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carbocyclic framework of the stealthins was synthesized.

Nitromethyl benzoate annulation reactions: a rapid construction of the stealthin skeleton

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ABSTRACT

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Quinones can be conveniently and regioselectively prepared using annulation reactions with phthalide 1 or ester 2 and suitable Michael acceptors.^{1–3} As part of a program to enhance natural antioxidants,⁴ we needed a flexible synthesis of the stealthin skeleton shown in Figure 1. Stealthin A (3) and stealthin C (4) are members of a class of tetracyclic natural products isolated from Streptomyces viridochromogenes.⁶ Stealthin A inhibited peroxylipid formation at IC₅₀ of 0.04 mg/mL versus 10.8 mg/mL for vitamin E.⁷ Syntheses of these compounds were reported by Snieckus and others.⁸ Although a cyclopentenone or an indenone would be the logical Michael acceptor, annulation reactions involving cyclopentenones and indenones can proceed in low yields due to their sensitivity to strongly basic conditions. Based on reports that nitroalkanes undergo efficient Michael addition to cyclopentenones,⁵ we envisaged a direct route to compounds such as **3** or **4** beginning from indenones. Our route generates the naphthol more efficiently than previous approaches.

Ester **5** was readily prepared in 32% yield from the methyl ester of ortho-toluic acid by bromination followed by displacement of the bromide with silver nitrite in ether.⁹ As shown in Scheme 1, ester **5** was treated with cyclopentenone in acetonitrile to form the Michael adduct which cyclized using NaOMe in THF in 54% yield. Conversion to phenol **6** proceeded smoothly using DDQ in toluene or air in DMSO.¹⁰

The successful annulation with cyclopentenone prompted us to examine indenone and other Michael acceptors. The reaction of **5**

with indenone, dicyclopentadienone,¹¹ tetrahydroindenone, and phenyl vinyl ketone afforded phenols **7**, **8**, **9**, and **10** in yields of 64%, 47%, 27%, and 22%, respectively, as shown in Figure 2. The melting point of **10** matched that reported in the literature.¹²

The annulation of 2-(nitromethyl)benzoates with enones gave 4-nitro-1-naphthols in good yields. The

Our strategy for assembling the stealthin skeleton, depicted below in Scheme 2, begins with ester **11**, readily available from ethyl acetoacetate.¹³ Benzylic bromination using NBS and AIBN in benzene produced the benzylic bromide in 75% yield. Treatment with silver nitrite in ether produced ester **12** in 50% yield.

Indenone **14** was prepared from 6-methyldihydrocoumarin (**13**) as shown in Scheme 3. An intramolecular Friedel–Crafts acylation generated the indanone in 71% yield. The phenol was protected as a benzenesulfonate in 87% yield. Treatment with HIO_3 in DMSO¹⁴ afforded indenone **14** in 87% yield.

As shown in Scheme 4, the reaction of **12** and **14** with DBU in acetonitrile produced a Michael addition product in 72% yield. The product was treated at -78 °C with 2.2 equiv of lithium diisopropylamide (LDA) to generate tetracyclic ketone **15** in 23% yield.











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Scheme 1. Reagents and conditions: (a) Bu₄NOH, MeOH 74%; (b) NaOMe, THF 54%; (c) air, DMSO 90%.



Figure 2. Adducts of 5 with enones.



Scheme 2. Reagents and conditions: (a) NBS, AlBN, benzene reflux, 75%; (b) AgNO₂, ether 0 °C–rt, 50%.



Scheme 3. Reagents and conditions: (a) AlCl₃, NaCl, 200 °C, 71%; (b) Et₃N, PhSO₂Cl, THF, rt, 87%; (c) HIO₃, DMSO 50 °C, 87%.



Scheme 4. Reagents and conditions: (a) DBU, CH_2Cl_2 , 0 °C, 72%; (b) 2.2 equiv LDA, THF, -78-0 °C, 23%.

In summary, the Michael addition/Claisen condensation sequence affords good overall yields of tricyclic and tetracyclic adducts.¹⁵ It provides an effective route to tetracyclic stealthin analogs. This pathway will permit the synthesis of an array of analogs similar to **15** for biological testing.

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References and notes

- (a) Kraus, G. A.; Cho, H.; Crowley, S.; Roth, B.; Sugimoto, H.; Prugh, S. J. Org. Chem. **1983**, 48, 3439–3444; (b) Kraus, G. A.; Sugimoto, H. Tetrahedron Lett. **1978**, 19, 2263–2266; (c) Kraus, G. A.; Sugimoto, H. Synth. Commun. **1977**, 7, 505–508; (d) Hauser, F. M.; Rhee, R. P. J. Am. Chem. Soc. **1979**, 101, 1628–1629.
- (a) Hauser, F. M.; Yin, H. Org. Lett. 2000, 2, 1045–1047; (b) Ray, S.; Patra, A.; Mal, D. Tetrahedron 2008, 64, 3253–3267.
- (a) Huang, X.; Xue, J. J. Org. Chem. 2007, 72, 3965–3968; (b) Wildeman, J.; Borgen, P. C.; Pluim, H.; Rouwette, P. H. F. M.; Van Leusen, A. M. Tetrahedron Lett. 1978, 19, 2213–2216.
- Kraus, G. A.; Kumar, G.; Phillips, G.; Michalson, K.; Mangano, M. Bioorg. Med. Chem. Lett. 2008, 18, 2329–2332.
- (a) Barr, L.; Easton, C. J.; Lee, K.; Lincoln, S. F. Org. Biomol. Chem. 2005, 3, 2990–2993; (b) Knobloch, K.; Koch, J.; Keller, M.; Eberbach, W. Eur. J. Org. Chem. 2005, 2715–2733; (c) Wang, L.; Meegalla, S. K.; Fang, C.-L.; Taylor, N.; Rodrigo, R. Can. J. Chem. 2002, 80, 728–738; (d) Hanessian, S.; Shao, Z.; Warrier, J. S. Org. Lett. 2006, 8, 4787–4790.
- Shinya, K.; Furihata, K.; Teshima, Y.; Hayakawa, Y.; Seto, H. Tetrahedron Lett. 1992, 33, 7025–7028.
- Seto H.; Hayakawa Y.; Shimazu A. (Kirin Brewery, Japan). Jpn. Kokai Tokkyo Koho, 12 pp. JP5085998, 1993; *Chem. Abstr.* 1993, 119:70550.
- (a) Koyama, H.; Kamikawa, T. J. Chem. Soc., Perkin Trans. 1 1998, 203–210; (b) Mohri, S.; Stefinovic, M.; Snieckus, V. J. Org. Chem. 1997, 62(21), 7072–7073; (c) Qabaja, G.; Jones, G. B. J. Org. Chem. 2000, 65, 7187–7194.
- (a) Hellwinkel, D.; Bohnet, S. Chem. Ber. **1987**, 120, 1151–1173; (b) Gandler, J. R.; Saunders, O. L.; Barbosa, R. J. Org. Chem. **1997**, 62, 4677–4682.
- 10. Danikiewicz, W.; Makosza, M. Tetrahedron Lett. 1985, 26, 3599-3600.
- 11. Firouzabadi, H.; Sharifi, A. Synthesis 1992, 10, 999-1002.
- 12. Compound 10: Ettlinger, M. G. J. Am. Chem. Soc. 1950, 72, 3666-3672.
- 13. Hauser, F. M.; Pogany, S. A. Synthesis 1980, 814-815.
- Nicolaou, K. C.; Montagnon, T.; Baran, P. S. Angew. Chem., Int. Ed. 2002, 41, 1386.
 Representative experimental:
- Representative synthesis of compound 7

To a solution of nitro ester 5 (0.196 g, 1.0 mmol) and 1H-inden-1-one (0.130 g, 1.0 mmol) in acetonitrile (3 mL) was added triton B solution in methanol (0.085 mL, 40% w/w, 0.2 equiv) at 0 °C. The temperature was allowed to reach rt and the reaction was monitored by TLC. Then the reaction was quenched with saturated NH₄Cl solution and extracted with EtOAc. Purification on column chromatography on silica gel using EtOAc/hexane as eluent gave 0.221 g of the open-chain intermediate in 68% yield. Intermediate: mixture of two diastereoisomers in a ratio of 3:2. Major isomer: ¹H NMR (CDCl₃, 400 MHz): 8.06 (d, J = 8.0 Hz, 1H), 7.91 (d, J = 8.0 Hz, 1H), 7.77 (m, 1H), 7.71 (m, 2H), 7.56 (m, 2H), 7.39 (m, H), 7.07 (d, J = 8.0 Hz, 1H), 4.42 (m, 1H), 3.73 (s, 3H), 2.81 (m, 2H), 2.42 (m, 1H), ¹³C NMR (CDCl₃, 100 MHz):203.7, 166.7, 151.7, 157.8 (m, 2H), 2.42 (m, 2H), 2.44 (m, 3H), 2.51 (III, 2H), 2.42 (III, 111). CHAIR (CDC), 165 (III), 252.57 (165, 1724.57), 137.8, 135.6, 134.8, 134.2, 133.4, 131.8, 130.2, 129.8, 127.6, 126.0, 124.2, 88.7, 63.0, 42.6, 40.9. Minor isomer: ¹H NMR (CDCl₃, 400 MHz): 7.99 (d, *J* = 8.0 Hz, 1H), 7.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, *J* = 8.0 Hz, 1H), 7.61 (m, 2H), 7.50 (m, 2H), 7.38 (m, 1H), 6.79 (d, J), 6.70 (m, 2H), 7.50 (m, J = 10.8 Hz, 1H), 6.66 (m, 1H), 4.63 (m, 1H), 3.82 (s, 3H), 2.80 (m, 2H), 2.41 (m, 1H). ¹³C NMR (CD₃COCD₃, 100 MHz): 203.2, 167.2, 152.9, 137.1, 135.0, 134.5, 134.0, 133.4, 131.4, 130.4, 129.3, 127.6, 126.1, 124.5, 88.4, 52.8, 42.0, 40.0. To a solution of intermediate (0.174 g, 0.535 mmol) in THF (20 mL) was added a solution of sodium methoxide in methanol (0.238 mL, 4.5 M, 2 equiv) at 0 °C, and it was stirred at rt. After the disappearance of the starting material on TLC, it was quenched with ice, followed by a saturated $\rm NH_4Cl$ solution. The crude product was extracted with EtOAc, dried over MgSO4, and the solvent was removed under vacuo. Then the crude product was dissolved in toluene (2 mL) and 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (0.123 g, 0.554 mmol. 1 equiv) was added in portion at 0 °C. The reaction was stirred at rt. After the reaction was finished ice was added and it was extracted with EtOAc and purification on column chromatography on silica gel gave 0.100 gram of compound 7 in 64% yield. Compound 7: ¹H NMR (CD₃COCD₃, 300 MHz): 8.25 (t, J = 7.8 Hz, 2H), 8.01 (d, J = 7.8 Hz, 1H), 7.70 (m, 2H), 7.62 (m, 2H), 7.31 (t, J = 7.8 Hz, 1H). (EI) m/z Calculated for $C_{17}H_9NO_4$: 291.0532, Found: 291.0536. Compound 6: ¹H NMR (CD₃COCD₃, 400 MHz): 8.46 (m, 2H), 7.96 (t, J = 8.0 Hz, 1H), 7.75 (t, J = 8.0 Hz, 1H), 3.58 (m, 2H), 2.58 (m, 2H). HRMS (EI) m/z Calculated for C₁₃H₉NO₄: 243.0532; Found: 243.0535. Compound 8: ¹H NMR (CDCl₃/CD₃COCD₃, 400 MHz): 8.40 (d, J = 8.4 Hz, 1H), 8.33 (d, J = 8.4 Hz, 1H), 7.81 (t, J = 8.0 Hz, 1H), 7.63 (t, J = 8.0 Hz, 1H), 5.98 (m, 1H), 5.61 (m, 1H), 4.44 (m, 1H), 3.43 (m, 2H), 2.68 (m, 1H), 1.80 (m, 2H). Compound 9: ¹H NMR (CDCl₃/CD₃COCD₃, 400 MHz): 8.38 (d, J = 8.0 Hz, 1H), 8.33 (d, J = 8.0 Hz, 1H), 7.76 (t, J = 7.6 Hz, 1H), 7.56 (t, J = 7.6 Hz, 1H), 5.85 (m, 1H), 5.71 (m, 1H), 4.23 (m, 1H), 3.09 (m, 1H), 2.59 (m, 2H), 2.42 (m, 2H). HRMS

(EI) m/z Calculated for $C_{17}H_{13}NO_4$:295.0845; Found: 295.0848. 10-hydroxy-2-methyl-11-oxo-11H-benzo[b][fluorene-4, 9-diyl dibenzenesulfonate

(15). To a stirred solution of ethyl 2-(nitromethyl)-6-((phenylsulfonyl) oxy)benzoate, 12 (968 mg, 2.65 mmol) and 6-methyl-1-oxo-1H-inden-4-yl benzenesulfonate, 14 (398 mg, 1.33 mmol) in 10 mL acetonitrile was added

DBU (0.4 ml, 2.65 mmol) at 0 °C and stirred at rt overnight, where it was worked up with 5% aqueous HCl and extracted with ethyl acetate three times. The combined organic extracts were washed with brine, dried over MgSO₄, and evaporated in vacuo. Purification via silica gel chromatography (40% ethyl acetate in hexanes) yielded the desired Michael adduct (637 mg, 72% yield). To a stirred solution of freshly prepared LDA (2.2 equiv) in THF at -78 °C was added a solution of the Michael adduct (405 mg, 0.61 mmol) in dry THF (10 mL). The reaction mixture was allowed to warm to rt, where it was stirred for 3 h. The reaction mixture was then worked-up with aqueous NH₄Cl and extracted with ethyl acetate three times. The combined organic extracts were

washed with brine, dried over $MgSO_4$, and evaporated in vacuo. Purification via silica gel chromatography (30% ethyl acetate in hexanes) yielded the desired cyclized product as a yellow compound (86 mg, 25% yield).

since get circulatographic (50% ethyl acctate in fickalies) yielded the desired cyclized product as a yellow compound (86 mg, 25% yield). ¹H NMR (300 MHz, CDCl₃): δ 7.94 (d, *J* = 7.2 Hz, 4H), 7.66–7.62 (m, 2H), 7.53–7.51 (m, 6H), 7.46–7.43 (m, 2H), 7.32 (s, 1H), 7.11 (d, *J* = 8.1 Hz, 1H), 7.09 (s, 1H), 2.38 (s, 3H). ¹³C NMR (300 MHz, CDCl₃): δ 21.62, 113.64, 116.39, 119.76, 121.49, 123.67, 128.61, 128.94, 129.22, 129.60, 130.13, 132.73, 134.47, 134.98, 136.19, 138.52, 141.35, 142.03, 145.56, 147.72, 156.65, 194.28.

HRMS (EI) m/z exact mass calculated for $\rm C_{30}H_{21}O_8S_2~[M+H]^*:$ 573.0672. Found 573.0677.