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A CONCISE SYNTHESIS OF THE AB-RING FRAGMENT OF (-)-GAMBIEROL

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Abstract – We describe herein a concise synthesis of the AB-ring fragment of gambierol, wherein silver(I) trifluoromethanesulfonate-catalyzed 6-*endo* cyclization of a hydroxy ynone was exploited for the formation of the A-ring.

This paper is dedicated to Professor Ei-ichi Negishi on the occasion of his 77th birthday.

(-)-Gambierol (1, Figure 1) is a structurally complex marine polycyclic ether natural product that was isolated from the cultured cells of the ciguatera causative dinoflagellate Gambierdiscus toxicus by Satake and co-workers.¹ The entire structure of **1** was elucidated by extensive 2D-NMR analysis and application of a chiral anisotropic reagent. Satake et al. have reported that gambierol exhibits potent lethal toxicity against mice with a minimal lethal dose value of 50 μ g/kg (ip) and that the neurological symptoms caused in mice resemble those shown by ciguatoxins, suggesting the possible role of 1 in ciguatera seafood poisoning. Unfortunately, detailed biological studies on 1 had been precluded for almost a decade, due to the extreme natural scarcity of **1**. Motivated by the structural complexity as well as the biological aspects, we have successfully completed the first total synthesis of 1 and enabled material supply for extensive biological investigations.^{2,3} Consequently, we have identified that 1 inhibits voltage-gated potassium channels (Kv channels) in mice taste cells in low nanomolar concentrations,⁴ while **1** acts as a weak partial agonist of voltage-gated sodium channels in human neuroblastoma cells.⁵ Snyders and co-workers have shown that 1 selectively inhibits the Kv1 and Kv3 subfamilies and that 1 binds to the previously undescribed binding site present between the S5 and S6 segments of Kv3.1 channels.⁶ In addition, we have recently found that the EFGH-ring domain of 1 plays a crucial role in inhibiting voltage-gated potassium currents and that, in an *in vitro* model of Alzheimer's disease, 1 and its truncated analogues are

able to lower the amyloid β (A β) and hyperphosphorylated tau levels with a possible implication of Kv channel inhibition.⁷ Undoubtedly, **1** represents an intriguing molecular probe for the functional analysis of Kv channels as well as for understanding downstream events of Kv channel inhibition.



Figure 1. Structure of (–)-gambierol (1)

Toward an even more practical synthesis of 1, we describe herein a concise synthesis of the AB-ring fragment of 1. We envisioned that the AB-ring fragment 2 would be readily accessible from dihydropyrone 3, which in turn would be obtainable from the known tetrahydropyran 4^8 via standard chemistry (Scheme 1).



Scheme 1. Synthesis plan toward the AB-ring fragment 2

Our initial synthetic approach toward **2** involved cyclodehydration of 1,3-diketone **5** (Scheme 2).⁹ Protection of **4** with triethylsilyl chloride (TESCl) and Et₃N in the presence of a catalytic amount of 4-(dimethylamino)pyridine (DMAP) gave TES ether **6** in 95% yield. Ozonolysis of the double bond delivered aldehyde **7** in 95% yield. Methylation with MeMgBr, followed by oxidation of the resultant alcohol with 2-iodoxybenzoic acid (IBX), provided methyl ketone **8** in 75% yield (two steps). Enolization of **8** with lithium hexamethyldisilazide (LHMDS) in the presence of ZnCl₂ followed by addition of 4-benzyloxybutanal (**9**) (THF, -78 to -20 °C) afforded β -hydroxy ketone **10** as a 1.8:1 mixture of diastereomers in 96% combined yield (stereochemistry not determined). Oxidation of **10** was most efficiently performed with IBX in refluxing EtOAc¹⁰ to give 1,3-diketone **5** in 86% yield. Unexpectedly, cyclodehydration of **5** was found to be rather problematic. Treatment of **5** with (+)-10-camphorsulfonic acid (CSA) in methanol under reflux resulted in decomposition of the material, possibly due to the instability of the diketone moiety. Accordingly, **5** was reacted with CSA initially at room temperature to cleave the silyl ether and then at 60 °C to effect cyclodehydration. This modified procedure gave

dihydropyrone **11** with concomitant loss of the benzylidene acetal in 41% yield, along with a mixture of several byproducts. Careful examination of the mixture revealed that the major byproduct was bicycle **12**. Running the reaction with pyridinium *p*-toluenesulfonate (PPTS) as an acid only provided a mixture of **12** and unidentified byproducts. The structure of **12** was assigned on the basis of extensive NMR studies on **12** and its acetate **13** as shown. Upon cleavage of the benzylidene acetal, the steric repulsion existing between the 1,3-diaxial methyl groups of the tetrahydropyran would facilitate ring flipping, hemiacetal formation, and dehydration to produce **12**.



Scheme 2. Initial efforts toward the synthesis of the A-ring

The disappointing outcome of the cyclodehydration of 1,3-diketone **5** led us to investigate 6-*endo* cyclization of hydroxy ynone **14** as the revised approach toward dihydropyrone **3** (Scheme 3). We have previously shown that a variety of substituted dihydropyrones could be accessed based on silver(I) trifluoromethanesulfonate (AgOTf)-catalyzed cyclization of the corresponding hydroxy ynones.^{11,12} Addition of a lithium acetylide generated from 5-benzyloxypentyne (**15**)¹³ to aldehyde **7** gave propargylic alcohol **16** as a 1.6:1 mixture of diastereomers in 96% combined yield. Without separation, this material was oxidized with IBX to deliver ynone **17** in 98% yield. Removal of the TES group was effected by exposure to tetra-*n*-butylammonium fluoride (TBAF) buffered with acetic acid (93%). The resultant

hydroxy ynone **14** was cyclized smoothly upon treatment with AgOTf in the presence of 2,6-di-*tert*-butylpyridine (DTBP),¹⁴ giving rise to dihydropyrone **3** in 88% yield.

Having established the reliable route to **3**, our focus was shifted toward elaboration of the AB-ring fragment **2** (Scheme 3). Hydrogenation of **3** proceeded in a stereoselective manner to give ketone **18** in 89% yield as a single stereoisomer. The newly generated stereogenic center was established through an NOE experiment as shown. Stereoselective reduction of **18** to establish the C6 stereogenic center turned out to be a challenging task. Even bulky reducing agents such as diisobutylaluminum hydride (DIBALH) or L-selectride[®] provided the desired alcohol **19a** with moderate diastereoselectivity (**19a**:**19b** = 1.4–1.8:1), while reduction with NaBH₄ or LiAlH₄ resulted in preferential formation of the undesired alcohol **19b** (**19a**:**19b** = ca. 1:6). However, we were fortunate to find that these diastereomeric alcohols were readily separable by flash chromatography on silica gel, and the undesired **19b** could be recycled by oxidation with IBX (80%).¹⁵ Finally, benzylation of **19a** afforded bis(benzyl) ether **2** in 85% yield, successfully intercepting our previous synthesis of **1**.



Scheme 3. Synthesis of the AB-ring fragment 2 via 6-endo cyclization of hydroxy ynone 14

In conclusion, we have devised a concise synthesis of the AB-ring fragment 2 of (-)-gambierol, by exploiting AgOTf-catalyzed 6-*endo* cyclization of hydroxy ynone 14. The present synthesis allows an efficient access to 2 from the B-ring tetrahydropyran 4 in nine steps, which compares favorably with our previous synthesis (17 steps). Further studies on the chemistry and biology of (-)-gambierol and its synthetic analogues are currently underway and will be reported shortly.

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REFERENCES AND NOTES

- M. Satake, M. Murata, and T. Yasumoto, J. Am. Chem. Soc., 1993, 115, 361; A. Morohashi, M. Satake, and T. Yasumoto, *Tetrahedron Lett.*, 1999, 40, 97.
- H. Fuwa, M. Sasaki, M. Satake, and K. Tachibana, *Org. Lett.*, 2002, 4, 2981; H. Fuwa, N. Kainuma, K. Tachibana, and M. Sasaki, *J. Am. Chem. Soc.*, 2002, 124, 14983.
- I. Kadota, H. Takamura, K. Sato, A. Ohno, K. Matsuda, and Y. Yamamoto, *J. Am. Chem. Soc.*, 2003, 125, 46; I. Kadota, H. Takamura, K. Sato, A. Ohno, K. Matsuda, M. Satake, and Y. Yamamoto, *J. Am. Chem. Soc.*, 2003, 125, 11893; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *J. Am. Chem. Soc.*, 2005, 127, 848; U. Majumder, J. M. Cox, H. W. B. Johnson, and J. D. Rainier, *Chem. Eur. J.*, 2006, 12, 1736; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *Chem. Eur. J.*, 2006, 12, 1736; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *Chem. Eur. J.*, 2006, 12, 1736; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *Chem. Eur. J.*, 2006, 12, 1736; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *Chem. Eur. J.*, 2006, 12, 1747; H. Furuta, Y. Hasegawa, and Y. Mori, *Org. Lett.*, 2009, 11, 4382; H. Furuta, Y. Hasegawa, M. Hase, and Y. Mori, *Chem. Eur. J.*, 2010, 16, 7586.
- V. Ghiaroni, M. Sasaki, H. Fuwa, G. P. Rossini, G. Scalera, T. Yasumoto, P. Pietra, and A. Bigiani, *Toxicol. Sci.*, 2005, 85, 657; V. Ghiaroni, H. Fuwa, M. Inoue, M. Sasaki, K. Miyazaki, M. Hirama, T. Yasumoto, G. P. Rossini, G. Scalera, and A. Bigiani, *Chem. Senses*, 2006, 31, 673.
- 5. M. C. Louzao, E. Cagide, M. R. Vieytes, M. Sasaki, H. Fuwa, T. Yasumoto, and L. M. Botana, *Cell. Physiol. Biochem.*, 2006, **17**, 257.
- E. Cuypers, Y. Abdel-Mottaleb, I. Kopljar, J. D. Rainier, A. L. Raes, D. J. Snyders, and J. J. Tytgat, *Toxicon*, 2008, **51**, 974; I. Kopljar, A. J. Labro, E. Cuypers, H. W. B. Johnson, J. D. Rainier, J. Tytgat, and D. J. Snyders, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 9896.
- E. Alonso, H. Fuwa, C. Vale, Y. Suga, T. Goto, Y. Konno, M. Sasaki, F. M. LaFerla, M. R. Vieytes, L. Giménez-Llort, and L. M. Botana, J. Am. Chem. Soc., 2012, 134, 7467.
- 8. K. C. Nicolaou, D. A. Nugiel, E. Couladouros, and C.-K. Hwang, *Tetrahedron*, 1990, 46, 4517.
- 9. K. Tsubone, K. Hashizume, H. Fuwa, and M. Sasaki, *Tetrahedron*, 2011, 67, 6600.
- 10. J. D. More and N. S. Finney, Org. Lett., 2002, 4, 3001.
- 11. H. Fuwa, S. Matsukida, and M. Sasaki, *Synlett*, 2010, 1239; H. Fuwa, K. Mizunuma, S. Matsukida, and M. Sasaki, *Tetrahedron*, 2011, **67**, 4995.
- For other reports on the AgOTf-mediated 6-endo cyclization of hydroxy ynones, see: S.-L. Shi, M. Kanai, and M. Shibasaki, Angew. Chem. Int. Ed., 2012, 51, 3932; C. R. Reddy and B. Srikanth,

Synlett, 2010, 1536; K. C. Nicolaou, M. O. Frederick, A. C. B. Burtoloso, R. M. Denton, F. Rivas, K.
P. Cole, R. J. Aversa, R. Gibe, T. Umezawa, and T. Suzuki, *J. Am. Chem. Soc.*, 2008, 130, 7466; C.
Wang and C. J. Forsyth, *Org. Lett.*, 2006, 8, 2997.

- Alkyne 15 was prepared from aldehyde 9 in two steps ((i) CBr₄, Ph₃P, CH₂Cl₂, 0 °C; (ii) *n*-BuLi, THF, -78 °C, 96% yield for the two steps).
- 14. Addition of DTBP was found to be essential for obtaining 3 in excellent yield.
- 15. All attempts at inverting the C6 stereogenic center by means of Mitsunobu reaction were unfruitful $(p-NO_2C_6H_4CO_2H, Ph_3P, diethyl azodicarboxylate, toluene or THF, reflux or$ *p* $-MeOC_6H_4CO_2H,$ *n* $-Bu_3P,$ *N*,*N*,*N'*,*N'*-tetramethyl azodicarboxamide,¹⁶ benzene, reflux).
- 16. T. Tsunoda, J. Otsuka, Y. Yamamiya, and S. Itô, Chem. Lett., 1994, 539.