FULL PAPERS

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The Quinine Thiourea-Catalyzed Asymmetric Strecker Reaction: An Approach for the Synthesis of 3-Aminooxindoles

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Abstract: An organocatalytic enantioselective Strecker reaction for the synthesis of 3-amino-3-cya-nooxindoles has been developed. Employing a quinine-derived thiourea catalyst, the nucleophilic addition of trimethylsilyl cyanide to *N*-Boc-ketimines affords 3-amino-3-canooxindoles in good to excellent yields (78–98%) and very good enantioselectivities (up to 94%). Furthermore, to the best of our knowl-

Introduction

Optically active α -amino nitriles are synthetically versatile intermediates for a wide range of natural products and pharmaceuticals.^[1] As well-known α -amino acid precursors, these useful compounds also provide access to valuable α -amino aldehydes, ketones, β amino alcohols and 1,2-diamines. Spurred on by the ever-increasing demand for these enantioenriched compounds, the asymmetric Strecker reaction has become one of the hottest topics in recent years.^[2]



Scheme 1. Synthesis of spirohydantoin (A) from 3-amino-3cyanooxindole.

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edge, this method also represents the first enantioselective organocatalyzed Strecker reaction with *N*-Boc-ketimines as electrophiles.

Keywords: 3-amino-3-cyanooxindoles; *N*-Boc-ketimines; organocatalysis; quaternary stereocenters; Strecker reaction

Substituted 3-amino-2-oxindoles have been recognized as a core structure in a variety of bioactive molecules.^[3] Among them, one classical example is the spirohydantoin (**A**) (Scheme 1) which was developed by AstraZeneca for potential use in the treatment of pain.^[4] Experimentally, this useful compound could be achieved by the conversion of optically active 3amino-3-cyanooxindoles (Figure 1). Considering the synthetic utility of this intermediate, Zhou's^[5] and Sacchetti's^[6] groups described the asymmetric addition of TMSCN to isatin ketimines, respectively, however, both with moderate yields and stereoselectivities.

Although there are several highly efficient catalytic enantioseletive systems for Strecker reaction with ketimines as the substrate,^[7,8] to the best of our knowledge no related Strecker reactions of *N*-Boc-ketimines have been reported.^[9] As part of our ongoing interest in the stereoselective synthesis of 3-aminooxindoles through the addition of carbon nucleophiles to isatin-derived ketimines,^[10] we herein report a facile method for the construction of optically active 3-amino-3-cyanooxindoles.

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Figure 1. Catalysts for screening.

Results and Discussion

Our investigation began with the addition of TMSCN to ketimine **1a** using natural *Cinchona* alkaloids as catalysts. As expected, the reactions gave products in excellent yields respectively, however, both with lower ee. Furthermore, long reaction times, even more than 3 days, were needed (entries 1 and 2, Table 1). The catalysts quinine-thiourea L3 and quinidine-thiourea L4 were involved in the next investigation. To our delight, catalyst L3 gave satisfactory results of 98% yield and 88% ee, and the enantiomer 2a' could be achieved in a similar manner to 2a with quinidine-thiourea L4 as catalyst. As shown in Table 1, good yields and faster reaction rates were observed when catalyst L5 was used, however, with moderate stereoselectivities.

Using L3 as the selected catalyst, a reaction optimization was carried out with the addition of TMSCN to 1a as the model reaction. The initial study focused

Table 1. Catalyst screening for the enantioselective Strecker reaction.[a]



Entry	Catalyst ^[b]	Time [h]	Yield [%]	ee [%] ^[c]
1	L1	108	95	34
2	L2	96	83	18
3	L3	24	96	88
4	L4	24	96	-86
5	L5	17	94	63

[a] Reaction conditions: (0.1 mmol). **TMSCN** 1a (0.15 mmol), CH₃C₆H₅ (1 mL) and at room temperature.

[b] Catalyst loading: 10 mol%.

[c] Determined by HPLC analysis. on the influence of solvents. As presented in Table 2, although toluene, diethyl ether and 1,2-dichloroethane gave similar results (entries 1,4 and 6, Table 2), a relatively shorter reaction time was observed with 1,2-dichloroethane as solvent. When the reaction was carried out at 0°C, the enantioselectivity improved slightly compared to that at room temperature (entry 10, Table 2). The Strecker product could be obtained in 98% yield and 93% ee when the reaction temperature was decreased to -25 °C (entry 11, Table 2). Meanwhile, we also found that further lowering the temperature could not improve the enantioselectivity (entry 12, Table 2). Then, the effects of molar ratio and concentration were investigated (entries 13-16, Table 2). The results indicated that satisfactory results could be obtained with 1.5 equiv. of TMSCN and in 1 mL of solvent. Aimed to test the reactivity, another efficient source CN⁻, NCCOOC₂H₅, was also examined, but no adduct was found even after four days.



t-Bu ∼*t-*Bu 10 mol% L3 TMSCN 2a

Entry	Solvent	Time [h]	Yield [%]	ee [%] ^[b]
1	toluene	24	96	88
2	DCM	16	83	86
3	CHCl ₃	16	98	84
4	DCE	12	98	88
5	THF	40	65	80
6	Et_2O	24	91	88
7	t-BuOMe	17	89	87
8	EtOAc	17	94	68
9	CH ₃ CN	17	98	83
10 ^[c]	DCE	16	98	91
11 ^[d]	DCE	24	96	93
12 ^[e]	DCE	48	72	93
13 ^[c,f]	DCE	48	94	93
14 ^[c,g]	DCE	18	96	92
15 ^[c,h]	DCE	36	91	92
16 ^[c,i]	DCE	42	98	93
17 ^[j]	DCE	_	_	-

[[]a] Reaction conditions: **1**a (0.1 mmol), TMSCN (0.15 mmol), in 1 mL solvent and at room temperature. [b] Determined by HPLC analysis.

[d] At -25°C.

[e] At -40°C.

[f] 2 mL DCM were used.

- [g] 0.5 mL DCM was used.
- [h] TMSCN (0.13 mmol) was used.
- [i] TMSCN (0.17 mmol) was used.
- [j] CNCOOC₂H₅ was used and no adduct was found.

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[[]c]

At 0°C.

Table 3. Effect of sdditives in the reaction.^[a]



Entry	Additive	Time [h]	Yield [%]	ee [%] ^[b]
1	MeOH	18	83	93
2	EtOH	18	90	93
3	<i>i</i> -PrOH	18	96	93
4 ^[c]	<i>i</i> -PrOH	60	_	_
5 ^[d]	<i>i</i> -PrOH	18	94	-93

[a] Reaction conditions: 1a (0.1 mmol), TMSCN (0.15 mmol), in 1 mL DCE, at -25 °C and with 0.15 mmol protic additive.

^[b] Determined by HPLC analysis.

^[c] With 1 mL *i*-PrOH as solvent.

^[d] With **L4** as catalyst.

To further improve this addition reaction, a series of alcoholic additives was tested.^[11] In general, the reaction rate was improved greatly with 0.15 mmol of a protic additive. And the reactions gave the products in similar enantioselectivity (entries 1–3, Table 3). In view of the best yield, *i*-PrOH was selected as the suitable additive for further investigation of the substrate scope. Partly because of the poor solubility of ketimine **1a** in *i*-PrOH, no adduct was observed even after 60 h with *i*-PrOH as solvent. In addition, the opposite enantiomer **2a'** of compound **2a** could be obtained in 94% yield and 93% *ee* under the optimal conditions with **L4** as catalyst.^[12]

After the establishment of the optimal reaction protocol, the effect of the substituent groups at the 1position of ketimines was investigated. As listed in Table 4, with CH₃, CH₂C₆H₅ and CH₂OCH₃ as the substituent group, the addition reactions were carried out smoothly and gave the products in high yields and excellent enantioselectivities, respectively (entries 1-3, Table 4). Although similar results were seen, a significantly slower reaction rate was observed with Nbenzyl-Boc-ketimine as the substrate. It was noteworthy that the ketimine without the substituent group at the 1-position showed good reactivity in this addition reaction and gave the product 5a in 95% yield and 90% ee (entry 4, Table 4).^[13] At the same time, lost reactivity was observed with the bulky substituent Boc or an acetyl group at the 1-position of ketimine. The above-mentioned results indicated that a negative impact on the reactivity of ketimines was made by the increase of the size or electron-withdrawing ability of the substituent.

After the end of the reactivity assessment of substituent groups at the ketimine 1-position, we then
 Table 4. Effect of substituent groups at ketimine 1-position.^[a]



^{a]} Reaction conditions: **1a** (0.1 mmol), TMSCN (0.15 mmol), at -25 °C, with 0.15 mmol *i*-PrOH and with 10 mol% catalyst loading.

^[b] Determined by HPLC analysis.

^[c] The *ee* was determined by HPLC after a simple chemical conversion and see also ref.^[13]

evaluated different N-Boc-ketimines to define the scope of this Strecker reaction. As summarized in Scheme 2, the reactions of a variety of structurally diverse N-Boc-ketimines bearing electron-withdrawing, electron-neutral or electron-donating substituents in the aromatic ring, with TMSCN were tested. We were delighted to obtain the corresponding adducts for all ketimines with good to excellent yields (78–98%) under the optimized conditions. The general conclusion drawn from the results was that the steric effect on reactivity and stereoselectivity was more pronounced than the electronic effect. For example, the reactions of ketimines containing Cl or Br at the 4-position gave poorer enatioselectivities (see 2b and 2c), while the same substituents at the 6-position gave the products in higher yields and enantioselectivities (see 2i and 2j). With halogen substituents at the 7-position, the regular response was obtained (see 2k, 2l and 2m).

As the increase of the substitutent's atomic radius, a downward trend of reactivity was shown (see 2k, 2l and 2m). The effect of the electronic nature in the aromatic ring was also observed. With the change from electron-withdrawing to electron-donating group in the 5-position the corresponding reaction time increased, while the similar yields and enantioselectivities were afforded (see 2d, 2e, 2f, 2g and 2h).

In addition, it turned out that this asymmetric addition of TMSCN to disubstituted ketimine also followed the same reaction pattern. When 7-methyl-5bromoketimine was used as the substrate, the corresponding product was obtained in 96% yield and 86% *ee.* As to 5,7-dibromoketimine, 96% yield and 86% *ee* were observed.

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[a] Reaction conditions: ketimine (0.1 mmol), TMSCN (0.15 mmol), 0.15 mmol *i*-PrOH, in 1 mL DCE, at -25 °C and with 10 mol% catalyst loading.

Scheme 2. Scope of the enantioselective addition reaction.

In order to obtain the precursor of spirohydantoin (**A**), several ketimines with allyl group at the 1-position of isatin were synthesized. The result turned out that *N*-allylketimines could also be efficiently applied under the present reaction conditions. As shown in Table 5, with $-CH_2CH=CH_2$, and $-CH_2CH=CH_2C_6H_5$ as the substituent groups the addition reaction was carried out smoothly and gave the products in high yields and excellent enantioselectivities, respectively (entries 1 and 2, Table 5). To our delight, with ketimine **1r** as the substrate, the reaction gave the precursor of spirohydantoin (**A**) in 98% yield and 91% *ee*.

To explore the role of the thiourea and tertiary amine in this cyanation reaction, we first compared **Table 5.** Cyanation reaction of N-allylketimines.^[a]



[a] Reaction conditions: ketimine (0.1 mmol), TMSCN (0.15 mmol), 0.15 mmol *i*-PrOH, in 1 mL DCE, at -25°C and with 10 mol% catalyst loading.

^[b] Determined by HPLC analysis.

the differences between the L1- and L3-catalyzed cyanation reactions. As demonstrated in Table 1, with quinine L1 as catalyst, although the cyanation of ketimine 1a was carried out smoothly, the lower reaction rate was shown. By contrast, with quinine-thiourea L3 as catalyst, the reaction need only 24 h. The different reaction rate was caused by the difference of the group at the quinine 9-position. In view of this point, the extra H bond donation ability played an important role in the activation of thiourea to ketimines.





Figure 2. Conditions for exploration of the mechanism,

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In the next step, we designed several reactions to discover the role of tertiary amine. As presented in Figure 2, taking cyanation of ketimine **1a** with TMSCN as the model reaction, quinine-thiourea L3 combined with 0.15 mmol benzoic acid was employed as catalyst (conditions 1, Figure 2). Surprisingly, no adduct was observed after 72 h even at room temperature. This showed that the bonding of tertiary amine with H⁺ made the catalyst L3 lose catalytic activity. To verify this result, L6, which has no alkali atoms in the molecule was used as catalyst (conditions 2, Figure 2). However, the substrate ketimine was quantitatively recovered after 72 h. For further verification of the role of the tertiary amine, the combination of 10 mol% L6 and 10 mol% DABCO was used as catalyst (conditions 3, Figure 2). The expected adduct 2a



Figure 3. Absolute configuration of 2m.

was obtained in 60% yield. The above experiments shown that the tertiary amine played a critical role in the activation of TMSCN.

The absolute configuration of product $2\mathbf{m}$ was ascertained as (R) on the basis of an X-ray crystal structure analysis (Figure 3).^[14] These data allowed us to propose a possible mechanism for the addition of TMSCN to Boc-ketimines derived from *N*-methylisatins. As shown in Figure 4, similar to the model suggested by Feng,^[7e] the catalyst serves the dual function of activating both reaction partners. The tertiary amine of the catalyst promotes the formation of HCN, and then, through coordination, holds it in close proximity, while the thiourea moiety binds and activates the ketimine through hydrogen bonds. Following an enantioselective addition of CN to ketimine from the Re-face, the (R) configurated product was obtained.

To further demonstrate the synthetic utility of the present system, a gram-scale reaction was carried out. As shown in Scheme 3, with 2 mmol **1a** as the substrate, similar to the 0.1 mmol scale, the addition reaction gave the product **2a** in 97% yield and 93% *ee* under optimized conditions. The conversion of the cyano group into amide and ester groups, respectively, was explored next. As expected, the corresponding Boc-amino amide **2ab** and amino ester **2ac** were readily achieved without the loss of stereochemical purity.



Figure 4. Proposed catalytic cycle for the Strecker reaction of ketimines.

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Scheme 3. Demonstration of the synthetic utility.

Conclusions

In conclusion, we have developed a highly practical method for the asymmetric synthesis of 3-amino-3-canooxindoles through the addition of TMSCN to isatin-derived *N*-Boc-ketimines with the catalysis of quinine-thiourea. In addition, a variety of ketimines bearing electron-withdrawing and electron-donating substituents at different positions on the aromatic ring were tested. The reactions provided the expected products with good yields and enantioselectivities.

Experimental Section

General

All commercially available reagents were used without further purification unless otherwise stated. Reactions were monitored by thin layer chromatography (TLC), column chromatography purifications were carried out using silica gel. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded on a 300 MHz spectrometer in CDCl₃ and carbon nuclear magnetic resonance (¹³C NMR) spectra were recorded on a 75 MHz spectrometer in CDCl₃ using tetramethylsilane (TMS) as internal standard. Data are presented as follows: chemical shift, integration, multiplicity (s= singlet, d = doublet, dd = doublet of doublets, t = triplet, m =multiplet) and coupling constant in Hertz (Hz). Dichloromethane and 1,2-dichloroethane was freshly distilled from CaH₂ before use; other solvents (THF, Et₂O, 1,4-dioxane, t-BuOMe and toluene) were freshly distilled from sodium. The thiourea catalysts were prepared according to the literature procedures.[10b][15]

General Procedure for the Synthesis of *N*-Bocketimines

The synthesis of *N*-Boc-ktimines was performed according to the literature procedure.^[10b] Characterization data of three new substrates are presented as below.

(*E*)-*tert*-Butyl (1-allyl-2-oxoindolin-3-ylidene)carbamate (1p): Ketimine 1p was obtained as a yellow solid; yield: 92%; mp 80–82°C. ¹H NMR (300 MHz, CDCl₃): δ =7.66 (d, *J*=6.9 Hz, 1H), 7.46 (td, *J*=7.8, 1.0 Hz, 1H), 7.09 (t, *J*= 7.5 Hz, 1H), 6.84 (d, *J*=7.9 Hz, 1H), 5.89–5.76 (m, 1H), 5.33–5.26 (m, 2H), 4.33 (d, *J*=4.0 Hz, 2H), 1.63 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ =160.5, 156.9, 153.1, 147.4, 135.3, 130.5, 124.4, 123.5, 119.3, 118.5, 110.2, 83.5, 42.5, 28.0; IR (KBr): v = 3434, 2980, 2933, 1740, 1722, 1681, 1613, 1469, 1371, 1350, 1249, 1164, 1146, 1097, 938, 763 cm⁻¹; HR-MS: m/z = 309.1209, calcd. for $C_{16}H_{18}N_2NaO_3$ [M+Na]⁺: 309.1210.

(*E*)-*tert*-Butyl (1-cinnamyl-2-oxoindolin-3-ylidene)carbamate (1q): Ketimine 1q was obtained as a yellow solid; yield: 91%; mp 130–132 °C.¹H NMR (300 MHz, CDCl₃): $\delta =$ 7.67 (d, *J*=7.0 Hz, 1H), 7.45 (t, *J*=7.4 Hz, 1H), 7.36–7.25 (m, 5H), 7.09 (t, *J*=7.3 Hz, 1H), 6.90 (d, *J*=7.8 Hz, 1H), 6.64 (d, *J*=15.9 Hz, 1H), 6.16 (dt, *J*=15.9, 6.0 Hz, 1H), 4.50 (d, *J*=5.2 Hz, 2H), 1.64 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta =$ 160.5, 157.0, 153.1, 147.4, 135.8, 135.4, 133.8, 128.7, 128.2, 126.5, 124.4, 123.5, 121.7, 119.4, 110.2, 83.6, 42.1, 28.1; IR (KBr): v =3432, 2985, 2935, 2361, 1740, 1719, 1680, 1618, 1470, 1348, 1251, 1154, 1098, 970, 750 cm⁻¹; HR-MS: *m/z*=385.1523, calcd. for C₂₂H₂₂N₂NaO₃ [M+Na]⁺: 385.1523.

(*E*)-tert-Butyl {1-[(*E*)-3-(3,4-dichlorophenyl)allyl]-2-oxoindolin-3-ylidene}carbamate (1r): Ketimine 1r was obtained as a yellow solid; yield: 94%; mp 80–82 °C. ¹H NMR (300 MHz, CDCl₃); δ =7.69 (d, *J*=7.1 Hz, 1H), 7.47 (td, *J*= 7.8, 1.1 Hz, 1H), 7.42 (d, *J*=2.0 Hz, 1H), 7.37 (d, *J*=8.3 Hz, 1H), 7.18–7.09 (m, 2H), 6.87 (d, *J*=7.9 Hz, 1H), 6.53 (d, *J*= 16.0 Hz, 1H), 6.18 (dt, *J*=16.0, 5.8 Hz, 1H), 4.49 (d, *J*= 4.9 Hz, 2H), 1.64 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ = 160.4, 157.0, 152.9, 147.1, 135.9, 135.4, 132.8, 131.8, 131.2 130.6, 128.2, 125.6, 124.6, 123.9 123.7 119.4, 109.9, 83.7 41.8, 28.0; IR (KBr): v=3436, 2980, 1738, 1350, 1253, 1151, 1100, 965, 752 cm⁻¹; HR-MS: *m*/*z*=453.0739, calcd. for C₂₂H₂₀Cl₂N₂NaO₃ [M+Na]⁺: 453.0743.

General Procedure for the Strecker Reaction of TMSCN with *N*-Boc-ketimines

In a 5-mL vial, *N*-Boc-ketimine (0.1 mmol) and quinine-derived thiourea catalyst **L3** 5.9 mg (0.01 mmol, 10 mol%) were placed. After an injection of 1,2-dichloroethane (1 mL), the reaction mixture was cooled to -25 °C, then, 20 µL (0.15 mmol) TMSCN were added, followed by the addition of 11.5 µL *i*-PrOH. After that the mixture was stirred until complete disappearance of the ketimine. The mixture was purified by flash chromatography (silica gel, hexane/ ethyl acetate) and afforded the resulting products as described below.

(*R*)-tert-Butyl (3-cyano-1-methyl-2-oxoindolin-3-yl)carbamate (2a): According to the general procedure, compound 2a was obtained as a pale yellow solid; yield: 27.6 mg (96%); mp 176–177 °C. 93%. The *ee* was determined by

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HPLC analysis (Daicel Chiralcel AD-H column, hexane/2propanol 60: 40, 1.0 mL min⁻¹); retention times: t_{major} =10.4 and t_{minor} =14.2 min. [α]_D²⁶: +68 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ =7.84 (d, *J*=7.5 Hz, 1H), 7.44 (td, *J*= 7.8, 1.2 Hz, 1H), 7.19 (td, *J*=7.5, 0.9 Hz, 1H), 6.91 (d, *J*= 7.8 Hz, 1H), 5.62 (s, 1H), 3.29 (s, 3H), 1.41 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ =167.9, 153.6, 143.0, 131.4, 126.2, 124.7, 124.3, 114.6, 109.3, 82.3, 54.6, 28.1, 27.3; IR (KBr): v=3315, 2978, 2927, 2247, 1742, 1719, 1612, 1492, 1473, 1369, 1255, 1160, 755 cm⁻¹; HR-MS (ESI): *m/z*= 310.1164, calcd. for C₁₅H₁₇N₃NaO₃ [M+Na]⁺: 310.1168.

(S)-tert-Butyl (3-cyano-1-methyl-2-oxoindolin-3-yl)carbamate (2a'): According to the general procedure, except for using L4 as the catalyst, compound 2a' was obtained as a pale yellow solid; yield: 27.0 mg (94%); mp 157-158°C. 93% ee was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 60: 40, 1.0 mL min⁻¹): retention times: $t_{\text{major}} = 10.5$ and $t_{\text{minor}} = 14.4$ min. $[\alpha]_{\text{D}}^{26}$: -66 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.83$ (d, J =7.4 Hz, 1 H), 7.44 (td, J=7.8, 1.3 Hz, 1 H), 7.18 (td, J=7.7, 1.0 Hz, 1 H), 6.91 (d, J=7.8 Hz, 1 H), 5.62 (s, 1 H), 3.29 (s, 3H), 1.41 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.9$, 153.6, 143.0, 131.4, 126.2, 124.7, 124.3, 114.6, 109.3, 82.3, 54.6, 28.1, 27.3; IR (KBr): v=3310, 2977, 2928, 2247, 1741, 1719, 1612, 1492, 1473, 1369, 1254, 1159, 755 cm⁻¹; HR-MS (ESI): m/z = 310.1163, calcd. for $C_{15}H_{17}N_3NaO_3$ [M+Na]⁺: 310.1162.

(*R*)-*tert*-Butyl (1-benzyl-3-cyano-2-oxoindolin-3-yl)carbamate (3a): According to the general procedure, compound 3a was obtained as a yellow solid; yield: 35.6 mg (98%); mp 190–192 °C. 93% *ee* was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 96: 4, 1.0 mLmin⁻¹); retention times: $t_{major} = 19.9$ and $t_{minor} =$ 18.5 min. $[\alpha]_{D}^{26}$: +46 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.83$ (d, J = 7.5 Hz, 1H), 7.37–7.27 (m, 6H), 7.13 (td, J = 7.7, 0.9 Hz, 1H), 6.76 (d, J = 7.8 Hz, 1H), 5.72 (s, 1H), 4.96 (s, 2H), 1.43 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) $\delta = 168.3$, 153.6, 142.2, 134.2, 131.2, 129.0, 128.1, 127.2, 126.3, 124.7, 124.3, 114.6, 110.3, 82.4, 54.8, 44.9, 28.1; IR (KBr): v =3218, 2979, 2928, 2247, 1720, 1611, 1487, 1370, 1278, 1255, 1160, 754, 700 cm⁻¹; HR-MS (ESI): m/z = 386.1481, calcd. for C₂₁H₂₁N₃NaO₃ [M+Na]⁺: 386.1481.

(R)-tert-Butyl [3-cyano-1-(methoxymethyl)-2-oxoindolin-**3-yl]carbamate (4a):** According to the general procedure compound 4a was obtained as a pale yellow solid; yield: 27.9 mg (88%); mp 149-151 °C. 91% ee was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2propanol 95: 5, 1.0 mLmin⁻¹): retention times: $t_{\text{major}} = 11.3$ and $t_{\text{minor}} = 14.6 \text{ min.} [\alpha]_{D}^{26}$: +62 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.75$ (d, J = 7.5 Hz, 1 H), 7.43 (td, J =7.8, 1.2 Hz, 1 H), 7.21 (td, J = 7.7, 0.9 Hz, 1 H), 7.11 (d, J =7.9 Hz, 1 H), 5.79 (s, 1 H), 5.19 (s, 2 H), 3.40 (s, 3 H), 1.40 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 168.7$, 153.3, 141.4, 131.5, 125.6, 124.7, 124.1, 114.5, 110.8, 82.4, 72.4, 56.8, 55.1, 28.1. IR (KBr): υ=3220, 2979, 2932, 2248, 1752, 1719, 1612, 1487, 1368, 1343, 1282, 1252, 1160, 1097, 756 cm⁻¹; HR-MS (ESI): m/z = 340.1272, calcd. for $C_{16}H_{19}N_3NaO_4$ [M+Na]⁺: 340.1273.

(*R*)-*tert*-Butyl (3-cyano-2-oxoindolin-3-yl)carbamate (5a): According to the general procedure compound 5a was obtained as a yellow solid; yield: 26.0 mg (95%); mp 88–90 °C. $[\alpha]_{D}^{26}$: +55 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta =$ 8.73 (s, 1H), 7.72 (d, J=7.6 Hz, 1H), 7.35 (td, J=7.8, 1.2 Hz, 1H), 7.15 (td, J=7.7, 0.9 Hz, 1H), 6.91 (d, J=7.8 Hz, 1H), 5.91 (s, 1H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 169.6$, 153.6, 140.3, 131.4, 125.9, 125.0, 124.2, 114.3, 111.2, 82.7, 55.0, 28.1; IR (KBr): $\upsilon = 3289$, 2979, 2928, 2249, 1746, 1620, 1474, 1369, 1254, 1158, 753 cm⁻¹; HR-MS (ESI): m/z = 296.1006, calcd. for C₁₄H₁₅N₃NaO₃ [M+Na]⁺: 296.1006.

(*R*)-*tert*-Butyl (4-chloro-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2b): According to the general procedure compound 2b was obtained as a white solid; yield: 31.2 mg (97%); mp 194–195 °C. 62% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mL min⁻¹): retention times: $t_{major} = 22.0$ and $t_{minor} =$ 19.6 min. $[\alpha]_D^{26}$: +38 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.38$ (t, J = 8.1 Hz, 1H), 7.10 (dd, J = 8.3, 0.5 Hz, 1H), 6.82 (dd, J = 7.9, 0.5 Hz, 1H), 5.89 (s, 1H), 3.30 (s, 3H), 1.32 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) $\delta = 167.1$, 152.4, 145.2, 132.4, 131.2, 124.4, 120.3, 112.8, 107.7, 82.3, 54.9, 28.0, 27.6; IR (KBr): v = 3307, 2979, 2927, 2249, 1751, 1707, 1607, 1459, 1290, 1122, 1033, 764 cm⁻¹. HR-MS (ESI): m/z = 344.0778, calcd. for C₁₅H₁₆ClN₃NaO₃ [M+Na]⁺: 344.0778.

(*R*)-*tert*-Butyl (4-bromo-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2c): According to the general procedure compound 2c was obtained as a yellow solid; yield: 28.6 mg (78%); mp 211–212 °C. 60% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention time: $t_{major} = 27.0$ and $t_{minor} =$ 23.2 min. $[\alpha]_D^{26}$: +29 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.34-7.25$ (m, 2H), 6.86 (dd, J = 7.2, 1.4 Hz, 1H), 5.92 (s, 1H), 3.30 (s, 3H), 1.32 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.0$, 152.3, 145.4, 132.6, 127.4, 122.0, 119.2, 112.7, 108.3, 82.3, 56.0, 28.0, 27.6; IR (KBr): v = 3306, 2979, 2926, 2247, 1747, 1708, 1603, 1457, 1363, 1287, 1157, 1113, 1025, 781, 756 cm⁻¹; HR-MS (ESI): m/z = 388.0271, calcd. for C₁₃H₁₆BrN₃NaO₃ [M+Na]⁺: 388.0273.

(*R*)-*tert*-Butyl (3-cyano-1-methyl-5-nitro-2-oxoindolin-3yl)carbamate (2d): According to the general procedure compound 2d was obtained as a yellow solid; yield: 31.2 mg (94%); mp 94–95°C. 89% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹); retention times: t_{major} =12.4 and t_{minor} =20.3 min. [α]_D²⁶: +50 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ =8.62 (d, *J*=2.2 Hz, 1H), 8.42 (dd, *J*= 8.7, 2.3 Hz, 1H), 7.05 (d, *J*=8.7 Hz, 1H), 5.85 (d, *J*=7.9 Hz, 1H), 3.39 (s, 3H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ =168.1, 153.2, 148.5, 144.4, 128.1, 125.2, 121.5, 113.3, 109.2, 83.2, 54.2, 28.1, 27.9; IR (KBr): v=3320, 2979, 2927, 2374, 1756, 1719, 1528, 1493, 1340, 1159, 756 cm⁻¹; HR-MS (ESI): *m*/*z*=355.1019, calcd. for C₁₅H₁₆N₄NaO₅ [M+Na]⁺: 355.1018.

(*R*)-*tert*-Butyl (5-bromo-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2e): According to the general procedure compound 2e was obtained as a pale yellow solid; yield: 35.9 mg (98%); mp 175–177°C. 92% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention times: t_{major} =8.4 and t_{minor} = 13.4 min. [α]_D²⁶: +54 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ =7.96 (s, 1H), 7.57 (dd, *J*=8.4, 2.0 Hz, 1H), 6.80 (d, *J*=8.4 Hz, 1H), 5.65 (s, 1H), 3.28 (s, 3H), 1.43 (s, 9H). ¹³C NMR (75 MHz, CDCl₃): δ =167.4, 153.5, 142.1, 134.3,

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129.4, 126.2, 116.8, 114.0, 110.8, 82.7, 54.4, 28.1, 27.5. IR (KBr): v=3315, 2977, 2927, 2249, 1747, 1609, 1487, 1358, 1255, 1160, 759 cm⁻¹; HR-MS (ESI): m/z=388.0271, calcd. for C₁₅H₁₆BrN₃NaO₃ [M+Na]⁺: 388.0273.

(*R*)-*tert*-Butyl (3-cyano-1,5-dimethyl-2-oxoindolin-3-yl)carbamate (2f): According to the general procedure compound 2f was obtained as a yellow solid; yield: 29.5 mg (98%); mp 204–205 °C. 90% *ee* was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention times: $t_{major} = 6.8$ and $t_{minor} =$ 6.2 min. [α]_D²⁶: +75 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.66$ (s, 1H), 7.26–7.19 (m, 1H), 6.79 (d, J =8.0 Hz, 1H), 5.57 (s, 1H), 3.26 (s, 3H), 2.37 (s, 3H), 1.44 (d, J = 9.6 Hz, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.8$, 153.6, 140.6, 134.2, 131.6, 126.8, 124.7, 114.7, 109.0, 82.2, 54.7, 28.1, 27.3, 21.1; IR (KBr): v = 3313, 2955, 2925, 2246, 1741, 1720, 1610, 1501, 1364, 1253, 1162 cm⁻¹; HR-MS (ESI): m/z = 324.1324, calcd. for C₁₆H₁₉N₃NaO₃ [M+Na]⁺: 324.1324.

(*R*)-*tert*-Butyl (3-cyano-5-methoxy-1-methyl-2-oxoindolin-3-yl)carbamate (2g): According to the general procedure compound 2g was obtained as a white solid; yield: 29.5 mg (93%); mp 158–159°C. 93% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mL min⁻¹): retention times: $t_{major} = 17.2$ and $t_{minor} =$ 14.1 min. $[\alpha]_D^{26}$: +67 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.49$ (s, 1H), 6.96 (dd, J = 8.6, 2.6 Hz, 1H), 6.82 (d, J = 8.6 Hz, 1H), 5.64 (s, 1H), 3.82 (s, 3H), 3.26 (s, 3H), 1.43 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.6$, 157.0, 153.7, 136.2, 125.8, 116.3, 114.6, 112.9, 109.9, 82.3, 56.0, 54.9, 28.1, 27.4; IR (KBr): v = 3312, 2977, 2930, 2248, 1738, 1607, 1500, 1366, 1285, 1162, 756 cm⁻¹. HR-MS (ESI): m/z =340.1273, calcd. for $C_{16}H_{19}N_3NaO_4$ [M+Na]⁺: 340.1273.

(R)-tert-Butyl [3-cyano-1-methyl-2-oxo-5-(trifluoromethoxy)indolin-3-yl]carbamate (2h): According to the general procedure compound 2h was obtained as a pale yellow solid; yield: 35.6 mg (96%); mp 133-134 °C. 90% ee was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention times: $t_{\text{major}} = 5.4$ and $t_{\text{minor}} = 8.3 \text{ min.} [\alpha]_{\text{D}}^{26}$: +55 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.78$ (s, 1 H), 7.33 (ddd, J=8.6, 2.4, 0.8 Hz, 1 H), 6.93 (d, J=8.6 Hz, 1 H), 5.71 (s, 1H), 3.31 (s, 3H), 1.43 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.7$, 145.6, 141.7, 125.9, 124.6, 122.1, 120.4, 118.7, 113.9, 110.0, 82.8, 54.5, 28.0, 27.5; IR (KBr): v=3314, 2980, 2928, 2250, 1748, 1721, 1621, 1498, 1368, 1258, 1218, 1165, 759 cm⁻¹; HR-MS (ESI): m/z = 394.0990, calcd. for $C_{16}H_{16}F_3N_3NaO_4 [M + Na]^+: 394.0991.$

(*R*)-*tert*-Butyl (6-chloro-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2i): According to the general procedure compound 2i was obtained as a yellow solid; yield: 31.5 mg (98%); mp 148–150 °C. 94% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mL min⁻¹): retention times: $t_{major} = 9.5$ and $t_{minor} =$ 15.4 min. $[\alpha]_D^{26}$: +65 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.75$ (d, J = 8.1 Hz, 1H), 7.16 (dd, J = 8.1, 1.9 Hz, 1H), 6.92 (d, J = 1.8 Hz, 1H), 5.69 (s, 1H), 3.28 (s, 3H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.9$, 153.5, 144.2, 137.4, 127.2, 124.2, 122.9, 114.1, 110.1, 82.6, 54.2, 28.1, 27.5; IR (KBr): v = 3318, 2979, 2928, 2249, 1747, 1720, 1610, 1493, 1368, 1253, 1160, 1073, 736 cm⁻¹; HR-MS (ESI): m/z =344.0777, calcd. for C₁₅H₁₆ClN₃NaO₃ [M+Na]⁺: 344.0778. (*R*)-*tert*-Butyl (6-bromo-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2j): According to the general procedure compound 2j was obtained as a yellow solid; yield: 35.9 mg (98%); mp 90–91°C. 93% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention times: $t_{major} = 8.4$ and $t_{minor} =$ 11.7 min. $[\alpha]_D^{26}$: +59 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.69$ (d, J = 8.0 Hz, 1H), 7.32 (dd, J = 8.1, 1.7 Hz, 1H), 7.07 (d, J = 1.7 Hz, 1H), 5.65 (s, 1H), 3.28 (s, 3H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.8$, 153.5, 144.3, 127.5, 127.2, 125.3, 123.4, 114.0, 112.9, 82.6, 54.3, 28.1, 27.5; IR (KBr): v = 3316, 2978, 2928, 2249, 1748, 1605, 1491, 1367, 1253, 1159, 759 cm⁻¹; HR-MS (ESI): m/z = 388.0272, calcd. for C₁₅H₁₆BrN₃NaO₃ [M+Na]⁺: 388.0273.

(*R*)-*tert*-Butyl (3-cyano-7-fluoro-1-methyl-2-oxoindolin-3yl)carbamate (2k): According to the general procedure compound 2k was obtained as a white solid; yield: 29.9 mg (98%); mp 124–125 °C. 91% *ee* was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 90: 10, 1.0 mLmin⁻¹); retention times: $t_{major} = 6.8$ and $t_{minor} =$ 7.7 min. [α]_D²⁶: +63 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.58$ (d, J = 7.0 Hz, 1 H), 7.21–7.08 (m, 2 H), 5.71 (s, 1 H), 3.50 (d, J = 2.8 Hz, 3 H), 1.41 (s, 9 H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.6$, 153.3, 149.4, 146.2, 130.0, 129.9, 127.0, 124.9, 124.8, 121.6, 119.5, 119.2, 114.1, 82.5, 54.6, 30.0, 29.9, 28.1; IR (KBr): v = 3319, 2980, 2929, 2251, 1750, 1720, 1631, 1487, 1367, 1246, 1161, 758, 734 cm⁻¹; HR-MS (ESI): m/z = 328.1074, calcd. for C₁₅H₁₆FN₃NaO₃ [M+Na]⁺: 328.1073.

(*R*)-*tert*-Butyl (7-chloro-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2l): According to the general procedure compound 2l was obtained as a yellow solid; yield: 31.5 mg (98%); mp 153–154°C. 91% *ee* was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 90: 10, 1.0 mLmin⁻¹): retention times: $t_{major} = 6.8$ and $t_{minor} =$ 7.9 min. [α]_D²⁶: +85 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.69$ (d, J = 7.3 Hz, 1H), 7.36 (dd, J = 8.3, 1.2 Hz, 1H), 7.10 (dd, J = 8.2, 7.6 Hz, 1H), 5.75 (s, 1H), 3.66 (s, 3H), 1.40 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 168.3$, 153.3, 139.1, 133.6, 127.1, 124.9, 124.3, 116.6, 114.0, 82.6, 54.4, 31.0, 28.1; IR (KBr): v = 3320, 2979, 2928, 2250, 1747, 1607, 1464, 1365, 1255, 1160, 1109, 759, 737 cm⁻¹; HR-MS (ESI): m/z = 344.0777, calcd. for C₁₅H₁₆ClN₃NaO₃ [M+Na]⁺: 344.0778.

(*R*)-*tert*-Butyl (7-bromo-3-cyano-1-methyl-2-oxoindolin-3yl)carbamate (2m): According to the general procedure compound 2m was obtained as a white solid; yield: 35.9 mg (98%); mp 163–164°C. 91% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 80: 20, 1.0 mLmin⁻¹): retention times: t_{major} =7.6 and t_{minor} = 9.2 min. [α]_D²⁶: +50 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ =7.73 (d, *J*=7.2 Hz, 1H), 7.54 (dd, *J*=8.2, 1.2 Hz, 1H), 7.03 (dd, *J*=8.1, 7.6 Hz, 1H), 5.77 (s, 1H), 3.67 (s, 3H), 1.39 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ =168.5, 153.3, 140.5, 137.0, 127.4, 125.3, 124.8, 114.0, 103.3, 82.6, 54.3, 31.2, 28.1; IR (KBr): v=3320, 2956, 2927, 2250, 1747, 1606, 1459, 1365, 1255, 1160, 1105, 737 cm⁻¹; HR-MS (ESI): *m*/*z*=388.0273, calcd. for C₁₅H₁₆BrN₃NaO₃ [M+Na]⁺: 388.0273.

(*R*)-*tert*-Butyl (5-bromo-3-cyano-1,7-dimethyl-2-oxoindolin-3-yl)carbamate (2n): According to the general procedure compound 2n was obtained as a pale yellow solid; yield:

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36.5 mg (96%); mp 204–205 °C. 86% *ee* was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2propanol 80: 20, 1.0 mLmin⁻¹): retention times: t_{major} =7.0 and t_{minor} =10.0 min. [α]_D²⁸: +52 (*c*=1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ =7.75 (d, *J*=1.5 Hz, 1H), 7.32 (dd, *J*= 2.0, 0.6 Hz, 1H), 5.60 (s, 1H), 3.54 (s, 3H), 2.55 (s, 3H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ =168.2, 153.3, 140.0, 137.5, 126.8, 126.7, 122.9, 116.5, 114.1, 82.6, 54.2, 30.9, 28.1, 18.8; IR (KBr): v=32316, 2957, 2926, 2250, 1740, 1720, 1602, 1463, 1342, 1257, 1161, 745 cm⁻¹; HR-MS (ESI): *m/z* = 402.0432, calcd. for C₁₆H₁₈BrN₃NaO₃ [M+Na]⁺: 402.0424.

(*R*)-*tert*-Butyl (5,7-dibromo-3-cyano-1-methyl-2-oxoindolin-3-yl)carbamate (20): According to the general procedure compound 20 was obtained as a white solid; yield: 43.2 mg (97%); mp 205–206°C. 87% *ee* was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 95: 5, 1.0 mLmin⁻¹); retention times: $t_{major} = 11.5$ and $t_{minor} =$ 13.5 min. [α]_D²⁶: +59 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.83$ (d, J = 1.3 Hz, 1H), 7.71 (d, J = 1.9 Hz, 1H), 5.77 (s, 1H), 3.65 (s, 3H), 1.42 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ 168.0, 153.2, 139.79, 138.9, 128.5, 127.9, 117.0, 113.5, 103.9, 83.0, 54.2, 31.2, 28.1; IR (KBr): v = 3312, 2956, 2925, 2377, 1750, 1719, 1460, 1370, 1335, 1253, 1159, 1112, 759, 740 cm⁻¹; HR-MS (ESI): m/z = 465.9375, calcd. for $C_{15}H_{15}Br_2N_3NaO_3$ [M+Na]⁺: 465.9372.

(1-allyl-3-cyano-2-oxoindolin-3-yl)carba-(R)-tert-Butyl mate (2p): According to the general procedure compound 2p was obtained as a pale yellow solid; yield: 27.5 mg (96%); mp 138-140°C. 91% ee was determined by HPLC analysis (Daicel Chiralcel OJ-H column, hexane/2-propanol 90:10, 1.0 mL min⁻¹): retention times: $t_{major} = 10.1$ and $t_{minor} =$ 12.5 min. $[\alpha]_D^{23}$: +50 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.85$ (d, J = 7.5 Hz, 1 H), 7.40 (td, J = 7.8, 1.2 Hz, 1 H), 7.17 (td, J = 7.7, 0.8 Hz, 1 H), 6.90 (d, J = 7.9 Hz, 1 H), 5.91-5.78 (m, 1H), 5.65 (s, 1H), 5.35-5.27 (m, 2H), 4.39 (d, J=1.8 Hz, 2 H), 1.43 (s, 10 H); ¹³C NMR (75 MHz, CDCl₃): δ 167.8, 153.6, 142.2, 131.2, 129.9, 126.3, 124.7, 124.2, 118.6, 114.6, 110.1, 82.3, 54.6, 43.4, 28.1; IR (KBr): v=3295, 2957, 2925, 1739, 1711, 1611, 1487, 1466, 1369, 1280, 1254, 1160, 1019, 755 cm⁻¹; HR-MS (ESI): m/z = 336.1321, calcd for $C_{17}H_{19}N_3NaO_3 [M+Na]^+: 336.1319.$

(R,E)-tert-Butyl (1-cinnamyl-3-cyano-2-oxoindolin-3-yl)carbamate (2q): According to the general procedure compound 2q was obtained as a colorless amorphous solid; yield: 34.7 mg (96%); 92% ee was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 90:10, 1.0 mL min⁻¹): retention times: $t_{\text{major}} = 23.0$ and $t_{\text{minor}} =$ 26.1 min. $[\alpha]_{D}^{23}$: +40 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.84$ (d, J = 7.5 Hz, 1 H), 7.41–7.24 (m, 6 H), 7.17 (t, J=7.6 Hz, 1H), 6.95 (d, J=7.9 Hz, 1H), 6.68 (d, J=16.0 Hz, 1 H), 6.18 (dt, J=15.9, 5.8 Hz, 1 H), 5.69 (s, 1 H), 4.54 (d, J = 5.5 Hz, 2H), 1.43 (s, 9H); ¹³C NMR (75 MHz, $CDCl_3$): δ 167.8, 153.6, 142.3, 135.8, 133.9, 131.3, 128.6, 128.2, 126.6, 126.3, 124.7, 124.3, 121.1, 114.6, 110.2, 82.4, 54.7, 43.1, 28.1; IR (KBr): v = 3408, 3311, 2958, 2925, 1744, 1711, 1611, 1487, 1467, 1364, 1281, 1223, 1162, 1018, 755,531 cm⁻¹; HR-MS (ESI): m/z = 412.1637, calcd. for $C_{23}H_{23}N_3NaO_3 [M + Na]^+: 412.1632.$

(*R*,*E*)-*tert*-Butyl {3-cyano-1-[3-(3,4-dichlorophenyl)allyl]-2-oxoindolin-3-yl}carbamate (2r): According to the general procedure compound 2r was obtained as a colorless amorphous solid; yield: 42.1 mg (98%). 91% *ee* was determined by HPLC analysis (Daicel Chiralcel IA column, hexane/2propanol 60:40, 1.0 mL min⁻¹): retention times: $t_{major} = 6.8$ and $t_{minor} = 6.1$ min. $[\alpha]_D^{23}$: +15 (*c* 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.79$ (d, J = 7.5 Hz, 1H), 7.43–7.34 (m, 3H), 7.19 (t, J = 7.8 Hz, 2H), 6.91 (d, J = 7.9 Hz, 1H), 6.60 (d, J = 16.0 Hz, 1H), 6.20 (dt, J = 16.0, 5.4 Hz, 1H), 5.78 (s, 1H), 4.55 (qd, J = 16.6, 5.0 Hz, 2H), 1.43 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 167.9$, 153.4, 142.1, 136.0, 132.7, 131.7, 131.4, 131.0, 130.5, 128.4, 125.9, 125.6, 124.6, 124.4, 123.1, 114.5, 109.9, 100.0, 82.4, 54.8, 42.7, 28.1; IR (KBr): v = 3318, 2957, 2926, 2853, 1741, 1718, 1612, 1471, 1369, 1279, 1255, 1160, 1026, 755 cm⁻¹; HR-MS (ESI): m/z =480.0860, calcd. for C₂₃H₂₁Cl₂N₃NaO₃ [M+Na]⁺: 480.0852.

Synthesis of 2ab from 2a

In a 25-mL round-bottom flask, 143.5 mg 2a (0.5 mmol) were placed. After an injection of 5 mL acetone, the mixture was stirred for 10 min. Then, 1.5 mL aqueous solution of Na₂CO₃ (1N, 3 equiv.) was added, followed by the addition of 1.6 mL of H_2O_2 (30%, 27 equiv.). After that the mixture was stirred for 2 h. After an evaporation of the volatile organic solvents, the resulting mixture was extracted with CH_2Cl_2 (3×10 mL). The combined organic layers were dried over Na₂SO₄. After filtration and evaporation of the solvent, the crude residue was purified by flash chromatography (silica gel, hexane/ethyl acetate), giving the product 2ab as a white solid; yield: 128 mg (84%); mp 163-164 °C. 93% ee, as determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 90:10, 1.0 mLmin⁻¹): retention times: $t_{\text{major}} = 28.5^{\circ}$ and $t_{\text{minor}} = 23.7 \text{ min.} [\alpha]_{\text{D}}^{26}$: -14 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.38$ (td, J = 7.7, 1.2 Hz, 1H), 7.30 (d, J=7.4 Hz, 1H), 7.12 (t, J=7.5 Hz, 1H), 6.92 (d, J=7.8 Hz, 1H), 6.39 (s, 1H), 5.82 (s, 1H), 5.53 (s, 1H), 3.27 (s, 3H), 1.25 (s, 9H); ¹³C NMR (75 MHz, $CDCl_3$): $\delta = 172.6, 167.0, 153.5, 144.0, 130.2, 127.5, 123.7,$ 123.6, 109.0, 80.6, 66.1, 28.0, 27.0; IR (KBr): v=3385, 3326, 3195, 2958, 2926, 2291, 1718, 1695, 1617, 1471, 1367, 1250, 1164, 1125, 753 cm⁻¹; HR-MS (ESI): m/z = 328.1272, calcd. for $C_{15}H_{19}N_3NaO_4$ [M + Na]⁺: 328.1268,.

Synthesis of 2ac from 2a

In a 25-mL round-bottom flask, 143.5 mg 2a (0.5 mmol) were placed. After an injection of 5 mL absolute methanol, the mixture was cooled to 0°C. Then, dried HCl was bubbled slowly. This process lasted for 15 min. After that, the resulting solution stirred at room temperature over ight. After an evaporation of the volatile organic solvents, 5 mL water were added followed by the addition of 2 mL saturated aqueous solution of NaHCO3. The resulting mixture was extracted with CH_2Cl_2 (3×10 mL). The combined organic layers were dried over Na₂SO₄. After filtration and evaporation of the solvent, the crude residue was purified by flash chromatography (silica gel, hexane/ethyl acetate), compound 2ac was obtained as a yellow solid; yield: 103.5 mg (94%); mp 109-110°C. 93% ee was determined by HPLC analysis (Daicel Chiralcel AD-H column, hexane/2-propanol 60: 40, 1.0 mLmin⁻¹): retention times: $t_{\text{major}} = 8.9$ and $t_{\text{minor}} =$ 10.5 min. $[\alpha]_{D}^{26}$: -122 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.37$ (td, J = 7.8, 1.2 Hz, 1 H), 7.30 (dd, J = 7.4, 0.7 Hz, 1H), 7.09 (td, J=7.6, 0.9 Hz, 1H), 6.89 (d, J=7.8 Hz, 1H), 3.68 (s, 3H), 3.25 (s, 3H), 2.28 (s, 2H);

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¹³C NMR (75 MHz, CDCl₃): δ = 174.7, 170.7, 144.2, 130.1, 128.9, 123.29, 123.26, 108.8, 65.2, 53.3, 26.7; IR (KBr): v = 3372, 3304, 2955, 2926, 1749, 1720, 1612, 1494, 1470, 1374, 1242, 1126, 1083, 1001, 755 cm⁻¹; HR-MS (ESI): m/z = 243.0746, calcd. for C₁₁H₁₂N₂NaO₃ [M+Na]⁺: 243.0746.

Synthesis of 5ab from 5a

In a 5-mL round-bottom flask, 10 mg NaH (0.25 mmol, 60% in mineral oil, 2.5 equiv.) were placed, after an injection of 0.5 mL THF, the mixture was cooled to 0°C, 26 mg compound 5a in 1 mL THF were dropwise added. The resulting solution was stirred 15 min, and 20 µL MOMCl were added. After that, the resulting solution stirred at room temperature overnight. Then, the solution was neutralized by the addition of a saturated aqueous solution of NH₄Cl and then the volatile organic solvents were removed under vacuum. The aqueous layer was extracted with CH_2Cl_2 (3×10 mL). The organic layer obtained was dried over Na₂SO₄, filtered and concentrated under reduced pressure. The crude residue was purified by flash chromatography (silica gel, hexane/ ethyl acetate), compound 5ab was obtained as a colorless oil; yield: 23 mg (64%). 90% ee was determined by HPLC analysis (Daicel Chiralcel OD-H column, hexane/2-propanol 95: 5, 1.0 mL min⁻¹): retention times: $t_{major} = 7.3$ and $t_{minor} =$ 8.3 min. $[\alpha]_{D}^{26}$: -5 (c 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 7.53$ (d, J = 7.5 Hz, 1 H), 7.42 (td, J = 7.8, 1.2 Hz, 1H), 7.18 (t, J=7.3 Hz, 1H), 7.10 (d, J=7.9 Hz, 1H), 5.31 (d, J=11.1 Hz, 1 H), 5.24 (d, J=11.0 Hz, 1 H), 5.13 (s, 1 H),5.05 (d, J=11.1 Hz, 1 H), 3.48 (s, 3 H), 3.43 (s, 3 H), 1.26 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 168.9$, 152.2, 141.6, 135.8, 131.0, 124.4, 124.3, 113.7, 110.4, 83.4, 78.6, 72.5, 60.6, 56.8, 56.0, 27.8; IR (KBr): v=3288, 2955, 2925, 1752, 1711, 1610, 1465, 1369, 1298, 1154, 1095, 757 cm⁻¹; HR-MS (ESI): m/z = 384.1530, calcd. for $C_{18}H_{23}N_3NaO_5$ [M+Na]⁺: 384.1530.

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References

- [1] For a review, see: D. Enders, J. P. Shilvock, *Chem. Soc. Rev.*, **2000**, *29*, 359–373.
- [2] For reviews, see: a) L. Yet, Angew. Chem. 2001, 113, 900–902; Angew. Chem. Int. Ed. 2001, 40, 875–877;
 b) H. Gröger, Chem. Rev. 2003, 103, 2795–2827; c) C. Spino, Angew. Chem. 2004, 116, 1796–1798; Angew. Chem. Int. Ed. 2004, 43, 1764–1766; d) H. Vogt, S. Bräse, Org. Biomol. Chem. 2007, 5, 406–430; e) S. J. Connon, Angew. Chem. 2008, 120, 1194–1197; Angew. Chem. Int. Ed. 2008, 47, 1176–1178; f) J. Wang, X. Liu, X. Feng, Chem. Rev. 2011, 111, 6947–6983.

- [3] For reviews, see: a) A. B. Dounay, L. E. Overman, *Chem. Rev.* 2003, 103, 2945–2963; b) C. V. Galliford, K. A. Scheidt, Angew. Chem. 2007, 119, 8902–8912; Angew. Chem. Int. Ed. 2007, 46, 8748–8758; c) F. Zhou, Y. L. Liu, J. Zhou, Adv. Synth. Catal. 2010, 352, 1381– 1407; d) J. J. Badillo, N. V. Hanhan, A. K. Franz, Curr. Opin. Drug Discov. Dev. 2010, 13, 758–766; e) J. E. M. N. Klein, R. J. K. Taylor, Eur. J. Org. Chem. 2011, 6821–6841; f) K. Shen, X. Liu, L. Lin, X. Feng, Chem. Sci. 2012, 3, 327–334; g) N. R. Ball-Jones, J. J. Badillo, A. K. Franz, Org. Biomol. Chem. 2012, 10, 5165–5181.
- [4] L. Horoszok, C. Leung, M. Tomaszewski, C. Walpole, *PCT Int. Appl.* WO2007091946, **2007**.
- [5] Y.-L. Liu, F. Zhou, J.-J. Cao, C.-B. Ji, M. Ding, J. Zhou, Org. Biomol. Chem. 2010, 8, 3847–3850.
- [6] A. Sacchetti, A. Silvani, F. G. Gatti, G. Lesma, T. Pilati, B. Trucchi, Org. Biomol. Chem. 2011, 9, 5515–5522.
- For examples of the enantioselective strecker reaction 171 of ketimines catalyzed by metal-complex, see: a) K. Yabu, S. Masumoto, S. Yamasaki, Y. Hamashima, M. Kanai, W. Du, D. P. Curran, M. Shibasaki, J. Am. Chem. Soc. 2001, 123, 9908-9909; b) S. Masumoto, H. Usuda, M. Suzuki, M. Kanai, M. Shibasaki, J. Am. Chem. Soc. 2003, 125, 5634-5635; c) N. Kato, T. Mita, M. Kanai, B. Therrien, M. Kawano, K. Yamaguchi, H. Danjo, Y. Sei, A. Sato, S. Furusho, M. Shibasaki, J. Am. Chem. Soc. 2006, 128, 6768-6769; d) J. P. Abell, H. Yamamoto, J. Am. Chem. Soc. 2009, 131, 15118-15119; e) J. Wang, W. Wang, W. Li, X. Hu, K. Shen, C. Tan, X. Liu, X. Feng, Chem. Eur. J. 2009, 15, 11642-11659; f) J. Wang, W. Li, Y. L. Liu, Y. Y. Chu, L. L. Lin, X. H. Liu, X. M. Feng, Org. Lett. 2010, 12, 1280-1283.
- [8] For examples of the enantioselective strecker reaction of ketimines catalyzed by organic catalyst, see: a) P. Vachal, E. N. Jacobsen, Org. Lett. 2000. 2, 867-870; b) P.V achal, E. N. Jacobsen, J. Am. Chem. Soc. 2002, 124, 10012-10014; c) M. Suginome, A. Yamamoto, Y. Ito, Chem. Commun. 2002, 1392-1393; d) X. Huang, J. L. Huang, Y. H. Wen, X. Feng, Adv. Synth. Catal. 2006, 348, 2579-2584; e) J. L. Huang, X. H. Liu, Y. H. Wen, B. Qin, X. Feng, J. Org. Chem. 2007, 72, 204-208; f) M. Rueping, E. Sugiono, S. A. Moreth, Adv. Synth. Catal. 2007, 349, 759-764; g) Z. R. Hou, J. Wang, X. H. Liu, X. Feng, Chem. Eur. J. 2008, 14, 4484-4486; h) K. Shen, X. H. Liu, Y. F. Cai, L. L. Lin, X. M. Feng, Chem. Eur. J. 2009, 15, 6008-6014; i) G.-W. Zhang, D.-H. Zheng, J. Nie, T. Wang, J.-A. Ma, Org. Biomol. Chem. 2010, 8, 1399-1405; j) D. Enders, K. Gottfried, G. Raabe Adv. Synth. Catal. 2010, 352, 3147-3152.
- [9] For the use of alpha-amido sulfones in the asymmetric Strecker reaction, see: a) R. P. Herrera, V. Sgarzani, L. Bernardi, F. Fini, D. Pettersen, A. Ricci, J. Org. Chem. 2006, 71, 9869–9872; b) T. Ooi, Y. Uematsu, J. Fujimoto, K. Fukumoto, K. Maruoka, Tetrahedron Lett. 2007, 48, 1337–1340; c) R. Reingrüber, T. Baumann, S. Dahmen, S. Bräse, Adv. Synth. Catal. 2009, 351, 1019–1024.
- [10] a) W. Yan, D. Wang, J. Feng, P. Li, R. Wang, J. Org. Chem. 2012, 77, 3311–3317; b) W. Yan, D. Wang, J. Feng, P. Li, D. Zhao, R. Wang, Org. Lett. 2012, 14,

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Adv. Synth. Catal. 0000, 000, 0-0

KK These are not the final page numbers!

¹⁰

2512-2515; c) J. Feng, W. Yan, D. Wang, P. Li, Q. Sun, R. Wang, Chem. Commun. 2012, 8003-8005.

- [11] For the use of alcohol as an additive or solvent in the Strecker reaction, see, for example: S. J. Zuend, M. P. Coughlin, M. P. Lalonde, E. N. Jacobsen, Nature 2009, 461, 968–970.
- [12] For recent examples on the use of bifunctional qunindine derived thiourea catalysts for enantioselective Strecker reaction, see: a) Y. Shao, S. Tian, Chem. Commun. 2012, 4899-4901; b) Y.-L. Liu, T.- D. Shi, F. Zhou, X.-L. Zhao, X. Wang, J. Zhou, Org. Lett, 2011, 13, 3826–3829.
- [13] After a simple chemical conversion, the ee of compound 5a was determined as 90%.



- [14] CCDC 888825 contains the supplementary crystallographic data for the compound 2m reported in this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [15] C. Gioia, L. Bernardi, A. Ricci, Synthesis 2010, 161-170.

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FULL PAPERS

12 The Quinine Thiourea-Catalyzed Asymmetric Strecker Reaction: An Approach for the Synthesis of 3-Aminooxindoles

Adv. Synth. Catal. 2013, 355, 1-12

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