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Synthesis and Ring Opening of Alkaloid-Type Compounds with a Novel Indolo[2,3-c][2]benzazepine Skeleton

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Abstract: Alkylation of the magnesium salts of 2,3-disubstituted indoles with 2-bromomethylbenzonitrile gave 3-(2-cyanobenzyl)-3H-indole derivatives. Reduction of the cyano group of N-methyl 3-(2-cyanobenzyl)-3H-indolium salts afforded previously unreported indolo[2,3-c][2]benzazepines, while acid hydrolysis gave the corresponding indolo[2,3-c][2]benzazepinones. The action of strong protonic acids on indolo[2,3-c][2]benzazepines causes opening of the benzazepine ring annelated to the indole system to form 3Hindolium salts.

Key words: alkaloids, indoles, indolo[2,3-c][2]benzazepines, ring closure, ring opening

The indole nucleus is a key structural feature in (bio)organic chemistry, as it constitutes the main structural element of tryptophane and its metabolites such as tryptamine, serotonine, and others. Fused heterocycles, bearing at the bedge of the indole nucleus a saturated azaheterocycle, are found in a wide range of biologically active indole alkaloids. The chemistry of hexahydropyrrolo[2,3-b]indoles, as exemplified by the calabar alkaloid (-)-physostigmine (1, Figure 1) and the analogues phenserine and (-)-eseroline,² has received considerable interest in drug research and for synthetic applications.³ Moreover, the hexahydropyrrolo[2,3-b]indole ring system is also present in akuammiline alkaloids, such as echitamine, which is a promising cytotoxic compound.⁴ Echibolines, incorporating the 1,2,3,4-tetrahydro-9a,4a-(iminoethano)-9H-carbazole structure 2,⁵ and 2,3,4,4a,9,9a-hexahydro-1*H*-pyrido[2,3-b] indoles,⁶ for example, compounds 3 and 4 with anti-inflammatory activity,⁷ also comprise structurally related classes of alkaloids with synthetic and medicinal potential. Benzo-annulated derivatives of hexahydro-1Hpyrido[2,3-b]indoles, that is, 5a,6,10b,11-tetrahydro-5Hquinindolines, like structure 5,8 are of interest within the synthetic study of communes in A (6a) and cytotoxic communesin B (**6b**),⁹ whereas the homologues of 2,3,4,4a,9,9a-hexahydro-1H-pyrido[2,3-b]indoles with a 1,2,3,4,5,5a,10,10a-octahydroazepino[2,3-b]indole (7) unit, have received some attention because of their sedative properties¹⁰ and as photodiscoloration prevention agents.11

In the present paper, the straightforward synthesis and study of the ring-chain transformations of tetracyclic compounds with a hitherto unreported 5,5a,6,7,12,12ahexahydroindolo[2,3-c][2]benzazepine (8) system is described. These syntheses, based on a 2-cyanobenzylation of indoles at C-3, followed by reductive or hydrolytic cyclization, can act as a model for the synthesis of alkaloid-



Figure 1

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like compounds since the 5,5a,6,7,12,12a-hexahydroindolo[2,3-*c*][2]benzazepine (**8**) structural element incorporates the aforementioned 1,2,3,4,5,5a,10,10a-octahydroazepino[2,3-*b*]indole (**7**) moiety as well as the 2,3,3a,4,9,10-hexahydro-1*H*-pyrrolo[2,3-*c*][2]benzazepine (**9**) scaffold which is also present in the structure of communesins **6**.^{9c}

The starting compounds, 2,3-dimethyl-1*H*-indole **10a** and 2,3,4,9-tetrahydro-1*H*-carbazole **10b**, prepared by Fischer indole synthesis, were alkylated at C-3 by treatment with ethylmagnesium iodide and subsequent reaction with 2-bromomethylbenzonitrile in benzene under reflux to give the corresponding 2-cyanobenzylated indoles **11a,b** in 53% and 76% yield, respectively (Scheme 1).¹² When the alkylation of 2,3-dimethylindole **10a** was performed in a mixture of benzene and diethyl ether, the yield of compound **11a** decreased (36% yield) and 12a-methyl-12,12a-dihydrobenzo[4,5]cyclohepta[1,2-*b*]indol-7(5*H*)-imine (**12a**) was formed as a minor product (17% yield).¹³

The 3*H*-indoles **11** were N-methylated in 78–91% yield upon reflux in iodomethane for 5 hours leading to 3*H*-indolium salts **13** which were crystallized from ethanol. These salts disclosed a typical N⁺=C signal at δ = ca. 195.0 ppm in the ¹³C NMR spectrum (DMSO-*d*₆).

The dehydroiodination of compounds **13** with sodium carbonate to afford the corresponding enamines, followed by reduction with LiAlH₄ in diethyl ether under reflux, resulted in cyclization of the intermediate amine across the enamine to afford indolo[2,3-c][2]benzazepines **14** (Scheme 1).¹⁴ The structure of 5,5a,12a-trimethylindo-lo[2,3-c][2]benzazepine **14a** was established by full spectroscopic investigations and the strong NOE interactions between the 5a-CH₃ and 12a-CH₃ groups suggested a *cis* relative configuration. It is necessary to point out that compounds **14a**,**b** in solution of CDCl₃ show a dynamic behavior, which leads to more or less broadening of most signals in the NMR spectra. The action of strong protonic acids on the indolo[2,3-c][2]benzazepines **14** leads to opening of the azepine ring with the formation of *3H*-in-

dolium salts **15**, which in turn form the initial cyclic adducts **14** under neutral conditions. The transformation of compounds **14a,b** to the open form dications **15a,b** in deuterated trifluoroacetic acid is indicated by the appearance of the characteristic signal of the indolium N⁺=C moiety in the ¹³C NMR spectrum at $\delta = 197.9$ and 199.3 ppm, respectively. The absorption maxima observed in the UV spectrum of compound **14a** in ethanol are specific for indoline derivatives [λ_{max} (log ε): 212 (2.99), 254 (2.86), 304 (2.51) nm]. The UV spectrum of compound **14a** in a 100:1 (v/v) mixture of ethanol with concd hydrochloric acid, when compared with the corresponding spectrum in pure ethanol, showed a hypsochromic displacement of the absorption bands [λ_{max} (log ε): 208 (3.05), 232 (2.84), 272 (2.73) nm], indicating that 3*H*-indolium salt **15** (R = Me, X = Cl) is formed.¹⁵

Hydrolysis of the cyano group of 3H-indole **11a** with concentrated sulfuric acid at 50 °C resulted in the formation of 2-(2,3-dimethyl-3H-indol-3-ylmethyl)benzamide (**16**) in 65% yield after basic aqueous work up (Scheme 2).

Applying the former hydrolytic conditions to compound **11b** afforded directly the indolo[2,3-*c*][2]benzazepinone **17**. The identification of the open amide **16** and the ringclosed lactam **17** was based on their full spectroscopic analysis with characteristic signals in the ¹³C NMR spectra at $\delta = 187.3$ (C=N) and 172.4 (C=O) ppm for the amide **16** (CDCl₃) and at $\delta = 79.5$ (aminal C-5a) and 171.5 (C=O) ppm for the lactam **17** (DMSO-*d*₆). In a solution of deuterated trifluoroacetic acid, the lactam **17** transforms to the 3*H*-indolium cation form **18**, possessing a characteristic N⁺=C signal at $\delta = 203.3$ ppm in the ¹³C NMR spectrum (TFA-*d*).

Acid hydrolysis of the substituted 3*H*-indolium iodides **13** followed by basic aqueous workup led to the formation of indolo[2,3-*c*][2]benzazepinones **19** (Scheme 3), the structure of which was fully proved by elemental and spectral analysis.¹⁶ The NOE between the 5a-CH₃ and 12a-CH₃ groups is indicative for the *cis*-configuration of structure **19a**.



Scheme 1

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18 X = CF₃COO or Cl

Scheme 2

Spectroscopic analysis of lactams **19** in deuterated trifluoroacetic acid, showing characteristic N⁺=C signals at δ = ca. 200.0 ppm, demonstrated again that treatment with protonic acids leads to cleavage of the C–N bond of the azepinone ring of compounds **19** with formation of the corresponding 3*H*-indolium salts **20**. Typical changes in the absorption bands of the UV spectra of compound **19a** in ethanol upon addition of concentrated hydrochloric acid also demonstrated the formation of the corresponding 3*H*-indolium chloride **20** (R = Me, X = Cl).



Scheme 3

In summary, it was found that alkylation of the magnesium salts of 2,3-disubstituted indoles with 2-bromomethylbenzonitrile and subsequent reduction or hydrolysis of the introduced cyano group resulted in a straightforward synthesis of a novel type of heterocyclic systems with an indolo[2,3-c][2]benzazepine and -benzazepinone skeleton, respectively. The structure and ring-chain transformations of the latter polycyclic alkaloid-type compounds were established by full spectroscopic and elemental analysis.

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References and notes

- Postdoctoral fellow of the Research Foundation-Flanders (FWO).
- (2) (a) Brossi, A.; Pei, X. F.; Greig, N. H. Austr. J. Chem. 1996, 49, 171. (b) Greig, N. H.; Pei, X. F.; Soncrant, T. T.; Ingram, D. K.; Brossi, A. Med. Res. Rev. 1995, 15, 3. (c) Robinson, B. Heterocycles 2002, 57, 1327. (d) Huang, A.; Kodanko, J. J.; Overman, L. E. J. Am. Chem. Soc. 2004, 126, 14043.
- (3) (a) Crich, D.; Banerjee, A. Acc. Chem. Res. 2007, 40, 151.
 (b) Takano, S.; Ogasawara, K. In *The Alkaloids*, Vol. 36; Brossi, A., Ed.; Academic Press: San Diego, 1989, 225–251.
- (4) Ramírez, A.; García-Rubio, S. Curr. Med. Chem. 2003, 10, 1891.
- (5) (a) Fritz, H.; Fischer, O. *Tetrahedron* 1964, 20, 1737.
 (b) Rees, J. M. H.; Cox, B.; Tanzil, S.; Newboult, L.; Kimber, K.; Robinson, B. *Adv. Biosci.* 1989, 75, 93.
 (c) Lévy, J.; Sapi, J.; Laronze, J. Y.; Royer, D.; Toupet, L. *Synlett* 1992, 601. (d) Dounay, A. B.; Humphreys, P. G.; Overman, L. E.; Wrobleski, A. D. *J. Am. Chem. Soc.* 2008, *130*, 5368. (e) Shen, L.; Zhang, M.; Wu, Y.; Qin, Y. *Angew. Chem. Int. Ed.* 2008, 47, 3618.
- (6) (a) Sunazuka, T.; Shirahata, T.; Tsuchiya, S.; Hirose, T.; Mori, R.; Harigaya, Y.; Kuwajima, I.; Omura, S. Org. Lett. 2005, 7, 941. (b) Snider, B. B.; Wu, X. Org. Lett. 2007, 9, 4913.
- (7) (a) Okamoto, T.; Akase, T.; Izumi, T.; Inaba, S.; Yamamoto, H. JP 47020196, **1972** *Chem. Abstr.* **1972**, *77*, 152142.
 (b) Cañas-Rodriquez, A.; Leeming, P. R. *J. Med. Chem.* **1972**, *15*, 762. (c) Matthews, N.; Franklin, R. J.; Kendrick, D. A. *Biochem. Pharmacol.* **1995**, *50*, 1053.
- (8) (a) May, J. A.; Zeidan, R. K.; Stoltz, B. M. *Tetrahedron Lett.* 2003, 44, 1203. (b) May, J. A.; Stoltz, B. *Tetrahedron* 2006, 62, 5262.
- (9) (a) George, J. H.; Adlington, R. M. Synlett 2008, 2093.
 (b) Fuchs, J. R.; Funk, R. L. J. Am. Chem. Soc. 2004, 126, 5068. (c) Siengalewicz, P.; Gaich, T.; Mulzer, J. Angew. Chem. Int. Ed. 2008, 47, 8170.
- (10) (a) Hester, J. B. J. Org. Chem. 1970, 35, 875. (b) Hester, J. B. US 3595874, 1971; Chem. Abstr. 1971, 75, 98552.
- (11) Kaneko, Y. JP 63163347, **1988**; Chem. Abstr. **1989**, 110, 144847.
- (12) For some examples on the alkylation of indolylmagnesium halides, see: (a) Jackson, A. H.; Smith, P. J. Chem. Soc. 1968, 1667. (b) Rodriguez, J. G.; San Andres, A. J. Heterocycl. Chem. 1991, 28, 1293. (c) Gruda, I.; Leblanc, R. M. Can. J. Chem. 1976, 54, 576.
- (13) Analytical and Spectroscopic Data for Compound 12a Yellow solid, mp >250 °C (from DMSO, with decomp.).
 ¹H NMR (500 MHz, DMSO-*d*₆): δ = 0.78 (3 H, s, CH₃), 3.04 (1 H, d, ²J = 14.2 Hz, 12-H), 3.09 (1 H, d, J = 14.2 Hz, 12-H), 3.40 (1 H, s, C=NH), 5.91 (1 H, s, 6-H), 6.34 (1 H, br s, NH), 6.96 (1 H, m, 2-H), 7.13 (2 H, m, 3-H, 4-H), 7.29 (1 H, d, J = 7.3 Hz, 1-H), 7.38–7.40 (3 H, m, 9-H, 10-H, 11-H),

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7.75 (1 H, m, 8-H). ¹³C NMR (125 MHz, DMSO- d_6): δ = 22.3 (CH₃), 39.7 (C-12), 51.2 (C-12a), 96.6 (C-6), 117.2 (C-4), 120.9 (C-1), 122.6 (C-2), 127.1 (C-9), 127.6 (C-3), 127.7 (C-8), 129.7 (C-10), 131.8 (C-11), 133.8 (C-7a), 136.4 (C-11a), 144.5 (C-12b), 151.8 (C-7), 155.3 (C-4a), 185.3 (C-5a). IR (KBr): 3430 (NH), 3310 (NH), 1655 (N=C) cm⁻¹. MS (ES⁺): m/z (%) = 262 (50) [M + 2H]⁺, 261 (100) [M + H]⁺. Anal. Calcd for C₁₈H₁₆N₂: C, 83.04; H, 6.19; N, 10.76. Found: C, 82.74; H, 5.99; N, 10.44.

(14) Typical Procedure for the Preparation of an Indolo[2,3c][2]benzazepine

A solution of 3*H*-indolium salt **13a** (0.5 g, 1.24 mmol) in EtOH (15 mL) was poured in a solution of 5% Na₂CO₃ (50 mL) and extracted with Et_2O (3 × 10 mL). The combined organic layers were washed with H₂O, dried over Na₂SO₄ and the solvent was evaporated under reduced pressure. The residue was dissolved in dry Et₂O (10 mL), LiAlH₄ (94 mg, 2.48 mmol) was added, and the mixture was refluxed under argon for 5 h. The reaction mixture was allowed to cool to r.t. and H₂O (1 mL) was dropped carefully into reaction flask. A finely suspended solid was filtered off using a fritted glass filter, and the solid material washed with Et₂O (20 mL). The filtrate was washed with H₂O, dried over Na₂SO₄ and concentrated under reduced pressure. The residue was purified by column chromatography (hexane-EtOAc, 7:1) to yield **14a** (0.19 g, 55%) as a viscous oil. ¹H NMR (500 MHz, CDCl₃): δ = 0.96 (3 H, s, 5a-CH₃), 1.15 (3 H, s, 12a-CH₃), 1.68 (1 H, s, NH), 2.36 (1 H d, ${}^{2}J$ = 14.3 Hz, 12-H), 2.72 (3 H, s, NCH₃), 3.56 (1 H, d, ${}^{2}J$ = 15.1 Hz, 7-H), 3.68 (1 H, d, $^{2}J = 14.3$ Hz, 12-H), 4.52 (1 H, d, $^{2}J = 15.1$ Hz, 7-H), 6.47 (1 H, d, J = 7.6 Hz, 4-H), 6.71 (1 H, t, J = 7.3 Hz, 2-H), 7.00 (1 H, d, J = 7.1 Hz, 1-H), 7.05 (1 H, m, 8-H), 7.11 (1 H, t, *J* = 7.6 Hz, 3-H), 7.14–7.19 (3 H, m, 9-H, 10-H, 11-H). ¹³C NMR (125 MHz, CDCl₃): δ = 18.7 (12a-CH₃), 18.9 (5a-CH₃), 27.0 (NCH₃), 44.7 (C-12), 45.3 (C-7), 46.2 (C-12a), 87.9 (C-5a), 106.5 (C-4), 117.5 (C-2), 121.1 (C-1), 126.3 (C-

8, C-9), 126.6 (C-10), 127.4 (C-3), 130.3 (C-11), 137.5 (C-12b), 138.4 (C-11a), 143.0 (C-7a), 149.1 (C-4a). ¹⁵N NMR (50.7 MHz, CDCl₃, ref.: MeNO₂): δ = -333.3 (N-6), -305.4 (N-3). IR (KBr): 3365 (NH) cm⁻¹. MS (ES⁺): *m/z* (%) = 280 (50) [M + 2H]⁺, 279 (100) [M + H]⁺. Anal. Calcd for C₁₉H₂₂N₂: C, 81.97; H, 7.97; N, 10.06. Found: C, 81.48; H, 7.60; N, 10.34.

- (15) Hinman, R. L.; Whipple, E. B. J. Am. Chem. Soc. 1962, 84, 2534.
- (16) Typical Procedure for the Preparation of an Indolo[2,3c][2]benzazepin-7(5H)-one

A solution of 3H-indolium salt 13a (0.5 g, 1.24 mmol) in concentrated H₂SO₄ (12 mL) was heated at 50 °C for 5 h. The mixture was poured onto crushed ice, neutralized with 10% KOH solution and extracted with Et₂O (3×15 mL). The combined organic layers were washed with H₂O, dried over Na₂SO₄, and the solvent was removed under reduced pressure. The residue was purified by column chromatography (hexane-EtOAc, 2:1) to yield 19a (0.195 g, 53%), mp 226–227 °C (from EtOH). ¹H NMR (500 MHz, CDCl₃): δ = 1.28 (3 H, s, 5a-CH₃), 1.45 (3 H, s, 12a-CH₃), 2.30 (3 H, s, NCH₃), 2.44 (1 H, br s, 12-H), 3.29 (1 H, d, ${}^{2}J$ = 12.7 Hz, 12-H), 5.85 (1 H, br s, 4-H), 6.54 (1 H, br s, 11-H), 6.66 (1H, br t, J = 7.4 Hz, 2-H), 6.82 (1 H, br s, NH), 6.94 (1 H, br t, J = 7.6 Hz, 3-H), 7.03 (1 H, br s, 10-H), 7.07 (1 H, d, J = 7.2 Hz, 1-H), 7.16 (1 H, br t, J = 7.3 Hz, 9-H), 7.57 (1 H, br d, J = 7.5 Hz, 8-H). ¹³C NMR (125 MHz, CDCl₃): δ = 19.5 (5a-CH₃), 21.5 (12a-CH₃), 26.1 (NCH₃), 47.6 (C-12), 55.1 (C-12a), 84.3 (C-5a), 104.5 (C-4), 117.1 (C-2), 121.9 (C-1), 126.5 (C-8, C-9), 128.1 (C-3), 128.9 (C-11), 130.1 (C-10), 131.8 (C-12b), 134.9 (C-7a), 136.3 (C-11a), 147.5 (C-4a), 172.8 (C=O). IR (KBr): 3180 (NH), 1650 (C=O) cm⁻¹. MS $(ES^+): m/z \ (\%) = 293 \ (100) \ [M + H]^+$. Anal. Calcd for C₁₉H₂₀N₂O: C, 78.05; H, 6.89; N, 9.58. Found: C, 78.31; H, 6.91; N, 9.65.