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Efficient Synthesis of New β-Lactams

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EFFICIENT SYNTHESIS OF NEW β -LACTAMS .

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Abstract : The synthesis of β -lactams was carried out by N-arylpropionamide cyclisation in a mixture of N,N-dimethylformamide and anhydrous sodium carbonate under nitrogen atmosphere with 50 to 90% yield .

Synthesis of β -lactams is accessible through several methods under anhydrous conditions 1 . Some β -lactam syntheses were shown to be carried out from β -haloacylchlorides and α -aminoacid derivatives in a mixture of benzene and 40% sodium hydroxide with a phase transfer catalysis (KAY 2). Yamazaki 3 performed preparation of β -lactams from N-alkyl- β -halocarboxamides 4 by using potassium hydroxide as phase

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1a:
$$R = CO_2H$$
; $R_1 = CI$

1b:
$$R = H$$
; $R_1 = CO_2H$

1c:
$$R = H$$
; $R_1 = CONH(CH_2)_2 - CO_2H$

1d:
$$R = H$$
; $R_1 = CONH(CH_2)_5 - CO_2H$

1e:
$$R = H$$
; $R_1 = (CH_2)_3$ - CO_2H

Scheme 1

transfer. In the present paper , we report the preliminary results of an easy synthesis leading to new β -lactams without the use of phase transfer catalysis for cyclisation . Firstly , we perfected a rapid synthesis of propionamide derivatives 2a - 2e. That was achieved with 60 to 70% yields by condensing diversely substituted aniline 1a - 1e and 3-chloropivaloyl chloride under reflux in dioxane for 1 hour in presence of one pyridine equivalent . The benzhydryl derivatives 4a - 4b, in presence of 3,3'-di-chloropivaloyl chloride and through the same reacting conditions led to expected propionamide compounds 5a - 5b. These compounds

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CI—C—CH₂—CI

H

CH₂—CI

H

CH₂—CI

H

CH₂—CI

H

O CH₃

NH—C—C—CH₂—CI

CH₂—CI

Aa : R = H

5b : R = CI

Fig. 1 R = H

CH₂—CI

Ab : R = H

$$\frac{5}{4}$$
 $\frac{5}{4}$
 $\frac{5}{4}$

were cyclized in β -lactams 3a - 3c and 5a - 5b by heating at 140°C in dimethylformamide in presence of one sodium carbonate equivalent with 70 to 90% yields after the solution was acidified at pH = 4 by diluted hydrochloric acid. The 3-chloromethyl-3-methyl-N-benzhydryl-2-oxoazetidines 6a - 6b heated in dimethylformamide with benzylpiperazine and sodium carbonate, led to dihydrochlorides 7a - 7b with 70% yields, after acidification.

The structures of these derivatives agreed with their ¹H NMR spectra while the elemental analysis agreed with the theoretical values with an 0,30%

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accuracy. Because these N- and C-substituted β -lactams have shown antibacterial effects against Gram⁺ bacteria, these preliminary results have to be completed by new synthesises of β -lactam derivatives having various chains with intent to improve the antibacterial activity.

Experimental:

Melting points were determined on Kofler type WME apparatus and are uncorrected. IR. spectra were recorded on Philips P.U.spectrometer . $^1\text{H NMR}$ spectra were recorded on Varian E.M.390 spectrometer at 90 MHz in hexadeuteriodimethylsulfoxide with tetramethylsilane as an internal reference . Chemical Shifts are expressed as δ (ppm) relative to TMS . The 3-chloropivaloyl chloride and 3,3'-dichloropivaloyl chloride started from the commercially Aldrich Limited .

N-(4-chloro-3-carboxyphenyl)-3-chloro-2,2-dimethylpropionamide **2a** : General procedure

A solution of 10 g (0.05 mole) of 5-amino-2-chlorobenzoic acid $\underline{1a}$ and 8.99 g (0.058 mole) of 3-chloropivaloyl chloride in 120 mL dioxane was refluxed with 5 ml pyridine under nitrogen atmosphere for 1 hr. The solvent was evaporated in vacuo and the residue dissolved in water and extracted with ether. The solvent was removed to leave a solid 12.50 g (75%), m.p.: 54° C of $\underline{2a}$ (Crystallized from ether-hexane). The yields and conditions for the isolated products $\underline{1a} - \underline{1e}$, $\underline{5a} - \underline{5b}$ are summarized in the table 1. 1 H NMR (DMSO-d₆): δ HAr: 7.56 and 7.36; δ NH: 9.16; δ CH₂: 3.83; δ (CH₃)₂: 1.40.

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Table 1

Entry	Reaction	ir (KBr)	m.p.	Yield	¹ H NMR/TMS
	time (hr)	(vcm ⁻¹)	(℃)	(%)	(δ) ppm.
<u>2b</u>	1	3300(NH); 1685, 1660(CO)	227	60	$δ$ NH: 9.66(s,1H); $δ$ Ar: 7.93 (m,4H); $δ$ CH $_2$: 3.93(s,2H); $δ$ (CH $_3$) $_2$: 1.33(s,6H); $δ$ OH: 12.16 (m,1H) .
<u>2¢</u>	1	3300(NH); 1690, 1645, 1610(CO)	190	70	δ NH: 9.33(s,1H), 8.26(t,1H); δ Ar: 7.60(m,4H); δ CH ₂ : 3.70 (s,2H); δ (CH ₂) ₂ : 3.30, 2.40 (m,4H); δ (CH ₃) ₂ : 1.20(s,6H); δ OH: 11.49 (s,1H) .
<u>2d</u>	1	3360, 3280 (NH); 1700, 1625(CO)	113	65	δNH: 9.46(s,1H), 8.30(t,1H); $δ$ Ar: 7.76(m,4H); $δ$ CH ₂ : 3.86 (s,2H); $δ$ (CH ₂) ₅ : 3.26, 2.20, 1.56(m,10H); $δ$ (CH ₃) ₂ : 1.30 (s,6H); $δ$ OH: 11.62 (s,1H) .
<u>2e</u>	1	3340(NH); 1720, 1660(CO)	124	78	δ NH: 9.13(s,1H); δ Ar: 7.43,7.00 (d.d,4H); δ CH ₂ CI: 3.76(s,2H); δ (CH ₂) ₃ : 2.46, 2.13, 1.76 (m,6H); δ (CH ₃) ₂ : 1.23(s,6H)
<u>5a</u>	1	3310(NH); 1640(CO)	176	70	$\begin{split} &\delta \text{NH: 8.60(d,1H); } \delta (\text{C}_6\text{H}_5)_2; \\ &7.13(\text{m,10H); } \delta \text{CH: 6.10 (d,1H); } \\ &\delta (\text{CH}_2)_2; 3.83(\text{d,4H); } \\ &\delta \text{CH}_3; 1.26(\text{s,3H)} \; . \end{split}$
<u>5b</u>	1	3310(NH); 1630(CO)	142	75	δ NH: 8.30(d,1H); δ C ₆ H ₅ ; C ₆ H ₄ :7.13(m,9H); δ CH: 6.06(d,1H); δ (CH ₂) ₂ : 3.71(d,4H); δ CH ₃ : 1.16(s,3H).

N-(3,3-dimethyl-2-oxoazetidinyl) phenyl-para-butyric acid **3e** : General procedure

A mixture of 10 g (0.0336 mole) of 4-[p- (3-chloro-2,2-dimethylpropionamidophenyl)] butyric acid $\underline{2e}$, sodium carbonate 8.19 g (0.0772 mole) and 30 mL of dimethylformamide was heated at 160°C with stirring under nitrogen for 1 hr . Water (100 mL) was added , and the mixture acidified (pH = 4) by slow addition of concentrated HCl under cooling with an icewater bath . The aqueous solution was extracted with ether (150 mL) , and evaporated to dryness . The dry residue was washed twice with etherhexane , to give in $\underline{3e}$ (6.20 g ; 71%) as a white solid , m.p. : 66°C .

¹H NMR (DMSO-d₆): δHAr: 7.20 ; δCH₂: 3.48 ; δ(CH₂)₃: 2.56, 2.23, 1.86 ; δCH₃: 1.36 ; δOH: 12.00 .

The isolated products 3a - 3d , 6a - 6b are summarized in the table 2 .

3-methyl-2-(4'-benzyl-1'-methylpiperazinyl)-N-(benzhydrylamido)-2-oxoazetidine dihydrochloride 7a.

To a solution of 5 g (0.0167 mole) of 3-methyl-3'-chloromethyl-N-(benzhydryl-amino)-2-oxoazetidine <u>6a</u> in 35 mL dry N,N-dimethylformamide was added 1.77 g (0.0167 mole) of sodium carbonate and 2.93 g (0.0167 mole) of benzylpiperazine. The mixture was stirred at 160°C for 1 hr . The solution was poured into ice/water (150 mL) and extracted with ethylacetate (150 mL) . The extract were dried (Na₂SO₄) , the solvent was removed under reduced pressure . The oil 5.25 g (0.0119 mole , 72 %) was dissolved in isopropanol (30 mL) and (5 mL) hydrochloric acid and was converted to the dihydrochloride salt <u>7a</u> (4.20 g ; 69%) as a white solid ,

Table 2

Entry	Reaction time (hr)	ir (KBr) (υcm ⁻¹)	m.p. (℃)	Yield (%)	¹ Η NMR/TMS (δ) ppm.
<u>3a</u>	1	1725, 1690(CO)	226	70	δAr: 7.66, 7.43(m,3H); δCH ₂ : 3.50(s,2H); δ(CH ₃) ₂ : 1.33 (s,6H); δOH: 12.22 (s,1H) .
<u>3b</u>	1	1750, 1670(CO)	235	90	δAr: 7.90(d,2H), 7.33(d,2H); δCH ₂ : 3.53(s,2H); δ(CH ₃) ₂ : 1.33 (s,6H); δOH: 12.56(s,1H).
<u>3c</u>	1	3300(NH); 1740, 1680(CO)	170	87	δAr: 7.80(d,2H); 7.33(d,2H); δNH: 8.40(t,1H); δOH: 12.20 (s,1H); δCH ₂ : 3.50(s,2H); δ(CH ₂) ₂ : 3.30, 2.40(m,4H); δ(CH ₃) ₂ : 1.30 (s,6H).
<u>3d</u>	1	3300(NH); 1730, 1700(CO)	140	85	δNH: 8.33(t,1H); δOH: 12.00 (s,1H); δAr: 7.83(d,2H); 7.33 (d,2H);δCH ₂ : 3.50 (s,2H); δ(CH ₂) ₅ : 3.16, 2.13, 1.60 (m,10H); δ(CH ₃) ₂ : 1.26(s,6H)
<u>6a</u>	1	1750(CO)	122	92	δAr: 7.20(m,10H); δCH: 5.95 (s,1H); δCH ₂ : 3.73(s,2H); δCH ₂ CI: 3.21, 2.90(q.q,2H); δCH ₃ : 1.25(s,3H).
<u>6b</u>	1	1750(CO)	88	83	δAr: 7.20(m,9H); δCH: 5.96 (s,1H); δCH ₂ : 3.73(s,2H); δCH ₂ Cl: 3.20, 2.96(q.q,2H); δCH ₃ : 1.23(s,3H).

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m.p. : 165^{\circ}C . ^{1}H NMR (DMSO-d<sub>6</sub>) : \deltaNH+: 3.96 ; \deltaHAr : 7.50 , 7.20 ; \deltaCH : 5.88 ; \deltaCH<sub>3</sub> : 1.33 ; \deltaCH<sub>2</sub> : 4.28 , 3.40 and 3.26 . The yield and the reaction condition for obtaining the 3-methyl-3-(4'-ben-
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zyl-1'-methylpiperazinyl)-N-(4-chlorobenzhydrylamido)-2-oxoazetidine dihydrochloride 7b (6 g , 72% and m.p.: 170°C) werethe same as 7a.

¹H NMR (DMSO-d₆): δHNH⁺: 3.96; δHAr: 7.56, 7.23; δCH: 5.90; δCH₃: 1.33; δCH₂: 4.30, 3.40 and 3.26.

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