Hydrolysis of Heptafluoropropylphosphonous Diiodide and Bisheptafluoropropylphosphinous Iodide. Formation of Bisheptafluoropropylphospine

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 \mathbf{H} eptafluoropropylphosphonous diiodide and bisheptafluoropropylphosphinous iodide formed by the interaction of heptafluoropropyl iodide with phosphorus were characterized as the N,N-dimethyl amides. Treatment of the mixture of iodides with water gave the expected heptafluoropropylphosphinic acid, and heptafluoropropane, but unexpectedly gave bisheptafluoropropylphosphine. The quantity of phosphine formed increased on refluxing with aqueous alkali, and was greatest when the acid iodides were treated with solid alkali. These results contrasted with those of Emeleus and Smith (1) who observed the formation of heptafluoropropylphosphinic acid and heptafluoropropane, but did not observe the formation of

bisheptafluoropropylphosphine under comparable experimental conditions.

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LITERATURE CITED

(1) Emeleus, H.J., Smith, J.D., J. Chem. Soc. 375, (1959).

Reactants HFPI + P		Table I. Summary of F			Analysis ^a		
	Conditions 190° C. 48 hours	$\begin{array}{c} \text{Products} \\ \text{C}_3\text{F}_7\text{PI}_2, 67\% \\ (\text{C}_3\text{F}_7)_2\text{PI}, 33\% \end{array}$	Boiling range ° C.	Yield, %		Calcd.	Found
			27-29 (for mixture)	47.5 25.7	%C %P %I	12.4 6.4 35.2	12.3 6.4 36.3
HFPP ₂ + bis HFPPI + (CH ₃) ₂ NH	Acid iodides added to cooled Pet. ether	C_3 FP[N(CH ₃) ₂] ₂	27-29	• • •	%C %H %P	29.2 4.2 10.8	28.7 4.6 10.5
	soln. of amine	$(C_3F_7)_2PN(CH_3)_2$	23-25		%N %C %H %P	9.7 23.1 1.5 7.5	9.4 24.6 2.2 7.9
HFPPI ₂ + bisHFPPI + H ₂ O	3-hour reflux	C_3F_7H	< room temperature	27	%N %C %H	$3.4 \\ 21.2 \\ 0.6$	4.9 22.2 0.8
		$C_3F_7P(O)H(OH)^b$	90 (l mm.)	7	%C % H %P	15.4 0.9 13.2	15.3 1.0 13.3
HFPPI ₂ + ois HFPPI + NaOH	Cooled	(C ₃ F ₇) ₂ PH	30-32		%C %H	19.5 0.3	19.7 0.4
$C_3F_7P(O)H(OH)$ $H_2O_2 + NaOH$	Neutralized to phenolphthalene	$C_3F_7P(O)(ONa)_2$	solid	31	%P %Na %C %P	8.4 15.7 12.2 10.5	8.0 15.9 12.3 10.7

A small amount of $(C_3F_7)_2PH$ was also isolated.