Tetrahedron: Asymmetry 23 (2012) 965-971

Contents lists available at SciVerse ScienceDirect

Tetrahedron: Asymmetry

journal homepage: www.elsevier.com/locate/tetasy

Catalytic asymmetric nitroaldol (Henry) reactions with copper(II)/cyclopropane-based bisoxazoline complexes

Jianyou Mao, Xin Nie, Min Wang, Qian Wang, Bing Zheng, Qinghua Bian*, Jiangchun Zhong*

Department of Applied Chemistry, China Agricultural University, 2 West Yuanmingyuan Road, Beijing 100193, PR China

ARTICLE INFO

Article history: Received 7 May 2012 Accepted 15 June 2012

ABSTRACT

A series of cyclopropane-based bisoxazolines were synthesized from (*R*)- and (*S*)-amino alcohols, and applied to copper-catalyzed enantioselective nitroaldol reactions between nitromethane and various aldehydes. The reactions generated β -hydroxy nitroalkanes in high yields (97%) and with good enantioselectivities (up to 87% ee). The effects of the oxazoline stereocenters constructed in the Henry reactions were also studied.

© 2012 Elsevier Ltd. All rights reserved.

Tetrahedron

1. Introduction

Chiral bisoxazolines have emerged as one of the most attractive ligands for asymmetric catalysis,¹ because they offer an advantageous combination of structural diversity, convenient synthesis, and excellent enantioselectivity for a wide range of reactions. Although a large variety of chiral bisoxazolines have been prepared from malonate,² tartrate,³ ferrocene or cyclohexane,⁴ biphenyl or binaphthyl,⁵ pyridine⁶, and other derivatives, formidable challenges remain in terms of the design and evaluation of novel bisoxazoline ligands. Such challenges increase the requirements for chiral catalysis. Cyclopropane-based ligands can exhibit high enantioselectivity for many catalytic reactions because of their highly variable platform and unusual bond angles; these ligands also offer a well-ordered chiral environment.⁷ Apart from our own efforts,8 however, very few efficient chiral cyclopropanebased ligands 1-5 have been reported,^{7,8} motivating us to incorporate cyclopropanes into bisoxazoline ligands to explore their potential utility in asymmetric catalytic reactions.



^{*} Corresponding authors. Tel.: +86 010 62731356; fax: +86 010 62820325 (Q.B.); tel.: +86 010 62731356; fax: +86 010 62820325 (J.Z.).



The asymmetric nitroaldol (Henry) reaction has become a powerful synthetic tool for the enantioselective construction of carbon–carbon bonds. The resulting β-hydroxy nitroalkanes provide efficient access to valuable bifunctional compounds, such as β-amino alcohols and α -hydroxy carboxylic acids.⁹ Given its significance, researchers have developed numerous asymmetric protocols, which involve the use of BINOL,¹⁰ bisoxazolidines,¹¹ cinchona alka-loids,¹² dinuclear zinc complexes,¹³ amino alcohols,^{8e,14} diamine,¹⁵ chiral Schiff bases, and tetrahydro-bisisoquinoline ligands.¹⁶ To the best of our knowledge, only a few reports on bisoxazoline catalysts for Henry reactions have been published.¹⁷ We previously reported the application of novel cyclopropane-based bisoxazolines 8a-8d in catalytic enantioselective Diels-Alder additions.¹⁸ To screen the effects of the configuration of the 4-substituent on the oxazoline and to develop novel chiral catalysts without privileged structures in asymmetric Henry reactions, we herein synthesized two new compounds 8e and 8f, which were the enantiomers of 8a and 8d, respectively, and describe in full detail the synthesis of chiral cyclopropane-based bisoxazolines and their applications as chiral ligands in copper-catalyzed asymmetric Henry reactions.



E-mail addresses: bianqinghua@cau.edu.cn (Q. Bian), dr_zhongjiangchun@ yahoo.com.cn (J. Zhong).

^{0957-4166/\$ -} see front matter © 2012 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.tetasy.2012.06.014

2. Results and discussion

Bisoxazoline ligands **8a–8f** were prepared by following the method reported by Evans (Scheme 1).¹⁹ Dicarbonyl chloride **6**



a: R=*i*-Pr (S), b: R=*t*-Bu (S), c: R=Bn (S), d: R=Ph (S), e: R=*i*-Pr (*R*), f: R=Ph (*R*)

Scheme 1. Synthesis of chiral cyclopropane-based bisoxazoline ligands 8a-8f.

Table 1

Ligand survey of the Henry reactions^a

	H CH ₃ NO ₂ Cu(OAc) ₂ ·H ₂ O (5mol%) bisoxazoline ligand (6 mol%)					
_	9a	EtOH, rt	-	10a		
Entry	Ligand	Time (h)	Yield ^b (%)	ee ^c (%)	Config ^d	
1	(S,S)- 8a	72	91	62	(<i>R</i>)	
2	(S,S)- 8b	98	66	6	_	
3	(S,S)- 8c	72	90	46	(<i>R</i>)	
4	(S,S)- 8d	72	58	4	_	
5	(R,R)- 8e	48	74	64	(S)	
6	(R,R)- 8f	48	72	4	-	

^a All reactions were performed at room temperature on a 0.5 mmol scale with 6 mol % of ligand and 5 mol % of Cu(OAc)₂·H₂O.

^b Values are isolated yields after chromatographic purification.

^c Enantiomeric excess was determined by HPLC using a Chiracel OD-H column. ^d The absolute configuration was assigned by comparing their specific rotations with data from the literature.^{17a}

Table 2

Optimization of the reaction conditions between nitromethane and benzaldehyde^a



Entry Lewis acid Catalyst (mol %) Temp (°C) Solvent Additive Yield^b (%) ee^c (%) 1 Cu(OAc)₂·H₂O 5 25 EtOH 74 64 Cu(OAc)₂·H₂O 25 i-PrOH 74 2 5 66 $Cu(OAc)_2 \cdot H_2O$ 5 70 3 25 MeOH _ 62 4 $Cu(OAc)_2$ 5 25 i-PrOH _ 75 72 5 $Cu(OTf)_2$ 5 25 i-PrOH 0 5 6 $Cu(OAc)_2$ 25 i-PrOH MS 72 69 5 Cu(OAc)₂ 25 *i*-PrOH 82 7 Et_3N 55 5 8 $Cu(OAc)_2$ 25 i-PrOH Pyridine 80 57 9 10 25 i-PrOH 72 68 Cu(OAc)₂ 10 3 25 i-PrOH 70 71 Cu(OAc)₂ _ 5 0 i-PrOH 70 11 $Cu(OAc)_2$ _ 85 _ 12 Cu(OAc)2·H2O 5 0 **EtOH** 50 82

^a All reactions were performed on a 0.5 mmol scale in aldehyde.

was first converted into new dihydroxy cyclopropane diamides **7a–7f** with (*R*)- and (*S*)-amino alcohols; subsequent reaction with *p*-toluenesulfonyl chloride and triethylamine in the presence of a catalytic amount of 4-(dimethylamino) pyridine produced **8a–8f**.

Ligands **8a–8f** were evaluated in copper-catalyzed Henry reactions between nitromethane and benzaldehyde **9a** (Table 1).^{17a} The results indicated that varying the substitution on the oxazoline ring had considerable effects on the enantioselectivity of the reactions. The bisoxazoline ligands with an *i*-Pr group **8a** presented higher ee values than did **8c** with a Bn group (entry 1 vs entry 3); the more hindered ligands **8b** and **8d** decreased the enantioselectivity greatly (entries 2 and 4). Furthermore, the absolute configuration of the Henry adducts could be controlled by the configuration of the 4-substituent on the oxazoline, **8a** with an (*S*)-*i*-Pr group furnished (*R*)-enantiomers, while **8e** with an (*R*)-*i*-Pr group generated (*S*)-enantiomers. Thus, bisoxazoline ligand **8e** was the ligand of choice, yielding a nitro alcohol product with 64% ee (entry 5).

The Henry reaction of benzaldehyde with nitromethane was tested to determine the optimal conditions for asymmetric catalytic reactions (Table 2). The ee values of product 10a remained almost unchanged in the three solvents examined (entries 1-3). Although replacing Cu(OAc)₂·H₂O with Cu(OAc)₂ slightly improved the enantioselectivity (entry 4 vs entry 2), the use of $Cu(OTf)_2$ yielded no product (entry 5). The addition of 4 Å molecular sieves or base promoters, such as pyridine and triethyl-amine, was not beneficial to the enantioselectivity; instead a loss of enantioselectivity was observed (entries 6-8). Further optimization of this process showed that the reaction may be performed with lower catalyst loading (3 mol %; entry 10). The temperature had a significant effect on the ee values; lowering the temperature of the reaction from 25 °C to 0 °C considerably increased the ee values (entry 4 vs entry 11). These experiments demonstrated that the best results can be obtained with 6 mol % of ligand 8e and 5 mol % of Cu(OAc)₂ in isopropanol at 0 °C (entry 11).

With the optimal ligand and conditions identified, we explored the scope of the reactions (Table 3). In general, good to high enantiomeric excesses (69–87% ee) were observed at 0 °C with various aldehydes. Strong electron withdrawing groups, such as NO₂ or F, significantly reduced enantioselectivity (entries 6 and 7). Except for **9f** and **9g**, other *ortho-*, *meta-*, and *para-*substituted benzaldehydes presented consistently good enantiomeric excesses (81– 87% ee, entries 2–5 and 8–14). In particular, a noteworthy result

Table 3

Henry reactions of nitromethane with various aldehydes^a



Entry	R	Product	Yield ^b (%)	ee ^c (%)
1	Ph	10a	70	85
2	3-ClC ₆ H ₄	10b	95	81
3	4-ClC ₆ H ₄	10c	96	82
4	3-BrC ₆ H ₄	10d	80	81
5	$4-BrC_6H_4$	10e	88	81
6	$2-FC_6H_4$	10f	84	73
7	4-NO ₂ C ₆ H ₄	10g	91	68
8	2-CH ₃ OC ₆ H ₄	10h	96	82
9	3-CH ₃ OC ₆ H ₄	10i	95	81
10	4-CH ₃ OC ₆ H ₄	10j	63	86
11	2-CH ₃ C ₆ H ₄	10k	93	83
12	3-CH ₃ C ₆ H ₄	101	65	86
13	4-CH ₃ C ₆ H ₄	10m	97	83
14	4-EtOC ₆ H ₄	10n	70	87
15	2,4-(MeO) ₂ C ₆ H ₃	10o	90	79
16	3,5-(MeO) ₂ C ₆ H ₃	10p	96	83
17	2,6-(MeO) ₂ C ₆ H ₃	10q	97	84
18	2-Naphthyl	10r	96	73
19	n-C ₇ H ₁₅	10s	60	69
20	PhCH=CH	10t	75	77
21	Cyclohexyl	10u	65	74

^a All reactions were performed at 0 $^{\circ}$ C on a 0.5 mmol scale with 6 mol % of ligand and 5 mol % of Cu(OAc)₂ in isopropanol.

^b Values are isolated yields after chromatographic purification.

^c Enantiomeric excess was determined by HPLC using Chiracel OD-H, Chiralpak AS-H or AD-H columns. The absolute configuration was assigned by comparing their specific rotations or the HPLC elution order with data from the literature.

(87% ee) was obtained with 4-ethoxybenzaldehyde. Even for the two substituted benzaldehydes, **90–9q**, good enantioselectivities were achieved (entries 15–17). The other aromatic aldehydes, such as 2-naphthaldehyde and cinnamaldehyde, also showed good ee values (entries 18 and 20). The substrate scope was not limited to aromatic aldehydes; aliphatic aldehydes **9s** and **9u** were effective substrates, generating nitroaldol products with 69% ee and 74% ee, respectively (entries 19 and 21).

3. Conclusion

A series of cyclopropane-based chiral bisoxazolines **8a–8f** have been conveniently prepared from (*R*)- and (*S*)-amino alcohols. Using the ligands without any additives, the Henry reactions gave the corresponding β -nitro alcohols in good to high enantioselectivities (up to 87% ee) from various aldehydes; even with low catalyst loading (3 mol %), good enantioselectivities were obtained. In particular, the two enantiomers of β -nitro alcohols could be obtained; bisoxazoline **8a** with an (*S*)-*i*-Pr group furnished the (*R*)-enantiomers; while **8e** with an (*R*)-*i*-Pr group generated the (*S*)-enantiomers. The application of these bisoxazolines in other asymmetric catalytic reactions is currently under investigation.

4. Experimental

4.1. General

All reactions were performed under an argon atmosphere. Solvents were dried according to standard procedures and distilled before use. All reagents were purchased commercially and used without further purification, unless stated otherwise. *cis*-3,3-Dimethylcyclopropane dicarbonyl chloride **6** was prepared following the methods in the literature.^{20a,b} Melting points were recorded using a Cole-Parmer Melt-Temp apparatus and are uncorrected.

4.2. General procedure for the preparation of dihydroxy cyclopropane diamides 7¹⁹

To a cold (0 °C) solution of amino alcohol (22.5 mmol) and Et₃N (10 mL, 72 mmol) in CHCl₃ (60 mL), a solution of dimethyl cyclopropane dicarbonyl chloride (1.8 g, 9 mmol) was slowly added in CHCl₃ (30 mL). The reaction mixture was allowed to warm to room temperature and was stirred for 24 h. The mixture was then washed with saturated brine, dried over Na₂SO₄, filtered, and concentrated. The residue was purified by silica gel chromatography (MeOH/EtOAc) to produce solid white powder **7**.

4.2.1. (1*R*,2*S*)-*N*,*N*'-Bis[(1*S*)-1-hydroxymethyl-2-methyl-propyl]-3,3-dimethyl-1,2-cyclopropanediamide 7a

White solid powder. Yield: 62%. Mp: $121-122 \,^{\circ}C. \, [\alpha]_D^{20} = -30.0$ (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 8.62 (d, *J* = 8.7 Hz, 1H), 8.33 (d, *J* = 9.0 Hz, 1H), 4.56 (t, *J* = 5.4 Hz, 1H), 4.51 (t, *J* = 5.7 Hz, 1H), 3.67–3.58 (m, 2H), 3.38–3.29 (m, 4H), 1.88–1.77 (m, 2H), 1.74 (s, 2H), 1.21 (s, 3H), 1.17 (s, 3H), 0.85–0.80 (m, 12H); ¹³C NMR (75 MHz, DMSO) δ : 169.8, 169.7, 61.5, 56.0, 55.91, 55.90, 34.83, 34.78, 28.7, 28.0, 27.9, 24.2, 19.85, 19.79, 17.8, 17.7, 15.7. El MS (70 eV) *m*/*z*: 328 [M]⁺, 298, 226, 140, 69, 41. HRMS: Calcd for C₁₇H₃₂N₂O₄ [M]⁺ 328.2362. Found: 328.2366.

4.2.2. (1*R*,2*S*)-*N*,*N*'-Bis[(1*S*)-1-hydroxymethyl-2,2-dimethylpropyl]-3,3-dimethyl-1,2-cyclopropane-diamide 7b

White solid powder. Yield: 66%. Mp: $190-192 \,^{\circ}C. \, [\alpha]_{0}^{20} = -29.6$ (*c* 2.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 8.65 (d, *J* = 9.0 Hz, 1H), 8.27 (d, *J* = 9.6 Hz, 1H), 4.40 (t, *J* = 5.7 Hz, 1H), 4.34 (t, *J* = 5.1 Hz, 1H), 3.68–3.52 (m, 4H), 3.34–3.23 (m, 2H), 1.78 (s, 2H), 1.23 (s, 3H), 1.20 (s, 3H), 0.86 (s, 9H), 0.84 (s, 9H). ¹³C NMR (75 MHz, DMSO) δ : 169.97, 169.95, 61.04, 61.02, 59.11, 59.07, 34.9, 33.7, 33.4, 28.8, 27.1, 27.0, 24.4, 16.1. EI MS (70 eV) *m/z*: 356 [M]⁺, 326, 299, 240, 140, 57, 43. Anal. Calcd C₁₉H₃₆N₂O₄: C, 64.01; H, 10.18; N, 7.86. Found: C, 63.80; H, 10.03; N, 7.67.

4.2.3. (1*R*,2*S*)-*N*,*N*'-Bis[(1*S*)-2-hydroxy-1-phenyl-methylethyl]-3,3-dimethyl-1,2-cyclopropanediamide 7c

White solid powder. Yield: 67%. Mp 142–143 °C. $[\alpha]_{\rm D}^{20} = -39.2$ (*c* 2.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 8.75 (d, *J* = 8.1 Hz, 1H), 8.44 (d, *J* = 8.4 Hz, 1H), 7.26–7.13 (m, 10H), 4.76 (t, *J* = 5.7 Hz, 1H), 4.73 (t, *J* = 4.2 Hz, 1H), 3.97–3.89 (m, 2H), 3.40–3.24 (m, 4H), 2.89–2.80(m, 2H), 2.65–2.50 (m, 2H), 1.60 (d, *J* = 3.0 Hz, 2H), 1.12 (s, 3H), 0.84 (s, 3H). ¹³C NMR (75 MHz, DMSO) δ : 169.4, 169.3, 139.31, 139.27, 129.2, 128.2, 128.1, 126.00, 125.97, 63.0, 62.6, 53.1, 52.7, 36.8, 36.7, 34.6, 34.3, 28.5, 24.3, 15.1. El MS (70 eV) *m*/*z*: 424 [M]⁺, 394, 274, 140, 91, 60, 41. Anal. Calcd C₂₅H₃₂N₂O₄: C, 70.73; H, 7.60; N, 6.60. Found: C, 70.66; H, 7.52; N, 6.53.

4.2.4. (1*R*,2*S*)-*N*,*N*'-Bis[(1*S*)-2-hydroxy-1-phenylethyl]-3,3dimethyl-1,2-cyclopropanediamide 7d

White solid powder. Yield: 46%. Mp 135–137 °C. $[\alpha]_D^{20} = +64.8$ (*c* 2.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 9.11 (d, *J* = 8.1 Hz, 1H), 9.01 (d, *J* = 8.4 Hz, 1H), 7.32–7.18 (m, 10H), 4.92–4.78 (m, 4H), 3.57–3.47 (m, 4H), 1.82 (s, 2H), 1.23 (s, 3H), 1.16 (s, 3H). ¹³C NMR (75 MHz, DMSO) δ : 169.5, 169.3, 141.21, 141.17, 128.14, 128.12, 127.1, 127.0, 126.8, 65.0, 64.9, 55.4, 55.2, 34.7, 34.3, 28.7,

24.7, 15.6. EI MS (70 eV) m/z: 366 [M-2CH₃]⁺, 347, 260, 243, 229, 140, 106, 96, 41. Anal. Calcd C₂₃H₂₈N₂O₄: C, 69.67; H, 7.12; N, 7.07. Found: C, 69.62; H, 7.07; N, 7.04.

4.2.5. (1R,2S)-N,N'-Bis[(1R)-1-hydroxymethyl-2-methylpropyl]-3,3-dimethyl-1,2-cyclopropanediamide 7e

White solid powder. Yield: 64%. Mp 121–122 °C. $[\alpha]_D^{25}=+35.0$ (c 1.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 8.64 (d, J = 8.7 Hz, 1H), 8.32 (d, J = 9.0 Hz, 1H), 4.56 (t, J = 5.4 Hz, 1H), 4.50 (t, I = 5.7 Hz, 1H), 3.67–3.58 (m, 2H), 3.39–3.27 (m, 4H), 1.99–1.77 (m, 2H), 1.74 (s, 2H), 1.21 (s, 3H), 1.17 (s, 3H), 0.85-0.73 (m, 12H). ¹³C NMR (75 MHz, DMSO) δ: 170.62, 170.56, 61.9, 56.4, 35.21, 35.16, 28.9, 28.44, 28.40, 24.9, 20.18, 20.15, 18.18, 18.14, 16.1. MS m/z: 329 [M+H]⁺, 311, 226, 208, 152, 140, 132. HRMS: Calcd for C₁₇H₃₃N₂O₄ [M+H]⁺ 329.2440. Found: 329.2435.

4.2.6. (1R.2S)-N.N-Bis[(1R)-2-hvdroxy-1-phenylethyl]-3.3dimethyl-1,2-cyclopropanediamide 7f

White solid powder. Yield: 45%. Mp 137–138 °C. $[\alpha]_D^{25} = -77.5$ (*c* 2.0, CHCl₃). ¹H NMR (300 MHz, DMSO) δ : 9.13 (d, *J* = 8.1 Hz, 1H), 9.03 (d, J = 8.1 Hz, 1H), 7.27-7.21 (m, 10H), 4.89-4.82 (m, 2H), 3.63-3.47 (m, 6H), 1.81 (s, 2H), 1.18 (s, 3H), 1.15 (s, 3H). ¹³C NMR (75 MHz, DMSO) *δ*: 169.6, 169.4, 141.2, 141.1, 128.2, 128.1, 127.1, 127.0, 126.8, 65.0, 64.8, 55.4, 55.3, 34.7, 34.3, 28.6, 24.8, 15.6. HRMS: Calcd for C₂₃H₂₉N₂O₄ [M+H]⁺ 397.2127. Found: 397.2124.

4.3. General procedure for the preparation of cyclopropanebased bisoxazoline ligands 8¹⁹

To a solution of dihydroxy diamide 7 (9.0 mmol), Et₃N (5.52 mL, 39.6 mmol) and a catalytic amount of 4-(dimethylamino)pyridine (0.121 g, 0.99 mmol) in CH₂Cl₂ (72 mL), a solution of *p*-toluenesulfonyl chloride (3.433 g, 18 mmol) in CH₂Cl₂ (42 mL) was added slowly. The reaction mixture was then stirred at room temperature for 27 h. The mixture was then diluted with CH₂Cl₂ (200 mL), and washed with saturated aqueous NH₄Cl, Na₂CO₃ and brine successively. The organic layer was dried over Na₂SO₄, filtered, and concentrated under reduced pressure to give a pale vellow viscous oil. The residue was purified by neutral Al₂O₃ column chromatography (pet ether/EtOAc) to afford 8 as colorless oil or white solid.

4.3.1. (1R,2S)-1,2-Bis[4(S)-isopropyloxazolin-2-yl]-3,3dimethylcyclopropane 8a

Colorless oil. Yield: 82%. $[\alpha]_D^{20} = -131.3$ (*c* 1.6, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 4.23–4.10 (m, 2H), 3.88–3.79 (m, 4H), 1.80 (d, J = 3.6 Hz, 2H), 1.76–1.65 (m, 2H), 1.42 (s, 3H), 1.23 (s, 3H), 0.97 (t, J = 7.2 Hz, 6H), 0.87 (t, J = 6.9 Hz, 6H). ¹³C NMR (75 MHz, CDCl₃) *δ*: 163.7, 163.2, 72.5, 72.2, 69.5, 69.4, 32.7, 32.5, 28.3, 26.6, 26.4, 24.4, 19.1, 19.0, 18.5, 18.1, 16.7. EI MS (70 eV) m/z: 292[M]⁺, 277, 249, 163, 136, 69, 41. HRMS: Calcd for C₁₇H₂₈N₂O₂ [M]⁺ 292.2151. Found: 292.2149.

4.3.2. (1R,2S)-1,2-Bis[4(S)-tert-butyloxazolin-2-yl]-3,3dimethylcyclopropane 8b

White solid. Yield: 52%. Mp 42–43 °C, $[\alpha]_D^{20} = -135.8$ (c 2.1, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 4.15-4.04 (m, 2H), 3.92-3.75 (m, 4H), 1.82 (d, J = 4.8 Hz, 2H), 1.43 (s, 3H), 1.23 (s, 3H), 0.90 (s, 9H), 0.87 (s, 9H). ¹³C NMR (75 MHz, CDCl₃) δ: 163.9, 163.2, 76.2, 75.7, 68.0, 67.9, 33.5, 28.5, 27.1, 26.4, 26.1, 25.9, 24.6, 16.8. EI MS (70 eV), m/z: 320 [M]⁺, 305, 279, 263, 205, 163, 136, 123, 57, 41. HRMS: Calcd for C₁₉H₃₂N₂O₂ [M]⁺ 320.2464. Found: 320.2462.

4.3.3. (1R,2S)-1,2-Bis[4(S)-benzyloxazolin-2-yl]-3,3dimethylcyclopropane 8c

Colorless oil. Yield: 39%. $[\alpha]_{D}^{20} = -56.5$ (*c* 0.78, CHCl₃). ¹H NMR (300 MHz, CDCl₃) *δ*: 7.30–7.16 (m, 10H), 4.37–4.30 (m, 2H), 4.11–

4.05 (m, 2H), 3.88-3.81 (m, 2H), 3.15-2.99 (m, 2H), 2.61-2.53 (m, 2H), 1.81 (d, J = 0.9 Hz, 2H), 1.44 (s, 3H), 1.25 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) *δ*: 164.1, 163.8, 138.4, 138.2, 129.1, 129.0, 128.43, 128.37, 126.32, 126.27, 71.0, 67.6, 67.5, 41.8, 41.6, 28.3, 26.5, 26.4, 24.6, 16.6. EI MS (70 eV), m/z: 388 [M]⁺, 297, 163, 91, 67, 41. HRMS: Calcd for C₂₅H₂₈N₂O₂ [M]⁺ 388.2151. Found: 388.2150.

4.3.4. (1R,2S)-1,2-Bis[4(S)-phenyloxazolin-2-yl]-3,3dimethylcyclopropane 8d

Colorless oil. Yield: 45%. $[\alpha]_D^{20} = -152.3$ (*c* 2.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.34-7.11 (m, 10H), 5.19-5.12 (m, 2H), 4.58-4.48 (m, 2H), 3.99-3.90 (m, 2H), 1.96 (d, J = 10.2 Hz, 2H), 1.58 (s, 3H), 1.32 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ: 165.0, 164.5, 142.7, 142.6, 128.6, 128.4, 127.3, 127.2, 127.0, 126.7, 74.1, 70.2, 70.0, 28.4, 26.8, 26.3, 25.1, 16.8. EI MS (70 eV), m/z: 360 [M]⁺, 345, 276, 242, 120, 104, 91, 43, 29, HRMS: Calcd for C₂₃H₂₄N₂O₂ [M]⁺ 360.1838. Found: 360.1853.

4.3.5. (1R,2S)-1,2-Bis[4(R)-isopropyloxazolin-2-yl]-3,3dimethylcyclopropane 8e

Colorless oil. Yield: 70%. $[\alpha]_{D}^{25} = +126.5$ (*c* 2.4, CHCl₃) ¹H NMR (300 MHz, CDCl₃) δ: 4.20-4.09 (m, 2H), 3.88-3.79 (m, 4H), 1.84-1.65 (m, 4H), 1.42 (s, 3H), 1.23 (s, 3H), 0.96 (t, J = 6.96 Hz, 6H), 0.87 (t, I = 6.87 Hz, 6H). ¹³C NMR (75 MHz, CDCl₃) δ : 163.7, 163.2, 72.6, 72.3, 69.6, 69.4, 32.8, 32.6, 28.4, 26.6, 26.4, 24.5, 19.1, 19.0, 18.5, 18.1, 16.7; EI MS m/z: 293 [M+H]⁺, 226, 208, 185, 168, 152, 140. HRMS: Calcd for C₁₇H₂₉N₂O₂ [M+H]⁺ 293.2229. Found: 293.2197.

4.3.6. (1R,2S)-1,2-Bis[4(R)-phenyloxazolin-2-yl]-3,3dimethylcyclopropane 8f

Colorless oil. Yield: 53%. $[\alpha]_D^{25} = +160.2$ (*c* 1.5, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.35–7.12 (m, 10H), 5.19–5.12 (m, 2H), 4.58-4.48 (m, 2H), 3.99-3.90 (m, 2H), 1.96 (d, J = 2.1 Hz, 2H), 1.58 (s, 3H), 1.32 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ: 164.9, 164.5, 142.6, 142.5, 128.5, 128.3, 127.3, 127.1, 126.9, 126.7, 74.0, 70.1, 69.8, 28.3, 26.7, 26.2, 25.1, 16.8, EI MS m/z: 360 [M]⁺, 345, 276, 199, 184, 155, 104, 91, 83, 56. HRMS: Calcd for C₂₃H₂₄N₂O₂ [M]⁺ 360.1838. Found: 360.1842.

4.4. General procedure for the asymmetric Henry reaction^{17a}

Ligand 8 (0.03 mmol) was mixed with $Cu(OAc)_2$ (4.54 mg, 0.025 mmol) under an argon atmosphere. Isopropanol (1.0 mL) was then added, and the mixture was stirred for 1 h. To the resulting clear solution, nitromethane (0.3 mL, 5 mmol) and the aldehyde (0.5 mmol) were added via syringe. Stirring was continued at 0 °C until TLC-control indicated the complete consumption of the aldehyde. The reaction mixture was diluted with saturated aqueous NH₄Cl (10 mL) and extracted with EtOAc (3×20 mL). The volatile components were removed under reduced pressure and the crude products were purified by column chromatography to give the desired products. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H, Chiralpak AD-H, or AS-H column, and the absolute configurations of the nitroaldol products were assigned by comparing their specific rotations or the HPLC elution order with data from the literature.

4.4.1. (5)-2-Nitro-1-phenylethanol 10a^{17a} Colorless oil. Yield 70%. $[\alpha]_D^{20} = +47.9$ (*c* 1.5, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.41-7.33 (m, 5H), 5.46-5.41 (m, 1H), 4.63-4.46 (m, 2H), 2.93 (d, J = 3.9 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 138.1, 128.8, 128.7, 125.8, 81.0, 70.8. HRMS (TOF): Calcd for C₈H₉NNaO₃ [M+Na]⁺: 190.0480, Found: 190.0476. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)enantiomer $t_{\rm R}$ = 8.3 min, major (S)-enantiomer $t_{\rm R}$ = 10.3 min; 85% ee.

4.4.2. (R)-2-Nitro-1-phenylethanol 10a^{17a}

Colorless oil. Yield 91%. $[\alpha]_D^{20} = -29.6$ (*c* 1.8, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.41-7.33 (m, 5H), 5.46-5.41 (m, 1H), 4.63-4.46 (m, 2H), 2.93 (d, J = 3.9 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 138.1, 128.8, 128.7, 125.8, 81.0, 70.8. HRMS (TOF): Calcd for C₈H₉NNaO₃ [M+Na]⁺: 190.0480, Found: 190.0474. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); major (R)enantiomer $t_{\rm R}$ = 8.7 min, minor (S)-enantiomer $t_{\rm R}$ = 10.5 min; 62% ee.

4.4.3. (S)-1-(3-Chlorophenyl)-2-nitroethanol 10b^{21a}

Colorless oil. Yield 95%. $[\alpha]_D^{20}=+27.3$ (c 1.6, CHCl_3). ^1H NMR (300 MHz, CDCl₃) δ: 7.42-7.25 (m, 4H), 5.46-5.41 (m, 1H), 4.61-4.47 (m, 2H), 3.09 (d, J = 3.94 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ: 140.0, 134.7, 130.2, 128.9, 126.1, 124.0, 80.8, 70.1. HRMS (TOF): Calcd for C₈H₈ClNNaO₃ [M+Na]⁺: 224.0090, Found: 224.0089. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/ min, 230 nm); minor (*R*)-enantiomer $t_{\rm R}$ = 8.4 min, major (*S*)-enantiomer $t_{\rm R}$ = 10.3 min; 81% ee.

4.4.4. (S)-1-(4-Chlorophenyl)-2-nitroethanol 10c^{17a}

Colorless oil. Yield 96%. $[\alpha]_D^{20} = +30.4$ (*c* 0.6, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.39–7.26 (m, 4H), 5.45–5.40 (m, 1H), 4.60– 4.45 (m, 2H), 3.08 (d, J = 3.5 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 136.5, 134.6, 129.0, 127.3, 80.8, 70.2. HRMS (TOF): Calcd for C₈H₈ClNKO₃ [M+K]⁺: 239.9830, Found: 239.9824. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (R)-enantiomer t_R = 8.1 min, major (*S*)-enantiomer t_R = 9.8 min; 82% ee.

4.4.5. (S)-1-(3-Bromophenyl)-2-nitroethanol 10d^{21b}

Colorless oil. Yield 80%. $[\alpha]_D^{20} = +23.4$ (*c* 1.6, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.58-7.25 (m, 4H), 5.46-5.41 (m, 1H), 4.61-4.47 (m, 2H), 3.03 (d, J = 3.9 Hz, 1H), ¹³C NMR (75 MHz, CDCl₃) δ : 140.3, 131.8, 130.5, 129.0, 124.5, 122.8, 80.8, 70.1. HRMS (TOF): Calcd for C₈H₈BrNNaO₃ [M+Na]⁺: 267.9585, Found: 267.9572. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)-enantiomer $t_{\rm R}$ = 9.0 min, major (*S*)-enantiomer $t_{\rm R}$ = 11.6 min; 81% ee.

4.4.6. (S)-1-(4-Bromophenyl)-2-nitroethanol 10e^{15a}

White solid. Yield 88%. $[\alpha]_{D}^{20} = +25.8$ (*c* 1.4, CHCl₃). ¹H NMR (300 MHz, CDCl₃) *δ*: 7.56–7.52 (m, 2H), 7.31–7.26 (m, 2H), 5.46– 5.41 (m, 1H), 4.61–4.46 (m, 2H), 2.90 (d, *J* = 3.9 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ: 137.0, 132.2, 127.6, 123.0, 80.9, 70.3. HRMS (TOF): Calcd for C₈H₈BrNKO₃ [M+K]⁺: 283.9325, Found: 283.9319. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (R)-enantiomer t_R = 8.9 min, major (S)-enantiomer *t*_R = 11.3 min; 81% ee.

4.4.7. (S)-1-(2-Fluorophenyl)-2-nitroethanol 10f^{21c}

Colorless oil. Yield 84%. $[\alpha]_{D}^{20} = +37.6$ (*c* 1.8, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.56-7.53 (m, 1H), 7.37-7.32 (m, 1H), 7.26-7.19 (m, 1H), 7.11–7.05 (m, 1H), 5.74 (dd, J = 8.0, 4.15 Hz, 1H), 4.65–4.54 (m, 2H), 3.07 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 160.9, 157.7, 130.4, 127.5, 125.0, 115.5, 79.6, 65.4. HRMS (TOF): Calcd for C₈H₈FNNaO₃ [M+Na]⁺: 208.0386, Found: 208.0387. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (95:5 n-hexanes/isopropanol, 0.8 mL/min, 230 nm); minor (*R*)-enantiomer $t_{\rm R}$ = 23.3 min, major (*S*)-enantiomer $t_{\rm R}$ = 24.6 min; 73% ee.

4.4.8. (S)-1-(4-Nitrophenyl)-2-nitroethanol 10g^{15a}

White solid. Yield 91%. $[\alpha]_{D}^{20} = +20.0$ (*c* 1.6, CHCl₃). ¹H NMR $(300 \text{ MHz}, \text{ CDCl}_3) \delta$: 8.29–8.24 (m, 2H), 7.63 (d, J = 8.6 Hz, 2H), 5.62 (dd, J = 7.7, 4.6 Hz, 1H), 4.66-4.55 (m, 2H), 3.20 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ: 148.1, 144.9, 126.9, 124.2, 80.6, 69.9. HRMS (TOF): Calcd for C₈H₈N₂NaO₅ [M+Na]⁺: 235.0331, Found: 235.0322. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/ min, 230 nm); minor (R)-enantiomer $t_{\rm R}$ = 12.8 min, major (S)-enantiomer *t*_R = 15.3 min; 68% ee.

4.4.9. (S)-1-(2-Methoxyphenyl)-2-nitroethanol 10h^{17a}

Yellow oil. Yield 96%. $[\alpha]_D^{20} = +40.9$ (c 1.8, CHCl₃). ¹H NMR $(300 \text{ MHz}, \text{ CDCl}_3) \delta$: 7.44 (dd, I = 7.5, 1.5 Hz, 1H), 7.36-7.30 (m, 1H), 7.04–7.00 (m, 1H), 6.91 (d, J = 8.3 Hz, 1H), 5.63 (d, J = 6.8 Hz, 1H), 4.68–4.53 (m, 2H), 3.88 (s, 3H), 3.14 (br s, 1H). $^{13}\mathrm{C}$ NMR (75 MHz, CDCl₃) δ: 156.0, 129.8, 127.2, 125.9, 121.1, 110.5, 79.8, 67.8, 55.4. HRMS (TOF): Calcd for $C_9H_{11}NNaO_4$ [M+Na]⁺: 220.0586, Found: 220.0572. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)-enantiomer t_R = 8.0 min, major (*S*)-enantiomer t_R = 9.0 min; 82% ee.

4.4.10. (S)-1-(3-Methoxyphenyl)-2-nitroethanol 10i^{21a}

Colorless oil. Yield 95%. $[\alpha]_{D}^{20} = +29.5$ (*c* 1.4, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.33-7.26 (m, 1H), 6.96-6.86 (m, 3H), 5.42 (d, J = 9.3 Hz, 1H), 4.62–4.46 (m, 2H), 3.81 (s, 3H), 2.97 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ: 159.8, 139.8, 129.9, 118.0, 114.2, 111.4, 81.0, 70.7, 55.2. HRMS (TOF): Calcd for $C_9H_{11}NNaO_4$ [M+Na]⁺: 220.0586, Found: 220.0584. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (R)-enantiomer $t_{\rm R}$ = 13.9 min, major (S)-enantiomer $t_{\rm R}$ = 18.2 min; 81% ee.

4.4.11. (S)-1-(4-Methoxyphenyl)-2-nitroethanol 10j^{21a}

Colorless oil. Yield 63%. $[\alpha]_D^{20} = +29.4$ (*c* 1.5, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.34-7.29 (m, 2H), 6.94-6.90 (m, 2H), 5.40 (dd, J = 9.5, 3.2 Hz, 1H), 4.64-4.45 (m, 2H), 3.81 (s, 3H), 2.75 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ: 160.0, 130.2, 127.3, 114.4, 81.2, 70.6, 55.3. HRMS (TOF): Calcd for C₉H₁₁NNaO₄ [M+Na]⁺: 220.0586. Found: 220.0577. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 n-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (R)-enantiomer t_R = 11.2 min, major (*S*)-enantiomer t_R = 13.6 min; 86% ee.

4.4.12. (S)-1-(2-Methylphenyl)-2-nitroethanol 10k^{17a}

Colorless oil. Yield 93%. $[\alpha]_D^{20} = +41.4$ (c 1.3, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.52-7.49 (m, 1H), 7.27-7.17 (m, 3H), 5.69-5.64 (m, 1H), 4.57-4.39 (m, 2H), 2.78-2.77 (m, 1H), 2.38 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ: 136.2, 134.4, 130.7, 128.5, 126.6, 125.5, 80.1, 67.8, 18.7. HRMS (TOF): Calcd for C9H11NNaO3 [M+Na]⁺: 204.0637. Found: 204.0632. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (85:15 n-hexanes/isopropanol, 0.8 mL/min, 215 nm); minor (R)-enantiomer $t_{\rm R}$ = 7.7 min, major (S)-enantiomer $t_{\rm R}$ = 11.8 min; 83% ee.

4.4.13. (5)-1-(3-Methylphenyl)-2-nitroethanol 10l^{15a} Colorless oil. Yield 65%. $[\alpha]_D^{20} = +36.4$ (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ: 7.31-7.15 (m, 4H), 5.42-5.37 (m, 1H), 4.62-4.45 (m, 2H), 2.89 (d, J = 3.8 Hz, 1H), 2.36 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ: 138.7, 138.1, 129.5, 128.8, 126.5, 122.9, 81.1, 70.9, 21.2. HRMS (TOF): Calcd for C₉H₁₁NNaO₃ [M+Na]⁺:

204.0637. Found: 204.0634. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)-enantiomer t_R = 7.4 min, major (*S*)-enantiomer t_R = 8.4 min; 86% ee.

4.4.14. (S)-1-(4-Methylphenyl)-2-nitroethanol 10m^{15a}

White solid. Yield 97%. $[\alpha]_D^{20} = +37.4$ (*c* 1.8, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.29–7.19 (m, 4H), 5.42 (dd, *J* = 9.44, 3.10 Hz, 1H), 4.64–4.45 (m, 2H), 2.77 (s, 1H), 2.36 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ : 138.9, 135.1, 129.6, 125.8, 81.2, 70.9, 21.1. HRMS (TOF): Calcd for C₉H₁₁NNaO₃ [M+Na]⁺: 204.0637, Found: 204.0635. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)-enantiomer *t*_R = 8.8 min, major (*S*)-enantiomer *t*_R = 10.7 min; 83% ee.

4.4.15. (+)-1-(4-Ethoxyphenyl)-2-nitroethanol 10n

White solid. Yield 70%. $[\alpha]_D^{20} = +32.5$ (*c* 1.1, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.31–7.26 (m, 2H), 6.93–6.88 (m, 2H), 5.42–5.36 (m, 1H), 4.63–4.44 (m, 2H), 4.03 (q, *J* = 7.0 Hz, 2H), 2.76 (dd, *J* = 3.6, 1.1 Hz, 1H). 1.41 (t, *J* = 7.0 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃) δ : 159.2, 130.1, 127.2, 114.8, 81.1, 70.5, 63.5, 14.6. HRMS (TOF): Calcd for C₁₀H₁₃NNaO₄ [M+Na]⁺: 234.0742. Found: 234.0736. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor enantiomeric t_R = 8.9 min, major enantiomeric t_R = 10.1 min; 87% ee.

4.4.16. (+)-1-(2,4-Dimethoxyphenyl)-2-nitroethanol 10o

Pale yellow solid. Yield 90%. $[\alpha]_{D}^{20} = +34.7$ (*c* 1.2, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.33–7.26 (m, 1H), 6.53–6.47 (m, 2H), 5.55 (dd, *J* = 7.2, 5.3 Hz, 1H), 4.63–4.53 (m, 2H), 3.85 (s, 3H), 3.81 (s, 3H), 3.07 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 161.1, 157.1, 128.0, 118.5, 104.6, 98.7, 80.0, 67.6, 55.38, 55.36. HRMS (TOF): Calcd for C₁₀H₁₃NNaO₅ [M+Na]⁺: 250.0691. Found: 250.0694. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor enantiomer $t_{\rm R}$ = 9.7 min, major enantiomer $t_{\rm R}$ = 13.3 min; 79% ee.

4.4.17. (S)-1-(3,5-Dimethoxyphenyl)-2-nitroethanol 10p^{15e}

White solid. Yield 96%. $[R]_D^{20} = +24.4$ (*c* 1.1, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 6.54 (d, *J* = 2.2 Hz, 2H), 6.43 (t, *J* = 2.2 Hz, 1H), 5.42–5.36 (m, 1H), 4.63–4.47 (m, 2H), 3.80 (s, 6H), 2.84 (d, *J* = 3.7 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 161.1, 140.5, 103.8, 100.5, 81.1, 70.9, 55.3. HRMS (TOF): Calcd for C₁₀H₁₃NNaO₅ [M+Na]⁺: 250.0691. Found: 250.0695. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); major (*S*)-enantiomer t_R = 8.0 min, minor (*R*)-enantiomer t_R = 9.6 min; 83% ee.

4.4.18. (-)-1-(2,6-Dimethoxyphenyl)-2-nitroethanol 10q

White solid. Yield 97%. $[\alpha]_D^{20} = -9.2$ (*c* 1.8, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.26 (dd, *J* = 10.0, 8.4 Hz, 1H), 6.59 (d, *J* = 8.4 Hz, 2H), 5.97–5.89 (m, 1H), 4.85 (dd, *J* = 11.9, 10.0 Hz, 1H), 4.46 (dd, *J* = 11.9, 3.6 Hz, 1H), 4.12 (d, *J* = 11.4 Hz, 1H), 3.88 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ : 157.8, 130.2, 113.4, 104.2, 79.8, 65.3, 55.8. HRMS (TOF): Calcd for C₁₀H₁₃NNaO₅ [M+Na]*: 250.0691. Found: 250.0686. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); major enantiomer t_R = 14.0 min, minor enantiomer t_R = 20.0 min, 84% ee.

4.4.19. (S)-1-(2-Naphthyl)-2-nitroethanol 10r^{21a}

Pale yellow solid. Yield 96%. $[\alpha]_D^{20} = +32.3$ (*c* 1.3, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.85–7.83 (m, 4H), 7.52–7.40 (m, 3H),

5.57 (d, J = 6.9 Hz, 1H), 4.68–4.51 (m, 2H), 3.04 (br s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 135.4, 133.3, 133.1, 128.8, 128.0, 127.7, 126.59, 126.56, 125.2, 123.1, 81.0, 71.0. HRMS (TOF): Calcd for C₁₂H₁₁NNaO₃ [M+Na]⁺: 240.0637, Found: 240.0639. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); minor (*R*)-enantiomer $t_R = 23.4$ min, major (*S*)-enantiomer $t_R = 33.4$ min; 73% ee.

4.4.20. (S)-1-Nitrononan-2-ol 10s^{1d}

Colorless oil. Yield 60%. $[\alpha]_D^{20} = +6.2$ (*c* 1.2, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 4.47–4.29 (m, 3H), 2.66 (d, *J* = 4.7 Hz, 1H), 1.57–1.29 (m, 12H), 0.89 (m, *J* = 6.5 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃) δ : 80.6, 68.7, 33.7, 31.6, 29.2, 29.0, 25.1, 22.5, 13.9. HRMS (TOF): Calcd for C₉H₁₉NNaO₃ [M+Na]⁺: 212.1263. Found: 212.1266. Enantiomeric excess was determined by HPLC with a Chiralpak AS-H column (98:2 *n*-hexanes/isopropanol, 1.0 mL/min, 210 nm); major (*S*)-enantiomer t_R = 7.0 min, minor (*R*)-enantiomer t_R = 7.8 min; 69% ee.

4.4.21. (S,E)-1-Nitro-4-phenyl-3-buten-2-ol 10t^{21a}

Pale yellow solid. Yield 75%. $[\alpha]_{D}^{20} = +3.7$ (*c* 1.2, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 7.41–7.26 (m, 5H), 6.79 (dd, *J* = 16.0, 1.2 Hz, 1H), 6.15 (dd, *J* = 15.9, 6.3 Hz, 1H), 5.09–5.03 (m, 1H), 4.56–4.47 (m, 2H), 2.62 (s, 1H). ¹³C NMR (75 MHz, CDCl₃) δ : 135.5, 133.7, 128.7, 128.5, 126.7, 124.9, 79.8, 69.6. HRMS (TOF): Calcd for C₁₀H₁₁NNaO₃ [M+Na]⁺: 216.0637, Found: 216.0635. Enantiomeric excess was determined by HPLC with a Chiralcel OD-H column (80:20 *n*-hexanes/isopropanol, 1.0 mL/min, 230 nm); major (*S*)enantiomer t_{R} = 18.6 min, minor (*R*)-enantiomer t_{R} = 21.3 min; 77% ee.

4.4.22. (S)-1-Cyclohexyl-2-nitroethanol 10u^{17a}

Colorless oil. Yield 65%. $[\alpha]_D^{20} = +13.7$ (*c* 1.2, CHCl₃). ¹H NMR (300 MHz, CDCl₃) δ : 4.52–4.38 (m, 2H), 4.13–4.06 (m, 1H), 2.60 (d, *J* = 5.2 Hz, 1H), 1.81–1.77 (m, 3H), 1.71–1.65 (m, 2H), 1.24–1.20 (m, 1H) 1.17–1.07 (m, 5H). ¹³C NMR (75 MHz, CDCl₃) δ : 79.3, 72.8, 41.3, 28.5, 27.8, 26.0, 25.8, 25.6. HRMS (TOF): Calcd for C₈H₁₅NNaO₃ [M+Na]⁺: 196.0950, Found: 196.0943. Enantiomeric excess was determined by HPLC with a Chiralpak AD-H column (90:10 *n*-hexanes/isopropanol, 1.0 mL/min, 210 nm); minor (*R*)-enantiomer t_R = 8.1 min, major (*S*)-enantiomer t_R = 8.9 min; 74% ee.

Acknowledgments

We thank the Twelfth Five-year Science and Technology Support Program (2012BAK25B03), China Postdoctoral Science Foundation Funded Project (2012M510621) and the National Basic Research Program of China (81102340) for their financial support.

References

- (a) Desimoni, G.; Faita, G.; Jørgensen, K. A. Chem. Rev. 2006, 106, 3561–3651;
 (b) Mcmanus, H. A.; Guiry, P. J. Chem. Rev. 2004, 104, 4151–4202; (c) Zhu, S. F.;
 Xu, B.; Wang, G. P.; Zhou, Q. L. J. Am. Chem. Soc. 2012, 134, 436–442; (d) Zhu, S.
 F.; Song, X. G.; Li, Y.; Cai, Y.; Zhou, Q. L. J. Am. Chem. Soc. 2010, 132, 16374–16376; (e) Lou, S.; Fu, G. C. J. Am. Chem. Soc. 2010, 132, 1264–1266.
- (a) Lowenthal, R. E.; Abiko, A.; Masamune, S. *Tetrahedron Lett.* **1990**, *31*, 6005–6008; (b) Corey, E. J.; Imai, N.; Zhang, H. Y. *J. Am. Chem. Soc.* **1991**, *113*, 728–729; (c) Evans, D. A.; Woerpel, K. A.; Hinman, M. M.; Faul, M. M. J. Am. Chem. Soc. **1991**, *113*, 726–728.
- (a) Harm, A. M.; Knight, J. G.; Stemp, G. Synlett **1996**, 677–678; (b) Bedekar, A. V.; Andersson, P. G. *Tetrahedron Lett.* **1996**, 37, 4073–4076; (c) Bedekar, A. V.; Koroleva, E. B.; Andersson, P. G. J. Org. Chem. **1997**, 62, 2518–2526.
- (a) Park, J.; Lee, S.; Ahn, K. H.; Cho, C. W. Tetrahedron Lett. **1995**, 36, 7263–7266;
 (b) Zhang, W.; Adachi, Y.; Hirao, T.; Ikeda, I. Tetrahedron: Asymmetry **1996**, 7, 451–460;
 (c) Kim, S. G.; Cho, C. W.; Ahn, K. H. Tetrahedron **1999**, 55, 10079–10086;
 (d) Kato, K.; Tanaka, M.; Yamamura, S.; Yamamoto, Y.; Akita, H. Tetrahedron Lett. **2003**, 44, 3089–3092.

- (a) Uozumi, Y.; Kyota, H.; Kitayama, K.; Hayashi, T. *Tetrahedron: Asymmetry* 1996, 7, 1603–1606; (b) Imai, Y.; Zhang, W. B.; Kida, T.; Nakatsuji, Y.; Ikeda, I. *Tetrahedron Lett.* 1997, 38, 2681–2684; (c) Meyers, A. I.; Price, A. J. Org. Chem. 1998, 63, 412–413; (d) Imai, Y.; Zhang, W. B.; Kida, T.; Nakatsuji, Y.; Ikeda, I. J. Org. Chem. 2000, 65, 3326–3333.
- (a) Nishiyama, H.; Yamaguchi, S.; Kondo, M.; Itoh, K. J. Org. Chem. 1992, 57, 4306–4309; (b) Nishiyama, H.; Soeda, N.; Naito, T.; Motoyama, Y. Tetrahedron: Asymmetry 1998, 9, 2865–2869; (c) Evans, D. A.; MacMillan, D. W. C.; Campos, K. R. J. Am. Chem. Soc. 1997, 119, 10859–10860; (d) Son, S.; Fu, G. C. J. Am. Chem. Soc. 2008, 130, 2756–2757.
- 7. Molander, G. A.; Burke, J. P.; Carroll, P. J. J. Org. Chem. 2004, 69, 8062-8069.
- (a) Zhong, J. C.; Wang, M. A.; Guo, H. C.; Bian, Q. H.; Wang, M. Synlett 2006, 1667–1670; (b) Zhong, J. C.; Wang, M. A.; Guo, H. C.; Yin, M. M.; Wang, M. Tetrahedron: Asymmetry 2007, 18, 734–741; (c) Zhong, J. C.; Hou, S. C.; Bian, Q. H.; Yin, M. M.; Na, R. S.; Zheng, B.; Li, Z. Y.; Liu, S. Z.; Wang, M. Chem. Eur. J. 2009, 15, 3069–3071; (d) Li, Z. Y.; Wang, M.; Bian, Q. H.; Zheng, B.; Mao, J. Y.; Li, S. N.; Liu, S. Z.; Wang, M. A.; Zhong, J. C.; Guo, H. C. Chem. Eur. J. 2011, 17, 5782–5786; (e) Zheng, B.; Wang, M.; Li, Z. Y.; Bian, Q. H.; Mao, J. Y.; Li, S. N.; Liu, S. Z.; Zhong, J. C.; Guo, H. C. Tetrahedron: Asymmetry 2011, 22, 1156– 1160.
- 9. Palomo, C.; Oiarbide, M.; Mielgo, A. Angew. Chem., Int. Ed. 2004, 43, 5442-5444.
- (a) Sasai, H.; Suzuki, T.; Arai, Š.; Arai, T.; Shibasaki, M. J. Am. Chem. Soc. 1992, 114, 4418–4420; (b) Choudary, B. M.; Ranganath, K. V. S.; Pal, U.; Kantam, M. L.;
- Sreedhar, B. J. Am. Chem. Soc. 2005, 127, 13167–13171.
 (a) Liu, S.; Wolf, C. Org. Lett. 2008, 10, 1831–1834; (b) Spangler, K. Y.; Wolf, C. Org. Lett. 2009, 11, 4724–4727.
- (a) Li, H.; Wang, B.; Deng, L. J. Am. Chem. Soc. 2006, 128, 732–733; (b) Palacio,
 C; Connon, S. J. Org. Lett. 2011, 13, 1298–1301; (c) Zielinska-Błajet, M.;
 Skarzewski, J. Tetrahedron: Asymmetry 2011, 22, 351–355.
- (a) Trost, B. M.; Yeh, V. S. C. Angew. Chem., Int. Ed. 2002, 41, 861–863; (b) Trost, B. M.; Yeh, V. S. C.; Ito, H.; Bremeyer, N. Org. Lett. 2002, 4, 2621–2623.
- (a) Palomo, C.; Oiarbide, M.; Laso, A. Angew. Chem., Int. Ed. 2005, 44, 3881– 3884; (b) Kim, H. Y.; Oh, K. Org. Lett. 2009, 11, 5682–5685; (c) Guo, Z. L.; Zhong, S.; Li, Y. B.; Lu, G. Tetrahedron: Asymmetry 2011, 22, 238–245; (d) Zhong, Y. W.; Tian, P.; Lin, G. Q. Tetrahedron: Asymmetry 2004, 15, 771–776.

- (a) Sanjeevakumar, N.; Periasamy, M. *Tetrahedron: Asymmetry* **2009**, *20*, 1842–1847; (b) Arai, T.; Watanabe, M.; Fujiwara, A.; Yokoyama, N.; Yanagisawa, A. *Angew. Chem., Int. Ed.* **2006**, *45*, 5978–5981; (c) Kodama, K.; Sugawara, K.; Hirose, T. *Chem. Eur. J.* **2011**, *17*, 13584–13592; (d) Arai, T.; Watanabe, M.; Yanagisawa, A. *Org. Lett.* **2007**, *9*, 3595–3597; (e) Bandini, M.; Piccinelli, F.; Tommasi, S.; Umani-Ronch, A.; Ventric, C. *Chem. Commun.* **2007**, 616–618; (f) Noole, A.; Lippur, K.; Metsala, A.; Lopp, M.; Kanger, T. *J. Org. Chem.* **2010**, *75*, 1313–1316.
- (a) Gan, C.; Lai, G.; Zhang, Z.; Wang, Z.; Zhou, M. M. Tetrahedron: Asymmetry 2006, 17, 725–728; (b) Xin, D. Y.; Ma, Y. D.; He, F. Y. Tetrahedron: Asymmetry 2010, 21, 333–338; (c) Colak, M.; Demirel, N. Tetrahedron: Asymmetry 2008, 19, 635–639; (d) Wei, Y.; Yao, L.; Zhang, B.; He, W.; Zhang, S. Y. Tetrahedron 2011, 67, 8552–8558; (e) Lai, G. Y.; Wang, S. J.; Wang, Z. Y. Tetrahedron: Asymmetry 2008, 19, 1813–1819; (f) Ji, Y. Q.; Qi, G.; Judeh, Z. M. A. Eur. J. Org. Chem. 2011, 25, 4892–4898.
- (a) Evans, D. A.; Seidel, D.; Rueping, M.; Lam, H. W.; Shaw, J. T.; Downey, C. W. J. Am. Chem. Soc. 2003, 125, 12692–12693;; (b) Christensen, C.; Juhl, K.; Hazell, R. G.; Jørgensen, K. A. J. Org. Chem. 2002, 67, 4875–4881;; (c) Christensen, C.; Juhl, K.; Jørgensen, K. A. Chem. Commun. 2001, 2222–2223; (d) Lang, K.; Park, J.; Hong, S. J. Org. Chem. 2010, 75, 6424–6435; (e) Lee, J. M.; Kim, J.; Shin, Y.; Yeom, C. E.; Lee, J. E.; Hyeon, T. Tetrahedron: Asymmetry 2010, 21, 285–291; (f) Subba Reddy, B. V.; George, J. Tetrahedron: Asymmetry 2011, 22, 1169–1175; (g) Lee, J. M.; Kim, J.; Shin, Y.; Yeom, C. E.; Lee, J. E.; Hyeon, T.; Kim, B. M. Tetrahedron Asymmetry 2010, 21, 285–291; (h) Du, D. M.; Lu, S. F.; Fang, T.; Xu, J. X. J. Org. Chem. 2005, 70, 3712–3715.
- 18. Bian, Q. H.; Liu, J.; Yin, M. M.; Wang, M. Chin. Chem. Lett. 2006, 17, 1033-1036.
- Evans, D. A.; Peterson, G. S.; Johnson, J. S.; Barnes, D. M.; Campos, K. R.; Woerpel, K. A. J. Org. Chem. 1998, 63, 4541–4544.
- (a) Montellano, P. R. O.; Dinizo, S. E. J. Org. Chem. 1978, 43, 4323–4328; (b) Blomquist, A. T.; Longone, D. T. J. Am. Chem. Soc. 1959, 81, 2012–2017.
- (a) Ji, Y. Q.; Qi, G.; Judeh, Z. M. A. Tetrahedron: Asymmetry 2011, 22, 929–935;
 (b) Vazquez-Villa, H.; Reber, S.; Ariger, M. A.; Carreira, E. M. Angew. Chem., Int. Ed. 2011, 50, 8979–8981;
 (c) Kowalczyk, R.; Sidorowicz, L.; Skarzewski, J. Tetrahedron: Asymmetry 2007, 18, 2581–2586;
 (d) Jin, W.; Li, X. C.; Huang, Y.; Wu, F.; Wan, B. Chem. Eur. J. 2010, 16, 8259–8261.