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Ring Expansive Routes to Quinolizidine Alkaloids: Formal Synthesis of (—)-Lasubine II

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ABSTRACT

The application of two nitrogen ring expansion reactions to lasubine alkaloid synthesis is reported. The approach involves a conjugate reduction/alkylation sequence carried out on triisopropylsilyl-protected (*S*)-4-(–)-hydroxycyclopentenone, the formation of the quinolizidone ring system through nitrogen ring expansion, and the addition of an arylmetallic species to the resulting lactam. This work resulted in the preparation of 2-epi-lasubine II and a formal synthesis of lasubine II.

Lasubine I (1) and its C10-epimer lasubine II (2) belong to a class of lythraceae alkaloids isolated from the leaves of

Lagerstroemia subcostata Koehne, widely distributed in the Amani islands, Taiwan, and China.¹ Although only of modest biological interest, these quinolizidine alkaloids have attracted significant interest among organic chemists for the validation of new methodologies for alkaloid synthesis. This has resulted in a number of syntheses of these alkaloids in either racemic or enantiomerically pure form.^{2,3}

We envisioned that a protected cyclopentenone **4**, a familiar precursor to prostagladins⁴ and other cyclopentane-containing natural products,⁵ would be a suitable chiral starting material for quinolizidine alkaloid synthesis (Scheme

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1). In the present case, this would require a conjugate reduction/alkylation sequence, in contrast to the conjugate addition/trapping protocol more commonly associated with such enones.^{4,6} In addition, an efficient ring adjustment sequence would also be required. Herein, we report the successful application of this strategy to lasubine alkaloid synthesis.

The TIPS-protected 4-(*S*)-(-)-hydroxy-2-cyclopentenone (4) was synthesized according to literature procedures established for the known TBS derivative. The change of protecting group was necessitated by the lability of the TBS group in a later Lewis acid-promoted step (see below). Reductive alkylation of 4 using platinum divinyltetramethyl disiloxane complex (Karstedt's catalyst) in the presence of triethylsilane afforded enol ether 5 as a single regioisomer in 96% yield (Scheme 2).6 Mukaiyama aldol reaction of 5

with aldehyde **6**⁸ followed by dehydration afforded the E enone **7** in 57% overall yield. 9,10 Catalytic hydrogenation of **7** with 10% Pd/C in ethanol resulted in reduction as well as

debenzylation to furnish alcohol **8** as a single stereoisomer (NMR) in 95% yield. NOE studies revealed the cis diastereomer of the alcohol as the predominant product.

Our original intention was to simultaneously convert the cyclopentanone into a six-membered ring while forming the second ring using the intramolecular Schmidt reaction developed in these laboratories.¹¹ To accomplish this, a modified Mitsunobu reaction on **8** afforded azide **9**; no epimerization was apparent in this step (Scheme 3).¹² When

9 was subjected to the standard Schmidt reaction conditions (TFA, BF₃•OEt₂, or TiCl₄ in CH₂Cl₂ at room temperature), no reaction was observed. Eventually, it was found that treatment with TiCl₄ in *refluxing* CH₂Cl₂ reluctantly afforded lactams **3a** and **3b** in a 1:1.3 ratio. Treatment of the diastereomerically pure 2-(4'-chlorobutyl) analogue of **9** resulted in complete epimerization at the α stereogenic center, which suggested that the formation of isomers **3a** and **3b** resulted from Lewis acid-promoted enolization and epimerization prior to the nitrogen insertion step (which has been shown to be stereoselective¹¹).

To avoid this Lewis acid-promoted epimerization, alternative ring expansion routes were considered. We have recently identified the photochemical rearrangement of endocyclic nitrones as an alternative to the intramolecular Schmidt reaction. ^{13,14} We hypothesized that such a reaction may proceed without epimerization because Lewis or protic acids are not required. Thus, alcohol 8 was converted into the bisbenzyloxylcarbonyl (Cbz)-protected hydroxylamine 10 under modified Mitsunobu conditions (Scheme 4). ¹⁵ Subsequent

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Scheme 4

8
$$\frac{\text{CbzNH}(\text{OCbz})}{\text{PPh}_3, \text{ DIAD}}$$
 $\frac{\text{PPh}_3, \text{ DIAD}}{\text{THF, 0 °C}}$
 $\frac{90\%}{90\%}$
 $\frac{\text{CbzO}}{\text{Cbz}}$
 $\frac{\text{OTIPS}}{\text{Cbz}}$
 $\frac{\text{N}}{\text{Cbz}}$
 $\frac{\text{OTIPS}}{\text{Chystoluene}}$
 $\frac{\text{N}}{\text{CH}_3\text{CN/toluene}}$
 $\frac{\text{N}}{\text{CH}_3\text{CN/toluene}}$
 $\frac{\text{N}}{\text{Chystoluene}}$
 $\frac{\text{N}}{\text{Chystoluene}}$

hydrogenation over 10% Pd—C effected both hydroxylamine deprotection and cyclization to afford the nitrone **11** directly. Photolysis of **11** gave a 13:1 mixture of **3a** and **3b** in 60% yield over two steps (GC).

A formal synthesis of lasubine II was completed as shown in Scheme 5. Thus, treatment of **3a** with a Grignard reagent derived from 4-bromoveratrole in the presence of anhydrous CeCl₃ and acidification, followed by NaBH₃CN reduction, afforded quinolizidine **12** as a single diastereomer in 85% yield. The CeCl₃ was included to prevent competing elimination of the OTIPS group. ¹⁶ The attack of hydride occurred from the pseudoaxial direction as predicted by the Stevens paradigm. ¹⁷ The relative stereochemistry of **12** was assigned on the basis of NOE experiments and Bohlmann bands in

the IR spectrum at 2786 and 2747 cm⁻¹. Finally, deprotection of TIPS group with TBAF in THF afforded (-)-2-*epi* lasubine II **13** in 87% yield. All spectroscopic and physical data of compound **13** were in agreement with the published data. This epimer can be readily converted to (-)-lasubine II via a Mitsunobu reaction as reported by Ma et al.³ⁱ

This work establishes cyclopentenone **4** as a precursor to quinolizidine alkaloids. In addition, the utility of a nitrone-based ring adjustment sequence as an alternative to the intramolecular Schmidt reaction has been demonstrated.

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Supporting Information Available: Experimental procedures and characterization of compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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