

[3 + 2] Cycloaddition reactions of the phospho-alkyne, Bu'CP, with a transition metal trisulphide containing terminal thio ligands: syntheses, crystal and molecular structures of $[W(\eta^5-(C_5Me_5)(S)(S_2PC'Bu)]_2$ and $[W(\eta^5-(C_5Me_5)(S)(S_2C_2Ph_2)]_2$

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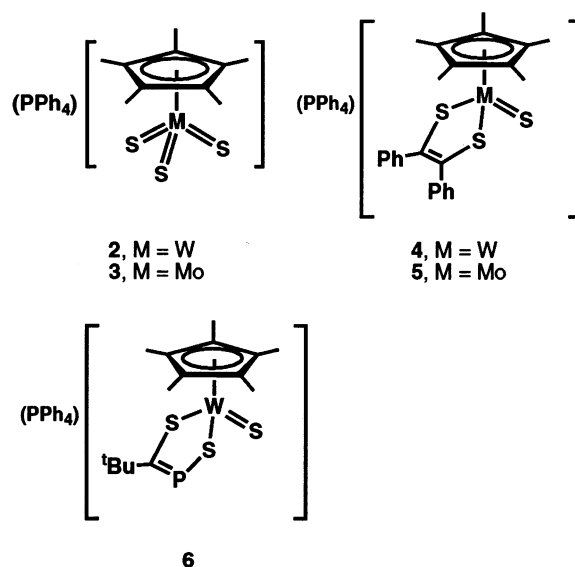
Received 5 March 1998; received in revised form 11 May 1998

Abstract

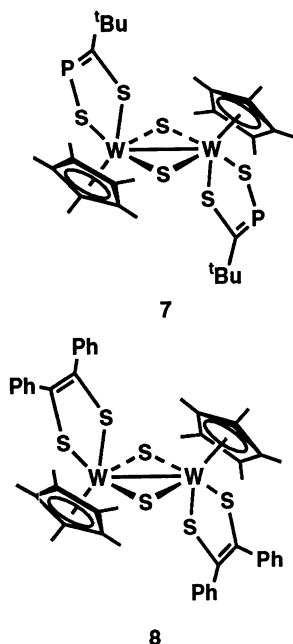
The trisulphido complex $[PPh_4][W(\eta^5-(C_5Me_5)(S)_3]$, reacts with the phospho-alkyne Bu'CP and O₂ to afford the dimeric complex $[W(\eta^5-(C_5Me_5)(S)(S_2PC'Bu)]_2$. Analogous reactions occur with the alkyne Ph₂C₂ to give $[W(\eta^5-(C_5Me_5)(S)(S_2C_2Ph_2)]_2$. The molecular structures of both complexes are presented and discussed. © 1998 Elsevier Science S.A. All rights reserved.

Keywords: Cycloaddition; Thio ligands; Trisulfide

There is considerable current interest in the study of metal sulphides, in view of their biological importance and their potential in the design of new materials [1,2]. The reactivity of certain transition metal sulphides towards alkenes [3] and alkynes [4,5] has been discussed, but a very recent report by Goodman and Rauchfuss [6], which describes the addition of nitriles to the tetrasulphido anion $[ReS_4]^-$, prompts us to report the first example of the ready addition of the phospho-alkyne, Bu'CP, **1**, to the anionic mononuclear trisulphido complex $[PPh_4][W(\eta^5-(C_5Me_5)(S)_3]$ **2**, containing three terminal W=S bonds.



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Previously [7,8], we showed that the addition of the alkyne Ph_2C_2 to complex **2** and its molybdenum analogue **3**, afforded mononuclear complexes **4** ($\text{M} = \text{W}$) and **5** ($\text{M} = \text{Mo}$), respectively, which were both fully characterised.

In view of the well-known similarity in behaviour of alkynes and phosphalkynes towards transition metals [9–12] and the previously reported synthesis of several five-membered heterocyclic compounds by [3 + 2] cycloaddition reactions involving phosphalkynes, it was of interest to study the reaction of the anionic trisulphide complex **2** with the phosphalkyne $\text{Bu}'\text{CP}$. Monitoring the r.t. reaction between **2** and one equivalent of $\text{Bu}'\text{CP}$ in acetonitrile by ^{31}P -NMR spectroscopy, showed the rapid development of a new signal ($\delta_{\text{P}} = 210.3$ ppm) assigned to complex **6**, which has not yet been possible to isolate. The same complex is obtained in the presence of five equivalents of $\text{Bu}'\text{CP}$.

However, when dry O_2 was passed through a solution of **6**, the green–brown complex $[\text{W}(\eta^5\text{-C}_5\text{Me}_5)(\text{S})(\text{S}_2\text{PC}'\text{Bu})]_2$ **7** was readily formed and isolated, ($\delta_{\text{P}} = 101.4$ ppm, $^2J(\text{WP}) = 26.8$ Hz) (52%) [13]. A single crystal X-ray diffraction study of the CH_2Cl_2 solvate, revealed the dimeric structure shown in Fig. 1 [14].

In a similar way, the green solution of **4** reacted with dioxygen, instantly turning purple. The purple intermediate has not yet been characterised but it seems to be paramagnetic. After 1 day at r.t., the colour of the solution gradually changed to green and deposited crystals (41%) of the dark-green dimeric complex **8** [15], whose structure, also determined by a single crystal X-ray diffraction study, is shown in Fig. 2 [16].

The close similarity in the structures of **7** and **8** is self-evident from an inspection of the bond length and bond angle data summarised in Figs. 1 and 2. The $\text{W}\cdots\text{W}$, $\text{W}-\text{S}$ and $\text{W}=\text{S}$ distances in the two complexes are identical within experimental error. The WS_2PC and WS_2C_2 ring systems are planar in **7** and **8**, respectively. This could be interpreted as involving some π -delocalisation as expected for 1,2-phospha-ene thiolato and 1,2-enethiolato coordination; however the $\text{C}(1)-\text{P}$ distance, $1.66(2)$ Å in **7**, and the $\text{C}(11)-\text{C}(18)$ distance, $1.358(5)$ Å in **8**, are consistent with localised $\text{P}=\text{C}$ and $\text{C}=\text{C}$ double bonds, respectively.

The ready [3 + 2] cycloaddition reaction of these anionic trithio complexes with phosphalkynes, suggests that other metal sulphides containing terminal $\text{M}=\text{S}$ bonds may also form this type of compound. Complex **6** also readily reacts with metal carbonyl compounds, in the presence of an excess of $\text{Bu}'\text{CP}$, to give the cage compound $\text{P}_2\text{C}_2\text{Bu}'_2\text{CO}$, (previously obtained by Cowley et al. [17] from the reaction of $\text{Bu}'\text{CP}$ and $[\text{Ti}(\eta^5\text{-C}_5\text{H}_5)(\text{CO})_2]$), and a mixture of the 1,2,4- and 1,3,4-thiadiphospholes $\text{P}_2\text{SC}_2\text{Bu}'_2$. The 1,2,4-thiadiphosphole was first reported by Lindner et al. [18] in an unusual reaction involving $\text{Bu}'\text{CP}$ and η^2 -thiophosphinato cobalt and manganese complexes, but the 1,3,4-isomer has not been previously reported, and it will be the subject of a separate publication [19].

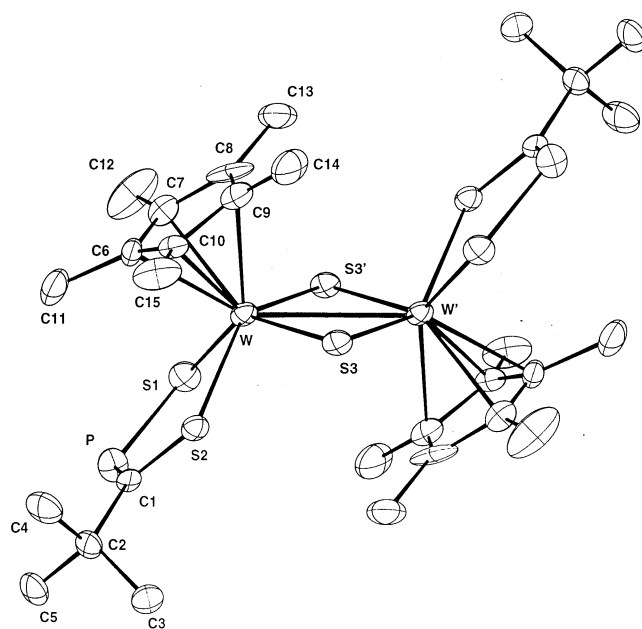


Fig. 1. Molecular structure of **7** together with selected bond length (Å) and bond angle ($^\circ$) data: $\text{W}\cdots\text{W}$ 3.068(2), $\text{W}-\text{S}(1)$ 2.404(6), $\text{W}-\text{S}(2)$ 2.403(4), $\text{W}-\text{S}(3)$ 2.337(5), $\text{W}-\text{S}(3')$ 2.324(5), $\text{S}(1)-\text{P}$ 2.039(7), $\text{S}(2)-\text{C}(1)$ 1.74(2), $\text{C}(1)-\text{P}$ 1.66(2); $\text{W}-\text{S}(3)-\text{W}$ 82.3(2), $\text{S}(2)-\text{W}-\text{S}(1)$ 79.9(2), $\text{W}-\text{S}(2)-\text{C}(1)$ 116.1(6), $\text{W}-\text{S}(1)-\text{P}$ 116.3(3), $\text{S}(2)-\text{C}(1)-\text{P}$ 122.4(9), $\text{S}(1)-\text{P}-\text{C}(1)$ 102.8(7).

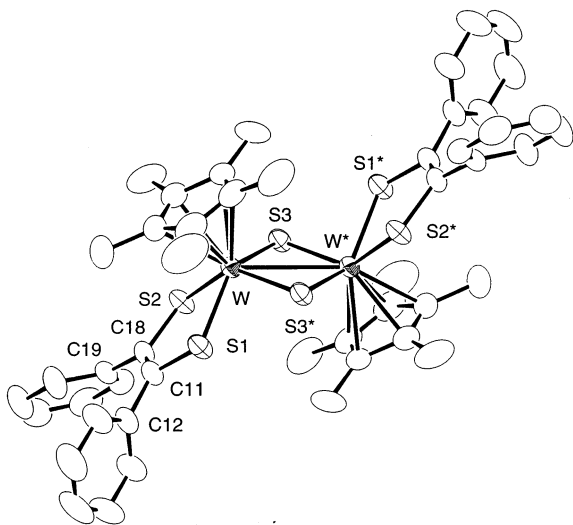


Fig. 2. Molecular structure of **8** together with selected bond length (Å) and bond angle ($^{\circ}$) data: W···W 3.0652(3), W–S(1) 2.402(1), W–S(2) 2.401(1), W–S(3) 2.348(1), W–S(3)* 2.344(1), S(1)–C(11) 1.736(4), S(2)–C(18) 1.740(4), C(11)–C(18) 1.358(5); W–S(3)–W* 81.58(3), S(2)–W–S(1) 78.07(3), W–S(2)–C(18) 111.3(1), W–S(1)–C(11) 111.1(1), S(2)–C(18)–C(11) 118.3(3), S(1)–C(11)–C(18) 118.9(3).

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- [13] **7**: $^1\text{H-NMR}$ (360 MHz, CD_3CN); δ 1.82 (s, C_5Me_5 , 30H), 0.61 (s, Bu^t , 18H).
- [14] Crystal data for **7**: $\text{C}_{31}\text{H}_{50}\text{Cl}_2\text{P}_2\text{S}_6\text{W}_2$, $M = 1115.6$, monoclinic, space group $C2/c$ (No. 15), $a = 28.794(6)$, $b = 9.435(6)$, $c = 15.074(4)$ Å, $\beta = 107.17(2)^{\circ}$, $U = 3913(3)$ Å 3 , $Z = 4$, $D_{\text{calc}} = 1.89$ g cm $^{-3}$, $F(000) = 2716$. Monochromated Mo– K_{α} radiation $\lambda = 0.71073$ Å, $T = 173(2)$ K. Data were collected on an Enraf-Nonius CAD 4 diffractometer using a crystal of $0.20 \times 0.20 \times 0.05$ mm. A total of 3423 independent reflections were measured for $2 < \theta < 25^{\circ}$ of which 2165 had $I > 2\sigma I$. The structure was solved by direct methods using SHELXS-86 and refined on F^2 with all non-H atoms anisotropic using SHELXL-93. Methyl-H atoms were included in riding mode, fixed at idealised geometry, but with the torsional angle defining the H atom positions refined and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. The final residues were $R_1 = 0.075$ (for $I > 2\sigma I$) and $wR_2 = 0.203$ for all data.
- [15] **8**: $^1\text{H-NMR}$ (500 MHz, CDCl_3); δ 2.02 (s, C_5Me_5 , 30H), 7.2 (m, Ph, 20H). UV-visible (λ_{max} , nm THF): 409, 542, 638. Found C, 49.01, H, 4.41, S, 15.89; $\text{C}_{48}\text{H}_{50}\text{W}_2\text{S}_6$ requires C, 48.57, H, 4.25, S, 16.21%.
- [16] Crystal data for **8**: $\text{C}_{48}\text{H}_{50}\text{S}_6\text{W}_2$, $M = 1186.98$, triclinic, space group $P\bar{1}$ (No. 2), $a = 10.5450(9)$, $b = 11.1888(8)$, $c = 9.898(1)$ Å, $\alpha = 96.359(8)^{\circ}$, $\beta = 91.321(8)^{\circ}$, $\gamma = 77.343(6)^{\circ}$, $U = 1132.4(2)$ Å 3 , $Z = 1$, $D_{\text{calc}} = 1.740$ g cm $^{-3}$, $F(000) = 582.00$. Monochromated Mo– K_{α} radiation $\lambda = 0.71069$ Å, $T = 296$ K. Data were collected on a Rigaku AFC7R diffractometer using a crystal of $0.20 \times 0.25 \times 0.15$ mm. A total of 5203 independent reflections were measured for $2 < \theta < 25^{\circ}$ of which 4430 had $I > 3\sigma I$. The structure was solved by Patterson methods and refined by full-matrix least-squares with all non-H atoms anisotropic using the TEXSAN package. The final residues were $R = 0.023$ and $R_w = 0.027$ (for $I > 3\sigma I$).
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