



A synthetic approach for (S)-(3-benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone, a selective CB2 receptor agonist

Zhushou Luo, Mohamed Naguib*

Anesthesiology Institute, Cleveland Clinic Foundation, 9500 Euclid Avenue, Cleveland, OH 44106, USA

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ABSTRACT

(S)-(3-Benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone, a selective CB2 receptor agonist, was obtained from 3-hydroxy-4-iodo benzoic acid in nine steps with 97.4% ee and 3.4% total yield, which involved palladium catalyzed tandem intramolecular Heck/Suzuki cross coupling reaction, chemical resolution with (+)-norephedrine and Wolf–Kishner reaction as the key steps. (S)-(3-Benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone will be evaluated in vivo studies and this approach will be applied in the optimization process of CB2 receptor agonist.

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The endogenous cannabinoid system is a complex system consisting of two cannabinoid (CB) receptors (CB1—expressed primarily in the brain and CB2—expressed primarily in the peripheral immune system¹ and in the central nervous system (CNS)²), seven endogenous endocannabinoid ligands (arachidonic acid derivatives) including anandamide (*N*-arachidonylethanolamine) and 2-arachidonoylglycerol (2-AG),³ and several proteins responsible for the regulation of endocannabinoid metabolic pathways, such as monoacylglycerol lipase and fatty acid amide hydrolase.⁴ CB1 and CB2 are seven transmembrane, G protein-coupled receptors, and they share 44% overall identity. Both receptors mediate the inhibition of adenylyl cyclases and stimulation of mitogen-activated protein kinases (MAPK). However, CB1 receptor agonists failed to show efficacy in patients with neuropathic pain compared with placebo, and their use has resulted in a high incidence of psychotropic adverse effects.⁵ In contrast, CB2 receptor agonists are neuroprotective and are emerging as treatments for neuropathic pain. Moreover, they lack the psychotropic adverse effects that normally occur with CB1 agonists.^{6,7} Increasing interest has been spurred in pharmaceutical companies and academia, and several classes of selective CB2 agonists for the treatment of inflammatory and neuropathic pain conditions have been reported.^{6,8}

We initiated a research program aimed at identifying novel selective CB2 agonists for the treatment of neuropathic pain, and we identified (S)-(3-Benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone **1** (Fig. 1) as a lead compound.^{6b} The absolute stereochemistry of compound **1** has also been

determined by X-ray crystal structure. In vitro *h*CB2 GTPγ[³⁵S] functional assay, compound **1** binds selectively to the CB2 receptor (EC₅₀: 108 nM on *h*CB2; >10 μM on *h*CB1), which is better than its corresponding racemate (EC₅₀: 128 nM on *h*CB2; >10 μM on *h*CB1) and *R* enantiomer (EC₅₀: 960 nM on *h*CB2, >10 μM on *h*CB1). Racemate also showed (i) dose-dependent reversal of neuropathic pain in spinal nerve ligation and paclitaxel-induced neuropathy rat models when administered intraperitoneally with ED₅₀ value of 7.5 and 24 mg/kg, respectively, and (ii) prevention of paclitaxel-induced neuropathy.^{6a} However, compound **1** was only obtained previously in very limited quantity using chiral HPLC separation of the corresponding racemate.^{6b} Thus, a scalable and effective approach is needed for the synthesis and optimization of compound **1**.

Compound **1** has a quaternary carbon stereocenter at the benzylic position in the ring system of 2,3-dihydro-1-benzofuran, and presents an interesting synthetic challenge.⁹ The key reaction for racemic synthesis of compound **1** and its analogs is shown in Scheme 1. The synthesis utilized palladium catalyzed tandem

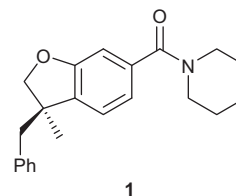
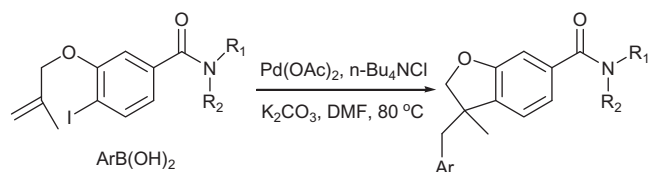


Figure 1. Structure of (S)-(3-benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone.

* Corresponding author. Tel.: +1 216 444 6328; fax: +1 216 636 2043.

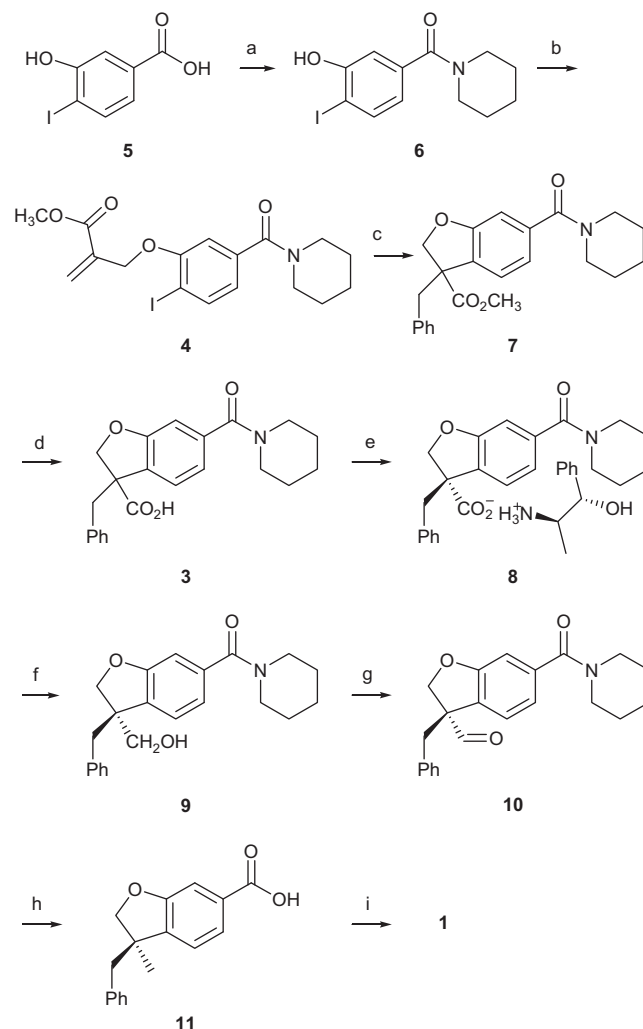
E-mail address: naguibm@ccf.org (M. Naguib).



Scheme 1. Synthesis of racemic **1** and its analogs.

intramolecular Heck/Suzuki coupling reactions.^{6b,10} Although asymmetric intramolecular Heck reaction has been widely used in organic synthesis including the formation of quaternary chiral carbon center,¹¹ our attempt of enantioselective synthesis was unsuccessful when we used different reaction conditions, chiral ligands, palladium sources, and appropriate additives. We then utilized chemical resolution approach using chiral resolving agent.¹² As shown in **Scheme 2**, we expect to obtain chiral separable solid salt **2** by the combination of appropriate chiral amine and racemic acid **3**, followed by further transformation to deliver compound **1**. Racemic acid **3** could be synthesized from tandem intramolecular Heck/Suzuki coupling reactions of compound **4** and phenylboronic acid.^{6b,10} Herein, we report our synthetic pathway of compound **1** based on this strategy.

Coupling of 3-hydroxy-4-iodo benzoic acid **5**¹³ with piperidine gave amide **6**, which reacted with methyl 2-bromoacrylate to afford **4** (**Scheme 3**). Compound **4** was subjected to Heck/Suzuki coupling conditions leading to compound **7** in 53% yield. Basic hydrolysis of **7** gave the racemic acid **3**. Among the screened chiral amine for chemical resolution, (+)-norephedrine gave desired results. Acid **3** and (+)-norephedrine were dissolved in ethanol under 75 °C, followed by slow addition of ethyl acetate to form a little cloudy solution. Then the solution was filtered, the filtrate cooled to room temperature, and stood at 4 °C overnight to give white precipitate, which was filtered to give salt **8** in good yields (24–33%) and high ee value (98.0%, see **supplementary data**). The absolute stereochemistry was not determined at this stage. Salt **8** was then directly transformed into mixed anhydride followed by NaBH₄ reduction to afford alcohol **9** in 70% yield. Initially, **9** was converted into its corresponding bromide followed by the treatment with tributyltin hydride and AIBN in refluxing benzene to afford only complex mixture. Finally Wolf–Kishner reaction¹⁴ was chosen to make this transformation. Swern oxidation of primary alcohol **9** gave aldehyde **10**, which reacted with hydrazine in diethylene glycol at 100 °C for 1 h followed by the addition of potassium hydroxide and heating at 195 °C for additional 4 h to



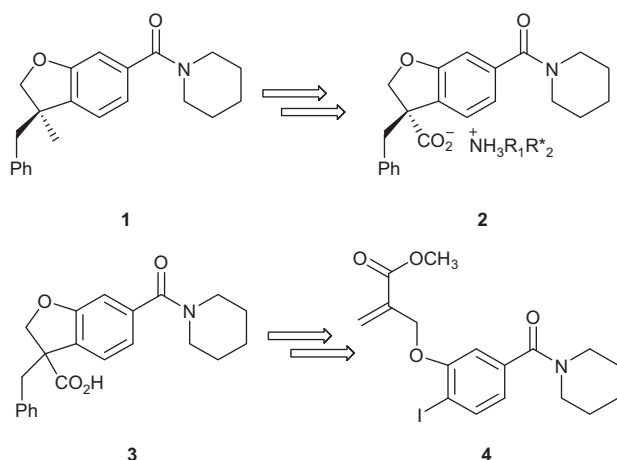
Scheme 3. Reagents and conditions: (a) piperidine, *i*Pr₂NEt, HATU, DMF, 23 °C, 16 h, 56%; (b) methyl 2-bromomethyl-acrylate, K₂CO₃, methyl ethyl ketone, reflux, 3 h, 86%; (c) phenylboronic acid, Pd(OAc)₂, *n*-Bu₄NCl, K₂CO₃, DMF, 80 °C, 16 h, 53%; (d) NaOH, dioxane/H₂O = 4/1, 23 °C, 48 h, 91%; (e) (+)-norephedrine, 24–33%; (f) methyl chloroformate, THF, 0–23 °C; then NaBH₄, ethanol, –78–0 °C, 70%; (g) (COCl)₂, DMSO, then Et₃N, CH₂Cl₂, –78–23 °C, 93%; (h) NH₂NH₂, diethylene glycol, 100 °C, 1 h, then KOH, 195 °C, 4 h, 73%; (i) piperidine, *i*Pr₂NEt, HATU, DMF, 23 °C, 16 h, 92%.

afford acid **11**. Acid **11** coupled with piperidine to give the final product **1**. Compound **1** was confirmed as the desired *S* configuration product with 97.4% ee by the comparison of all spectroscopic data with authentic sample, which was previously purified by chiral HPLC and assigned as *S* configuration by X-ray crystallography.^{6b}

In conclusion, we have developed an effective approach to synthesize (*S*)-(3-benzyl-3-methyl-2,3-dihydro-benzofuran-6-yl)-piperidin-1-yl-methanone **1**, a selective CB₂ receptor agonist with 3.4% total yield and high ee (97.4%). This approach involved tandem intramolecular Heck/Suzuki coupling reaction of intermediate **4**, chemical resolution of intermediate acid **3** and Wolf–Kishner reaction for the transformation of intermediate **9** into **1**. Currently compound **1** is evaluated in vivo studies and this approach is used in our lab for the optimization of compound **1**.

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Scheme 2. Synthetic strategy for compound **1**.

carried out at the supporting facilities at the chemistry department of Case Western University. HPLC analysis was done at Mass Spectrometry II core lab at the Lerner Research Institute of Cleveland Clinic Foundation. We thank Professor Gary Sulikowski for his suggestions.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2012.04.076>.

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