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The De Novo Design and Synthesis of Cyclic Urea Inhibitors of Factor Xa: Optimization of the S4 Ligand[†]

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Abstract—In this report refinements to the S4 ligand group leads to compound **19**, an inhibitor of fXa with good potency in vitro and an improved pharmacokinetic profile in rabbit. The X-ray crystallographic study of a representative analogue confirms our binding model for this series. © 2000 DuPont Pharmaceuticals Company. Published by Elsevier Science Ltd. All rights reserved.

Our effort to develop oral anticoagulants has focused on inhibitors of the blood coagulation protease Factor Xa $(fXa)^{1-6}$ because of its role as the chief agent of thrombin generation in the common pathway of coagulation^{7,8} and reports suggesting that inhibitors of fXa exert effective control of thrombogenesis with a minimal effect upon bleeding.^{9,10} The current drug of choice for oral anticoagulant therapy, Coumadin[®], is highly effective and being utilized in a growing number of antithrombotic indications.^{11,12} However, it is a narrow therapeutic index drug that requires careful dose titration, stabilization, and periodic monitoring for each patient.¹³ Our objective has been to discover an orally bioavailable drug with an optimal pharmacokinetic profile amenable to once- or twice-daily administration.

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In a previous report,14 we disclosed the design and synthesis of a new class of fXa inhibitors utilizing a cyclic urea¹⁵ 2,4-diazepin-3-one core. The initial compound of this series, 1, utilized the 2,4-diazepin-3-one ring as a rigid platform upon which to tether ligand groups: a meta-benzamidine to interact with the S1 pocket and a para-benzamidine to interact with the S4 pocket of the fXa enzyme. It was anticipated that a dicationic species such as 1 would be poorly adsorbed in vivo, hence the discovery of 2, which has the substantially less basic N-benzylpiperidine as a ligand for the S4 pocket, was considered an important advance and used as a lead for the development of this series. In this paper, we report on further refinements to the S4 ligand group which gives a 10-fold increase of in vitro activity over 2, the in vivo pharmacokinetic profile of selected analogues, and an X-ray crystallographic study that confirms our binding model for these compounds.

Synthesis

The compounds for this study **6–19** were prepared according to the methods outlined in Scheme 1.¹⁶ Intermediate **4** is prepared in good (75%) yield by stirring the protected 4-aminopiperidine 3^{17} with distilled 3-cyanophenylisocyanate in DMF at ambient temperature. The 2,4-diazepin-3-one ring is formed by di-*N*-alkylation of the urea functionality of **4** with 1,4-dibromobutane and sodium hydride in hot DMF in 65% yield; compound **5** is obtained by Boc deprotection

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of the dialkylation product with HCl in ether. Compounds 6, 14–16 and 18 were prepared by reaction with the appropriate alkylchloride in THF with triethylamine as base, then the amidine was formed using Pinner's method¹⁸ by reaction of the nitrile substituent with anhydrous HCl gas saturated in methyl acetate: methanol (5:1) at 0°C followed by treatment with excess ammonium carbonate in methanol at ambient temperature. The amides 7 and 8 were obtained by acylation with benzovl or phenylacetyl chloride in triethylamine/THF solution followed by amidine formation as described above. Sulfonamides 9, 10, 17 and 19 were derived from 5 by reaction with the corresponding sulfonylchloride in triethylamine/THF then amidine formation. Compounds 11–13 were prepared by sulfonylation of 5 with 2-, 3- or 4-nitrophenyl sulfonylchloride followed by amidine formation, then the nitro group was reduced by catalytic hydrogenation to give the amines.

Results and Discussion

Table 1 illustrates the effect of varying the S4 ligand upon in vitro biological activity. The fXa inhibition of this series shows a dependence on the type of bond used to link the piperidine group to the terminal aryl ring. Compared to the *N*-alkylpiperidines **2** and **6**, the *N*acylpiperidine analogues **7** and **8** are substantially less potent while the *N*-sulfonylpiperidines **9** and **10** are 4and 22-fold more active than their related alkyl homologues. The sulfonyl anilines **11–13** represent an unsuccessful attempt to develop a new hydrogen bonding interaction with the S4 subsite. The series of picolinyl



Scheme 1. Synthesis of 2,4-diazepin-3-one analogues. Reagents: (i) 3-cyanophenylisocyanate, DMF, 18 h; (ii) 3 equiv NaH, 2 equiv BrCh₂CH₂CH₂CH₂Br, DMF, 70 °C, 3 h, then HCl:Et₂O; (iii) **R**-Cl (see Table 1), Et₃N, THF, then HCl(g), CH₃CO₂CH₃:CH₃OH (5:1), 0–10 °C, 18 h, then 5 equiv NH₄CO₃, CH₃OH, 18 h; (iv) **R**-Cl (2-, 3-, 4-nitrophenylsulfonyl chlorides), Et₃N, THF, then HCl(g), CH₃CO₂-CH₃CH₃OH (5:1), 0–10 °C, 18 h, then 5 equiv (NH₄)₂CO₃, CH₃OH, 18 h, then H₂, Pd-C, MeOH.

analogues 14–16 shows that 14 is exceptionally active, perhaps due to protonation of the pyridine nitrogen. However, as demonstrated by 17, further improvement in inhibition is not observed when this modification is combined with a sulfonamide linkage. Thiophenes 18 and 19 follow the same trends observed for the carbocyclic analogues. Finally, while the more potent compounds 9–11, 14, 17 and 19 have 50- to 220-fold selectivity for fXa over thrombin, the selectivity against trypsin is 3- to 10-fold.

The iv pharmacokinetic profiles of 10, 17 and 19 were determined in rabbits after a 1 h infusion of compound at 5 mg/kg (Table 2). While the benzylsulfonamide 10 and picolinylsulfonamide 17 are rapidly cleared and have half-lives of about 2 min, they do achieve an appreciable C_{max} with small apparent volumes of distribution (V_d) . The thiophene sulfonamide 19 has a more encouraging profile; clearance of this compound is approximately 10-fold lower than for 10 and 17, resulting in a half-life of 1 h, a 10-fold increase in C_{max} and little change in the volume of distribution. In a rabbit arterio-venous shunt model of thrombosis,^{3,4} compound 19 was shown to inhibit thrombus formation with an $ID_{50} = 7 \mu mol/kg/h$ upon iv administration. The potential for transport across the intestinal mucosa and in vivo absorption of 19 was evaluated by measuring permeation across Caco-2 monolayers.²¹ While 19 was soluble in the aqueous vehicle used for this assay,²² the measured permeability coefficient (2×10^{-7} cm/s) suggests poor oral absorption.

Because of the general difficulty in obtaining the X-ray crystal structure of fXa bound with an inhibitor, an X-ray crystal structure of 10 complexed with the related enzyme thrombin was determined (Fig. 1).23 This thrombin structure confirms the proposed binding model¹⁴ for these compounds: the *m*-benzamidine group binds in the S1 pocket by forming a bidentate interaction with Asp189, the N-(benzylsulfonyl)piperidine group reaches into the S4 pocket to form an edge-toface interaction with Trp215 and the urea carbonyl forms an H-bond acceptor interaction with the Gly216 amide NH. Asp189, Trp215 and Gly216 have analogous residues in fXa. Minimization²⁴ of the thrombin-bound conformation of 10 in the crystal structure of human des(1-45) fXa²⁵ (Fig. 2) gives comparable distances between the amide NH of Gly216 and the urea carbonyl of 10 in fXa (2.97 Å) versus the corresponding distance in the thrombin structure (2.87 Å). Likewise, the distances observed for the Asp189 to benzamidine saltbridge (fXa model: 2.69, 2.8 Å; thrombin structure: 2.85, 2.97 Å) and the centroid separation²⁶ observed for
 Table 1. In vitro inhibitory activity¹⁹



	R	x	FXa Thrombin Trun		
	K	Λ	$K_{\rm i}$ (nM)	$K_{\rm i}$ (nM)	$K_{\rm i} ({\rm nM})$
2	~~~~	2•Cl [−]	102	2000	282
6		2•C1-	220	3600	1000
7		TFA ^{-a}	7800	>21,000	_
8	p ^{r^r}	TFA-	900	11,800	_
9	ξ-0 ₂ 8-√	TFA-	24	1600	60
10	}-0 ₂ \$	TFA ⁻	10	500	80
11	H ₂ N {-O ₂ S	2•TFA ⁻	34	2400	130
12	§-02S-√NH5	2•TFA [−]	200	2200	
13	§-028-√-NH2	2•TFA-	71	6300	190
14	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	2•TFA ⁻	12	2600	120
15	sort N	2•TFA ⁻	670	5300	
16	N	2•TFA [−]	200	6800	_
17	[₹] -0 ₂ \$_	2•TFA [−]	10	500	57
18	store S	TFA-	80	1400	170
19	§-0₂ S→	Cl-	12	1500	52

^aTFA⁻, trifluoroacetate anion.

 Table 2.
 Pharmacokinetics of selected inhibitors²⁰

	Clearance (L/h/kg)	$T_{1/2}$ (h)	$C_{\max} \left(\mu \mathbf{M} \right)$	V _d (L/kg)
10	4.1	0.04	4.8	0.23
17	7.9	0.03	2.4	0.34
19	0.68	1.1	36	1.1



Figure 1. Electron density of 10 complexed to thrombin. Good density is observed over all of the inhibitor except for the carbon next to the sulfonamide.



Figure 2. Model of the thrombin-bound conformation of compound 10 after minimization in the crystal structure of human des(1-45) fXa.²³

the Trp215 to benzylsulfonyl edge-to-face interaction (fXa model: 5.28 Å; thrombin structure: 4.97 Å) are similar.

Conclusions

In our previous report,¹⁴ we established by modeling and SAR that the seven-membered 2,3-diazepin-3-one core is the optimium ring size for this series. In this work, an X-ray crystal structure for the thrombin complex of **10** confirms the assumptions made in our binding model for these inhibitors. Furthermore, the binding affinity for fXa is increased substantially through the introduction of an N-sulfonyl linkage in the S4 ligand group, while the best pharmacokinetic profile in rabbits is observed for **19**, an analogue with an N-(sulfonylthiophen-2-yl)piperidine group as the S4 ligand.

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24. Compound **10** was modeled into a crystal structure of Factor Xa (ref 25) using InsightII (MSI, San Diego, CA). The starting conformation of the inhibitor was from the crystal structure of **10** in thrombin. The inhibitor was minimized in a rigid protein using the CFF98 forcefield.

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