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Enhanced Reactivity by Torsional Strain of Cyclic Diaryliodonium in Cu-Catalyzed Enantioselective Ring-Opening Reaction



An efficient method is developed for preparing functionalized axially chiral compounds via the ring-opening reaction of cyclic diaryliodonium salts. Two conformers of the cyclic diaryliodonium salt observed in the crystal structure underwent quick equilibration. The distortion of the diaryliodonium salts significantly increased reactivity toward the chiral copper catalyst, which enabled the reaction to take place in mild conditions to furnish the products in enantioselectivities up to 99.5:0.5 (*R*:*S*).



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HIGHLIGHTS

Significant reactivity improvement by torsional strain of cyclic diaryliodoniums

Rapid access to optically active amino aryliodide atropisomers

Highly atom-economic synthesis of aryliodide atropisomers via ring-opening reaction

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Enhanced Reactivity by Torsional Strain of Cyclic Diaryliodonium in Cu-Catalyzed Enantioselective Ring-Opening Reaction

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SUMMARY

Atropisomers are stereoisomers arising from the restricted rotation around a single bond. In particular, biaryl atropisomers represent an important class of compounds, as they are widely present in natural products, ligands and pharmaceutical molecules. However, the preparation of structurally diverse biaryl atropisomers under mild conditions is a significant challenge. Here, we describe a Cu-bis(oxazolinyl)pyridine-catalyzed asymmetric ring-opening amination reaction of cyclic diaryliodoniums. Increasing the torsional strain of these cyclic compounds significantly improved the reactivity of cyclic diaryliodoniums. Computational investigation indicated that the two conformers of the cyclic diaryliodoniums had a low rotational barrier, and generally the reaction achieved high yields and high enantioselectivity (up to >99% ee). Furthermore, this ring-opening amination reaction also featured high atom economy in comparison with traditional reactions involving diaryliodonium. Finally, we propose a catalytic cycle and a mechanistic model that accounts for the observed enantioselectivity.

INTRODUCTION

Atropisomerism is a type of stereoisomerism where rotation via a single bond is restricted. Atropisomers are frequently seen in amides, biaryls, and diaryl ethers.^{1,2} In some cases, the atropisomeric architecture is the key feature in bioactive natural products, pharmaceuticals (i.e., vancomycin) (Scheme 1B), and privileged ligands or catalysts (i.e., [1,1'-binaphthalene]-2,2'-diol [BINOL] and KenPhos) (Scheme 1C).³⁻⁶ In contrast to stereogenic sp³ carbons bearing four distinct substituents, biaryl atropisomerism requires two conditions: (1) a restricted rotation around a stable axis, typically a bond, and (2) different substituents on both sides of the axis (A \neq B and C \neq D; Scheme 1A). Thus, the axis of a stable biaryl atropisomer is usually sterically congested, and its construction is challenging. Because of their importance and widespread applications in both medicinal and synthetic chemistry, a number of strategies for the construction of biaryl atropisomers have been developed, asymmetric cross-coupling,^{7–11} including metal-catalyzed asymmetric "aromatic substitution,"¹² and *de novo* aromatic ring formation (such as asymmetric [2 + 2 + 2] cycloaddition).^{13–15} In addition, modification of the existing biaryl structures enables access to enantio-enriched atropisomers via either a desymmetrization reaction or (dynamic) kinetic resolution (Scheme 1D).¹⁶⁻²⁵ Furthermore, the construction of benzylic center chirality, followed by center-to-axial chirality transfer aromatization is also an effective approach to access enantio-enriched biaryl atropisomers.^{26,27} Recently, there have been a number of reports on organocatalyzed asymmetric atropisomer synthesis.^{28–36} Ring opening of strained biaryl compounds

The Bigger Picture

Axially chiral biaryls represent an important class of stereoisomers that arise from restricted rotation around a single bond. They can be found among natural products, ligands, and pharmaceutical molecules. Divergent and atomeconomic synthetic approaches toward optically active biaryl atropisomers, in particular ones generating little to no waste, constitute a significant challenge. Here, we describe a highly enantioselective coppercatalyzed ring-opening reaction of cyclic diaryliodoniums where increasing the torsional strain of the diaryliodonium salts improved the relative poor reactivity of the cyclic diaryliodoniums. This method enabled us to access enantio-enriched amino aryliodides with higher atom economy than traditional reactions involving diaryliodonium. These axially chiral molecules are important precursors for divergent synthesis of axially chiral compounds, which are potentially useful catalysts or ligands in industry.

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is a powerful method for the construction of functionalized biaryl atropisomers. Bringmann and co-workers reported the asymmetric reduction of biaryl lactones to form a series of axially chiral 2'-(hydroxymethyl)-[1,1'-biphenyl]-2-ol analogs, and a number of atropisomeric natural products have been efficiently synthesized by this protocol.^{3,4} Nickel-catalyzed ring opening of dinaphthothiophene to afford axially chiral 1,1'-binaphthyls was described by Hayashi and colleagues.³⁷ In addition, the same group reported an elegant palladium-catalyzed asymmetric carbonylation reaction of dinaphthaleneiodonium, albeit with low enantioselectivity.³⁸ It is of significance to develop synthetic methods to access atropisomers that can serve as "platform molecules", i.e., compounds that can be readily converted to structurally diverse atropisomers. Recently, copper-catalyzed aminations of diaryliodonium have emerged as important methods for the synthesis of aniline derivatives.^{39–45} Diaryliodoniums act as powerful reagents for arylation of nucleophiles in both asymmetric and non-asymmetric reactions, most notably in contributions from the Gaunt group.^{46–53} We reasoned that the ring-opening reaction of cyclic diaryl iodoniums would give atropisomeric aryl iodides, which represent an ideal class of precursors for the divergent synthesis of biaryl atropisomers (Scheme 1E).^{54–59} The key challenges, however, are (1) how to increase the reactivity, because cyclic diaryliodoniums showed significantly lower activity than noncyclic reagents,⁵⁷ and (2) how to avoid the formation of diamination products or carbazoles. These requirements prompted us to establish proper reaction conditions that are mild enough to keep the aryl-iodine bonds in the products intact. Here, we report on the realization of this ring-opening amination process that directly converts cyclic iodonium salts to valuable atropisomeric products with very high enantiocontrol. The merits of this transformation include (1) a very high level of enantiocontrol, (2) high atom economy, and (3) high synthetic utility. The resulting aryl iodides are not only precursors for further diverse elaboration but also possible catalysts in hypervalent iodine oxidation reactions.⁵⁹

RESULTS AND DISCUSSION

As a result of the extreme importance of the C–N bond formation reaction in chemistry,⁶⁰ our initial trials started with the ring-opening reaction between cyclic diaryliodonium 1a (Figures S9-S13) and p-methylaniline with a copper salt as catalyst. X-ray single-crystal analysis found that iodonium 1a has two atropisomeric conformers (Figure 1; also see Data S1). Computational calculations indicated that the two conformers have a very low rotation barrier ($\Delta_r^{\neq} G_m^{\theta}$ = 36.1 kJ mol^{-1}) and are thus configurationally dynamic with a half-life around 1.2 \times 10⁻⁷ s under ambient conditions (Figures S2-S4). The ring-opening product 2a bearing four ortho substituents exhibits a much higher rotation barrier and should be configurationally stable. Our preliminary survey focused on identifying a class of suitable chiral ligands for asymmetric induction. The reactions with picolinamide derivative L1, PHOX (L2), and (R)-BINAP (L3) as ligands gave no atropselectivity, albeit virtually quantitative yields of 2a were obtained (Scheme 2, entries 1-3). The reaction with bis(oxazoline) L4 as ligand gave a complex mixture (entry 4). The Cu(OAc)₂/(ligand L5) combination delivered 2a in only 10% yield with 37% enantiomeric excess (ee) (entry 5). Displacing bis(oxazoline) ligand with tri-dentated bis(oxazolinyl)pyridine L6 boosted the yield, and the enantioselectivity was also increased to 54% (entry 6). Phenylsubstituted ligand L7 further improved the enantiomeric excess (entry 7). The use of a cationic copper source Cu(OTf)₂ or CuOTf·0.5PhH boosted the enantioselectivity to 85%, but the valence of the copper pre-catalysts had no influence on either reactivity or selectivity (entries 8 and 9). A decrease in loading of the ligand L7 $(Cu(OTf)_2/L7 = 1:1)$ did not result in deterioration of the yield, whereas the

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Scheme 1. Strategies in Biaryl Atropisomer Synthesis

- (A) Model structure of biaryl atropisomers.
- (B) Structure of vancomycin.
- (C) Ligands with a biaryl atropisomeric skeleton.
- (D) Strategies for the synthesis of asymmetric biaryl atropisomers.

enantioselectivity was significantly reduced (entry 10). Further screening of the ligands revealed that benzyl- or 2,3-dihydro-1*H*-indene-derived ligands L8 and L9 improved the enantiomeric excess to 99% with almost quantitative yields (entries 11 and 12). Copper(I) complexes A1 or A2 with a composition of Cu:ligand = 1:2

⁽E) This work.

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Figure 1. Structure and Reactivity Properties of Cyclic Diaryliodoniums Reaction conditions: the reaction was catalyzed by $Cu(Bn-PyBox)_2PF_6$ (A1) in the presence of

 $BnNH_2$ and Na_2CO_3 in dichloromethane.

prepared independently according to the literature⁶¹ gave more consistent results for the reactions wherein (sub)milligrams of catalyst was used. In addition, the loading of the catalyst A1 or A2 could be decreased to 5 mol % without compromising the yield or enantioselectivity of 2a (entries 13 and 14) (Figures \$98, \$99, and \$188–\$189).

A single crystal of 1a was obtained by slow evaporation of the solvent (methanol/ $H_2O = 10:1$) at room temperature. The crystal structure of 1a revealed that the distortion of C5-C6/C7-C12 was over 30°, and both (S) and (R) conformers were found in the crystal packing (Figure 1; also see Data S1). However, the corresponding structures without ortho methyl (1b)⁶² or with one ortho methyl group (1c) barely have distortion (0.3° and 3.6°, respectively) (Figure 1; also see Data S2), and all the atoms of the biaryl moiety are in approximately the same plane (Figures S14–S18). In sharp contrast to compound 1a, there are no stereoisomers ((R) or (S) conformers) in the crystal structures of 1b or 1c (Figure 1). Furthermore, the angles of C5–C6–C7 and C6–C7–C12 of 1a were slightly larger than those of 1b. It was reported that the five-membered cyclic diaryliodonium showed much lower reactivity than noncyclic [Ar₂I]⁺ for aromatic nucleophilic substitution reactions.^{57,63} Our studies found that the reactivity of these cyclic diaryliodonium depended significantly on their structural properties. Substrates with more distorted structure exhibited higher reactivity (Figure 1). Specifically, under the optimized catalytic system, the reaction of compound 1b and benzylamine gave 5% yield of the corresponding product at 0°C after 24 hr and 97% yield at 40°C after 5 hr. However,

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Entry	Cu source	Ligand/mol%	Temp./ºC	Yield/%	ee%
1	Cu(OAc) ₂ , 20 mol%	L1 , 40 mol%	40	99	0
2	Cu(OAc) ₂ , 20 mol%	L2 , 40 mol%	40	99	0
3	Cu(OAc) ₂ , 20 mol%	L3 , 40 mol%	40	99	0
4	Cu(OAc) ₂ , 20 mol%	L4 , 40 mol%	40	complex	-
5	Cu(OAc) ₂ , 20 mol%	L5 , 40 mol%	40	10	37
6	Cu(OAc) ₂ , 20 mol%	L6 , 40 mol%	40	99	54
7	Cu(OAc) ₂ , 20 mol%	L7 , 40 mol%	40	99	60
8	Cu(OTf) ₂ , 20 mol%	L7 , 40 mol%	40	99	85
9	CuOTf·0.5PhH, 20 mol%	L7 , 40 mol%	40	99	85
10	Cu(OTf) ₂ , 20 mol%	L7 , 20 mol%	40	99	26
11	Cu(OTf) ₂ , 20 mol%	L8 , 40 mol%	0	99	99
12	Cu(OTf) ₂ , 20 mol%	L9 , 40 mol%	0	99	99
13	A1 , 5 mol%	-	0	99	98
14	A2 , 5 mol%	-	0	99	98

Scheme 2. Optimization of the Reaction Conditions

The enantiomeric excess (ee) was determined by chiral high-performance liquid chromatography analysis.



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Scheme 3. Substrate Scope with Various Amines

Reaction conditions: cyclic diaryliodonium **1a** (1.0 equiv), amine (1.2 equiv), copper catalyst (5.0 mol %) (A1 for anilines, A2 for benzylamines), and Na₂CO₃ (3.0 equiv) in dichloromethane (0.050 mol L⁻¹) at 0°C or indicated otherwise. [†]ent-A1 was used; [‡]benzidine (0.5 equiv) was used; [#]benzylamine was used as a limiting reagent.

the reaction of **1c** only produced the desired product in 30% yield after 5 hr at 40°C, possibly as a result of enhanced steric hindrance. Pleasingly, the ring-opening reaction of the more torsionally strained substrate **1a** proceeded uneventfully at 0°C and afforded the corresponding product in 98% yield after 24 hr.

With the optimal conditions for asymmetric ring-opening reaction in hand, we examined the scope for the substrate (Scheme 3). First, a panel of substituted aniline derivatives were tested (2a-2x) (Figures S98-S148 and S188-S231). Alkyl substituents had a negligible effect on either the yield or enantioselectivity (2a-2c, 2h). Electron-donating groups on the aniline ring diminished the enantioselectivity slightly (2d and 2e). Electron-withdrawing groups significantly decreased the reactivity of this ring-opening reaction, although almost perfect enantioselectivity could still be achieved (2f, 2g, and 2m). The bulkiness of ortho substituents on the aniline had significant effects on the reactivity (2j-2m). For instance, with naphthalen-1-amine and 2-methylaniline as the substrates, the reaction became sluggish and either increased reaction temperature or prolonged reaction time was necessary to get satisfactory conversion (2I and 2j). Taking advantage of this ortho-substituent effect, the catalyst was able to differentiate two amino groups with different steric environments. Compounds 2n and 2o were formed as the key products with attractive yields and enantioselectivity. In addition, trifluoromethanesulfonamide was also amenable to the copper-catalyzed ring-opening reaction, and a lower reaction temperature was applied to get satisfactory enantioselectivity (2p). Starting from benzidine, double ring-opening product 2q was formed with 50:1 dr and fantastic selectivity. Benzylamines displayed higher reactivity than anilines, and the reactions of benzylamines were conducted at a slightly lower reaction temperature to gain satisfactory ee values (2r-2x). The structure of the products was confirmed by single-crystal X-ray diffraction analysis of 2v, and the absolute configuration of the major enantiomer was determined to be R(Data S3). The extant chirality in 1-phenylethanamine imparted negligible effects on the yield or selectivity. With optically pure (R)- and (S)-1-phenylethanamine as the amination reagents, compounds 2w and 2x were formed with excellent diastereoselectivities regardless of the existed center chirality, underlining the high catalyst-directed stereocontrol. Finally, to demonstrate the practicability, the copper-catalyzed asymmetric ring-opening reaction was facilely performed on a gram scale of 1a (1.14 g, 2.50 mmol) with p-methylaniline, and product 2a was obtained in 99% yield with 98% ee (see Scheme 3, 2a).

Subsequently, a study was initiated to further explore the substrate scope regarding the diaryliodonium salts (Scheme 4; Figures S19–S97). The reaction smoothly delivered the corresponding products with enantiomeric excesses ranging from 95% to 97% when additional methyl groups were introduced to the *meta* or *para* positions (2y–2cc) (Figures S149–S158 and S232–S241). Pleasingly, the asymmetric ring-opening reaction was applicable to the 2,2'-diethyl-1,1'-biphenyl-derived cyclic iodonium salt, and the reaction proceeded uneventfully with minimal reduction in stereocontrol (2dd and 2ee) (Figures S159–S162 and S242–S245). For non-symmetric cyclic diaryliodoniums, a group adjacent to the iodine atom was introduced to accomplish exclusive regioselectivity (2ff–2nn) (Figures S163–S181 and S246–S263). Furthermore, the reaction was not limited to 1,1'-biphenyl substrates, and compound 2jj



Scheme 4. Substrate Scope with Cyclic Diaryliodoniums

Reaction conditions: cyclic diaryliodonium **1a** (1.0 equiv), amine (1.2 equiv), copper catalyst (5.0 mol %) (**A1** for anilines and **A2** for benzylamines), and Na₂CO₃ (3.0 equiv) in dichloromethane (0.05 mol L^{-1}) at 0°C or indicated otherwise.

bearing a naphthyl-phenyl skeleton was isolated in 97% yield with excellent stereoinduction. Incorporation of a fluorine or chlorine atom, or a methoxyl group adjacent to the iodine atom resulted in a decrease of enantioselectivity, and the corresponding products 2kk-2mm were delivered in 84%–86% yields and synthetically valuable enantioselectivity. High enantiocontrol could still be achieved when the substrate had an electron-withdrawing group at the *meta* position of the iodine atom (compare 2nn with 2gg). Notwithstanding the sluggish reaction rate of the 2,2-biphenol derivative, the high ee value of 200 still showcased the high potential of this reaction to apply in *ortho* oxygenated biaryls (Figures \$182–\$183 and

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Figure 2. Mechanistic Studies

(A) Enantiomeric excess of 2a versus the ratio of Ph-PyBox/Cu(OTf)2.
(B) Correlation between the enantiomeric excesses of 2a and the ee values of Bn-PyBox.
(C) Plot of the initial rate versus the concentration of A1.

S264–S265). The reaction of diaryliodonium with the 1,1'-binaphthalene skeleton also proceeded uneventfully to afford high enantioselectivity (**2pp**) (Figures S184–S185 and S266–S267). However, introducing an *iso*-propyl group at the 6-position resulted in a decrease in enantioselectivity (88% ee) of the corresponding aminoio-dide (**2qq**) (Figures S186–S187 and S268–S269).

To get preliminary mechanistic insights of this copper-catalyzed ring-opening reaction, we examined the effect of the ratio of L7/Cu(OTf)₂ on the enantioselectivity. The enantiomeric excess gradually increased when the ratio of L7/Cu(OTf)₂ was increased before reaching the highest value when L7/Cu(OTf)₂ was 2:1 (Figure 2A). Further increasing the loading of ligand L7 did not improve the ee values. Analysis of the ee values of the product 2a produced by Pybox-Bn ligand L8 with various degrees of reduced enantiomeric excess revealed a linear correlation between the ee values of the product and the ligand (Figure 2B), suggesting that monomeric CuL* (L* = Pybox-Bn) might be the active species. Catalysts containing one copper bearing two bis(oxazolinyl)pyridine ligands (i.e., complex A1), dimeric forms, or higher-order aggregates were not supposed to be active species. These results corroborate the requirement of excess ligand for high enantiocontrol, which allows effective suppression of the dissociation of bis(oxazolinyl)pyridine ligand from the metal center.

Although a precise understanding of the catalytic rationale is beyond reach at this point, a plausible reaction pathway was proposed on the basis of the above studies (Scheme 5). Disproportionation of copper species occurred in a solution of pre-catalyst A1 and amine in dichloromethane. The catalyst A1 is an inactive species, which would quickly dissociate to give A3. Species A3 was postulated as the real catalytically active species in this reaction. On the basis of the results presented in Figure 2B, the dissociation of A1 to A3 was much faster than the rate of the A3-catalyzed ring-opening reaction. The fast dissociation of A1 to A3 was further confirmed by the first-order dependence of the reaction rate on the concentration of catalyst A1, which was measured by monitoring the initial reaction rates at various catalyst loadings (Figure 2C) (Tables S1–S4; Figure S1). In addition, in the presence of amines, the copper complex A3 undergoes ligand displacement to form an amino-copper complex, which, because of the lack of chiral ligand, catalyzes the reaction to give products with no stereoinduction. The independent reactions of Cu(OTf)₂ or CuOTf·0.5PhH in the absence of bis(oxazolinyl)pyridine

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Scheme 5. The Rationale of the Copper-Catalyzed Ring-Opening Reaction

ligand proceeded faster than the reactions with ligands. The amount of amino-copper complex may decrease significantly via coordination-dissociation equilibrium upon increasing the loading of bis(oxazolinyl)pyridine. The interaction between A3 and cyclic iodonium would give A4 and A4', where the two conformers of iodonium moieties should rapidly interconvert. However, A4' is sterically disfavored because of steric repulsion between the benzyl group of bis(oxazolinyl)pyridine and the methyl in cyclic diaryliodonium, which would quickly invert to A4 and undergo a ring-opening reaction to give A5 to establish axial chirality. Subsequently, the coordination of aniline to the Cu(III) center would give intermediate A6. Deprotonation of the amine moiety of A6 followed by reductive elimination delivers the major enantiomer (R)-2a and regenerates the catalyst A3 to complete the catalytic cycle.

Computational calculations on the ring-opening step with A3 as the catalyst were performed (Figures S4–S8). It was found that (1) the two diastereomers of the ring-opening copper complexes have significant energy differences (30.4 kJ/mol for 1b-derived complexes and 18.8 kJ/mol for 1a-derived complexes) (Scheme 6A), and (2) the oxidation addition is a fast step. The reaction of Cu(I)/PyBox with 1a proceeds much faster than the reaction of Cu(I)/PyBox with 1b (3.7 kJ/mol versus 19.3 kJ/mol) (Schemes 6B and 6C), but oxidative addition could not be the rate-determining step. The computational activation energies are consistent with the results reported by Canty and Sanford and their co-workers, who found that the oxidative addition of Cu(I) with [(Mes)PhI]⁺ had a very low activation energy.^{64,65}

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Energies are given in kJ/mol. [Cu] = Cu(I)/Bn-PyBox

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[Cu]

⊕ A4_Me

H₃C

H₃C



 H_3C

H₃C

A5_Me

-51.5

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Scheme 6. Computational Results

(A) Energies for copper complexes.(B) Energy changes of the oxidative addition step with 1b.(C) Energy changes of the oxidative addition step with 1a.

Conclusion

A catalytic enantioselective synthesis of atropisomeric 2'-iodo-[1,1'-biphenyl]-2amines by a highly efficient ring-opening amination reaction of cyclic diaryliodoniums was developed. The relatively low reactivity of cyclic diaryliodoniums was overcome by increasing the distortion of the cyclic substrates. In this reaction, the chiral Cu-bis(oxazolinyl)pyridine complex likely discriminated the two conformers of cyclic diaryliodonium salts to give products with very high enantiomeric excess. A linear correlation between ee values of **2a** and the ee values of Pybox-Bn was observed, indicating that monomeric Cu(Pybox-Bn)⁺ was more likely to be the actual catalyst in this reaction. The protocol offers a general method for the construction of axially chiral biaryliodides under mild conditions, which are expected to be useful precursors for further elaboration to structurally diverse biaryl atropisomers.

EXPERIMENTAL PROCEDURES

Full experimental procedures are provided in the Supplemental Information.

DATA AND SOFTWARE AVAILABILITY

The crystallography data have been deposited at the Cambridge Crystallographic Data Center (CCDC) under accession numbers CCDC: 1565987 (1a), CCDC: 1565988 (1c), and CCDC: 1565989 (2v) and can be obtained free of charge from www.ccdc.cam.ac.uk/getstructures.

SUPPLEMENTAL INFORMATION

Supplemental Information includes Supplemental Experimental Procedures, 269 figures, 4 tables, and 3 data files and can be found with this article online at https://doi.org/10.1016/j.chempr.2018.01.017.

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AUTHOR CONTRIBUTIONS

K.Z. initiated this project. K.Z., L.D., and S.X. performed the experiments and analyzed data for all compounds. Z.G. designed and directed the project. Z.G. wrote the manuscript. J.J. and Y.F. performed the computational calculation.

DECLARATION OF INTERESTS

The authors declare no competing interests.

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