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# A Bulky Chiral N-Heterocyclic Carbene Palladium Catalyst Enables **Highly Enantioselective Suzuki-Miyaura Cross-Coupling Reactions** for the Synthesis of Biaryl Atropisomers

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ABSTRACT: Axially chiral biaryl scaffolds are essential structural units in chemistry. The asymmetric Pd-catalyzed Suzuki-Miyaura cross-coupling reaction has been widely recognized as one of the most practical methods for constructing atropisomers of biaryls. However, longstanding challenges remain in this field. For example, substrate scope is often narrow and specialized, functional groups and heterocycles can lead to reduced reactivity and selectivity, bulky ortho-substituents are usually needed, and reported methods are generally inapplicable to tetra-ortho-substituted biaryls. We have developed an unprecedented highly enantioselective N-heterocyclic carbene (NHC)-Pd catalyzed Suzuki-Miyaura cross-coupling reaction for the synthesis of atropisomeric biaryls. These reactions enable efficient coupling of aryl halides (Br, Cl) or aryl triflates with various types of aryl boron compounds (B(OH)<sub>2</sub>, Bpin, Bneo, BF<sub>3</sub>K), tolerate a remarkably broad scope of functional groups and heterocycles (>41 examples), employ low loading of catalyst (0.2-2 mol%), and proceed under mild conditions. The protocol provided general and efficient access to various atropisomeric biaryls and heterobiaryls in excellent enantioselectivities (up to >99% ee) with no need of using bulky ortho-substituted substrates and was effective for the synthesis of tetra-ortho-substituent biaryls. Moreover, the method was successfully applied to the diastereo- and enantioselective synthesis of atropisomeric ternaphthalenes. Critical to the success of the reaction is the development and application of an extremely bulky  $C_2$ -symmetric chiral NHC, (*R*,*R*,*R*)-DTB-SIPE, as the ligand for palladium. To the best of our knowledge, this is the first highly enantioselective (>90% ee) example of a chiral NHC-metal catalyzed C(sp<sup>2</sup>)-C(sp<sup>2</sup>) cross-coupling reaction.

#### **INTRODUCTION**

The axially chiral biaryl skeletons represent key structural elements found in a wide variety of biologically active natural products and drugs.<sup>1,2</sup> These include gossypol, korupensamine A, and BI-224436, a quinoline-based biaryl serves as the allosteric HIV-1 integrase inhibitor.<sup>3</sup> Moreover, atropisomeric biaryls are widely present in many privileged ligands and catalysts such as BINOL, BINAP, and QUINAP, and have shown remarkable ability of stereoinduction in tremendous asymmetric reactions catalyzed by organometallics and organocatalysts (Figure 1).4



Figure 1. Examples of natural products and ligands bearing axially chiral biaryls.

Accordingly, extensive effort has been devoted to the synthesis of the axially chiral biaryl fragments in the past decades.<sup>5,6</sup> Among these reported methods, the Pd-catalyzed Suzuki-Miyaura cross-coupling reaction<sup>7,8</sup> has been recognized as one of the most practical methods for constructing the atropisomers because of the stability to moisture and oxygen and easy availability of both organo-boron and aryl halide Suzuki-Miyau ACS Paragon Plus Environment

coupling partners. Although Suzuki-Miyaura coupling is one of the most frequently utilized reactions in modern medicinal chemistry, the atroposelective variant of this Nobel prizewinning reaction has not been well-developed. In 2000, the pioneering studies on Pd-catalyzed asymmetric Suzuki-Miyaura reactions were disclosed by the groups of Buchwald<sup>8a</sup> and Cammidge<sup>8c</sup> using KenPhos and a chiral ferrocene ligand, respectively (Scheme 1A). Since then, important contributions in this realm have been made from the group of Fernandez and Lassaletta,<sup>8e-f</sup> Uozumi,<sup>8g,u</sup> Suginome,<sup>8h</sup> Senanayake,<sup>8s</sup> Tang,<sup>8i-j</sup> Lin,<sup>8k</sup> and others,<sup>8m-t</sup> through the development of chiral ligands including bishydrazones, a resin-supported phosphine, a helically chiral polymeric phosphine, biaryl monophosphines, and dienes. However, despite recent achievements, longstanding challenges and significant limitations remain in the field of enantioselective metal-catalyzed  $C(sp^2)-C(sp^2)$ cross-couplings (Scheme 1B). First, the general construction of heterobiaryls represents an unmet challenge. The use of heterocyclic substrates poses additional challenges due to problems associated with their strong coordination ability,9 weak reactivity or stability of reagents, and the reduced configurational stability of the products.<sup>6a-d,r</sup> Second, a general catalyst that allows the synthesis of atropisomeric biaryls with a wide array of functional groups and substitution patterns is highly desirable and remains unknown. Third, the preparation of tetra-ortho-substituted biaryls through an asymmetric Suzuki-Miyaura reaction is still challenging.<sup>8s</sup> Fourth, the high

enantioselectivity outcomes commonly relied on the use of substrates possessing a bulky *ortho*-substituent,<sup>8</sup> which led to limited variations of the *ortho*-substituents. Furthermore, reactions that performed with low catalyst loading under mild conditions are still rare. Finally, we noted that only a limited range of chiral phosphine ligands provide the products with good enantioselectivity, and limits the further development of enantioselective coupling reactions.

#### Scheme 1. Pd-Catalyzed Asymmetric Suzuki-Miyaura Cross-Coupling: Synthesis of Atropisomeric Biaryls

A. Representative chiral ligands for efficient Suzuki-Miyaura couplings:



In this context, we felt that the development of new chiral ligand would be a fundamental strategy to address the abovementioned problems. In particular, we envision that the use of a highly electron-donating NHC ligand would form a robust palladium catalyst<sup>10</sup> to facilitate the oxidative addition of aryl halides and to suppress the deactivating coordination of substrates, thus resulting in the tolerance to various heterocycles and functional groups. However, the enantiocontrol of NHC-Pd-catalyzed C(sp2)-C(sp2) crosscoupling is notoriously difficult.8p-r,11 Indeed, the best enantioselectivity so far obtained is 80% ee, despite several decades of effort from the chemical community to develop more selective ligand.<sup>8</sup> In this regard, we recently developed a family of C<sub>2</sub>-symmetric chiral NHC,<sup>12</sup> namely ANIPE and SIPE type ligands (Scheme 1C).<sup>13</sup> We envisage that the modular nature of our NHCs would allow the introduction of bulky and tunable C<sub>2</sub>-symmetric substituents on nitrogen to enhance the level of enantiocontrol. Moreover, the sterically demanding property of these NHCs is expected to accelerate the critical elementary reductive elimination step. As part of our continuing efforts in NHC-metal catalysis, we disclose herein the development of a scalable route to a novel bulky chiral NHC ((R,R,R)-DTB-SIPE) and its successful application to a general, efficient, and highly enantioselective Pd-catalyzed Suzuki-Miyaura reaction for the synthesis of bi- and teraryl atropisomers with remarkably broad scope and tolerance for functional groups and heterocycles.

#### RESULTS AND DISCUSSION

**Reaction Optimization And Ligand Design.** We thus commenced our study by using 1-bromo-2-methoxynaphthalene (1a) and naphthalen-1-ylboronic acid (2a) as model substrates for the synthesis of axially chiral biaryl (3a) in the presence of palladium-NHC pre-catalysts at ambient temperature. At first, the use of our previously reported ANIPE ligand (L1) gave product 3a in nearly quantitative yield and 44% ee (Table 1, entry 1).

#### Table 1. Reaction Optimization<sup>a</sup>

Br OMe + (		B(OH) <sub>2</sub> [Po	(NHC)(n <sup>3</sup> -cin)Cl] (0.5 mmol%) base (2.0 equiv) solvent (0.2 M) 30 °C, 12 h 3a		
Entry	NHC	Solvent	Base	Yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	L1	EtOH	K <sub>2</sub> CO <sub>3</sub>	99	44
2	L2	EtOH	K <sub>2</sub> CO <sub>3</sub>	89	74
3	L3	EtOH	K <sub>2</sub> CO <sub>3</sub>	70	80
4	L4	EtOH	K <sub>2</sub> CO <sub>3</sub>	98	54
5	L5	EtOH	K <sub>2</sub> CO <sub>3</sub>	72	84
6	L6	EtOH	K <sub>2</sub> CO <sub>3</sub>	66	96
7	L7	EtOH	K <sub>2</sub> CO <sub>3</sub>	99	40
8	L6	THF	K <sub>2</sub> CO <sub>3</sub>	<2	nd
9	L6	<sup>c</sup> hexane	K <sub>2</sub> CO <sub>3</sub>	<2	nd
10	L6	toluene	K <sub>2</sub> CO <sub>3</sub>	59	90
11	L6	<sup>i</sup> PrOH	K <sub>2</sub> CO <sub>3</sub>	73	94
12	L6	<sup>t</sup> BuOH	K <sub>2</sub> CO <sub>3</sub>	76	97
13	L6	<sup>t</sup> BuOH	NaOH	<2	nd
14	L6	<sup>t</sup> BuOH	KO <sup>t</sup> Bu	45	97
15	L6	<sup>t</sup> BuOH	K <sub>3</sub> PO <sub>4</sub>	98	96
16	L6	<sup>t</sup> BuOH	КОН	98	97
			igands		



<sup>a</sup>Reactions were performed on a 0.1 mmol scale. <sup>b</sup>Determined by NMR analysis using crude samples. <sup>c</sup>Determined by HPLC analysis with a chiral stationary phase.

Using a bulkier DM-ANIPE ligand (L2) with 3,5-xylyl groups on the sidearm phenyl groups afford **3a** in 74% ee (entry 2).

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Sterically more demanding DTB-ANIPE possessing 3,5-di-tertbutvl phenyl groups (L3) further improved enantioselectivity to 80% (entry 3). Next, we examined our recently disclosed SIPE type ligand. The use of saturated SIPE (L4) having the same N-substituents with L1 gave 3a in quantitative yield and 54% ee (entry 4). Bulkier DM-SIPE ligand (L5) shown dramatically increased enantioselectivity of 84% (entry 5). Based on these observations, we concluded that SIPE type ligands were superior to the corresponding ANIPE type ligands in terms of enantioselectivity. The higher enantioselectivity stems from the incorporation of a bulkier flanking group to the ligands. Therefore, we anticipated that an extremely bulky DTB-SIPE ligand (L6) should be promising to improve the level of enantiocontrol. However, the preparation of L6 turned out to be problematic (Scheme 2). Although we could prepare the corresponding chiral aniline<sup>11j</sup> in a 13.4 gramscale employing very low loading of rhodium catalyst (0.3 mol%), a typical synthetic route to the carbene precursor (an imidazolium salt, L6/HCl) involving a diimine intermediate resulted in very low yield. Eventually, we were able to efficiently synthesize L6/HCl in gram-scale through a newly designed route using a bis-oxalamide intermediate.<sup>14</sup> As we expected, the use of L6 did significantly improve the enantioselectivity to 96% ee, although the yield was moderate due to the formation of hydrodebromination byproduct (entry 6). Unsurprisingly, the less bulky unsaturated IPE ligand (L7) gave 3a in very low enantioselectivity of 40% (entry 7). Importantly, a survey of solvents revealed that 'BuOH was a superior solvent, as the hydrodebromination side reaction was completely suppressed in these cases (entries 8-12). Finally, through extensive screening of base effects, KOH was identified as the most effective base affording product 3a in 97% ee and 98% yield (entries 13-16).





**Reaction Scope.** With the optimized reaction conditions in hand, we first examined the scope of this asymmetric cross-coupling using simple bromoarene substrates. As shown in Table 2, a wide variety of biaryl atropisomers were obtained with high yields and excellent enantioselectivities (**3a-3r**, 85-99% ee). Notably, large coordination groups at *ortho*-position of haloarenes such as phosphonate<sup>8a,h,m-o</sup>, amide<sup>8b</sup>, or carbonyl-

benzooxazolidinone<sup>8i-j</sup> previously used for optimal stereoinduction are not necessary for our protocol. For example, the chiral ortho-hydroxyl biaryls, important structures commonly existed in numerous natural products, were obtained in high enantioselectivities without the assistance of a special bulky protection<sup>8j</sup>, but using simple ether protecting groups (**3c**, 3d) or non-protected substrates (3e-3h) even with lower loading of catalyst (0.5 mol%). Remarkably, these mild conditions tolerated a wide array of functional groups, including ethers (3c), hemiacetals (3d), esters (3h), amides (3i), aldehydes (3j), free anilines (3k), free phenols (3e-3h), ketones (3p), a trifluoromethyl (31), nitro (3n), cyano (3o) group and fluoride (3m). In addition to aryl bromides, aryl chlorides (3v) and triflates (3q) were competent coupling partners. For the arylboron component, various bench-stable and commonly used organoboron sources, including the boronic acids, pinacol and neopentylglycol boronate esters (Bpin, Bneo), as well as the potassium organotrifluoroborates, all delivered products in excellent enantioselectivities and chemical yields (3r). These expansions further highlighted the generality of the current method.

Encouraged by the above outcomes, we next surveyed the scope of heteroaryl substrates. Due to their strong coordination to metals and lower configurational stability of products, it's much more challenging to achieve high levels of enantiocontrol. Quinolines, core structures in bioactive molecules and ligands, were chosen for evaluation. To our delight, high to excellent levels of enantiocontrol were obtained when we used bromo-quinoline and -isoquinoline substrates with nitrogen atoms at all possible positions (3s-3y, 87-95%) ee). Moreover, bromoguinoline substrates were effective for coupling with several different arylboronic acids (3z-4b). Aside from quinoline and isoquinolines, other electron-deficient heterocycles such as pyridines (4c) and quinazolines (4d), and electron-rich heterocycles including carbazole (4e), indole (4f), and benzothiophene (4g), were all compatible, affording heterobiaryls in high yields and enantioselectivities. It bears mentioning that challenging heterobiaryls with lower rotation energy barriers, for example, 3y and 4g with less bulky 8quinoline and benzothiophene fragments, respectively, could be generated in high enantioselectivities by conducting the reaction at ambient temperature. Interestingly, a heteroaryl bromide could be coupled with a heteroaryl boronate in excellent yield and enantioselectivity (4h). Importantly, this asymmetric coupling method could apply to the synthesis of challenging tetra-ortho-substituted biaryls from both aryl and heteroaryl bromides, furnishing products in excellent enantioselectivities (4i-4l). Furthermore, a gram-scale reaction (4 mmol) was successfully performed in the presence of very low loading of catalyst (0.2 mol%) to deliver the product (**3p**) in high yield and enantioselectivity as before, highlighting the practicality of this method. Finally, the absolute configuration of 3d was determined to be the (R)-form by X-ray crystallographic diffraction analysis.

**Synthesis Of Axially Chiral Ternaphthalenes.** To further demonstrate the utility of our protocol, we applied it to the synthesis of atropisomeric ternaphthalenes (Scheme 3). Ternaphthalenes and oligonaphthalenes, which are higher homologs of binaphthalene, have played essential roles in supramolecular chemistry and materials science.<sup>15</sup> However, the asymmetric catalytic synthesis of these homologs has rarely been reported, presumably due to the difficulty in control of both diastereoselectivity and enantioselectivity.<sup>16</sup> To our

delight, the subjection of 1,4-dibromonaphthalene (5) and boronic acids (6 or 7) to our coupling conditions smoothly delivered the optically active ternaphthalenes (8 or 9) in

excellent enantioselectivity (>99% ee) and in high to excellent ratio with their meso-isomers (8' or 9').



<sup>a</sup>Yields of isolated products on a 0.2 mmol scale. <sup>b</sup>40 °C. <sup>c</sup>30 °C. <sup>d</sup>0.5 mol% catalyst, KO'Bu (1.3 equiv), toluene/H<sub>2</sub>O (9:1), rt, 24 h. <sup>e</sup>2 mol% catalyst, Cs<sub>2</sub>CO<sub>3</sub> (2.5 equiv), 'BuOH/H<sub>2</sub>O (9:1), 60 °C, 24 h. <sup>f</sup>2 mol% catalyst. <sup>g</sup>0.5 mol% catalyst.

#### Scheme 3. Synthesis of Axially Chiral Ternaphthalenes

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Proposed Stereoinduction Models. To get insight into the origin of excellent stereoinduction of the newly designed ligand, we obtained the X-ray crystal structure of  $Pd(L6)(\eta^3-cin)Cl$ (Figure 2a).<sup>17</sup> The steric map of this complex shows a pronounced C<sub>2</sub>-symmetric binding pocket with two accessible guadrants.<sup>18</sup> Based on the X-ray crystal structure, we proposed stereoinduction models for the oxidative addition and transmetalation intermediates of L6-Pd complex, as shown in Figure 2b. Notably, due to the steric repulsions between coupling partners and the bulky 3,5-di-tert-butyl phenyl group on the ligand, the reductive elimination would proceed through the favored transition state  $(TM_1)$ , in which the two naphthyl substituents occupy the two vacant quadrants provided by L6-Pd complex, thereby furnishing biaryls 3d in the (R)configuration. These results are consistent with our initial hypothesis that the bulkier C2-symmetric N-substituents on NHCs would lead to a more efficient enantiodiscrimination and facilitate the key reductive elimination step.



b) Proposed stereoinduction models



**Figure 2.** a) X-ray crystal structure and corresponding steric map of  $Pd(L6)(\eta^3-cin)Cl$ . b) Oxidative addition (OA<sub>1-2</sub>) and transmetalation (TM<sub>1-4</sub>) intermediates to binaphthyls (**3d**).

#### CONCLUSIONS

In conclusion, we have developed the first highly enantioselective NHC-Pd catalyzed (Suzuki-Miyaura) C(sp<sup>2</sup>)-C(sp<sup>2</sup>) cross-coupling reaction for the synthesis of atropisomeric biaryls. A diverse variety of axially chiral biaryls, heterobiaryls. tetra-ortho-substituted biaryls, and ternaphthalenes, were efficiently prepared in high yields with excellent levels of enantiocontrol from readily available and stable substrates. These reactions tolerate a remarkable scope of heterocycles and functional groups, employ low catalyst loading, and proceed under mild conditions. Key to the success of the reaction was the development and application of a very bulky C2-symmetric chiral NHC for the Pd catalyst. Efforts to further explore this NHC-metal catalysis are underway in our laboratory.

#### ASSOCIATED CONTENT

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures, spectroscopic data, and NMR spectra of all products (PDF)

Crystallographic data for 3d (CCDC 1921079) (CIF)

Crystallographic data for Pd(L4)( $\eta^3$ -cin)Cl (CCDC 1921080) (CIF)

Crystallographic data for Pd(L6)( $\eta^3$ -cin)Cl (CCDC 1921081) (CIF)

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Notes

The authors declare the following financial interests: A patent (WO2019096209) for some of the ligands (including L6) in this Communication has been filed.

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