



A facile route for the synthesis of novel S-linked 1,3,5-triazine tethered peptidomimetics



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ABSTRACT

An efficient one-pot synthesis of N^{α} -protected S-linked 1,3,5-triazine tethered peptidomimetics is described. The protocol involves a three-component condensation reaction employing N^{α} -protected amino alkyl isothiouronium salt, formaldehyde and amino acid ester or aryl amine as reactants. Various aryl amines with substitutions and amino acids with simple as well as bifunctional side chains were employed to obtain triazine tethered peptidomimetics in good yield.

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Peptidomimetics have aroused great interest due to their wide utility in developing new therapeutic agents and in drug design.¹ One of the useful approaches for the synthesis of such peptidomimetics involves the incorporation of heterocyclic units at one or more peptide bonds. In this context, triazoles, tetrazoles, thiazoles, oxazoles, oxadiazoles and many other biologically important heterocycles have been incorporated into the peptide backbone and their biological properties have been studied.²

1,3,5-Triazine is an important heterocyclic unit that has found useful applications as antimicrobial and antitumor agents.³ It was also used as supramolecular agent,⁴ and in the synthesis of dyes,⁵ as well as DNA cleavage reagent.⁶ Triazine derivatives also showed non-peptidic prokineticin antagonist⁷ and analgesic properties.⁸ The S-linked 1,3,5-triazine derivatives also exhibit a wide range of pharmacological properties. Brown and co-workers⁹ described the synthesis of MAC13243, **1** a new antibacterial compound which inhibits the activity of LolA protein, a crucial component of the lipoprotein targeting pathway in bacteria (Fig. 1). Belonging to the similar class, 5-chlorouracil-linked pyrazolo 1,3,5-triazines, **2** serve as thymidine phosphorylase inhibitor.¹⁰

In an earlier report, Ralbovsky et al.,¹¹ described the synthesis of triazinedione through two different methods. In the first method, commercially available sulfonic acid salt of 2-methyl isothiourea was treated with isocyanate in the presence of NaOH to form an

intermediate which was then cyclized using methyl chloroformate and Et₃N at –10 °C to rt to obtain the triazinediones. Parallelly, commercially available thiourea was treated with CH₃Cl in the presence of MeOH to prepare isothiourea as an intermediate which was then cyclized with N-chlorocarbonylisocyanate to yield triazinedione. Kong et al.,¹² reported the synthesis of S-linked 1,3,5-triazine-2,4-diones through cyclization of the intermediate obtained by the reaction of ethoxycarbonylisocyanate, methyl

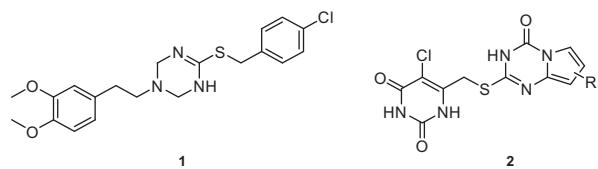


Figure 1. Examples of Biologically active S-linked 1,3,5-triazines.

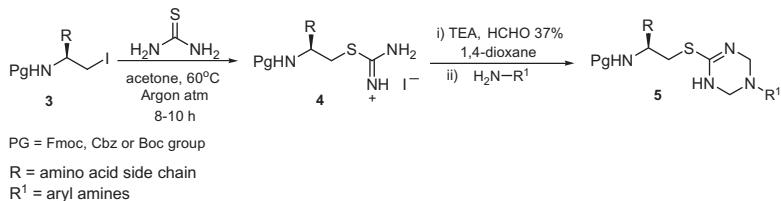
Table 1

Solvent screening for the synthesis of S-linked 1,3,5-triazine tethered peptidomimetics **5**

Entry	Solvent	Yield (%)
a	THF	38
b	Acetonitrile	65
c	EtOH	40
d	1,4-Dioxane	89
e	Methanol	30

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Scheme 1. Synthesis of S-linked 1,3,5-triazine tethered peptidomimetics **5**.

Table 2
List of S-linked 1,3,5-triazine tethered peptidomimetics 5

Entry	Reactant 4	Products 5	Yield ^a (%)
a	A chemical structure showing a benzyl group attached to a central carbon atom, which is also bonded to an Fmoc-amino group (-CH2-CH(NH2)-), a thiomethyl group (-CH2-S-), and an amino group (-NH2). A counterion, I-, is shown as NH2+I-.	A chemical structure showing the product of the cyclization of reactant 4a. It features a benzyl group attached to a central carbon, which is bonded to an Fmoc-amino group, a thiomethyl group, and a 1,3-dihydroimidazol-2-ylidene group (-C(=O)N1CCN(Cc1ccccc1)CC1=NN=C(S-)-C=C1).	89
b	A chemical structure showing a benzyl group attached to a central carbon atom, which is bonded to an Fmoc-amino group, a thiomethyl group, and an amino group. A counterion, I-, is shown as NH2+I-.	A chemical structure showing the product of the cyclization of reactant 4b. It features a benzyl group attached to a central carbon, which is bonded to an Fmoc-amino group, a thiomethyl group, and a 1,3-dihydroimidazol-2-ylidene group where the 5-position is substituted with a 4-chlorophenyl group.	83
c	A chemical structure showing a branched cyclohexylmethyl group attached to a central carbon atom, which is bonded to a Cbz-amino group, a thiomethyl group, and an amino group. A counterion, I-, is shown as NH2+I-.	A chemical structure showing the product of the cyclization of reactant 4c. It features a branched cyclohexylmethyl group attached to a central carbon, which is bonded to a Cbz-amino group, a thiomethyl group, and a 1,3-dihydroimidazol-2-ylidene group.	81
d	A chemical structure showing a tert-butylmethyl group attached to a central carbon atom, which is bonded to a Cbz-amino group, a thiomethyl group, and an amino group. A counterion, I-, is shown as NH2+I-.	A chemical structure showing the product of the cyclization of reactant 4d. It features a tert-butylmethyl group attached to a central carbon, which is bonded to a Cbz-amino group, a thiomethyl group, and a 1,3-dihydroimidazol-2-ylidene group.	79
e	A chemical structure showing a cyclopentylmethyl group attached to a central carbon atom, which is bonded to a Boc-amino group, a thiomethyl group, and an amino group. A counterion, I-, is shown as NH2+I-.	A chemical structure showing the product of the cyclization of reactant 4e. It features a cyclopentylmethyl group attached to a central carbon, which is bonded to a Boc-amino group, a thiomethyl group, and a 1,3-dihydroimidazol-2-ylidene group.	80

^a Yields correspond to the isolated pure S-linked 1,3,5-triazine-tethered peptidomimetics.

amine and benzyl bromide in the presence of NaH in THF followed by treatment with isocyanate at 60 °C. Lam and co-workers¹³ described the preparation of S-linked triazines using Mukaiyama reagent in the presence of DMAP at about 60–110 °C. These methods inherit certain limitations wherein the isocyanates employed for the reaction are not commercially available and are moisture sensitive and toxic, which otherwise poses difficulty while

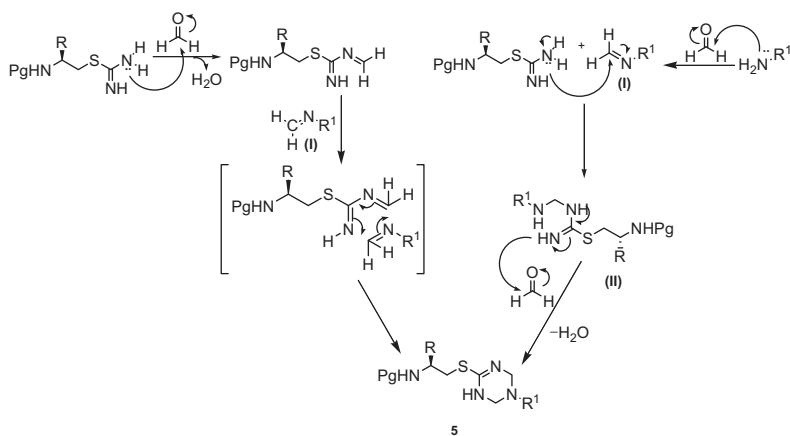
handling. Also the methods described involve long reaction duration and harsh reaction conditions. Zhao et al.¹⁴ reported the synthesis of N²-Cbz-amino acyl derived 3,5,6-trisubstituted 1,2,4-triazines using Cbz-protected amino acyl hydrazides and 1,2-diones in the presence of excess NH₄OAc at 180 °C.

Our group reported several new classes of peptidomimetics possessing heterocycles such as 1,3,4,-oxadiazole,¹⁵ 1,2,4,-oxadiazole,¹⁶ 1,3,4,-thiadiazole,¹⁷ triazoles¹⁸ and tetrazoles.¹⁹ This list also includes Fmoc-protected amino alkyl S/Se-linked tetrazoles²⁰ and Z/Boc-protected S-linked oxadiazole tethered peptidomimetics.²¹ Thus, with the continuing interest in designing heterocycle tethered peptidomimetics we envisaged the synthesis of the S-linked 1,3,5-triazine moiety incorporated into the peptide backbone.

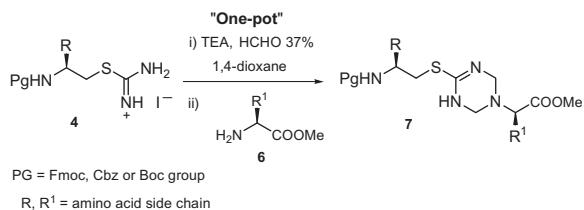
The starting material, N^α-protected amino alkyl isothiouronium salt **4** required for the present study was prepared through the literature protocol as reported earlier by our group.²² The N^α-protected amino acid was subjected to reduction and then to Mitsunobu conditions to form iodo compound. Thus obtained iodo compound was then treated with thiourea in acetone under argon atmosphere at reflux temperature for about 8–10 h to obtain corresponding isothiouronium salt, which was used as such for the next step.

In an initial study, isothiouronium salt derived from Fmoc-Phe-OH, that is, Fmoc-Phe- ψ [CH₂-SC(NH)NH₂]H^I **4a** was made to react with formaldehyde (37 wt % solution in water) and benzylamine as a test case in the presence of TEA in THF at rt. The product **5a** was obtained *albeit* in very low yield. The four solvents viz acetonitrile, EtOH, 1,4-dioxane and MeOH (**Table 1**) were screened in parallel reactions. An impressive yield of 89% of **5a** was isolated when the reaction was carried out using 1,4-dioxane as solvent, in the presence of 1.5 equiv of TEA, 2.0 equiv of HCHO, 1.5 equiv of amine and 1.0 equiv of isothiouronium salt (**Scheme 1**).

With the optimized reaction conditions in hand, the generality of the present protocol was tested by using aryl amines which afforded the title molecules in good yield (Table 2). A plausible reaction mechanism was proposed for the formation of compound 5 (Scheme 2).



Scheme 2. Plausible pathways for the formation of N²-protected S-linked 1,3,5-triazine tethered peptidomimetics 5.



Scheme 3. Synthesis of N^α-protected S-linked 1,3,5-triazine tethered dipeptidomimetics 7.

In the first step of the reaction, the amine was condensed with formaldehyde producing an imine intermediate [I], which subsequently reacts with a molecule of N-protected amino alkyl isothiourea, giving an intermediate [II]. This intermediate [II] in the presence of formaldehyde undergoes dehydrocyclization to form N^α-protected S-linked 1,3,5-triazine tethered peptidomimetics. The formation of the desired product can also be explained through another possible pathway wherein deprotonated isothiouronium salt reacts with HCHO to give the corresponding imine-like diunsaturated intermediate. This will then react with the intermediate

Table 3
List of N^α-protected S-linked 1,3,5-triazine tethered dipeptidomimetics 7

Entry	Reactant 4	Products 7	Yield (%)
a			81
b			80
c			83
d			85
e			82
f			76
g			72
h			79
i			75

(I) in an aza-Diels Alder like process to afford the product **5** as illustrated in **Scheme 2**.

In the next part of the study, we focused our attention towards the synthesis of dipeptidomimetics containing S-linked 1,3,5-triazines. In a typical experiment, Fmoc-Phe- ψ [CH₂-SC(NH)NH₂-HI] **4d** in 1,4-dioxane was treated with HCHO (37% wt, in solution of water), TEA and followed by the addition of a solution of H-Leu-OMe (deprotonation of HCl·H-Leu-OMe was carried out using zinc dust) in 1,4-dioxane. The reaction mixture was initially heated until the turbid solution turns into a clear solution. Then it is stirred at rt for about 6 h and the completion of the reaction was monitored through TLC. A simple work-up resulted in the corresponding product **7d**. The crude product was then purified through column chromatography using CHCl₃ and MeOH as eluent (95:5) to afford the pure product in 85% yield (**Scheme 3**). The efficacy of this protocol was demonstrated by the synthesis of a series of compounds employing several N^o-protected amino alkyl isothiouronium salts and amino acid methyl esters in moderate to good yields (**Table 3**).²³ All the compounds were isolated as stable ones and characterized through mass, ¹H and ¹³C NMR analyses.²⁴

The possibility of racemization, if any, during the synthesis of S-linked 1,3,5-triazine tethered peptidomimetics via the present protocol was assessed through RP-HPLC analysis of the intentionally made diastereomers, that is, Fmoc-L-Phg-triazine-(R)-PEA **5f** and Fmoc-L-Phg-triazine-(S)-PEA **5g**.²⁵ From these results it was found that the protocol was racemization-free and yielded optically pure products.

In the present Letter, we have demonstrated an application of N^o-protected amino alkyl isothiouronium salts as precursor units for the preparation of S-linked 1,3,5-triazine tethered peptidomimetics in one pot. The synthetic protocol implemented was straightforward, mild and avoids the usage of isothiocyanates used in earlier reports, which were albeit toxic and hazardous and handling of such molecules is difficult. The products were obtained in good yields and characterized by mass, ¹H and ¹³C analyses.

Acknowledgements

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2014.08.075>.

References and notes

- (a) Farmer, P. S. In *Drug Design*; Ariens, E. J., Ed.; Academic Press: San Diego, 1980; Vol. 10, pp 119–143; (b) Farmer, P. S.; Ariens, E. J. *Trends Pharmacol. Sci.* **1982**, *3*, 362–365.
- (a) Abel, A. D. *Lett. Pept. Sci.* **2002**, *8*, 267–272; (b) Welsch, M. E.; Snyder, S. A.; Stockwe, B. R. *Curr. Opin. Chem. Biol.* **2010**, *14*, 347–361; (c) Kharb, R.; Rana, M.; Sharma, P. C.; Yar, M. S. *J. Chem. Pharm. Res.* **2011**, *3*, 173–186; (d) Kim, S. J.; Lin, C. C.; Pan, C. M.; Ranaware, D. P.; Ramsey, D. M.; McAlpine, S. R. *Med. Chem. Comm.* **2013**, *4*, 406–410; (e) Mann, E.; Kessler, H. *Org. Lett.* **2003**, *5*, 4567–4570; (f) Banerjee, S.; Ganguly, S.; Sen, K. K. *J. Adv. Pharm. Educ. Res.* **2013**, *3*, 494–511; (g) Pedersen, D. S.; Abell, A. *Eur. J. Org. Chem.* **2011**, *2399–2411*; (h) Angell, Y. L.; Burgess, K. *Chem. Soc. Rev.* **2007**, *36*, 1674–1689; (i) Boeglin, D.; Cantel, S.; Heitz, A.; Martinez, J.; Ferrentz, J. A. *Org. Lett.* **2003**, *5*, 4465–4468; (j) Niu, T. F.; Wen-bin, Y. L.; Cai, C. *ACS Comb. Sci.* **2012**, *14*, 309–328; (k) Yu, K. L.; Johnson, R. L. *J. Org. Chem.* **1987**, *52*, 2051–2059; (l) Ko, E.; Liu, J.; Perez, L. M.; Lu, G.; Schaefer, A.; Burgess, K. *J. Am. Chem. Soc.* **2011**, *133*, 462–477.
- (a) Zhou, C.; Min, J.; Liu, Z.; Young, A.; Deshazer, H.; Gao, T.; Chang, Y. T.; Kallenbach, N. R. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 1308–1311; (b) Srinivas, K.; Bhanuprakash, K.; Harakishore, K.; Murthy, U. S. N.; Rao, V. J. *J. Med. Chem.* **2006**, *41*, 1240–1246; (c) Paquin, I.; Raeppe, S.; Leit, S.; Gaudette, F.; Zhou, N.; Moradei, O.; Saavedra, O.; Bernstein, N.; Raeppe, F.; Bouchain, G.; Frechette, S.; Woo, S. H.; Vaisburg, A.; Fournel, M.; Kalita, A.; Robe, M. F.; Lu, A.; Trachy-Bourget, M. C.; Yan, P. T.; Liu, J.; Rahil, J.; MacLeod, A. R.; Besterman, J. M.; Li, Z.; Delorme, D. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 1067–1071; (d) Zheng, M.; Xu, C.; Ma, J.; Sun, Y.; Du, F.; Liu, H.; Lin, L.; Li, C.; Ding, J.; Chen, K.; Jiang, H. *Bioorg. Med. Chem.* **2007**, *15*, 1815–1827; (e) Mandal, S.; Berube, G.; Asselin, E.; Mohammad, I.; Richardson, V. J.; Gupta, A.; Pramanik, S. K.; Williams, A. L.; Mandal, S. K. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 4955–4960.
- (a) Seki, T.; Yagai, Y.; Karatsu, T.; Kitamura, A. *J. Org. Chem.* **2008**, *73*, 3328–3335; (b) Whitesides, G. M.; Simanek, E. E.; Mathias, J. P.; Seto, C. T.; Chin, D.; Mammen, M.; Gordon, D. M. *Acc. Chem. Res.* **1995**, *28*, 37–44; (c) Zerkowski, J. A.; Seto, C. T.; Whitesides, G. M. *J. Am. Chem. Soc.* **1992**, *114*, 5473–5475.
- Um, S. I.; Kang, Y.; Lee, J. K. *Dyes Pigment.* **2007**, *75*, 681–686.
- Chen, J.; Wang, X.; Shao, Y.; Zhu, J.; Zhu, Y.; Li, Y.; Xu, Q.; Guo, Z. *Inorg. Chem.* **2007**, *46*, 3306–3312.
- Coats, S. J.; Dyatkin, A. B.; He, W.; Lisko, J.; Ralovsky, J. L.; Schultz, M. J.; Patent WO 2006104713, 2006.
- (a) Brown, E. D. *Eur. Pat. Appl.* EP 300756, 1979; (b) Balboni, G.; Lazzari, I.; Trapella, C.; Negri, L.; Lattanzi, R.; Giannini, E.; Nicotra, A.; Melchiorri, P.; Visentini, S.; Nuccio, C. D.; Salvador, S. *J. Med. Chem.* **2008**, *51*, 7635–7639.
- Pathania, R.; Zlitni, S.; Barker, C.; Das, R.; Gerritsma, D. A.; Lebert, J.; Awuah, E.; Melacini, G.; Capretta, F. A.; Brown, E. D. *Nat. Chem. Biol.* **2009**, *5*, 849–856.
- Sun, L.; Li, J.; Bera, H.; Dolzenko, A. V.; Chiu, G. N. C. *Eur. J. Med. Chem.* **2013**, *70*, 400–410.
- Ralovsky, J. L.; Lisko, J. G.; Palmer, J. M.; Mabus, J.; Chevalier, K. M.; Schulz, M. J.; Dyatkin, A. B.; Miskowski, T. A.; Coats, S. J.; Hornby, P.; Wei, H. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 2661–2663.
- Kong, K. H.; Tan, C. K.; Lam, Y. J. *Comb. Chem.* **2009**, *11*, 1050–1060.
- Kong, K. H.; Tan, C. K.; Lin, X.; Lam, Y. *Chem. Eur. J.* **2012**, *18*, 1476–1486.
- Zhao, Z.; Leister, W. H.; Strauss, K. A.; Wisnoski, D. D.; Lindsley, C. W. *Tetrahedron Lett.* **2003**, *44*, 1123–1127.
- (a) Prabhu, G.; Sureshbabu, V. V. *Tetrahedron Lett.* **2012**, *53*, 4232–4234; (b) Lamani, R. S.; Nagendra, G.; Sureshbabu, V. V. *Tetrahedron Lett.* **2010**, *51*, 4705–4709.
- Sureshbabu, V. V.; Hemantha, H. P.; Naik, S. A. *Tetrahedron Lett.* **2008**, *49*, 5133–5136.
- Nagendra, G.; Ravi, S. L.; Narendra, N.; Sureshbabu, V. V. *Tetrahedron Lett.* **2010**, *51*, 6338–6341.
- (a) Narendra, N.; Vishwanatha, T. M.; Sureshbabu, V. V. *Int. J. Pept. Res. Ther.* **2010**, *16*, 283–290; (b) Sureshbabu, V. V.; Narendra, N.; Hemantha, H. P. *Protein Pept. Lett.* **2010**, *17*, 499–506.
- Sureshbabu, V. V.; Venkataramanarao, R.; Naik, S. A.; Chennakrishnareddy, G. *Tetrahedron Lett.* **2007**, *48*, 7038–7041.
- Sureshbabu, V. V.; Vasantha, B.; Hemantha, H. P. *Synthesis* **2011**, *9*, 1447–1455.
- Sureshbabu, V. V.; Vasantha, B.; Nagendra, G. *Tetrahedron Lett.* **2012**, *53*, 1332–1336.
- Sureshbabu, V. V.; Vishwanatha, T. M.; Vasantha, B. *Synlett* **2010**, 1037–1042.
- General procedure for the synthesis of compounds **(5a–e)** and **(7a–f)**
To a solution of isothiouronium salt, **4** (1 equiv), TEA (1.5 equiv) and HCHO (2.0 equiv, 37 wt % solution in water) in 1,4-dioxane was added a solution of aryl amine or amino acid methyl ester (1.5 equiv, deprotonation of hydrochloride salt of amino acid ester was carried out using zinc dust). The reaction mixture was heated to reflux to obtain a clear solution. It was allowed to stir for about 6 h at room temperature. After the completion of the reaction as indicated by TLC analysis, the reaction mixture was diluted with water (50 mL) and extracted into ethyl acetate (3 × 15 mL). The combined organic extracts were then washed with 5% HCl, water and brine solution, dried over anhydrous Na₂SO₄ and evaporated in vacuo. The crude product was purified through silica gel column chromatography (100–200 mesh) using CHCl₃/MeOH (95:5) as eluent.
- Characterization data for selected compounds **5a–e** and **7a–f**
*Spectroscopic data for compound **5a**:* ¹H NMR (400 MHz, CDCl₃): δ 2.52 (s, 2H), 2.89 (s, 4H), 2.92 (s, 2H), 3.48–3.56 (m, 2H), 3.98–4.02 (m, 1H), 4.14–4.18 (m, 1H), 4.44–4.46 (m, 2H), 5.09 (br s, 1H), 5.36 (br s, 1H), 7.16–7.73 (m, 18H); ¹³C NMR (100 MHz, CDCl₃): δ 29.8, 40.5, 47.0, 52.9, 54.6, 58.0, 65.4, 67.1, 125.2, 125.5, 126.0, 126.2, 126.3, 126.4, 126.5, 127.3, 128.2, 128.4, 128.7, 128.9, 132.1, 132.5, 134.8, 135.6, 136.1, 138.1, 142.9, 153.1, 155.8 ppm; ESI-MS: m/z [M+H]⁺ calcd for C₃₄H₃₄N₄O₂S: 563.24; found: 563.20.
*Spectroscopic data for compound **7a**:* ¹H NMR (400 MHz, CDCl₃): δ 1.58 (d, J = 7.0 Hz, 3H), 2.68 (s, 2H), 3.10 (s, 4H), 3.52–3.55 (m, 2H), 3.62 (s, 3H), 3.70–3.81 (m, 1H), 4.12–4.20 (m, 1H), 4.32 (t, J = 6.6 Hz, 1H), 4.47 (d, J = 6.4 Hz, 2H), 5.55 (br s, 1H), 5.69 (br s, 1H), 6.97–7.68 (m, 13H); ¹³C NMR (100 MHz, CDCl₃): δ 18.9, 31.9, 33.7, 47.8, 49.7, 50.8, 58.1, 59.0, 61.9, 64.1, 125.0, 125.5, 126.0, 126.2, 126.5, 127.5, 127.8, 128.6, 128.6, 130.8, 132.1, 132.8, 134.9, 135.9, 138.7, 141.0, 152.9, 156.0, 171.5 ppm; ESI-MS: m/z [M+H]⁺ calcd for C₃₁H₃₄N₄O₂S: 559.23; found: 559.20.
- Racemisation study: The RP-HPLC analysis of the pure products **5f** and **5g** showed single peaks at different R_t values, that is, at R_t = 16.16 and 16.94 min, respectively. These studies showed that the prepared compounds, **5f** and **5g** contain a single optically pure isomer and consequently it was proved that the present protocol was free from racemization. (Method for RP-HPLC analysis: Gradient 0.1% TFA, acetonitrile 30–100% in 30 min with a flow rate of 0.5 mL/min. λ_{max} = 254 nm.)