 1 H), 3.68 (s, 3 H), 1.55 (d, J = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.9, 165.7, 148.1, 139.5, 138.0, 135.6, 133.1, 131.0, 129.8, 129.3, 128.9, 128.5, 125.7, 124.4, 117.6, 116.9, 73.8, 52.4, 16.5.

MS (EI): m/z (%) = 407 ([M]⁺, 73), 376 ([M - OCH₃]⁺, 4), 364 (100).

HRMS (ESI-TOF): m/z [M + Na]⁺ calcd for $C_{23}H_{18}CINNaO_4$: 430.0817; found: 430.0822.

Methyl 2-Methyl-4-(1-naphthyl)-3-oxo-3,4-dihydro-2*H*-1,4benzoxazine-6-carboxylate (5h)

Light pink solid; yield: 0.59 g (85%); mp 133–135 °C; $R_f = 0.50$ (EtOAc–PE, 1:4).

¹H NMR (400 MHz, DMSO- d_6): δ = 8.16–8.09 (m, 4 H), 7.75–7.45 (m, 12 H), 7.28 (d, J = 7.2 Hz, 1 H), 7.26 (d, J = 7.2 Hz, 1 H), 6.66 (d, J = 2.0 Hz, 1 H), 6.59 (d, J = 2.0 Hz, 1 H), 5.35 (q, J = 6.8 Hz, 1

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H), 5.15 (q, *J* = 6.8 Hz, 1 H), 3.60 (s, 6 H), 1.64 (d, *J* = 6.8 Hz, 3 H), 1.60 (d, *J* = 6.8 Hz, 3 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 165.9, 165.7, 165.2, 147.9, 147.7, 134.2, 129.7, 128.8, 128.7, 128.0, 127.7, 127.6, 127.3, 126.9, 126.8, 126.3, 126.3, 125.5, 125.4, 122.2, 121.3, 117.3, 117.2, 116.4, 116.4, 73.8, 73.5, 52.0, 16.4, 16.2.

MS (EI): m/z (%) = 347 ([M]⁺, 69), 316 ([M - OCH₃]⁺, 4), 304 (100), 77 (5).

HRMS (ESI-TOF): $m/z [M + Na]^+$ calcd for $C_{21}H_{17}NNaO_4$: 370.1050; found: 370.1054.

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