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Arynes and Cyclic Alkynes as Synthetic Building Blocks for Stereodefined Quaternary Centers

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ABSTRACT: We report a facile method to synthesize stereodefined quaternary centers from reactions of arynes and related strained intermediates using β -ketoester-derived substrates. Conversion of β -ketoesters to chiral enamines is followed by reaction with in situ generated strained arynes or cyclic alkynes. Hydrolytic workup provides the arylated or alkenylated products in enantiomeric excesses as high as 96%. We also describe the one-pot conversion of a β -ketoester substrate to the corresponding enantioenriched α -arylated product. Computations show how chirality is transferred from the *N*-bound chiral auxiliary to the final products. These are the first theoretical studies of aryne trapping by chiral nucleophiles to set new stereocenters. Our approach provides the most general known solution to the challenging problem of stereoselective β -ketoester arylation/alkenylation, with formation of a quaternary center.

INTRODUCTION

Arynes have historically been avoided as synthetic intermediates as a result of their high reactivity.^{1,2} However, recent studies have demonstrated that arynes can be generated under mild reaction conditions,³ trapped regioselectively using predictive models,⁴ and employed in a host of synthetic applications. The utility of arynes is evident, as they have now been used to synthesize natural products, ligands, materials, agrochemicals, and pharmaceutical agents (e.g., **1–3**, Figure 1a).^{1,5}

The majority of reported synthetic applications of arynes are intermolecular reactions that lead to achiral or racemic products.⁶ We questioned if arynes and related strained intermediates could instead serve as building blocks to generate enantioenriched products bearing quaternary centers. Only two methodologies leading to intermolecular, stereoselective aryne trappings have been reported and are limited to the synthesis of tertiary stereocenters.^{7,8}

We considered the reaction manifold in which β-ketoesters 4⁹ would be trapped with strained alkynes 5, to give their corresponding α -arylated products **6** with formation of a quaternary stereocenter (Figure 1b).¹⁰ As prior efforts to achieve this direct functionalization in a racemic sense were accompanied by an undesired C-C bond fragmentation,¹¹ we considered a two-step, alternative approach. First, β -ketoesters 4 would be treated with amines 7 to afford the corresponding enamines **8**.¹² Trapping of the enamines **8** with in situ-generated arynes (or strained cyclic alkynes) would give the α -arylated or alkenylated products 6 after hydrolysis in the same pot.¹³ The use of a chiral amine (i.e., 7) in this process would ultimately give rise to enantioenriched products 6 bearing quaternary stereocenters.^{10,14} It should be noted that the enantioselective α-arylation of β-ketoesters has remained a challenging svnthetic problem.¹⁵ Promising developments include the use of hypervalent iodine reagents (racemic or modest enantioenrichment),¹⁶ the Cu-catalyzed, enantioselective coupling of 2methylacetates with 2-iodotrifluoroacetanilides,¹⁷ and the Pdcatalyzed α -arylation of malonates and cyanoacetates (racemic).¹⁸ A general method for the stereocontrolled α -arylation or -alkenylation of β -ketoesters has not been disclosed.

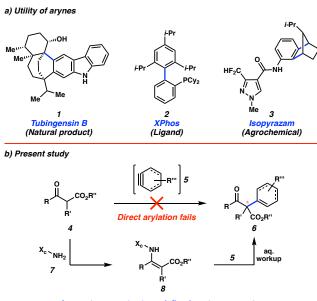




Figure 1. Synthetic applications of arynes and strategy for the stereoselective arylation of β -ketoesters.

We report the development of the synthetic sequence shown in Figure 1b, which provides a facile method to achieve the stereoselective α -arylation/alkenylation of β -ketoesters.⁹ In addition to providing access to adducts bearing stereodefined quaternary centers, this methodology demonstrates that highly reactive arynes and related intermediates can serve as building blocks to access enantioenriched products by intermolecular trapping. In addition, the origins of stereoselectivity have been revealed by a computational investigation of these reactions.

RESULTS AND DISCUSSION

benzvlamine

Na₂SO₄

benzene

(quant. yield)

D₂Et

13

Development of a Racemic and Stereospecific Reaction to Generate Quaternary Centers. To commence our studies, we selected β -ketoester 9 as an initial substrate for the twostep arylation procedure (Figure 2). As the use of enamines and arynes to construct quaternary stereocenters was unknown, we first pursued a racemic transformation. Benzylamine was condensed with ketoester 9 to yield enamine 10 quantitatively. Next, enamine 10 was used to trap benzyne, which was generated in situ from silyl triflate 11 (1.5 equiv) in DME at 30 °C (6 h). After quenching with 1 M HCl_(aq), we were delighted to obtain the desired α -arylated product 12 in 92% yield with introduction of a quaternary center.¹⁹ Furthermore, we surveyed several other highly reactive intermediates to gauge the possibility of utilizing substituted benzynes and cyclic alkynes. The use of fused arynes 2,3-naphthalyne and *N*-Boc-4,5-indolyne²⁰ provided arylated products **13** and **14**, respectively.²¹ In addition, trapping with known heterocyclic alkynes²² delivered tetrahydropyridine 15 and dihydropyran 16 in 67% and 74% yields, respectively. Regioselectivities for the formation of 14-16 were in accord with the distortion/interaction model.^{20,22} These results represent a facile means to install aryl and vinyl moieties onto a cyclic βketoester with quaternary center formation.

i, CsF, DME, 30 °C

ii. 1 M HCI_(aq)

15

(92% yield)

12

16

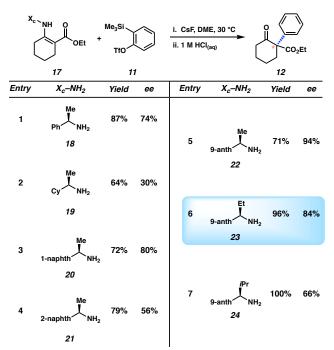
92% yield 42% yield ^a 67% yield 74% yield Figure 2. Discovery of methodology for the arylation/vinylation of β-ketoesters in racemic fashion. Conditions for enamine formation: ketoester 9 (1.0 equiv), benzylamine (1.5 equiv), Na₂SO₄ (5:1 by wt.), benzene (0.7 M), 80 °C, 16 h. Conditions for arylation/alkenylation unless otherwise stated: i. enamine 10 (1.0 equiv), silyl triflate 11 (1.5 equiv), CsF (7.5 equiv), DME (0.1 M), 30 °C, 6 h; ii. 1 M HCl_(aq), 23 °C, 30 min. Yields reflect the average of two isolation experiments. ^a Aryne trapping performed for 3 h.

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Having developed the racemic arylation/alkenylation reaction, we turned our attention to the discovery of a diastereoselective variant to access enantioenriched products (Table 1).²³ Thus, a series of enantioenriched chiral amines, readily prepared using Ellman auxiliary chemistry (i.e., **18–24**),^{24,25} were condensed with ketoester **9** to access enamines **17**. Subsequent arylation under the conditions depicted in Figure 2 furnished **12** in enantioenriched form. Utilization of phenyl derivative **18** resulted in the formation of **12** in good yield and 74% enantiomeric excess (ee) (entry 1). Employing amine **19**, bearing a cyclohexyl moiety, gave the desired product in a lower ee of 30% (entry 2). Recognizing the importance of the aryl fragment, we examined 1- and 2-naphthyl derived amines **20** and **21**, which provided **12** in 80% and 56% ee (entries 3 and 4, respectively). With improved results in the case of **20**, we examined anthracenyl amines **22–24** (entries 5–7). As the use of ethyl derivative **23** furnished **12** with the best combination of yield and ee (entry 6), **23** was selected for subsequent studies. It should be noted that the Ellman-approach provides both enantiomers of **23**, which, in turn, permits access to each enantiomer of the products depicted subsequently.^{26,27}

 Table 1 Survey of chiral auxiliaries to give optically enriched ketone 12.^a



^{*a*} Reaction conditions: i. enamine **17** (1.0 equiv), silyl triflate **11** (1.5 equiv), CsF (7.5 equiv), DME (0.1 M), 30 °C, 6 h; ii. 1 M $HCl_{(aq)}$, 23 °C, 30 min.

Scope of Methodology. With a suitable chiral amine identified, we evaluated several cyclic alkynes in the stereoselective arylation/alkenylation reaction to form quaternary stereocenters (Figure 3). The reaction was tolerant of substituted benzyne intermediates and extended aryl units, giving rise to arylated products **28** and **13**, respectively.²⁸ Moreover, trapping of an indolyne intermediate delivered heterocycle-containing product **14**. When applied to non-aromatic, strained alkynes, the methodology provided alkenylated products in good yields and stereoselectivities. For example, trapping of cyclohexyne²⁹ provided cyclohexene derivative **29** in good yield and 86% ee. Additionally, by employing heterocyclic alkynes, products **15** and **16** were obtained in excellent yields and comparable stereoselectivities.

As shown in Figure 4, the methodology is also tolerant of variation in the nucleophilic component. For example, replacement of the ethyl ester with a benzyl ester in the parent substrate gave rise to arylated product **32** in 71% yield and 86% ee. Furthermore, piperidinone and tetrahydropyranone derivatives could be employed to access heterocyclic products (i.e., **33–35**). Enamines derived from 7-membered ring β -ketoesters could also be utilized, as shown by the formation of arylated products **36** and **37** with excellent stereoselectivity.

Lastly, the formation of ketoester **38** demonstrates the viability of utilizing this methodology for the α -arylation of acyclic β -ketoesters.

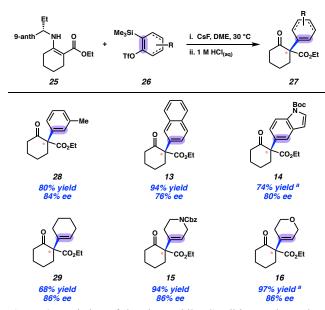


Figure 3. Variation of the electrophile. Conditions unless otherwise stated: i. enamine **25** (1.0 equiv), silyl triflate **26** (1.5 equiv), CsF (7.5 equiv), DME (0.1 M), 30 °C, 6 h; ii. 1 M HCl_(aq), 23 °C, 30 min. Yields reflect the average of two isolation experiments. ^{*a*} Aryne or cyclic alkyne trapping performed for 3 h.

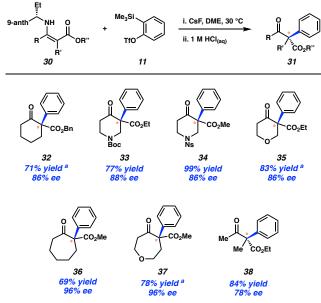


Figure 4. Variation of the nucleophilic component 30 in the trapping with 11. Conditions unless otherwise stated: i. enamine 30 (1.0 equiv), silyl triflate 11 (1.5 equiv), CsF (7.5 equiv), DME (0.1 M), 30 °C, 3 h; ii. 1 M HCl_(aq), 23 °C, 30 min. Yields reflect the average of two isolation experiments. ^{*a*} Aryne trapping performed for 6 h.

One-pot, Stereoselective Arylation. As one final application of this methodology, we developed a one-pot variant of the methodology to convert ketoester substrate **39** to α - arylated product **36**, with recovery of the chiral auxiliary (Figure 5). β-Ketoester **39** was reacted with amine **23** to generate enamine **40** in situ. Addition of CsF and silyl triflate **11**, followed by stirring at 30 °C for 6 h, and subsequent acid-mediated hydrolysis yielded the desired α-arylated product **36**. When performed on mmol scale, the reaction gave **36** in 68% yield and 92% ee, in addition to 67% recovered amine **23**. This protocol provides a promising means to achieve the direct, asymmetric α-arylation of β-ketoesters.

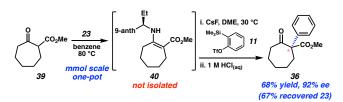


Figure 5. One-pot, mmol-scale arylation reaction to furnish 36. Conditions for enamine formation: ketoester 39 (1.0 equiv), amine 23 (1.0 equiv), benzene (0.7 M), 80 °C, 16 h, followed by evaporation of benzene solvent. Conditions for arylation: i. silyl triflate 11 (1.5 equiv), CsF (7.5 equiv), DME (0.1 M), 30 °C, 6 h; ii. 1 M $HCl_{(aq)}$, 23 °C, 12 h.

Computational Analysis of Chirality Transfer. Density functional theory (DFT) calculations were performed to understand how stereochemical information is transferred from the chiral auxiliary to the newly formed quaternary stereocenter. Our laboratories have studied reactions of arynes in nucle-ophilic additions using computations,⁴ but no theoretical studies of aryne trapping by chiral nucleophiles to set new stereocenters have been reported. All calculations described here utilize the M06-2X³⁰/def2-TZVPP–SMD³¹ (diethyl ether)//B3LYP³²/6-31+G(d,p) level of theory (see the SI for a discussion of the computational methods and results with other density functionals).

We first calculated the stereo-controlling transition structures for the reaction of benzvne and the S enantiomer of the chiral enamine derived from amine 20, which possesses the 1naphthyl group at the chiral center. The stereochemistrycontrolling transition structures are shown in Figure 6. Each pathway has a low barrier ($\Delta G^{\ddagger} = 9.6$ and 11.6 kcal/mol, respectively). TS1 leads to the experimentally preferred stereoisomer, (S)-12, whereas TS2 yields the minor enantiomer, (R)-12.³³ The difference in free energy of activation ($\Delta\Delta G^{\ddagger}$) is 2.0 kcal/mol, within error of the experimentally observed selectivity of 80% ee ($\Delta\Delta G^{\ddagger}$ = 1.3 kcal/mol). In both **TS1** and **TS2**, an intramolecular hydrogen bond between the NH and ester carbonyl is present. Axial-attack by benzyne occurs in both cases, as expected from the preference the forming bond to be staggered with respect to the allylic CH bonds (known previously as the Fürst-Plattner rule).³⁴ While attack is axial in both cases, and the chiral group is in its favored conformation, the interaction of the CH at the stereogenic center is disfavorable in the minor TS.

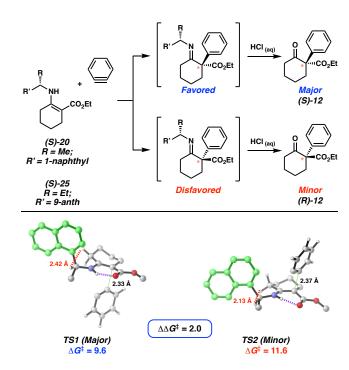


Figure 6. Lowest-energy transition structures TS1 and TS2 for the addition of benzyne and the chiral enamine derived from amine 20 (M06-2X/def2-TZVPP–SMD (diethylether)//B3LYP/6-31+G(d,p)). Free energy activation barriers (ΔG^{\ddagger}) are compared to separated intermediates. The difference in free energies of activation ($\Delta \Delta G^{\ddagger}$), relative to TS1, are reported in kcal/mol.

In **TS2**, there is a close-contact H–H interaction of 2.1 Å between the chiral center of the enamine and methylene of the six-membered ring. This contact is alleviated in **TS1**, with an H–H interaction distance of 2.4 Å. Our laboratory has previously examined the transmission of chirality in the reaction of a similar chiral enamine with acrylonitrile, which similarly revealed the importance of torsional interactions between forming bonds and allylic bonds.³⁴ In that case, the same conformations and their energies were found for the chiral enamine with a phenyl ring instead of naphthyl. Torsional strain³⁵ controls the stereoselectivity of this reaction, where the enamine conformations remain the same for both stereoisomeric transition states.

One might expect that the stereoselectivity cannot be modulated by the size of the substituent, but as found here, enamine 25 has improved enantioselectivity with the larger 9anthracenvl substituent. We calculated the stereochemistrycontrolling transition structures for the reaction of chiral enamine 25 and benzyne using methyl groups in place of ethyl groups to simplify computations.³⁶ The two lowest-energy transition structures leading to the major and minor stereoisomers are shown in Figure 7. TS3 leads to the experimentally preferred stereoisomer, (S)-12, whereas TS4 yields the minor enantiomer. The difference in free energy of activation ($\Delta \Delta G^{\ddagger}$) is 2.5 kcal/mol, within error of the experimentally observed selectivity of 84% ee ($\Delta\Delta G^{\ddagger}$ = 1.5 kcal/mol) and 0.5 kcal/mol higher than observed with 20. Axial-attack by benzyne again occurs for the two half-chair conformers of the cyclohexene. However, here the conformation of the enamine stereogenic group differs between the stereoisomeric transition structures. Whereas the conformation of **TS3** is analogous to that in **TS1**, the enamine in TS4 is in a higher-energy conformation because the face being blocked to form the (R)-isomer is obstructed by the anthracene group. This conformation yields a higher-energy penalty than the torsional strain found in **TS2**, which enables an increase in enantiospecificity.

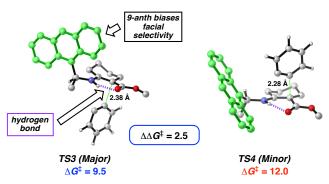


Figure 7. Lowest-energy transition structures **TS3** and **TS4** for the addition of benzyne and chiral enamine **25** (M06-2X/def2-TZVPP–SMD (diethylether)//B3LYP/6-31+G(d,p)). Free energy activation barriers (ΔG^{\ddagger}) are compared to separated intermediates. The difference in free energies of activation ($\Delta \Delta G^{\ddagger}$), relative to **TS1**, are reported in kcal/mol.

CONCLUSIONS

We have developed the first methodology that allows for arynes and related strained intermediates to be trapped intermolecularly for the formation of stereodefined quaternary centers. The strategy relies on the facile conversion of βketoesters to chiral enamines, which undergo nucleophilic trapping of in situ generated strained arynes or cyclic alkynes. Hydrolysis in the same pot provides the arylated products in good to excellent enantiomeric excesses (up to 96% ee). This strategy circumvents a previously known undesired C-C bond fragmentation, while providing the most general solution to the challenging problem of stereoselective β-ketoester arylation/alkenylation, with formation of a quaternary center. In addition, a one-pot procedure for the conversion of a β ketoester substrate to the corresponding enantioenriched α arylated product was developed. Finally, computations show how chirality transfer is achieved from the chiral auxiliary to the final products, a type of conformational transmission operating in the trapping of arynes by chiral nucleophiles. We expect these studies will enable further developments of intermolecular, stereoselective reactions of highly reactive aryne and cyclic alkyne intermediates.

ASSOCIATED CONTENT

Supporting Information Available. Detailed experimental procedures, compound characterization data, and computational analysis. This material is available free of charge via the Internet at <u>http://pubs.acs.org</u>.

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Author Contributions

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REFERENCES

¹ For reviews regarding benzyne and related reactive intermediates, see: a) Pellissier, H.; Santelli, M. *Tetrahedron* 2003, *59*, 701–730. b) Wenk, H. H.; Winkler, M.; Sander, W. *Angew. Chem., Int. Ed.* 2003, *42*, 502–528. c) Sanz, R. *Org. Prep. Proced. Int.* 2008, *40*, 215–291. d) Bronner, S. M.; Goetz, A. E.; Garg, N. K. *Synlett* 2011, 2599–2604. e) Tadross, P. M.; Stoltz, B. M. *Chem. Rev.* 2012, *112*, 3550–3557. f) Gampe, C. M.; Carreira, E. M. *Angew. Chem., Int. Ed.* 2012, *51*, 3766–3778. g) Bhunia, A.; Yetra, S. R.; Biju, A. T. *Chem. Soc. Rev.* 2012, *41*, 3140–3152. h) Yoshida, H.; Takaki, K. *Synlett* 2012, 1725–1732. i) Dubrovskiy, A. V.; Markina, N. A.; Larock, R. C. *Org. Biomol. Chem.* 2013, *11*, 191–218. j) Wu, C.; Shi, F. *Asian J. Org. Chem.* 2013, *52*, 2655–2656. l) Goetz, A. E.; Garg, N. K. *J. Org. Chem.* 2014, *79*, 846–851. m) Goetz, A. E.; Shah, T. K.; Garg, N. K. *Chem. Commun.* 2015, *51*, 34–45. n) Yoshida, S.; Hosoya, T. *Chem. Lett.* 2015, *44*, 1450–1460. o) Bhojgude, Sachin Chem. And Schemer 2017. Chem. 2016. Chem. Context.

Suresh; Bhunia, A.; Biju, A. T. Acc. Chem. Res. 2016, 49, 1658–1670.
 ² For the quantification of benzyne's high electrophilicity, see: Fine Nathel, N. F.; Morrill, L. A.; Mayr, H.; Garg, N. K. J. Am. Chem. Soc.

2016, 138, 10402–10405.
 ³ For Kobayashi's generation of arynes from *o*-silyltriflate precursors, see:

Himeshima, Y.; Sonoda, T.; Kobayashi, H. Chem. Lett. 1983, 1211-1214. 35 For the aryne distortion/interaction model, see: a) Cheong, P. H.-Y.; 36 Paton, R. S.; Bronner, S. M.; Im, G.-Y. J.; Garg, N. K.; Houk, K. N. J. 37 Am. Chem. Soc. 2010, 132, 1267-1269. b) Im, G.-Y. J., Bronner, S. M.; Goetz, A. E.; Paton, R. S.; Cheong, P. H.-Y.; Houk, K. N.; Garg, N. K. J. 38 Am. Chem. Soc. 2010, 132, 17933-17944. c) Goetz, A. E.; Bronner, S. M.; 39 Cisneros, J. D.; Melamed, J. M.; Paton, R. S.; Houk, K. N.; Garg, N. K. 40 Angew. Chem., Int. Ed. 2012, 51, 2758–2762. d) Bronner, S. M.; Mackey, J. L.; Houk, K. N.; Garg, N. K. J. Am. Chem. Soc. 2012, 134, 13966-41 13969. e) Medina, J. M.; Mackey, J. L.; Garg, N. K.; Houk, K. H. J. Am. 42 Chem. Soc. 2014, 136, 15798-15805. f) Picazo, E.; Houk, K. N.; Garg, N. 43 K. Tetrahedron Lett. 2015, 56, 3511-3514.

⁵ For select examples of synthetic applications, see: a) Mauger, C. C.; Mignani, G. A. Org. Proc. Res. Dev. 2004, 8, 1065–1071. b) Lin, J. B.; Shah, T. K.; Goetz, A. E.; Garg, N. K.; Houk, K. N. J. Am. Chem. Soc. 2017, 139, 10447–10455. c) Surry, D. S.; Buchwald, S. L. Angew.
47 Chem., Int. Ed. 2008, 47, 6338–6361. d) Ross, S. P.; Hoye, T. R. Nat. Chem. 2017, 9, 523–530. e) Corsello, M. A.; Kim, J.; Garg, N. K. Nat. Chem. 2017, 9, 944–949.

⁶ Intramolecular aryne trappings, although less common, have been used to construct complex natural product frameworks. See references 1, 5, and the following recent examples: a) Goetz, A. E.; Silberstein, A. L.; Corsello, M. A. J. Am. Chem. Soc. 2014, 136, 3036–3039. b) Neog, K.; Borah, A.; Gogoi, P. J. Org. Chem. 2016, 81, 11971–11977. c) Corsello, M. A.; Kim, J. Garg, N. K. Nat. Chem. 2017, 9, 944–949. d) Ross, S.P.; Hoye, T.R. Nat. Chem. 2017, 9, 523–530. e) Neumeyer, M.; Kopp, J.; Brückner, R. Eur. J. Org. Chem. 2017, 2883–2915.
⁷ For the intermediate transmiss of heavien with dimension the Org.

⁷ For the intermolecular trapping of benzyne with dienes bearing the Oppolzer sultam (with later cleavage of the auxiliary) to generate tertiary

stereocenters, see: a) Dockendorff, C.; Sahli, S.; Olsen, M.; Milhau, L.; Lautens, M. *J. Am. Chem. Soc.* **2005**, *127*, 15028–15029. b) Webster, R.; Lautens, M. *Org Lett.* **2009**, *11*, 4688–4691.

⁸ For the intermolecular trapping of arynes with Schöllkopf reagents (with later hydrolysis) to generate tertiary stereocenters, see: a) Jones, E. P.; Jones, P.; Barrett, A. G. M. *Org. Lett.* **2011**, *13*, 1012–1015. b) Jones, E. P.; Jones, P.; White, A. J. P.; Barrett, A. G. M. *Beilstein J. Org. Chem.* **2011**, *7*, 1570–1576.

 9 β-keto esters are commonly seen in bioactive molecules, so methods to synthesize functionalized derivatives are valuable. A Reaxys search reveals there are 53,683 known β-keto esters with biological activity. January 24, 2018.

¹⁰ Methodologies to generate quaternary stereocenters, especially with control of absolute stereochemistry, remain highly sought after. For pertinent reviews, see: a) Quasdorf, K. W.; Overman, L. E. *Nature* **2014**, *516*, 181–191. b) Liu, Y.; Han, S.-J.; Liu, W.-B.; Stoltz, B. M. *Acc. Chem. Res.* **2015**, *48*, 740–751. c) Shockley, S. E.; Holder, J. C.; Stoltz, B. M. *Org. Process Res. Dev.* **2015**, *19*, 974–981. d) Zeng, X.-P.; Cao, Z-Y.; Wang, Y-H.; Zhou, F.; Zhou, J. *Chem. Rev.* **2016**, *116*, 7330–7396.

¹¹ For the α -arylation of β -ketoesters, with concomitant C–C bond fragmentation, see: Tambar, U. K.; Stoltz, B. M. J. Am. Chem. Soc. **2005**, *127*, 5340–5341.

 12 Enamines derived from β -ketoesters were considered well suited, as they are stable on neutral alumina and can be purified prior to use.

¹³ For pioneering studies of racemic enamine arylations, see: a) Kuehne, M. E. J. Am. Chem. Soc. **1962**, 82, 837–847. For the α -arylation of enamines with arynes to give functionalized achiral enamines, see: b) Ramtohul, Y. K.; Chartrand, A. Org. Lett. **2007**, 9, 1029–1032. c) Li, R.; Wang, X.; Wei, Z.; Wu, C.; Shi, F. Org. Lett. **2013**, 15, 4366–4369.

¹⁴ For select previous studies on the enantioselective functionalization of enamines bearing chiral auxiliaries, see: a) Tomioka, K.; Ando, K.; Takemasa, Y.; Koga, K. J. Am. Chem. Soc. **1984**, 106, 2718–2719. b) Christoffers, J.; Mann, A. Chem. Eur. J. **2001**, 7, 1014–1027. c) Camara, C.; Joseph, D.; Dumas, F.; d'Angelo, J.; Chiaroni, A. Tetrahedron Lett. **2002**, 43, 1445–1448. d) Fujimoto, T.; Endo, K.; Tsuji, H.; Nakamura, M.; Nakamura, E. J. Am. Chem. Soc. **2008**, 130, 4492–4496.

¹⁵ A related breakthrough, albeit not using β-ketoesters, is the α-arylation of oxindoles using Pd catalysis; see: a) Altman, R. A.; Hyde, A. M.; Huang, X.; Buchwald, S. L. J. Am. Chem. Soc. **2008**, 130, 9613–9620. b) Taylor, A. M.; Altman, R. A.; Buchwald, S. L. J. Am. Chem. Soc. **2009**, 131, 9900–9901. c) Li, P.-F.; Buchwald, S. L. Angew. Chem., Int. Ed. **2011**, 50, 6396–6400.

¹⁶ a) Beringer, F. M.; Forgione, P. S.; Yudis, M. D. *Tetrahedron* **1960**, *8*, 49–63. b) Ochiai, M.; Kitagawa, Y.; Takayama, N.; Takaoka, Y.; Shiro, M. J. Am. Chem. Soc. **1999**, *121*, 9233–9234. c) Oh, C. H.; Kim, J. S.; Jung, H. H. J. Org. Chem. **1999**, *64*, 1338–1340.

¹⁷ Xie, X.; Chen. Y.; Ma. D. J. Am. Chem. Soc. 2006, 128, 16050–16051...

¹⁸ For the Pd-catalyzed α -arylation of malonates, see: Beare, M. A.; Hartwig, J. F. *J. Org. Chem.* **2002**, *67*, 541–555.

¹⁹ By employing the *N*-deutero derivative of **10** in this transformation, we observe deuterium incorporation on the ortho position of the aromatic ring of **12**. For details, see the SI.

²⁰ For the synthesis and regioselective trappings of indolynes accessed from silyltriflate precursors, see references 4a and 4b.

²¹ For reasons not fully understood, the racemic arylation of the indolyne was lower yielding than the corresponding stereoselective arylation reaction.

²² For the synthesis and regioselective trappings of these strained cyclic alkynes, see: a) McMahon, T. C.; Medina, J. M.; Yang, Y.-F.; Simmons, B. J.; Houk, K. N.; Garg, N. K. *J. Am. Chem. Soc.* 2015, *137*, 4082–4085.
b) Shah, T. K.; Medina, J. M.; Garg, N. K. *J. Am. Chem. Soc.* 2016, *138*, 4948–4954.

²³ Other strategies for this transformation were pursued, such as the use of Cu/BOX as a catalyst for the arylation reaction. Additionally, many other classes of amines, such as amino acids and amino acid derivatives were examined in the arylation reaction, but led to either poor yields and/or poor stereochemical outcomes.

²⁴ For a review on Ellman's chiral sulfinamides, see: Ellman, J. A.; Owens, T. D.; Tang, T. P. *Acc. Chem. Res.* **2002**, 35, 984. Procedure followed to synthesize chiral anthracenyl amines: Rodriguez–Hernandez, R.; Hernandez–Castillo, T.; Huizar–Trejo, K. E. *Synthesis* **2011**, *17*, 2817–2821.

²⁵ Chiral auxiliaries have seen widespread use to install important stereocenters in drugs and complex targets. For a recent review, see: Farina, V.; Reeves, J. T.; Senanayake, C. H.; Song, J. J. Chem. Rev. 2006, 106, 2734-2793. For examples using chiral sulfinamides, see: a) Pflum, D. A.; Krishnamurthy, D.; Han, Z.; Wald, S. A.; Senanayake, C. H. Tetrahedron Lett. 2002, 43, 923-926. b) Han, Z. S.; Herbage, M. A.; Mangunuru, H. P. R.; Xu, Y.; Zhang, L.; Reeves, J. T.; Sieber, J. D.; Li, Z.; DeCroos, P.; Zhang, Y.; Li, G.; Li, N.; Ma, S.; Frinberg, N.; Wang, X.; Goyal, N.; Krishnamurthy, D.; Lu, B.; Song, J. J.; Wang, G.; Senanayake, C. H. Angew. Chem., Int. Ed. 2013, 52, 6713-6717. For examples of chiral auxiliaries used in drug development see: Zhang, W.-Y.; Sun, C.; Hunt, D.; He, M.; Deng, Y.; Zhu, Z.; Chen, C.-L.; Katz, C. E.; Niu, J.; Hogan, P. C.; Xiao, X.-Y.; Dunwoody, N.; Ronn, M. Org. Process Res. Dev. 2016, 20, 284-296

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- ²⁶ For the crystallographic data used to assign the absolute stereochemistry 13 of amine 23 and its enantiomer, see the SI. 14
- ²⁷ The choice of enantiomer of 23 used in subsequent experiments was made arbitrarily, as both were prepared in gram quantities. 15
- ²⁸ The regioselective formation of **28** is consistent with previously ob-16 served trends in the trapping of 3-methylbenzyne; see: Aithagani, S. K.; 17 Dara, S.; Munagala, G.; Aruri, H.; Yadav, M.; Sharma, S.; Vishwakarma, 18 R. A.; Singh, P. P. Org. Lett. 2015, 17, 5547-5549.
- ²⁹ For a study involving the generation and trapping of cyclohexyne, see: 19 Medina, J. M.; McMahon, T. C.; Jiménez-Osés, G.; Houk, K. N.; Garg, N. 20 K. J. Am. Chem. Soc. 2014, 136, 14706-14709. 21
 - ³⁰ Zhao, Y.; Truhlar, D. G. *Theor. Chem. Acc.* **2008**, *120*, 215–241.
- ³¹ Marenich, A. V.; Cramer, C. J.; Truhlar, D. G. J. Phys. Chem. B 2009, 22 23 113.6378-6396.
- a) Vosko, S. H.; Wilk, L.; Nusair, M. Can. J. Phys. 1980, 58, 24 1200-1211. b) Lee, C.; Yang, W.; Parr, R. G. Phys. Rev. B: Condens. 25 Matter Mater. Phys. 1988, 37, 785-789. c) Becke, A. D. J. Chem. Phys. 1993, 98, 5648-5652. d) Stephens, P. J.; Devlin, F. J.; Chabalowski, C. F.; 26 Frisch, M. J. J. Phys. Chem. 1994, 98, 11623-11627. 27
- ³³ The stereochemical outcome predicted by the computational analysis 28 was validated experimentally. For the determination of product absolute 29 stereochemistry, see the SI.
- Lucero, M. J.; Houk, K. N. J. Am. Chem. Soc. 1997, 119, 826-827. 30
 - ³⁵ Wu, Y. D.; Houk, K. N.; Paddon-Row, M. N. Angew. Chem., Int. Ed. Engl. 1992, 31, 1019-1021.
 - ³⁶ From the results provided in Table 1, entries 5 and 6, the use of methyl vs ethyl has a negligible impact on stereoselectivity.

SYNOPSIS TOC

