Asymmetric Sequential Allylic Transfer Strategy for the Synthesis of (–)-Adaline and (–)-Euphococcinine

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Received 29 February 2009

Abstract: Enantioselective synthesis of the piperidine alkaloids, (–)-adaline (1) and (–)-euphococcinine (2), were achieved from 7 in a six-step sequence through stepwise allylic transfer reactions. Dramatic additive effect of Bu_3SnF for the conversion of 3 into 9 was observed to expedite the process to afford the cyclized products in good yields.

Key words: alkaloid, allylation, chiral, cyclization, stereoselectivity

Development of stereocontrolled sequential processes can offer advantages over the stepwise transformations by increasing chemical efficacy due to a simple operation.¹ In the course of our research program aimed at finding new synthetic methods in the stereoselective construction of cyclic system via stepwise allylic transfer reactions,² we disclosed our investigations on the transition-metal-catalyzed intramolecular allylation mainly between allene and carbonyl functionality to afford cyclic compounds in high levels of stereoselectivity.³ With these observations in hand, we became quite interested in designing an enantioselective synthetic route for naturally occurring the piperidine alkaloids (–)-adaline (1) and (–)-euphococcinine (2) possessing bridged cyclic system as shown in Scheme 1.



Scheme 1

(–)-Adaline (1) was isolated from secretion of the ladybugs *Cryptolaenus mintrouzieri* and *Adalia bipunctata*.⁴ A methyl analogue, (+)-euphococconine (ent-2), has been found from the plant and animal kingdoms. It was first isolated from *Euphorbia acto* and has also been found in the defense secretion of ladybugs *Cryptolaenus mintrouzieri* and *Epilachna varivestis*.⁵ These bridged piperidine alkaloids⁶ were found to exhibit repulsive activity against different insect. The biological behaviors coupled with their interesting structural features have stimulated considerable synthetic efforts to construct bicyclic piperidine systems.⁷ It was envisaged that the asymmetric sequential allylic transfer reactions starting from **5** and **6**⁸ leading to the synthesis of **1** and **2** could be realized by the four-step sequence as described in Scheme 1.

With this issue in mind, the 5-oxoaldehydes 5 were prepared according to the modification of known procedure.⁹ Treatment of 7 with a mixture of *n*-BuLi and KOt-Bu in the presence of TMEDA in pentane at -78 °C to -20 °C for 1 hour, followed by reaction with corresponding alkyl iodides in THF at -20 °C for 4 hours, and then acidic cleavage of acetal afforded 5 as described in Scheme 2. The stage was thus set for the allylic transfer reaction of 5 with **6** catalyzed by a chiral Lewis acid catalyst.^{2a} The allylic transfer reactions were conducted by the dropwise addition of 6 in PhCF₃ at -20 °C to a mixture of (S)-BINOL-Ti(IV)[OCH(CF₃)₂]₂ (5 mol%) and 5 in PhCF₃. After 12 hours at -20 °C, the mixture was quenched by addition of a saturated aqueous NaHCO₃. Workup and chromatography on triethylamine-treated silica gel gave 4 in good yields with high levels of enantioselectivity as shown in Scheme 2.



SYNLETT 2009, No. 9, pp 1498–1500 Advanced online publication: 13.05.2009 DOI: 10.1055/s-0029-1217172; Art ID: U01609ST © Georg Thieme Verlag Stuttgart · New York

Scheme 2

The next stage was the synthesis of a cyclic imine **3** as a precursor for an intramolecular allylic transfer reaction to establish a bridged piperidine system. Compound **4** was converted into the azido-ketone **8** using Mitsunobu conditions in good yields.¹⁰ Treatment of **4a** with DIAD and HN₃ (benzene solution) in THF at 20 °C for 2 hours gave **8a** in 81% purified yield. With the azido-ketone **8**, we tried to cyclize **8** to a tetrahydropyridine **3** via an aza-Wittig reaction.¹¹ In the event, the azido-ketones **8** were converted into the corresponding tetrahydropyridines **3** by treating with Ph₃P at 20 °C for 4 hours in diethyl ether in good yields. However, attempts to a direct conversion of **4a** into **3a** under same conditions except the use of excess Ph₃P (4 equiv) gave only marginal due to low chemical yield (Scheme 3).





With the tetrahydropyridines **3** in hand, we had reached a suitable intermediate for the construction of the required piperidine 9 through an intramolecular allylic transfer reaction under proper reaction conditions. Several attempts to cyclization of 3a (R = *n*-pentyl) with Lewis acids such as $BF_3 \cdot OEt_2$, $SnCl_4$, and $TiCl_4$ indicated that the conversion into the corresponding 9a could not be realized under various conditions. The use of TMSOTf turned out to be a limited outcome. We observed the formation of only trace of product (ca 10%) except the decomposition of the intermediate. We then turned to other reagents especially Brønsted acids that could promote this conversion. A series of such reagents was tested without success. Finally, we were delighted to find that CF₃SO₃H could be a proper reagent for this purpose. After surveying numerous conditions as summarized in Table 1, several key findings emerged: i) the use of CF_3SO_3H turned out to be marginal but encouraging; ii) the introduction of Bu₃SnOMe as an additive increased chemical yield; iii) a dramatic additive effect was observed by introducing Bu₃SnF in terms of reactivity and chemical yield presumably due to an exchange of silane to stannane; iv) the use of toluene proved to be most suitable solvent compared to others such as CH₂Cl₂ and MeCN.

Under optimal conditions, the reaction was conducted by an addition of CF_3SO_3H (1.1 equiv) in toluene at 0 °C to a solution of **3a** (1 equiv) in toluene. After 5 minutes, Bu₃SnF (1.2 equiv) was slowly added to the resulting reaction mixture. After stirring at 0 °C for 1 hour, the reaction mixture was warmed to 20 °C. Reaction was then allowed to proceed for additional 12 hours. The reaction was quenched by addition of saturated aqueous NaHCO₃. The aqueous layer was extracted with diethyl ether. After removal volatile materials under reduced pressure, final purification was effected by column chromatography to yield **9a** in 81% yield.

Table 1 Synthesis of Compounds 9 under Various Conditions



Entry	Additive ^a	Solvent	Yield (%) ^b
1	none	toluene	21
2	Bu ₃ SnOMe	toluene	38
3	Bu ₃ SnF	CH_2Cl_2	66
4	Bu ₃ SnF	toluene	81

^a 1.2 equiv.

^b Refer to purified yields.

The final transformation was an oxidative cleavage of an olefin to a ketone. We were surprised that the ozonolysis of **9a** followed by treating with Me₂S did not afford the corresponding **1** except the decomposition of starting material to a mixture of unidentified compounds. Acetylation of **9a** gave the amide **10a**, which was converted into the corresponding ketone **11a** by the same ozonolysis process. After examined with several reagents, the final step for the synthesis of **1** {64%, $[\alpha]_D^{20}$ –12.2 (CHCl₃, *c* 1.1)} and **2** {71%, $[\alpha]_D^{20}$ –6.1 (*c* 1.3, MeOH)} was achieved by the reaction of **9** with KIO₄ in the presence of OsO₄ catalyst (3 mol%) in a mixture of THF and H₂O at 20 °C as shown in Scheme 4.¹²

In summary, this paper describes the enantioselective synthesis of the piperidine alkaloids (–)-adaline (1) and (–)-euphococcinine (2) starting from the commercially available 7 in a six-step sequence through the stepwise sequential asymmetric allylic transfer reactions as key transformations.



Acknowledgment

Generous financial support from the Korea Research Foundation (KRF-2006-312-C00234; KRF-2005-005-J11901) and the Korea Science & Engineering Foundation (R01-2007-000-20315-0) is gratefully acknowledged.

References

- For recent examples, see: (a) Flamme, E.; Roush, W. R. J. Am. Chem. Soc. 2002, 124, 13644. (b) Wang, X.; Meng, Q.; Nation, A. J.; Leighton, J. L. J. Am. Chem. Soc. 2002, 124, 10672. (c) Nakamura, M.; Hatakeyama, T.; Hara, K.; Hukudome, H.; Nakamura, E. J. Am. Chem. Soc. 2004, 126, 14344. (d) Halland, N. H.; Aburel, P. S.; Jørgensen, K. A. Angew. Chem. Int. Ed. 2004, 43, 1272; and references cited therein.
- (2) (a) Yu, C.-M.; Lee, J.-Y.; So, B.; Hong, J. Angew. Chem. Int. Ed. 2002, 41, 161. (b) Yu, C.-M.; Kim, J.-M.; Shin, M.-S.; Yoon, M.-O. Chem. Commun. 2003, 1744.
- (3) (a) Kim, S. H.; Oh, S.-J.; Ho, P.-S.; Kang, S.-C.; O, K.-J.; Yu, C.-M. Org. Lett. 2008, 10, 265. (b) Kim, S.-H.; Oh, S.-J.; Kim, Y.; Yu, C.-M. Chem. Commun. 2007, 5025.
 (c) Yu, C.-M.; Youn, J.; Jung, J. Angew. Chem. Int. Ed. 2006, 45, 1553. (d) Yu, C.-M.; Youn, J.; Jung, H.-K. Bull.

Korean Chem. Soc. **2006**, *27*, 463. (e) Yu, C.-M.; Youn, J.; Lee, M.-K. *Org. Lett.* **2005**, *7*, 3733. (f) Yu, C.-M.; Youn, J.; Hong, Y.-T.; Yoon, S.-K. *Org. Lett.* **2005**, *7*, 4507. (g) Yu, C.-M.; Hong, Y.-T.; Lee, J. *J. Org. Chem.* **2004**, *69*, 8506.

- (4) (a) Tursch, B.; Braekman, J.-C.; Daloze, D.; Hootele, C.; Losman, D.; Karlsson, R.; Pasteels, J.-M. *Tetrahedron Lett.* **1973**, 201. (b) Lognay, G.; Hemptinne, J. L.; Chan, F. Y.; Gaspar, C. H.; Marlier, M.; Braekman, J. C.; Daloze, D.; Pasteels, J. M. *J. Nat. Prod.* **1996**, *59*, 510.
- (5) (a) Hart, N. K.; Johns, S. R.; Lamberton, J. A. Aust. J. Chem. 1967, 20, 561. (b) Brown, W. V.; Moore, B. P. Aust. J. Chem. 1982, 35, 1255. (c) Eisner, T.; Goetz, M.; Aneshausley, D.; Ferstandig-Arnold, G.; Meinwald, J. Experientia 1986, 42, 204.
- (6) For a review, see: King, A. G.; Meinwald, J. Chem. Rev. 1996, 96, 1105.
- (7) (a) Itoh, T.; Yamazaki, N.; Kibayashi, C. Org. Lett. 2002, 4, 2469. (b) Mechelke, M. F.; Meyers, A. I. Tetrahedron Lett. 2000, 41, 4339. (c) Davison, E. C.; Holmes, A. B.; Forbes, I. T. Tetrahedron Lett. 1995, 36, 9047. (d) Yue, C.; Royer, J.; Husson, H.-P. J. Org. Chem. 1992, 57, 4211.
- (8) (a) Kang, K.-T.; U, J. S.; Park, D.; Kim, J. G.; Kim, W. J. Bull. Korean Chem. Soc. 1995, 16, 464. (b) Keck, G. E.; Covel, J. A.; Schiff, T.; Yu, T. Org. Lett. 2002, 4, 1189.
 (c) Keck, G. E.; Truong, A. P. Org. Lett. 2005, 7, 2153.
 (d) Wender, P. A.; Verma, V. A. Org. Lett. 2008, 10, 3331.
 (e) Kang, K.-T.; Sung, T. M.; Jung, H. C.; Lee, J. G. Bull. Korean Chem. Soc. 2008, 29, 1669.
- (9) (a) Yu, C.-M.; Choi, H.-S.; Lee, J.-K.; Yoon, S.-K. J. Org. Chem. 1997, 62, 6687. (b) Yu, C.-M.; Jung, W.-H.; Choi, H.-S.; Lee, J.; Lee, J.-K.; Yoon, S.-K. Tetrahedron Lett. 1995, 36, 8255.
- (10) (a) Mitsunobu, O. Synthesis 1981, 1. (b) Dembinski, R. Eur. J. Org. Chem. 2004, 2763.
- (11) (a) Loiseleur, O.; Ritson, D.; Nina, M.; Crowley, P.; Wagner, T.; Hanessian, S. *J. Org. Chem.* 2007, 72, 6353.
 (b) Stork, G.; Niu, D.; Fujimoto, A.; Koft, E. R.; Balkovec, J. M.; Tata, J. R.; Dake, G. R. *J. Am. Chem. Soc.* 2001, *123*, 3239.
- (12) Baggiolini, E. G.; Iacobelli, J. A.; Hennessy, B. M.; Batcho, A. D.; Sereno, J. F.; Uskokovic, M. R. *J. Org. Chem.* **1986**, *51*, 3098.

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