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Kinetic Resolution of Tertiary 2-Alkoxycarboxamido-allylic Alcohols by Chiral Phosphoric Acid Catalyzed Intramolecular Transesterification

Subramani Rajkumar[‡], Shunlong He[‡] and Xiaoyu Yang*

Abstract: A highly enantioselective kinetic resolution of tertiary 2alkoxycarboxamido-allylic alcohols has been achieved through chiral phosphoric acid catalyzed intramolecular transesterification reaction. Both alkyl,aryl- and dialkyl-substituted tertiary allylic alcohols can be well-resolved by the optimal conditions with excellent efficiencies, affording both recovered tertiary alcohols and transesterification products with high enantioselectivities (with s factor up to 164.6). The gram-scale reaction with 1 mol % catalyst loading and facile conversion of chiral products into useful chiral building blocks, such as chiral oxazolidinones and β -amino alcohols, well demonstrated the value of this reaction.

The kinetic resolution of racemic secondary alcohols has been extensively investigated in the past few decades and demonstrated as one of the most efficient and practical approaches to obtain chiral secondary alcohols¹. In contrast, the corresponding kinetic resolution of tertiary alcohols has been less studied and only a handful of catalytic examples have been reported, including both biocatalytic processes² and nonenzymatic methods³. However, due to the relative challenging demands for spatial differentiation among three R groups^{3a, 3h, 3i} and the intrinsic reaction requirements^{3b, 3c}, kinetic resolution of tertiary alcohols with high efficiencies that can provide both unreacted starting material and products with high enantioselectivities are still guite limited. Among them, List and co-workers reported chiral Brønsted acid catalyzed kinetic resolution of homoaldols and diols possessing tertiary hydroxyl groups via intra- and intermolecular acetalizations with excellent performance, which could generate both tertiary alcohol starting materials and acetal products with high enantioselectivities^{3d, 3f} Zhao group^{3e} and Smith group^{3g} reported highly efficient kinetic resolution of tertiary heterocyclic alcohols via asymmetric acylation reactions enabled by NHC catalyst and isothiourea catalyst, respectively. Recently, Oestreich and co-workers developed kinetic resolution of tertiary progargylic alcohols by enantioselective Cu-H-catalyzed Si-O coupling with a broad substrate scope^{3j}.

Recently, the combination of 2-amidoallylic alcohol substrates (could also be termed as α -hydroxy enamides) with chiral Brønsted acid catalysis has become a useful strategy for asymmetric synthesis of functionalized enamide derivatives⁴. Our group⁵ and Kartika group⁶ independently reported the asymmetric synthesis of β -indolyl cyclopentenamides from α -hydroxy cyclopentenamides upon chiral phosphoric acid

catalysis (Figure 1, a). On the other hand, Chan group reported dehydrative Nazarov-type electrocyclization of a-hydroxy enamides to provide access to chiral 1H-indene derivatives under chiral Brønsted acid catalysis⁷ (Figure 1, b). In these examples, the chiral anion-paired 2-amidoallyl cations (INT A) were proposed as the key intermediates, in which the chiral center from the starting material was eliminated. During our continuous study on asymmetric reactions of a-hydroxy enamides under chiral Brønsted acid catalysis, we recently found that no 2-amidoallyl cation intermediate was generated by treatment of a-hydroxy encarbamates 1 with phosphoric acid catalyst; instead an intramolecular transesterification reaction⁸ proceeded smoothly to give the oxazolidinone derivatives 2 (Figure 1, c). Herein, we reported a highly efficient kinetic resolution of tertiary 2-alkoxycarboxamido-allylic alcohols (ahydroxy encarbamates) by utilizing this reaction under the catalysis of chiral phosphoric acid catalyst⁹. A range of racemic tertiary 2-alkoxycarboxamido-allylic alcohols (including both aryl,alkyl- and alkyl,alkyl-substituted allylic alcohols) could be well resolved by the optimal conditions, generating both transesterification products and unreacted tertiary alcohols with high enantioselectivities. The obtained chiral products could be readily converted into useful chiral building blocks, such as oxazolidinones¹⁰ and β -amino alcohols¹¹, which have wide applications as chiral auxiliaries, chiral ligands and biologically active molecules.



Previous work: asymmetric reactions of α-hydroxy enamides catalyzed by chiral Bronsted acid catalysts



useful chiral building blocks

Figure 1. a) Asymmetric addition of indoles to α -hydroxy cyclopentenamides under chiral Brønsted acid catalysis. b) Asymmetric Nazarov-type electrocyclization of α -hydroxy enamides under chiral Brønsted acid catalysis. c) Kinetic resolution of tertiary 2-alkoxycarboxamido-allylic alcohols via chiral phosphoric acid catalyzed intramolecular transesterification.

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Table 1. Optimization of reaction conditions.^a



^aReactions were run with 1 (0.1 mmol) with catalyst (0.01 mmol, 10 mol %) in solvent (1 mL) with 5 Å molecular sieves (30 mg). ^bConversion (C) = ee_s/(ee_s+ee_p). ^cDetermined by HPLC analysis on a chiral stationary phase. ^ds = ln[(1-C)(1-ee_p)]/ln[(1-C)(1+ee_p)].

We initiated our studies by using racemic *N*-Boc protected tertiary 2-alkoxycarboxamido-allylic alcohol **1a**' as model substrate and chiral phosphoric acid as catalyst. While using 10 mol % of TRIP catalyst (cat **A1**) in toluene at ambient temperature, the kinetic resolution of tertiary alcohol **1a**' proceeded with a promising selectivity factor (s)¹² of 7.2, giving the transesterification product **2a** with 60% enantiomeric excess (ee) and recovered starting material **1a**' with 60% ee (Table 1, entry 1). Subsequently, a variety of BINOL- and H8-BINOL-derived chiral phosphoric acid catalysts were examined (entries 2-7), and the TCYP catalyst (cat **A6**) turned out to be the optimal catalyst, giving the best *s* factor of 26.6 (entry 6). Next, a range of solvents were also explored (entries 8-11), and PhCI was found to be the optimal one, in which the *s* factor could be further improved to 37.5 (entry 11). The leaving groups on the

carbamate moiety were also investigated, and we observed that replacing of the tBu group by either Et and Bn substituents led to faster reaction rate, however, with poor kinetic resolution performance as well (entries 12-13). Interestingly, using an iPr group instead of tBu group provided an improved s factor of 41.8 and also reduced reaction time (entry 14). With the faster reaction rate found for substrate **1a** at ambient temperature, we further decreased the reaction temperature to 10 °C and 0 °C, and found the s factor could be further improved to 63.7 and 78.4 respectively, albeit with longer reaction time (entries 15-16).

With the optimal conditions established, we next sought to investigate the substrate scope of this kinetic resolution reaction. A range of different substitutions at the 3-position of allylic alcohols were firstly evaluated (Table 2). A variety of substituted phenyl groups (bearing different substitutions at the *para-, meta-*

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and ortho-positions) were well tolerated under the optimal conditions, providing the recovered 2-alkoxycarboxamido-allylic alcohols 1 and oxazolidinone derivatives 2 with high enantioselectivities (1a-1i), as well as a 2-naphthyl group (1j, s factor = 133.4). The absolute configurations of the products were determined as (R) by analogy to product 2c, whose structure was unambiguously confirmed by X-ray crystallography¹³. A linear alkyl group at the terminal alkene carbon of the allylic alcohol substrate was also compatible with the optimal conditions, albeit providing the products and recovered SM with slightly diminished enantioselectivities (1k).

Table 2. Substrate scope regarding to the 3-position of 2-alkoxycarboxamidoallylic alcohols.^a

HO Ph rac-1	(S)-cat A6 (10 mol %) PhCl, 5 A MS, temperature	HO, Ph (S)-1	Pr O + O Me ^w	-NH R h (<i>R</i>)-2
IPrOOCHN Ho Ph (S)-1a, 47%, 98% ee (R)-2a, 45%, 89% ee s = 78.4	iPrOOCHN HO Ph (S)-1c, 469 (R)-2c, 509 s = 63	6, 89% ee 6, 91% ee 3.5	X-Ray stru	cture of 2c
iPrOOCHN HO Ph	iPrOOCHN HO Ph	OMe	iPrOOCHN HO Ph	CF3
(S) -1b , 49%, 92% ee (<i>R</i>)- 2b , 48%, 93% ee <i>s</i> = 94.1	^d (S)- 1d , 47%, 95% ee (<i>R</i>)- 2d , 48%, 94% <i>s</i> = 120.6		(S)-1e, 50%, 85% ee (R)-2e, 46%, 85% ee s =33.2	
iProochN HO Ph	iPrOOCHN HO Ph	OMe	iPrOOCHN HO Ph	F
(S)-1f, 48%, 92% ee (R)-2f, 48%, 85% ee s = 40.2	(S) -1g , 48%, 8 (R)- 2g , 49%, 9 s = 61.8	8% ee 1% ee	(S) -1h , 48%, (<i>R</i>)- 2h , 51%, s = 47.0	92% ee 87% ee
iPrOOCHN HO Ph CI	iPrOOCHN HO Ph		HO Ph	OiPr
^d (S)- 1i , 47%, 87% ee (<i>R</i>)- 2i , 47%, 78 % ee	(S) -1j , 43%, 93 (<i>R</i>)- 2j , 48%, 95	9% ee 5% ee	^e (S) -1k , 44%, (<i>R</i>)- 2k , 45%,	80% ee 77% ee

^aUnless otherwise noted, reactions were performed with 0.2 mmol rac-1, 0.01 mmol (S)-cat A6 catalyst and 60 mg 5 Å MS in 2 ml PhCl at 0 $^\circ\text{C}.$ The reactions were monitored by HPLC using a chiral stationary phase. ^bThe yields are isolated yields. ^cThe ee values were determined by HPLC analysis on a chiral stationary phase. ^dThese reactions were performed at 10 °C. ^eThis reaction was performed at 25 °C.

18.6 s =

s = 133.4

s = 22.6

With the well compatibility of various substitutions at the 3position of the 2-alkoxycarboxamido-allylic alcohol substrates, we moved our attention to the scope of tertiary alcohol moiety of the substrates (Table 3). Replacing the phenyl group of the tertiary alcohols in 1a with a range of substituted phenyl groups (with various different substitutions at the para-, meta- and ortho-positions) gave highly efficient kinetic resolution of the 2alkoxycarboxamido-allylic alcohol substrates (11-1r). lt's interesting to find that an electron-donating -OMe substitution at different sites of the phenyl groups all gave excellent kinetic resolution performance (1n, 1p and 1r), with s factors > 98. Other aryl and heteroaryl tertiary alcohols were also well tolerated with the optimal conditions (1s and 1t). Other alkyl groups than methyl group were also evaluated. The Et,Phsubstituted substrate (1u) still gave high selectivity factor, while the iPr,Ph-substituted substrate (1v) gave a decreased s factor, probably due to the small size differentiation between these two groups. Tertiary alcohols with two alkyl groups were also investigated under the standard conditions and a range of Me,alkyl-substituted 2-amidoallylic alcohol substrates provided excellent performance under the optimal kinetic resolution conditions. The Me, Bu-substituted allylic alcohol gave the best s factor of 164.6 (1w), affording both recovered alcohol and cyclized product with > 95% ee. The Me,iPr- and Me,Cysubstituted allylic alcohols were also amenable with the optimal kinetic resolution conditions (1x and 1y). It's surprising to find

Table 3. Substrate scope regarding to the tertiary alcohol moiety of 2alkoxycarboxamido-allylic alcohols.



performed at 10 °C. °These reactions were performed at 25 °C.

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that primary-alkyl,Me-disubstituted allylic alcohols could also be well-resolved by these conditions (**1z** and **1aa**), providing a rare access for the asymmetric synthesis of these tertiary alcohols.

To evaluate the practicability of this kinetic resolution reaction, gram-scale reaction of rac-1a was conducted (Scheme 1, a). While performing this reaction at ambient temperature, the catalyst loading of (R)-cat A6 could be further reduced to 1 mol %, still providing comparable s factor value and enantioselectivities for (R)-1a and (S)-2a. The substrate scope of this kinetic resolution reaction was further extended to secondary 2-alkoxycarboxamido-allylic alcohol substrate (Scheme 1, b). With racemic secondary alcohol 3a used as substrate, the kinetic resolution with the same conditions proceeded smoothly to provide (S)-3a in 33% isolated yield with 85% ee without optimization, with selectivity factor of 10. However. under these standard conditions, the transesterification product underwent further isomerization to give 2-oxazolone 4a as product, with the chiral center eliminated.

Gram-scale reaction using 1 mol % catalyst:



Kinetic resolution of secondary 2-amidoallylic alcohol:



Scheme 1. Gram-scale reaction and kinetic resolution of secondary 2alkoxycarboxamido-allylic alcohol.

To demonstrate the utility of this reaction, the derivatizations of the chiral products were also studied. Base promoted cyclization of (*S*)-**1a** readily yielded (*S*)-oxazolidinone **2a**, thus providing an enantiodivergent route for the asymmetric synthesis of both enantiomers of **2a** (Scheme 2, a). Hydrogenation of oxazolidinone **2a** with Raney Ni as catalyst afforded oxazolidinone **5a** in 80% yield with > 25:1 dr without erosion of the enantioselectivity, which could be used as chiral auxiliaries in asymmetric synthesis¹⁰. The relative configuration of compound **5a** was determined by NOE analysis¹³. Further hydrolysis of **5a** with aqueous NaOH solution cleanly removed the protecting group, affording free β -amino alcohol **6a** in high yield (Scheme 2, b).





Scheme 2. Derivatizations of the chiral products.

In conclusion, we have disclosed a highly efficient kinetic resolution of tertiary 2-alkoxycarboxamido-allylic alcohols via intramolecular transesterification reaction under the catalysis of chiral phosphoric acid catalyst. A rang of different tertiary alcohols, including aryl,alkyl- and alkyl,alkyl-disubstituted 2-alkoxycarboxamido-allylic alcohols are well-compatible with the optimal kinetic resolution conditions, providing access to both unreacted tertiary alcohols and cyclized oxazolidinone products with high enantioselectivities. The gram-scale reaction with reduced catalyst loading and facile transformation of the chiral products into useful chiral building blocks well demonstrated the potential of this method.

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Conflict of interest

The authors declare no conflict of interest.

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Subramani Rajkumar, Shunlong He and Xiaoyu Yang*

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by Chiral Phosphoric Acid Catalyzed Intramolecular Transesterification

A highly enantioselective kinetic resolution of tertiary 2-alkoxycarboxamido-allylic alcohols has been achieved through chiral phosphoric acid catalyzed intramolecular transesterification reaction. Both alkyl,aryl- and dialkyl-substituted tertiary allylic alcohols can be resolved with excellent efficiencies, with *s* factor up to 164.6. This method affords expedient access to enantioenriched oxazolidinones and β -amino alcohols.