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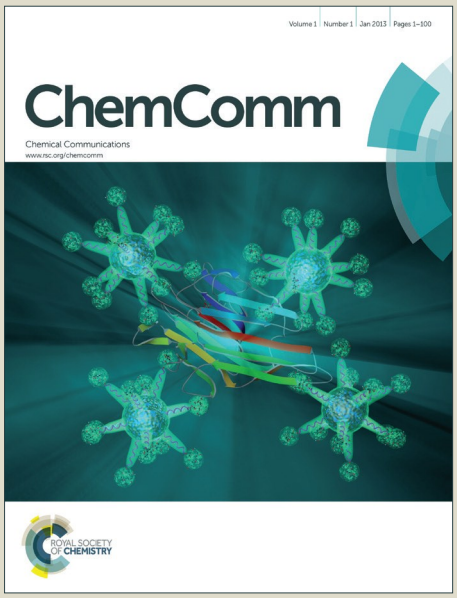

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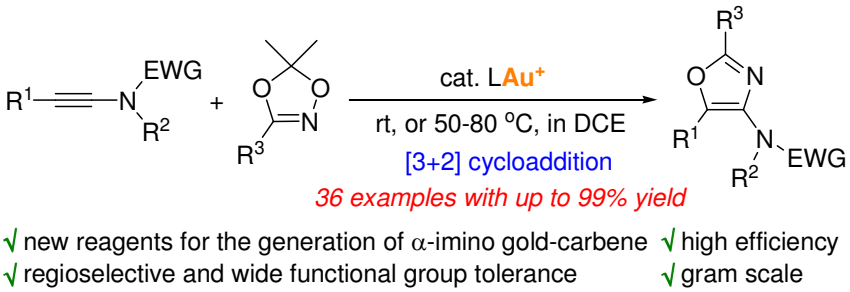

Abstract: Gold-catalyzed regioselective [3+2] cycloaddition of ynamides with 1,4,2-dioxazoles offers an efficient approach to functionalized oxazoles under mild reaction conditions.

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COMMUNICATION

Dioxazoles, A New Mild Nitrene Transfer Reagent in Gold Catalysis: Highly Efficient Synthesis of Functionalized Oxazoles

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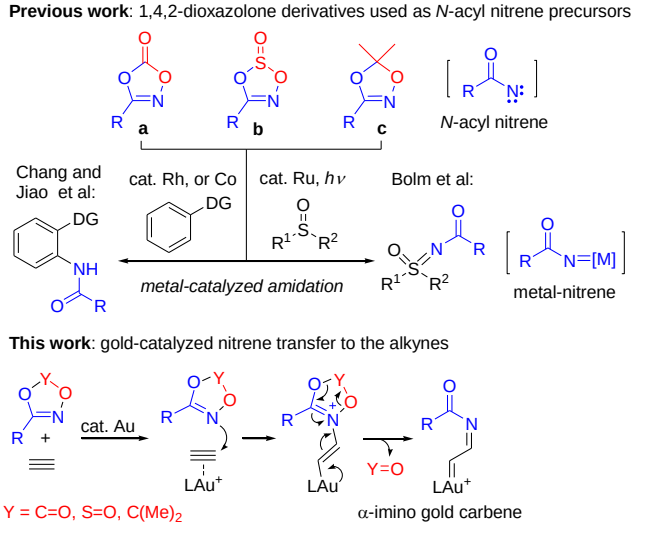
A gold-catalyzed regioselective [3+2] cycloaddition of ynamides with 1,4,2-dioxazoles has been developed, which offers a novel approach to highly functionalized oxazoles under mild reaction conditions. 1,4,2-Dioxazole was found to act as an efficient *N*-acyl nitrene equivalent to trigger a facile generation of α-imino gold-carbene intermediate through the elimination of a ketone.

In recent years, gold-carbene mediated reactions have attracted considerable attention since they serve as promising intermediates in the synthesis of various types of carbo- or heterocycles.[1] Compared with α-carbonyl gold carbenes,[2] generations and reactions of α-imino gold carbenes have less been explored.[3] These highly reactive gold-species are mainly accessed through gold-catalyzed nitrene transfer to alkynes using azides as the nitrene equivalent reported by Toste,[4a] Gagosz,[4b] Zhang[4c-e] and others.[4] Recently, 2*H*-azirines,[5] *N*-iminopyridium ylides,[6] isoxazoles,[7] benzoisoxazoles[8] and triazapentalene[9] have also been used as nitrene equivalents. Despite the impressive progress, the development of new methods for the generation of α-imino gold carbenes that utilization of less reactive/sensitive nitrene transfer reagents with high chemo- and regioselectivities under milder reaction conditions is still highly desired. 1,4,2-Dioxazol-5-one **a**, a cyclic carbonate of hydroxamic acids, and its derivative of 1,4,2-dioxazol-5-thione **b**, were found in 1968 to undergo thermal or photo-induced decomposition leading to highly reactive *N*-acyl nitrene intermediates via elimination of CO_2 or SO_2.[10] 1,4,2-Dioxazole **c** decomposed similarly at elevated temperatures (above 150 °C) into isocyanates and ketones.[11] These attractive and easily accessible heterocyclic compounds are potentially useful as the *N*-acyl nitrene precursors in place of hazardous acyl azides, which may produce the *N*-acyl nitrene or *N*-acyl nitrenoid intermediates under mild reaction conditions such as in the presence of a metal catalyst. Recently, Bolm et al. described an elegant light-induced ruthenium-catalyzed synthesis of *N*-acyl sulfoximines and sulfimides at room temperature via a ruthenium *N*-acyl nitrene intermediate using dioxazolone **a** as the nitrene precursors.[12] More recently, Chang and others[13] revealed that the substrates **a**-**c** could also be used as amidating reagents in metal-catalyzed C-H amidation reactions, in which metal-nitrene complex is proposed to be involved (Scheme 1). During our continuous work on gold-catalyzed oxidative reactions, we hypothesized that these five-membered heterocycles might be employed as a nucleophilic nitrene equivalent to trigger an efficient generation of α-imine gold-carbene species through nucleophilic attack of the gold-activated alkyne followed by expulsion of a leaving group. In this design, no metal-nitrene complex is formed, which is different from other metal-catalyzed reactions shown in above. Herein, we describe a novel reactivity of dioxazole derivatives, which acts as a new type of nitrene transfer reagent and undergoes gold-catalyzed [3+2] cycloaddition with ynamides leading to a facile synthesis of highly functionalized oxazoles.[6b-d]

To test our hypothesis, we initially investigated the reactions of mesylamide-derived ynamide **1a** with three different types of dioxazole derivatives **2a**-**2c** in the presence of 5 mol%



Scheme 1. Metal-catalyzed reactions involing nitrene equivalents of 1,4-2-dioxazolone derivatives and the design of gold-catalyzed nitrene transfer reactions

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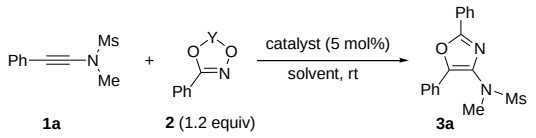
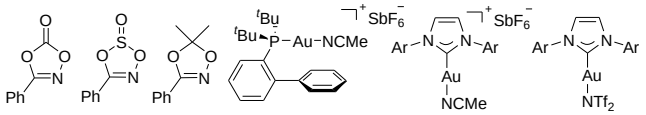
Johnphos(MeCN)AuSbF$_6$ (catalyst **A**) in DCE at room temperature. However, in the case of dioxazolone **2a**, a non-clean reaction mixture was resulted with significant remaining of **2a**, possibly the rapid self-reaction of ynamide occurred under gold-catalyzed conditions.[14] No desired cyclization product was observed also in the case of dioxathiazole **2b** (Table 1, entries 1-2). Considering the lower nucleophilicity of **2a** and **2b**, we reasoned that employing more nucleophilic dioxazole might be feasible for the successful transformation. Gratifyingly, employing dioxazole **2c** led to the desired 4-amino-oxazole **3a** in 92% yield within 2 h (entry 3). The results implied that an efficient [3+2] cycloaddition of ynamide with dioxazole took place, and the self-reaction of yanmide was mostly suppressed. A similar reaction outcome was found when *N*-heterocyclic carbene gold(I) complex **B** (IPrAu(MeCN)SbF$_6$) or **C** (IPrAuNTf$_2$)[15] was used as the catalyst (IPr = 1,3-bis(2,6-diisopropylphenyl)imidazol-2-ylidene) (entries 4-5). Various commonly used gold catalysts also catalyze the desired cyclization efficiently, furnishing **3a** in lower yields of 72-87% (entries 6-8). The reaction could also be performed smoothly in the solvents of DCM, THF, toluene or CH_3CN (entries 9-12). No reaction was observed catalyzed by IPrAuCl or AgNTf$_2$ alone or in the absence of any catalyst (entries 13-15).

Encouraged by these results, we next investigated the substrate scope of the reaction. The scope of ynamides was first investigated using dioxazole **2c** as the reaction partner under the reaction conditions given in Table 1, entry 5. The results are shown in Table 2. The effects of the electron-withdrawing groups on nitrogen were first examined. The reactions proceeded very well with tosyl, *para*-bromobenzenesulfonyl (Bs) and a stronger electron-withdrawing *para*-nitrobenzenesulfonyl (*p*-Ns) moiety, furnishing **3c**-**3e** in 72-86% yields. More electron-rich ynamide with an oxazolidine group also afforded the corresponding oxazole **3f** in 89% yield. *N*-aryl mesylamide, whenever bearing an electron-neutral, electron-deficient CF_3, or electron-rich MeO substituent on its aromatic ring, tolerated well in this reaction, leading to **3b** and **3g**-**3h** in 96-99% yields. *N*-benzyl mesylamide was also suitable to provide **3i** in 85% yield. Next, the effect of R^1 group on the alkyne terminus was examined. For aryl substituted alkynes, a wide range of functionalities such as F, Cl, CF_3, Me and MeO on aromatic rings were compatible, furnishing **3j**-**3n** in good to high yields. It was noted that when *p*-MeO-substituted aryl alkyne was used, partial of the product precipitated during the reaction process at room temperature, which appeared to interfere with the reaction process. Then a higher reaction temperature (50 °C) was required to achieve a better conversion. Ynamide with an 1,3,5(10)-estratrien-3-ol-17-one derivative also reacted efficiently to produce the oxazole **3o** in 97% yield. 1-Naphthyl and 2-thienyl-substituted alkynes converted into the corresponding **3p** and **3q** in excellent yields. Cyclohexenyl-substituted alkyne transformed to the corresponding **3r** in

Table 1. Optimization of reaction conditions.



1a **2** (1.2 equiv) **3a**



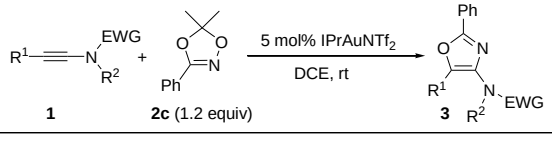
2a **2b** **2c** **A** **B** **C**

Ar = 2,6-(^{i}Pr)$_2$C$_6$H$_3$

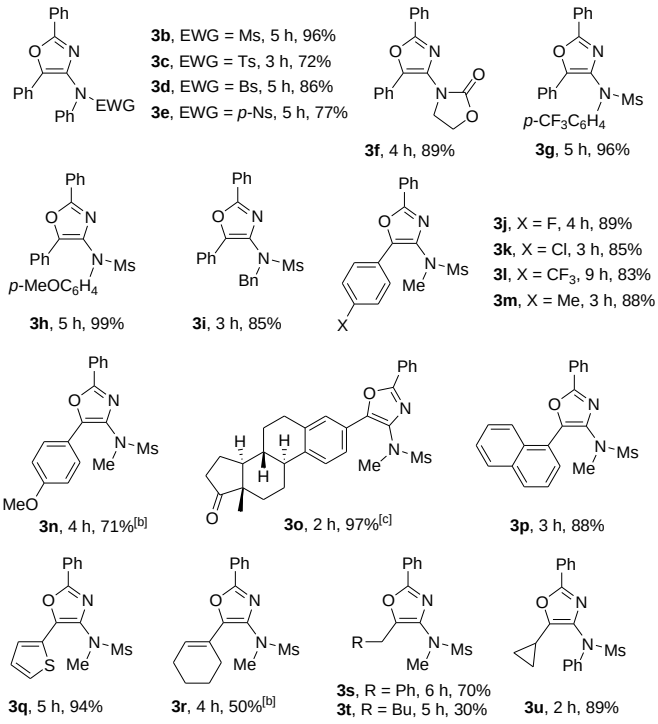
entry	substrate	catalyst	solvent	time (h)	yield (%)[a]
1	**2a**	**A**	DCE	3 h	-
2	**2b**	**A**	DCE	3 h	-
3	**2c**	**A**	DCE	2 h	92
4	**2c**	**B**	DCE	2 h	90
5	**2c**	**C**	DCE	2 h	95
6	**2c**	PPh_3AuNTf_2	DCE	2 h	85
7	**2c**	PPh_3AuSbF_6	DCE	3 h	87
8	**2c**	PPh_3AuOTf	DCE	3 h	72
9	**2c**	**C**	DCM	3 h	91
10	**2c**	**C**	THF	3 h	86
11	**2c**	**C**	toluene	3 h	88
12	**2c**	**C**	MeCN	3 h	92
13	**2c**	IPrAuCl	DCE	3 h	-(99)
14	**2c**	$AgNTf_2$	DCE	3 h	-(98)
15	**2c**	none	DCE	12 h	-(99)

[a] Isolated yields. Ms = methanesulfonyl. The yileds of recovered **1a** are shown in parentheses.

Table 2. Scope of ynamides.[a]



1 **2c** (1.2 equiv) **3**

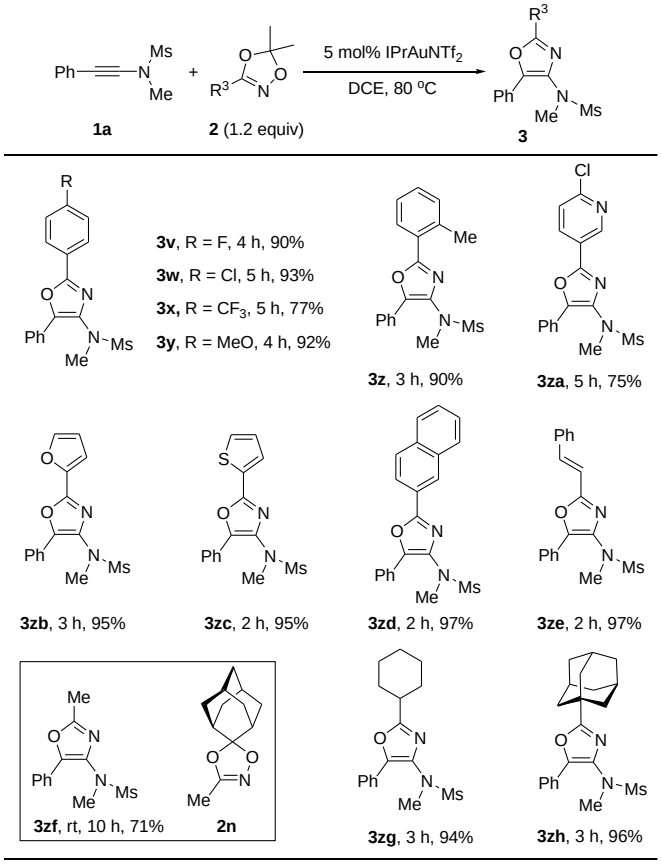


[a] Isolated yields. Ts = toluene-4-sulfonyl, Bs = *para*-bromobenzenesulfonyl. *p*-Ns = *para*-nitrobenzenesulfonyl. [b] 50 °C. [c] 80 °C.

moderate yield. Alkyl-substituted alkynes such as benzyl- or cyclopropyl-substituted one underwent the reaction smoothly to give **3s** and **3u** in 70% and 89% yields, respectively. However, a pentyl-substituted ynamide afforded **3t** only in 30% yield. No alkene product derived from 1,2-C-H insertion of gold-carbene intermediate was observed in these cases. The results indicated that intramolecular nucleophilic attack of *N*-acyl group to gold-carbene is much faster than 1,2-C-H insertion due to the ease of aromatization.

The scope of dioxazoles was also investigated using ynamide **1a** as the reaction partner. Due to the lower solubility of the products in DCE, all the reactions were carried out at 80 °C. Under this reaction condition, we were pleased to see that the reactions were quite general with substituted dioxazoles, since aryl, heteroaryl, alkenyl as well as alkyl substituted one were all suitable for this reaction, leading to the highly functionalized oxazoles in good to excellent yields. The reaction efficiency was affected by the nature of aryl substituents: *p*-FC_6H_4 (**3v**, 90%), *p*-ClC_6H_4 (**3w**, 93%), *p*-$CF_3C_6H_4$ (**3x**, 77%), *p*-$MeOC_6H_4$ (**3y**, 92%). Sterically encumbered *o*-Me-substituted aryl dioxazole reacted efficiently to afford **3z** in 90% yield, suggesting the steric hindrance had little effect on the reaction course. Heteroaryl-substituted dioxazoles, such as pyridyl, furanyl and thienyl-substituted one transformed to **3za**-**3zc** successfully in 75-95% yields. High product yields were also observed in 2-naphthyl or alkenyl-substituted dioxazoles.

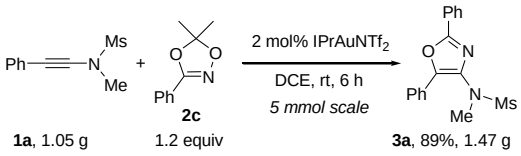
Table 3. Scope of dioxazoles.[a]



[a] Isolated yields.

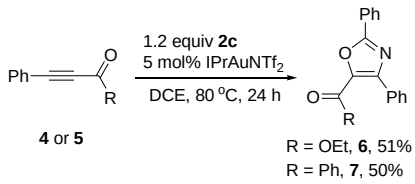
Alkyl-substituted dioxazoles such as methyl, cyclohexyl or even bulky adamantyl turned out to be also perfect substrates to afford **3zf**-**3zh** in 71-96% yields. It was noted that in the case of **3zf**, dioxazole **2n** was used instead of 3,5,5-trimethyl-1,4,2-dioxazole since it is not convenient to prepare the latter with a lower boiling point. Oxazoles constitute important classes of natural products, drugs, and biologically active substances. These compounds are commonly prepared by cyclization of an acyclic precursors or ring derivatization. However, construction of oxazoles through convergent and one-pot methods from readily available substrates is still limited.[16] Our method provided a mild and efficient route to these compounds.

To demonstrate the practicality of our method, a gram scale reaction was performed. It was found that using only 2 mol% of $IPrAuNTf_2$, the reaction of **1a** with dioxazole **2c** at 5 mmol scale delivered oxazole **3a** in high yield of 89% (Scheme 2).



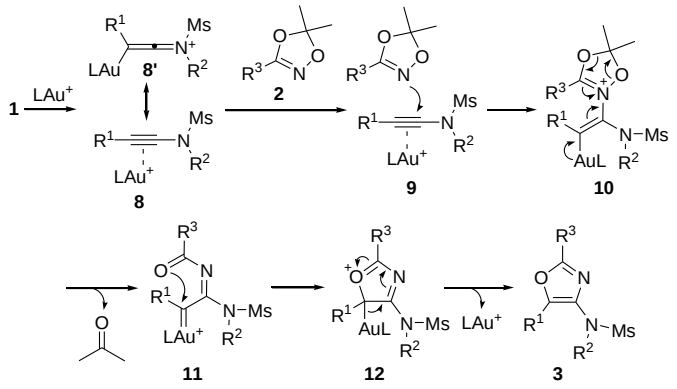
Scheme 2. Gram scale synthesis of **3a**.

The reaction can be extended to other activated alkynes. As shown in Scheme 3, gold-catalyzed reactions of alkynyl ester **4** or alkynyl ketone **5** with **2c** afforded functionalized oxazole **6** or **7** in 51% and 50% yields, respectively. However, when terminal alkyne such as phenylacetylene was used, no clean reaction was observed.



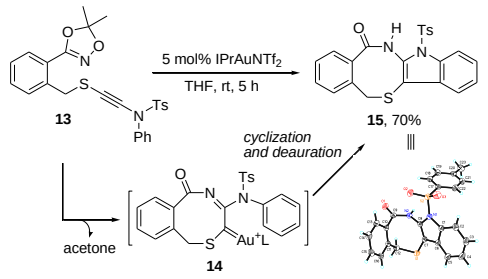
Scheme 3. Reactions of dioxazole with activated alkynes.

We propose the following reaction mechanism for this novel transformation (Scheme 4). Initially, dioxazole **2** attacks on the gold-coordinated ynamide **8** or **8'** regioselectively at the carbon adjacent to the nitrogen due to the polarity of the ynamide to afford iminium ion intermediate **10**. Subsequently, ring fragmentation of **10** generates α-imino gold-carbene **11** with concomitant elimination of acetone. In fact, acetone was formed quantatively and could be detected in the crude reaction mixture.[17] Intermediate **11** may prefer an *E*-form of C=N bond due to the steric repulsion between R^3 substituent on dioxazole with amino moiety,[6a] resulting a *cis* orientation of *N*-acyl group with gold-carbene. Nucleophilic attack of the acyl oxygen in **11** to gold-carbene[18] followed by elimination of the gold catalyst leads to the oxazole products **3**. The reaction pathway involving the formation of *N*-acylaziridine via gold-nitrene followed by cyclization is unlikely, since the oxazole with different regioselectivity would possibly be resulted.[6b,19]



Scheme 4. Possible reaction mechanism

To understand the reaction mechanism, we also tried to trap the α-imino gold-carbene intermediate via an intramolecular cyclization of dioxazole-ynamide **13**, since the C-O bond formation can be avoided in such case. To our delight, **13** cyclized efficiently to give the fused indole derivative **15**[15] in 70% yield (Scheme 5). The results indicated that the α-imino gold-carbene **14** was likely generated in the process, which could be trapped by *N*-aryl ring followed by deauration to furnish the cyclized product.



Scheme 5. Trapping of α-imino gold-carbene intermediate

In summary, we have disclosed that 1,4,2-dioxazole can be used as an efficient nitrene equivalent in gold-catalyzed nitrene transfer reactions to ynamides. The reaction proceeds under mild reaction conditions to afford highly functionalized oxazoles in good to excellent yields via likely the formation of an α-imino gold carbene intermediate followed by cyclization. This method offers several advantages such as easily accessible starting materials, high regioselectivity, wide functional group compatibility and high efficiency. Further investigations on the detailed reaction mechanism and application of this chemistry are in progress.

We thank the National Natural Science Foundation of China (Grant Nos. 21421091, 21372244, 21572256) for financial support.

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