A Modular Synthesis of Salvileucalin B Structural Domains

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A modular strategy leading to the salvileucalin B core structure has been accomplished. The developed synthetic strategy featured a bioinspired intramolecular Diels—Alder reaction to construct domain "A", REDOX chemistry to functionalize domain "B", and a palladium-mediated cross-coupling to install domain "C". This flexible approach should facilitate further chemical and biological investigations of this fascinating class of compounds.

Architecturally complex natural products continue to serve as a testing ground for state-of-the-art synthetic organic chemistry and as a powerful vehicle for the invention of novel synthetic technologies and strategies. In conjunction with the biological potential possessed by these novel molecular entities, their value in advancing chemistry and biology cannot be understated.¹ Among the recent isolates, salvileucalins A and B (1 and 3, respectively, Scheme 1) were obtained from the aerial parts of Salvia leucantha Cav. and characterized by their neoclerodane skeleton after extensive NMR spectroscopic and X-ray crystallographic analysis.² The minor of the two isolates, salvileucalin B (3), possesses a highly unusual norcaradiene core which has been hypothesized to originate biosynthetically from salvileucalin A (1) via an enzymatic oxidation and followed by a subsequent intramolecular

Diels–Alder reaction $(1 \rightarrow 2 \rightarrow 3)$, Scheme 1).² Furthermore, salvileucalin B (3) was found to exhibit cytotoxic activities against A549 (human lung adenocarcinoma) and HT-29 (human colon adenocarcinoma) cells.² The unprecedented molecular architecture of salvileucalin B (3) and its biological potential has sparked interest from the scientific community,³ with the first total synthesis of (+)salvileucalin B elegantly accomplished by the Reisman group in early 2011.^{3b} Here, we report the realization of our objective to develop a modular strategy to access potentially both the natural and designed salvileucalins.

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In order to fulfill our ultimate objective of systematically investigating the chemistry and biology of salvileucalin B (3), and potentially accessing the naturally occurring substance itself [i.e., (3)], we opted for a strategy that retrosynthetically divided the global structure of salvileucalin B (3) into three domains (Scheme 1, A, B, and C). As outlined in Scheme 1, we envisaged a transition-metal-mediated

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Scheme 1. a



^{*a*} Molecular structure of salvileucalin A (1), proposed biosynthetic intermediate 2, and salvileucalin B (3) indicating the three key structural domains A, B, and C. Retrosynthetic analysis of salvileucalin B (3) leading to stannane 4 and synthetic intermediates 5, 6, and 7. TBS = *tert*-butyldimethylsilyl.

cross-coupling reaction to install subdomain C [for example, a Stille coupling reaction⁴ engaging stannane 4 to install the furanyl moiety of salvileucalin B (3)]. Upon executing this opening retrosynthetic maneuver, our attention immediately turned to the most challenging aspect of the foreseeable synthetic campaign, namely the construction of subdomain A containing the norcaradiene core of salvileucalin B (3). Inspired by the biosynthetic proposal, we were intrigued with the possibility of a carefully designed triene system (e.g., $\mathbf{6}$) to undergo an intramolecular Diels–Alder reaction,⁵ thereby furnishing the tetracyclic core structure represented by advanced intermediate 5. Indeed, several examples of an intramolecular Diels-Alder reaction of 5-vinyl-1,3-cyclohexadiene have been documented.⁶ Furthermore, a recent quantum mechanical calculation study has unveiled the influence of functional groups present in both the substrate and the possible enzyme active site that could accelerate the intramolecular Diels-Alder reaction leading to the formation of the norcaradiene core of salvileucalin B (3).⁷ Finally, the spirocyclic motif within hydroxy methyl ester 6 could be conceived from a Conia-ene reaction⁸ engaging alkynyl β -diketone 7 (represented in its enol form).

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Scheme 2. Synthesis of Pentacyclic Hydroxy Lactone 24^a



 a LDA = lithium diisopropyl amide, DMP = Dess-Martin periodinane, Dibal-H = diisobutylaluminium hydride, KHMDS = potassium hexamethyldisilazide, Tf = trifluoromethanesulfonyl, OTf = trifluoromethanesulfonate, py = pyridine, HMDS = hexamethyldisilazane.

The realization of our synthetic strategy began with the preparation of the proposed intramolecular Diels-Alder precursor 6, and thereby accessing the key synthetic intermediate 5 harboring subdomain A, as shown in Scheme 2. Thus, a lithium aldol reaction (LDA) between cyclohexenone (9) and alkynyl aldehyde 8 afforded β -hydroxy ketone 10, which was further oxidized to diketone 7 (represented in its enol form) under DMP conditions in 75% overall yield. A Conia-ene reaction⁸ engaging alkynyl diketone 7 was examined under a variety of reaction conditions, and ultimately ZnI₂ was revealed as the most effective promoter for this transformation, giving spirocyclic diketone 11 in 85% yield. Chemo- and stereoselective reduction of keto enone 11 under the Dibal-H conditions led to hydroxy enone 12 as a single diastereoisomer, which was subsequently protected as its TBS ether (TBSOTf, 13, 85% yield over the two steps). In preparation for the proposed intramolecular Diels-Alder reaction, enone 13 was converted to methyl ester 15 through the intermediacy of triflate 14 (KHMDS, PhNTf₂, 76% yield) and its reaction under the standard carboxymethylation

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conditions [Pd(PPh₃)₄, CO, MeOH, 82% yield]. While TBS ether **15** failed to undergo the much anticipated intramolecular Diels–Alder reaction, gratifyingly, its desilylated version (**6**, HF•py, 100% yield) underwent the desired transformation to deliver tetracycle **5** in 79% yield. After extensive experimentation, this reaction was best carried out through iterative microwave heating⁹ with HMDS as an additive in 1,2-dichlorobenzene. The structural validity of the Diels–Alder product **5** was confirmed through an X-ray crystallographic analysis of its enone derivative (**5a**),¹⁰ as shown in Figure 1.¹¹



Figure 1. X-ray derived ORTEP representation of 5a.¹¹

With a synthetic route successfully charted for accessing domain A and a sufficient quantity of the crucial building block 5 secured, we began to investigate the chemistry associated with the structural domain B [see Scheme 1]. As such, we first assembled the α -methylene γ -lactone moiety required for salvileucalin B (3). Thus, a two-step procedure involving silvl protection of the secondary alcohol 5 (TBSOTf) and reduction of the so-obtained methyl ester 16 (Dibal-H) furnished homoallylic alcohol 17 in 92% overall yield. A hydroxyl-directed hydroboration of 17 was envisaged for the functionalization at C₄. In this instance, while the reaction proceeded with excellent conversion (95% yield), only a modest level of regioselectivity in favor of the desired regioisomer 19 (19:18 ca. 1.4:1, chromatographic separable) was observed. Next, a four-step transformation smoothly delivered the one-carbon homologated α , β -unsaturated methyl ester 23 from diol 19, through the intermediacy of bis-TBS ether 20 (TBSCl-Et₃N, 85% yield), ketone 21 (DMP, 80% yield), enol triflate 22 (KHMDS, PhNTf₂, 89% yield), and reaction of the latter compound under the carboxymethylation conditions [Pd(PPh₃)₄, CO, MeOH, 78% yield]. Finally, casting of the γ -lactone moiety was effected through global desilvlation of the bis-TBS ether 23 (HF•py), to afford hydroxy lactone 24 in 90% yield. With the construction of the α -methylene γ -lactone moiety for salvileucalin B (3) successfully secured, we should mention that intermediates

5 and **16–24** are all valuable building blocks in our ensuing salvileucalin studies.¹²

Now, we turned our attention to the final missing piece of the puzzle, namely the assembly of structural domain C (see Scheme 1). In accordance with our original proposal (Scheme 1), we first attempted to introduce the furanyl moiety onto the growing pentacyclic γ -lactone **24** through a Stille⁴ coupling reaction engaging enol derivative **27** and stannane **4**.¹³ However, after extensive experimentation, neither the preparation of enol derivative **27** nor its Stille coupling with stannane **4** could be accomplished (Scheme 3).

Scheme 3. Initial Attempts in the Installation of the Furanyl Moiety through Stille Coupling Engaging Enol Derivative 27 and Stannane 4 a^{a}



The unsuccessful union between enol derivatives 27 and stannane 4^{13} generally resulted in the production of an unidentified byproduct where the alkenyl γ -lactone moiety was either absent or significantly compromised. Thus, a revised strategy that involved the introduction of the furanyl moiety in the absence of the $C_4/C_5 \gamma$ -lactone was pursued in earnest (Scheme 4). In this instance, the readily accessible Diels-Alder product 5 was conveniently transformed to β -keto ester **30** through oxidation (DMP, 82%) vield) and α -acylation (LHMDS, Mander's reagent¹⁴). Conversion of the so-obtained β -keto ester **30** to enol triflate 31 (Comin's reagent, NaH, 81% yield from 29) proceeded smoothly, where the latter compound (31) underwent palladium-catalyzed Stille⁴ coupling with stannane 4^{13} to afford furanyl derivative **32** in a pleasing 80% yield as a mixture of diastereoisomers (ca. 2:1) epimeric at C_{12} . Finally, liberation of the TBS guarded secondary alcohol in 32 (HF•py) resulted in spontaneous lactonization to furnish the hexacyclic, γ -lactone containing methyl ester 33.

⁽⁹⁾ For experimental details, see Supporting Information.

⁽¹⁰⁾ For preparation of enone 5a from Diels–Alder product 5, see Supporting Information.

⁽¹¹⁾ CCDC-793468 contains the supplementary crystallographic data for **5a**. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

⁽¹²⁾ The synthesis and biological evaluation of natural and designed salvileucalin compounds will be reported in the full account of this work.

⁽¹³⁾ For preparation of furanyl stannane **4**, see Supporting Information.

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The successful preparation of triflate **31** and its facile participation under the palladium-mediated cross-coupling reactions (e.g., Stille coupling)⁴ should enable a variety of furan substitutes to be introduced.

In order to examine the synthetic feasibility of the furanyl bearing γ -lactone 33 in accessing domain B analogues of salvileucalin B (3), particularly in view of the potential liability of the furanyl and γ -lactone functionalities, further synthetic transformations were carried out as shown in Scheme 4. In this context, γ -lactone containing methyl ester 33 was subjected to a two-step reduction (Dibal-H)/selective-oxidation (TEMPO, PIDA) sequence to afford primary alcohols 34 and 12-epi-34 (ca. 2:1, chromatographically separated) in 72% combined yield (from 32). In an analogous fashion to the preparation of ketone 21 (Scheme 2), chemoselective hydroboration of homoallylic alcohol 34 followed by oxidative workup afforded a chromatographically separable mixture of diols 35 and 36 (35:36 ca. 1:3.5 dr), where selective silvlation of the latter compound (36) was achieved through the action of TBSCl and imidazole. Final oxidation of the hydroxy TBS ether 37 afforded ketone 38, an intermediate that should be amenable for both the preparation of domain **B** analogues or salvileucalin B (3) through the late-stage chemistry reported by Reisman^{3b} and in accordance to Scheme 2.

In summary, a modular synthetic strategy with flexibility in addressing the salvileucalin B structural domains A, B, and C, as outlined in Scheme 1, has been demonstrated. The developed synthetic sequence featured a ZnI₂-mediated Conia–ene spirocyclization, a bioinspired intramolecular Diels–Alder reaction, and a palladium-mediated crosscoupling reaction. Furthermore, key intermediate 12 has been prepared in its optically active form through an analogous ZnI₂-mediated Conia–ene spirocyclization reaction.¹⁵ The synthetic technology and strategy disclosed herein illustrated an entry to the entire carbocyclic framework salvileucalin B, which is therefore path-determining for the preparation of both designed and natural substances for further chemical and biological investigations.¹² Scheme 4. Synthesis of Ketone 38^a



 a DME = ethylene glycol dimethyl ether; NMP = *N*-methyl-2pyrrolidone; TEMPO = 2,2,6,6-tetramethyl-1-piperidinyloxy; PIDA = phenyliodine diacetate.

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Supporting Information Available. Experimental procedures, compound characterization (PDF), and cif file for compound **5a**. This material is available free of charge via Internet at http://pubs.acs.org.

⁽¹⁵⁾ An asymmetric synthesis of spirocyclic hydroxy enone **12** has been accomplished. For details, see Supporting Information.