Carbohydrate Research 396 (2014) 48-53

Contents lists available at ScienceDirect

Carbohydrate Research

journal homepage: www.elsevier.com/locate/carres

Useful approach to the synthesis of aryl thio- and selenoglycosides in the presence of rongalite



Department of Organic Chemistry, Indian Institute of Science, Bangalore 560 012, India

ARTICLE INFO

ABSTRACT

Article history: Received 21 June 2014 Received in revised form 15 July 2014 Accepted 15 July 2014 Available online 21 July 2014

Keywords: Thioglycoside Phenyl selenoglycoside Diaryl dichalcogenide Rongalite Reductive cleavage

Thioglycosides play a vital role as glycosyl donors for the synthesis of oligosaccharides.^{1–4} The potential ability of thioglycosides as activators toward various reagents⁴ makes them attractive and stable glycosyl donors. Moreover, the exceptional stability of thioglycosides is useful in medicinal and biochemical applications.⁵ In the literature, a number of methods are available for the synthesis of thioglycosides.^{6–18,20–24} Most commonly, the synthesis of thioglycosides involves the treatment of peracetylated sugars with the appropriate thiol in the presence of a Lewis acid or bases.^{7–12} The other methods include the generation of nucleophilic thiolate anion in situ from disulfides using Zn dust/ZnCl₂¹³ or Lewis acids¹⁴ during the reaction with glycosyl halides or peracetylated sugars and the reaction of S-glycosyl isothiouronium intermediates with alkyl halides in the presence of a mild base.^{10,15,16} Despite the usefulness of the above methods, they have limitations such as the use of foul smelling thiols, use of expensive reagents, long reaction times, lower yields, and harsh reaction conditions. In this context, developing a convenient and green method would be of greater importance. From our laboratory, we reported earlier that rongalite (HOCH₂SO₂Na), **1** is a useful reagent for the cleavage of diaryl dichalcogenides to generate the corresponding chalcogenide anion in situ and it has been utilized for the ring opening of aziridines and epoxides.¹⁷ Rongalite, **1** is a cheap and commercially available reagent and is used in many organic transformations.¹⁸ The

advantage of rongalite, **1** over the other reducing agents includes its cost effectiveness, mild reaction conditions, and simple work up procedure. In continuation of our study to explore the scope and limitation on the use of this reagent, we anticipated that the in situ generated organo chalcogen anion from diaryl dichalcogenides and rongalite, **1** can be used in reaction with glycosyl halides (Scheme 1) to furnish the corresponding thioand selenoglycosides.

A simple, mild, and cost effective methodology has been developed for the synthesis of aryl thio-and sele-

noglycosides from glycosyl halides and diaryl dichalcogenides. Diaryl dichalcogenides undergo reductive

cleavage in the presence of rongalite (HOCH₂SO₂Na) to generate a chalcogenide anion in situ followed by

reaction with glycosyl halides to furnish the corresponding aryl thio- and selenoglycosides in excellent

yields. Using this protocol, synthesis of 4-methyl-7-thioumbelliferyl-β-D-cellobioside (MUS-CB), a

fluorescent non-hydrolyzable substrate analogue for cellulases has been achieved.

Accordingly, a reaction was carried out with 2,3,4,6-tetra-O-acetyl- α -D-glucopyranosyl bromide **2a** and diphenyl disulfide **3a** in the presence of rongalite **1**. Addition of **2a** (1 equiv) to a well-stirred solution of diphenyl disulfide **3a** (0.5 equiv), potassium carbonate (2 equiv), and rongalite **1** (3 equiv)¹⁹ in *N*,*N*-dimethylformamide (DMF) at room temperature resulted in the formation of the corresponding phenyl thioglycoside **4a** (1 h) in 93% yield.

The stereochemical outcome of the reaction was confirmed by NMR spectroscopy. The ¹H NMR of the product **4a** showed a doublet at δ 4.71 for H-1, with $J_{1,2} \sim 10.0$ Hz and the signal at δ 5.23 (t, $J_{2,3} \sim 10.0$ Hz) for H-2 which clearly demonstrated that the product **4a** has a 1,2-*trans* configuration without any 1,2-*cis* linkage. The mechanism of the reaction involves the base promoted decomposition of rongalite to form formaldehyde and HSO₂⁻¹⁷ The HSO₂⁻ transfers a single electron to disulfide **3a** to form a thiolate ion in situ which reacts at the anomeric carbon (C-1) of **2a** in an S_N2 fashion to give the desired product **4a** with 1,2-*trans* configuration.

After the successful synthesis of thioglycoside **4a**, we decided to extend the scope of the method with other disulfides **3b**-**3f** using



Note



© 2014 Elsevier Ltd. All rights reserved.



^{*} Corresponding author. Tel.: +91 80 22932404; fax: +91 80 23600529. *E-mail address:* scn@orgchem.iisc.ernet.in (S. Chandrasekaran).



Scheme 1. General scheme for the synthesis of aryl thioglycosides and selenoglycosides.

2a as the partner. In all the cases the corresponding thioglycosides **4b–4f**, respectively, were obtained in good to excellent yields. The results are summarized in Table 1.

From the Table 1, it can be seen that, irrespective of the nature of the substituent on the phenyl moiety of the disulfide, the corresponding thioglycosides **4b**-**4e** were obtained in good to excellent yields (Table 1, entries 1–4). 2,2'-Dipyridyl disulfide **3f** underwent reductive cleavage under these conditions followed by its reaction with **2a** to furnish the corresponding thioglycoside **4f** in moderate yield (65%) after 2 h (Table 1, entry 5). However, aliphatic disulfides such as dibenzyl disulfide **3g** failed to react under these conditions to form the corresponding thioglycoside **4g** even after 6 h (Table 1, entry 6).

Then we further screened the reactivity of **3a** with other glycosyl halides **2b–2f** (Scheme 3). The glycosyl bromides **2b**, **2c**, and **2f** reacted smoothly (1 h) with **3a** in the presence of **1** to form the corresponding thioglycosides **4h**, **4i**, and **4l**, respectively, in excellent yields (Table 2, entries 2, 3, 6) under similar conditions described in Scheme 3. The glucosamine derived glycosyl chloride **2e** took 1 h to form the corresponding thioglycoside **4k** in 83% yield (Table 2, entry 5) whereas the glycosyl bromide **2d** having bromine at C-6 furnished the desired product **4j** in 89% yield after 3 h (Table 2, entry 4).

Like diaryl disulfides, diphenyl diselenide **3h** also showed the ability to cleave in a reductive fashion with rongalite **1** (Scheme 2). The reaction of glycosyl halides **2a–2f** with diphenyl diselenide **3h** in the presence of rongalite (25 °C) resulted in the formation of the corresponding selenoglycosides **5a–5f**, respectively, in good to excellent yields. The results are presented in Table 2. While the glucose, galactose, and lactose derived anomeric bromides **2a–2c**, **2f**, respectively, reacted with ease (1 h) with diselenide **3h** in the presence of **1** (Table 2, entries 1–3, 6). The reaction of other glycosyl halides **2d** and **2e** were slower (2–4 h) (Table 2, entries 4, 5).



Synthesis of various thioglycosides 4b-4f derived from 2a



Table 2

Synthesis of phenyl thio- and selenoglycosides



Scheme 2. Synthesis of phenyl thioglycoside 4a derived from 2a and 3a.



Scheme 3. Reaction of glycosyl halides 2 with diphenyl disulfide (3a)/diselenide (3h) in the presence of rongalite 1.

Finally, the synthetic potential of this method has been demonstrated in the preparation of peracetylated derivative of 4-methyl-7-thioumbelliferyl- β -D-cellobioside **4m** which is the key precursor for **MUS-CB**, a fluorescent non-hydrolyzable substrate analogue for cellulases.²⁰ Thus treatment of anomeric bromide **2g** with disulfide of **3i** derived from 7-mercaptocoumarin²¹

in the presence of rongalite **1** furnished the corresponding peracetylated **MUS-CB 4m** in excellent yield (86%) with exclusive β -selectivity (Scheme 4). The deacetylation of **4m** in the presence of NaOMe/MeOH furnished **MUS-CB**.⁶

In summary, we have presented a useful, alternative approach to the synthesis of aryl thio- and selenoglycosides. The method



Scheme 4. Synthesis of 4-methyl-7-thioumbelliferyl-β-D-cellobioside (**MUS-CB**).

involves the reductive cleavage of diaryl dichalcogenides using rongalite to generate the chalcogenide anion in situ which reacts with glycosyl halides. The significance of this method is its cost effectiveness and avoiding the use of foul smelling organo chalcogenols.

1. Experimental section

1.1. General information

All the reactions were performed in an oven dried apparatus and stirred magnetically. Chemicals and solvents were either purchased or purified by standard techniques. Analytical TLC was performed on commercial plates coated with silica gel GF254 (0.25 mm). Silica gel (230-400 mesh) was used for column chromatography. Melting point values reported are uncorrected. Infrared spectra were recorded using the JASCO FTIR instrument and the frequencies are reported in wave numbers (cm^{-1}) and intensities of the peaks are denoted as s (strong), w (weak), m (medium), broad (br). ¹H and ¹³C NMR spectra were recorded on a Jeol 400 MHz (100 MHz, ¹³C) NMR spectrometer and calibrated using tetramethylsilane (TMS) for (¹H) or residual undeuterated solvent (CDCl₃) as an internal reference. Chemical shifts are reported in parts per million and multiplicity is indicated using the following abbreviations: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), bs (broad singlet) etc. Coupling constants are reported wherever it is necessary in Hertz (Hz). Highresolution mass spectra (HRMS) were recorded on micromass **Q-TOF** electrospray.

1.2. General procedure for the synthesis of aryl thio- and selenoglycosides

To a well stirred solution of the aryl disulfides/diphenyl diselenide (0.2 mmol, 0.5 equiv) in DMF (2 mL), was added rongalite **1** (1.2 mmol, 3 equiv) and K₂CO₃ (0.8 mmol, 2 equiv). The reaction mixture was stirred for 10 min at 25 °C followed by the addition of glycosyl halide (0.4 mmol, 1 equiv) and stirring was continued. The completion of the reaction was monitored by TLC till the consumption of disulfides/diselenide. Then the reaction mixture was quenched with water and extracted with ethyl acetate (20 mL × 2). The organic layer was separated, then washed with brine, extracted with ethyl acetate (5 mL × 2), and dried over anhydrous Na₂SO₄. The solvent was evaporated under vacuum and the crude product was purified by column chromatography on silica gel (230–400 mesh) using ethyl acetate/petroleum ether as eluent.

1.2.1. Phenyl 2,3,4,6-tetra-O-acetyl-1-thio- β -D-glucopyranoside (4a)^{11,15}

Yield 93%; white solid; mp: 111–113 °C; $[\alpha]_D^{25}$ –20.9 (*c* 0.8, CHCl₃); FTIR (neat): 1746 (s), 1440 (w), 1372 (m), 1254 (m), 1226 (s), 1088 (w), 1043 (m), 913 (w), 743 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.51–7.49 (m, 2H), 7.33–7.30 (m, 3H), 5.23 (t, *J* = 9.3 Hz, 1H), 5.01 (t, *J* = 9.3 Hz, 1H), 4.98 (t, *J* = 10.0, 1H), 4.71 (d, *J* = 10.1, 1H), 4.25–4.16 (m, 2H), 3.75–3.71 (m, 1H), 2.09–1.99 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 170.2, 169.4, 169.3, 133.2, 131.7, 129.0, 128.5, 85.8, 75.8, 74.0, 70.0, 68.2, 62.2, 20.8, 20.8, 20.6; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₄O₉SNa (M+Na)⁺: 463.1039; found: 463.1035.

1.2.2. 4-Tolyl-phenyl 2,3,4,6-tetra-O-acetyl-1-thio-β-Dglucopyranoside (4b)^{1,11}

Yield 93%; white solid; mp: 115–117 °C; $[\alpha]_D^{25}$ –19.0 (*c* 1.0, CHCl₃); FTIR (KBr): 3449 (s), 1750 (s), 1382 (w), 1228 (s), 1041 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.39 (d, *J* = 8.0, 2H), 7.13 (d, *J* = 7.9 Hz, 2H), 5.21 (t, *J* = 9.4 Hz, 1H), 5.03 (t, *J* = 9.8 Hz, 1H), 4.94 (t, *J* = 9.6 Hz, 1H), 4.64 (d, *J* = 10.0 Hz, 1H), 4.24–4.16 (m, 2H), 3.73–3.68 (m, 1H), 2.35–1.99 (5s, 15H); ¹³C NMR (100 MHz, CDCl₃): δ 170.7, 170.3, 169.5, 169.3, 138.9, 133.9, 129.7, 127.5, 85.8, 75.8, 74.0, 69.9, 68.2, 62.2, 21.2, 20.8, 20.8, 20.6, 20.6; HRMS (ESI-TOF) *m/z*: calcd for C₂₁H₂₆O₉SNa (M+Na)⁺: 477.1195; found: 477.1196.

1.2.3. 4-Methoxy-phenyl 2,3,4,6-tetra-O-acetyl-1-thio-β-Dglucopyranoside (4c)¹¹

Yield 88%; white solid; mp: 95–97 °C; $[\alpha]_D^{25}$ –47.6 (*c* 2.6, CHCl₃); FTIR (KBr): 1749 (s), 1592 (w), 1495 (w), 1384 (m), 1287 (m), 1226 (s), 1037 (s), 920 (w), 832 (m), 605 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.44 (d, *J* = 8.7 Hz, 2H), 6.85 (d, *J* = 8.7 Hz, 2H), 5.19 (t, *J* = 9.4 Hz, 1H), 4.99 (t, *J* = 9.8 Hz, 1H), 4.89 (t, *J* = 9.6 Hz, 1H), 4.56 (d, *J* = 10.0 Hz, 1H), 4.19 (bs, 2H), 3.81 (s, 3H), 3.69–3.66 (m, 1H), 2.11–1.98 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.5, 170.2, 169.4, 169.2, 160.4, 136.5, 120.9, 114.4, 85.6, 75.7, 74.0, 69.8, 68.1, 62.0, 55.3, 20.8, 20.7, 20.6, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₁H₂₆O₁₀SNa (M+Na)⁺: 493.1144; found: 493.1146.

1.2.4. 4-Chlorophenyl 2,3,4,6-tetra-O-acetyl-1-thio-β-Dglucopyranoside (4d)¹¹

Yield 86%; white solid; mp: 112–114 °C; $[\alpha]_D^{22}$ –41.1 (*c* 4.3, CHCl₃); FTIR (KBr): 3293 (m), 2234 (m), 1751 (s), 1610 (w), 1382 (w), 1225 (s), 1167 (w), 1040 (m), 919 (w), 836 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.44 (d, *J* = 8.3 Hz, 2H), 7.29 (d, *J* = 8.5 Hz, 2H), 5.22 (t, *J* = 9.3 Hz, 1H), 5.02 (t, *J* = 9.8 Hz, 1H), 4.94

(t, *J* = 9.6 Hz, 1H), 4.65 (d, *J* = 10.0 Hz, 1H), 4.21–3.70 (m, 3H), 2.09– 1.99 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.5, 170.1, 169.3, 169.2, 134.9, 129.4, 129.0, 85.1, 75.8, 73.8, 69.7, 68.0, 62.0, 20.7, 20.5; HRMS (ESI-TOF) *m*/*z*: calcd for C₂₀H₂₃ClO₉SNa (M+Na)⁺: 497.0649; found: 497.0643.

1.2.5. 4-Nitro-phenyl 2,3,4,6-tetra-O-acetyl-1-thio- β -D-glucopyranoside (4e)⁶

Yield 79%; yellow solid; mp: 180–182 °C; $[\alpha]_D^{22}$ –54.4 (*c* 3.0, CHCl₃); FTIR (KBr): 3448 (w), 1754 (s), 1737 (s), 1515 (m), 1373 (m), 1341 (m), 1242 (s), 1221 (s), 1093 (m), 1044 (s), 913 (w), 851 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 8.17 (d, *J* = 8.7 Hz, 2H), 7.60 (d, *J* = 8.7 Hz, 2H), 5.28 (t, *J* = 9.3 Hz, 1H), 5.07 (q, *J* = 10.4 Hz, 2H), 4.88 (d, *J* = 10.0 Hz, 1H), 4.26 (dd, *J* = 12.4, 5.3 Hz, 1H), 4.20 (dd, *J* = 12.3, 2.0 Hz, 1H), 3.86–3.81 (m, 1H), 2.11–2.01 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.4, 170.0, 169.3, 169.2, 147.0, 141.7, 131.0, 123.8, 84.3, 76.1, 73.6, 69.5, 68.0, 62.0, 20.7, 20.6, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₃NO₁₁SNa (M+Na)⁺: 508.0890, found: 508.0896.

1.2.6. 2-Pyridyl 2,3,4,6-tetra-O-acetyl-1-thio- β -D-glucopyranoside (4f)²²

Yield 65%; yellow solid; mp: 120–121 °C; $[\alpha]_D^{25}$ –10.2 (*c* 0.4, CHCl₃); FTIR (KBr): 3468 (w), 1752 (s), 1375 (w), 1254 (m), 1225 (s), 1047 (m), 915 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 8.45 (m, 1H), 7.57–7.52 (m, 1H), 7.22 (d, *J* = 7.6 Hz, 1H), 7.09–7.06 (m, 1H), 5.84 (d, *J* = 10.4 Hz, 1H), 5.36 (t, *J* = 9.3 Hz, 1H), 5.25–5.14 (m, 2H), 4.27 (dd, *J* = 12.4, 4.4 Hz, 1H), 4.10 (dd, *J* = 12.4, 1.9 Hz, 1H), 3.91–3.87 (m, 1H), 2.04–2.02 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 170.1, 169.5, 169.4, 155.2, 149.6, 136.5, 123.3, 120.8, 81.6, 75.9, 74.1, 69.4, 68.2, 61.9, 20.7, 20.6, 20.6, 20.6; HRMS (ESI-TOF) *m/z*: calcd for C₁₉H₂₃NO₉SNa (M+Na)⁺: 464.0991; found: 464.0990.

1.2.7. Phenyl 2,3,4,6-tetra-O-acetyl-1-thio- β -D-galacto pyranoside (4h)^{11,23}

Yield 92%; white solid; mp: 61–63 °C; $[\alpha]_D^{26}$ +4.5 (*c* 1.9, CHCl₃); FTIR (neat): 1751 (s), 1369 (w), 1223 (s), 1082 (w), 1054 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.53–7.51 (m, 2H), 7.33–7.31 (m, 3H), 5.42 (d, *J* = 3.2 Hz, 1H), 5.25 (t, *J* = 10.0 Hz, 1H), 5.06 (dd, *J* = 9.9, 3.3 Hz, 1H), 4.72 (d, *J* = 10.0 Hz, 1H), 4.19 (dd, *J* = 11.4, 7.0 Hz, 1H), 4.12 (dd, *J* = 11.4, 6.2 Hz, 1H), 3.94 (t, *J* = 6.5 Hz, 1H), 2.12– 1.98 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 170.2, 170.0, 169.4, 132.5, 132.4, 128.9, 128.1, 86.6, 74.4, 72.0, 67.2, 67.2, 61.6, 20.8, 20.6, 20.6, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₄O₉SNa (M+Na)⁺: 463.1039; found: 463.1036.

1.2.8. Phenyl 2,3,4,6-tetra-O-benzyl-1-thio- β -D-glucopyranoside $(4i)^{24}$

Yield 93%; pale yellow solid; mp: 88–90 °C; $[\alpha]_D^{27}$ –43.2 (*c* 1.9, CHCl₃); FTIR (KBr): 3029 (w), 2904 (w), 1139 (m), 1062 (s), 755 (w), 733 (m), 695 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.59–7.57 (m, 2H), 7.40–7.18 (m, 23H), 4.91–4.81 (m, 4H), 4.73 (d, *J* = 10.2 Hz, 1H), 4.67 (d, *J* = 9.8 Hz, 1H), 4.62–4.52 (m, 3H), 3.80–3.62 (m, 4H), 3.53–3.49 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 138.4, 138.3, 138.0, 133.8, 131.9, 128.9, 128.4, 128.4, 128.3, 128.2, 127.9, 127.8, 127.8, 127.7, 127.6, 127.5, 127.4, 87.4, 86.7, 80.8, 79.1, 77.8, 75.8, 75.4, 75.0, 73.4, 69.0; HRMS (ESI-TOF) *m/z*: calcd for C₄₀H₄₀O₅SNa (M+Na)⁺: 655.2494; found: 655.2496.

1.2.9. Methyl 2,3,4-tri-O-acetyl-6-deoxy-6-phenylthio- α -D-glucopyranoside $(4j)^{25}$

Yield 89%; gummy mass; $[\alpha]_D^{22}$ +162.3 (*c* 3.4, CHCl₃); FTIR (neat): 1750 (s), 1370 (w), 1245 (m), 1123 (s), 1044 (m), 742 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.36–7.18 (m, 5H), 5.44 (t, *J* = 9.7 Hz, 1H), 4.97 (t, *J* = 9.6 Hz, 1H), 4.92–4.87 (m, 2H),

4.00–3.95 (m, 1H), 3.39 (s, 3H), 3.11 (dd, J = 13.4, 6.7 Hz, 1H), 3.02 (dd, J = 13.7, 8.5 Hz, 1H), 2.06–2.00 (3s, 9H); ¹³C NMR (100 MHz, CDCl₃): δ 170.1, 170.0, 169.8, 136.1, 129.4, 129.0, 126.3, 96.4, 72.0, 70.9, 70.0, 68.2, 55.2, 35.8, 20.7, 20.6; HRMS (ESI-TOF) m/z: calcd for C₁₉H₂₄O₈SNa (M+Na)⁺: 435.1090; found: 435.1090.

1.2.10. Phenyl 2-acetamido-3,4,6-tri-O-acetyl-1,2-di-deoxy-1-thio- β -D-glucopyranoside (4k)²⁶

Yield 83%; white solid; mp: 195–197 °C; $[\alpha]_D^{22}$ –14.1 (*c* 4.3, CHCl₃); FTIR (KBr): 3304 (w), 1750 (s), 1662 (m), 1559 (w), 1543 (m), 1375 (m), 1243 (s), 1219 (s), 1087 (w), 1047 (m), 913 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.51–7.48 (m, 2H), 7.30–7.28 (m, 3H), 6.07 (d, *J* = 9.1 Hz, 1H), 5.25 (d, *J* = 9.8 Hz, 1H), 5.05 (d, *J* = 9.7 Hz, 1H), 4.89 (d, *J* = 10.4 Hz, 1H), 4.23–4.14 (m, 2H), 4.05 (q, *J* = 9.9 Hz, 1H), 3.78 (m, 1H), 2.07–1.98 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 171.0, 170.6, 170.2, 169.3, 132.5, 132.3, 128.9, 128.0, 86.5, 75.6, 74.0, 68.5, 62.4, 53.2, 23.2, 20.7, 20.7, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₅NO₈SNa (M+Na)⁺: 462.1199; found: 462.1199.

1.2.11. Phenyl 2,3,6-tri-O-acetyl-4-O-(2',3',4',6'-tetra-O-acetyl-β-D-galactopyranosyl)-l-deoxy-l-thio-β-D-glucopyranoside (4l)^{11,26}

Yield 90%; white solid; mp: 160-162 °C; $[\alpha]_D^{25}$ +35.4 (*c* 3.6, CHCl₃); FTIR (KBr): 1750 (s), 1372 (m), 1231 (s), 1049 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.48–7.29 (m, 5H), 5.34 (d, *J* = 3.1 Hz, 1H), 5.22 (t, *J* = 9.1 Hz, 1H), 5.10 (dd, *J* = 10.2, 8.0 Hz, 1H), 4.95 (dd, *J* = 10.4, 3.4 Hz, 1H), 4.90 (t, *J* = 9.6 Hz, 1H), 4.68 (d, *J*=10.1 Hz, 1H), 4.54–4.47 (m, 2H), 4.14–4.05 (m, 3H), 3.87 (t, *J* = 6.7 Hz, 1H), 3.75 (t, *J* = 9.5 Hz, 1H), 3.67–3.63 (m, 1H), 2.15–1.96 (7s, 21H); ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 170.2, 170.1, 170.0, 169.7, 169.5, 169.0, 133.0, 131.7, 128.8, 128.3, 101.0, 85.4, 76.1, 73.8, 70.9, 70.6, 70.2, 69.0, 66.6, 62.10, 60.8, 20.8, 20.7, 20.6, 20.4; HRMS (ESI-TOF) *m/z*: calcd for C₃₂H₄₀O₁₇SNa (M+Na)⁺: 751.1884; found: 751.1884.

1.2.12. Phenyl 2,3,4,6-tetra-O-acetyl-1-seleno- β -D-glucopyranoside (5a)¹⁴

Yield 86%; white solid; mp: 97–99 °C; $[\alpha]_D^{22}$ –54.5 (*c* 1.5, CHCl₃); FTIR (KBr): 1749 (s), 1372 (s), 1253 (w), 1227 (m), 1085 (m), 1044 (s), 910 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.61 (d, *J* = 7.48 Hz, 2H), 7.37–7.28 (m, 3H), 5.19 (t, *J* = 9.3 Hz, 1H), 5.06 (q, *J* = 9.6 Hz, 2H), 4.89 (d, *J* = 10.1 Hz, 1H), 4.20–3.67 (m, 3H), 2.07–1.99 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 170.2, 169.4, 169.3, 135.2, 129.0, 128.6, 126.9, 80.9, 76.8, 73.8, 70.7, 68.1, 62.1, 20.8, 20.7, 20.6, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₄O₉SeNa (M+Na)⁺: 511.0483; found: 511.0485.

1.2.13. Phenyl 2,3,4,6-tetra-O-acetyl-1-seleno-β-D-galactopyranoside (5b)¹⁴

Yield 86%; gummy mass; $[\alpha]_D^{26}$ –5.3 (*c* 2.1, CHCl₃); FTIR (neat): 1750 (s), 1369 (w), 1221 (s), 1081 (w), 1055 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.64–7.62 (m, 2H), 7.36–7.25 (m, 3H), 5.41 (d, *J* = 3.2 Hz, 1H), 5.27 (t, *J* = 10.0 Hz, 1H), 5.02 (dd, *J* = 9.9, 3.3 Hz, 1H), 4.91 (d, *J* = 10.1 Hz, 1H), 4.17 (dd, *J* = 11.3, 6.9 Hz, 1H), 4.10 (dd, *J* = 11.3, 6.2 Hz, 1H), 3.91 (t, *J* = 6.6 Hz, 1H), 2.09–1.97 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 170.1, 170.0, 169.5, 134.8, 128.9, 128.3, 127.7, 81.8, 75.4, 71.7, 68.0, 67.3, 61.6, 20.8, 20.6, 20.6; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₄O₉Se+Na (M+Na)⁺: 511.0483; found: 511.0485.

1.2.14. Phenyl 2,3,4,6-tetra-O-benzyl-1-seleno- β -D-glucopyranoside (5c)²⁴

Yield 82%; white solid; mp: 77–79 °C; $[\alpha]_D^{-26}$ –13.6 (*c* 0.9, CHCl₃); FTIR (KBr): 3448 (s), 1664 (s), 1135 (s), 1065 (s), 732 (m), 694 (m) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.68 (d, *J* = 7.1 Hz,

2H), 7.39–7.16 (m, 23H), 4.90–4.80 (m, 5H), 4.72 (d, J = 10.1 Hz, 1H), 4.62–4.51 (m, 3H), 3.79–3.63 (m, 4H), 3.54–3.49 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 138.3, 138.3, 138.0, 137.9, 134.4, 129.0, 128.5, 128.4, 128.4, 128.3, 128.2, 127.9, 127.8, 127.8, 127.8, 127.7, 127.6, 127.5, 86.8, 82.9, 81.3, 80.1, 77.7, 75.8, 75.2, 75.0, 73.4, 68.9; HRMS (ESI-TOF) *m/z*: calcd for C₄₀H₄₀O₅SeNa (M+Na)⁺: 703.1939; found: 703.1935.

1.2.15. Methyl 2,3,4-tri-O-acetyl-6-deoxy-6-phenylseleno- α -D-glucopyranoside (5d)

Yield 79%; gummy mass; $[\alpha]_D^{26}$ +143.3 (*c* 3.3, CHCl₃); FTIR (neat): 1755 (s), 1371 (m), 1247 (s), 1224 (s), 1067 (m), 1045 (s), 739 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.51–7.49 (m, 2H), 7.28–7.24 (m, 3H), 5.44 (t, *J* = 9.6 Hz, 1H), 4.97–4.87 (m, 3H), 4.05–4.00 (m, 1H), 3.42 (s, 3H), 3.06–2.97 (m, 2H), 2.07 (s, 3H), 2.00 (2s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 170.1, 170.0, 169.8, 132.6, 130.4, 129.1, 127.1, 96.5, 72.6, 71.0, 70.0, 68.8, 55.4, 29.2, 20.7, 20.7, 20.7; HRMS (ESI-TOF) *m/z*: calcd for C₁₉H₂₄O₈SeNa (M+Na)⁺: 483.0534; found: 483.0530.

1.2.16. Phenyl 2-acetamido-3,4,6-tri-O-acetyl-1,2-di-deoxy-1-seleno- β -p-glucopyranoside (5e)²⁷

Yield 79%; pale yellow solid; mp: 194–196 °C; $[\alpha]_D - 29.9$ (*c* 8.8, CHCl₃); FTIR (KBr): 3448 (bs), 1745 (s), 1661 (w), 1375 (w), 1243 (s), 1227 (s), 1046 (m), 738 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.62 (d, *J* = 6.7 Hz, 2H), 7.32–7.26 (m, 3H), 6.15–6.11 (m, 1H), 5.18 (t, *J* = 9.7 Hz, 1H), 5.07–5.02 (m, 2H), 4.21–4.11 (m, 3H), 3.72–3.68 (m, 1H), 2.05–1.97 (4s, 12H); ¹³C NMR (100 MHz, CDCl₃): δ 170.1, 170.6, 170.2, 169.3, 134.6, 128.9, 128.2, 128.0, 82.6, 73.6, 68.5, 62.4, 53.9, 23.2, 20.7, 20.7, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₂₀H₂₅NO₈SeNa (M+Na)⁺: 510.0643; found: 510.0645.

1.2.17. Phenyl 2,3,6-tri-O-acetyl-4-O-(2',3',4',6'-tetra-O-acetyl-βp-galactopyranosyl)-l-deoxy-l-seleno-β-p-glucopyranoside (5f)¹⁴

Yield 83%; white solid; mp: $77-79 \,^{\circ}$ C; $[\alpha]_{D}^{23} -29.6$ (*c* 5.5, CHCl₃); FTIR (KBr): 1751 (s), 1371 (m), 1231 (s), 1054 (m), 910 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.59 (d, *J* = 7.8 Hz, 2H), 7.36–7.27 (m, 3H), 5.34 (d, *J* = 1.9 Hz, 1H), 5.18 (t, *J* = 9.0 Hz, 1H), 5.09 (dd, *J* = 10.1, 8.2 Hz, 1H), 4.96–4.85 (m, 3H), 4.52 (d, *J* = 11.8 Hz, 1H), 4.46 (d, *J* = 7.9 Hz, 1H), 4.14–4.04 (m, 3H), 3.86 (t, *J* = 6.7 Hz, 1H), 3.73 (t, *J* = 9.5 Hz, 1H), 3.63–3.59 (m, 1H), 2.14–1.96 (6s, 21H); ¹³C NMR (100 MHz, CDCl₃): δ 170.3, 170.2, 170.1, 170.0, 169.7, 169.6, 169.0, 135.2, 128.9, 128.4, 126.9, 100.9, 80.7, 77.7, 76.0, 73.6, 71.1, 70.9, 70.6, 69.0, 66.5, 62.0, 60.7, 20.8, 20.8, 20.7, 20.6, 20.6, 20.5; HRMS (ESI-TOF) *m/z*: calcd for C₃₂H₄₀O₁₇SeNa (M+Na)⁺: 799.1328; found: 799.1326.

1.2.18. (2*R*,3*R*,4*S*,5*R*,6*S*)-2-(Acetoxymethyl)-6-{[(2*R*,3*R*,4*S*,5*R*,6*S*)-4,5-diacetoxy-2-(acetoxymethyl)-6-[(4-methyl-2-oxo-2*H*-chromen-7-yl)thio]tetrahydro-2*H*-pyran-3-yl]oxy}tetrahydro-2*H*-pyran-3,4,5-triyl triacetate (4m)⁶

Yield 86%; white solid; mp: 205–207 °C; $[\alpha]_D^{25}$ –4.8 (*c* 1.0, CHCl₃); FTIR (KBr): 3481 (br), 2959 (w), 1750 (s), 1604 (w), 1374 (m), 1230 (s), 1168 (w), 1041 (s), 955 (w) cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 7.50 (d, *J* = 8.6 Hz, 2H), 7.27–7.25 (m, 1H), 6.28 (s, 1H), 5.24 (t, *J* = 8.9 Hz, 1H), 5.16 (t, *J* = 9.3 Hz, 1H), 5.07 (t, *J* = 9.6 Hz, 1H), 5.01–4.92 (m, 2H), 4.81 (d, *J* = 10.1 Hz, 1H), 4.59 (d, *J* = 11.7 Hz, 1H), 4.52 (d, *J* = 8.0 Hz, 1H), 4.38 (dd, *J* = 12.5,

4.1 Hz, 1H), 4.13 (dd, J = 12.0, 5.6 Hz, 1H), 4.04 (dd, J = 12.5, 1.7 Hz, 1H), 3.80–3.66 (m, 3H), 2.42–1.99 (6s, 24H); ¹³C NMR (100 MHz, CDCl₃): δ 170.6, 170.5, 170.3, 169.7, 169.5, 169.4, 169.2, 160.3, 153.6, 151.9, 138.0, 126.4, 124.7, 119.2, 118.3, 115.2, 100.9, 84.7, 77.3, 76.3, 73.5, 73.0, 72.1, 71.6, 69.8, 67.8, 62.1, 61.6, 20.9, 20.8, 20.7, 20.6, 20.5, 18.6; HRMS (ESI-TOF) m/z: calcd for C₃₆H₄₂O₁₉SNa (M+Na)⁺: 833.1939; found: 833.1940.

Acknowledgements

CVR thanks the CSIR for a Shyama Prasad Mukherjee (SPM) Senior Research Fellowship, VG thanks the Department of Science and Technology (DST), New Delhi for a fellowship under the Women Scientists Programme, and SCN thanks the Department of Science and Technology (DST) for the award of SERB Distinguished Fellow and the JNCASR, Jakkur for the Hindustan Lever Professorship.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.carres.2014.07. 011.

References

- 1. Tai, C.-A.; Kulkarni, S. S.; Hung, S.-C. J. Org. Chem. 2003, 68, 8719–8722.
- Dinkelaar, J.; van den Bos, L. J.; Hogendorf, W. F. J.; Lodder, G.; Overkleeft, H. S.; Codée, J. D. C.; van der Marel, G. A. Chem. Eur. J. 2008, 14, 9400–9411.
- Tatai, J.; Osztrovszky, G.; Kajtár-Peredy, M.; Fügedi, P. Carbohydr. Res. 2008, 343, 596–606.
- 4. Zhu, X.; Schmidt, R. R. Angew. Chem., Int. Ed. 2009, 48, 1900–1934.
- 5. Witczak, Z.; Culhane, J. Appl. Microbiol. Biotechnol. 2005, 69, 237–244. and references cited there in.
- 6. Brachet, E.; Brion, J.-D.; Messaoudi, S.; Alami, M. Adv. Synth. Catal. 2013, 355, 477–490.
- 7. Agnihotri, G.; Tiwari, P.; Misra, A. K. Carbohydr. Res. 2005, 340, 1393–1396.
- 8. Chen, H.-M.; Withers, S. G. Carbohydr. Res. 2010, 345, 2596–2604.
- Das, S. K.; Roy, J.; Reddy, K. A.; Abbineni, C. Carbohydr. Res. 2003, 338, 2237– 2240.
- Ibatullin, F. M.; Shabalin, K. A.; Jänis, J. V.; Shavva, A. G. Tetrahedron Lett. 2003, 44, 7961–7964.
- 11. Weng, S.-S. Tetrahedron Lett. 2009, 50, 6414–6417.
- 12. Weng, S.-S.; Lin, Y.-D.; Chen, C.-T. Org. Lett. 2006, 8, 5633-5636.
- Mukherjee, C.; Tiwari, P.; Misra, A. K.; Sau, A. Tetrahedron Lett. 2006, 47, 441– 445.
- 14. Sau, A.; Misra, A. K. Synlett 2011, 1905–1911.
- 15. Ibatullin, F. M.; Selivanov, S. I.; Shavva, A. G. Synthesis 2001, 0419–0422.
- 16. Valerio, S.; Iadonisi, A.; Adinolfi, M.; Ravidà, A. J. Org. Chem. 2007, 72, 6097-
 - 6106.
 - 17. Ganesh, V.; Chandrasekaran, S. Synthesis 2009, 3267–3278.
 - 18. Kotha, S.; Khedkar, P. Chem. Rev. 2011, 112, 1650–1680.
 - 19. When we decrease the equivalents of rongalite (<3 equiv), we found a gradual decrease in the yield of the product.
 - 20. Barr, B. K.; Holewinski, R. J. Biochemistry 2002, 41, 4447-4452.
 - Chen, Y.; Zhou, J.; Wang, H.; Xia, Y.; Yang, Z. Y.; Xia, P. Chin. Chem. Lett. 2008, 19, 925–927.
 - 22. Mereyala, H. B.; Reddy, G. V. Tetrahedron 1991, 47, 6435-6448.
 - 23. Patil, P. R.; Kartha, K. P. R. Green Chem. 2009, 11, 953–956.
 - France, R. R.; Rees, N. V.; Wadhawan, J. D.; Fairbanks, A. J.; Compton, R. G. Org. Biomol. Chem. 2004, 2, 2188–2194.
 - 25. González, F. S.; Baer, H. H. Carbohydr. Res. 1990, 202, 33–47.
 - Guilbert, B.; Davis, N. J.; Pearce, M.; Aplin, R. T.; Flitsch, S. L. Tetrahedron: Asymmetry 1994, 5, 2163–2178.
 - 27. Grant, L.; Liu, Y.; Walsh, K. E.; Walter, D. S.; Gallagher, T. Org. Lett. 2002, 4, 4623–4625.