

Improved Total Synthesis of Tubulysins and Design, Synthesis, and **Biological Evaluation of New Tubulysins with Highly Potent** Cytotoxicities against Cancer Cells as Potential Payloads for Antibody–Drug Conjugates

K. C. Nicolaou,^{*,†} Rohan D. Erande,[†] Jun Yin,[†] Dionisios Vourloumis,^{†,‡} Monette Aujay,[#] Joseph Sandoval,[#] Stefan Munneke,[#] and Julia Gavrilyuk[#]

[†]Department of Chemistry, BioScience Research Collaborative, Rice University, 6100 Main Street, Houston, Texas 77005, United States

[‡]Laboratory of Chemical Biology of Natural Products & Designed Molecules, N.C.S.R "Demokritos", 153 10 Agia Paraskevi, Athens, Greece

[#]AbbVie Stemcentrx, LLC, 450 East Jamie Court, South San Francisco, California 94080, United States

S Supporting Information

ABSTRACT: Improved, streamlined total syntheses of natural tubulysins such as V (Tb45) and U (Tb46) and pretubulysin D (PTb-D43), and their application to the synthesis of designed tubulysin analogues (Tb44, PTb-D42, PTb-D47-PTb-D49, and Tb50-Tb120), are described. Cytotoxicity evaluation of the synthesized compounds against certain cancer cell lines revealed a number of novel analogues with exceptional potencies [e.g., **Tb111**: $IC_{50} = 40 \text{ pM}$ against MES SA (uterine sarcoma) cell line; $IC_{50} = 6$ pM against HEK 293T (human embryonic kidney cancer) cell line; and $IC_{50} =$ 1.54 nM against MES SA DX (MES SA with marked multidrug resistance) cell line]. These studies led to a set of valuable structure-activity relationships that provide guidance to



further molecular design, synthesis, and biological evaluation studies. The extremely potent cytotoxic compounds discovered in these investigations are highly desirable as potential payloads for antibody-drug conjugates and other drug delivery systems for personalized targeted cancer chemotherapies.

1. INTRODUCTION

The tubulysins are among the most potent cytotoxic compounds ever discovered from Nature. $^{1-3}$ Their mechanism of action involves depolymerization of microtubules with disintegration of the cytoskeleton as a consequence.⁴⁻⁶ Isolated from the myxobacteria Archangium gephyra and Angiococcus disciformis,^{7,8} these natural products elicited intense research efforts directed toward their total synthesis, analogue design and synthesis, and biological investigations as part of anticancer drug discovery and development programs.⁹⁻¹⁴ Thus, total syntheses of the naturally occurring tubulysins A,¹⁵ B,¹⁵ C,¹⁶ D,^{16–18} G, I, U (**Tb46**, Figure 1),^{16,19–22} and V (**Tb45**, Figure 1)^{16,20-24} and pretubulysin D (**PTb-D43**, Figure 1),^{8a,11} as well as numerous analogues, have been accomplished.²⁵⁻⁴⁹ From the latter, N^{14} -desacetoxytubulysin H (Tb1, Figure 1) is distinguished for its methyl, instead of the acyl methyl, substituent on N14 of tubulysins $A-I^{11,15-18,25}$ and its high potency.^{6,25} We have recently published a total synthesis of N^{14} -desacetoxytubulysin H (Tb1, Figure 1) and several of its analogues (e.g., Tb32, Figure 1) and their biological

evaluation.¹¹ In this Article we report on a more extensive study that included (a) streamlined total syntheses of the natural tubulysins V (Tb45, Figure 1) and U (Tb46, Figure 1) and pretubulysin D (PTb-D43, Figure 1); (b) design and synthesis of numerous novel tubulysin analogues (i.e., PTb-D42, Tb44, PTb-D47-PTb-D49, and Tb50-Tb120, Figure 2); and (c) biological evaluation of the synthesized compounds. These investigations led to the discovery of a number of exceptionally potent antitumor agents particularly suited as payloads for antibody-drug conjugates (ADCs)^{50,51} and other delivery systems.52

2. RESULTS AND DISCUSSION

Having developed our C-H activation-based strategy for the synthesis of the tubuvaline residue,¹¹ and in order to devise a practical synthesis of tubulysins V (Tb45) and U (Tb46), pretubulysin D (PTb-D43), and their analogues, we decided to

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Figure 1. Molecular structures of naturally occurring tubulysins V (Tb45) and U (Tb46), pretubulysin D (PTb-D43), N¹⁴-desacetoxy-tubulysin H (Tb1), and previously synthesized potent tubulysin analogue (Tb32).

improve and apply our synthetic technologies to that end. In pursuing new tubulysin analogues, we applied rational ligand design based on our previously developed preliminary structure–activity relationships (SARs)¹¹ and the recently reported X-ray crystallographic analysis regarding the binding requirements of tubulysin-like molecules to microtubules, their biological target.⁵³

2.1. Total Synthesis of Tubulysin V (Tb45), Tubulysin U (Tb46) and Its Methyl Ester (Tb44). Our newly developed, streamlined total synthesis of the naturally occurring tubulysins V (Tb45) and U (Tb46), and the methyl ester of the latter (Tb44), proceeded along an appropriately modified and improved synthetic route as shown in Scheme 1.¹¹ Thus, aldehyde $1^{11,54}$ was subjected to C–H activation coupling with thiazole derivative 2 [PhI(OCOCF₃)₂, TMSN₃],^{11,55} furnishing coupling product 3 in 56% yield. Stereoselective reduction of the thiazolyl ketone moiety within 3 with (S)-CBS catalyst in the presence of $BH_3 \cdot Me_2 S^{11,56}$ produced alcohol 4 in 83% yield and as a single diastereoisomer after chromatographic purification. Elaboration of intermediate 4 to acetoxy carboxylic acid 5 was achieved through a sequence involving deacetylation (K₂CO₃, MeOH), two-step selective oxidation of the so generated primary alcohol (TEMPO, BAIB; then NaClO₂), and acetylation of the secondary alcohol (Ac₂O, Et₃N) in 78% overall yield. Coupling of carboxylic acid 5 with ammonium salt 6^{11} in the presence of HATU and Et₃N led to amide 7 (94%) yield). Removal of the Boc group from the latter through the action of TFA, followed by coupling of the resulting amine with carboxylic acid 8,¹¹ produced peptide 9 (HATU, Et₃N, 92% yield) as shown in Scheme 1. Cleavage of the Boc protecting group from 9 (TFA) and coupling of the resulting amine with N-methyl-D-pipecolic acid (10) afforded tubulysin U methyl ester (Tb44, 85% overall yield). Conversion of Tb44 to tubulysin U (Tb46) via tubulysin V (Tb45) required sequential

treatment with $Me_3SnOH^{11,S7}$ (cleavage of both methyl ester and acetate moieties, 68% yield), and re-acetylation of the resulting hydroxy carboxylic acid (Ac₂O, pyridine, 79% yield) as shown in Scheme 1.

2.2. Improved Total Synthesis of Pretubulysin D (PTb-D43) and Its Methyl Ester (PTb-D42). In an effort to streamline our original synthesis of pretubulysin D (PTb-D43), and since the adjacent to the thiazole carbonyl moiety was not needed in this case [cf. intermediates 3 (Scheme 1) and 14 (Scheme 2)], we decided to employ the commercially available valine derivative 11 as the starting material. Thus, and as shown in Scheme 2, exposure of 11 to TMSCHN₂ followed by LiAlH₄ reduction of the resulting methyl ester furnished the corresponding primary alcohol, whose reaction with CBr₄ and PPh₃ led to bromide **12** in 62% overall yield for the three steps. Coupling of the anion generated from thiazole 13 (see Supporting Information for preparation), through the action of *n*-BuLi, with bromide 12 furnished 14 in 78% yield.⁵⁸ Transformation of TBS-ether 14 to the desired carboxylic acid (15) was achieved through desilylation (TBAF) followed by two-step oxidation of the resulting alcohol (DMP; then NaClO₂), in 78% overall yield. Coupling of carboxylic acid 15 with aminoester 6^{11} in the presence of HATU and Et₂N led to amide **16** (82% yield). Removal of the Boc protecting group from the latter (TFA) followed by coupling of the resulting amine with acid fluoride 17¹¹ furnished peptide 18 (i-Pr₂NEt, 95% overall yield for the two steps) as shown in Scheme 2. Cleavage of the Fmoc-group from 18 under basic conditions $[N(CH_2CH_2NH_2)_3]$ and coupling of the so formed amine with *N*-methyl-D-pipecolic acid $(10)^{11}$ provided pretubulysin D precursor PTb-D42 in 72% overall yield. Conversion of PTb-D42 to pretubulysin D (PTb-D43) was accomplished using the previously reported conditions (LiOH, 91% yield)¹¹ presented in Scheme 2.

2.3. Synthesis of Pretubulysin D Analogues PTb-D47, PTb-D48, and PTb-D49. Scheme 3 summarizes the synthesis of pretubulysin analogues PTb-D47, PTb-D48, and PTb-D49 from known intermediates 18¹¹ and 16,¹¹ respectively. Thus, removal of the Fmoc protecting group from 18 $[N(CH_2CH_2NH_2)_3]$ and coupling of the resulting amine with carboxylic acid 19 (for its synthesis see Supporting Information) provided pretubulysin analogue PTb-D47 in 82% overall yield, as shown in Scheme 3A. Removal of the Boc group from 16 (TFA) followed by coupling of the resulting amine with acid fluoride 20^{11} gave peptide 21 (*i*-Pr₂NEt, 95% overall yield) as shown in Scheme 3B. Cleavage of the Fmoc group from 21 through the action of $N(CH_2CH_2NH_2)_3$ and coupling of the resulting amine with either N-methyl-Dpipecolic acid $(10)^{11}$ or *n*-butyl-substituted pipecolic acid 19 provided pretubulysin D analogues PTb-D48 and PTb-D49 in 81% and 76% overall yields, respectively.

2.4. Synthesis of N^{14} -Desacetoxytubulysin H Analogues Tb50–Tb120. Given that N^{14} -methyl-substituted tubulysins (such as N^{14} -desacetoxytubulysin H, $N^{11,25}$ Tb1) have been proven more potent than their N^{14} -H and N^{14} -acetoxymethyltubulysin (such as tubulysin H) counterparts, we focused considerable efforts on designing and synthesizing a number of N^{14} -methyl-substituted tubulysins. Scheme 4 summarizes the synthesis of N^{14} -methyl-substituted tubulysins Tb50 and Tb51, in which the pipecolic acid residue of the molecule is replaced with N-Me-substituted pyrrole and imidazole structural motifs, respectively. Thus, cleavage of the Fmoc protecting group from previously synthesized inter-



Figure 2. continued

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Figure 2. Molecular structures of synthesized naturally occurring tubulysins [pretubulysin D (PTb-D43), tubulysins U (Tb46) and V (Tb45)] and synthesized designed tubulysin analogues (Tb44, PTb-D42, PTb-D47, PTb-D48, PTb-D49, and Tb50–Tb120).

mediate **22**¹¹ through the action of *N*,*N*-bis(2-aminoethyl)-1,2ethanediamine followed by coupling of the so generated amine with 1-methyl-1*H*-pyrrole-2-carboxylic acid (**23**) and 1-methyl-1*H*-imidazole-2-carboxylic acid (**24**) provided N^{14} -desacetoxytubulysin analogues **Tb50** and **Tb51**, in 74% and 76% yields, respectively, as summarized in Scheme 4.

Tubulysin analogues Tb52–Tb55, in which changes in the two end structural motifs were made while keeping the proven to be desirable N^{14} -Me and the *i*-Pr moieties on the isoleucine residue, were synthesized as shown in Scheme 5. Thus, coupling of carboxylic acid 25^{11} with commercially available ammonium salt 26 in the presence of HATU furnished

dipeptide 27 (84% yield). Exposure of this protected dipeptide to TFA resulted in removal of the Boc group to afford the corresponding amine, whose coupling with acid fluoride 20 in the presence of *i*-Pr₂NEt in DMF led to the formation of tripeptide 28 (92% overall yield). Removal of the Fmoc group from 28 [N(CH₂CH₂NH₂)₃], followed by coupling of the resulting amine with *N*-methyl-D-pipecolic acid (10) and *n*butyl-substituted pipecolic acid 19 (see Supporting Information for preparation) under HATU conditions, resulted in the formation of tubulysin analogues Tb52 (72% yield) and Tb54 (77% yield), respectively, as shown in Scheme 5 (see Supporting Information for further details). Finally, the Scheme 1. Syntheses of Tubulysin U Methyl Ester Tb44 and Tubulysins V (Tb45) and U (Tb46)^a



^aReagents and conditions: (a) 1 (2.0 equiv), 2 (1.0 equiv), TMSN₃ (1.5 equiv), PIFA (1.5 equiv), benzene, 23 °C, 12 h; then 1 (2.0 equiv), TMSN₃ (1.5 equiv), PIFA (1.5 equiv), 23 °C, 12 h, 56%; (b) (S)-CBS (0.2 equiv; 1.0 M in toluene), $BH_3 \cdot Me_2S$ (5.0 equiv; 2.0 M in THF), $0 \rightarrow 23 \,^{\circ}$ C, 18 h, 83%; (c) K₂CO₃ (4.0 equiv), MeOH, 23 $^{\circ}$ C, 3 h, 95%; (d) TEMPO (0.1 equiv), BAIB (1.0 equiv), CH₂Cl₂, 23 °C, 16 h, 98%; (e) NaClO₂ (5.4 equiv), NaH₂PO₄·H₂O (12.5 equiv), 2methyl-2-butene (7.5 equiv), t-BuOH, THF, H₂O, 23 °C, 12 h; (f) Ac₂O (3.0 equiv), Et₃N (3.0 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 15 h, 78% for the two steps; (g) 6 (2.0 equiv), HATU (3.2 equiv), Et_3N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 14 h, 94%; (h) TFA (1.75 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 6 h; (i) 8 (2.0 equiv), HATU (4.0 equiv), Et₃N (10 equiv), DMF, $0 \rightarrow 23$ °C, 12 h, 92% for the two steps; (j) TFA (1.75 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 6 h; (k) 10 (2.2 equiv), HATU (3.2 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 85% for the two steps; (1) Me₃SnOH (10 equiv), CH₂Cl₂, reflux, 12 h, 68%; (m) Ac₂O (7.5 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 79%. TMS = trimethylsilyl; PIFA = phenyliodine(III)bis(trifluoroacetate); (S)-CBS = (3aS)-tetrahydro-1methyl-3,3-diphenyl-1H,3H-pyrrolo[1,2-c][1,3,2]oxazaborole; TEMPO = (2,2,6,6-tetramethylpiperidin-1-yl)oxyl; BAIB = bis-(acetoxy)iodo benzene; Ac = acetyl; py = pyridine; THF = tetrahydrofuran; HATU = 1-[bis(dimethylamino)methylene]-1H-1,2,3-triazolo[4,5-b]pyridinium 3-oxid hexafluorophosphate; DMF = N,N-dimethylformamide; TFA = trifluoroacetic acid; Boc = tertbutyloxycarbonyl.

Scheme 2. Total Synthesis of Pretubulysin D (PTb-D43) and Its Methyl Ester Precursor PTb-D42 a



^aReagents and conditions: (a) TMSCHN₂ (1.2 equiv; 2.0 M in diethyl ether), toluene:methanol (2:1, ν/ν), 0 \rightarrow 23 °C, 0.5 h, 74%; (b) LiAlH₄ (2.0 equiv; 2.0 M in THF), THF, 0 °C, 1 h, 98%; (c) CBr₄ (2.0 equiv), PPh₃ (2.0 equiv), benzene, $0 \rightarrow 10 \,^{\circ}\text{C}$, 1 h, 81%; (d) 13 (1.2 equiv), *n*-BuLi (1.2 equiv; 2.5 M in hexanes), THF, $-78 \rightarrow 0$ °C, 2.5 h, 78%; (e) TBAF (2.0 equiv; 1.0 M in THF), THF, 0 °C, 1 h, 94%; (f) DMP (1.5 equiv), CH₂Cl₂, 23 °C, 15 min, 90%; (g) NaClO₂ (5.4 equiv), NaH₂PO₄·H₂O (12.5 equiv), 2-methyl-2-butene (7.5 equiv), t-BuOH, THF, H₂O, 23 °C, 1 h; 92%; (h) 6 (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 82%; (i) TFA (45 equiv), CH₂Cl₂, 23 °C, 2 h; (j) 17 (4.1 equiv), *i*-Pr₂NEt (6.2 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 95% for the two steps; (k) $N(CH_2CH_2NH_2)_3$ (16 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^{\circ}C$, 2 h; (l) 10 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 72% for the two steps; (m) LiOH·H₂O (5.0 equiv), THF, H₂O, 23 °C, 24 h, 91%. TBAF = tetra-n-butylammonium fluoride; DMP = Dess-Martin periodinane; Fmoc = fluorenylmethyloxycarbonyl.

corresponding methyl esters were converted to their carboxylic acid counterparts **Tb53** and **Tb55**, respectively, through the sequential action of Me₃SnOH^{11,57} (cleavage of methyl ester and acetate moieties) and Ac₂O/pyridine (re-acetylation of hydroxy group) in 68% and 74% overall yield, respectively, as presented in Scheme 5.

The syntheses of tubulysin analogues **Tb56** and **Tb57**, where the "right end" (Tup) and the "left end" (Mep) amino acid residues of N^{14} -desacetoxy tubulysin **Tb1**¹¹ were replaced with structural motifs represented by fragments 33 or 34¹¹ and 19 (for their synthesis see Supporting Information), respectively, are presented in Scheme 6. Thus, removal of the Boc group Scheme 3. Synthesis of Pretubulysin D Analogues PTb-D47, PTb-D48, and PTb-D49 a



⁴⁷Reagents and conditions: (a) $N(CH_2CH_2NH_2)_3$ (16 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 2 h; (b) **19** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 0 \rightarrow 23 °C, 24 h, 82% for the two steps; (c) TFA (40 equiv), CH_2Cl_2 , 23 °C, 2 h; (d) **20** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, 0 \rightarrow 23 °C, 18 h, 95% for the two steps; (e) $N(CH_2CH_2NH_2)_3$ (16 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 2 h; (f) **10** or **19** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 0 \rightarrow 23 °C, 24 h, 81% for the two steps for **PTb-D48** and 76% for the two steps for **PTb-D49**.

from 29¹¹ (TFA) followed by reaction of the resulting amine with acid fluoride 17 in the presence of *i*-Pr₂NEt in DMF led to the formation of dipeptide 30 (75% overall yield). The latter was further treated with N(CH₂CH₂NH₂)₃ to remove the Fmoc group, and the resulting amine was coupled with pipecolic acid derivative 19 (HATU, Et₃N) to furnish tripeptide 31 in 82% overall yield. Tripeptide 31 was then converted to its carboxylic acid counterpart (32) through sequential treatment with Me₃SnOH and Ac₂O as described above for the conversion of Tb52 to Tb53, in 78% overall yield. Finally, coupling of 32 with ammonium salt 33 or 34 under HATU conditions furnished tubulysin analogues Tb56 (71% yield) and Tb57 (76% yield), respectively, as shown in Scheme 6 (see Supporting Information for further details).

Tubulysin analogue Tb59 in which the acetoxy group of Tb2 was replaced with a carbonyl group, was synthesized from the previously reported acetoxy ester analogue $Tb2^{11}$ through hydroxy tubulysin Tb58 as summarized in Scheme 7. Thus, exposure of Tb2 to Me₃SnOH furnished Tb58 in 78% yield.

Scheme 4. Synthesis of N^{14} -Desacetoxytubulysin Analogues Tb50 and Tb51^{*a*}



^aReagents and conditions: (a) $N(CH_2CH_2NH_2)_3$ (16 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) **23** or **24** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 74% for the two steps for **Tb50** and 76% for the two steps for **Tb51**.

The latter was converted to the desired keto acid analogue **Tb59** (81% yield) through the action of DMP, as shown in Scheme 7.

Tubulysin analogues Tb60, Tb61, Tb62, and Tb63, containing a valine instead of an isoleucine residue, were synthesized as summarized in Scheme 8. Specifically, the previously reported analogue Tb32¹¹ was converted to its carboxylic acid counterpart Tb60 through exposure to Me₃SnOH^{11,57} (cleavage of methyl ester and acetate moieties, 70% yield) followed by re-acetylation of the hydroxy acid to Tb61 (Ac₂O/pyridine, 61% yield). Tb60 was converted to its keto acid counterpart Tb62, in 78% yield, by DMP oxidation as shown in Scheme 8. Methyl ester formation from the latter using TMSCHN₂ furnished Tb63 in 71% yield.

Scheme 9 summarizes the synthesis of tubulysin analogues Tb64 and Tb65, in which the thiazole moiety was replaced with a pyridine structural motif (while maintaining all the other structural features of Tb32, one of our most promising tubulysin analogues).¹¹ Their synthesis was initiated with the removal of the Boc group from dipeptide 35^{11} (TFA) followed by coupling of the liberated amine with Fmoc-protected acid fluoride 20^{11} to provide tripeptide 36 (99% yield for the two steps) as shown in Scheme 9. Cleavage of the Fmoc group [N(CH₂CH₂NH₂)₃] from this intermediate followed by coupling of the so generated amine with *N*-methyl-D-pipecolic acid (10) led to tubulysin analogue Tb64 (75% overall yield). Finally, analogue Tb64 was converted to its carboxylic acid counterpart Tb65 through the sequential action of Me₃SnOH^{11,57} and Ac₂O/pyridine in 68% overall yield, as shown in Scheme 9.

Tubulysin analogues Tb66, Tb67, and Tb68, in which the thiazole moiety carries a methyl group, were synthesized from the known and readily available aldehyde $37^{11,54}$ as summarized in Scheme 10. Thus, C–H activation-based coupling of aldehyde 37 with methyl thiazole acetate 38,¹¹ under the previously reported conditions [PhI(OCOCF₃)₂, TMSN₃],^{11,55} provided ketone 39 in 75% yield. Reduction of thiazolyl ketone 39 with (*S*)-CBS catalyst in the presence of BH₃·Me₂S^{11,56} produced alcohol 40 in 72% yield as a single diastereoisomer

Scheme 5. Synthesis of Tubulysin Analogues Tb52-Tb55^a



^aReagents and conditions: (a) **26** (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 23 °C, 18 h, 84%; (b) TFA (40 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 2 h; (c) **20** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 92% for the two steps; (d) N(CH₂CH₂NH₂)₃ (16 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 2 h; (e) **10** (3.0 equiv) or **19** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 72% for the two steps for **Tb52**, 77% for the two steps for **Tb54**; (f) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (g) Ac₂O (6.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 68% for the two steps for **Tb53**, 74% for the two steps for **Tb55**.

after chromatographic purification. The required elaboration of alcohol 40 to acetoxy carboxylic acid 41 was achieved through a sequence involving deacetylation (K₂CO₃, MeOH), selective oxidation of the resulting primary alcohol (TEMPO, BAIB; then NaClO₂), and acetylation (Ac₂O, pyridine) of the remaining secondary alcohol, in 61% overall yield for the four steps. Coupling of carboxylic acid **41** and ammonium salt 6^{11} in the presence of HATU and Et₃N led to amide 42 (88% yield). The Boc protecting group was cleaved from the latter compound (TFA), and the resulting amine was coupled with acid fluoride 20¹¹ (*i*-Pr₂NEt, 91%) to afford peptide 43 as shown in Scheme 10. Removal of the Fmoc group from 43 $[N(CH_2CH_2NH_2)_3]$ followed by coupling of the so generated amine with N-methyl-D-pipecolic acid (10) provided tubulysin analogue Tb66 (65% overall yield). Tubulysin analogue Tb67 was formed from Tb66 through methyl ester hydrolysis (Me_3SnOH) and acetylation $(Ac_2O, pyridine)$ of the resulting Scheme 6. Synthesis of Tubulysin Analogues Tb56 and Tb57 a



^aReagents and conditions: (a) TFA (40 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^\circ$ C, 2 h; (b) 17 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 18 h, 75% for the two steps; (c) N(CH₂CH₂NH₂)₃ (16 equiv), CH₂Cl₂, $0 \rightarrow 23 \ ^\circ$ C, 2 h; (d) 19 (1.5 equiv), HATU (1.5 equiv), Et₃N (3.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 24 h, 82% for the two steps; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (6.0 equiv), pyridine, $0 \rightarrow 23 \ ^\circ$ C, 12 h, 78% for the two steps; (g) 33 or 34 (1.2 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 18 h, 71% for the two steps for Tb56 and 76% for the two steps for Tb57.

hydroxy acid (62% overall yield), as shown in Scheme 10. Keto acid tubulysin analogue **Tb68** was obtained from **Tb66** by treatment with Me₃SnOH followed by oxidation of the resulting hydroxy acid with Dess–Martin periodinane in 64% overall yield (Scheme 10).

Scheme 11 summarizes the synthesis of tubulysin analogues Tb69 (lacking the *N*-Me substituent), Tb70, and Tb71, the latter two containing the *N*-methyl pyrrolidine structural motif as a substitution for the piperidine residue. Thus, advanced intermediate 43 (for preparation see Scheme 10) was converted to its amino counterpart through the action of $N(CH_2CH_2NH_2)_3$, and the latter was coupled with Fmocprotected pipecolic acid 44 and *N*-methyl-D-proline (45) to afford tubulysin analogues Tb69 and Tb70 in 62% and 82% overall yields, respectively. Finally, methyl ester Tb70 was converted to its carboxylic acid counterpart Tb71 through the sequential action of Me₃SnOH^{11,57} and Ac₂O/pyridine in 74% overall yield, as shown in Scheme 11.

Scheme 12 depicts the synthesis of tubulysin analogues Tb72 and Tb73, both of which feature an isopropyl group on the Scheme 7. Synthesis of Tubulysin Analogues Tb58 and Tb59 $\!\!\!^a$



"Reagents and conditions: (a) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h, 78%; (b) DMP (1.5 equiv), CH_2Cl_2 , 23 °C, 30 min, 81%.





"Reagents and conditions: (a) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h, 70%; (b) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 61%; (c) DMP (1.5 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 30 min, 78%; (d) TMSCHN₂ (1.2 equiv; 2.0 M in diethyl ether), toluene:methanol (2:1, ν/ν), $0 \rightarrow 23$ °C, 1 h, 71%.

thiazole structural motif. Thus, commercially available bromothiazole ester derivative **46** was reduced to the corresponding primary alcohol (LiBH₄) and the latter was silylated (TBSCl, imidazole, 86% yield for the two steps) to afford bromothiazole **47**. The lithio derivative generated from bromide **47** and *n*-BuLi was then reacted with Weinreb amide **48** forming ketone **49**, whose asymmetric reduction with (*S*)-CBS catalyst and BH₃·Me₂S gave, stereoselectively, hydroxy compound **50**. The latter was elaborated to acetoxy carboxylic Scheme 9. Synthesis of Tubulysin Analogues Tb64 and Tb65 a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , 23 °C, 2 h; (b) **20** (4.0 equiv), *i*- Pr_2NEt (6.0 equiv), DMF, 0 \rightarrow 23 °C, 18 h, 99% for the two steps; (c) N($CH_2CH_2NH_2$)₃ (15 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 0 \rightarrow 23 °C, 24 h, 75% for the two steps; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, 0 \rightarrow 23 °C, 12 h, 68% for the two steps.

acid **51** through a sequence involving acetylation (Ac₂O, pyridine, 82% yield), desilylation (TBAF, 98% yield), and oxidation (DMP, 89% yield; then NaClO₂, 2-methyl-2-butene, 98% yield). Carboxylic acid **51** was coupled with ammonium salt **6** under HATU conditions furnishing Boc-protected segment **52**, whose deprotection (Boc removal, TFA) and union of the resulting amine with acid fluoride **20** under standard conditions led to fragment **53** in 84% overall yield. Deprotection of the latter [N(CH₂CH₂NH₂)₃] followed by coupling of the resulting amine with *N*-methyl-D-pipecolic acid (**10**) under HATU conditions furnished coveted tubulysin analogue **Tb72** (81% overall yield), and its carboxylic acid counterpart **Tb73** (72% overall yield) upon sequential ester cleavage and re-acetylation under the standard conditions mentioned above and summarized in Scheme **12**.

Tubulysins Tb74 and Tb75 carry oxygenated pipecolic acid residues as well as an isopropyl group on their thiazole moiety, as shown in their structures (see Scheme 13). They were synthesized from advanced intermediate 53 (for preparation, see Scheme 12) as shown in Scheme 13. Thus, Fmoc derivative 53 was deprotected $[N(CH_2CH_2NH_2)_3]$, and the resulting amine was reacted with hydroxy *N*-methyl pipecolic acid 54 under HATU conditions to afford Tb74 in 69% overall yield. Analogue Tb75 was generated from Tb74 by DMP oxidation in 78% yield as shown in Scheme 13.

Retaining the valine moiety instead of the isoleucine residue just like their Tb74 and Tb75 siblings but lacking the isopropyl group on their thiazole ring, tubulysins Tb76 and Tb77 feature oxygenated *N*-methyl pipecolic acid structural motifs and an ethyl, rather than a methyl, ester group at the other end of the molecule. Their synthesis proceeded from *N*-Boc-protected thiazolyl carboxylic acid 25^{11} as summarized in Scheme 14. Scheme 10. Synthesis of Tubulysin Analogues Tb66–Tb68^a



^aReagents and conditions: (a) 37 (2.0 equiv), 38 (1.0 equiv), TMSN₃ (2.0 equiv), PIFA (2.0 equiv), benzene, 23 °C, 16 h; then 37 (2.0 equiv), TMSN₃ (2.0 equiv), PIFA (2.0 equiv), 23 °C, 12 h, 75%; (b) (S)-CBS (0.2 equiv; 1.0 M in toluene), $BH_3 \cdot Me_2S$ (1.0 equiv; 2.0 M in THF), $0 \rightarrow 23 \,^{\circ}$ C, 18 h, 72%; (c) K₂CO₃ (4.0 equiv), MeOH, 23 $^{\circ}$ C, 3 h, 93%; (d) TEMPO (0.1 equiv), BAIB (1.0 equiv), CH₂Cl₂, 23 °C, 16 h, 91%; (e) NaClO₂ (5.4 equiv), NaH₂PO₄·H₂O (12.2 equiv), 2methyl-2-butene (7.5 equiv), t-BuOH, THF, H₂O, 23 °C, 12 h; (f) Ac₂O (3.0 equiv), pyridine (3.0 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 15 h, 72% for the two steps; (g) 6 (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 88%; (h) TFA (45 equiv), CH₂Cl₂, 0 \rightarrow 23 °C, 2 h; (i) **20** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, 0 \rightarrow 23 °C, 18 h, 91% for the two steps; (j) $N(CH_2CH_2NH_2)_3$ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (k) 10 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 65% for the two steps; (1) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (m) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 62% for the two steps; (n) DMP (1.5 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 30 min, 64% for the two steps.

Thus, **25** was coupled to ammonium salt **55** in the presence of HATU and Et_3N to afford dipeptide **56** (81% yield), whose exposure to TFA led to the corresponding amine. Coupling of the latter with acid fluoride **20** was facilitated by *i*-Pr₂NEt

Scheme 11. Synthesis of Tubulysin Analogues Tb69–Tb71^a



^aReagents and conditions: (a) $N(CH_2CH_2NH_2)_3$ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) 44 or 45 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 62% for the two steps for **Tb69** and 82% for the two steps for **Tb70**; (c) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (d) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 74% for the two steps.

leading to tripeptide **57**. Analogue **Tb76** was smoothly generated from **57**, upon liberation of its amino group $[N(CH_2CH_2NH_2)_3]$ and union of the resulting amine substrate with hydroxypipecolic acid **54** under the influence of HATU and Et₃N (96% yield for the two steps). Finally, silylation of the resulting alcohol with TBDMSOTf and 2,6-lutidine furnished analogue **Tb77** in 87% overall yield, as presented in Scheme 14. The latter analogue was meant to test the effect of increased lipophilicity of the TBS-bearing pipecolic acid residue.

Tubulysin analogue **Tb78**, whose novel structural motif is the pentyl spirocycle moiety instead of the isoleucine residue, was synthesized as shown in Scheme 15. Thus, removal of the Boc group from previously reported dipeptide 58^{11} (TFA) and coupling of the liberated amine with Fmoc-protected acid fluoride 59 (see Supporting Information for preparation) under standard conditions provided tripeptide 60 (56% yield for the two steps). Cleavage of the Fmoc group [N(CH₂CH₂NH₂)₃] from the latter followed by coupling of the generated amine with *N*-methyl-D-pipecolic acid (10) led to the targeted tubulysin analogue Tb78 in 69% overall yield.

Scheme 16 summarizes the synthesis of tubulysin analogues Tb79 and Tb80, both of which feature a hexafluoro isopropyl unit⁵⁹ as opposed to their isoleucine residue. Their synthesis began with removal of the Boc group from the previously reported dipeptide 58^{11} and proceeded with coupling of the liberated amine with Fmoc-protected acid fluoride 61 (prepared from its amino acid counterpart by sequential exposure to FmocCl and DAST; see Supporting Information for details) followed by cleavage of the Fmoc group [N(CH₂CH₂NH₂)₃] to afford amine 62 (27% yield for the three steps) as shown in Scheme 16 (see Supporting Information for further details). Coupling of the so generated

Scheme 12. Synthesis of Tubulysin Analogues Tb72 and Tb73 $\!\!\!\!^a$



^aReagents and conditions: (a) LiBH₄ (1.53 equiv; 2 M in THF), MeOH (1.55 equiv), THF, $0 \rightarrow 23$ °C, 12 h; (b) TBSCl (1.23 equiv), imidazole (1.23 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 0.5 h, 86% for the two steps; (c) n-BuLi (1.44 equiv; 2.5 M in hexanes), 48 (1.0 equiv), THF, $-78 \rightarrow -50$ °C, 3 h, 71%; (d) (S)-CBS (0.1 equiv, 1.0 M in toluene), $BH_2 \cdot Me_2S$ (1.0 equiv; 2.0 M in THF), $0 \rightarrow 23$ °C, 36 h, 74%; (e) Ac₂O (3.0 equiv), Et₃N (4.0 equiv), $0 \rightarrow 23$ °C, 2 h, 82%; (f) TBAF (2.0 equiv; 1 M in THF), THF, $0 \rightarrow 23$ °C, 30 min, 98%; (g) DMP (1.5 equiv), CH₂Cl₂, 23 °C, 1 h, 89%; (h) NaClO₂ (5.4 equiv), NaH₂PO₄·H₂O (12.2 equiv), 2-methyl-2-butene (7.5 equiv), t-BuOH, THF, H₂O, 23 °C, 1 h, 98%; (i) 6 (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 99%; (j) TFA (45 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 2 h; (k) 20 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 84% for the two steps; (1) N(CH₂CH₂NH₂)₃ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^{\circ}C$, 2 h; (m) 10 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 81% for the two steps; (n) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (o) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 72% for the two steps.

amine 62 with *N*-methyl-D-pipecolic acid (10) resulted in the formation of tubulysin analogue Tb79 (87% yield). Finally, time controlled exposure of Tb79 to $Me_3SnOH^{11,57}$ (5 h; cleavage of acetate only) furnished analogue Tb80 in 87% yield, as shown in Scheme 16.

Scheme 13. Synthesis of Tubulysin Analogues Tb74 and Tb75 a



^aReagents and conditions: (a) $N(CH_2CH_2NH_2)_3$ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) 54 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 69% for the two steps; (c) DMP (1.5 equiv), CH_2Cl_2 , 23 °C, 30 min, 78%.

Scheme 17 summarizes the synthesis of tubulysin analogues Tb81, Tb82, Tb83, and Tb84, which incorporate a trifluoro ethyl moiety,⁵⁹ instead of the isoleucine residue. Their synthesis began with removal of the Boc group from the previously reported dipeptide 58¹¹, and coupling of the liberated amine with Fmoc-protected acid fluoride 63 (prepared from its amino acid counterpart by sequential exposure to FmocCl, and DAST; see Supporting Information for details) to provide tripeptide 64 (70% yield for the two steps), as shown in Scheme 17 (see Supporting Information for further details). Cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from this intermediate afforded free amine 65 (82% yield), which was coupled with either N-methyl-D-pipecolic acid (10) or N-methyl-D-proline (45) to give tubulysin analogues Tb81 (79% yield) or Tb82 (66% yield), respectively. Finally, exposure of Tb81 and Tb82 to Me₃SnOH^{11,57} furnished analogues Tb83 (82% yield) and Tb84 (79% yield), respectively, as shown in Scheme 17.

Scheme 18 summarizes the synthesis of tubulysin analogues Tb85, Tb86, and Tb87, which incorporate an (R)-hexafluoro isopropyl moiety⁵⁹ instead of the (S)-isoleucine residue found in many of the other designed analogues. Their synthesis began with removal of the Boc group (TFA) from the previously reported dipeptide 58^{11} followed by coupling of the so obtained amine with Fmoc-protected acid fluoride 66 (prepared from its amino acid counterpart by sequential exposure to FmocCl and DAST) to provide tripeptide 67, upon cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ (38% yield for the three steps), as shown in Scheme 18 (see Supporting Information for further details). Coupling of the latter with either N-methyl-D-pipecolic acid (10) or N-methyl-D-proline (45) under HATU conditions led to tubulysin analogues Tb85 (89% yield) and Tb86 (88% yield), respectively. Sequential treatment of Tb85 with Me₃SnOH^{11,57} and Ac₂O/pyridine then gave analogue Tb87 in 89% overall yield as shown in Scheme 18.

Tubulysin analogue **Tb88**, lacking isoleucin's side chain, was constructed as shown in Scheme 19. Thus, removal of the Boc group from dipeptide **56** (for preparation, see Scheme 14) with Scheme 14. Synthesis of Tubulysin Analogues Tb76 and Tb77 a



^aReagents and conditions: (a) **55** (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 23 °C, 24 h, 81%; (b) TFA (45 equiv), CH₂Cl₂, 0 \rightarrow 23 °C, 2 h; (c) **20** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, 0 \rightarrow 23 °C, 18 h, 99% for the two steps; (d) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, 0 \rightarrow 23 °C, 2 h; (e) **54** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, 0 \rightarrow 23 °C, 24 h, 96% for the two steps; (f) TBDMSOTf (2.0 equiv), 2,6-lutidine (3.0 equiv), CH₂Cl₂, 0 \rightarrow 23 °C, 0.5 h, 87%.

TFA followed by coupling of the resulting amine with Fmocprotected acid fluoride **68** in the presence of *i*-Pr₂NEt afforded tripeptide **69** (86% yield for the two steps). Cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from the latter followed by coupling of the generated amine with *N*-methyl-D-pipecolic acid (**10**) under HATU conditions led to the targeted tubulysin analogue **Tb88** (72% overall yield).

Scheme 20 summarizes the synthesis of tubulysin analogues Tb89 and Tb91, both featuring an alanine in place of their isoleucine residue, and Tb90 and Tb92, which furthermore feature the proline counterpart (as represented by building block 45) of the pipecolic acid residue. The synthesis of these tubulysin analogues started with Boc-protected dipeptide 58¹¹ and proceeded through tripeptide 71. Thus, exposure of 58 to TFA generated the corresponding free amine, which was coupled with acid fluoride 70 (prepared from its amino acid precursor by sequential treatment with FmocCl and DAST; see Supporting Information for further details) in the presence of *i*-Pr₂NEt to furnish 71 in 92% overall yield. Removal of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from this intermediate followed by union of the resulting amine with either N-methyl-D-pipecolic acid (10) or *N*-methyl-D-proline (45) in the presence of HATU led to tubulysin analogues Tb89 (89% overall yield) and Tb90

Scheme 15. Synthesis of Tubulysin Analogue Tb78^a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) **59** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 56% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 69% for the two steps.





^{*a*}Reagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \,^{\circ}C$, 2 h; (b) **61** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \,^{\circ}C$, 18 h; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 \,^{\circ}C$, 2 h, 27% for the three steps; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \,^{\circ}C$, 24 h, 87%; (e) Me₃SnOH (20 equiv), 1,2-dichloroethane, reflux, 5 h, 87%.

(88% overall yield), respectively. Finally, **Tb89** and **Tb90** were converted to their carboxylic acid counterparts **Tb91** and **Tb92** through the sequential action of $Me_3SnOH^{11,57}$ and $Ac_2O/$ pyridine in 82% and 85% overall yield, respectively, as presented in Scheme 20.

Tubulysin analogues **Tb93** and **Tb94**, featuring an ethyl group instead of the isobutyl group at their isoleucine residue,

Scheme 17. Synthesis of Tubulysin Analogues Tb81-Tb84^a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 °C$, 2 h; (b) **63** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 °C$, 18 h, 70% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 °C$, 2 h, 82%; (d) **10** or **45** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 °C$, 24 h, 79% for **Tb81** and 66% for **Tb82**; (e) Me₃SnOH (20 equiv), 1,2-dichloroethane, reflux, 5 h, 82% for the **Tb83** and 79% for the **Tb84**.

were synthesized as summarized in Scheme 21. Thus, removal of the Boc group from fragment 56 (for preparation, see Scheme 14) with TFA and coupling of the so obtained amine with Fmoc-protected acid fluoride 72 in the presence of *i*-Pr₂NEt provided tripeptide 73 (85% yield for the two steps). Cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from the latter followed by coupling of the resulting amine with *N*-methyl-D-pipecolic acid (10) gave tubulysin analogue Tb93 (86% overall yield). Methyl ester Tb93 was then converted to its carboxylic acid counterpart Tb94 through sequential use of Me₃SnOH^{11,57} and Ac₂O/pyridine in 76% overall yield, as shown in Scheme 21.

Scheme 22 summarizes the synthesis of tubulysin analogues **Tb95** and **Tb96**, whose primary feature is the *n*-butyl group in place of their isoleucine side chain. Their synthesis began with removal of the Boc group from intermediate **56** (for preparation, see Scheme 14) and coupling of the liberated amine with Fmoc-protected acid fluoride **74** (prepared from its

Scheme 18. Synthesis of Tubulysin Analogues Tb85-Tb87^a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \, ^\circ C$, 2 h; (b) **66** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \, ^\circ C$, 18 h; (c) N($CH_2CH_2NH_2$)₃ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23 \, ^\circ C$, 2 h, 38% for the three steps; (d) **10** or **45** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \, ^\circ C$, 24 h, 89% for Tb**85** and 88% for Tb**86**; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23 \, ^\circ C$, 12 h, 89% for the two steps.

Scheme 19. Synthesis of Tubulysin Analogue Tb88^a



"Reagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) **68** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 86% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 72% for the two steps.

amino acid counterpart by sequential exposure to FmocCl and DAST) to provide tripeptide **75** (98% yield for the two steps)

Scheme 20. Synthesis of Tubulysin Analogues Tb89-Tb92^a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \,^{\circ}C$, 2 h; (b) 70 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \,^{\circ}C$, 18 h, 92% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 \,^{\circ}C$, 2 h; (d) 10 or 45 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \,^{\circ}C$, 24 h, 89% for the two steps for Tb89 and 88% for the two steps for Tb90; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23 \,^{\circ}C$, 12 h, 82% for the two steps for Tb91 and 85% for the two steps for Tb92.

as shown in Scheme 22 (see Supporting Information for further details). Cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from this intermediate, followed by coupling with either *N*-methyl-D-pipecolic acid (10) or its proline sibling 45 led to tubulysin analogues Tb95 (90% overall yield) and Tb96 (87% overall yield), respectively, as shown in Scheme 22.

Tubulysin analogues **Tb97** and **Tb98**, whose novel feature is their 3-methylbutyl moiety as opposed to their isoleucine residue, were constructed from dipeptide fragment **56** (for preparation see Scheme 14). Thus, removal of the Boc group from dipeptide **56** and coupling of the so generated amine with Fmoc-protected acid fluoride **76** (prepared from its amino acid counterpart by sequential exposure to FmocCl and DAST) provided tripeptide **77** (93% yield for the two steps) as shown in Scheme **23** (see Supporting Information for further details). Cleavage of the Fmoc group $[N(CH_2CH_2NH_2)_3]$ from the latter followed by coupling with either *N*-methyl-D-pipecolic acid (**10**) or *N*-methyl-D-proline (**45**) led to tubulysin analogues **Tb97** (79% overall yield) or **Tb98** (74% overall yield), as summarized in Scheme **23**. Scheme 21. Synthesis of Tubulysin Analogues Tb93 and Tb94^{*a*}



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^\circ$ C, 2 h; (b) 72 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 18 h, 85% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 \ ^\circ$ C, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 24 h, 86% for the two steps; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23 \ ^\circ$ C, 12 h, 76% for the two steps.

Scheme 22. Synthesis of Tubulysin Analogues Tb95 and Tb96 a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^\circC$, 2 h; (b) 74 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circC$, 18 h, 98% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 \ ^\circC$, 2 h; (d) 10 or 45 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circC$, 24 h, 90% for the two steps for Tb95 and 87% for the two steps for Tb96.

Scheme 23. Synthesis of Tubulysin Analogues Tb97 and Tb98 a



^{*a*}Reagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23$ °C, 2 h; (b) 76 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 93% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23$ °C, 2 h; (d) 10 or 45 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 79% for the two steps for Tb97 and 74% for the two steps for Tb98.

Tubulysins **Tb99–Tb101** are characterized with rather drastic modifications at their isoleucine and tubuphenylalanine residues (i.e., cyclopropyl, cubane, and [1.1.1]bicyclopentane moieties). Their syntheses are shown in Scheme 24. Thus, the previously synthesized cyclopropyl-carrying intermediate 78^{11} was coupled with amino esters 79,^{11,60} 80,^{11,60} and 81^{11} under the influence of HATU and Et₃N to afford amides **Tb99** (70% yield), **Tb100** (72% yield) and **Tb101** (75% yield), respectively.

Tubulysin analogues **Tb102** and **Tb103** (both featuring a [1.1.1]bicyclopentane structural motif at the "right edge" of the molecule instead of the tubuphenylalanine residue), **Tb104** (featuring the bulkier naphthalene instead of the phenyl moiety on its tubuphenylalanine residue), and **Tb105** (missing the methyl group on its tubuphenylalanine residue) were synthesized as highlighted in Scheme 25. Thus, key intermediate 82^{11} was coupled with amino acid methyl ester 83 (see Supporting Information for its preparation) under the influence of HATU and Et₃N to afford **Tb103** (79% yield). Similarly carboxylic acid 82^{11} was joined with amino acid methyl ester 80^{11,60} leading to **Tb102** (75% yield). **Tb104** and **Tb105** were synthesized through similar couplings of 82^{11} with fragments 84 (see Supporting Information for preparation) (78% yield) as summarized in Scheme 25.

The next series of tubulysin analogues (i.e., **Tb106–Tb109**, Schemes 26 and 27) were intended to probe the effect of shape, but mainly volume of the lipophilic substituent of the isoleucine residue on the potency of the tubulysin molecule. Thus, tubulysins **Tb106** and **Tb107**, carrying a tertiary butyl group on their isoleucine residue, were synthesized from dipeptide Article





^aReagents and conditions: (a) 79 or 80 or 81 (5.0 equiv), HATU (5.0 equiv), Et₃N (10 equiv), DMF, $0 \rightarrow 23$ °C, 16 h, 70% for Tb99, 72% for Tb100, and 75% for Tb101.

fragment **56** (prepared as shown in Scheme 14) as summarized in Scheme 26. Thus, deprotection of the amino group (TFA) of **56** and coupling of the resulting amine with acid fluoride **86** (see Supporting Information for preparation) in the presence of *i*-Pr₂NEt afforded tripeptide **87** (81% overall yield for the two steps). Removal of the Fmoc group from the latter [N(CH₂CH₂NH₂)₃] and coupling of the so generated amine with carboxylic acid **10**¹¹ (HATU, Et₃N) led first to **Tb106** (76% overall yield) and then to **Tb107** upon ester hydrolysis (Me₃SnOH)^{11,57} and re-acetylation (Ac₂O, pyridine), in 84% yield for the two steps.

Tubulysin analogues **Tb108** and **Tb109**, carrying a 3,3dimethylpentanoic group at their isoleucine residue were similarly synthesized from **56** as depicted in Scheme 27. Thus, deprotection of **56** as described above (i.e., TFA) followed by *i*-Pr₂NEt facilitated coupling of the resulting amine with acid fluoride **88** (see Supporting Information for preparation) furnished intermediate tripeptide **89** (72% overall yield from **56**). Fmoc removal from **89** with [N-(CH₂CH₂NH₂)₃] and coupling of the so obtained amine with carboxylic acid **10** facilitated by HATU and Et₃N led, in 74% overall yield, to analogue **Tb108**. Finally, exposure of **Tb108** to Me₃SnOH furnished the corresponding hydroxy carboxylic acid, which was acetylated (Ac₂O, pyridine) to afford analogue **Tb109**, in 70% overall yield from **Tb108**, as shown in Scheme 27.

In an attempt to decipher further SARs within the tubulysin family of compounds we designed and synthesized tubulysin analogues Tb110–Tb113 (Scheme 28) equipped with benzyloxy ethyl and hydroxy ethyl groups on the thiazole ring. To this end, bromothiazole methyl ester 90^{61} (for preparation, see Supporting Information) was reduced with DIBAL-H to the corresponding alcohol, which was silylated (TBSOTf, 2,6-lutidine) to afford TBS-ether **91** (86% overall

Scheme 25. Synthesis of Tubulysin Analogues Tb102–Tb105 a



^aReagents and conditions: (a) **80** or **83** or **84** or **85** (1.2 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 75% for Tb102, 79% for Tb103, 78% for Tb104, and 78% for Tb105.

yield). Bromide 91 was treated with *n*-BuLi, and to the resulting lithio derivative was added Weinreb amide 92¹¹ to afford ketone 93 in 70% yield. Reduction of the latter with BH₃·Me₂S in the presence of CBS catalyst^{11,56} furnished stereoselectively hydroxy compound 94 (66% yield). The latter compound was elaborated to acetoxy carboxylic acid 95 through a sequence involving acetylation (Ac2O, Et3N, 88% yield), desilylation (TBAF, 99% yield), and oxidation (DMP, 91% yield; then NaClO₂, 99% yield). Coupling carboxylic acid 95 with ammonium salt 55 through the action of HATU and Et₃N furnished fragment 96 in 93% yield. Removal of the Boc group (TFA) from the latter followed by coupling of the so generated amine with acid fluoride 20 gave tripeptide 97 (92% overall yield). Finally, cleavage of the Fmoc group from 97 and coupling of the resulting amine with carboxylic acid 10 facilitated by HATU and Et₃N led to analogue Tb110 in 75% overall yield. Tubulysin analogue Tb111 was obtained from Tb110 through the standard procedure of hydrolysis (acetate and ethyl ester) with LiOH¹¹ followed by reacetylation of the resulting hydroxy acid with Ac₂O/pyridine (77% overall yield). Finally, Tb110 and Tb111 were converted to Tb112 and Tb113 in 71% and 65% yield, respectively, through hydrogenolysis $[Pd(OH)_2/C \text{ catalyst, } H_2]$ as shown in Scheme 28.

We finally explored modifications at the tubuphenylalanine (Tup) site of the tubulysin molecule as demonstrated with the

Scheme 26. Synthesis of Tubulysin Analogues Tb106 and Tb107 a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^\circC$, 2 h; (b) **86** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circC$, 18 h, 81% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, 0 $\rightarrow 23 \ ^\circC$, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circC$, 24 h, 76% for the two steps; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23 \ ^\circC$, 12 h, 84% for the two steps.

Scheme 27. Synthesis of Tubulysin Analogues Tb108 and Tb109^a



^aReagents and conditions: (a) TFA (45 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^\circ$ C, 2 h; (b) **88** (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 18 h, 72% for the two steps; (c) N(CH₂CH₂NH₂)₃ (15 equiv), CH₂Cl₂, $0 \rightarrow 23 \ ^\circ$ C, 2 h; (d) **10** (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23 \ ^\circ$ C, 24 h, 74% for the two steps; (e) Me₃SnOH (50 equiv), 1,2-dichloroethane, reflux, 12 h; (f) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23 \ ^\circ$ C, 12 h, 70% for the two steps.

Scheme 28. Synthesis of Tubulysin Analogues Tb110–Tb113^{*a*}



^aReagents and conditions: (a) DIBAL-H (3.0 equiv; 1.0 M in diethyl ether), diethyl ether, $-78 \rightarrow 23$ °C, 1 h, 87%; (b) TBSOTf (1.2 equiv), 2,6-lutidine (2.0 equiv), CH_2Cl_2 , $0 \rightarrow 23 \,^{\circ}C$, 0.5 h, 99%; (c) *n*-BuLi (1.44 equiv; 2.5 M in hexanes), 92 (1.0 equiv), THF, $-78 \rightarrow$ -50 °C, 3 h, 70%; (d) (S)-CBS (0.1 equiv; 1.0 M in toluene), BH₃. Me₂S (1.0 equiv; 2.0 M in THF), $0 \rightarrow 23$ °C, 36 h, 66%; (e) Ac₂O (3.0 equiv), Et_3N (4.0 equiv), $0 \rightarrow 23$ °C, 2 h, 88%; (f) TBAF (2.0 equiv; 1.0 M in THF), THF, $0 \rightarrow 23$ °C, 30 min, 99%; (g) DMP (1.5 equiv), CH₂Cl₂, 23 °C, 0.5 h, 91%; (h) NaClO₂ (5.4 equiv), NaH₂PO₄·H₂O (12.2 equiv), 2-methyl-2-butene (7.5 equiv), t-BuOH, THF, H₂O, 23 °C, 1 h, 99%; (i) 55 (1.5 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 93%; (j) TFA (45 equiv), CH_2Cl_2 , 0 \rightarrow 23 °C, 2 h; (k) 20 (4.0 equiv), *i*-Pr₂NEt (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 92% for the two steps; (l) N(CH₂CH₂NH₂)₃ (15 equiv), CH_2Cl_2 , $0 \rightarrow 23 \ ^{\circ}C$, 2 h; (m) 10 (3.0 equiv), HATU (3.0 equiv), Et₃N (6.0 equiv), DMF, $0 \rightarrow 23$ °C, 24 h, 75% for the two steps; (n) LiOH·H₂O (5.0 equiv), THF:H₂O (5:1, v/v), 23 °C, 24 h; (o) Ac₂O (4.0 equiv), pyridine, $0 \rightarrow 23$ °C, 12 h, 77% for the two steps; (p) Pd(OH)₂/C (20 wt%), H₂, MeOH, 23 °C, 18 h, 71% for Tb112 and 65% for Tb113. DIBAL-H = diisobutylaluminum hydride.

Scheme 29. Synthesis of Tubulysin Analogue Tb114^a



"Reagents and conditions: (a) 98 (1.2 equiv), HATU (1.2 equiv), Et_3N (2.4 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 54%.

Scheme 30. Synthesis of Tubulysin Analogues Tb115–Tb118^{*a*}



^aReagents and conditions: (a) **99** (ca. 4:1 dr, 1.2 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 26% for **Tb115** and 52% for **Tb116**; (b) 10% Pd/C (50 wt%), H₂, MeOH, 23 °C, 20 h, 88%; (c) 10% Pd/C (50 wt%), H₂, wet EtOH, 23 °C, 20 h, 77%.

structures of analogues Tb114–Tb118 (Schemes 29 and 30). Tubulysin analogue Tb114, possessing a C_2 -symmetric malonic acid type structural motif at its "right end" (resembling Meldrum's acid structure) was synthesized from carboxylic acid 82¹¹ and ammonium salt 98⁶² (see Supporting Information for details) through the action of HATU in the presence of Et₃N in 54% yield, as shown in Scheme 29.

The amino containing tubulysin analogues Tb115-Tb118 were prepared from the previously synthesized fragment 82^{11} and amino acid derivative 99⁶³ (diastereomeric mixture ca. 4:1, see Supporting Information for preparation) as summarized in Scheme 30. Thus, coupling of carboxylic acid 82 with ammonium salt 99 in the presence of HATU and Et₃N yielded Tb115 and Tb116 as a mixture of diastereoisomers (78% yield, ca. 1:2, separated by silica gel column chromatography). Tb116 was subjected to hydrogenolysis in MeOH (10% Pd/C, 50 wt%, H₂, 23 °C, 20 h) to afford dimethylamino tubulysin analogue Tb117 in 88% yield. Similar treatment of Tb116 in EtOH led to ethyl amino tubulysin analogue Tb118 in 77% yield. Apparently, this is a known outcome of hydrogenolytic Cbz group cleavage from primary amines under strong catalyst and prolonged time conditions.⁶⁴ As demonstrated here, this reaction provides practical means for accessing substituted amines of considerable complexity.

Scheme 31 summarizes the synthesis of Tb119 and Tb120, both containing a hydroxymethyl group adjacent to their





"Reagents and conditions: (a) 33 (1.2 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 60% for Tb119 and 18% for Tb120.

carboxylate moiety (instead of a methyl group) of the tubuphenylalanine residue. Thus, tripeptide carboxylic acid **82**¹¹ was coupled with ammonium salt **33** (ca. 4:1 dr, see Supporting Information for preparation) under the influence of HATU and Et₃N to afford tubulysin analogues **Tb119** and **Tb120** as a mixture of diastereoisomers, which were separated by HPLC to give pure **Tb119** (60% yield) and **Tb120** (18% yield).⁶⁴

Stereoselective syntheses of tubulysin analogues **Tb119** and **Tb120** were also developed starting with readily available phenylalanine derivative **100**, as shown in Schemes 32 and 33. Thus, substrate **100** was converted to its azide counterpart **101** through a two-step sequence (TFA; TfN₃, CuSO₄ catalyst, 81% overall yield, Scheme 32). The latter was reacted with chiral auxiliary **102** [(*S*)-4-benzyl-2-oxazolidinone] in the presence of PivCl, Et₃N, and LiCl to afford oxazolidinone **103** (56% yield), which was hydroxymethylated stereoselectively with trioxane in the presence of TiCl₄, leading to hydroxy azide oxazolidinone





^aReagents and conditions: (a) TFA (45 equiv), CH₂Cl₂, 0 °C, 3 h, quant; (b) TfN₃ (3.0 equiv), 0.57 M in CH₂Cl₂), CuSO₄·SH₂O (0.1 equiv), K₂CO₃ (2.0 equiv), MeOH, H₂O, 23 °C, 12 h, 81% yield for the two steps; (c) Et₃N (1.8 equiv), LiCl (1.7 equiv), PivCl (1.5 equiv), (S)-4-benzyl-2-oxazolidinone **102** (1.7 equiv), -20 °C, 2 h, 56%; (d) TiCl₄ (1.1 equiv), *i*-Pr₂NEt (1.1 equiv), 1,3,5-trioxane (1.1 equiv), 0 °C, 3.5 h, 54% for **104** and 28% for **105**; (e) NaOMe (1.0 equiv), MeOH, CH₂Cl₂, -78 \rightarrow 0 °C, 2 h, 95%; (f) 10% Pd/C (50 wt %), H₂, HCl (1.2 equiv; 1 M in MeOH), MeOH, 23 °C, 0.5 h, 99%; (g) **82** (0.9 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, 0 \rightarrow 23 °C, 18 h, 95%; (h) NaOMe (1.0 equiv), MeOH, CH₂Cl₂, -78 \rightarrow 0 °C, 2 h, 78%.

104 (54% yield) and, unexpectedly, its C2 epimeric sibling 105 (28% yield). Besides NMR spectroscopic analyses revealing its general structure, the absolute stereochemistry of the latter compound was established by converting it to hydroxy ester 108 (NaOMe; 78% yield), whose data matched those obtained from another sample of the same compound prepared from oxazolidinone 111 as shown in Scheme 33. Hydroxy azide methyl ester 106 was then generated from oxazolidinone 104 through the action of NaOMe (95% yield). Reduction $(H_2/Pd$ catalyst, HCl, MeOH, 99% yield) of the azide group within the latter provided hydroxy ammonium salt 107, which was smoothly coupled with tripeptide carboxylic acid 82^{11} under HATU conditions to afford the targeted tubulysin analogue Tb119 in 95% yield, as summarized in Scheme 32. The stereoselective synthesis of diastereomeric analogue Tb120 proceeded through the same sequence, starting with carboxylic acid 101 and via intermediates 110, 111, 108, and 112, by utilizing the enantiomeric chiral auxiliary [109: (R)-4-benzyl-2oxazolidinone] in similar yields and without the formation of





^aReagents and conditions: (a) Et₃N (1.8 equiv), LiCl (1.7 equiv), PivCl (1.5 equiv), **109** (1.7 equiv), -20 °C, 2 h, 55%; (b) TiCl₄ (1.1 equiv), *i*-Pr₂NEt (1.1 equiv), 1,3,5-trioxane (1.1 equiv), 0 °C, 3.5 h, 61%; (c) NaOMe (1.0 equiv), MeOH, CH₂Cl₂, $-78 \rightarrow 0$ °C, 2 h, 94%; (d) 10% Pd/C (50 wt%), H₂, HCl (1.2 equiv; 1 M in MeOH), MeOH, 23 °C, 0.5 h, 98%; (e) **82** (0.9 equiv), HATU (1.2 equiv), Et₃N (2.4 equiv), DMF, $0 \rightarrow 23$ °C, 18 h, 94%.

epimeric side products during the stereoselective hydroxymethylation step (110 \rightarrow 111) as shown in Scheme 33.

2.5. Biological Evaluation of Tubulysin Analogues and Structure-Activity Relationships. The synthesized analogues were evaluated for their activity against MES SA (uterine sarcoma cells), MES SA DX (multidrug-resistant uterine sarcoma cells), and HEK 293T (human embryonic kidney cancer cells). As shown in Table 1, several of these compounds exhibited picomolar potencies, with the most notable (highlighted with yellow background) being Tb94, Tb95, and Tb102, and even more potent (highlighted with green background) Tb64, Tb67, Tb73, Tb93, Tb106, Tb108, Tb112, Tb117, Tb119, Tb120, and most potent (highlighted with blue background) Tb107, Tb109, Tb111, Tb115, Tb116, and Tb118. The latter group (i.e., Tb107, Tb109, Tb111, Tb115, Tb116, and Tb118) is also notable for their relatively potent (low nanomolar) cytotoxicities against the marked drugresistant cancer cell line MES SA DX.

With the large number of tubulysins synthesized and tested in our laboratories, and guided by the insights recently obtained through X-ray crystallographic studies⁵³ on tubulin binding molecules, we were in a position to explain and develop clearer SARs within the tubulysin family of compounds. The X-rayderived structures of N^{14} -desacetoxytubulysin H (**Tb1** in this study; tubulysin M in ref 53) and its peptide-like relatives HTI-286⁵³ and monomethyl auristatin E (MMAE)⁵³ revealed a binding model that included a number of binding sites on the tubulysin molecule, including (from "left" to "right", see Figure 3): (1) basic nitrogen on the "left domain" (protonated form, binding through a salt bridge to a carboxylate moiety of tubulin); (2) one amide NH moiety binding through Hbonding to a carbonyl O of the receptor; (3) a hydrophobic moiety (i.e., 2-methyl butyl group) binding to a hydrophobic pocket within tubulin; (4) carbonyl O binding through Hbonding to an amide NH moiety of tubulin; (5) a second hydrophobic group (i.e., isopropyl moiety) binding to a different hydrophobic pocket within the tubulin unit; (6) the thiazole N and the adjacent carbonyl O, both serving as Hbond acceptors from a H-bond donor on the tubulin unit; (7) the phenyl moiety of the tubuphenyl alanine residue fitting snuggly into a hydrophobic cavity within a tubulin unit; and (8) the carboxylate unit forming a salt bridge with a counterpart within the tubulin receptor. This model seems to be, more or less, in accordance with our findings, correlating well structural motifs with potencies within the family of compounds synthesized and tested in this study (see Figure 2 and Table 1).

As the pipecolic acid residue of the tubulysin molecule perfectly occupies its binding site on tubulin, according to the X-ray-generated model,⁵³ it was not surprising that certain modifications made to this fragment led to only insignificant or low biological activity. This phenomenon is clearly demonstrated by tubulysin analogues Tb50 and Tb51 (in which the pipecolic acid was exchanged for 1-methyl-1H-pyrrole-2carboxylic acid). It was also evident in the cases of PTb-D47, PTb-D49, Tb54, Tb55, and Tb77, in which the pipecolic acid residue was modified at the N-atom with the larger *n*-butyl in place of the methyl group or oxygenated on one of the carbons of the ring, changing its steric and/or hydrophobic requirements that apparently do not fit the binding site of the tubulin receptor. In addition, when the pipecolic acid moiety was replaced with its five-membered proline counterpart, as in Tb70 and Tb71, significant loss of potency was observed (see Table 1), providing further support for the strict and crucial requirements of the pipecolic acid binding site within the tubulin receptor, although other novel substituents on this nitrogen-bearing residue may prove fruitful.

As can be seen in Figure 3, the isoleucine (Ile) residue of the tubuysin ligand provides a hydrogen-bonding opportunity (one acceptor and one donor) and one hydrophobic moiety (the 3methybutyl group) that fits snuggly into the α 2 tubulin subunit. As shown from our results, the tolerance of the hydrophobic acceptor site is rather limited. Thus, tubulysins equipped with an isopropyl (e.g., Tb61), tertiary butyl (e.g., Tb106, Tb107), and the one-carbon higher 1,1-dimethylpropyl moieties (e.g., Tb108, Tb109) exhibited exceptional potencies. On the other hand, the longer n-butyl (e.g., Tb95, Tb96) groups and 3methylbutyl (e.g., Tb97, Tb98) instead of the isoleucine side chain are not tolerated, as evident from the lack of or significantly lower cytotoxicities of the corresponding analogues (see Table 1). Furthermore, smaller size group substitutions at the isoleucine side chain position, as in Tb88 (hydrogen) and Tb89-Tb92 (methyl group), led, surprisingly, to no significant activity. Interestingly, the ethyl group containing tubulysin analogues Tb93 and Tb94 exhibited significant potencies, indicating perhaps the lower limit of lipophilicity and steric demand required at that position for potent activity. The fluorinated tubulysin analogues Tb79-Tb87 carrying substituted ethyl or isopropyl moieties on the isoleucine residue, were disappointing in that none exhibited subnanomolar potencies, although some had significant activities (see Table 1). This observation may be attributed to the polarization of

Table 1. Cytotoxicity Data against Cancer Cell Lines MES SA, MES SA DX, and HEK 293T^a for Tubulysins PTb-D47–PTb-D49 and Tb50–Tb120

compound	MES SA	MES SA DX	HEK 293T	compound	MES SA	MES SA DX	HEK 293T
Tb1	0.34	>10	0.02	Tb83	>1000	>1000	>1000
Tb32	0.012	1.29	0.002	Tb84	>1000	>1000	>1000
PTb-D47	>1000	>1000	>1000	Tb85	37.32	>1000	32.4
PTb-D48	>1000	>1000	>1000	Tb86	>1000	>1000	>1000
PTb-D49	>1000	>1000	>1000	Tb87	233.5	>1000	>1000
Tb50	>1000	>1000	>1000	Tb88	457.70	>2500	493.80
Tb51	>1000	>1000	>1000	Tb89	>1000	>1000	>1000
Tb52	6.164	66.55	3.83	Tb90	>1000	>1000	636.4
Tb53	>1000	>1000	>1000	Tb91	>1000	>1000	>1000
Tb54	>1000	>1000	>1000	Tb92	>1000	>1000	>1000
Tb55	>1000	>1000	>1000	Tb93	0.15	31.93	0.20
Tb56	>1000	>1000	>1000	Tb94	0.46	89.77	0.40
Tb57	>1000	>1000	>1000	Tb95	0.937	>1000	0.53
Tb58	>1000	>1000	>1000	Tb96	10.5	>1000	5.26
Tb59	>1000	>1000	>1000	Tb97	4.65	>1000	2.87
Tb60	14.02	>2500	6.34	Tb98	41.8	393.2	26.5
Tb61	6.07	>1000	5.99	Tb99	>1000	>1000	>1000
Tb62	>1000	>1000	>1000	Tb100	15.16	>1000	16.51
Tb63	>1000	>1000	>1000	Tb101	>1000	>1000	>1000
Tb64	0.22	108.70	0.10	Tb102	0.926	54.12	0.355
Tb65	2.44	278.80	2.04	Tb103	2.74	22.57	1.793
Tb66	1.034	>1000	0.773	Tb104	9.5	>1000	12.83
Tb67	0.836	71.52	0.1435	Tb105	2.94	>1000	1.53
Tb68	>1000	>1000	>1000	Tb106	0.12	2.73	0.13
Tb69	30.82	>1000	>1000	Tb107	0.01	4.05	0.02
Tb70	3.113	>1000	2.32	Tb108	0.95	6.08	0.38
Tb71	1.422	>1000	0.408	Tb109	0.36	1.39	0.01
Tb72	11.70	>70	4.02	Tb110	3.091	>400	1.87
Tb73	1.28	44.69	0.16	Tb111	0.04	1.54	0.006
Tb74	5.32	>400	2.46	Tb112	0.10	>400	0.09
Tb75	400.00	>400	126.60	Tb113	5.79	>400	0.315
Tb76	4.88	>2500	1.37	Tb114	1.225	>400	0.524
Tb77	848	>2500	>400	Tb115	0.020	13.850	0.010
Tb78	7.18	780.5	5.77	Tb116	0.007	6.002	0.003
Tb79	39.09	>1000	36.37	Tb117	0.099	4.629	0.059
Tb80	>1000	>1000	>1000	Tb118	0.14	5.60	0.015
Tb81	5.84	>1000	4.98	Tb119	0.093	>14	0.041
Tb82	47.53	>1000	51.60	Tb120	0.059	>14	0.039

 a IC₅₀ = 50% inhibitory concentration of compound against cell growth; MES SA = uterine sarcoma cell line; MES SA DX = MES SA cell line with marked multidrug resistance; HEK 293T = human embryonic kidney cancer cell line; ^bSee Supporting Information for further details.





Figure 3. Binding interactions of N^{14} -desacetoxytubulysin H [Tb1, (tubulysin M)^{53a}] as deciphered from X-ray crystallographic analysis.^{53a}

the bonds within these moieties that contribute negatively to their fitting into the hydrophobic pocket of the receptor.

A more complex explanation may be necessary for the effect of the spiro cyclopentyl moiety within the isoleucine residue leading to the lower potency of **Tb78**, the latter structural motif most likely changing the overall tertiary conformation of the tubulysin molecule, thereby decreasing its complementarity to its binding pocket.

According to the crystallographic analysis,⁵³ the acetate group of the Tuv fragment sits at a narrow channel within the interface of the α and β tubulin subunits, with no recognizable H-bonding interactions. The significant reduction in potency that accompanies the removal of this acetate (as in **PTb-D42**,

PTb-D43, and **DTb-D48**), its deprotection to the naked hydroxyl group (as in **Tb58** and **Tb60**), and its oxidation to the corresponding ketone (as in **Tb59**, **Tb62**, **Tb63**, and **Tb68**), may suggest an unknown structural or biochemical function of this moiety (e.g., facilitating entrance of the molecule into the cell).

The thiazole component of Tuv forms two H-bonding interactions emanating from the thiazole nitrogen atom and the adjacent carbonyl to the backbone of the β 1 tubulin subunit, thus stabilizing the overall conformation of the central region of the bound tubulysin molecule. Consequently, any aromatic functionality that maintains these interactions should be tolerated as long as it does not contribute to additional steric or electronic constraints within the binding channel, as demonstrated by analogues Tb64 and Tb65 (pyridine instead of thiazole). The 5-position (i.e., H-substituted position) of the thiazole ring is oriented toward an open space⁵³ and away from the interface of the two tubulin monomers, although some steric constraints could be imposed by substituents in the near neighborhood of the thiazole ring.53 The recognition of this open space inspires and provides guidance for further refinement of the tubulysin molecule as potential payloads for ADCs. Thus, methyl substitution at the 5-position in analogues Tb66 and Tb67 results in some loss of potency, whereas the presence of an isopropyl group in analogues Tb72 and Tb73 led to significant loss of potency. In contrast, the longer linear chains, as in Tb110-Tb113, proved beneficial as demonstrated by their generally increased cytotoxicity potency.

The tubulysin phenylalanine (Tup) domain positioned as it is at the "right end" of the molecule binds, according to the Xray data,⁵³ at the edge of the binding channel of tubulin. As such, it is free to rotate as long as structural changes do not disturb other binding interactions. Thus, the [1.1.1]bicyclopentane containing analogues Tb100, Tb102, and Tb103 and the cyclohexyl carrying analogue Tb52 are marginally tolerated, with Tb102 showing subnanomolar potencies against two of the cell lines tested (see Table 1). The bulkier naphthalene substituent at this position, as in Tb104, is also barely tolerated as concluded from its modest potency compared to the most active compounds (see Table 1). Although *para*-substitution of the aromatic ring of the Tup phenylalanine residue is tolerated as evidenced by a number of active natural tubulysins (e.g., A–C, G, and I) $^{15-18}$ that contain a phenolic moiety at that position, the presence of a fluorine residue in this aromatic ring, as in analogue Tb101, is not, leading to loss of activity (see Table 1). These data suggest the importance of a potential $\pi - \pi$ interaction of this moiety with a binding site in the receptor.

The Tup carboxylic acid moiety is involved in a very important interaction with Arg278 of the tubulin receptor, forming a salt bridge that provides additional stabilization of the ligand-receptor complex. The adjacent methyl group appears to be in an open space region not limited by any apparent steric or electronic constraints. Removal of this methyl group, as in **Tb105**, results in considerable loss of activity (see Table 1). Most importantly, however, the replacement of this methyl group of the Tup residue with nitrogen-containing substituents, as in **Tb115-Tb118**, translates into high potencies, with **Tb117** and **Tb118** being the most impressive.

Figure 4 summarizes the conclusions drawn from these studies on the effect of structural changes within the four



Figure 4. Structure-activity relationships (SARs).

domains Mep [(*N*-methyl-D-pipecolic acid or pipecolic acid)], Ile (L-isoleucine), Tuv (tubuvaline), and Tup (tubuphenyl alanine) on the cytotoxicity potencies of the N^{14} -desacetoxytubulysin H molecule (**Tb1**). These SARs are consistent with the X-ray crystallographic studies of the tubulysin–tubulin complex and provide guidance for the design of new optimized tubulysin analogues.

3. CONCLUSION

The described synthetic endeavors culminated in short and efficient chemical processes for the synthesis of natural tubulysins V (Tb45) and U (Tb46) and pretubulysin D (PTb-D43) and allowed rapid and efficient syntheses of a number of tubulysin analogues (Tb44, PTb-D42, PTb-D47–PTb-D49, and Tb50–Tb120). Biological evaluation of the

synthesized compounds led to the identification of extremely potent tubulysin analogues (e.g., **Tb107**, **Tb111**, **Tb115**, and **Tb116**, see **Table 1** and **Figure 2**) equipped with hydroxyl, amino, and carboxylate handles for conjugation to appropriate targeting systems such as antibodies^{50,51} and other delivery systems.⁵² Such conjugates are highly desirable for personalized targeted chemotherapies, a currently intensely pursued and rapidly emerging paradigm for cancer treatment. The newly developed structure–activity relationships described herein are in line with recently reported X-ray crystallographic analysis results obtained from a tubulysin–tubulin complex,⁵³ and provide strong pathpointing guidance for further optimization of the latest tubulysin analogues.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.7b12692.

Experimental procedures and characterization data for all compounds; biological evaluation and data, HEK 293T, MES SA, and MES SA DX (AbbVie Stemcentrx) (PDF)

AUTHOR INFORMATION

Corresponding Author

*kcn@rice.edu

ORCID [©]

K. C. Nicolaou: 0000-0001-5332-2511

Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Dömling, A.; Richter, W. Mol. Diversity 2005, 9, 141.

(2) Sasse, F.; Steinmetz, H.; Heil, J.; Höfle, G.; Reichenbach, H. J. Antibiot. 2000, 53, 879.

- (3) Sandmann, A.; Sasse, F.; Müller, R. Chem. Biol. 2004, 11, 1071.
 (4) Khalil, M. W.; Sasse, F.; Lünsdorf, H.; Elnakady, Y. A.;
- Reichenbach, H. ChemBioChem 2006, 7, 678.

(5) Kubicek, K.; Grimm, S. K.; Orts, J.; Sasse, F.; Carlomagno, T. Angew. Chem., Int. Ed. 2010, 49, 4809.

(6) Steinmetz, H.; Glaser, N.; Herdtweck, E.; Sasse, F.; Reichenbach, H.; Höfle, G. Angew. Chem., Int. Ed. 2004, 43, 4888.

(7) Chai, Y.; Pistorius, D.; Ullrich, A.; Weissman, K. J.; Kazmaier, U.; Müller, R. *Chem. Biol.* **2010**, *17*, 296.

(8) (a) Ullrich, A.; Chai, Y.; Pistorius, D.; Elnakady, Y. A.; Herrmann, J. E.; Weissman, K. J.; Kazmaier, U.; Müller, R. Angew. Chem., Int. Ed. 2009, 48, 4422. (b) Braig, S.; Wiedmann, R. M.; Liebl, J.; Singer, M.; Kubisch, R.; Schreiner, L.; Abhari, B. A.; Wagner, E.; Kazmaier, U.; Fulda, S.; Vollmar, A. M. Cell Death Dis. 2014, 5, e1001. (c) Herrmann, J.; Elnakady, Y. A.; Wiedmann, R. M.; Ullrich, A.; Rohde, M.; Kazmaier, U.; Vollmar, A. M.; Müller, R. PLoS One 2012, 7, e37416. (d) Kubisch, R.; von Gamm, M.; Braig, S.; Ullrich, A.; Burkhart, J. L.; Colling, L.; Hermann, J.; Scherer, O.; Müller, R.; Werz, O.; Kazmaier, U.; Vollmar, A. M. J. Nat. Prod. 2014, 77, 536. (e) Hoffmann, J.; Gorges, J.; Junk, L.; Kazmaier, U. Org. Biomol. Chem. 2015, 13, 6010. (9) Murray, B. C.; Peterson, M. T.; Fecik, R. A. Nat. Prod. Rep. 2015, 32, 654.

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(10) Neri, D.; Fossati, G.; Zanda, M. *ChemMedChem* **2006**, *1*, 175. (11) Nicolaou, K. C.; Yin, J.; Mandal, D.; Erande, R. D.; Klahn, P.; Jin, M.; Aujay, M.; Sandoval, J.; Gavrilyuk, J.; Vourloumis, D. *J. Am. Chem. Soc.* **2016**, *138*, 1698.

- (12) Xu, X.; Friestad, G. K.; Lei, Y. Mini-Rev. Med. Chem. 2013, 13, 1572.
- (13) Kazmaier, U.; Ullrich, A.; Hoffmann, J. Open Nat. Prod. J. 2013, 6, 12.
- (14) Höfle, G.; Glaser, N.; Leibold, T.; Karama, U.; Sasse, F.; Steinmetz, H. Pure Appl. Chem. 2003, 75, 167.
- (15) Pando, O.; Dörner, S.; Preusentanz, R.; Denkert, A.; Porzel, A.; Richter, W.; Wessjohann, L. Org. Lett. 2009, 11, 5567.
- (16) (a) Shibue, T.; Hirai, T.; Okamoto, I.; Morita, N.; Masu, H.; Azumaya, I.; Tamura, O. *Chem. - Eur. J.* **2010**, *16*, 11678. (b) Shankar, P. S.; Bigotti, S.; Lazzari, P.; Manca, I.; Spiga, M.; Sani, M.; Zanda, M. *Tetrahedron Lett.* **2013**, *54*, 6137.
- (17) Sasse, F.; Menche, D. Nat. Chem. Biol. 2007, 3, 87.
- (18) Peltier, H. M.; McMahon, J. P.; Patterson, A. W.; Ellman, J. A. J. Am. Chem. Soc. 2006, 128, 16018.
- (19) Yang, X.; Dong, C.; Chen, J.; Ding, Y.; Liu, Q.; Ma, X.; Zhang, Q.; Chen, Y. *Chem. Asian J.* **2013**, *8*, 1213.
- (20) Sani, M.; Fossati, G.; Huguenot, F.; Zanda, M. Angew. Chem., Int. Ed. 2007, 46, 3526.
- (21) Dömling, A.; Beck, B.; Eichelberger, U.; Sakamuri, S.; Menon,
- S.; Chen, Q.-Z.; Lu, Y.; Wessjohann, L. A. Angew. Chem., Int. Ed. 2006, 45, 7235.
- (22) Balasubramanian, R.; Raghavan, B.; Begaye, A.; Sackett, D. L.; Fecik, R. A. J. Med. Chem. **2009**, *52*, 238.
- (23) Wang, R.; Tian, P.; Lin, G. Chin. J. Chem. 2013, 31, 40.
- (24) Tao, W.; Zhou, W.; Zhou, Z.; Si, C. M.; Sun, X.; Wei, B. G. *Tetrahedron* **2016**, *72*, 5928.
- (25) Wipf, P.; Wang, Z. Org. Lett. 2007, 9, 1605.
- (26) Raghavan, B.; Balasubramanian, R.; Steele, J. C.; Sackett, D. L.; Fecik, R. A. J. Med. Chem. **2008**, *51*, 1530.
- (27) Vlahov, I. R.; Wang, Y.; Vetzel, M.; Hahn, S.; Kleindl, P. J.; Reddy, J. A.; Leamon, C. P. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 6778.
- (28) Floyd, W. C., III; Datta, G. K.; Imamura, S.; Kieler-Ferguson, H. M.; Jerger, K.; Patterson, A. W.; Fox, Me. E.; Szoka, F. C.; Fréchet, J.
- M. J.; Ellman, J. A. ChemMedChem 2011, 6, 49.
- (29) Patterson, A. W.; Peltier, H. M.; Sasse, F.; Ellman, J. A. Chem. -Eur. J. 2007, 13, 9534.
- (30) Rath, S.; Liebl, J.; Fürst, R.; Ullrich, A.; Burkhart, J. L.; Kazmaier, U.; Herrmann, J.; Müller, R.; Günther, M.; Schreiner, L.; Wagner, E.;
- Vollmar, A. M.; Zahler, S. Br. J. Pharmacol. 2012, 167, 1048.
- (31) Eirich, J.; Burkhart, J. L.; Ullrich, A.; Rudolf, G. C.; Vollmar, A.; Zahler, S.; Kazmaier, U.; Sieber, S. A. *Mol. BioSyst.* **2012**, *8*, 2067.
- (32) Burkhart, J. L.; Müller, R.; Kazmaier, U. Eur. J. Org. Chem. 2011, 2011, 3050.
- (33) Ullrich, A.; Chai, Y.; Pistorius, D.; Elnakady, Y. A.; Herrmann, J. E.; Weissman, K. J.; Kazmaier, U.; Müller, R. *Angew. Chem., Int. Ed.* **2009**, *48*, 4422.
- (34) Shibue, T.; Okamoto, I.; Morita, N.; Morita, H.; Hirasawa, Y.; Hosoya, T.; Tamura, O. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 431.
- (35) Pando, O.; Stark, S.; Denkert, A.; Porzel, A.; Preusentanz, R.; Wessjohann, L. A. J. Am. Chem. Soc. 2011, 133, 7692.
- (36) Shankar P, S.; Sani, M.; Saunders, F. R.; Wallace, H. M.; Zanda, M. Synlett **2011**, 2011, 1673.
- (37) Wang, Z.; McPherson, P. A.; Raccor, B. S.; Balachandran, R.; Zhu, G.; Day, B. W.; Vogt, A.; Wipf, P. *Chem. Biol. Drug Des.* **2007**, *70*, 75.
- (38) Shankar, P. S.; Jagodzinska, M.; Malpezzi, L.; Lazzari, P.; Manca, I.; Greig, I. R.; Sani, M.; Zanda, M. Org. Biomol. Chem. **2013**, *11*, 2273.
- (39) Burkhart, J. L.; Kazmaier, U. RSC Adv. 2012, 2, 3785.
- (40) Yang, X.; Dong, C.; Chen, J.; Liu, Q.; Han, B.; Zhang, Q.; Chen, Y. *Tetrahedron Lett.* **2013**, *54*, 2986.
- (41) Balasubramanian, R.; Raghavan, B.; Steele, J. C.; Sackett, D. L.; Fecik, R. A. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 2996.
- (42) Patterson, A. W.; Peltier, H. M.; Ellman, J. A. J. Org. Chem. 2008, 73, 4362.

(43) Park, Y.; Lee, J. K.; Ryu, J. S. Synlett 2015, 26, 1063.

- (44) (a) Vlahov, I. R.; Leamon, C. P.; Wang, Y. U.S. Patent US 2010/ 0240701A1, 2010. (b) Dömling, A.; Henkel, B.; Beck, B.; Illgen, K.; Sakamuri, S.; Menon, S. Int. Patent WO 2004/005327, 2004. (c) Cheng, H.; Cong, Q.; Gangwar, S.; Zhang, Q. U.S. Patent US 2011/0027274A1, 2011. (d) Dömling, A.; Henkel, B.; Beck, B.; Illgen, K.; Sakamuri, S.; Menon, S. U.S. Patent US 7,816,377B2, 2010. (e) Ellman, J. A.; Patterson, A. W.; Peltier, H.; Sasse, F. Int. Patent WO 2009/012958A2, 2009. (f) Vlahov, I. R.; Wang, Y.; Leamon, C. P. Int. Patent WO 2009/055562A1, 2009. (g) Müller; Chai, Y.; Kazmaier, U.; Ullrich, A. Eur. Patent EP 2 174 947A1, 2010. (h) Vlahov, I. R.; Santhapuram, H. K. R; Kleindl, P. J.; Leamon, C. P.; You, U. Int. Patent WO 2013/149185A1, 2013. (i) Wessiohann, L. A.: Pando, M. O. Eur. Patent EP 2 409 983A1, 2012. (j) Pando, M. O.; Wessjohann, L. A. Int. Patent WO 2012/010287A1, 2012. (k) Groaning, M.; Kleindl, P. J.; Leamon, C. P.; Ritter, A.; Santhapuram, H. K. R; Stanford, K. M.; Vlahov, I. R.; Wang, Y.; Xu, L. C.; You, F. Int. Patent WO 2012/019123A1, 2012. (1) Dömling, A.; Weber, L. Int. Patent WO 2004/005326A2, 2004. (m) Wipf, P.; Wang, Z. Int. Patent WO 2008/106080A2, 2008. (n) Vlahov, I. R.; Fei, Y.; Kleindl, P. J.; Vetzel, M.; Reddy, J. A.; Leamon, C. P. Int. Patent WO 2017/031209A1, 2017. (o) Yurkovetskiy, A.; Lowinger, T. B. Int. Patent WO 2014/ 160360A1, 2014. (p) Perez, H. L.; Wei, D.; Borzilleri, R. M.; Gangwar, S.; Schroeder, G. M.; Cheng, H.; Schmidt, R. J. U.S. Patent US 2016/ 0130299A1, 2016.
- (45) Friestad, G. K.; Banerjee, K.; Marie, J. C.; Mali, U.; Yao, L. J. Antibiot. 2016, 69, 294.
- (46) Colombo, R.; Wang, Z.; Han, J.; Balachandran, R.; Daghestani, H. N.; Camarco, D. P.; Vogt, A.; Day, B. W.; Mendel, D.; Wipf, P. J. Org. Chem. **2016**, *81*, 10302.
- (47) Leverett, C. A.; Sukuru, A. C. K.; Vetelino, B. C.; Musto, S.; Parris, K.; Pandit, J.; Loganzo, F.; Varghese, A. H.; Bai, G.; Liu, B.; Liu, D.; Hudson, S.; Doppalapudi, V. R.; Stock, J.; O'Donnell, C. J.; Subramanyam, C. ACS Med. Chem. Lett. **2016**, *7*, 999.
- (48) Sani, M.; Lazzari, P.; Folini, M.; Spiga, M.; Zuco, V.; De Cesare, M.; Manca, I.; Dall'Angelo, S.; Frigerio, M.; Usai, I.; Testa, A.; Zaffaroni, N.; Zanda, M. *Chem. Eur. J.* **2017**, *23*, 5842.
- (49) Park, Y.; Bae, S. Y.; Hah, J. M.; Lee, S. K.; Ryu, J. S. *Bioorg. Med. Chem.* **2015**, 23, 6827.
- (50) (a) Chari, R. V. J.; Miller, M. L.; Widdison, W. C. Angew. Chem., Int. Ed. 2014, 53, 3796. (b) Perez, H. L.; Cardarelli, P. M.; Deshpande, S.; Gangwar, S.; Schroeder, G. M.; Vite, G. D.; Borzilleri, R. M. Drug Discovery Today 2014, 19, 869. (c) Sievers, E. L.; Senter, P. D. Annu. Rev. Med. 2013, 64, 15.
- (51) Tumey, L. N.; Leverett, C. A.; Vetelino, B.; Li, F.; Rago, B.; Han, X.; Loganzo, F.; Musto, S.; Bai; Sukuru, S. C. K.; Graziani, E. I.; Puthenveetil, S.; Casavant, J.; Ratnayake, A.; Marquette, K.; Hudson, S.; Doppalapudi, V. R.; Stock, J.; Tchistiakova, L.; Bessire, A. J.; Clark, T.; Lucas, J.; Hosselet, C.; O'Donnell, C.; Subramanyam, C. ACS Med. Chem. Lett. **2016**, *7*, 977.
- (52) (a) Polu, K. R.; Lowman, H. B. *Expert Opin. Biol. Ther.* **2014**, *14*, 1049. (b) Desnoyers, L. R.; Vasiljeva, O.; Richardson, J. H.; Yang, A.; Menendez, E. E. M.; Liang, T. W.; Wong, C.; Bessette, P. H.; Kamath, K.; Moore, S. J.; Sagert, J. G.; Hostetter, D. R.; Han, F.; Gee, J.; Flandez, J.; Markham, K.; Nguyen, M.; Krimm, M.; Wong, K. R.; Liu, S.; Daugherty, P. S.; West, J. W.; Lowman, H. B. *Sci. Transl. Med.* **2013**, *5*, 207ra144. (c) Cohen, R.; Vugts, D. J.; Visser, G. W. M.; Stigter-van Walsum, M. S.; Bolijn, M.; Spiga, M.; Lazzari, P.; Shankar, P. S.; Sani, M.; Zanda, M.; van Dongen, G. A. M. S. *Cancer Res.* **2014**, *74*, 5700.
- (53) (a) Wang, Y.; Benz, F. W.; Wu, Y.; Wang, Q.; Chen, Y.; Chen, X.; Li, H.; Zhang, Y.; Zhang, R.; Yang, J. Mol. Pharmacol. 2016, 89, 233. (b) Zeino, M.; Zhao, Q.; Eichhorn, T.; Herrmann, J.; Müller, R.; Efferth, T. J. Biosci. Med. 2013, 3, 37. (c) Cormier, A.; Marchand, M.; Ravelli, R. B. G.; Knossow, M.; Gigant, B. EMBO Rep. 2008, 9, 1101. (54) (a) Sohtome, Y.; Shin, B.; Horitsugi, N.; Takagi, R.; Noguchi, K.; Nagasawa, K. Angew. Chem., Int. Ed. 2010, 49, 7299. (b) In, J. K.; Lee, M. S.; Lee, M. W.; Kwak, J. H.; Lee, H.; Hong, J. T.; Chung, Y.; Choi, Y.; Jung, J. K. Arch. Pharmacal Res. 2007, 30, 695.

(55) (a) Matcha, K.; Antonchick, A. P. Angew. Chem., Int. Ed. 2013, 52, 2082. (b) Khemnar, A. B.; Bhanage, B. M. Synlett 2014, 25, 110.
(c) Chatgilialoglu, C.; Crich, D.; Komatsu, M.; Ryu, I. Chem. Rev. 1999, 99, 1991. (d) Yeung, C. S.; Dong, V. M. Chem. Rev. 2011, 111, 1215.

(56) (a) Corey, E. J.; Bakshi, R. K.; Shibata, S. J. Am. Chem. Soc.
1987, 109, 5551. (b) Deloux, L.; Srebnik, M. Chem. Rev. 1993, 93, 763.
(c) Corey, E. J.; Helal, C. J. Angew. Chem., Int. Ed. 1998, 37, 1986.
(57) Nicolaou, K. C.; Estrada, A. A.; Zak, M.; Lee, S. H.; Safina, B. S. Angew. Chem., Int. Ed. 2005, 44, 1378.

(58) Altman, L. J.; Richheimer, S. L. *Tetrahedron Lett.* 1971, *12*, 4709.
(59) (a) Eberle, M. K.; Keese, R. *Helv. Chim. Acta* 2010, *93*, 1583.
(b) Eberle, M. K.; Keese, R.; Stoeckli-Evans, H. *Helv. Chim. Acta* 1998, *81*, 182. (c) Buckman, B.; Nicholas, J. B.; Serebryany, V.; Seiwert, S. D. U.S. Patent US 20110312996A1, 2011. (d) Lee, J. C.; Oh, Y. S.; Cho, S. H.; Lee, J. D. Org. Prep. Proced. Int. 1996, *28*, 480.

(60) (a) Nicolaou, K. C.; Vourloumis, D.; Totokotsopoulos, S.; Papakyriakou, A.; Karsunky, H.; Fernando, H.; Gavrilyuk, J.; Webb, D.; Stepan, A. F. ChemMedChem 2016, 11, 31. (b) Wlochal, J.; Davies, R. D. M.; Burton, J. Org. Lett. 2014, 16, 4094. (c) Falkiner, M. J.; Littler, S. W.; McRae, K. J.; Savage, G. P.; Tsanaktsidis, J. Org. Process Res. Dev. 2013, 17, 1503. (d) Ingalsbe, M. L.; St. Denis, J. D.; Gleason, J. L.; Savage, G. P.; Priefer, R. Synthesis 2010, 2010, 98. (e) Stepan, A. F.; Subramanyam, C.; Efremov, I. V.; Dutra, J. K.; O'Sullivan, T. J.; DiRico, K. J.; McDonald, W. S.; Won, A.; Dorff, P. H.; Nolan, C. E.; Becker, S. L.; Pustilnik, L. R.; Riddell, D. R.; Kauffman, G. W.; Kormos, B. L.; Zhang, L.; Lu, Y.; Capetta, S. H.; Green, M. E.; Karki, K.; Sibley, E.; Atchison, K. P.; Hallgren, A. J.; Oborski, C. E.; Robshaw, A. E.; Sneed, B.; O'Donnell, C. J. J. Med. Chem. 2012, 55, 3414. (f) Pätzel, M.; Sanktjohanser, M.; Doss, A.; Henklein, P.; Szeimies, G. Eur. J. Org. Chem. 2004, 2004, 493.

(61) (a) Barton, A.; Breukelman, S. P.; Kaye, P. T.; Meakins, G. D.; Morgan, D. J. *J. Chem. Soc., Perkin Trans.* 1 **1982**, 159. (b) Cui, Y. M.; Huang, Q. Q.; Xu, J.; Chen, L. L.; Li, J. Y.; Ye, Q. Z.; Li, J.; Nan, F. J. *Bioorg. Med. Chem. Lett.* **2005**, 15, 4130.

(62) (a) Kerr, M. S.; Read de Alaniz, J.; Rovis, T. J. Org. Chem. 2005, 70, 5725. (b) Smrcina, M.; Majer, P.; Majerová, E.; Guerassina, T.; Eissenstat, M. A. Tetrahedron 1997, 53, 12867. (c) Ullrich, A.; Herrmann, J.; Müller, R.; Kazmaier, U. Eur. J. Org. Chem. 2009, 2009, 6367. (d) Hin, B.; Majer, P.; Tsukamoto, T. J. Org. Chem. 2002, 67, 7365.

(63) (a) Schmidt, U.; Griesser, H.; Leitenberger, V.; Lieberknecht, A.; Mangold, R.; Meyer, R.; Riedl, B. Synthesis 1992, 1992, 487.
(b) Reetz, M. T.; Kayser, F. Tetrahedron: Asymmetry 1992, 3, 1377.
(c) Soroka, A.; Van der Veken, P.; De Meester, I.; Lambeir, A. M.; Maes, M. B.; Scharpé, S.; Haemers, A.; Augustyns, K. Bioorg. Med. Chem. Lett. 2006, 16, 4777.

(64) (a) Guillena, G.; Ramon, D. J.; Yus, M. Chem. Rev. 2010, 110, 1611. (b) Liu, X.; Hermange, P.; Ruiz, J.; Astruc, D. ChemCatChem 2016, 8, 1043.