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## The Synthesis of Some New Fused and Substituted Chromenes

**Ahmed M. El-Sayed**

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*Some new chromeno[2,3-b]pyrimidines, chromeno[3,2-c]pyridines, chromeno[2,3-b]pyridines and 3-chromenyl-1,3-thiazines were synthesized via the synthetic studies of the reaction of 2-imino-2H-chromen-3-thiocarboxamide with some aromatic aldehydes, active nitriles, and their ylidenes derivatives.*

**Keywords** Chromenopyridines; chromenothiazines; malononitrile; ylidene nitriles

### INTRODUCTION

Fused coumarins comprise a very interesting class of compounds for their significant antibacterial<sup>1–7</sup> and novobiocin<sup>8,9</sup> activities. For these reasons we continue our previous work,<sup>6,7</sup> which deals with the synthesis of chromenopyrazoles, chromenopyridines, and chromenoazepines. We report here the synthesis of some new chromenopyrimidines, chromenopyridines, and 3-chromenyl-1,3-thiazines.

### RESULTS AND DISCUSSION

The treatment of 2-imino-2H-chromene-3-thiocarboxamide **1**<sup>10</sup> with different aromatic aldehydes, namely benzaldehyde, p-anisaldehyde, p-nitrobenzaldehyde, and 2-naphthaldehyde, in a 1:1 molar ratio in refluxing tetrahydrofuran containing a catalytic amount of piperidine gave the cyclized products 2-phenyl-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione, 2-(4-methoxyphenyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione, 2-(4-nitrophenyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione, and 2-(2-naphthyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thiones **2a–d**. The effect of a cyclic ketone on compound **2b** was tested via its reaction with cyclohexanone in refluxing dioxan in the presence of triethylamine as a catalyst where

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### TABLE I Analytical and Spectral Data of the Prepared Compounds

Compound No.	M.P. (°C) <sup>a</sup>	Crys. Solv.	Yield %	M.F. (M.wt)	Analytical Data <sup>b</sup>				I.R. (KBr) <sup>c</sup> (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ) <sup>d</sup> (δ ppm)
					Calc.	Found	%			
					C	H	N	S		
<b>2a</b>	274–276		68	C <sub>17</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> S (292.35)	69.84 (70.10)	4.14 (4.07)	9.58 (9.75)	10.97 (10.86)	3146(NH); 1188 (C=S)	8.12(s, 1H, C <sub>5</sub> -H); 8.00–7.12 (m, 11H, arom. + NH + C <sub>2</sub> -H)
<b>2b</b>	197–198	THF	54	C <sub>18</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S (322.38)	67.06 (67.22)	4.38 (4.30)	8.69 (8.88)	9.95 (10.12)	3139(NH); 1173 (C=S)	8.42(s, 1H, C <sub>5</sub> -H); 8.13–6.94 (m, 9H, arom. + C <sub>2</sub> -H); 6.21(s, 1H, NH); 3.94(s, 3H, OCH <sub>3</sub> )
<b>2c</b>	184–186	methanol	91	C <sub>17</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub> S (337.35)	60.52 (60.70)	3.29 (3.35)	12.46 (12.60)	9.51 (9.36)	3183(NH); 1177 (C=S)	8.64(s, 1H, C <sub>5</sub> -H); 8.41–7.10 (m, 10H, arom. + C <sub>2</sub> -H + NH)
<b>2d</b>	252–254	EtOH/dioxane	56	C <sub>21</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S (342.41)	73.66 (73.80)	4.12 (4.08)	8.18 (8.38)	9.36 (9.14)	3153(NH); 1199 (C=S)	8.81(s, 1H, C <sub>5</sub> -H); 8.20–7.12 (m, 13H, arom. + C <sub>2</sub> -H + NH)
<b>3b</b>	261–263	dioxane	73	C <sub>24</sub> H <sub>22</sub> N <sub>2</sub> O <sub>2</sub> S (402.51)	71.61 (71.89)	5.51 (5.60)	6.96 (6.75)	7.97 (8.14)	3140(NH)	9.80(s, 1H, NH); 8.22–6.75 (m, 9H, arom. + CH <sub>pyrim</sub> ); 3.82(s, 3H, OCH <sub>3</sub> ); 3.69(s, 1H, CH <sub>pyrim</sub> ); 2.31–1.20(m, 8H, 4CH <sub>2</sub> Cyclohex)
<b>4</b>	238–239	AcOH/EtOH	85	C <sub>12</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> S (246.29)	58.52 (58.69)	4.09 (4.15)	11.37 (11.554)	13.02 (12.82)	3288, 3146(2NH); 1697(C=O); 1143 (C=S)	9.40(s, 1H, NH <sub>acetyl</sub> ); 8.32(s, 1H, C <sub>4</sub> -H); 7.83–7.13(m, 5H, arom.+NH); 2.40(s, 3H, COCH <sub>3</sub> )
<b>5<sup>a</sup></b>	267–269	DMSO	79	C <sub>12</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> S (228.27)	63.14 (63.02)	3.53 (3.60)	12.27 (12.48)	14.05 (13.94)	1150(C=S)	8.50(s, 1H, C <sub>5</sub> -H); 7.65–7.05 (m, 4H, arom.); 2.30(s, 1H, CH <sub>3</sub> )
<b>6</b>	243	EtOH/dioxane	68	C <sub>11</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> S (232.26)	56.88 (56.72)	3.47 (3.41)	12.06 (12.26)	13.81 (13.65)	3289, 3146(2NH); 1697(C=O); 1143 (C=S)	11.80(s, 1H, NH <sub>formyl</sub> ); 8.71(s, 1H, C <sub>4</sub> -H); 8.24(s, 1H, CHO); 7.70–7.10(m, 5H, arom. + NH)

7	203–205 EtOH	53	C <sub>11</sub> H <sub>6</sub> N <sub>2</sub> O <sub>8</sub> (214.24)	61.67 (61.46)	2.82 (2.90)	13.08 (13.28)	14.97 (14.78)	1170 (C≡S)	8.85(s, 1H, C <sub>5</sub> -H); 8.30 (s, 1H, C <sub>2</sub> -H); 7.85–7.27(m, 4H, arom.)
8	>300 EtOH/dioxane	64	C <sub>13</sub> H <sub>7</sub> N <sub>3</sub> O <sub>8</sub> (253.28)	61.65 (61.46)	2.79 (2.84)	16.59 (16.82)	12.66 (12.54)	3344, 3210(2NH); 2202(CN); 1237 (C=S)	10.00 (s, 1H, C <sub>5</sub> -H); 8.92 (s, 1H, NH); 8.61(s, 1H, NH); 7.90–6.82 (m, 4H, arom)
9	>300 EtOH	75	C <sub>13</sub> H <sub>6</sub> N <sub>3</sub> O <sub>2</sub> S (271.29)	57.55 (57.82)	3.34 (3.39)	15.49 (15.69)	11.82 (11.70)	3320, 3195(2NH); 2211 (CN); 1690 (C≡O)	12.84 (s, 1H, NHCO); 9.61(s, 1H, C <sub>4</sub> -H); 8.90 (s, 1H, =NH); 7.85–7.00(m, 4H, arom.); 4.80 (s, 2H, CH <sub>2</sub> ).
10	>300 EtOH/dioxane	84	C <sub>13</sub> H <sub>6</sub> N <sub>3</sub> O <sub>2</sub> S <sub>2</sub> (287.36)	54.34 (54.61)	3.16 (3.20)	14.62 (14.44)	22.32 (22.41)	3176 (NH); 2650 (SH); 2171 (CN); 1213(C=S)	9.93 (s, 1H, NH); 8.70 (s, 1H, C <sub>4</sub> -H); 7.44–6.60(m, 4H, arom.); 3.11 (s, 2H, CH <sub>2</sub> ); 1.60 (s, 1H, SH)
11	296–298 EtOH/dioxane	71	C <sub>13</sub> H <sub>7</sub> N <sub>3</sub> O <sub>8</sub> (253.28)	61.65 (61.74)	2.79 (2.89)	16.59 (16.70)	12.66 (12.48)	2211 (CN); 1157 (C=S)	8.90(s, 1H, C <sub>5</sub> -H); 7.86–7.10 (m, 4H, arom.); 4.82–4.09 (d, 2H, CH <sub>2</sub> )
12	>300 (dec.) EtOH/THF	51	C <sub>13</sub> H <sub>6</sub> N <sub>4</sub> O (236.23)	66.10 (66.31)	3.41 (3.48)	23.72 (23.54)	— —	3431, 3348, 3181 (NH, NH <sub>2</sub> ); 2205 (CN)	8.80 (s, 1H, C <sub>5</sub> -H); 8.60(s, 1H, NH); 7.84–6.90 (m, 4H, arom.); 6.73 (s, 2H, NH <sub>2</sub> )
13	134–135 EtOH	24	C <sub>15</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub> (283.28)	63.60 (63.44)	4.63 (4.57)	14.83 (14.71)	— —	3398, 3183 (2NH); 2202(CN); 1737 (C≡O)	11.72–11.50 (br, 2H, 2NH); 8.70 (s, 1H, C <sub>4</sub> -H); 8.12–7.24 (m, 4H, arom.); 4.60 (s, 1H, CH); 4.73–4.20 (q, 2H, CH <sub>2</sub> ); 1.60–1.30 (t, 3H, CH <sub>3</sub> )
14	>300 EtOH/dioxane	62	C <sub>13</sub> H <sub>7</sub> N <sub>3</sub> O <sub>2</sub> (237.21)	65.82 (65.68)	2.97 (3.03)	17.71 (17.58)	— —	3450 (OH); 3173 (NH); 2214 (CN)	12.90 (s, 1H, NH); 12.73 (s, 1H, OH); 8.90 (s, 1H, C <sub>5</sub> -H); 7.88–7.40 (m, 4H, arom.)
15	>300 (sub.) EtOH	63	C <sub>13</sub> H <sub>7</sub> N <sub>3</sub> O <sub>8</sub> (253.28)	61.65 (61.90)	2.79 (2.71)	16.59 (16.75)	12.66 (12.48)	3164 (NH); 2600 (SH); 2209 (CN)	8.73(s, 1H, NH) 7.80(s, 1H, C <sub>5</sub> -H); 7.63–6.80 (m, 4H, arom.); 3.40 (s, 1H, SH)

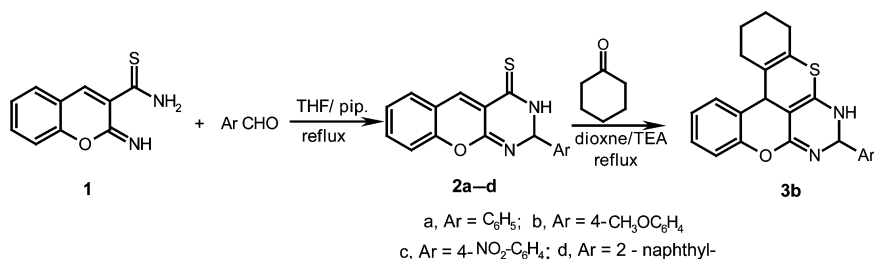
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Table I Analytical and Spectral Data of the Prepared Compounds (Continued)

Compound No.	M.P. (°C) <sup>a</sup> Solv.	Crys. Yield %	M.F. (M.wt)	Analytical data <sup>b</sup>				I.R. (KBr) <sup>c</sup> (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (DMSO-d <sub>6</sub> ) <sup>d</sup> (δ ppm)
				Calc.	Found	%			
				C	H	N	S		
<b>16<sub>a</sub></b>	190–192 EtOH	54	C <sub>20</sub> H <sub>14</sub> N <sub>4</sub> O <sub>5</sub>	67.02	3.94	15.63	8.95	3437, 3353, 3244 (NH, NH <sub>2</sub> ); 2213 (CN)	9.90(s, 1H, NH); 8.14(s, 1H, C <sub>4</sub> -H); 7.82–6.94(m, 9H, arom.); 6.67 (s, 2H, NH <sub>2</sub> ); 6.10 (s, 1H, CH)
			(358.42)	(67.30)	(3.89)	(15.49)	(8.76)		10.11 (s, 1H, NH); 8.65–7.13 (m, 10H, C <sub>4</sub> -H + arom.); 6.80 (s, 2H, NH <sub>2</sub> ); 5.78 (s, 1H, CH)
<b>16<sub>b</sub></b>	199 EtOH	86	C <sub>20</sub> H <sub>13</sub> N <sub>5</sub> O <sub>3</sub> S	59.55	3.25	17.36	7.95	3452, 3361, 3242 (NH, NH <sub>2</sub> ); 2197 (CN)	9.80 (s, 1H, NH); 7.65–6.80 (m, 9H, arom + C <sub>4</sub> -H); 6.53 (s, 2H, NH <sub>2</sub> ); 5.81 (s, 1H, CH); 3.72 (s, 3H, OCH <sub>3</sub> )
			(403.41)	(59.38)	(3.33)	(17.20)	(8.15)		10.60 (s, 1H, NH); 8.77–6.83(m, 9H, arom. + C <sub>4</sub> -H); 6.62 (s, 2H, NH <sub>2</sub> ); 6.00(s, 1H, CH); 4.89–4.40 (q, 2H, CH <sub>2</sub> ); 1.38–1.10(t, 3H, CH <sub>3</sub> )
<b>16<sub>c</sub></b>	270–273 EtOH/pet. ether	73	C <sub>21</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub> S	64.93	4.15	14.42	8.25	3437, 3351, 3241 (NH, NH <sub>2</sub> ); 2209 (CN)	11.22 (s, 1H, NH); 10.53(s, 1H, OH); 8.53–7.40 (m, 12H, arom. + C <sub>4</sub> -H); 6.10 (s, 1H, CH)
			(388.44)	(64.78)	(4.25)	(14.61)	(8.04)		10.30 (s, 1H, NH); 8.35–7.23 (m, 12H, arom. + C <sub>4</sub> -H); 6.76(s, 2H, NH <sub>2</sub> ); 6.30(s, 1H, CH); 4.74– 4.25 (q, 2H, CH <sub>2</sub> ); 1.40–1.00 (t, 3H, CH <sub>3</sub> )
<b>16<sub>d</sub></b>	60–80 218–219 EtOH	81	C <sub>22</sub> H <sub>18</sub> N <sub>4</sub> O <sub>5</sub> S	58.66	4.03	12.44	7.12	3416, 3341, 3237 (NH, NH <sub>2</sub> ); 1744 (C=O)	9.90(s, 1H, NH); 8.14(s, 1H, C <sub>4</sub> -H); 7.82–6.94(m, 9H, arom.); 6.67 (s, 2H, NH <sub>2</sub> ); 6.10 (s, 1H, CH)
			(450.47)	(58.86)	(4.10)	(12.61)	(6.90)		10.11 (s, 1H, NH); 8.65–7.13 (m, 10H, C <sub>4</sub> -H + arom.); 6.80 (s, 2H, NH <sub>2</sub> ); 5.78 (s, 1H, CH)
<b>16<sub>e</sub></b>	192–194 EtOH/(CHCl <sub>3</sub> )	28	C <sub>24</sub> H <sub>15</sub> N <sub>3</sub> O <sub>2</sub> S	70.40	3.69	10.26	7.83	3437 (OH); 3320 (NH); 2213 (CN)	9.90(s, 1H, NH); 8.14(s, 1H, C <sub>4</sub> -H); 7.82–6.94(m, 9H, arom.); 6.67 (s, 2H, NH <sub>2</sub> ); 6.10 (s, 1H, CH)
			(409.46)	(70.61)	(3.58)	(10.37)	(7.68)		10.30 (s, 1H, NH); 8.35–7.23 (m, 12H, arom. + C <sub>4</sub> -H); 6.76(s, 2H, NH <sub>2</sub> ); 6.30(s, 1H, CH); 4.74– 4.25 (q, 2H, CH <sub>2</sub> ); 1.40–1.00 (t, 3H, CH <sub>3</sub> )
<b>16<sub>f</sub></b>	159–161 Et. Acetate/pet. ether 60–80	45	C <sub>26</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> S	68.55	4.65	9.22	7.04	3440, 3380, 3260 (NH, NH <sub>2</sub> ); 1730 (C=O)	9.90(s, 1H, NH); 8.14(s, 1H, C <sub>4</sub> -H); 7.82–6.94(m, 9H, arom.); 6.67 (s, 2H, NH <sub>2</sub> ); 6.10 (s, 1H, CH)
			(455.53)	(68.4 0)	(4.50)	(9.31)	(7.20)		10.11 (s, 1H, NH); 8.65–7.13 (m, 10H, C <sub>4</sub> -H + arom.); 6.80 (s, 2H, NH <sub>2</sub> ); 5.78 (s, 1H, CH)

<b>17</b>	280–283 (char.) EtOH/CHCl <sub>3</sub>	20	C <sub>14</sub> H <sub>8</sub> N <sub>4</sub> OS (280.31)	59.99 (60.15)	2.88 (2.79)	19.99 (19.80)	11.44 (11.63)	3339 (NH); 2640 (SH); 2198 (CN)	9.00 (s, 1H, NH); 8.92 (s, 1H, =CH); 7.75–6.94 (m, 5H, C <sub>4</sub> -H + arom.); 3.50 (s, 1H, SH)
<b>18</b>	187–189 EtOH/pet.0 ether 60–8	48	C <sub>14</sub> H <sub>8</sub> N <sub>4</sub> OS (280.31)	59.99 (60.14)	2.88 (2.77)	19.99 (2083)	11.44 (11.61)	3319; 3202 (2NH); 2196 (CN)	11.70 (s, 1H, NH); 10.40 (s, 1H, NH); 8.74 (s, 1H, CH <sub>thiazyl</sub> ); 8.32 (s, 1H, C <sub>4</sub> -H); 7.75–6.83 (m, 4H, arom.)
<b>19</b>	> 300 EtOH	26	C <sub>15</sub> H <sub>10</sub> N <sub>4</sub> OS (294.33)	61.21 (61.05)	3.42 (3.50)	19.04 (19.20)	10.89 (10.65)	3178 (NH); 2630 (SH); 2201 (CN)	10.60 (s, 1H, NH); 7.86–7.16 (m, 4H, arom.); 6.82 (s, 1H, C <sub>4</sub> -H); 4.00 (s, 1H, SH); 2.80 (s, 3H, CH <sub>3</sub> )
<b>20</b>	234–236 EtOH/pet.	51	C <sub>15</sub> H <sub>10</sub> N <sub>4</sub> OS (294.33)	61.21 (61.08)	3.42 (3.50)	19.04 (19.23)	10.89 (10.68)	3281; 3174 (2NH); 2204 (CN)	12.64 (s, 1H, NH); 11.60 (s, 1H, NH); 8.20 (s, 1H, C <sub>4</sub> -H); 7.55–6.84 (m, 4H, arom.); 2.82 (s, 3H, CH <sub>3</sub> )
<b>21</b>	ether 60–80 195–196 EtOH/pet.	35	C <sub>16</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub> S (327.36)	58.70 (58.55)	4.00 (4.18)	12.84 (12.98)	9.80 (9.62)	3169 (NH); 2645 (SH); 2206 (CN); 1734 (C=O)	10.30 (s, 1H, NH); 8.95 (s, 1H, =CH); 7.86–6.94 (m, 5H, C <sub>4</sub> -H + arom.); 5.21–4.82 (q, 2H, CH <sub>2</sub> ); 4.20 (s, 1H, SH); 1.82–1.34 (t, 3H, CH <sub>3</sub> )
<b>22</b>	198–199 EtOH	48	C <sub>14</sub> H <sub>7</sub> N <sub>3</sub> O <sub>2</sub> S (281.29)	59.78 (59.59)	2.51 (2.59)	14.94 (14.72)	11.40 (11.59)	3291 (NH); 2202 (CN); 1697 (C=O)	10.40 (s, 1H, NH); 8.94 (s, 1H, CH <sub>thiazyl</sub> ); 8.45 (s, 1H, C <sub>4</sub> -H); 7.90–7.14 (m, 4H, arom.)

<sup>a</sup>Uncorrected.<sup>b</sup>Satisfactory microanalyses, obtained: (C ± 0.30%; H ± 0.11%; N ± 0.35%; S ± 0.20%).<sup>c</sup>Measured on Nicolet 710 FTIR spectrophotometer.<sup>d</sup>Measured with Varian EM 360 L spectrometer at 400 MHz using TMS as internal standard.<sup>e</sup>TFA used as a solvent for <sup>1</sup>H-NMR measurement.

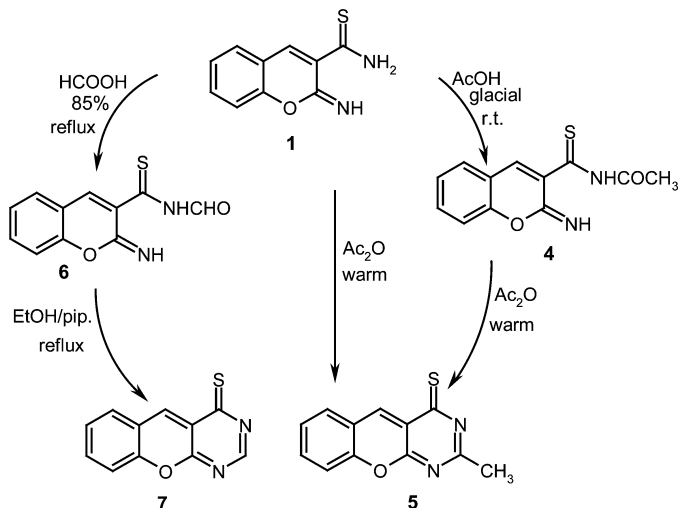


### SCHEME 1

4-(8,8a,10,11,12,13-hexahydro-7H-5-oxa-9-thia-6,8-diazanaphtho [2,-3,4-de]anthracen-7-yl)phenyl methyl ether **3b** was obtained (Scheme 1, Table I).

The formation of compound **3b** is assumed to proceed through the addition of the active methylene group of cyclohexanone on C<sub>3</sub> = C<sub>4</sub> of the chromene moiety followed by the elimination of a water molecule from the interaction of the tautomeric SH group and the carbonyl group.

The acetylation of compound **1** was achieved by treating compound **1** with glacial acetic acid at r.t. to give N-[(2-imino-2H-chromen-3-yl)carbonothioyl]acetamide **4**, which in turn cyclized to 2-methyl-4H-chromeno[2,3-d] pyrimidine-4-thione **5** in an 81% yield on heating with acetic anhydride. The same compound **5** was obtained directly by warming compound **1** with an excess amount of acetic anhydride in 79% (Scheme 2, Table I). Also, N-formyl-2-imino-2H-chromene-3-thiocarboxamide **6** was formed through the reaction of our starting

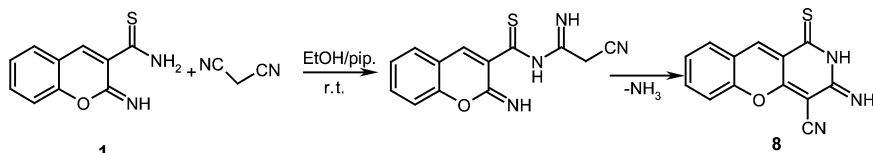


### SCHEME 2



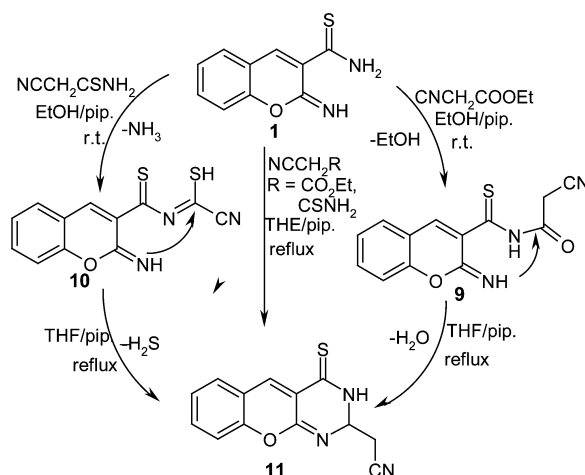
material with warmed formic acid, which underwent intramolecular cyclization into benzo[*g*]quinazoline-4(10*H*)-thione **7** in refluxing ethanol containing piperidine as a catalyst (Scheme 2, Table I).

The action of active nitriles on our starting material was studied. So, on treating compound **1** with malononitrile at r.t. in ethanol using piperidine as a catalyst, 3-imino-1-thioxo-2,3-dihydro-1*H*-chromeno-[3,2-*c*]pyridine-4-carbonitrile **8** was yielded (Scheme 3, Table I).

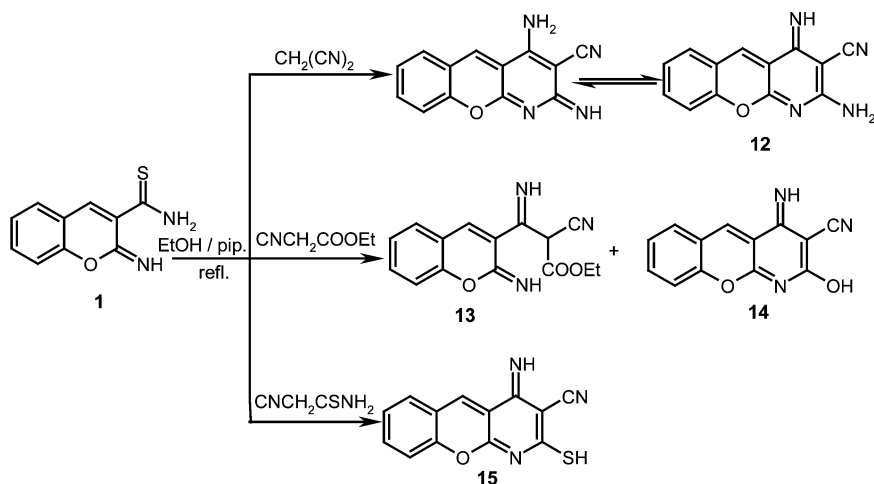


**SCHEME 3**

Also, the treatment of compound **1** with ethyl cyanoacetate or cyanothio-acetamide in ethanol containing a few drops of piperidine at r.t. afforded 2-cyano-*N*-[(2-imino-2*H*-chromen-3-yl)carbonothioyl]acetamide **9** or 2-cyano-*N*-[(2-imino-2*H*-chromen-3-yl)carbono-thioyl]ethanimidothioic acid **10**. Both compounds **9** and **10** gave the same cyclized product (4-thioxo-4*H*-chromeno[2,3-*d*]pyrimidin-2-yl)-acetonitrile **11** in 83% and 80% yields in refluxing tetrahydrofuran along with few drops of piperidine. Compound **11** was obtained directly in a 71% yield by refluxing compound **1** with ethyl cyanoacetate or cyanothioacetamide in tetrahydrofuran along with piperidine as a catalyst (Scheme 4, Table I).



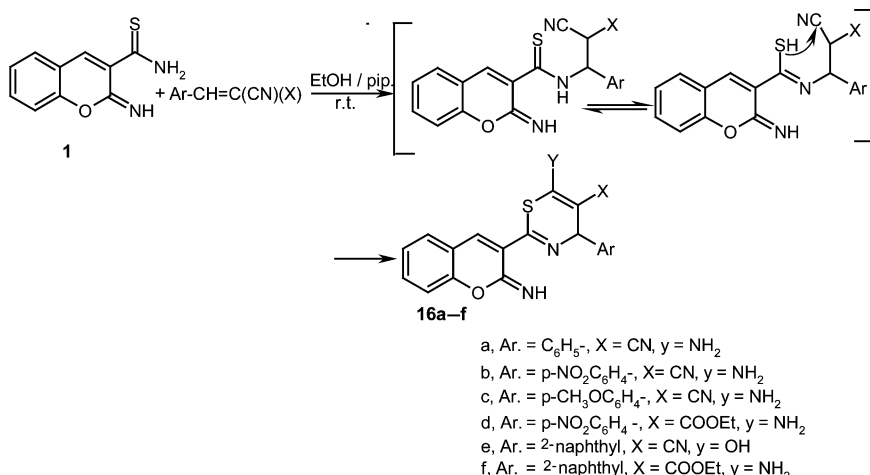
**SCHEME 4**



SCHEME 5

It has been reported<sup>11</sup> that the reaction of compound **1** with malononitrile or ethyl cyanoacetate along with ammonium acetate in a 1:1:1.5 molar ratio, respectively, in refluxing ethanol afforded the corresponding benzopyrano[3,4-c]pyridine-4H-thiones. Herein we report that the reaction of compound **1** with malononitrile, ethyl cyanoacetate, or cyanothioacetamide in refluxing ethanol using piperidine as a catalyst afforded 2-amino-4-imino-2H-chromeno[2,3-b]pyridine-3-carbonitrile **12**, ethyl 2-cyano-3-imino-3-(2-imino-2H-chromen-3-yl)propanoate **13**, 2-hydroxy-4-imino-4H-chromeno[2,3-d]pyridine-3-carbonitrile **14**, or 4-imino-2-mercapto-4H-chromeno[2,3-b]pyridine-3-carbonitrile **15**, respectively (Scheme 5, Table I). The reaction pathway is believed to be a nucleophilic attack of the methylene anion at the thione group with elimination of  $\text{H}_2\text{S}$  molecule to give the condensation product (this was confirmed by isolating the intermediate **13**) followed by intramolecular cyclization through the nucleophilic attack of the imino group at the cyano, carbonyl or thione group respectively.

The reaction of compound **1** with arylidene(benzylidene, p-nitrobenzylidene, p-methoxybenzylidene)malononitrile, ethyl p-nitrobenzylidenecyanoacetate, and ethyl 2-naphthylidenecyanoacetate at r.t. in the presence of piperidine as a catalyst afforded 6-amino-2-(2-imino-2H-chromen-3-yl)-4-phenyl-4H-1,3-thiazine-5-carbonitrile **16a**, 6-amino-2-(2-imino-2H-chromen-3-yl)-4-(4-nitrophenyl)-4H-1,3-thiazine-5-carbonitrile **16b**, 6-amino-2-(2-imino-2H-chromen-3-yl)-4-(4-methoxyphenyl)-4H-1,3-thiazine-5-carbonitrile **16c**, ethyl 6-amino-2-(2-imino-2H-chromen-3-yl)-4-(4-nitrophenyl)-4H-1,3-thiazine-5-carboxylate **16d**,

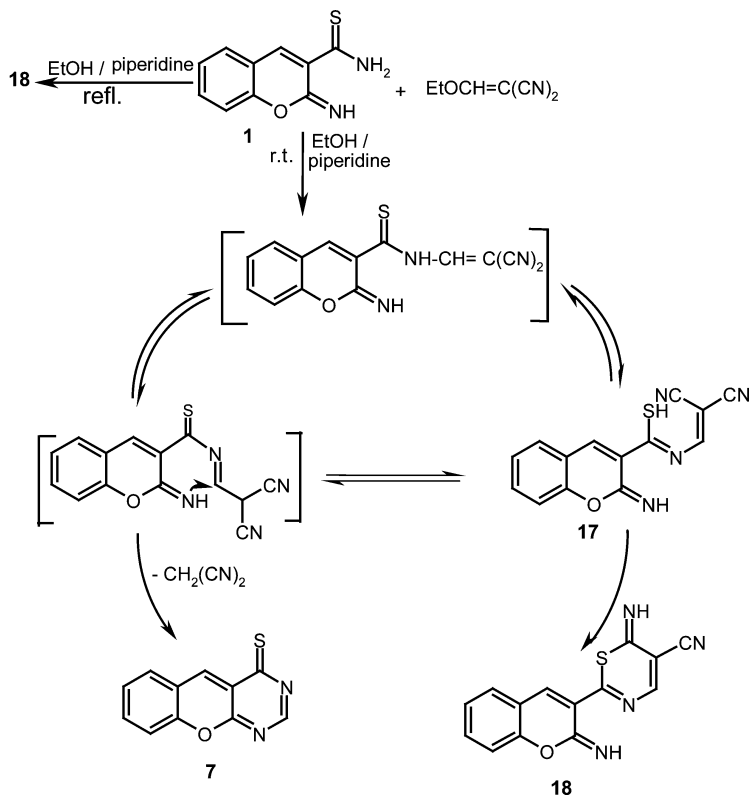


## SCHEME 6

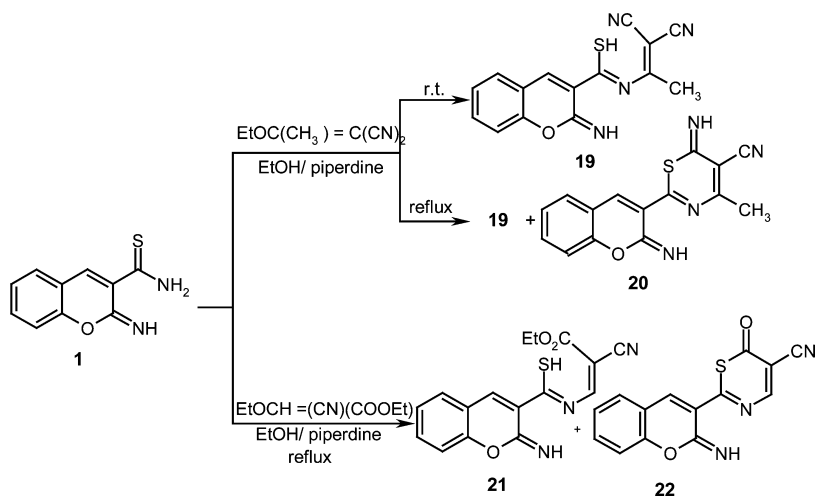
6-hydroxy-2-(2-imino-2H-chromen-3-yl)-4-(2-naphthyl)-4H-1,3-thiazine-5-carbonitrile **16e**, and ethyl 6-amino-2-(2-imino-2H-chromen-3-yl)-4-(2-naphthyl)-4H-1,3-thiazine-5-carboxylate **16f** (Scheme 6, Table I).

Moreover, compound **1** was subjected to react with ethoxymethylene-malononitrile in ethanol using piperidine as a catalyst at r.t. in which a mixture of N-(2,2-dicyanovinyl)-2-imino-2H-chromene-3-carbimidothioic acid **17**, benzo[g]quinazoline-4(10H)-thione **7**, and 6-imino-2-(2-imino-2H-chromen-3-yl)-6H-1,3-thiazine-5-carbonitrile **18** were isolated in 20%, 28%, and 48% yields, respectively. Carrying out the same reaction in refluxing ethanol afforded compound **18** only in an 80% yield. (Scheme 7, Table I).

Similarly, the reaction of compound **1** with (1-ethoxyethylidene)-malononitrile in ethanol in the presence of piperidine as a catalyst at r.t. gave the open structure **19** in a 26% yield. But the reaction of the same reagents in refluxing ethanol afforded a mixture of N-(2,2-dicyano-1-methylvinyl)-2-imino-2H-chromene-3-carbimidothioic acid **19** and 6-imino-2-(2-imino-2H-chromen-3-yl)-4-methyl-6H-1,3-thiazine-5-carbonitrile **20** in 30% and 51% yields, respectively. Also, the treatment of compound **1** with ethyl ethoxymethylenecyanoacetate in refluxing ethanol using a piperidine catalyst afforded N-[2-cyano-3-ethoxy-3-oxoprop-1-enyl]-2-imino-2H-chromene-3-carbimidothioic acid **21** and 2-(2-imino-2H-chromen-3-yl)-6-oxo-6H-1,3-thiazine-5-carbonitrile **22** (Scheme 8, Table I).



SCHEME 7



SCHEME 8

## CONCLUSION

This synthetic study reveals that the reaction of the starting material 2-imino-2H-chromene-3-thiocarboxamide with the active nitriles and their ylidenе derivatives depends on reaction conditions (temperature and time).

## EXPERIMENTAL

The synthesis of 2-phenyl-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione, 2-(4-methoxyphenyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione, 2-(4-nitrophenyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione and 2-(2-naphthyl)-2,3-dihydro-4H-chromeno[2,3-d]pyrimidine-4-thione **2a**–**d**

### General Procedure

To a solution of compound **1** (1.02 g, 0.005 mole) and benzaldehyde (0.50 mL), p-anisaldehyde (0.57 mL), p-nitrobenzaldehyde (0.75 g) or 2-naphthaldehyde (0.79 g) in tetrahydrofuran (50 mL), and 3 drops of piperidine were added. The reaction mixture was refluxed for 4 h. Compound **2d** was precipitated on heating, collected by filtration, and crystallized. The reaction mixture was left to cool where compounds **2a** and **2b** were separated, collected by filtration, and crystallized. The reaction mixture was evaporated, and the solid residue was washed with a water/ethanol mixture and crystallized where compound **2c** was obtained (Table I).

### The Synthesis of 4-(8,8a,10,11,12,13-Hexahydro-7H-5-oxa-9-thia-6,8-diazanaphtho[2,3,4-de]anthracen-7-yl)phenyl Methyl Ether **3b**

A mixture of compound **2b** (0.80 g, 0.0025 mole), cyclohexanone (0.26 mL, 0.0025 mole), and 2 drops of triethylamine and dioxane (20 mL) was stirred at r.t. for 1 h and then refluxed for 2 h. The reaction mixture was evaporated, and the residual solid was washed with a water/ethanol mixture and crystallized (Table I).

### The Synthesis of N-[(2-imino-2H-chromen-3-yl)-carbonothioyl]acetamide **4**

Compound **1** (1.02 g, 0.005 mole) was dissolved in glacial acetic acid (20 mL) and stirred for 2 h at r.t. The separated solid was collected by filtration, washed with ethanol, and crystallized (Table I).

### **The Synthesis of 2-Methyl-4H-chromeno[2,3-d]-pyrimidine-4-thione 5**

Compound **4** (0.615 g, 0.0025 mole) or compound **1** (1.02 g, 0.005 mole) was refluxed in acetic anhydride (10 mL) for 10 min. The reaction mixture was poured into an ice-water mixture, and the separated solid was filtered off, washed with water, and crystallized (Table I).

### **The Synthesis of N-formyl 2-Imino-2H-chromene-3-thiocarboxamide 6**

A solution of compound **1** (1.02 g, 0.005 mole) in 10 mL of formic acid (85%) was heated (60–65°C) for 10 min. The reaction mixture was evaporated in vacuo. The residual solid was washed with water and crystallized (Table I).

### **The Synthesis of Benzo[g]quinazoline-4(10H)-thione 7**

A mixture of compound **6** (0.697 g, 0.003 mole) and 2 drops of piperidine and 15 mL of ethanol was refluxed for 2 h. The reaction mixture was evaporated. The residual solid was washed with water and crystallized (Table I).

### **The Synthesis of 3-Imino-1-thioxo-2,3-dihydro-1H-chromeno[3,2-c]pyridine-4-carbonitrile 8, 2-Cyano-N-[(2-imino-2H-chromen-3-yl)carbonothioyl]acetamide 9, and 2-Cyano-N-[(2-imino-2H-chromen-3-yl)carbonothioyl]-ethanimidothioic Acid 10: General Procedure**

A solution of compound **1** (1.02 g, 0.005 mole) and malononitrile, (0.33 g, 0.005 mole), ethyl cyanoacetate (0.503 mL, 0.005 mole), or cyanothioacetamide (0.50 g, 0.005 mole) in ethanol (50 mL) was treated with 2 drops of piperidine. The reaction mixture was stirred for 15 min and left for 72 h. The precipitate was collected by filtration and crystallized from the proper solvent (Table I).

### **The Synthesis of (4-Thioxo-4H-chromeno[2,3-d]pyrimidin-2-yl)-acetonitrile 11**

A mixture of compound **1**, **9**, or **10** (0.002 mole), tetrahydrofuran (40 mL), and 2 drops of piperidine was refluxed for 3 h. The solvent was evaporated in vacuo. The solid residue was washed with water

followed by ethanol and crystallized from an ethanol/dioxane mixture (Table I).

**The Synthesis of 4-Amino-2-imino-2H-chromeno[2,3-b]-pyridine-3-carbonitrile 12, Ethyl 2-Cyano-3-imino-3-(2-imino-2H-chromen-3-yl)propanoate 13, 2-Hydroxy-4-imino-4H-chromeno[2,3-d]pyridine-3-carbonitrile 14, and 4-Imino-2-mercapto-4H-chromeno[2,3-b]pyridine-3-carbonitrile 15:**  
**General Procedure**

To a mixture of compound **1** (1.02 g, 0.005 mole), ethanol (50 mL), 2 drops of piperidine, malononitrile, and ethyl cyanoacetate or cyanothioacetamide (0.005 mole) was added. The reaction mixture was refluxed until the evolution of H<sub>2</sub>S gas ceased (5 h). Compound **13** was precipitated on heating, which was collected by filtration and crystallized. Compounds **12** and **15** were precipitated after cooling, collected by filtration, and crystallized. Compound **14** was isolated via the evaporation of the filtrate of compound **13** and washing the solid residue with water followed by crystallization (Table I).

**The Synthesis of Compounds 16a–f: General Procedure**

A mixture of compound **1** (0.51 g, 0.0025 mole) and ethanol (40 mL) containing 3 drops of piperidine was treated with benzylidenemalononitrile (0.39 g), p-nitrobenzylidenemalononitrile, (0.59 g), p-methoxybenzylidenemalononitrile (0.42 g), ethyl p-nitrobenzylidenecyanoacetate (0.555 g), or ethyl 2-naphthylidenecyanoacetate (0.527 g). The reaction mixture was stirred for 15 min and left for 48 h at r.t. The reaction mixture was evaporated in vacuo. The solid residue was washed with water and ethanol followed by crystallization to give compounds **16a**, **16b**, and **16d**. In the case of compounds **16c** and **16e**, the solid residue was treated with a CHCl<sub>3</sub>/pet. ether (40–60°C) mixture, and the precipitate was collected by filtration and crystallized. The filtrate of **16e** was evaporated, and the solid residue was crystallized to give **16f** (Table I).

**The Synthesis of N-(2,2-dicyanovinyl)-2-imino-2H-chromene-3-carbimidothioic Acid 17 and 6-Imino-2-(2-imino-2H-chromen-3-yl)-6H-1,3-thiazine-5-carbonitrile 18**

A mixture of compound **1** (2.04 g, 0.01 mole), ethoxymethylenemalononitrile (1.22 g, 0.01 mole), ethanol (50 mL), and 3 drops of piperidine was stirred for 30 min and left for 4 h at r.t. The precipitate was collected

by filtration, washed with ethanol, and crystallized to give compound **7**. The filtrate was left for 6 h, and the separated solid was filtered off, washed with ethanol, and crystallized to give compound **18**. The second filtrate was evaporated in vacuo, and the solid residue was washed with water, and pet. ether (60–80°C) and crystallized to give compound **17** (Table I).

### Synthesis of Compound 18: Direct Method

To a solution of compound **1** (0.005 mole) and ethoxymethylenemalononitrile (0.005 mole) in ethanol (30 mL), 3 drops of piperidine was added. The reaction mixture was refluxed for 2 h and left to cool. The precipitate was filtered off, washed with ethanol, and crystallized.

### The Synthesis of N-(2,2-dicyano-1-methylvinyl)-2-imino-2H-chromene-3-carbimidothioic Acid **19**

A solution of compound **1** (0.51 g, 0.0025 mole) and 1-ethoxyethylidenemalononitrile (0.33 g, 0.0025 mole) in ethanol (30 mL) was treated with two drops of piperidine. The reaction mixture was stirred for 6 h at r.t. and left for 24 h. The precipitate was collected by filtration, washed with ethanol, and crystallized.

### The Synthesis of and 6-Imino-2-(2-imino-2H-chromen-3-yl)-4-methyl-6H-1,3-thiazine-5-carbonitrile **20**

A mixture of compound **1** and 1-ethoxyethylidenemalononitrile (0.0025 mole), ethanol (40 mL), and 2 drops of piperidine was refluxed for 3 h and left to cool. The precipitate was filtered off, washed with ethanol, and crystallized to give compound **19**. The filtrate was evaporated, and the solid residue was washed with water and pet. (60–80°C) and crystallized to give compound **20**.

### The Synthesis of N[2-cyano-3-ethoxy-3-oxoprop-1-enyl]-2-imino-2H-chromene-3-carbimidothioic Acid **21** and 2-(2-Imino-2H-3-chromen-3-yl)-6-oxo-6H-1,3-thiazine-5-carbonitrile **22**

A solution of compound **1** (0.51 g, 0.005 mole) and ethyl ethoxymethylenecyanoacetate (0.42 g, 0.005 mole) in ethanol (40 mL) was treated with 2 drops of piperidine. The reaction mixture was refluxed for 5 h. The reaction mixture was concentrated to half its volume and left to cool. The precipitate was collected by filtration and crystallized to give compound **22**. The filtrate was evaporated. The residual



solid was washed with pet. ether (60–80°C) and crystallized to give compound **21**.

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