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# Assembly of polycyclic N-heterocycles *via* coppercatalyzed cycloamination of indolylquinones and aromatic amines<sup>†</sup>

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The copper-catalyzed cycloamination of indolylquinones and various (hetero)aromatic amines under ligand-free conditions for the synthesis of polycyclic N-heterocycles has been developed. This method allows facile access to polycyclic N-heterocycles with the tolerance of chloride, bromide, amino, thio, *etc.* groups in moderate to high yields (60–89%).

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# Introduction

The nitrogen-containing heterocycle nucleus is an important structural scaffold frequently found in drugs, functional materials and natural products.<sup>1</sup> Among nitrogen-containing heterocycles, polycyclic N-heterocycles are widespread in nature and exhibit several interesting biological activities.<sup>2</sup> The representative compounds include the quinone-fused polycyclic N-heterocycles pixantrone (experimental antineoplastic drug with fewer toxic effects on cardiac tissue)<sup>3</sup> and ascidide-min (possessing prominent cytotoxic properties)<sup>4</sup> and the indole-fused carbazole rebeccamycin (topoisomerase I inhibitor),<sup>5</sup> and the naphtho[*a*]carbazole (a potential candidate for cancer treatment)<sup>6</sup> (Fig. 1). On the other hand, polycyclic N-heterocycles exhibit interesting properties for application as fluorescent probes<sup>7</sup> and optoelectronic materials.<sup>8</sup>

Pentacyclic complexes bearing a quinone skeleton have been used for metal ion recognition and have also been proven to be suitable receptors for the colorimetric sensing of certain anions.<sup>9</sup> Owing to their various applications, numerous methods for the synthesis of annellated polycyclic complexes bearing a quinone skeleton have been developed. The representative approaches usually include the classical Friedel– Crafts reactions,<sup>10</sup> Lewis acid catalyzed intramolecular cyclization with a Lewis acid as a catalyst,<sup>11</sup> introduction of the 9,10carbonyl functions by the oxidation of the corresponding hydrocarbon,<sup>12</sup> transition-metal-catalyzed oxidative cyclization,<sup>13</sup> and thermal or Lewis acid catalyzed Diels–Alder reaction followed by aromatization.<sup>14</sup> Despite their merits, many existing synthetic approaches have several shortcomings. However, some of them rely heavily on the use of expensive transition metal catalysts, and cannot avoid harsh conditions, multistep synthesis, poor yield of the proposed product and the preparation of the prerequisite functional groups.

Transition-metal-catalyzed C–H amination is a step-economical and straightforward synthetic methodology to form aromatic C–N bonds.<sup>15</sup> In particular, transition-metal catalyzed cross-dehydrogenative coupling (CDC) amination is highly desirable owing to aromatic amines being utilized as reactants.<sup>16</sup> Recently, we discovered that polycyclic N-heterocyclic complexes are synthesized by cobalt-catalyzed cycloamination reaction between indolylquinones and aromatic amines and



Fig. 1 A representative polycyclic N-heterocyclic complex.

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Scheme 1 Copper-catalyzed cycloamination for the synthesis of polycyclic N-heterocycles.

*t*-BuOK mediated oxidative coupling amination of 1,4-naphthoquinone and related 3-indolylnaphthoquinones with amines.<sup>17</sup> Further studies revealed that commercially available CuCl is very efficient for Cu-catalyzed cycloamination reaction between indolylquinones and aromatic amines (Scheme 1). It is noteworthy that unactivated indolylquinones and simple aromatic amines were employed as the starting materials with good functional group tolerance. Herein, we wish to disclose our results.

### **Results and discussion**

We selected the reaction of readily available indolylnaphthoquinone (1a) and aniline (2a) as the model reaction for optimizing the CDC aromatic amination conditions (see Table 1, as well as Tables S1–S5 in the ESI†). To our delight, under the reaction of CuCl (10 mol%), *t*-BuOK (2.0 equiv.), and DMF



<sup>*a*</sup> Reaction conditions: **1a** (0.3 mmol), **2a** (0.6 mmol, 2.0 equiv.), [cat.] (10 mol%), base (0.6 mmol, 2.0 equiv.), solvent (2.0 mL), 16 h, air, 120 °C. <sup>*b*</sup> Isolated yield. DMF = N,N-dimethylformamide; DMAC = N,N-dimethylacetamide; NMP = N-methylpyrrolidone; DMSO = dimethyl sulfoxide; DMAP = 4-dimethylaminopyridine; N.R. = no reaction.

(2 mL), at 120 °C, for 16 h (Table 1, entry 1), the desired product 3aa was obtained in 87% isolated yield. Other metal salts including CuBr, CuI, CuBr<sub>2</sub>, Cu(OAc)<sub>2</sub>, CuSO<sub>4</sub>·5H<sub>2</sub>O, CuCl<sub>2</sub>·2H<sub>2</sub>O and Cu(OTf)<sub>2</sub> did not provide better results for the cycloamination (entries 2-6, see the ESI<sup>†</sup>). Subsequently, the variation of t-BuOK to t-BuONa, KOH, CH<sub>3</sub>ONa, K<sub>2</sub>CO<sub>3</sub>, Et<sub>3</sub>N, or DMAP did not show any improvement (entries 7-9, see the ESI<sup>†</sup>). The reaction performed under air atmosphere in the absence of a catalyst or a base afforded very low yield of the products or no product (entries 14 or 15), which showed that the catalyst or base played a pivotal role in obtaining the desired product. The use of DMF as a solvent was crucial, as the reaction gave poor results in other solvents such as DMAC, NMP, DMSO, CH<sub>3</sub>NO<sub>2</sub>, DCE or CH<sub>3</sub>CN (Table 1, entries 10-13, see the ESI<sup>†</sup>). It should be noted that increasing or decreasing the reaction temperature gave slightly inferior results, indicating that the transformation was sensitive to temperature (see the ESI<sup>†</sup>).

With the optimized reaction conditions in hand, subsequently, the scope of the anilines was investigated (Table 2). A variety of substituted anilines were compatible with the Cucatalyzed cycloamination, regardless of the electronic properties of the substrates (3ab-3ai). It is noteworthy that the valuable groups (NH<sub>2</sub>, SH, F, Cl, and Br) could be readily tolerated, which provides an opportunity for further elaboration. Although ortho-monosubstituted aromatic amines are sterically hindered even for cyclization, the ortho-substituted anilines underwent reactions successfully to give the corresponding aminated products (3aj-3am). For disubstituted anilines, the reaction was also found to proceed smoothly (3an-3ap). In particular, naphthylamine and aminopyrene were amenable under our reaction conditions and provided the expected cyclization products 3aq and 3ar. Even heteroanilines are well tolerated in this reaction. The use of aminopyridine provides moderate yields of the desired product (3as). The strongly coordinating groups (pyridine), which were employed as reagents for direct C-H functionalization, were fully tolerwith high chemoselectivity and regioselectivity. ated Unfortunately, no desired product can be obtained when aliphatic amines are used.

Next, a series of substituted indolylnaphthoquinones was tested for the cycloamination sequence (Table 3). As summarized in Table 3, the reaction was compatible with a variety of indole moieties (1a-1i) bearing electron-donating and electron-withdrawing substituents to produce the desired polycyclic N-heterocyclic products (3aa-3ia) in moderate to good yields (62-87%). Notably, the gram-scale synthesis afforded 1.50 g of 3ba in 83% yield. To our delight, the reaction also showed good compatibility with a wide range of valuable functional groups such as fluoro (3ha), chloro (3fa), and bromo (3ga) groups. Tolerance to the halogen atoms was noteworthy since they have been frequently used for further modifications. N-Methylindoles and N-benzylindole naphthoquinone can smoothly react to give the corresponding products in good yields (3aa and 3ba). Unfortunately, no desired product can be obtained when 1c is used as the substrate. Moreover, we were

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#### Table 2Scope for anilines<sup>a</sup>



<sup>a</sup> Reaction conditions: **1a** (0.3 mmol), **2** (0.6 mmol), CuCl (10 mol%), t-BuOK (2.0 equiv.), DMF (2.0 mL), 16 h, air, 120 °C. Isolated yield.





<sup>*a*</sup> Reaction conditions: **1** (0.3 mmol), **2a** (0.6 mmol), CuCl (10 mol%), *t*-BuOK (2.0 equiv.), DMF (2.0 mL), 16 h, air, 120 °C. Isolated yield. N. R. = no reaction. <sup>*b*</sup> In a 5 mmol scale.

pleased to find that the position of the substituent on the indole moiety showed no obvious influence on the reaction outcome, and substituents at C4-(3da), C5-(3ea-3ga), C6-(3ha) or C7-(3ia) were all well tolerated in the reaction. 1,4-Anthraquinone-substituted indole 1j was also employed, affording the corresponding product 3ja in 83% yield.

Control experiments were conducted to clarify the cycloamination reaction pathway (Scheme 2). The studies revealed that the radical scavenger TEMPO (2,2,6,6-tetramethyl-1-piperidinyloxy) did not inhibit the reaction under standard conditions, ruling out the radical mechanism (Scheme 2a). At 40 °C for 2 h, the reaction of model compounds **1a** and **2a** generated the coupling product **4a** in a yield of 63% (Scheme 2b). The reaction of model compounds **1a** and **2a** generated the coupling



Scheme 2 Control experiments.

product **4a** without CuCl in a yield of 68% (Scheme 2c). The studies revealed the Michael addition process of the aniline to the quinone in the presence of a base at first. Under the standard conditions, the reaction of compound **4a** generated the product **3aa** in a yield of 81% (Scheme 2d). We surmised that the coupling product **4a** should be a key intermediate in this reaction. Without CuCl, the reaction of compound **4a** generated the product **3aa** in just 12% yield (Scheme 2e). The reaction of compound **4a** under N<sub>2</sub> delivered only a trace amount of the product (Scheme 2f). The studies demonstrated the importance of CuCl and O<sub>2</sub>.

On the basis of this and previous reports, a possible reaction mechanism was proposed (Scheme 3). Initially, the Michael addition of indolylnaphthoquinone (1a) and aniline (2a) in the presence of a base gave the intermediate **A**, which was immediately oxidized to intermediate 4a by  $O_2$  or the oxidative naphthoquinone.<sup>18</sup> On the other hand, a copper salt was oxidized to the Cu(m) species in the presence of  $O_2$ , which then reacted with intermediate 4a giving the Cu(m) species **B**.



Scheme 3 Plausible reaction mechanism.

Finally, the intermediate **B** underwent reductive elimination producing the desired product **3aa** and regenerating Cu(III) with the oxidant (O<sub>2</sub>).<sup>19</sup>

### Conclusions

In conclusion, we have successfully demonstrated Cu(1)-catalyzed cycloamination reaction between indolylquinones and various (hetero)aromatic amines to accomplish the polycyclic N-heterocyclic molecules. The significant aspects of our work allow modest functional group tolerance, including electrondonating and electron-withdrawing groups, which were compatible under the current methodology.

# Conflicts of interest

There are no conflicts to declare.

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