Sequential Aza-Claisen Rearrangement and Ring-Closing Metathesis as a Route to 1-Benzazepine Derivatives

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Abstract: A synthetic strategy based on sequential application of aza-Claisen rearrangement and ring-closing metathesis reaction as key steps has been developed for the synthesis of various 1-benza-zepine derivatives of pharmaceutical relevance.

Key words: aza-Claisen rearrangement, benzazepine, ring-closing metathesis

Recent studies have revealed that N-substituted 1-benzazepines can act as powerful vasopressin V2 receptor antagonists, and therefore have the potential for use in the treatment of heart diseases.¹ Some of the most promising derivatives of this class are OPC-31260² (1, Figure 1) and tolvaptan³ (2), currently undergoing clinical trials. In addition, several other 1-benzazepine derivatives display promising biological activities towards a number of important targets such as enzymes (e.g., compound 3), ion channels and other G-protein-coupled receptors (e.g., compound 4).⁴ Consequently, several synthetic strategies for 1-benzazepines have been developed.⁵ 1-Benzazepine derivatives are most commonly accessed through Beckmann⁶ or Schmidt⁷ ring expansion of derivatives of 1-tetralone or intramolecular Claisen-type condensation.⁸ Some recent methods involve metal-catalyzed oxidative cyclization of amino alcohols,9 intramolecular Heck-type arylation,¹⁰ metal-catalyzed intramolecular amination,¹¹ and ring-closing metathesis^{12,13} of appropriate dienes. However, a simple and general entry to this pharmaceutically important scaffold is desirable. We have recently reported¹⁴ the synthesis of various benz- and heteroannulated oxepin and oxocin derivatives involving tandem application of Claisen rearrangement and ring-closing diene/enyne metathesis as key steps. Herein, we report a synthetic strategy to N-substituted 1-benzazepine derivatives using similar application.

The thermal rearrangement of *N*-allylanilines require drastic conditions compared to corresponding aryl allyl ethers.¹⁵ However, such rearrangements usually proceed with significant rate enhancement under protic or Lewis acid catalysis.¹⁶ It has further been observed that the pres-

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Figure 1 Pharmaceutically important 1-benzazepine derivatives

ence of an additional *N*-alkyl group in the *N*-allylaniline moiety greatly enhances the rate of an aromatic aza-Claisen rearrangement.¹⁷ This led us to consider selective monorearrangement of a *N*,*N*-diallylaniline derivative as a possible mean for the direct preparation of a *C*,*N*-diallyl aniline derivative to be used as a diene precursor of a RCM reaction for the ultimate preparation of the corresponding 1-benzazepine derivative.

Accordingly, we first prepared N,N-diallylaniline 6a (Scheme 1) by straightforward allylation of aniline with allyl bromide for studying its selective monorearrangement. After considerable experimentation, it was found that the desired monorearrangement of 6a proceeds best in refluxing chlorobenzene in the presence of $BF_3 \cdot OEt_2$ (1.5 equiv) within a reasonably short period of time (5-6 h) to provide N,2-diallylaniline (7a, 64%) as the only product, no double rearrangement being observed. Attempted optimization of the reaction using other high-boiling solvents or by changing the nature/amount of the Lewis acid or prolonging the reaction time, either singly or in combination, had deleterious effects on the outcome of the reaction since varied amount (up to ca. 40%) of 2,6-diallylaniline formed along with some unidentified products at the expense of the desired monorearrangement product 7a. Similarly, the rearrangement of each of the diallylanilines



Scheme 1 Reagents and conditions: (i) allyl bromide, Et_3N , MeCN, reflux, 18 h; (ii) BF₃·OEt₂ (1.5 equiv), PhCl, reflux, 5–6 h; (iii) TsCl, Et_3N , CH_2Cl_2 , r.t., 24 h, or (CF₃CO)₂O, DMAP, Et_3N , CH_2Cl_2 , r.t., 24 h, or benzoyl chloride, Et_3N , CH_2Cl_2 , 12 h; (iv) 9 (5 mol%), CH_2Cl_2 (0.01 M), r.t., 2–3 h.

6b–e under the developed conditions led to the corresponding rearranged products **7b–e** in acceptable yields. In each of the rearrangement reactions, minor amount (ca. 10%) of the unreacted starting material was also recovered. The electronic nature of the substituents in the starting anilines **6b–e** was found to have little effect on the outcome of the reaction as would be consistent with the concerted nature of the reaction.

Having access to the required aniline derivatives 7a-e, we then focused on their RCM reaction. To this end, the corresponding *N*-tosyl derivatives **8a–d** and the *N*-benzoyl derivatives 8e-g were prepared under conventional conditions. The nitroaniline derivative 7e, however, did not undergo clean N-tosylation or N-benzoylation under some of the attempted conditions. Consequently, it was converted to the corresponding trifluoroacetamide derivative 8h. Ring-closing metathesis of the diene 8a proved to be extremely facile with Grubbs first-generation catalyst benzylidene bistricyclohexylphosphinoruthenium(IV) dichloride (9) at room temperature, and a high-yield conversion to the azepine derivative **10a** was realized. Similarly, the dienes **8b-h** all underwent smooth RCM to provide the corresponding tetrahydrobenzazepine derivatives 10b-h in very good to excellent yields.

We next considered incorporation of functionalities in the 5-position of the prepared benzazepine derivatives as relevant to compounds 1 and 2. Attempted benzylic oxidation of compound 10a under a range of conditions proved to be unsuccessful. However, the corresponding saturated derivative 11a underwent oxidation with pyridinium chlorochromate¹⁸ in refluxing benzene leading to the cor-



ethers.^{14a-d} The bias for an angular rearrangement against a linear rearrangement of allyl ethers accommodated in an appropriate coumarin, quinolone, or carbazole scaffold is well documented. However, it is interesting to note that such a bias in aza-Claisen rearrangement in these systems is also operative. The rearranged products **14**, **18**, and **22** were then converted to the corresponding *N*-tosyl derivatives **15**, **19**, and **23**, respectively. Ring-closing metathesis of each of these dienes separately under the developed conditions then led to the corresponding tetrahydroazepine derivatives **16**, **20**, and **24**, respectively, in very good yields.

In summary, we have developed a suitable protocol for the selective monorearrangement of a *N*,*N*-diallylaniline derivative for the preparation of a N-tethered diene as precursor of the corresponding 1-benzazepine derivative having three points of diversity relevant to drug design. The methodology is simple, general in nature, and it involves the use of easily available starting materials and reagents. The predetermined mode of ring closure and its tolerance to electron-withdrawing groups may make the process complementary to those existing in the literature,



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responding ketone derivative **12a** in moderate yield. Similarly, the keto derivatives **12b–d** were prepared (Scheme 2).¹⁹

While several 1-benzazepine derivatives have been pre-

pared and evaluated, little is known about heteroannulated

azepine derivatives. Various common ring-sized azacycle-

annulated coumarin,²⁰ 2-quinolone²¹ and carbazole²² de-

rivatives display interesting biological properties. We

thus considered extending the present study to 6-amino-

coumarin, 6-amino-1-methyl-2-quinolone, and 3-amino-

carbazole with a view to prepare corresponding azepine-

annulated constructs. Thus, N,N-diallyl-6-aminocoumarin

(13, Table 1) was prepared by allylation of 6-aminocoumarin. Pleasingly, compound 13 underwent smooth monorearrangement under the developed conditions to

provide the corresponding rearrangement product 14 as

the only regioisomer in good yield. Similarly, the N.N-di-

allylquinolone derivative 17 and the diallylamino-

carbazole derivative 21 also underwent selective

monorearrangement to the corresponding angularly ally-

lated compounds 18 and 22, respectively, in good yields.



^a Appropriate amine, allyl bromide, Et₃N, MeCN, reflux, 18 h.

^b BF₃·OEt₂ (1.5 equiv), ClPh, reflux, 2–6 h.

^c TsCl (1.5 equiv), Et₃N (2 equiv), CH₂Cl₂, r.t., 24 h.

^d Grubbs catalyst 9 (5 mol%), CH₂Cl₂ (0.01 M), r.t., 2–3 h.

in particular, for the preparation of substituted 1-benzazepines, which are otherwise difficult to access. It may also find application in the synthesis of other mediumsized azacycle-annulated aromatic compounds of interest. Work will be continued along these directions.

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References and Notes

- (1) Blakeney, J. S.; Reid, R. C.; Le, G. T.; Fairlie, D. P. *Chem. Rev.* **2007**, *107*, 2960.
- (2) (a) Ogawa, H.; Yamashita, H.; Kondo, K.; Yamamura, Y.; Miyamoto, H.; Kan, K.; Kitano, K.; Tanaka, M.; Nagaya, K.; Nakamura, S.; Mori, T.; Tominaga, M.; Yabuuchi, Y. *J. Med. Chem.* **1996**, *39*, 3547. (b) Mayanoff, B. E. Acc. *Chem. Res.* **2006**, *39*, 831.
- (3) (a) Miyazaki, T.; Fujiki, H.; Yamamura, Y.; Nakamura, S.; Mori, T. *Cardiovasc. Drug Rev.* 2007, 25, 1. (b) Torisawa, Y.; Furuta, T.; Nishi, T.; Aki, S.; Minamikawa, J. *Bioorg. Med. Chem. Lett.* 2007, 17, 6455. (c) Cordero-Vargas, A.; Quiclet-Sire, B.; Zard, S. Z. *Bioorg. Med. Chem.* 2006, 14, 6165.
- (4) (a) Caggiano, T. J. *Drugs Future* 2002, *27*, 248.
 (b) Mayanoff, B. E. *Drug Discovery and Development*, Vol. 1; Chorghade, M. S., Ed.; Wiley: New Jersey, 2006, 313.

- (5) (a) Proctor, G. R. Azepines, In Heterocyclic Compounds, Vol. 43; Wiley: New York, **1984**, 637. (b) Smalley, R. K. Azepines, In Comprehensive Heterocyclic Chemistry, Vol. N7; Katritzky, A. R.; Rees, C. W., Eds.; Pergamon Press: Oxford, **1984**, 491. (c) Yet, L. Chem. Rev. **2000**, 100, 2963.
- (6) Maruoka, K.; Miyazaki, T.; Ando, M.; Matsumura, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. J. Am. Chem. Soc. 1983, 105, 283.
- (7) Grunwald, G. L.; Dahanukar, V. H.; Ching, P.; Kriscione, K. R. J. Med. Chem. **1996**, *39*, 3539.
- (8) (a) Learmonth, D. A.; Proctor, G. R.; Scopes, D. I. C. J. Chem. Soc., Perkin Trans. 1 1997, 2569. (b) Ikemoto, T.; Ito, T.; Nishiguchi, A.; Miura, S.; Tomimatsu, K. Org. Process Res. Dev. 2005, 9, 168.
- (9) Fujita, K.; Yamamoto, K.; Yamaguchi, R. Org. Lett. 2002, 4, 2691.
- (10) (a) Gibson, S. E.; Middleton, R. J. J. Chem. Soc., Chem. Commun. 1995, 1743. (b) Cropper, E. L.; White, A. J. P.; Ford, A.; Hii, K. K. J. Org. Chem. 2006, 71, 1732.
- (11) (a) Omar-Amrani, R.; Thomas, A.; Brenner, E.; Schneider, R.; Fort, Y. Org. Lett. 2003, 5, 2311. (b) Margolis, B. J.; Swidorski, J. J.; Rogers, B. N. J. Org. Chem. 2003, 68, 644.
 (c) Qadir, M.; Priestley, R. E.; Rising, T. W. D. F.; Gelbrich, T.; Coles, S. J.; Hursthouse, M. B.; Sheldrake, P. W.; Whittall, N.; Hii, K. K. Tetrahedron Lett. 2003, 44, 3675.
- (12) (a) Kaim, L. E.; Grimaud, L.; Oble, J. J. Org. Chem. 2007, 72, 5835. (b) Qadir, M.; Cobb, J.; Sheldrake, P. W.; Whittall, N.; White, A. J. P.; Hii (Mimi), K. K.; Horton, P. N.; Hursthouse, M. B. J. Org. Chem. 2005, 70, 1545.
 (c) Dolman, S. J.; Schrock, R. R.; Hoveyada, A. H. Org. Lett. 2003, 5, 4899. (d) Kotha, S.; Sha, V. R. Eur. J. Org. Chem. 2008, 1054.

Synlett 2008, No. 19, 3011-3015 © Thieme Stuttgart · New York

- (13) For a recent review on medium-ring heterocycle formation by RCM, see: Chattopadhyay, S. K.; Karmakar, S.; Biswas, T.; Majumdar, K. C.; Rahaman, H.; Roy, B. *Tetrahedron* **2007**, *63*, 3919.
- (14) (a) Chattopadhyay, S. K.; Maity, S.; Panja, S. *Tetrahedron Lett.* 2002, *43*, 7781. (b) Chattopadhyay, S. K.; Biswas, T.; Neogi, K. *Chem. Lett.* 2006, *35*, 376. (c) Chattopadhyay, S. K.; Dey, R.; Biswas, S. *Synthesis* 2005, 403.
 (d) Chattopadhyay, S. K.; Roy, S. P.; Ghosh, D.; Biswas, G. *Tetrahedron Lett.* 2006, *47*, 6895. (e) Chattopadhyay, S. K.; Biswas, T.; Maity, S. *Synlett* 2006, 2211.
- (15) For some recent reviews on Claisen rearrangement, see:
 (a) Nubbemeyer, U. *Synthesis* 2003, 961. (b) Castro, A. M. M. *Chem. Rev.* 2004, *104*, 2939.
- (16) Nubbemeyer, U. Top. Curr. Chem. 2005, 244, 149.
- (17) Krowichi, K.; Paillous, N.; Riviere, M.; Lattes, A. *J. Heterocycl. Chem.* **1976**, *13*, 555.
- (18) Rathore, R.; Saxena, N.; Chandrasekaran, S. Synth. Commun. 1986, 16, 1493.
- (19) Representative Procedure for the Sequence of Reactions in Scheme 1: N,N-Diallyl-4-methylaniline (6b) Allyl bromide (2.7 mL, 31.4 mmol) was added dropwise to a solution of $\mathbf{5b}$ (1.7 g, 15.7 mmol) and Et_3N (4.4 mL, 31.4 mmol) in dry MeCN (25 mL), and the mixture was heated to reflux for 18 h. It was then allowed to come to r.t., concentrated under reduced pressure, and the residual mass was extracted with EtOAc (50 mL). The extract was washed successively with H₂O (25 mL), brine (25 mL), and then dried (Na₂SO₄). It was filtered, concentrated under reduced pressure, and the residue was purified by chromatography over SiO₂ using PE as eluent to afford **6b** as a pale yellow viscous liquid (2.09 g, 71%). IR(neat): 1642, 1619, 1521, 1235, 1182 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.01 (d, 2 H, J = 8.2 Hz), 6.62 (d, 2 H, J = 7.8 Hz), 5.89–5.80 (m, 2 H), 5.20–5.12 (m, 4 H), 3.88 (d, 4 H, J = 4.8 Hz), 2.23 (s, 3 H). Anal. Calcd for C₁₃H₁₇N: C, 83.37; H, 9.15; N, 7.48. Found: C, 83.34; H, 9.28; N, 7.39.

N,2-Diallyl-4-methylaniline (7b)

Boron trifluoride etherate (1.8 mL, 15 mmol) was slowly added to a solution of 6b (1.9 g, 10 mmol) in PhCl (15 mL) under nitrogen, and the mixture was heated to reflux for 5 h. It was then allowed to come to r.t., quenched with sat. aq NaHCO₃ solution (20 mL), and the aqueous layer was extracted with EtOAc (2×25 mL). The combined organic mixture was washed successively with H₂O (25 mL), brine (25 mL), and then dried (Na2SO4). It was filtered, concentrated under reduced pressure, and the residue was purified by chromatography over SiO₂ using PE as eluent to give starting **6b** (0.17g, 9%) followed by the product **7b** (1.31g, 69%) as a pale yellow viscous liquid. IR(neat): 3442, 3387, 1636, 1618, 1515, 1313 cm⁻¹. ¹H NMR (300 MHz, $CDCl_3$): $\delta = 6.94$ (d, 1 H, J = 8.1 Hz), 6.87 (s, 1 H), 6.55 (d, 1 H, J = 8.1 Hz), 5.98–5.92 (m, 2 H), 5.28–5.06 (m, 5 H), 3.76 (dt, 2 H, J = 5.4, 1.5 Hz), 3.28 (d, 2 H, J = 6.2 Hz), 2.24 (s, 3 H). Anal. Calcd for C₁₃H₁₇N: C, 83.37; H, 9.15; N, 7.48. Found: C, 83.40; H, 9.26; N, 7.43.

N-Allyl-*N*-(2-allyl-4-methylphenyl)-4-methylbenzenesulfonamide (8b)

p-Toluenesulfonyl chloride (1.71 g, 9 mmol) was added to a solution of **7b** (1.1g, 5.9 mmol) and Et₃N (1.7 mL, 12 mmol) in dry CH₂Cl₂ (20 mL), and the reaction mixture was stirred at r.t. for 12 h. It was then diluted with CH₂Cl₂ (20 mL), and the solution was washed successively with HCl (1 N, 2 × 25 mL), H₂O (25 mL), brine (25 mL), and then dried (Na₂SO₄). It was filtered, concentrated under reduced pressure, and the residue was purified by chromatography over SiO₂ using EtOAc–PE (1:19) as eluent to give the product as a colorless

viscous liquid (1.7 g, 85%). IR (CHCl₃): 1638, 1598, 1497, 1349, 1219, 1164, 1062 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.58$ (d, 2 H, J = 8.2 Hz), 7.26 (d, 2 H, J = 8.1 Hz), 7.09 (s, 1 H), 6.85 (d, 1 H, J = 7.9 Hz), 6.46 (d, 1 H, J = 8.0 Hz), 5.97–5.84 (m, 1 H), 5.80–5.68 (m, 1 H), 5.14–5.08 (m, 2 H), 5.01–4.93 (m, 2 H), 4.30 (dd, 1 H, J = 14.0, 5.6 Hz), 3.85 (dd, 1 H, J = 14.0, 7.6 Hz), 3.55 (dd, 1 H, J = 15.5, 6.5 Hz), 3.45 (dd, 1 H, J = 15.4, 6.5 Hz), 2.44 (s, 3 H), 2.30 (s, 3 H). Anal. Calcd for C₂₀H₂₃NO₂S: C, 70.35; H, 6.79; N, 4.10. Found: C, 70.48; H, 6.88; N, 4.29.

7-Methyl-1-tosyl-2,5-dihydro-1*H*-benzo[*b*]azepine (10b) Catalyst 9 (14 mg, 5 mol%) was added to a solution of 8b (0.11g, 0.32 mmol) in dry, degassed CH₂Cl₂ (30 mL) under nitrogen, and the reaction mixture was stirred at r.t. for 2 h. It was then concentrated under reduced pressure and the residue was chromatographed over SiO₂ using EtOAc-PE (1:13) as eluent to give the product 10b (83 mg, 83%) as a colorless solid; mp 114 °C. IR (CHCl₃): 1598, 1496, 1343, 1157 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.64 (d, 2 H, J = 8.2 Hz), 7.24 (d, 2 H, J = 8.5 Hz), 7.16 (d, 1 H, J = 8.0Hz), 7.00 (d, 1 H, J = 7.9 Hz), 6.87 (s, 1 H), 5.66–5.60 (m, 1 H), 5.45–5.41 (m, 1 H), 4.35 (br s, 2 H), 2.92 (br s, 2 H), 2.42 (s, 3 H), 2.29 (s, 3 H). ¹³C NMR (75 MHz, CDCl₃): $\delta = 143.1, 140.8, 138.7, 138.3, 136.0, 129.9, 129.7, 129.4,$ 128.0, 127.0, 125.8, 125.3, 49.1, 32.2, 21.5, 21.0. Anal. Calcd for C₁₈H₁₉NO₂S: C, 68.98; H, 6.11; N, 4.47. Found: C, 69.13; H, 6.18; N, 4.58. MS (TOFMS ES⁺): *m/z* = 336 [M⁺ + Nal.

Selected Data

Compound **10c**: Mp 128 °C. IR (KBr): 1602, 1500, 1341, 1159 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.63 (d, 2 H, *J* = 8.2 Hz), 7.24 (d, 2 H, *J* = 8.1 Hz), 7.19 (d, 1 H, *J* = 8.8 Hz), 6.71 (dd, 1 H, *J* = 8.6, 2.9 Hz), 6.58 (d, 1 H, *J* = 2.7 Hz), 5.65–5.59 (m, 1 H), 5.45–5.41 (m, 1 H), 4.35 (br s, 2 H), 3.78 (s, 3 H), 2.88 (br s, 2 H), 2.42 (s, 3 H). ¹³C NMR (75 MHz, CDCl₃): δ = 159.2, 143.1, 142.5, 138.6, 131.2, 131.1, 129.4, 127.0, 125.9, 124.9, 114.6, 112.1, 55.3, 49.2, 32.4, 21.5. Anal. Calcd for C₁₈H₁₉NO₃S: C, 65.63; H, 5.81; N, 4.25. Found: C, 65.80; H, 5.98; N, 4.43. MS (TOFMS ES⁺): m/z = 352 [M⁺ + Na].

Compound **12b**: Mp 135 °C. IR (KBr): 1715, 1596, 1491, 1352, 1168 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.91 (d, 2 H, *J* = 8.3 Hz), 7.41 (d, 1 H, *J* = 8.0 Hz), 7.32 (d, 2 H, *J* = 8.1 Hz), 7.18 (d, 1 H, *J* = 8.1 Hz), 7.03 (s, 1 H), 2.48–2.44 (m, 2 H), 2.38 (s, 3 H), 2.23 (s, 3 H), 2.10–2.04 (m, 3 H), 1.79–1.77 (m, 1 H). ¹³C NMR (75 MHz, CDCl₃): δ = 172.7, 144.8, 139.4, 136.6, 135.8, 133.4, 129.7, 129.3, 129.1, 128.8, 127.9, 34.4, 29.1, 27.3, 21.7, 21.1. Anal. Calcd for C₁₈H₁₉NO₃S: C, 65.63; H, 5.81; N, 4.25. Found: C, 65.78; H, 6.04; N, 4.48. MS (TOFMS ES⁺): *m/z* = 352 [M⁺ + Na].

Compound **16**: IR (CHCl₃): 1735, 1597, 1342, 1160, 1109 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.91 (d, 1 H, *J* = 10.0 Hz), 7.68 (d, 2 H, *J* = 8.2 Hz), 7.35 (d, 1 H, *J* = 8.8 Hz), 7.30 (d, 2 H, *J* = 8.1 Hz), 7.17 (d, 1 H, *J* = 8.7 Hz), 6.43 (d, 1 H, *J* = 9.9 Hz), 5.79–5.71 (m, 1 H), 5.54–5.50 (m, 1 H), 4.37 (br s, 2 H), 3.34 (d, 2 H, *J* = 4.0 Hz), 2.45 (s, 3 H). ¹³C NMR (75 MHz, CDCl₃): δ = 159.7, 153.7, 143.6, 140.3, 139.6, 138.0, 135.2, 132.6, 129.7, 126.9, 126.7, 123.5, 116.5, 116.3, 115.5, 48.7, 24.6, 21.4. Anal. Calcd for C₂₀H₁₇NO₄S: C, 65.38; H, 4.66; N, 3.81. Found: C, 65.66; H, 4.83; N, 3.96. MS (TOFMS ES⁺): *m/z* (%) = 390(100) [M + Na], 368(41) [M + H].

Compound **20**: mp 224 °C IR (KBr): 1654, 1578, 1455, 1333, 1158, 1123 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 7.90 (d, 1 H, *J* = 10.0 Hz), 7.68 (d, 2 H, *J* = 8.2 Hz), 7.47 (d, 1 H, *J* = 9.0 Hz), 7.30–7.22 (m, 3 H), 6.72 (d, 1 H,

- $J = 10.0 \text{ Hz}, 5.80-5.73 \text{ (m, 1 H)}, 5.52-5.48 \text{ (m, 1 H)}, 4.13 \text{ (br s, 2 H)}, 3.71 \text{ (s, 3 H)}, 3.35 \text{ (d, 2 H, } J = 4.3 \text{ Hz}), 2.44 \text{ (s, 3 H)}. {}^{13}\text{C} \text{ NMR} (75 \text{ MHz, CDCl}_3): \delta = 161.6, 143.5, 140.2, 140.1, 138.4, 134.5, 133.4, 131.6, 129.7, 126.9, 124.1, 121.7, 118.1, 113.2, 49.0, 29.8, 24.6, 21.5. Anal. Calcd for C_{21}H_{20}N_2O_3S: C, 66.29; H, 5.30; N, 7.36. Found: C, 66.36; H, 5.41; N, 7.24. MS (TOFMS ES⁺): <math>m/z = 403 \text{ [M + Na]}.$
- (20) (a) O'Kennedy, R.; Thornes, R. D. Coumarins: Biology, Applications and Mode of Action; Wiley and Sons: Chichester, **1997**. (b) Fylaktakidou, K. C.; Hadjipavlou-Litina, D. J.; Litinas, K. E.; Nicolaides, D. N. Curr. Pharm. Des. **2004**, 10, 3813.
- (21) (a) Thorsett, E. D.; Latimer, L. H. *Curr. Opin. Chem. Biol.* 2000, *4*, 377. (b) Chevalier, J.; Atifi, S.; Eyraud, A.; Mahamoud, A.; Barbe, J.; Pages, J.-M. *J. Med. Chem.* 2001, *44*, 4023.
- (22) (a) Knölker, H.-J.; Reddy, K. R. Chem. Rev. 2002, 102, 4303. (b) Gallagher, P. T. In Science of Synthesis (Houben-Weyl), Vol. 10; Thomas, E. J., Ed.; Thieme: Stuttgart, 2001, 693. (c) Knölker, H.-J. Top. Curr. Chem. 2005, 244, 115.