

Available online at www.sciencedirect.com



CARBOHYDRATE RESEARCH

Carbohydrate Research 338 (2003) 2149-2152

Note

www.elsevier.com/locate/carres

A simple one-pot method for the synthesis of partially protected mono- and disaccharide building blocks using an orthoesterification– benzylation–orthoester rearrangement approach

Balaram Mukhopadhyay, Robert A. Field*

Centre for Carbohydrate Chemistry, School of Chemical Sciences and Pharmacy, University of East Anglia, Norwich NR4 7TJ, UK

Received 15 May 2003; accepted 25 June 2003

Abstract

A simple one-pot method is reported for making partially protected glycosyl acceptors from O-methyl or S-alkyl/aryl glycosides of D-glucose, D-galactose, D-arabinose, L-rhamnose, L-fucose and lactose via orthoester formation, benzylation and selective hydrolysis.

© 2003 Elsevier Ltd. All rights reserved.

Keywords: Orthoester; Benzylation; Rearrangement; One-pot reaction

A basic challenge in oligosaccharide synthesis is the preparation of partially protected building blocks where the protecting groups can be manipulated such that each can be selectively removed during the course of the synthetic route. In connection with the synthesis of various biologically active oligosaccharides, we had a need for a practical approach to partially protected building blocks. In the present note we describe a simple one-pot method for the synthesis of some useful glycosyl acceptors and donors based on an orthoesterification–benzylation–orthoester rearrangement approach.

The well-known formation of cyclic orthoesters involving *cis*-hydroxyl groups and selective rearrangement thereof, has been successfully used as the key reaction in this strategy. A selection of methyl glycopyranosides (Table 1) were treated with triethylorthoacetate and catalytic *p*-toluenesulfonic acid in dry acetonitrile¹ to form the corresponding orthoesters, which were then benzylated² in the same pot to protect the remaining free hydroxyl groups (Scheme 1).[†] The reaction mixture was diluted with dichloromethane and shaken with 1M HCl

E-mail address: r.a.field@uea.ac.uk (R.A. Field).

solution to effect orthoester rearrangement, giving the expected products.⁴ The same approach could also be applied for the thioglycosides, but it worked equally well only when N,N-dimethylformamide has been used as solvent, as noted previously⁵ (Table 1, entries 5–7).

In conclusion, we have shown that a number of useful glycosyl building blocks can be derived from *O*-methyl or *S*-alkyl/aryl glycosides by executing several reaction steps in a single pot and purifying only at the final stage by flash chromatography. The strategy significantly reduces the time for making these building blocks, and overall yields are higher compared to those obtained by stepwise reactions. Scale-up reactions have been performed with compounds **1**, **3** and **10**, which showed that this procedure can be applied on a 10-g (50 mmol) scale (Table 1).

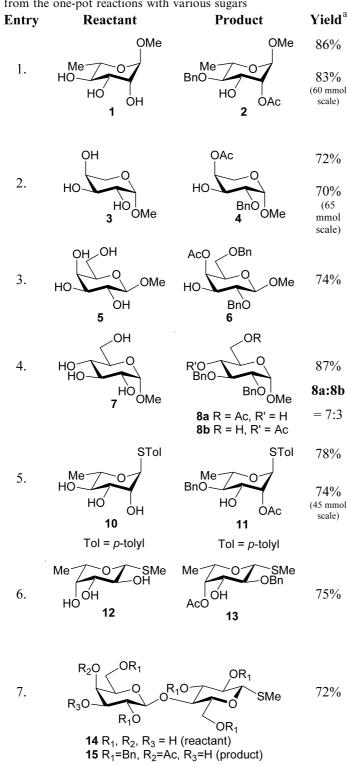
1. Experimental

1.1. General procedure

1.1.1. Methyl glycosides. To a suspension of the methyl glycoside (1 mmol) in dry MeCN (10 mL), triethylorthoacetate (1.5 mmol) was added, followed by addition of a catalytic amount of p-TsOH. The mixture was allowed to stir for 45–60 min. After complete conver-

^{*} Corresponding author. Fax: +44-1603-592003.

[†] For the two-pot synthesis of the corresponding 2,4-di-*O*-benzoyl derivative, see Ref. 3.



Results obtained from the one-pot reactions with various sugars

Comments

Compound 2 might be utilized for the synthesis of rhamnogalacturonan fragments found in plant cell wall extracellular matrix.¹¹ Compound 4 might be a useful precursor for plant saponin synthesis.¹²

Compound 6 could be used for the synthesis of Trypanosoma cruzi mucin oligosaccharides.13

Compounds 8a and 8b could be utilized as glycosyl acceptors for amylopectin branch point synthesis.¹⁴

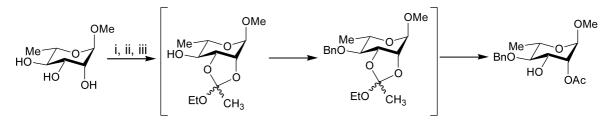
Compound 11 has been successfully utilized in this lab for the synthesis of а tetrasaccharide related to E. coli K-12 Oantigen.15

Compound 13 might be useful for the synthesis of NODfactor IX. where fucose is а nonterminal sugar.¹⁶

Compound 15 might be a useful precursor of sialylated lactoside based the on ganglioside GM_3 structure.¹⁷

^aUnless otherwise stated, yields refer to reactions performed on a 1 mmol scale.

Table 1



Scheme 1. (i) Triethyl orthoacetate, p-TsOH, CH₃CN; (ii) BnBr, NaH; (iii) 1 N HCl.

sion thin layer chromatography (TLC) [2:1 *n*-hexane-EtOAc], Et₃N was added to neutralize the solution. NaH (1.5 mmol, 60% dispersion in mineral oil) was added, followed by BnBr [1.2 mmol (2.2 mmol for glucose derivative)], and the mixture was allowed to stir for 1 h at room temperature. When TLC (3:1 *n*-hexane-EtOAc) showed complete conversion, MeOH (1 mL) was carefully added to destroy excess NaH, and the mixture was diluted with CH₂Cl₂ (20 mL). The organic layer was shaken with M HCl $(3 \times 15 \text{ mL})$, followed by washing with satd NaHCO₃ solution $(3 \times 15 \text{ mL})$ and water $(3 \times 15 \text{ mL})$. Finally the organic layer was separated, dried (Na₂SO₄) and evaporated to syrup. The crude product was purified by flash chromatography using 2:1 *n*-hexane–EtOAc as eluent. The yields are given in Table 1.

1.1.2. *S*-Alkyl/aryl glycosides. The same experimental procedure described for *O*-glycosides (above) was followed, except that N,N-dimethylformamide has been used as solvent instead of MeCN.⁵

Compounds $2,^{6} 6,^{7} 8a^{8,9}$ and $8b^{10}$ are known; specific rotation and NMR data were in accord with the literature. Specific rotation, NMR and HRMS data of new compounds 4, 11, 13 and 15 are given below.

1.2. Methyl 4-*O*-acetyl-2-*O*-benzyl-β-Larabinopyranoside (4)

[α]_D²³ +125.1° (*c* 2.1, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 7.38–7.26 (m, 5 H, aromatic protons), 5.14 (m, 1 H, H-4), 4.73, 4.64 (2d, 2 H, CH₂Ph), 4.67 (d, 1 H, $J_{1,2}$ 3.9 Hz, H-1), 4.11 (m, 1 H, H-3), 3.75 (dd, 1 H, $J_{5a,5b}$ 12.9 Hz, H-5a), 3.71 (dd, 1 H, $J_{1,2}$, $J_{2,3}$ 9.9 Hz, H-2), 3.62 (dd, 1 H, $J_{5a,5b}$, H-5b), 3.33 (s, 3 H, O–CH₃), 2.60 (bs, 1 H, OH), 2.11 (s, 3 H, COCH₃). ¹³C NMR (100 MHz, CDCl₃): δ 171.0 (COCH₃), 138.1, 128.5, 128.0 (aromatic carbons), 98.2 (C-1), 76.6, 72.9, 71.3, 67.0, 60.2, 55.4 (OCH₃), 20.9 (COCH₃). HRMS: Calcd for C₁₅H₂₄NO₆ (M+NH₄): 314.1598; found: *m/z* 314.1597.

1.3. *p*-Tolyl 2-*O*-acetyl-4-*O*-benzyl-1-thio-α-L-rhamnopyranoside (11)

 $[\alpha]_{D}^{23}$ -148.1° (*c* 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 7.45-7.12 (m, 9 H, aromatic protons), 5.42

(dd, 1 H, $J_{1,2}$ 1.6 Hz, $J_{2,3}$ 2.8 Hz, H-2), 5.41 (d, 1 H, $J_{1,2}$, H-1), 4.93, 4.77 (2d, 2 H, J 11.2 Hz, CH₂Ph), 4.30 (m, 1 H, H-5), 4.12 (dd, 1 H, $J_{2,3}$, $J_{3,4}$ 9.6 Hz, H-3), 3.50 (t, 1 H, $J_{3,4}$, H-4), 3.19 (bs, 1 H, OH), 2.34 (s, 3 H, S– PhCH₃), 2.17 (s, 3 H, COCH₃), 1.42 (d, 3 H, $J_{5,6}$ 6.4 Hz, H-6). ¹³C NMR (100 MHz, CDCl₃): δ 170.6 (COCH₃), 137.9, 137.5, 132.0, 129.8 129.6, 128.2, 127.6 (aromatic carbons), 85.9 (C-1), 81.4, 74.9, 74.2, 70.3, 68.4, 20.8 (COCH₃), 20.8 (S–C₆H₄–CH₃), 17.6 (C–CH₃). HRMS: Calcd for C₂₂H₃₀NO₅S (M+NH₄): 420.1839; found: m/z 420.1843.

1.4. Methyl 4-*O*-acetyl-2-*O*-benzyl-1-thio-β-L-fucopyranoside (13)

[α]²³_D -11.3° (*c* 1.9, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 7.39–7.23 (m, 5 H, aromatic protons), 5.02 (d, 1 H, $J_{3,4}$ 3.2 Hz, H-4), 4.87, 4.67 (2d, 2 H, *J* 10.4 Hz, CH₂Ph), 4.27 (d, 1H, $J_{1,2}$ 9.6 H_z, H-1) 3.70 (dd, 1 H, $J_{2,3}$ 9.2 Hz, $J_{3,4}$, H-3), 3.51 (q, 1 H, $J_{5,6}$ 6.4 Hz, H-5), 3.44 (t, 1 H, $J_{1,2}$ 9.2 Hz, $J_{2,3}$, H-2), 3.01 (bs, 1 H, OH), 2.20 (s, 3 H, S–CH₃), 2.10 (s, 3 H, COCH₃), 1.11 (d, 3 H, $J_{5,6}$ 6.4 Hz, H-6). ¹³C NMR (100 MHz, CDCl₃): δ 171.2 (COCH₃), 137.8, 128.2, 128.0, 127.7 (aromatic carbons), 84.9 (C-1), 78.1, 75.2, 73.2, 72.9, 72.7, 20.6 (COCH₃), 16.3 (S–CH₃), 12.8 (C–CH₃). HRMS: Calcd for C₁₆H₂₆NO₅S (M+NH₄): 344.1526; found: *m*/*z* 344.1528.

1.5. Methyl 4-*O*-acetyl-2,6-di-*O*-benzyl- β -D-galactopyranosyl- $(1 \rightarrow 4)$ -2,3,6-tri-*O*-benzyl-1-thio- β -D-glucopyranoside (15)

[α]²³_D +21.3° (*c* 1.2, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 7.47–7.22 (m, 25 H, aromatic protons), 5.37 (dd, 1 H, $J_{4',5'}$ 3 Hz, H-4'), 5.15–4.27 (10d, 10 H, 5 CH₂Ph), 4.51 (d, 1 H, $J_{1',2'}$ 6.2 Hz, H-1'), 4.41 (d, 1 H, $J_{1,2}$ 6.6 Hz, H-1), 4.11 (dd, 1 H, $J_{2',3'}$ 6.4 Hz, H-3'), 3.87–3.81 (m, 3 H, H-5', H-6'_a, H-6'_b), 3.70–3.61 (m, 2 H, H-4, H-5), 3.59–3.43 (m, 3 H, H-2, H-2', H-3), 3.40 (m, 2 H, H-6_a, H-6_b). ¹³C NMR (100 MHz, CDCl₃): δ 170.9 (COCH₃), 139.1, 138.3, 138.2, 137.9, 137.8, 128.3–127.2, 102.3 (C-1'), 84.4 (C-1), 85.1, 80.2, 79.9, 76.0, 75.9, 75.3, 75.2, 73.2, 72.9, 72.5, 71.8, 69.5, 67.2, 20.5(COCH₃), 12.3 (S–CH₃). HRMS: Calcd for

 $C_{50}H_{60}NO_{11}S$ (M+NH₄): 882.3882; found: *m*/*z* 882.3887.

Acknowledgements

This work was supported by the EPSRC. We gratefully acknowledge the EPSRC Mass Spectrometry Service Centre, Swansea for invaluable support.

References

- 1. Auzanneau, F. I.; Bundle, D. R. Can. J. Chem. 1993, 71, 534–548.
- 2. Brimacombe, J. S. Methods Carbohydr. Chem. 1972, 6, 376–378.
- 3. Wessel, H. P.; Bundle, D. R. Carbohydr. Res. 1983, 124, 301–311.
- Bouchra, M.; Calinaud, P.; Gelas, J. Carbohydr. Res. 1995, 267, 227–237.
- 5. Auzanneau, F.-I.; Bundle, D. R. Carbohydr. Res. 1991, 212, 13-24.
- Wessel, H. P.; Bundle, D. R. J. Chem. Soc. Perkin Trans. 1 1995, 2251–2260.

- 7. Paulsen, H.; Hasenkamp, T.; Paal, M. Carbohydr. Res. 1985, 144, 45–56.
- Paulsen, H.; Buensch, H. Chem. Ber. 1981, 114, 3126– 3145.
- Boren, H. B.; Ekborg, G.; Eklind, K.; Garegg, P. J.; Pilotti, A.; Swahn, C. G. Acta Chem. Scand. 1973, 27, 2639–2644.
- 10. Wolflehner, W. Carbohydr. Res. 1978, 65, 132-137.
- O'Neil, M. A.; Warrenfeltz, D.; Kates, K.; Pellerin, P.; Doco, T.; Darvill, A. G.; Albersheim, P. J. Biol. Chem. 1996, 271, 22923–22930.
- Haralampidis, K.; Trojanowska, M.; Osbourn, A. E. Adv. Biochem. Eng. Biotechnol. 2002, 75, 31–49.
- Todeschini, A. R.; Mendonça, L.; Previato, J. O.; Varki, A.; van Halbeek, H. *Glycobiology* 2000, 10, 213–221.
- Buleon, A.; Colonna, P.; Planchot, V.; Ball, S. Int. J. Biol. Macromol. 1998, 23, 85–112.
- 15. Mukhopadhyay, B.; Field, R. A.; (Manuscript in preparation).
- (a) Wang, Y.; Spellman, M. W. J. Biol. Chem. 1998, 273, 8112–8118;
 (b) Moloney, D. J.; Lin, A. I.; Haltiwanger, R. S. J. Biol. Chem. 1997, 272, 19046–19050;
 (c) Stults, N. L.; Cummings, R. D. Glycobiology 1993, 3, 589–596.
- 17. Duclos, R. I. Carbohydr. Res. 2000, 328, 489-507.