Synthesis of Some Novel *N*-Alkyl/Acyl/Aroyl 2-(Chroman/6-bromochroman-2-yl)-1*H*-benzimidazoles Using Ionic Liquids and Their Antibacterial Activity

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Synthesis of some novel N-substituted 2-(chroman/6-bromochroman-2-yl)-1H-benzimidazoles by the condensation of 3,4-dihydro-2H-chroman-2-carboxylic acid and 6-bromo-3,4-dihydro-2H-chroman-2-carboxylic acid with o-phenylenediamine in ionic liquid (IL) [bmim]BF₄ and subsequent reactions at the benzimidazole-NH with different types of electrophiles in ILs [bmim]BF₄ = 1-butyl-3-methylimidazolium tetrafluoroborate, [bmim]PF₆ = 1-butyl-3-methylimidazolium hexafluorophosphate and [buPy]BF₄ = butylpyridinium tetrafluoroborate in the presence of sodium hydroxide as a base have been reported. All the synthesized compounds were screened for their antibacterial activity. Some compounds exhibited promising antibacterial activity against Staphylococcus aureus and Salmonella typhimurium when compared to Cephalexin as a reference standard.

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INTRODUCTION

There are a vast number of pharmacologically active heterocyclic compounds, many of which are in regular clinical use. Of the wide variety of heterocyclic systems known till date, the nitrogen heterocycles are of great importance and benzimidazole is one amongst such important heterocycles because of its synthetic utility and broad spectrum of pharmacological activity [1-10]. Various substituted benzimidazoles are known to have varied biological activities and among these, 2-substituted benzimidazoles are found to be more potent [11]. The biological activities of benzimidazoles containing compounds have been well documented [12-13]. Despite their wide applicability, available routes for their synthesis are limited. The reported synthesis of benzimidazoles included reactions of aryl acid with o-phenylenediamine (OPDA) in conventional [14,15], microwave-assisted [16], and ionic liquids (ILs) [17] methods. The N-alkylation and acylation of benzimidazoles has been reported to be accomplished by treatment with an appropriate base such as sodium hydride, sodium hydroxide, potassium carbonate, pyridine, etc. followed by reaction of the resulting salt with an alkylating reagent in various solvents, e.g. acetone, acetonitrile, pyridine, DMF, THF, etc. [18–20].

In the rapidly developing field of synthetic organic chemistry, an efficient, simple, and highly selective synthetic method for widely used organic compounds from readily available reagents is one of the major challenges. ILs are proving to be increasingly promising as viable media not only for potentially 'green' synthesis and separations, but also for novel applications. The unique property set of the IL materials provides new options based on different chemical and physical properties. The room temperature ILs are of special interest as 'green' recyclable alternative to the classical molecular solvents in the synthetic organic chemistry [21-24]. ILs are the best choice for N-alkylation of heterocyclic compounds bearing an acidic hydrogen attached to nitrogen. The reports on great improvement in the yields and rates of reaction using ILs [25] prompted us to study the Nalkylation and acylation of benzimidazoles in ILs. Hence it was thought that it would be worthwhile to design and synthesize the N-substituted benzimidazoles in ILs under environment-friendly conditions and screen them for potential biological activity.

Scheme 1. Reagents, i) Diethyl Oxalate, NaOEt, aq.HCl ii) AcOH, H_2 , Pd/C, 175psi iii) $[bmim]BF_4$, $100^{\circ}C$ iv) Ethanolic-HCl v) AcOH, Br_2 vi) EtOH/NaOH vii) $[bmim]BF_4$, $100^{\circ}C$

Scheme 2

i) IL, NaOH, R-X ii) IL, NaOH, R'-X

RESULTS AND DISCUSSION

Compound 3 was synthesized according to literature procedure in good yield [26]. Condensation of 3,4-dihydro-2H-chroman-2-carboxylic acid (3) with OPDA (4) was carried out in IL [bmim]BF₄ at 100° C for 6 h to obtain compound 5 in excellent yield (scheme 1).

For the synthesis of compound **3c**, 3,4-dihydro-2*H*-chroman-2-carboxylic acid (**3**) was treated with ethanolic-HCl at 85°C for 3 h to afford the corresponding ethyl ester derivative **3a** in a reasonable yield. Compound **3a** was brominated using bromine in glacial acetic acid at 25°C for 2 h to obtain 6-bromo-3,4-dihydro-2*H*-chroman-2-carboxylic acid ethyl ester (**3b**), which

Sr. No	Substrate	R/R'	Ionic liquid	Reaction conditions	Product	Yield (%)	Mp (°C)
1	5	CH ₃ -O-CO-	[bmim]PF ₆	50°C, 3 h	6a	75	132–133
2	5	C ₂ H ₅ —O—CO—	[bupy]BF ₄	50°C, 3 h	6b	71	105-107
3	5	(CH ₃) ₂ —CH—CH ₂ —O—CO—	[bmim]BF ₄	50°C, 3 h	6c	77	90-92
4	5	C ₆ H ₅ —O—CO—	[bmim]PF ₆	50°C, 3 h	6d	85	120-121
5	5	CH ₃ —SO ₂ —	[bmim]BF ₄	60°C, 2 h	6e	82	190-192
6	5	(p) — CH_3 — C_6H_4 — SO_2 —	[buPy]BF ₄	60°C, 2 h	6f	73	172-173
7	5	C ₆ H ₅ —CH ₂ —	[bmim]BF ₄	75°C, 5 h	6g	80	128-130
8	5	$(p)F-C_6H_4-CH_2-$	[buPy]BF ₄	75°C, 6 h	6h	75	120-122
9	5	(p) Br $-C_6H_4-CH_2-$	[bmim]PF ₆	75°C, 5 h	6i	76	155-157
10	5	$(p)CH_3-C_6H_4-CH_2-$	[bmim]BF ₄	75°C, 6 h	6 j	82	176-178
11	5	(p) Tert.butyl— C_6H_4 — CH_2 —	[bmim]BF ₄	75°C, 5 h	6k	68	150-151
12	5a	CH ₃	[bmim]BF ₄	50°C, 5 h	6'a	70	138-140
13	5a	C_2H_5	[bmim]BF ₄	50°C, 5 h	6′b	80	125-127
14	5a	CH ₃ -O-CO-	[bmim]PF ₆	50°C, 3 h	6'c	73	140-141
15	5a	C ₂ H ₅ —O—CO—	[buPy]BF ₄	50°C, 3 h	6'd	78	148-150
16	5a	Isobutyl—O—CO—	[bmim]BF ₄	50°C, 3 h	6′e	75	120-121
17	5a	CH ₃ —SO ₂ —	[bmim]PF ₆	60°C, 2 h	6′f	89	115-116
18	5a	$(p) CH_3 - C_6H_4 - SO_2 -$	[bupy]BF ₄	60°C, 2 h	6'g	86	140-141
19	5a	C ₆ H ₅ —CH ₂ —	[bmim]BF ₄	75°C, 5 h	6'h	78	180-182
20	5a	$(p) \text{ FC}_6 \text{H}_4 \text{CH}_2 $	[buPy]BF ₄	75°C, 6 h	6'i	85	210-212
21	5a	(p) Br—C ₆ H ₄ —CH ₂ —	[bmim]PF ₆	75°C, 5 h	6′j	80	178-180
22	5a	(p) CH ₃ —C ₆ H ₄ —CH ₂ —	[bmim]BF ₄	75°C, 6 h	6′k	84	225-226
23	5a	(p) Tert.butyl— C_6H_4 — CH_2 —	[bmim]BF ₄	75°C, 5 h	6′1	78	166-167
24	5a	C ₆ H ₅ —O—CO—	[bmim]BF ₄	50°C, 3 h	6′m	82	170–172

 $\label{eq:compound} \textbf{Table 1b}$ Recycling of [bmim]BF4 for the compound 6g.

Number of cycles	Yield (%)
1	78
2	75
3	71

was further hydrolyzed in ethanol, sodium hydroxide, and 5N HCl to give 6-bromo-3,4-dihydro-2H-chroman-2-carboxylic acid (3c) as a brownish solid. Condensation of 3c with 4 was carried out in IL [bmim]BF₄ at 100° C for 6 h yielded compound 5a as a white solid.

The *N*-alkylation and acylation of **5** and **5a** with various electrophilic reagents in ILs to obtain the *N*-alkylated/acylated derivatives **6a-6k** and **6'a-6'm** (Scheme 2). The recovered IL was reused successfully with only a slight loss in yield (Table 1b). The structures of newly synthesized compounds were confirmed by IR, ¹H NMR, ¹³C NMR, and mass spectrometry. The physical and spectral data of the compounds **6a-6k** and **6'a-6'm** is presented in experimental section. The synthesized compounds were tested for their antimicrobial activity against two different bacterial species namely, *Staphy*-

Table 2

Antibacterial activity of synthesized compounds 5, 5a, 6a–6k, and 6'a–6'm against *S. aureus* NCIM 5021.

Commound	Concentration (µg/mL)						
Compound No.	0.1 1		10	100	200	500	App. MIC
5	++	++	+	Р	_	_	10
6a	++	+	+	P	_	_	10
6b	++	++	++	+	_	_	100
6c	++	++	+	+	_	_	10
6d	++	++	+	P	_	_	10
6e	++	++	++	+	_	_	100
6f	++	++	++	+	_	_	100
6g	++	++	+	P	_	_	10
6h	++	++	++	+	P	_	100
6i	++	++	+	+	_	_	100
6 j	++	++	+	P	_	_	10
6k	++	++	++	+	_	_	100
5a	++	++	+	+	P	_	10
6'a	++	++	++	+	P	_	100
6′b	++	++	++	+	P	_	100
6'c	++	++	++	+	P	_	100
6'd	++	++	++	+	P	_	100
6'e	++	++	++	+	P	_	100
6′f	++	++	+	+	P	_	10
6'g	++	+	+	P	P	_	1
6'h	++	++	+	+	P	_	10
6'i	++	++	+	+	P	_	10
6′j	++	+	P	_	_	_	1
6′k	++	++	++	+	P	_	100
6′l	++	++	+	+	P	_	10
6′m	++	+	+	+	P	_	1
Cephalexin	+	_	_	_	_	_	0.1

Table 3

Antibacterial activity of synthesized compounds 5, 5a, 6a–6k, and 6'a–6'm against *S. typhimurium* NCIM 2501.

Compound	Concentration (µg/mL)						
Compound No.	0.1	1	10	100	200	500	App. MIC
5	++	++	+	P	_	_	10
6a	++	++	+	+	_	_	10
6b	++	++	++	+	_	P	100
6c	++	++	++	+	_	_	100
6d	++	++	++	+	_	_	100
6e	++	++	++	+	_	_	100
6f	++	++	++	+	_	_	100
6g	++	++	+	P	_	_	10
6h	++	++	++	+	_	_	100
6i	++	++	++	+	_	_	100
6 j	++	++	++	+	_	_	100
6k	++	++	+	P	_	_	10
5a	++	++	++	++	+	_	200
6'a	++	++	++	+	_	_	100
6′b	++	++	++	+	_	_	100
6'c	++	++	++	++	+	_	200
6′d	++	++	++	+	_	_	100
6'e	++	++	++	+	P	_	100
6'f	++	++	+	P	_	_	10
$6'\mathbf{g}$	++	++	++	+	_	_	100
6'h	++	++	++	+	P	_	100
6′i	++	++	++	++	+	_	200
6′j	++	++	++	+	P	_	100
6'k	++	++	++	+	P	_	100
6′1	++	++	++	+	P	_	100
6′m	++	++	++	+	P	_	100
Cephalexin	+	_	_	_	_	_	0.1

⁻, Total inhibition, no growth of organism; P, Poor growth compared to controls; +, Medium growth compared to controls; ++, Confluent growth, no inhibition.

lococcus aureus NCIM 5021 and Salmonella typhimurium NCIM 2501.

BIOLOGICAL ACTIVITY

All the compounds prepared herein were screened for their antibacterial activity against *Staphylococcus aureus* NCIM 5021 (Gram positive) and *Salmonella typhimurium* NCIM 2501 (Gram negative) bacterial strains. Cephalexin was used as a reference standard. Antibacterial activity result of compounds 5, 5a, 6a-6k, and 6'a-6'm is summarized in Table 2 and 3. Some of the compounds found to have good antibacterial activity against *S. aureus*; however, they were found to have less activity against *S. typhimurium* when compared to Cephalexin as a reference standard.

CONCLUSION

In conclusion, we have successfully synthesized a novel series of *N*-substituted 2-(chroman/6-bromochroman-2-yl)-1*H*-benzimidazole derivatives by the

condensation of 3,4-dihydro-2*H*-chroman-2-carboxylic acid and 6-bromo-3,4-dihydro-2*H*-chroman-2-carboxylic acid with OPDA and subsequent reactions at the benzimidazole-NH with different electrophilic reagents under different reaction conditions in ILs and tested for antibacterial activity. Some of the compounds **6'g**, **6'j**, **6'm** showed the most potent inhibition at 1 µg/mL, where as compounds **5**, **6a**–**6d**, **6g**, **6j** and **5a**, **6'f**, **6'h**, **6'i**, **6'l** were found to possess good activity at 10 µg/mL against *S. aureus* and compounds **5**, **6a**, **6g**, **6k**, **6'f** showed the good activity at 10 µg/mL, where as other compounds showed minimal activity against a *S. typhimurium*.

EXPERIMENTAL

All the solvents were of commercial grade and OPDA, alkylating and acylating agents were obtained from Aldrich. Melting points were recorded on a MRVIS series, Lab India Instrument and are uncorrected. IR spectra were recorded on a Perkin-Elmer Spectrum One FTIR spectrometer in potassium bromide pellets unless otherwise stated. ¹H and ¹³C NMR spectra were recorded on a Varian Mercury VX SWBB 300 MHz spectrometer. Elemental analysis was carried out on a PerkinElmer Series-II C H N S O Analyzer 2400. Chemical shifts are reported in ppm from internal tetramethylsilane (TMS) standard and are given δ units. The solvent for NMR spectra was CDCl3 unless otherwise mentioned. Mass spectra were recorded on hp 1100 LC/MSD mass spectrometer (positive and negative APCI ion source, 50-200 V, nitrogen). ILs [bmim]BF₄, [bmim]PF₆, and [buPy]BF₄ were synthesized in the laboratory according to reported procedures [27].

Synthesis of 6-bromo-3,4-dihydro-2*H*-chroman-2-carboxylic acid (3c). A solution of (34 mmol) of 3a dissolved in glacial acetic acid (50 mL) was cooled to 10–15°C. The compound 3a was brominated by slowly adding a solution of bromine (33 mmol) in glacial acetic acid (25 mL). After the addition was complete, the solution was stirred at 25–30°C for 3 h. The reaction mass was then diluted with water (100 mL). The product was isolated by extraction with ethyl acetate (50 mL × 3). The ethyl acetate layer was washed with 5% aqueous sodium bicarbonate solution (50 mL × 2) and water (50 mL × 2). The ethyl acetate layer was dried over anhydrous sodium sulfate and solvent was removed under reduced pressure to obtain 6-bromo-3,4-dihydro-2*H*-chroman-2-carboxylic acid ethyl ester (3b).

A solution of (67 mmol) of **3b** dissolved in ethanol (25 mL) was hydrolysed by 0.5M sodium hydroxide solution (25 mL) at $25-30^{\circ}$ C for 1 h. The solution was concentrated to about its half volume and acidified with 5N hydrochloric acid. The resultant solid was filtered to give **3c**, 4.5 g, Yield 85%; mp $170-172^{\circ}$ C; ¹H NMR (CDCl₃) (300 MHz): δ 2.13–2.25 (m, 2H, CH₂), 2.29–2.37 (m, 1H, CH), 2.72–2.91 (m, 2H, CH₂), 4.79 (dd, 1H, J = 7.8, 7.8 Hz, CH), 6.81 (d, 1H, J = 9 Hz, Ar-H), 7.19–7.26 (m, 2H, Ar-H), 11.63 (s, 1H, COOH).

Synthesis of 2-(chroman-2-yl)-1H-benzimidazole using IL (5). A solution of (10 mmol) of 3 in IL [bmim]BF₄ (2 mL) and (12 mmol) of OPDA (4) was heated at 100°C for 6 h (as monitored by TLC). After 6 h the mixture was cooled to 25°C

and diluted with water (10 mL). The product was extracted with ethyl acetate (10 mL \times 2). The ethyl acetate layer was washed with 5N HCl (10 mL), 5% aqueous sodium bicarbonate solution (10 mL) and water (10 mL). The ethyl acetate layer was dried over sodium sulfate and evaporated under reduced pressure to afford crude product, which was recrystalized from ethyl acetate to obtain as a white solid compound 5. After isolation of the product in ethyl acetate IL was in aqueous layer, which was further washed with ethyl acetate (10 mL) and dried under vacuum. The suspension was filtered to remove insoluble and the recovered IL was recycled.

2-(Chroman-2-yl)-1*H***-benzimidazole** (5). mp 225–227°C; ir (KBr): 3434, 2950, 1487, 1231 cm⁻¹; ¹H NMR (CDCl₃): δ 2.20–2.35 (m, 2H, CH₂), 2.81–3.04 (m, 2H, CH₂), 5.47 (dd, J = 8.6, 8.9 Hz, 1H, CH), 6.87–6.91 (m, 4H, Ar-H), 7.08–7.18 (m, 2H, Ar-H), 7.46 (d, J = 7.4 Hz, 1H, Ar-H), 7.71 (d, J = 7.6 Hz, 1H, Ar-H), 9.7 (bs, 1H, NH); ms: m/z 249.1 (M⁺+1); Anal. Calcd. for C₁₆H₁₄N₂O: C, 76.78; H, 5.64; N, 11.19. Found: C, 76.52; H, 5.52; N, 11.36.

Synthesis of 2-(6-bromochroman-2-yl)-IH-benzimidazole using IL (5a). A mixture of 3c (10 mmol), OPDA (4) (12 mmol) and IL [bmim]BF₄ (2 mL) was heated to 100° C for 6 h (as monitored by TLC). The reaction mixture was then cooled to room temperature and diluted with water (10 mL). The product was extracted with ethyl acetate (10 mL \times 2) and ethyl acetate layer was washed with 5N HCl (10 mL), 5% aqueous sodium bicarbonate solution (10 mL) and water (10 mL). The ethyl acetate layer was dried over sodium sulfate and solvent was evaporated under reduced pressure to obtain crude product, which was recrystallized from ethyl acetate to yield the pure compound 5a. The IL was recovered by the procedure described earlier.

2-(6-Bromochroman-2-yl)-*1H***-benzimidazole (5a).** mp 210–211°C; ir (KBr): 3428, 2919, 1476, 1232 cm⁻¹; ¹H NMR (DMSO-d₆): δ 2.26–2.38 (m, 2H, CH₂), 2.86–3.01 (m, 2H, CH₂), 5.46 (dd, J = 8.7, 8.7 Hz, 1H, CH), 6.86 (d, J = 9 Hz, 1H, Ar-H), 7.15–7.34 (m, 4H, Ar-H), 7.48 (d, J = 6.9 Hz, 1H, Ar-H), 7.62 (d, J = 6.9 Hz, 1H, Ar-H) 12.62 (bs, 1H, NH) ppm; ¹³C NMR (DMSO-d₆) (75 MHz): δ 23.36 (CH₂), 25.60 (CH₂), 72.25 (OCH), 111.77, 112.11, 119,01, 119.24, 121.64, 122.63, 125.01, 130.18, 132.18, 134.43, 142.95, 153.02, 153.23 (aromatic carbons) ppm; ms: m/z 329.32 (M⁺+1); Anal. Calcd. for C₁₆H₁₃BrN₂O: C, 58.38; H, 3.98; N, 8.51. Found: C, 58.26; H, 3.75; N, 8.66.

General procedure for the synthesis of compound 6'a-6'b using IL. A solution (2 mmol) of 5a, IL (2 mL), sodium hydroxide (4 mmol) and followed by the addition of respective alkylting reagents (3 mmol) at 25–30°C. After the addition was complete, the solution was heated to 50° C for 5 h (as monitored by TLC). The reaction mixture was then cooled to 25° C and diluted with water (10 mL). The product was extracted with ethyl acetate (10 mL \times 2) and ethyl acetate layer was washed with water (10 mL \times 2). The organic layer was dried over sodium sulfate and the solvent was evaporated under reduced pressure to obtain the corresponding *N*-substituted derivatives. The crude products were recrystallized from ethanol to give pure compounds 6'a and 6'b, respectively. The IL was recovered by the procedure described earlier.

Methyl 2-(6-bromochroman-2-yl)-1H-benzimidazole (6'a). ir (KBr): 3435, 2948, 1474, 1231 cm⁻¹; 1H NMR (DMSO-d₆): δ 2.40–2.46 (m, 2H, CH₂), 2.97–3.01 (m, 2H, CH₂), 3.91

(s, 3H, CH₃), 5.63 (dd, J=8.1, 8.4 Hz, 1H, CH), 6.81(d, J=8.4 Hz, 1H, Ar-H), 7.28–7.35 (m, 4H, Ar-H), 7.59–7.66 (m, 2H, Ar-H) ppm; 13 C NMR (DMSO-d₆) (75 MHz): δ 23.07 (CH₂), 23.59 (CH₂), 29.41 (NCH₃), 70.10 (CH), 108.11, 111.84, 117.24, 118.84, 121.01, 121.89, 122.08, 128.93, 130.95, 134.91, 140.60, 149.96, 151.55 (aromatic carbons) ppm; ms: m/z 343.61 (M⁺+1); Anal. Calcd. for C₁₇H₁₅BrN₂O: C, 59.49; H, 4.41; N, 8.56. Found: C, 59.38; H, 4.62; N, 8.32.

Ethyl 2-(6-bromochroman-2-yl)-1H-brenzimidazole (6'b). ir (KBr): 3435, 2983, 1473, 1234 cm⁻¹; ¹H NMR (DMSO-d₆): δ 1.4 (t, J = 6.9 Hz, 3H, CH₃), 2.42–2.50 (m, 2H, CH₂), 3.03–3.07 (m, 2H, CH₂), 4.41 (q, J = 6.8 Hz, 2H, CH₂), 5.62 (dd, J = 8.1, 8.3 Hz, 1H, CH), 6.78 (d, J = 9 Hz, 1H, Ar-H), 7.24–7.36 (m, 4H, Ar-H), 7.61–7.67 (m, 2H, Ar-H) ppm; ¹³C NMR (DMSO-d₆) (75 MHz): δ 15.25 (CH₃), 24.25 (CH₂), 24.86 (CH₂), 39.23 (NCH₂), 71.24 (CH), 109.61, 113.08, 118.51, 120.24, 122.23, 123.12, 124.08, 130.22, 132.27, 135.11, 142.18, 150.80, 152.93 (aromatic carbons) ppm; ms: m/z 357.66 (M⁺+1); Anal. Calcd. for C₁₈H₁₇BrN₂O: C, 60.52; H, 4.80; N, 7.84. Found: C, 60.38; H, 4.97; N, 7.63.

General procedure for the synthesis of compound (6a-6f, 6'c-6'g, and 6'm) using ILs. To a solution of 5 and 5a (2 mmol) in ILs (2 mL), pyridine (10 mmol), followed by the addition of appropriate acyl or arylsulfonyl chloride (3 mmol) were stirred (for reaction conditions, Table 1a). The reaction mixture was then cooled to 25°C and diluted with water (10 mL). The product was extracted with ethyl acetate (10 mL × 2), ethyl acetate layer was washed with 5% aqueous sodium bicarbonate solution (10 mL) followed by washed with water (10 mL). The ethyl acetate layer was dried over sodium sulfate and the solvent was evaporated to obtain the crude products, which were recrystallized from ethanol, to afford pure compounds (6a-6f), (6'c-6'g and 6'm), respectively. The ILs was recovered by the procedure described earlier.

Methyl 2-(chroman-2-yl)-1H-benzimidazole-1-carboxylate (6a). ir (KBr): 3453, 2922, 1759, 1455 cm $^{-1}$; 1 H NMR (CDCl₃): δ 2.40–2.51 (m, 2H, CH₂), 2.92–3.07 (m, 2H, CH₂), 4.13 (s, 3H, CH₃), 5.86 (dd, J=8.4, 8.6 Hz, 1H, CH), 6.87–6.94 (m, 2H, Ar-H), 7.12 (d, J=6.9 Hz, 2H, Ar-H), 7.37–7.38 (m, 2H, Ar-H), 7.82 (d, J=6.9, Hz, 1H, Ar-H), 7.95 (d, J=7.2 Hz, 1H, Ar-H); ms: m/z 309.1 (M $^{+}$ +1); Anal. Calcd. for C₁₈H₁₆N₂O₃: C, 70.12; H, 5.23; N, 9.09. Found: C, 70.23; H, 5.35; N, 9.25.

Ethyl 2-(chroman-2-yl)-1H-benzimidazole-1-carboxylate (6b). ir (KBr): 3453, 2972, 1744, 1455 cm $^{-1}$; 1 H NMR (CDCl $_{3}$): δ 1.53 (t, J=7.2 Hz, 3H, CH $_{3}$), 2.40–2.51 (m, 2H, CH $_{2}$), 2.92–3.06 (m, 2H, CH $_{2}$), 4.57 (q, 2H, CH $_{2}$), 5.89 (dd, J=8.7 Hz, 1H, CH), 6.87–6.95 (m, 2H, Ar-H), 7.1 (d, J=6.9 Hz, 2H, Ar-H), 7.36–7.39 (m, 2H, Ar-H), 7.82 (d, J=8.1 Hz, 1H, Ar-H), 7.98 (d, 1H, J=6 Hz, Ar-H); ms: m/z 323.1 (M $^{+}$ +1); Anal. Calcd. for C $_{19}$ H $_{18}$ N $_{2}$ O $_{3}$: C, 70.79; H, 5.63; N, 8.69. Found: C, 70.92, H, 5.48; N, 8.85.

Isobutyl 2-(chroman-2-yl)-1H-benzimidazole-1-carboxylate (6c). ir (KBr): 3272, 1738, 1456 cm⁻¹; ¹H NMR (CDCl₃): δ 1.57 (d, J=6.6 Hz, 6H, CH₃), 2.16–2.19 (m, 1H, CH), 2.44–2.52 (m, 2H, CH₂), 2.91–3.06 (m, 2H, CH₂), 4.32 (d, J=6.6 Hz, 2H, CH2), 5.91 (dd, J=8.1, 8.1 Hz, 1H, CH), 6.89–6.94 (m, 2H, Ar-H), 7.11 (d, J=6.6 Hz, 2H, Ar-H), 7.39 (m, 2H, Ar-H), 7.82 (d, J=6 Hz, 1H, Ar-H), 7.97 (d, J=5.7 Hz, 1H, Ar-H); ms: m/z 351.2 (M⁺+1); Anal. Calcd. for C₂₁H₂₂N₂O₃: C, 71.98; H, 6.33; N, 7.99. Found: C, 71.86; H, 6.43; N, 8.10.

Phenyl 2-(chroman-2-yl)-1H-benzimidazole-1-carboxylate (6d). ir (KBr): 3460, 2938, 1762 cm⁻¹; ¹H NMR (CDCl₃): δ 2.52–2.57 (m, 2H, CH₂), 2.99–3.05 (m, 2H, CH₂), 5.92 (dd, J = 8.6, 8.7 Hz, 1H, CH), 6.86–6.97 (m, 2H, Ar-H), 7.12 (d, J = 7.5 Hz, 2H, Ar-H), 7.26–7.52 (m, 7H, Ar-H), 7.87 (d, J = 6.3 Hz, 1H, Ar-H), 8.09 (d, J = 5.7 Hz, 1H, Ar-H); ms: m/z 371.1 (M⁺+1); Anal. Calcd. for C₂₃H₁₈N₂O₃: C, 74.58; H, 4.90; N, 7.56. Found: C, 74.69; H, 4.78; N, 7.66.

2-(Chroman-2-yl)-1-methanesulfonyl-1H-benzimidazole (6e). ir (KBr): 3436, 1583, 1489 cm⁻¹; 1 H NMR (CDCl₃): δ 2.59–2.69 (m, 2H, CH₂), 3.02–3.08 (m, 2H, CH₂), 3.56 (s, 3H, CH₃), 5.82 (dd, J=7.9, 8.1 Hz, 1H, CH), 6.76 (d, J=8.1 Hz, 2H, Ar-H), 6.89–6.94 (m, 2H, Ar-H), 7.41–7.44 (m, 2H, Ar-H), 7.82 (d, J=7.8 Hz, 1H, Ar-H), 7.97 (d, J=6.3 Hz, 1H, Ar-H); ms: m/z 329.1 (M⁺+1); Anal. Calcd. for C₁₇H₁₆N₂O₃S: C, 62.18; H, 4.91; N, 8.53. Found: C, 62.32; H, 4.80; N, 8.49.

2-(Chroman-2-yl)-1-tosyl-1H-benzimidazole (6f). ir (KBr): 3399, 1485, 1374 cm⁻¹; ¹H NMR (CDCl₃): δ 2.41 (s, 3H, CH₃), 2.53–2.61 (m, 2H, CH₂), 2.96–3.10 (m, 2H, CH₂), 5.97 (dd, J=7.7, 7.9 Hz, 1H, CH), 6.7 (d, J=9 Hz, 1H, Ar-H), 6.88–6.93 (m, 1H, Ar-H), 7.08–7.15 (m, 2H, Ar-H), 7.3–7.40 (m, 4H, Ar-H), 7.76 (d, J=7.8 Hz, 1H, Ar-H), 7.97 (d, J=8.1 Hz, 2H, Ar-H), 8.05 (d, J=8.1 Hz, 1H, Ar-H); ms: m/z 405.1 (M⁺+1); Anal. Calcd. for C₂₃H₂₀N₂O₃S: C, 60.30; H, 4.98; N, 6.93. Found: C, 60.15; H, 5.20; N, 6.78.

Methyl-2-(6-bromochroman-2-yl)-1H-benzimidazole-1-carboxylate (6'c). ir (KBr): 3434, 2926, 1753, 1481, 1357 cm⁻¹; ¹H NMR (CDCl₃): δ 2.42–2.52 (m, 2H, CH₂), 2.95–3.08 (m, 2H, CH₂), 4.13 (s, 3H, CH₃), 5.88 (dd, J=8.4, 8.4 Hz, 1H, CH), 6.83 (d, J=8.7 Hz, 1H, Ar-H), 7.19–7.4 (m, 4H, Ar-H), 7.80–7.95 (m, 2H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 23.22 (CH₂), 23.87 (CH₂), 53.52 (NCH₃), 70.76 (OCH), 111.55, 113.68, 117.56, 119.39, 122.77, 123.47, 128.85, 130.66, 130.95, 131.31, 140.59, 149.28, 151.85, 152.26 (aromatic carbons) ppm; ms: m/z 389.52 (M⁺+1); Anal. Calcd. for C₁₈H₁₅BrN₂O₃: C, 55.83; H, 3.90; N, 7.23. Found: C, 55.94; H, 3.78; N, 7.34.

Ethyl-2-(6-bromochroman-2-yl)-1H-benzimidazole-1-carboxylate (6'd). ir (KBr): 3436, 2945, 1745, 1480, 1330 cm⁻¹;

¹H NMR (CDCl₃): δ 1.51 (t, J = 7.2 Hz, 3H, CH₃), 2.45–2.50 (m, 2H, CH₂), 2.94–3.02 (m, 2H, CH₂), 4.58 (q, J = 5.4 Hz, 2H, CH₂), 5.88, (dd, J = 8.7 and 8.7 Hz, 1H, CH), 6.82 (d, J = 8.4 Hz, 1H, Ar-H), 7.19–7.39 (m, 4H, Ar-H), 7.8–7.97 (m, 2H, Ar-H) ppm;

¹³C NMR (DMSO-d₆) (75 MHz): δ 13.1 (CH₃), 63.52 (NCH₂), 23.29 (CH₂), 24.94 (CH₂), 70.87 (OCH), 111.58, 113.78, 117.62, 119.44, 122.76, 123.43, 128.90, 130.71, 130.98, 131.52, 140.66, 148.78, 151.89, 152.29 (aromatic carbons) ppm; ms: m/z 403.46 (M⁺+1); Anal. Calcd. for C₁₉H₁₇BrN₂O₃: C, 56.87; H, 4.27; N, 6.98. Found: C, 56.70; H, 4.38; N, 7.15.

Isobutyl-2-(6-bromochroman-2-yl)-1H-benzimidazole-1-carboxylate (6'e). ir (KBr): 3429, 2927, 1730, 1663, 1474 cm⁻¹;

¹H NMR (DMSO-d₆): δ 1.07 (d, J = 6.6 Hz, 6H, CH₃), 1.50–1.58 (m, 1H, CH), 2.17–2.24 (m, 2H, CH₂), 2.98–3.04 (m, 2H, CH₂), 4.32 (d, J = 6.6 Hz, 2H, CH₂), 5.9 (dd, J = 8.9, 8.9 Hz, 1H, CH), 6.82 (d, J = 8.4 Hz, 1H, Ar-H), 7.18–7.26 (m, 2H, Ar-H), 7.37–7.39 (m, 2H, Ar-H), 7.81 (d, J = 7.8 Hz, 1H, Ar-H), 7.96 (d, J = 6.6 Hz, 1H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 19.21, 19.21, 24.38 (CH₃), 26.06 (CH₂), 27.81, 71.92 (CH₃), 74.46, 112.65, 114.80, 118.73, 120.59,

123.83, 124.52, 125.32, 129.98, 131.79, 132.59, 141.79, 150.10, 153.10, 153.38 (aromatic carbons) ppm; ms: m/z 429.51 (M⁺+1); Anal. Calcd. for $C_{21}H_{21}BrN_2O_3$: C, 58.75; H, 4.93; N, 6.53. Found: C, 58.82; H, 4.83; N, 6.64.

2-(6-Bromochroman-2-yl)-1-methanesulfonyl-1H-benzimidazole (6'f). ir (KBr): 3435, 3030, 2927, 1475, 1372 cm⁻¹; 1 H NMR (DMSO-d₆): δ 2.58–2.66 (m, 2H, CH₂), 2.96–3.04 (m, 2H, CH₂), 3.52 (s, 3H, CH₃), 5.8 (dd, J=8.7, 8.9 Hz, 1H, CH), 6.65 (d, J=8.7 Hz, 1H, Ar-H), 7.18–7.44 (m, 4H, Ar-H), 7.82 (d, J=5.4 Hz, 1H, Ar-H), 7.96 (d, J=6.3 Hz, 1H, Ar-H) ppm; 13 C NMR (DMSO-d₆) (75 MHz): δ 24.25 (CH₂), 24.83 (CH₂), 42.62 (SCH₃), 70.69 (OCH), 113.32, 116.09, 117.88, 120.99, 124.17, 125.02, 126.03, 129.82, 130.11, 132.30, 141.04, 150.75, 152.76 (aromatic carbons) ppm; ms: m/z 409.23 (M⁺+1); Anal. Calcd. for C₁₇H₁₅BrN₂O₃S: C, 50.13; H, 3.71; N, 6.88. Found: C, 50.30; H, 3.59; N, 7.11.

2-(6-Bromochroman-2-yl)-1-tosyl-1H-benzimidazole (6'g). ir (KBr): 3432, 2938, 1474, 1370 cm⁻¹; ¹H NMR (CDCl₃): δ 2.41 (s, 3H, CH₃) 2.42–2.52 (m, 2H, CH₂), 2.92–3.01 (m, 2H, CH₂), 5.96 (dd, J=8.4 and 8.4 Hz, 1H, CH), 6.56 (d, J=8.4 Hz, 1H, Ar-H), 7.18 (d, J=7.5 Hz, 1H, Ar-H), 7.26–7.42 (m, 4H, Ar-H), 7.76 (d, J=6.9 Hz, 1H, Ar-H), 7.94 (d, J=8.1 Hz, 2H, Ar-H), 8.04 (d, J=7.8 Hz, 2H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 21.70 (CH₃), 24.38 (CH₂), 26.08 (CH₂), 70.77 (OCH), 112.98, 113.78, 118.35, 120.95, 124.07, 124.94, 125.83, 127.29, 127.29, 130.04, 130.04, 130.09, 132.17, 132.91, 135.21, 141.46, 146.12, 151.62, 153.21 (aromatic carbons) ppm; ms: m/z 483.38 (M⁺+1); Anal. Calcd. for C₂₃H₁₉BrN₂O₃S: C, 57.15; H, 3.96; N, 5.80. Found: C, 57.31; H, 3.85; N, 5.62.

General procedure for the synthesis of compound (6g-6k and 6'h-6'l) using ILs. A solution of (2 mmol) of 5 and 5a, (4 mmol) of sodium hydroxide in ILs (2 mL), followed by the addition of appropriate (3 mmol) benzyl bromides at 25–30°C. After the addition was complete, the solution was heated to 75°C for 5–6 h (as monitored by TLC). The reaction mass was then cooled to 25°C and diluted with water (25 mL). The resultant solid was filtered and washed with water (10 mL) to afford crude products. The crude products were recrystallized from ethanol, to obtain pure compounds 6g-6k and 6'h-6'l, respectively. The aqueous layer contains the ILs, which was recovered as described earlier procedure.

2-(Chroman-2-yl)-1-benzyl-1H-benzimidazole (6g). ir (KBr): 3432, 2930, 1583, 1487, 1232 cm⁻¹; ¹H NMR (CDCl₃): δ 2.55–2.63 (m, 2H, CH₂), 2.94–3.01 (m, 2H, CH₂), 5.36 (dd, J = 9.3, 9.4 Hz, 1H, CH), 5.62 (s, 2H, CH₂), 6.64 (d, J = 8.1 Hz, 1H, Ar-H), 6.85–6.90 (m, 1H, Ar-H), 7.03–7.13 (m, 4H, Ar-H), 7.26–7.28 (m, 6H, Ar-H), 7.83 (d, J = 6, 6 Hz, 1H, Ar-H); ms: m/z 341.2 (M⁺+1); Anal. Calcd. for C₂₃H₂₀N₂O: C, 81.15; H, 5.92; N, 8.23. Found: C, 81.27; H, 5.82; N, 8.35.

2-(Chroman-2-yl)-1-(4-fluorobenzyl)-1H-benzimidazole (6h). ir (KBr): 3454, 2945, 1485 cm $^{-1}$; 1 H NMR (CDCl₃): δ 2.56–2.60 (m, 2H, CH₂), 2.90–3.02 (m, 2H, CH₂), 5.35 (dd, J = 8.7, 9.0 Hz, 1H, CH), 5.58 (s, 2H, CH₂), 6.62 (d, J = 7.5 Hz, 1H, Ar-H), 6.86–6.91 (m, 1H, Ar-H), 6.99–7.11 (m, 5H, Ar-H), 7.24–7.28 (m, 4H, Ar-H), 7.82 (d, J = 6.6 Hz, 1H, Ar-H); ms: m/z 360.2 (M $^{+}$ +1); Anal. Calcd. for C₂₃H₁₉FN₂O: C, 77.08; H, 5.34; N, 7.82. Found: C, 77.26; H, 5.42; N, 7.68.

2-(Chroman-2-yl)-1-(4-bromobenzyl)-1H-benzimidazole (6i). ir (KBr): 3432, 2926, 1584, 1459, 1231cm⁻¹; ¹H NMR

(CDCl₃): δ 2.54–2.60 (m, 2H, CH₂), 2.99–3.04 (m, 2H, CH₂), 5.34 (dd, J=7.8, 8.4 Hz, 1H, CH), 5.56 (s, 2H, CH₂), 6.59 (d, J=7.2 Hz, 1H, Ar-H), 6.88 (m, 2H, Ar-H), 6.99–7.11 (m, 2H, Ar-H), 7.19–7.31 (m, 4H, Ar-H), 7.42 (d, J=7.8 Hz, 2H, Ar-H), 7.83 (d, J=7.2 Hz, 1H, Ar-H); ms: m/z 421.1 (M⁺+1); Anal. Calcd. for C₂₃H₁₉BrN₂O: C, 65.88; H, 4.57; N, 6.68. Found: C, 65.98; H, 4.45; N, 6.78.

2-(Chroman-2-yl)-1-(4-methylbenzyl)-1H-benzimidazole (6j). ir (KBr): 3431, 1582, 1456, 1230 cm⁻¹; ¹H NMR (CDCl₃): δ 2.31 (s, 3H, CH₃), 2.55–2.61 (m, 2H, CH₂), 2.91–3.01 (m, 2H, CH₂), 5.33 (dd, J=7.5 Hz, 1H, CH), 5.57 (s, 2H, CH₂), 6.7 (d, J=8.1 Hz, 1H, 5' Ar-H), 6.86–6.90 (m, 1H, Ar-H), 7.01–7.11 (m, 6H, Ar-H), 7.30–7.32 (m, 3H, Ar-H), 7.82 (d, J=7.2 Hz, 1H, Ar-H); ms: m/z 355.2 (M⁺+1); Anal. Calcd. for C₂₄H₂₂N₂O: C, 81.33; H, 6.26; N, 7.90. Found: C, 81.19; H, 6.37; N, 8.11.

2-(Chroman-2-yl)-1-(4-tert-butylbenzyl)-1H-benzimidazole (6k). ir (KBr): 3428, 2958, 1581, 1463 cm⁻¹; ¹H NMR (CDCl₃): δ 1.28 (s, 9H, tert butyl), 2.55–5.62 (m, 2H, CH₂), 2.97–3.01 (m, 2H, CH₂), 5.34 (dd, J=9.1, 9.1 Hz, 1H, CH), 5.52–5.65 (s, 2H, CH₂), 6.63–6.66 (m, 1H, Ar-H), 6.85–6.90 (m, 1H, Ar-H), 7.03–7.10 (m, 3H, Ar-H), 7.26–7.31 (m, 6H, Ar-H), 7.82 (d, J=6.2 Hz, 1H, Ar-H); ms: m/z 397.2 (M⁺+1); Anal. Calcd. for C₂₇H₂₈N₂O: C, 81.78; H, 7.12; N, 7.06. Found: C, 81.69; H, 7.31; N, 7.20.

2-(6-Bromochroman-2-yl)-1-benzyl-1H-benzimidazole (6'h). ir (KBr): 3446, 2926, 1574, 1482, 1234 cm⁻¹; ¹H NMR (CDCl₃): δ 2.46–2.55 (m, 2H, CH₂), 2.94–3.01 (m, 2H, CH₂), 5.38 (dd, J=7.2, 8.1 Hz, 1H, CH), 5.60 (s, 2H, CH₂), 6.47 (d, J=9 Hz, 1H, Ar-H), 7.02–7.15 (m, 2H, Ar-H), 7.21–7.28 (m, 8H, Ar-H), 7.86 (d, J=7.8 Hz, 1H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 22.99 (CH₂), 23.68 (CH₂), 46.54 (NCH₂), 70.08 (OCH), 108.89, 111.84, 117.25, 118.97, 121.24, 122.23, 122.73, 125.07, 125.07, 126.51, 127.57, 128.87, 130.88, 134.55, 134.89, 140.74, 150.15, 151.41 (aromatic proton) ppm; ms: m/z 421.60 (M⁺+1); Anal. Calcd. for C₂₃H₁₉BrN₂O: C, 65.88; H, 4.57; N, 6.68. Found: C, 65.70; H, 4.68; N, 6.77.

2-(6-Bromochroman-2-yl)-1-(4-fluorobenzyl)-1H-benzimidazole (*6'i*). ir (KBr): 3435, 2928, 1605, 1484, 1227 cm⁻¹; 1 H NMR (CDCl₃): δ 2.54–2.58 (m, 2H, CH₂), 2.96–3.04 (m, 2H, CH₂), 5.38 (dd, J=8.4, 8.4 Hz, 1H, CH), 5.56 (s, 2H, CH₂), 6.47 (d, J=9 Hz, 1H, Ar-H), 6.96–6.99 (m, 2H, Ar-H), 7.12–7.22 (m, 3H, Ar-H), 7.26–7.44 (m, 4H, Ar-H), 7.85 (d, J=9 Hz, 1H, Ar-H) ppm; 13 C NMR (CDCl₃) (75 MHz): δ 24.02 (CH₂), 24.72 (CH₂), 47.07 (NCH₂), 71.28 (OCH), 110.01, 113.17, 115.63, 115.92, 118.39, 120.29, 122.60, 123.57, 123.98, 128.03, 128.14, 130.18, 131.91, 132.22, 135.64, 142.03, 151.33, 152.62, 160.58 (aromatic carbons) ppm; ms: m/z 437.7 (M⁺+1); Anal. Calcd. for C₂₃H₁₈BrFN₂O: C, 63.17; H, 4.15; N, 6.41. Found: C, 63.28; H, 4.31; N, 6.28.

2-(6-Bromochroman-2-yl)-1-(4-bromobenzyl)-1H-benzimid-azole (6'j). ir (KBr): 3444, 2932, 1574, 1480, 1234 cm⁻¹; 1 H NMR (CDCl₃): δ 2.51–2.57 (m, 2H, CH₂), 2.92–3.02 (m, 2H, CH₂), 5.36 (dd, J=9, 9 Hz, 1H, CH), 5.54 (s, 2H, CH₂), 6.46 (d, J=7.8 Hz, 1H, Ar-H), 6.44 (d, J=9 Hz, 2H, Ar-H), 6.97 (d, J=7.8 Hz, 2H, Ar-H) 7.19 (m, 3H, Ar-H), 7.44 (d, J=8.4 Hz, 2H, Ar-H), 7.83 (d, J=7.8 Hz, 1H, Ar-H) ppm; 13 C NMR (CDCl₃) (75 MHz): δ 23.95 (CH₂), 24.96 (CH₂), 47.11 (NCH₂), 71.13 (OCH), 109.81, 113.05, 118.19, 120.09, 121.48, 122.49, 123.45, 123.76, 127.88, 127.88, 129.96, 131.71,

131.71, 131.91, 134.97, 135.34, 141.74, 151.06, 152.34 (aromatic carbons) ppm; ms: m/z 499.64 (M⁺+1); Anal. Calcd. for $C_{23}H_{18}Br_2N_2O$: C, 55.45; H, 3.64; N, 5.62. Found: C, 55.34; H, 3.76; N, 5.49.

2-(6-Bromochroman-2-yl)-1-(4-methylbenzyl)-1H-benzimidazole (6'k). ir (KBr): 3427, 2929, 1513, 1262 cm⁻¹; ¹H NMR (CDCl₃): δ 2.32 (s, 3H, CH₃), 2.51–2.55 (m, 2H, CH₂), 2.94–3.0 (m, 2H, CH₂), 5.41 (dd, J = 8.6, 8.6 Hz, 1H, CH), 5.56 (s, 2H, CH₂), 6.55 (d, J = 8.7 Hz, 1H, Ar-H), 6.98–6.70 (m, 2H, Ar-H), 7.09–7.16 (m, 4H, Ar-H), 7.22–7.29 (s, 3H, Ar-H), 7.85 (d, J = 7.2 Hz, 1H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 21.16 (CH₃), 24.19 (CH₂), 24.88 (CH₂), 47.52 (NCH₂), 71.25 (OCH), 110.12, 112.99, 118.46, 119.9, 120.10, 122.35, 123.33, 123.93, 126.26, 129.39, 129.39, 130.04, 132.05, 132.97, 135.46, 137.46, 141.90, 151.28, 152.67 (aromatic carbons) ppm; ms: m/z 435.75 (M⁺+1); Anal. Calcd. for C₂₄H₂₁BrN₂O: C, 66.52; H, 4.88; N, 6.46. Found: C, 66.40; H, 5.13; N, 6.58.

2-(6-Bromochroman-2-yl)-1-(4-tert-butylbenzyl)-1H-benzi-midazole (6'l). ir (KBr): 3446, 2962, 1475, 1411, 1219 cm⁻¹; ¹H NMR (CDCl₃): δ 1.28 (s, 9H, tert. butyl), 2.50–2.54 (m, 2H, CH₂), 2.94–2.99 (m, 2H, CH₂), 5.36 (dd, J = 9.9 Hz, 1H, CH), 5.57 (s, 2H, CH₂), 7.02 (d, J = 7.2 Hz, 1H, Ar-H), 7.11–7.13 (m, 1H, Ar-H), 7.21–7.29 (m, 8H, Ar-H), 7.85 (d, J = 7.8 Hz, 1H, Ar-H) ppm; ¹³C NMR (CDCl₃) (75 MHz): δ 24.06 (CH₂), 24.75 (CH₂), 31.30, 31.30, 31.30 (CH₃), 34.51 (tert.butyl), 47.29 (NCH₂), 71.06, (OCH), 110.04, 112.90, 118.39, 120.05, 122.27, 123.25, 123.87, 125.56, 125.56, 125.91, 125.91, 129.91, 131.95, 133.03, 135.73, 141.86, 150.57, 151.24, 152.56 (aromatic carbons) ppm; ms: m/z 477.40 (M⁺+1); Anal. Calcd. for C₂₇H₂₇BrN₂O: C, 68.21; H, 5.72; N, 5.89. Found: C, 68.10; H, 5.80; N, 6.10.

Phenyl-2-(6-bromochroman-2-yl)-1H-benzimidazole-1-carboxylate ($6^{\prime}m$). ir (KBr): 3435, 2925, 1766, 1475, 1355, 1232 cm⁻¹; 1 H NMR (CDCl₃): δ 2.50–2.54 (m, 2H, CH₂), 2.95–3.05 (m, 2H, CH₂), 5.60 (s, 2H, CH₂), 5.94 (dd, J=9.0, 9.1 Hz, 1H, CH), 6.90 (d, J=8.1 Hz, 1H, Ar-H), 7.09–7.14 (m, 2H, Ar-H), 7.26–7.52 (m, 5H, Ar-H), 7.86–7.88 (m, 1H, Ar-H), 8.07–8.09 (m, 1H, Ar-H) ppm; ms: m/z 421.72 (M⁺+1); Anal. Calcd. for C₂₃H₁₇BrN₂O₃: C, 61.48; H, 3.81; N, 6.23. Found: C, 61.58; H, 3.92; N, 6.31.

REFERENCES AND NOTES

- [1] Sheehan, D. J.; Hitchcock, C. A.; Sibley, C. M. Clin Microbiol Rev 1999, 12, 40.
- [2] Kazimerczuk, Z.; Andrzejewska, M.; Klimesova, V. Eur J Med Chem 2005, 40, 203.

- [3] Sukalovic, V.; Andric, D.; Roglic, G.; Sladjana, K. R.; Schratteholz, A.; Soski, V. Eur J Med Chem 2005, 40, 481.
- [4] Mariana, B.; Mercedes, G. Mini Rev Med Chem 2005, 5, 409
- [5] Khalafi-Nezhad, A.; Soltani Rad, M. N.; Mohabatkar, H.; Ansari, Z.; Hemmateenejad, B. Bioorg Med Chem Lett 2003, 13, 1931.
- [6] Bin, S.; Jincheng, H.; Qun, S.; Valenzano, K. J.; Lori, S.; Scott, N. Bioorg Med Chem Lett 2005, 15, 719.
- [7] Li, Y.; Kataoka, M.; Tatsuta, M.; Yasoshima, K.; Yura, T.; Urbahns, K.; Kiba, A.; Yamamoto, N.; Gupta, J. B.; Hashimoto, K. Bioorg Med Chem Lett 2005, 15, 805.
- [8] Garg, Y.; Samota, M. K.; Seth, G. Asian J Chem 2005, 17, 615
- [9] Teague, S. J.; Barber, S.; King, S.; Stein, L. Tetrahedron Lett 2005, 46, 4613.
- [10] Hashimoto, K.; Tatsuta, M.; Yasoshima, K.; Shogase, Y.; Shimazaki, M.; Yura, T.; Li, Y.; Urbahns, K.; Yamamoto, N.; Gupta, J. B. Bioorg Med Chem Lett 2005, 15, 799.
- [11] (a) Preston, P. N. Chem Rev 1974, 74, 279; (b) Preston, P.
 N. Merck Index 11th Edn. 1794.
- [12] Preston, P. N.; In Benzimidazoles and Congeneric Tricyclic Compounds; Wiley Interscience: New York, 1980; Part 2, Chapter 10, pp 531.
- [13] Daniel, L. A. Strategiies for Organic Drug Synthesis and Design; Wiley Interscience: 1998; p 300.
- [14] Eaton, P. E.; Carlson, G. R.; Lee, J. T. J Org Chem 1973, 38, 4071.
 - [15] Phillips, M. A. J Chem Soc 1928, 172, 2393.
- [16] Dubey, R.; Hari Narayana Moorthy, N. S. Chem Pharm Bull 2007, 55, 115.
- [17] Maradolla M. B.; Allam S. K.; Mandha A.; Chandramouli G. V. P. Arkivoc 2008, 15, 42.
- [18] Vinod Kumar, R.; Vaidya, S. D.; Siva Kumar, B. V.; Bhise U. N.; Bhirud S. B.; Mashelkar, U. C. Indian J Hetrocycl Chem 2005, 14, 197.
- [19] Vaidya, S. D.; Siva Kumar, B. V.; Vinod Kumar, R.; Bhirud S. B.; Mashelkar, U. C. Euro J Med Chem 2008, 43, 986.
- [20] Siva Kumar, B. V.; Vaidya, S. D.; Vinod Kumar, R.; Bhirud S. B.; Mane R. B. Euro J Med Chem 2006, 41, 599.
 - [21] Welton, T. Chem Rev 1999, 99, 2071.
- [22] Wasserscheid, P.; Keim, W. Angew Chem Int Ed Engl 2000, 39, 3772.
 - [23] Sheldon, R. Chem Commun 2001, 1, 2399.
 - [24] Gordon, C. M. Appl Catal A 2001, 222, 101.
- [25] Wasserscheid. P.; Welton, T., Eds. Ionic Liquids in Synthesis; VCH-Wiley: Weinheim, 2002.
 - [26] Connell, R. D. U. S. Pat.6,469,031 (2002).
- [27] Owens, G. S.; Abu-Omer, M. M. J Mol Catal A 2002, 187, 211.