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Enantioselective Total Synthesis of the Potent Antitumor Agent (–)-Mucocin Using a Temporary Silicon-Tethered Ring-Closing Metathesis Cross-Coupling Reaction

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The potent antitumor agent mucocin (1) was isolated from the leaves of *Rollinia mucosa* (jacq.) Baill. (Annonaceae) by McLaughlin and co-workers in 1995.^{1–3} This agent has exquisite selectivity for the inhibition of A-549 (lung cancer) and PACA-2 (pancreatic cancer) solid tumor cell lines with potency 10,000 times that of adriamycin (doxorubicin). Annonaceous acetogenins selectively inhibit cancerous cells through the blockage of the mitochondrial complex I (NADH-ubiquinone oxidoreductase), and the inhibition of the plasma membrane NADH oxidase, which depletes ATP and induces apoptosis (programmed cell death) in malignant cells.⁴

In a program directed toward the construction of nonadjacent tetrahydrofuran containing acetogenins, we have developed a new approach to the construction of C_2 -symmetrical 1,4-diols, using a temporary silicon-tethered (TST) ring-closing metathesis (RCM) homo-coupling reaction.⁵ Herein, we now describe a novel and expeditious synthesis of mucocin (1), which utilizes the TST-RCM cross-coupling reaction (Scheme 1).^{6,7} This approach capitalizes on the localized C_2 -symmetry and thereby permits the construction of 2 and 3 from a common synthetic intermediate, the known homoallylic epoxide 5.8 We further envisioned that the C4-C5 bond could be formed by enantioselective addition of the alkyne 3 to the aldehyde 4, thereby providing a new strategic disconnection for this class of biologically important molecules.9 The key feature of this approach is the utilization of a triply convergent strategy, that can be adapted to facilitate the synthesis of related annonaceous acetogenins, resulting in one of the most expeditious syntheses of a complex acetogenin developed to date.

The synthesis of the 3-hydroxy-2,6-disubstituted tetrahydropyran 2 was accomplished using the novel six-step strategy outlined in Scheme 2. Mitsunobu inversion of the allylic alcohol 5 using *p*-methoxyphenol afforded the requisite aryl ether.¹⁰ Regiospecific ring opening of the epoxide with the homoenolate equivalent¹¹ derived from tert-butyldimethylsilyl protected divinyl carbinol, followed by an *in situ* protection of the resultant secondary alcohol, afforded the differentially protected triene 7 in 96% overall yield. Chemoselective Sharpless asymmetric dihydroxylation of the triene 7 using AD-mix- β furnished the hydroxy ketone 8 in 70% yield $(ds \ge 99:1 \text{ by HPLC})$, after recycling the recovered triene 7 $(2 \times)$.¹² The alkyl side chain was then introduced via the conjugate addition of the cuprate derived from octylmagnesium bromide and copper cyanide to furnish the ketone 9 and thereby set the stage for the reductive etherification. Treatment of 9 with bismuth tribromide and tert-butyldimethylsilane in acetonitrile, followed by in situ protection of the secondary alcohol, furnished the tert-butyldimethylsilyl ether **10** in 93% yield ($ds \ge 19:1$ by NMR).¹³ Finally, the *p*-methoxyphenyl ether was oxidatively cleaved with ceric ammonium nitrate (CAN) to complete the construction of 2.10

The construction of the tetrahydrofuran 3 was also initiated from the homoallylic epoxide 5, as outlined in Scheme 3. Mitsunobu inversion of 5 followed by regiospecific ring opening of the epoxide

Scheme 1





^{*a*}(a)*p*-MeOC₆H₄OH, DIAD, PPh₃, THF, 0°C, 80%; (b) (CH₂=CH)₂CHOTBS, ^sBuLi, THF, -78 °C, then TBSOTf, 2,6-lutidine, -78 to 0 °C, 96%; (c) AD-mix- β , 'BuOH/H₂O, MeSO₂NH₂, 0 °C (3×), 70%; (d) "octylMgBr, CuCN, THF, -78 °C, 65%; (e) BiBr₃, 'BuMe₂SiH, MeCN, 0 °C, then 2,6-lutidine, TBSOTf, 0 °C, 93%; (f) (NH₄)₂Ce(NO₃)₆, MeCN/H₂O, -5 °C, 91%.



^{*a*} (a)*p*-MeOC₆H₄OH, DIAD, PPh₃, THF, 0 °C, 80%; (b) CH₂=CHCH₂MgBr, CuCN, Et₂O, -78 °C, 90%; (c) Co(modp)₂, O₂, 'BuOOH, 'PrOH, 60 °C, 83%; (d) Tf₂O, Et₃N, CH₂Cl₂, -78 °C, 86%; (e) TMSC≡C(CH₂)₄MgBr, CuI, THF, -20 to -10 °C; then MeOH, TBAF, -20 °C to room temperature, 73%.

(*cf.* Scheme 2) with the cuprate derived from allylmagnesium bromide and catalytic copper cyanide afforded the secondary alcohol, which was subjected to a cobalt(II) catalyzed oxidative cyclization to afford the *trans*-2,5-tetrahydrofuran **11** in 75% overall yield (*ds* \geq 19:1).^{2d,14} Conversion of the primary alcohol **11** to triflate, followed by cuprate displacement and *in situ* deprotection of trimethylsilyl group, furnished the B-ring fragment **3**.

The synthesis of butenolide fragment 4 commenced with the regioselective ring opening of commercially available (*S*)-propylene oxide **6** (Scheme 4). Treatment of **6** with the carbanion derived from the alkyne **12** afforded the secondary alcohol, which was

converted to the selenocarbonate **13** using phosgene and phenylselenol.¹⁵ The selenocarbonate **13** was subjected to standard free radical conditions, to afford the γ -butyrolactone in 80% yield. Metal-catalyzed isomerization of the *exo*-cyclic olefin and subsequent hydrolysis of the diethyl acetal furnished the requisite aldehyde **4** in good overall yield.

Scheme 4^a



^{*a*} (a) *S*-Propylene oxide **6**, "BuLi, HMPA, THF, -30 °C; (b) COCl₂, Et₃N, C₆H₆, 0 °C to room temperature, then PhSeH, pyridine, THF/C₆H₆, 0 °C to room temperature, 60% overall yield from **12**; (c) "Bu₃SnH, AIBN, C₆H₆, Δ, 80%; (d) RhH(CO)(PPh₃)₃, C₆H₆, 85 °C, 84%; (e) HCOOH, pentane, 0 °C, 90%.

Scheme 5^a



^{*a*} (a) **3**, Et₂Zn, PhMe, Δ, then (*R*)-BINOL, Ti(O'Pr)₄, THF, **4**, 0 °C, 81%; (b) TIPSOTf, pyridine, DMAP, THF, 0 °C, 96%; (c) (NH₄)₂Ce(NO₃)₆, MeCN/H₂O, -10 °C, 91%; (d) **2**, 'Pr₂SiCl₂ (xs), CH₂Cl₂, imidazole, 0 °C to room temperature, then **14**, imidazole, 0 °C to room temperature, 74%; (e) Grubbs' catalyst (1.8 equiv), 1,2-DCE, Δ, 83%; (f) HF/MeCN, CH₂Cl₂, room temperature, 91%; (g) TsNHNH₂, NaOAc, 1,2-DME/H₂O, Δ, 95%.

Scheme 5 outlines the manner in which the three fragments were assembled to complete the synthesis of mucocin (1). The enantioselective addition of the alkynyl zinc reagent derived from 3 to the aldehyde 4 furnished the propargylic alcohol in 81% yield with excellent selectivity (ds = 20:1 by HPLC).^{9,16} Protection of the alcohol as the triisopropylsilyl ether followed by deprotection of the *p*-methoxyphenyl ether afforded the allylic alcohol 14^{10} and thereby set the stage for the TST-RCM cross-coupling reaction. The construction of the mixed bis-alkoxy silane was achieved from the allylic alcohol 2 through the treatment with excess diisopropyldichlorosilane to afford the mono-alkoxychlorosilane, followed by the removal of the excess silvlating agent and addition of the second allylic alcohol 14. Ring-closing metathesis of the silicon-tethered diene using stoichiometric Grubbs' catalyst furnished 15 in 83% yield and completed the construction of the carbon skeleton of mucocin (1) (Scheme 5).¹⁷ The synthesis was concluded with the fluoride-mediated deprotection of 15, followed by chemoselective reduction with diimide.¹⁸ The spectroscopic data and optical rotation of synthetic mucocin (1) were identical in all respects to the values reported for the natural substance $[{}^{1}H/{}^{13}C$ NMR, IR, $[\alpha]^{26}_{D}$ -16.0 $(c = 0.25, CH_2Cl_2)].$

In conclusion, we have accomplished an enantioselective total synthesis of the annonaceous acetogenin (-)-mucocin (1) using a

triply convergent 12-step sequence (longest linear sequence) in 13.6% overall yield. This approach represents the first application of the temporary silicon-tethered (*TST*) ring-closing metathesis (*RCM*) cross-coupling reaction and the enantioselective alkyne/ aldehyde addition in the synthesis of a complex annonaceous acetogenin. Finally, the synthesis highlights the utility of the bismuth tribromide-mediated reductive etherification for the construction of 3-hydroxy-2,6-disubstituted tetrahydropyrans.

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Note Added after ASAP. In the version posted 11/5/03, in Scheme 2 the absolute configuration for the secondary *tert*-butyldimethylsilyl ether in **7**, **8**, and **9** was incorrect. The version posted 11/11/03 and the print version are correct.

Supporting Information Available: Spectral data and detailed experimental procedures for all of the synthetic intermediates (PDF). This material is available free of charge via the Internet at http:// pubs.acs.org.

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