

Asymmetric Aza-Diels–Alder Reactions of in Situ Generated β,β -Disubstituted α,β -Unsaturated N–H Ketimines Catalyzed by Chiral Phosphoric Acids

Shunlong He, Huanchao Gu, Yu-Peng He,* and Xiaoyu Yang*

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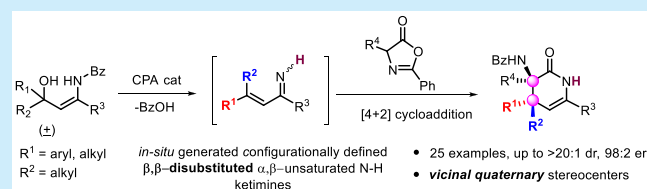
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ABSTRACT: A novel asymmetric synthesis of dihydropyridinones with vicinal quaternary stereocenters has been realized by asymmetric aza-Diels–Alder reactions of 3-amido allylic alcohols with oxazolones enabled by chiral phosphoric acid catalysis. A series of aryl/alkyl- and alkyl/alkyl-disubstituted 3-amido allylic tertiary alcohols and 4-substituted oxazolones could be well tolerated in these reactions, producing dihydropyridinones with excellent diastereoselectivities and high enantioselectivities. Mechanistic study and control experiments were performed to shed light on the reaction mechanism, in which a configurationally defined β,β -disubstituted α,β -unsaturated N–H ketimine was proposed as the key intermediate.



Dihydropyridinones are a type of privileged six-membered azacycles, which are widely present in a series of biologically active small molecules and natural products.¹ Among various synthetic methods for preparation of these heterocycles, the inverse-electron-demand aza-Diels–Alder (ADA) reaction² is one of the most efficient methods, which was pioneered by Boger and co-workers.³ The asymmetric catalytic versions⁴ of these reactions have also drawn considerable research interest due to the significance of chiral dihydropyridinone scaffolds, and numerous of elegant enantioselective methods have been developed in the past two decades. In general, an enolate-type (or enolate analog) intermediate is generated under diverse catalytic systems, which undergoes the enantioselective aza-Diels–Alder reactions with α,β -unsaturated imines to generate the chiral dihydropyridinone products.⁵ Since the landmark work of using chiral N-heterocyclic carbene (NHC) catalysts in the asymmetric ADA reactions between enals and α,β -unsaturated imines by Bode and co-workers,⁶ asymmetric NHC catalysis has become a versatile approach for the synthesis of enantioenriched dihydropyridinones, using ketenes,⁷ α -chloroaldehyde,⁸ and alkylacetic ester⁹ as enolate precursors by Ye and Chi group, respectively. The Smith group disclosed enantioselective synthesis of dihydropyridinones through asymmetric ADA reactions between α,β -unsaturated imines and arylacetic acids¹⁰ and N-acyl imidazoles¹¹ via chiral isothiourea catalysis. The Gong group reported the asymmetric three-component ADA reaction between enals, anilines, and oxazolones under chiral Brønsted acid catalysis¹² (Figure 1a, top). On the other hand, asymmetric ADA reactions have also been well employed in the asymmetric synthesis of tetrahydropyridine derivatives via cycloadditions between

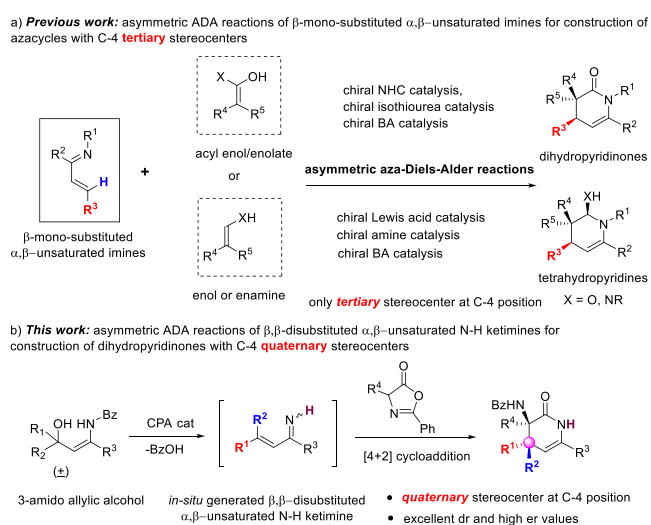


Figure 1. Asymmetric aza-Diels–Alder reactions of α,β -unsaturated imines.

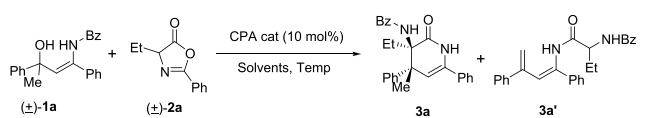
enols/enamines and α,β -unsaturated imines, which could be enabled by chiral Lewis acid catalysis,¹³ chiral amine catalysis,¹⁴ and chiral Brønsted acid catalysis (Figure 1a, bottom).¹⁵

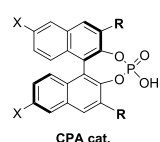
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Despite the fact that a series of different asymmetric catalytic methods have been developed for asymmetric synthesis of dihydropyridinones and tetrahydropyridines via enantioselective ADA reactions, the other coupling partner for cycloaddition in these reactions was limited to α,β -unsaturated imines possessing a mono- β -substitution, thus leading to the generation of azacycles with only a tertiary stereocenter at the C-4 position.^{5f,g} To the best of our knowledge, no example of β,β -disubstituted α,β -unsaturated imine precursors has been utilized in asymmetric ADA reactions, probably due to their relatively low reactivities and challenges in stereoselectivities control. With our continuous interest in developing novel asymmetric reactions of amido allylic alcohols,¹⁶ herein we report a novel asymmetric synthesis of dihydropyridinones with vicinal quaternary stereocenters via enantioselective aza-Diels–Alder reactions between 3-amido allylic alcohols and oxazolones¹⁷ enabled by chiral phosphoric acid (CPA) catalysis,¹⁸ in which an *in situ* generated β,β -disubstituted α,β -unsaturated N–H ketimine¹⁹ was proposed as the key intermediate (Figure 1b).

We commenced our study by choosing racemic 3-amido allylic alcohol **1a** and oxazolone **2a** as model substrates under CPA catalysis (Table 1). The initial attempt between these two

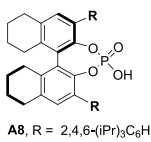
Table 1. Optimization of Reaction Conditions^a





CPA cat.

A1, X = H, R = 2,4,6-(iPr)₃C₆H₂
 A2, X = H, R = 1-naphthyl
 A3, X = H, R = 2-naphthyl
 A4, X = H, R = 9-anthracenyl
 A5, X = H, R = 2,4,6-(Me)₃C₆H₂
 A6, X = H, R = 2,4,6-(Cy)₃C₆H₂
 A7, X = nC₆H₁₇, R = 2,4,6-(iPr)₃C₆H₂
 A8, R = 2,4,6-(iPr)₃C₆H₂



CPA cat.

entry ^a	cat.	solvents	yield ^b	dr ^c	er ^d
1	A1	toluene	30	>20:1	91:9
2	A1	DCM	11	>20:1	85:15
3	A1	CHCl ₃	23	>20:1	89.5:10.5
4	A1	CCl ₄	37	>20:1	93.5:6.5
5	A2	CCl ₄	16	>20:1	78:22
6	A3	CCl ₄	24	>20:1	62.5:37.5
7	A4	CCl ₄	13	>20:1	93.5:6.5
8	A5	CCl ₄	24	>20:1	91.5:8.5
9	A6	CCl ₄	28	>20:1	91:9
10	A7	CCl ₄	44	>20:1	93:7
11	A8	CCl ₄	34	>20:1	92.5:7.5
12 ^e	A7	CCl ₄	56	>20:1	93.5:6.5
13 ^{e,f}	A7	CCl ₄	61	>20:1	95:5

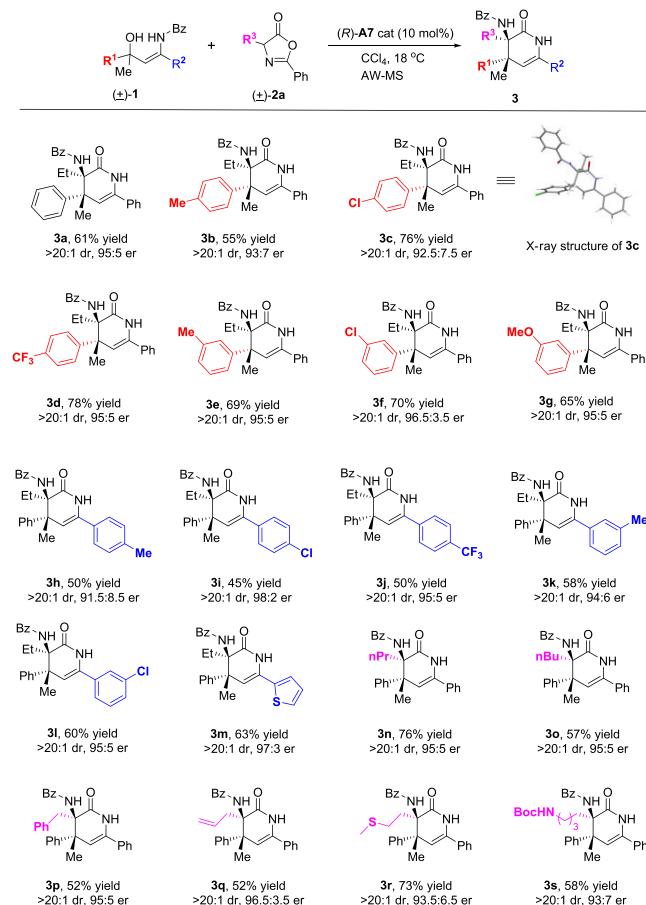
^aReactions were performed with **1a** (0.1 mmol), **2a** (0.1 mmol), CPA catalyst (0.01 mmol), solvents (1 mL), AW-300 MS (10 mg) at 25 °C. ^bIsolated yield. ^cDr value was determined by NMR analysis. ^dEr value was determined by HPLC analysis on a chiral stationary phase. ^e**1a** (0.15 mmol) was used. ^fAt 18 °C.

substrates (1:1 mol ratio) under the catalysis of TRIP catalyst (cat **A1**, 10 mol %) in toluene in the presence of acid-wash molecular sieves (AW-300 MS) at 25 °C successfully produced the dihydropyridinone **3a** in 30% yield with >20:1 diastereoselectivity and 91:9 enantiomeric ratio (er) (entry 1). Interestingly, the *N*-1 benzoyl group of the product was removed under these conditions. Next, a series of solvents were first examined (entries 2–4), indicating CCl₄ was the optimal

one, in which an improved er was afforded (entry 4). Subsequently, a range of BINOL-derived CPA catalysts were screened; however, unfortunately, none of them could provide improved results (entries 5–9). Interestingly, using C₈-TRIP (cat **A7**) as catalyst led to an improved yield with retained er (entry 10), while switching the chiral scaffold to H8-BINOL type did not provide better results (entry 11). Due to the vulnerability of **1a** under these conditions, increasing the amount of **1a** to 1.5 equiv provided significantly improved yield (entry 12). Finally, decreasing the reaction temperature to 18 °C generated the dihydropyridinones **3a** in 61% yield with >20:1 dr and 95:5 er (entry 13; see Table S1 in the Supporting Information (SI) for more optimization conditions).²⁰ It is worth mentioning that the oxazolone ring-opening addition product **3a'** was isolated as the major byproduct in 27% yield with 55.5:44.5 er under these optimal conditions.

With the optimal conditions in hand, we sought to explore the scope of this reaction (Scheme 1). A series of aryl groups at the 1-position of 3-amido allylic alcohols **1** were first examined, and we found that various *para*- and *meta*-substituted phenyl groups could be well tolerated under the optimal conditions

Scheme 1. Scope for Enantioselective Synthesis of Dihydropyridinones with Vicinal Chiral Quaternary Stereocenters^a

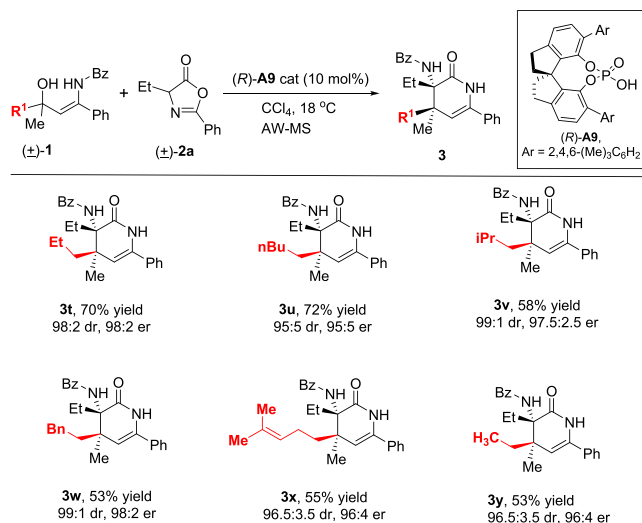


^aReactions were performed with **1** (0.15 mmol), **2** (0.1 mmol), (*R*)-**A7** cat (0.01 mmol), CCl₄ (1 mL), AW-300 MS (10 mg) at 18 °C for 40 h. Yields were isolated yield. Dr values were determined by NMR analysis. Er values were determined by HPLC analysis on a chiral stationary phase.

(3b–3g), generating the dihydropyridinone products with excellent diastereoselectivities and high enantioselectivities. The absolute configurations of the dihydropyridinone products were confirmed by analogy to product 3c, whose structure was unambiguously assigned by X-ray crystallography (CCDC 1993830).²⁰ Next, we investigated the scope of the substitutions at the 3-position of allylic alcohols. Encouragingly, a variety of substituted phenyl groups were amenable with the standard conditions (3h–3l), as well as a thienyl substituent (3m). Finally, the scope of the 4-substitutions of oxazolone 2 was also explored, and a range of alkyl groups were well compatible with the optimal conditions (3n–3o), as well as the benzyl (3p) and allyl groups (3q). In addition, the functional group-containing substituents were also well tolerated, producing the products with high stereoselectivities (3r and 3s).

To further extend the scope of these reactions, *n*Pr/Me-disubstituted 3-amido allylic alcohol 1t was examined under the optimal conditions, however, which provided the dihydropyridinone product with poor enantioselectivity.²⁰ Encouragingly, after brief optimizations of the CPA catalysts (see Table S2 in SI), the 2,4,6-trimethylphenyl substituted SPINOL-derived phosphoric acid (*R*)-A9 was determined to be the optimal catalyst for this reaction, which generated the product in 70% yield with 98:2 dr and 98:2 er (Scheme 2, 3t).

Scheme 2. Scope for Enantioselective Synthesis of Dihydropyridinones from Dialkyl-Substituted 3-Amido Allylic Alcohols^a



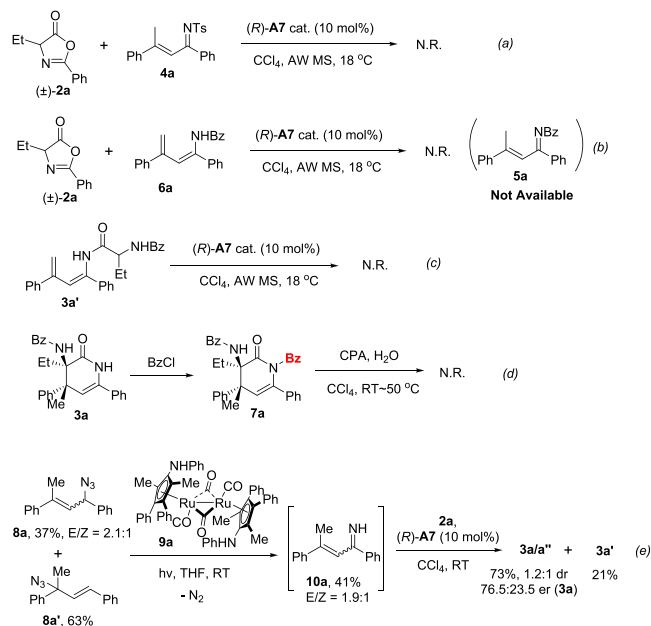
^aReactions were performed with 1 (0.15 mmol), 2a (0.1 mmol), (*R*)-A9 cat (0.01 mmol), CCl₄ (1 mL), AW-300 MS (10 mg) at 18 °C for 40 h. Yields were isolated yields. Dr and er values were determined by HPLC analysis on a chiral stationary phase.

Subsequently, a series of alkyl groups were examined at the 1-position of the allylic alcohols, which provided the dihydropyridinones with high diastereo- and enantioselectivities (3u and 3v), including the aryl- and olefine-containing substituents (3w and 3x). It is worth mentioning that even the Et/Me-disubstituted 3-amido allylic alcohol could afford the product with excellent stereoselectivity under the standard conditions without exception (3y).

To shed light on the reaction mechanism and demonstrate the uniqueness of these reactions, some control experiments

were performed (Scheme 3). Treatment of oxazolone 2a with the β,β -disubstituted α,β -unsaturated *N*-tosyl ketimine 4a⁹ did

Scheme 3. Control Experiments and Mechanistic Studies

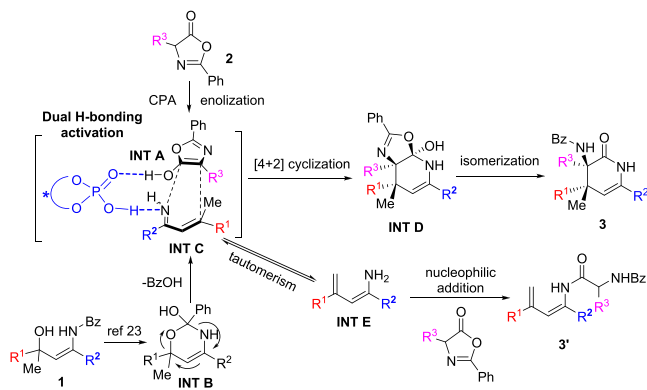


not afford any desired dihydropyridinone product under the standard conditions (Scheme 3a). In an attempt to prepare the β,β -disubstituted α,β -unsaturated *N*-benzoyl imine 5a via condensation⁹ between the corresponding enone and BzNH₂, only the dienamide product 6a was obtained. Theoretically, this dienamide could still undergo isomerization to give the α,β -unsaturated imine 5a under CPA catalysis conditions; however, no dihydropyridinone product was detected in the reaction between dienamide 6a and oxazolone 2a under the standard conditions, as well as no oxazolone ring-opening addition product (Scheme 3b). Treatment of the oxazolone ring-opening addition product 3a' with the standard conditions could not provide any dihydropyridinone product, which indicated 3a' was probably not an intermediate in these cycloaddition reactions (Scheme 3c). The absence of the *N*-1 Bz group in the products is a key feature of these reactions, which is also very important to understand the reaction mechanism. Selective benzoylation of the dihydropyridinone product 3a gave the *N*-1 benzoylated product 7a, whose structure was confirmed by 2D NMR analysis. Treatment of 7a with CPA catalyst A7 and H₂O did not give any debenzoylation product, even at higher reaction temperature, which suggested that 7a was also not an intermediate of these reactions²¹ (Scheme 3d). All these results clearly suggested that the α,β -unsaturated *N*-H ketimine 5a is probably not the intermediate in these reactions. Accordingly, we envisioned that the relatively rare α,β -unsaturated *N*-H ketimine may be the key intermediate of these cycloaddition reactions. To prove this hypothesis, irradiation of the mixture of α - and γ -allyl azides²² (8a and 8a') by ruthenium catalysis (9a) under fluorescent light in THF led to the formation of α,β -unsaturated *N*-H ketimine (10a) in 41% conversion, which existed as an *E/Z* mixture.²³ After removal of the solvent, the obtained crude mixture was directly subjected into the standard reaction conditions with oxazolone 2a, which finally provided the desired dihydropyridinone 3a in 39% yield with

76.5:23.5 er, with the diastereomer **3a''** in 34% yield and oxazolone ring-opening product **3a'** in 21% yield (Scheme 3e).²⁰

Based on these results and previous reports, a plausible reaction mechanism was proposed (Scheme 4). First, the

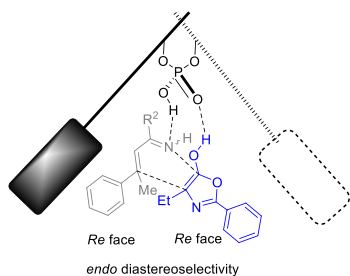
Scheme 4. Plausible Reaction Mechanism



oxazolone **2** was activated by CPA catalyst to generate its active enol form **INT A**,^{12,17k,o} at the same time, the CPA also mediated the addition of the hydroxyl group to amide group in **1** to generate the orthoester **INTB**,²⁴ which then underwent the rearrangement to give the configurationally defined α,β -unsaturated N–H ketimine **INT C**. Subsequently, dual hydrogen bonding activation of the enol form of oxazolone (**INT A**) and α,β -unsaturated N–H ketimine **INT C** led to the facile [4 + 2] cycloaddition,²⁵ generating the bicyclic intermediate **INT D**, which then went through the isomerization process to give the dihydropyridinone product **3**. On the other hand, the α,β -unsaturated N–H ketimine **INT C** could also undergo the imine–enamine tautomerism to generate the primary dienamide **INT E**, which would attack the oxazolone to give the ring-opening addition byproduct **3'**.

To account for the observed stereoselectivities, a possible stereochemical model was depicted (Scheme 5). Through the

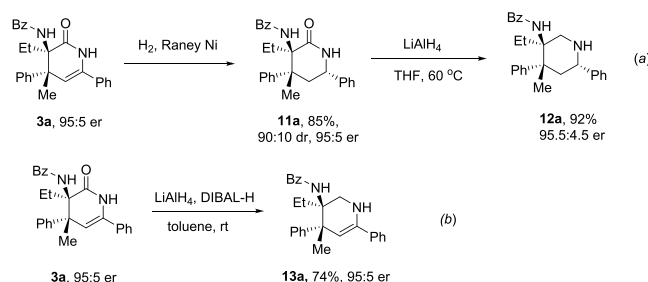
Scheme 5. Possible Stereochemical Model



bifunctional activation of the α,β -unsaturated N–H ketimine intermediate and the enol form of oxazolone via dual hydrogen-bonding activation with the CPA catalyst, the *Re* face of the enol form of oxazolone attacks the *Re* face of α,β -unsaturated N–H ketimine in an *endo*-type fashion to generate the (3*R*,4*R*)-dihydropyridinone products.

To demonstrate the applicability of these reactions, the derivatizations of the chiral products were studied (Scheme 6). Catalytic hydrogenation of dihydropyridinone **3a** using Raney Ni as the catalyst generated the pyridinone **11a** in 85% yield with 90:10 dr, whose relative configuration was assigned by

Scheme 6. Derivatizations of the Chiral Products



NOE analysis.²⁰ Further reduction of **11a** with LiAlH_4 afforded the piperidine **12a** in 92% yield, without erosion of the optical purity (Scheme 6a). Direct subjection of dihydropyridinone **3a** into the reduction with the mixture of DIBAL-H and LiAlH_4 ²⁶ produced the tetrahydropyridine **13a** in 74% yield with retained ee (Scheme 6b).

In conclusion, we have disclosed a novel asymmetric synthesis of dihydropyridinones bearing vicinal chiral quaternary stereocenters via enantioselective aza-Diels–Alder reaction between 3-amido allylic alcohols and oxazolones enabled by chiral phosphoric acid catalysis. A wide array of substituents at the 1- and 3-position of 3-amido allylic tertiary alcohols and 4-position of oxazolones could be well tolerated, generating dihydropyridinones with excellent diastereoselectivities and high enantioselectivities, especially highlighting the compatibility of the 1,1-dialkyl substituted 3-amido allylic alcohols. Control experiments were performed to elucidate the mechanism of these reactions, in which a configurationally defined β,β -disubstituted α,β -unsaturated N–H ketimine was proposed as the key intermediates for these asymmetric [4 + 2] cycloaddition reactions. Further application of this protocol for *in situ* generation of reactive α,β -unsaturated N–H ketimine intermediates in CPA catalyzed asymmetric reactions was under investigation in our lab, especially for the asymmetric construction of chiral quaternary stereocenters.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c01994>.

Full experiment procedures, spectroscopic characterizations, NMR and HPLC spectra (PDF)

■ Accession Codes

CCDC 1993830 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Authors

Yu-Peng He – College of Chemistry, Chemical Engineering and Environmental Engineering, Liaoning Shihua University, Fushun 113001, China; Email: yupeng.he@lnpu.edu.cn

Xiaoyu Yang – School of Physical Science and Technology, ShanghaiTech University, Shanghai 201210, China;
orcid.org/0000-0002-0756-0671; Email: yangxy1@shanghaitech.edu.cn

Authors

Shunlong He – School of Physical Science and Technology, ShanghaiTech University, Shanghai 201210, China; College of Chemistry, Chemical Engineering and Environmental Engineering, Liaoning Shihua University, Fushun 113001, China

Huanchao Gu – School of Physical Science and Technology, ShanghaiTech University, Shanghai 201210, China

Complete contact information is available at:

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Notes

The authors declare no competing financial interest.

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