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Design, Synthesis, and Insecticidal Activity of Novel Doramectin Derivatives Containing Acylurea and Acylthiourea Based on Hydrogen Bonding

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23 ABSTRACT

Our recent investigation on the insecticidal activities of several doramectin 24 25 derivatives preliminarily revealed that the presence of hydrogen bonds at the C4" position of the molecule with target protein GABA receptor was crucial for retaining 26 high insecticidal activity. As a continuation of our research work on the development 27 of new insecticides, two series of novel acylurea and acylthiourea doramectin 28 derivatives were designed and synthesized. The bioassay results indicated that the 29 newly synthesized compounds (50, 5t, and 6t) exhibited higher insecticidal activity 30 31 against diamondback moth, oriental armyworm, and corn borer than the controls compounds doramectin, commercial avermectins, chlorbenzuron, and the lead 32 compound 3g in our laboratory. Specifically, compound 5t was identified as the most 33 34 promising insecticide against diamondback moth with a final mortality rate of 80.00% at the low concentration of 12.50 mg/L, showing approximately 7.75-fold higher 35 potency than the parent doramectin (LC₅₀ value of 48.1547 mg/L), 6.52-fold higher 36 potency than commercial avermeetins (LC₅₀ value of 40.5507 mg/L), and 3.98-fold 37 higher potency than 3g (LC₅₀ value of 24.7742 mg/L). Additionally, molecular 38 docking simulations revealed that **5t** (2.17, 2.20, 2.56, and 2.83 Å) displayed stronger 39 hydrogen bond action in binding with GABA receptor, better than that of **50** (1.64 and 40 2.15 Å) and 6t (2.20 and 2.31 Å) at the C4" position. This work demonstrated that 41 these compounds containing hydrogen bond groups might contribute to the 42 43 improvement of insecticidal activity and supplies certain hints toward structure optimization design for the development of new insecticides. 44

KEYWORDS: Acylurea, acylthiourea, insecticidal activities, molecular docking, 45 oriental armyworm, diamondback moth, and corn borer. 46

47 **INTRODUCTION**

Currently, Lepidoptera (diamondback moth, oriental armyworm, corn borer, etc.) is 48 becoming a major threat to the agricultural system.¹ To obtain higher food production, 49 many chemical and biological pesticides have been widely applied for control of 50 insect pests.²⁻⁸ In the quest for new insect control agents, natural products have been 51 and remain an excellent source of novel chemistry and inspiration for insecticides. 52 53 Among natural product-based insecticides, avermeetin, a macrocyclic lactone with high efficiency and without cross-resistance, is an important insecticide that has been 54 commercialized in crop protection and plays an important role in modern agriculture 55 56 by increasing both crop quality and yield while improving living standards due to a unique mechanism of action.⁹ To further identify more stable, efficient, low-toxicity 57 and broad-spectrum chemical compounds, scientists have made a large number of 58 59 structural modifications using natural avermectin components as parent compounds, and obtained many highly active derivatives.¹⁰⁻¹⁵ Among the avermectin-based 60 insecticides, doramectin, a third generation of the avermectins family with a 61 cyclohexyl group at C25 position in lieu of the sec-butyl or isopropyl of avermectins, 62 can aid in the identification of new mode of action and might be used as a lead 63 structure to derive new potent pesticides. Therefore, doramectin might be a promising 64 65 template for the discovery of new insecticide candidates.

66

The gamma-aminobutyric acid (GABA) receptor (GABAR) is a crucial target for

insecticide action. Studies suggested that the target of avermectin and doramectin was 67 the GABA-gated chloride channel, and this targeting caused hyperpolarization of 68 69 nerve membrane potential and inhibited nerve membrane, thus blocking normal nerve conduction and paralyzing insects.¹⁶ In previous papers from our laboratory, we 70 reported that with the introduction of carbamate, ester, sulfonate active groups to 71 replace the hydroxyl group at the C4" position of doramectin, compound **3g** (Figure 1) 72 containing cyclopropyl carbamate exhibited excellent insecticidal activities.¹⁷ A 73 molecular docking study indicated that 3g could form the N-H...O H bond, which 74 75 was crucial to the binding of the macrolide and GABAR, and the NH group on the C4" position of doramectin was necessary to retaining higher insecticidal activity. In 76 fact, Yamamoto,¹⁸ Kagabu,¹⁹ Casida,²⁰ and Qian²¹ demonstrated the importance of 77 78 the hydrogen bond between pesticide molecules and target proteins in various modes. To the best of our knowledge, use of acylureas and acylthioureas has reported for 79 many years in insecticidal, herbicidal, fungicidal applications for crop protection.²²⁻²⁴ 80 Acylureas and acylthioureas contain two NH groups, which are also highly 81 susceptible to additional hydrogen bonds with the target protein.^{25, 26} With this 82 characteristic in mind, we speculated whether the insecticidal activity could be 83 enhanced by forming additional hydrogen bonds through the introduction of acylurea 84 and acylthiourea moieties. 85

Herein, as a continuation of our work and to verify our hypothesis, a series of novel acylurea and acylthiourea derivatives were rationally designed and synthesized by replacing the hydroxyl group at the C4" position with acylurea and acylthiourea

active groups (Figure 1). The insecticidal activities of these target compounds were 89 tested against diamondback moth, oriental armyworm, and corn borer. Furthermore, 90 91 structure-activity relationship (SAR) studies and docking analysis were also discussed. 92

93

MATERIALS AND METHODS

Chemicals. Doramectin purchased from Chongqing Oiantai 94 was Biopharmaceutical Co., Ltd. (Chongqing, 95 China). Imidazole, triphosgene, tert-butyldimethylsiyl chloride (TBDMS-Cl), and N, N-dimethypyridin-4-amine 96 97 (DMAP) were purchased from Energy Chemical Ltd. (Beijing, China). All chemical chlorides, phenyl dichlorophosphate, materials amides and acid sodium 98 cyanoborohydride, ammonium thiocyanate, oxalyl chloride, and dry dichloromethane 99 100 (DCM) were purchased from J&K Chemical Ltd. (Beijing, China). Triethylamine, p-toluenesulfonic acid (TsOH), sodium carbonate, magnesium sulfate, methanol, 101 dichloromethane (DCM), and dimethyl sulfoxide (DMSO) were purchased from 102 103 Kelong Chemical Reagent Co., Ltd. (Chengdu, China). All reagents and solvents were reagent grade without further purification. 104

Instruments. The melting points were determined on an X-6 precision 105 micro-melting point apparatus (Beijing Fukui Technology Development Co., Ltd). ¹H 106 NMR (400 MHZ) and ¹³C NMR (100 MHZ) spectra were obtained using a Bruker 107 400 spectrometer in CDCl₃ solution with tetramethylsilane as the internal standard. 108 109 Reaction progress was monitored according to thin-layer chromatography (TLC) on silica gel GF254 with ultraviolet (UV) detection. High resolution mass spectra 110

111 (HRMS) data were recorded on a Micro Q-TOF II mass spectrometer (HR-ESI-MS,

112 Bruker, Germany) in the negative ion detection mode.

General Synthesis. The silica gel chromatography was performed with a column of 254 mm × 26 mm i.d. (Synthware glass Co. Ltd., Beijing, China) using 100-140 mesh silica gel (Sinopharm Chemical reagent Co. Ltd., Shanghai, China). The general synthetic methods for doramectin derivatives containing acylurea (5a-5v) and acylthiourea (6a-6v) groups are shown in Schemes 1 and Schemes 2, and their structures are listed in S-1 (Supporting Information). Acyl isocyanate and acyl isothiocyanate were prepared according to the methods in the literature.^{27, 28}

120 General Procedure for Synthesis of Acylurea Derivatives of Doramectin 121 (5a-5v) (Scheme 1).

Synthesis of [5-O-(tert-butyldimethylsilyl)doramectin](2).29 To a solution of 122 doramectin (20.00 g, 22.24 mmol) in dry dichloromethane (200.00 mL), imidazole 123 (15.10 g, 222.00 mmol), N, N-dimethypyridin-4-amine (DMAP, 271.00 mg, 2.22 124 125 mmol), and tert-butyldimethylsiyl chloride (TBDMS-Cl, 11.72 g, 77.77 mmol) were added. The mixture was stirred for 15 h at room temperature. When the reaction was 126 completed according to TLC analysis, water (200.00 mL) and dichloromethane 127 (200.00 mL) were added to the mixture. The aqueous phase was extracted with 128 dichloromethane $(3 \times 100.00 \text{ mL})$, and the combined organic phases were washed 129 with saturated brine $(3 \times 100.00 \text{ mL})$, dried over anhydrous magnesium sulfate, 130 filtered, and concentrated by evaporation under vacuum to yield a white solid. The 131 residue was purified by column chromatography on silica gel eluted with petroleum 132

133	ether/ethyl acetate (2: 1, v/v) to produce 19.48 g (86 %) of compound 2 as a white
134	foamy solid. ¹ H NMR (400 MHz, CDCl ₃) & 5.88-5.80 (m, 1H, H9), 5.79-5.66 (m, 3H,
135	H10, H11, H23), 5.53 (dd, J = 9.9, 2.6 Hz, 1H, H22), 5.41-5.32 (m, 3H, H3, H19,
136	H1"), 5.04-4.97 (m, 1H, H15), 4.78 (dd, J = 4.0, 1.3 Hz, 1H, H1'), 4.68 (dd, J = 14.5,
137	2.4 Hz, 1H, H8a-a), 4.58 (dd, J = 14.5, 2.3 Hz, 1H, H8a-b), 4.46-4.39 (m, 1H, H4"),
138	4.12 (d, J = 13.1 Hz, 1H, H13), 3.93 (s, 1H, H7-OH), 3.90-3.72 (m, 4H, H17, H5',
139	H5", H5), 3.61 (ddd, <i>J</i> = 11.0, 8.4, 4.6 Hz, 1H, H3'), 3.51-3.44 (m, 1H, H3"), 3.41 (d,
140	J = 2.9 Hz, 6H, H3'-OMe, H3"-OMe), 3.39 (d, J = 2.4 Hz, 1H, H6), 3.33-3.11 (m, 3H,
141	H2, H25, H4'), 2,58 (s, 1H, H4"-OH), 2.55-2.47 (m, 1H, H12), 2.37-2.19 (m, 5H,
142	H16, H18a, H2'a, H2"a), 2.04-1.96 (m, 1H, H20a), 1.83-1.75 (m, 6H, H4a-CH ₃ , H27a,
143	H30a, H31a), 1.73-1.55 (m, 5H, H24, H28a, H26, H29), 1.52-1.44 (m, 4H,
144	H14a-CH ₃ , H20b), 1.32-1.10 (m, 15H, H2"b, H27b, H28b, H30b, H31b, H2b,
145	H5'-Me, H5"-Me, H12a-CH ₃), 0.98-0.79 (m, 13H, H24a-CH ₃ , H18b, H-C(CH ₃) ₃),
146	0.13 (s, 6H, H-Si(CH ₃) ₂). ¹³ C NMR (100 MHz, CDCl ₃) δ 173.9, 140.1, 137.6, 137.5,
147	136.1, 135.1, 127.7, 124.7, 119.4, 118.2, 117.2, 98.4, 95.7, 94.8, 81.8, 80.4, 80.2,
148	80.1, 79.3, 78.2, 77.2, 76.1, 69.5, 68.3, 68.2, 68.1, 67.9, 67.2, 56.5, 56.4, 45.7, 40.3,
149	39.6, 38.7, 36.6, 34.6, 34.4, 34.1, 31.4, 30.0, 27.0, 26.9, 26.6, 26.5, 25.8(3-C), 25.5,
150	20.3, 20.0, 18.4, 17.6, 16.6, 15.2, -4.5, -4.8.

151 *Synthesis of [4"-O-5-O-(tert-butyldimethylsilyl)doramectin*].³⁰ Dimethyl 152 sulfoxide (DMSO, 25.00 mL) and triethylamine (TEA, 60.00 mL) were added to a 153 solution of compound **2** (15.00 g, 14.80 mmol) in 100.00 mL of dry 154 dichloromethane. A solution of PhOPOCl₂ (22.00 mL) in CH₂Cl₂ (100.00 mL) was added dropwise over 60 minutes to the mixture at -15 °C, which was stirred at -15 °C overnight until TLC indicated that the reaction was completed. The reaction solution was poured into 1% H₃PO₄, the aqueous phase was extracted with dichloromethane $(3 \times 100.00 \text{ mL})$, and the combined organic layers were washed with saturated brine $(3 \times 100.00 \text{ mL})$, dried over anhydrous magnesium sulfate, filtered, and evaporated to yield a yellow solid (14.21 g), and the product was not purified for use in the next step.

Synthesis of [4"-epi-NH₂-5-O-(tert-butyldimethylsilyl)doramectin].³⁰ A mixture of 162 compound 3 (14.00 g, 13.8 mmol), methanol (200.00 mL), and ammonium acetate 163 (5.32 g, 69.00 mmol) was stirred 1 h at room temperature, and sodium 164 cyanoborohydride (4.33 g, 69.00 mmol) was added. The mixture was stirred for 4 h at 165 166 room temperature until the reaction was completed according to TLC analysis. Water (100.00 mL), saturated sodium carbonate (100.00 mL), and ethyl acetate (200.00 mL) 167 were added, the aqueous phase was extracted with ethyl acetate $(3 \times 100.00 \text{ mL})$, and 168 169 the combined organic phases were washed with saturated sodium chloride solution (3 × 100.00 mL), dried over anhydrous magnesium sulfate, filtered, and evaporated to 170 yield a yellow solid. The residue was purified by column chromatography on silica 171 gel eluted with dichloromethane/methanol (20: 1, v/v) to produce 5.38 g (38 %) of 172 intermediate compound 4 as a yellow solid. 173

Synthesis of (4"-N- benzoylurea doramectin) (5a).^{27, 29} A mixture of benzoylamide
(1.21 g, 10.00 mmol), 1,2-dichloroethane (1.75 mL, 20.00 mmol), and oxalyl chloride
(1.75 mL, 20.00 mmol) was stirred overnight at 84 °C under an N₂ atmosphere. The

residue was concentrated in vacuo to benzoyl isocyanate (1.12 g) as a colorless liquidand was not purified for use in the next step.

Benzoyl isocyanate (218.00 mg, 1.48 mmol) was added to a solution of compound 179 4 (300.00 mg, 0.29 mmol) and N, N-dimethypyridin-4-amine (DMAP, 3.60 mg) in dry 180 dichloromethane (20.00 mL). The mixture was stirred for 2 h at room temperature 181 until the reaction was completed according to TLC analysis. Saturated sodium 182 bicarbonate (20.00 mL) was added, the aqueous phase was extracted with 183 dichloromethane $(3 \times 20.00 \text{ mL})$, and the combined organic phases were washed with 184 185 saturated sodium chloride solution $(3 \times 20.00 \text{ mL})$, dried over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The residue was purified by column 186 chromatography on silica gel eluted with petroleum ether/ethyl acetate (4: 1, v/v) to 187 188 produce 280.00 mg as a yellow solid. A deprotection reagent solution of 7.50 mL of p-toluenesulfonic acid-methanol complex (0.02 g/mL) was added dropwise to a 189 solution of yellow solid (280.00 mg) in methanol (15.00 mL). The mixture was stirred 190 191 1 h at room temperature until the reaction was completed according to TLC analysis. Saturated sodium bicarbonate (30.00 mL) and dichloromethane (30.00 mL) were 192 added, the aqueous phase was extracted with dichloromethane $(3 \times 10.00 \text{ mL})$, and 193 the combined organic phases were washed with saturated brine $(3 \times 10.00 \text{ mL})$, dried 194 over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The crude 195 product was purified by column chromatography on silica gel eluted with petroleum 196 ether/ethyl acetate (4: 1, v/v) to produce 195.21 mg (78 %) of compound 5a. ¹H NMR 197 (400 MHz, CDCl₃) δ 8.95 (d, J = 9.8 Hz, 1H, CONHCO), 8.45 (s, 1H, CONH), 7.86 198

199	(d, <i>J</i> = 7.1 Hz, 2H, Ph), 7.63-7.56 (m, 1H, Ph), 7.49 (dd, <i>J</i> = 8.4, 7.0 Hz, 2H, Ph), 5.90
200	(d, J = 10.1 Hz, 1H, H9), 5.83-5.68 (m, 3H, H10, H11, H23), 5.54 (dd, J = 9.9, 2.5 Hz,
201	1H, H22), 5.50-5.32 (m, 3H, H3, H19, H1"), 5.01 (d, <i>J</i> = 10.6 Hz, 1H, H15), 4.79 (d,
202	J = 3.8 Hz, 1H, H1'), 4.69 (t, J = 2.9 Hz, 2H, H8a), 4.42-4.35 (m, 1H, H4"), 4.30 (t, J
203	= 7.3 Hz, 1H, H5), 4.16-4.09 (m, 1H, H5"), 4.07 (s, 1H, H7-OH), 3.97 (d, <i>J</i> = 6.2 Hz,
204	1H, H6), 3.94 (s, 1H, H13), 3.92-3.81 (m, 2H, H17, H5'), 3.74 (dt, <i>J</i> = 12.2, 4.4 Hz,
205	1H, H3"), 3.66-3.57 (m, 1H, H3'), 3.44 (s, 3H, H3'-OMe), 3.43 (s, 3H, H3"-OMe),
206	3.35-3.20 (m, 3H, H2, H25, H4'), 2.59-2.46 (m, 1H, H12), 2.37 (d, <i>J</i> = 8.2 Hz, 1H,
207	H5-OH), 2.33-2.19 (m, 4H, H16, H24, H2"a), 2.09-1.97 (m, 2H, H2'a, H20a), 1.88 (d,
208	<i>J</i> = 2.1 Hz, 3H, H4a-CH ₃), 1.83-1.72 (m, 3H, H27a, H30a, H31a), 1.74-1.64 (m, 2H,
209	H18a, H28a), 1.56 (d, <i>J</i> = 11.1 Hz, 3H, H26, H29), 1.49 (d, <i>J</i> = 8.1 Hz, 4H, H14a-CH ₃ ,
210	H20b), 1.31-1.14 (m, 15H, H2"b, H27b, H28b, H30b, H31b, H2'b, H5'-Me, H5"-Me,
211	H12a-CH ₃), 0.93 (d, $J = 7.1$ Hz, 3H, H24a-CH ₃), 0.88 (d, $J = 12.4$ Hz, 1H, H18b). ¹³ C
212	NMR (100 MHz, CDCl ₃) δ 173.5, 168.1, 154.8, 139.5, 138.0, 137.8, 136.2, 135.0,
213	132.9, 132.5, 128.7(2-C), 127.8(2-C), 124.7, 120.5, 118.2, 118.1, 98.7, 95.7, 94.9,
214	81.8, 81.0, 80.4, 79.2, 73.9, 68.4, 68.3, 68.2, 67.7, 67.1, 65.4, 56.7, 56.0, 52.5, 49.8,
215	45.7, 41.1, 40.4, 39.7, 38.6, 36.6, 34.6, 34.3, 31.8, 31.4, 30.0, 29.7, 26.9, 26.6, 26.5,
216	25.5, 20.2, 19.9, 18.3, 17.3, 16.6, 15.1. HRMS (ESI) m/z calcd. for $C_{58}H_{80}O_{15}N_2Na$:
217	(M+Na) ⁺ , 1067.5451; found 1067.5416.

The target compounds (**5b-5v**) were synthesized according to a procedure similar to that used for compound **5a.** Their HRMS, ¹H NMR, and ¹³C NMR data can be found in Supporting Information.

General Procedure for Synthesis of Acylthiourea Derivatives of Doramectin (6a-6v) (Scheme 2).

Synthesis of benzoyl isothiocyanate.²⁸ A mixture of benzoyl chloride (703.00 mg, 5.00 mmol), acetone (15.00 mL), and ammonium thiocyanate (381.00 mg, 5.00 mmol) was stirred 1 h at room temperature under an N_2 atmosphere. The mixture was filtered, and concentrated in vacuo to yield 600.00 mg of light yellow liquid, which was not purified for use in the next step.

Synthesis of (4"-N- benzoylthiourea doramectin) (6a).^{28, 29} Benzoyl isothiocyanate 228 (322.32 mg, 1.98 mmol) was added to a solution of compound 4 (400.00 mg, 0.39 229 mmol) and N, N-dimethypyridin-4-amine (DMAP, 4.80 mg, 0.04 mmol) in dry 230 dichloromethane (20.00 mL). The mixture was stirred for 1 h at room temperature 231 232 until the reaction was completed according to TLC analysis. Saturated brine (20.00 mL) was added, the aqueous phase was extracted with dichloromethane (3×20.00) 233 mL), and the combined organic phases were washed with saturated brine (3×20.00) 234 235 mL), dried over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The residue was purified by column chromatography on silica gel eluted with 236 petroleum ether/ethyl acetate (7:1, v/v) to produce 401.00 mg as a yellow solid. A 237 deprotection reagent solution of 15.00 mL of p-toluenesulfonic acid-methanol 238 complex (0.02 g/mL) was added dropwise to a solution of yellow solid (401.00 mg) in 239 methanol (15.00 mL). The mixture was stirred for 1 h at room temperature until the 240 reaction was completed according to TLC analysis. Saturated sodium bicarbonate 241 (50.00 mL) and dichloromethane (50.00 mL) were added, the aqueous phase was 242

243	extracted with dichloromethane (3 \times 10.00 mL), and the combined organic phases
244	were washed with saturated brine (3 \times 10.00 mL), dried over anhydrous magnesium
245	sulfate, filtered, and concentrated in vacuo. The crude product was purified by column
246	chromatography on silica gel eluted with petroleum ether/ethyl acetate (2.5: 1, v/v) to
247	produce 254.00 mg (70 %) of compound 6a . ¹ H NMR (400 MHz, CDCl ₃) δ 11.11 (d,
248	<i>J</i> = 9.8 Hz, 1H, CONHCS), 9.07 (s, 1H, CSNH), 7.88-7.82 (m, 2H, Ph), 7.64-7.59 (m,
249	1H, Ph), 7.51 (t, <i>J</i> = 7.6 Hz, 2H, Ph), 5.92-5.85 (m, 1H, H9), 5.83-5.67 (m, 3H, H10,
250	H11, H23), 5.57-5.50 (m, 2H, H22, H1"), 5.47-5.36 (m, 2H, H3, H19), 5.09 (dd, <i>J</i> =
251	10.0, 3.6 Hz, 1H, H4"), 5.03-4.98 (m, 1H, H15), 4.79 (d, J = 3.8 Hz, 1H, H1'),
252	4.73-4.62 (m, 2H, H8a), 4.33-4.24 (m, 1H, H5), 4.22-4.13 (m, 1H, H5"), 4.06 (s, 1H,
253	H7-OH), 3.97 (d, <i>J</i> = 6.2 Hz, 1H, H6), 3.94 (s, 1H, H13), 3.92-3.78 (m, 3H, H17, H5',
254	H3"), 3.62 (ddd, <i>J</i> = 11.2, 8.5, 4.8 Hz, 1H, H3'), 3.52 (s, 3H, H3"-OMe), 3.43 (s, 3H,
255	H3'-OMe), 3.34-3.21 (m, 3H, H2, H25, H4'), 2.57-2.49 (m, 1H, H12), 2.41 (s, 1H,
256	H5-OH), 2.27 (dddd, <i>J</i> = 18.9, 11.4, 5.8, 2.2 Hz, 4H, H16, H24, H2"a), 2.14 (dd, <i>J</i> =
257	13.7, 5.0 Hz, 1H, H2'a), 2.03-1.97 (m, 1H, H20a), 1.87 (t, <i>J</i> = 2.0 Hz, 3H, H4a-CH ₃),
258	1.84-1.74 (m, 3H, H27a, H30a, H31a), 1.71-1.63 (m, 2H, H18a, H28a), 1.56 (d, J =
259	8.2 Hz, 3H, H26, H29), 1.49 (d, J = 7.9 Hz, 4H, H14a-CH ₃ , H20b), 1.30-1.14 (m,
260	15H, H2"b, H27b, H28b, H30b, H31b, H2'b, H5'-Me, H5"-Me, H12a-CH ₃), 0.92 (d,
261	$J = 7.2$ Hz, 3H, H24a-CH ₃), 0.87-0.82 (m, 1H, H18b). ¹³ C NMR (100 MHz, CDCl ₃) δ
262	181.2, 173.5, 166.9, 139.5, 138.0, 137.8, 136.2, 135.0, 133.4, 131.8, 129.0(2-C),
263	127.7, 127.5(2-C), 124.7, 120.5, 118.2, 118.1, 98.6, 95.7, 94.8, 81.8, 81.0, 80.4, 79.2,
264	79.2, 73.8, 68.4, 68.2, 68.2, 67.7, 67.0, 65.7, 57.1, 56.6, 55.7, 45.7, 40.3, 39.7, 38.6,

36.7, 34.6, 34.3, 32.3, 31.4, 30.0, 29.6, 26.9, 26.6, 26.5, 25.5, 20.2, 19.9, 18.3, 17.2,
16.6, 15.1. HRMS (ESI) m/z calcd. for C₅₈H₈₀O₁₄SN₂Na: (M+Na)⁺, 1083.5222; found
1083.5187.

The target compounds (**6b-6v**) were prepared by following the same procedure as for **6a.** Their HRMS, ¹H NMR, and ¹³C NMR data are list in Supporting Information.

Biological Assay and Insecticidal Test. Diamondback moth, oriental armyworm, 270 and corn borer were used to examine the insecticidal activities of the target 271 compounds (5a-5v and 6a-6v) according to the previously reported methods.³¹⁻³³ For 272 273 each compound, 30 larvae (10 larvae per group) were used. The percentage of mortalities was evaluated 2 days after treatment. Each bioassay was repeated in 274 triplicate at 25 ± 1 °C. For comparative purposes, doramectin, commercial 275 276 avermectins, chlorbenzuron, diafenthiuron, and our previous work 3g were tested under the same condition. Assessments were made on a dead/alive basis, and 277 mortality rates were corrected using Abbott's formula.³⁴ Percentage mortalities were 278 279 based on a percentage scale of 0-100, where 0 was no activity, 100 was total kill.

Homology Modeling and Molecular Docking.¹⁷ In our previously reported study, the crystal structure of human glycine receptor alpha-3 (PDB ID 5TIN) was chosen as the template to build the 3D structure of *Plutella xylostella* GABA receptor. Schrodinger-Glide was used to investigate the binding mode between compounds and the target protein. The best-scoring pose judged by the Glide docking score was chosen and analyzed using Schrodinger software.

286 **RESULTS AND DISCUSSION**

Chemical Synthesis. As shown in Scheme 1 and Scheme 2, two series of novel 287 acylurea and acylthiourea derivatives of doramectin (5a-5v and 6a-6v) modified at 288 289 position C4"-OH were prepared. As an initial material, doramectin 1 was reacted with tert-butylchlorodimethylsilane (TBDMS-Cl) for selective protection of the hydroxyl 290 group at the 5-position of doramectin to prepare 5-O-TBDMS doramectin 2, as 291 described previously.²⁹ In an attempt to introduce the C4" acyl isocyanate and acyl 292 isothiocyanate side chain, the protected intermediate 2 was first oxidized using 293 diphenyl phosphate dichloride (PhOPOCl₂) in DMSO at low temperature to produce 294 295 4"-oxo-5-O- TBDMS doramectin (3). Subsequently, C4"=O was reductively aminated to C4"-NH₂ with ammonium acetate and sodium cyanoborohydride (NaCNBH₃) to 296 produce the key intermediate 4"-epi-NH2-5-O-TBDMS doramectin (4) at 38 % yield. 297 The amine intermediate 4 was first reacted with different newly prepared acvl 298 isocyanate (RCONCO) isothiocyanate (RCONCS) or acyl using N, 299 *N*-dimethypyridin-4-amine (DMAP) as a catalyst in a solution of dichloromethane, 300 which was further deprotected using *p*-toluenesulfonic acid-methanol complex, 301 according to published procedures.³⁰ All target compounds were purified by column 302 chromatography, and their structures were characterized by ¹H NMR, ¹³C NMR, and 303 HRMS. 304

Insecticidal Activities. Forty-four newly synthesized acylurea and acylthiourea doramectin derivatives (5a-5v and 6a-6v) were evaluated for insecticidal activities against diamondback moth and oriental armyworm, and a few representative compounds 5l, 5o, 5t, 6l, 6o, 6n, 6p, and 6t were further evaluated for insecticidal efficacy against corn borer, and the results are listed in **Tables 1**, **2** and **3**. The LC_{50} values were calculated using five different concentrations (200.00 mg/L, 100.00 mg/L, 50.00 mg/L, 25.00 mg/L, and 12.50 mg/L) and are listed in **Table 4**. Doramectin, commercial avermectins, chlorbenzuron, diafenthiuron, and **3g** were used as positive controls. Leaves treated with acetone were used as a blank control group.

Insecticidal Activity against Diamondback Moth. As shown in Table 1, acylurea 315 and acylthiourea derivatives of doramectin except for 5r, 5u, 6b, 6e, 6h, 6i, 6p, 6q, 316 and **6u**, exhibited insecticidal activities against diamondback moth at the 317 concentration of 100.00 mg/L. Moreover, the acylurea derivatives (5a-5e, 5g-5j, 318 5m-5q, and 5t-5v) had higher insecticidal activities than the corresponding 319 320 acylthiourea series (6a-6e, 6g-6j, 6m-6q, and 6t-6v). Furthermore, among all derivatives, compounds 5j, 5o, 5t, 6l, and 6o exhibited insecticidal activities equal to 321 or higher than that of the positive control, doramectin, commercial avermectins, and 322 323 chlorbenzuron. Particularly, 5j, 5o, 5t, and 6l displayed high insecticidal activities (\geq 80%) against diamondback moth at a concentration of 100.00 mg/L. When the 324 concentration was reduced to 12.50 mg/L, 50 and 5t exhibited desirable insecticidal 325 activities, and compound 5t still had 80.00% mortality. The insecticidal activity of 5t 326 against diamondback moth was much higher than that of the control compounds, 327 including doramectin, commercial avermectins, chlorbenzuron, and 3g (the most 328 329 active compound in our group), which was parallel to that of diafenthiuron at the same concentration of 12.50 mg/L. The LC₅₀ value of 5t (LC₅₀= 6.2123 mg/L) was 330

6.52-fold that of avermeetins (LC₅₀= 40.5507 mg/L), 7.75-fold that of dorameetin

332 (LC₅₀= 48.1547 mg/L), and 3.98-fold that of **3g** (LC₅₀= 24.7742 mg/L).

Insecticidal Activity against Oriental Armyworm. The insecticidal activities of all 333 compounds against oriental armyworm were evaluated, and the results are shown in 334 Table 2. Acylurea and acylthiourea doramectin derivatives, except for 5a, 5d, 5e, 5f, 335 5h, 5k, 5r, 5s and 5v, exhibited insecticidal activities against oriental armyworm at 336 concentrations of 200.00-50.00 mg/L. Unlike the insecticidal efficacy against 337 diamondback moth, the acylthiourea series were more active than the corresponding 338 339 acylurea series at the same concentration. Most acylthiourea derivatives displayed good to excellent insecticidal activities at high concentration (\geq 50.00 mg/L) and 340 certain insecticidal activities at low concentration (12.50 mg/L), whereas a few 341 342 acylurea derivatives (5e, 5f, 5k, and 5v) displayed no insecticidal activities against oriental armyworm at 200.00 mg/L, and only three compounds (51, 50, and 5t) had 343 insecticidal activities at the concentration of 12.50 mg/L. In addition, compounds 51, 344 345 50, 5t, 6l, 6n, and 6p exhibited good insecticidal activities against oriental armyworm with mortality rates of 56.67%, 50.00%, 60.00%, 56.67%, 53.67%, and 50.00%, 346 respectively, at 12.50 mg/L, which were superior to those of doramectin (20.00%), 347 avermectins (23.33%), and chlorbenzuron (40.00%) under the same concentration. 348 Interestingly, compounds 60 and 6t with mortality rates of 66.67% and 73.33% 349 showed a much more potent inhibitory effect against oriental armyworm than the 350 most effective 3g (60.00%), as reported in our previous paper. The insecticidal 351 activity of compound 6t against oriental armyworm was the highest, and the LC_{50} 352

value of **6t** (LC₅₀= 6.2816 mg/L) was 5.08-fold that of avermeetins (LC₅₀= 31.9488 mg/L), 6.21-fold that of dorameetin (LC₅₀= 39.0069 mg/L), and 1.29-fold that of **3g** (LC₅₀= 8.1391 mg/L).

Insecticidal Activity against Corn Borer. From Table 1 and Table 2, several 356 compounds (51, 50, 5t, 6l, 60, 6n, 6p, and 6t) exhibited excellent insecticidal activities 357 against diamondback moth or oriental armyworm among all synthesized compounds. 358 For further evaluation of the insecticidal efficacy of these compounds, the results of 359 insecticidal activities against corn borer are shown in Table 3. From Table 3, we 360 found that all compounds had insecticidal efficacy at concentrations from 200.00 to 361 25.00 mg/L, and only compound 6p lost insecticidal activity at a concentration of 362 12.50 mg/L. Additionally, compounds 50, 5t and 6t exhibited insecticidal activities 363 364 higher than those of doramectin, avermectins, chlorbenzuron and diafenthiuron. Specifically, compound **50** showed the highest insecticidal activity with a mortality 365 rate of 60.00% at a concentration of 12.50 mg/L, which was superior to those of 5t 366 (43.33%), 6t (40.00%), and the contrast 3g (36.67%) at the same concentration. The 367 LC₅₀ values of compounds 50, 5t, 6t and 3g against corn borer were 10.0931, 368 22.7966, 26.7018 and 23.0991 mg/L, respectively. Compound 50 exhibited better 369 insecticidal activity than 3g. 370

371 Structure-Activity Relationship (SAR). A comparison of the insecticidal 372 activities of different types of doramectin acylurea and acylthiourea derivatives is 373 shown in Table 1, Table 2, and Table 3. The overall contrasting trends are given as 374 follows: (1) insecticidal activities against diamondback moth: acylurea > acylthiourea

375	groups, such as, $5t >> 6t$, $5o > 6o$; 4-aromatic-substituted compounds (5g, 5j) are
376	superior to 2-aromatic-substituted compounds (5b-5d) and 3-aromatic-substituted
377	compounds (5e, 5f); cyclopropyl-substituted compound (50) is better than other
378	alkyl-substituted compounds (51-5n, 5p, 5q); (2) insecticidal activities against oriental
379	armyworm: acylthiourea > acylurea groups, such as, $6t > 5t$, $6o > 5o$, $6p >> 5p$;
380	alkyl-substituted acylthiourea compounds (6l, 6n-6p) are superior to monosubstituted
381	aromatic compounds (6a-6k); compound 6t has the highest activity; (3) insecticidal
382	activities against corn borer: $50 > 5t > 6t$.

Docking Analysis. To better elucidate the effect of the acylurea and acylthiourea 383 moieties on insecticidal activity against diamondback moth, we selected the sequence 384 of *Plutella xylostella* Rdl-1 for homology modeling to investigate the binding mode 385 386 between three compounds (50, 5t, and 6t) and the target protein using Schrodinger-Glide. Compounds 50, 5t, and 6t were docked into human glycine 387 receptor alpha-3, and the detailed interactions of the compounds and receptor are 388 shown in Figure 2. Due to the different substituents at C4" position, different types of 389 hydrogen bonds were formed. The acylureas and acylthioureas contained two NH 390 groups and were highly susceptible to more hydrogen bonds with the target protein. 391 As shown in Figure 2, two N–H···O H-bonds (1.64 and 2.15 Å for 50; 2.20 and 2.30 392 Å for 6t) were formed between two NH groups on acylurea and the oxygen atom of 393 Asp165. Impressively, four hydrogen bonds were formed between the acylurea of 394 compound 5t and the target protein. Two NH groups on acylurea of compound 5t 395 formed N-H…O H-bonds (2.17 and 2.56 Å) with the oxygen atom of Asp165, and 396

two CO groups on acylurea also formed O···H H-bonds (2.20 and 2.83 Å) with the hydrogen atom of Asn164. Among the three compounds, the hydrogen bonds between compound **5t** and the receptor protein in the docking simulation were the most numerous and strongest. Accordingly, compound **5t**, with the strongest H-bonding strengths, exhibited the highest insecticidal activity against diamondback moth compared with **5o** and **6t**.

To better explain the structure-activity relationship, we also calculated the binding 403 free energy, as shown in Table 5, and compared the binding conformations of 50 and 404 405 5t with those of the lead compound 3g (described in our previous study) and 6t, respectively. The results (Table 5) showed that the binding free energy of 50 (-3.637 406 kcal mol⁻¹) was comparable to that of **3g** (-3.964 kcal mol⁻¹), and the binding free 407 408 energy of 5t (-1.758 kcal mol⁻¹) was much lower than that of 6t (-0.821 kcal mol⁻¹). Additionally, the H-bonding strengths of **50** (1.64 and 2.15 Å) were comparable to 409 those of 3g (2.12 and 2.31 Å), whereas the H-bonding strengths of 5t (2.17, 2.20, 2.56 410 and 2.83 Å) were much better than those of 6t (2.20 and 2.30 Å) at the C4" position. 411 The docking results of compounds 50, 5t, 6t and 3g were in good agreement with our 412 insecticidal assays; the activity of 50 with acylurea was comparable to that of 3g with 413 carbamate and the activity of 5t with acylurea was better than that of 6t. From the 414 results above, it can be preliminarily concluded that the insecticidal activity of our 415 synthesized compounds had a positive correlation with the H-bonds at the C4" 416 417 position.



acylthiourea groups were designed and synthesized based on hydrogen bonds. The 419 results from bioassays of the insecticidal activities were evaluated. Compounds 5t, 6t, 420 421 and 50 exhibited excellent insecticidal activities against diamondback moth, oriental armyworm and corn borer, respectively. Molecular docking also showed that 422 compound **5t** formed stronger hydrogen bonds with the target protein, and the docking 423 result was consistent with the corresponding insecticidal activity test result. Therefore, 424 the rationality of the designed novel doramectin derivatives based on hydrogen 425 bonding was fully supported in our study. A future detailed study on the structural 426 427 optimization and mechanism of action are in progress at our laboratory.

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432 Supporting Information

Supplementary data associated with this article can be found in Supporting
Information. ¹H NMR and ¹³C NMR data and spectrums of compounds 5a-5v, 6a-6v
were available free of charge via the Supporting Information.

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- 556 **Figure captions**
- 557 **Figure 1.** Design of target compounds.
- 558 Figure 2. Molecular docking of compounds 50, 5t, and 6t.
- 559 Scheme 1. General Procedure for Synthesis of Acylurea Derivatives of Doramectin
- 560 (**5a-5v**).
- 561 Scheme 2. General Procedure for Synthesis of Acylthiourea Derivatives of
- 562 Doramectin (**6a-6v**).



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Figure 2. Molecular docking of compounds 50, 5t, and 6t. The H-bonding distance of 50 (1.64, 2.15 Å), 5t (2.17, 2.20, 2.56, 2.83 Å), and 6t (2.20, 2.30 Å) at the C4" position.

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581 Scheme 1. General Procedure for Synthesis of Acylurea Derivatives of 582 Doramectin (5a-5v).



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Table 1. Insecticidal Activities of Doramectin Derivatives Against Diamondback Moth.

Compd.	200.00 (mg/L)	100.00 (mg/L)	50.00 (mg/L)	25.00 (mg/L)	12.50 (mg/L)
5a	50.00%	36.67%	30.00%	10.00%	0
5b	56.67%	33.33%	0	0	0
5c	70.00%	60.00%	30.00%	23.33%	0
5d	100.00%	60.00%	30.00%	0	0
5e	40.00%	10.00%	0	0	0
5f	43.33%	20.00%	13.33%	0	0
5g	80.00%	60.00%	50.00%	36.66%	30.00%
5h	46.67%	20.00%	10.00%	0	0
5i	76.67%	60.00%	33.33%	20.00%	10.00%
5j	100.00%	83.33%	70.00%	40.00%	30.00%
5k	30.00%	10.00%	0	0	0
51	33.33%	20.00%	0	0	0
5m	90.00%	80.00%	63.33%	30.00%	0
5n	70.00%	50.00%	36.67%	23.33%	10.00%
50	90.00%	80.00%	63.33%	50.00%	40.00%
5p	63.33%	43.33%	40.00%	23.33%	10.00%
5q	40.00%	20.00%	10.00%	0	0
5r	0	0	0	0	0
5 s	36.67%	20.00%	0	0	0
5t	100.00%	100.00%	100.00%	90.00%	80.00%
5u	0	0	0	0	0
5v	50.00%	20.00%	0	0	0
6a	30.00%	26.67%	20.00%	0	0
6b	0	0	0	0	0
6c	30.00%	13.67%	0	0	0
6d	50.00%	30.00%	0	0	0
6e	0	0	0	0	0
6f	70.00%	53.33%	40.00%	33.33%	0
6g	73.33%	40.00%	30.00%	10.00%	0
6h	0	0	0	0	0
6i	0	0	0	0	0
6j	40.00%	33.33%	0	0	0
6k	63.33%	46.67%	20.00%	10.00%	0
61	90.00%	80.00%	53.33%	30.00%	20.00%
6m	50.00%	36.33%	30.00%	0	0
6n	56.67%	50.00%	10.00%	0	0
60	76.67%	50.00%	40.00%	36.33%	20.00%
6р	0	0	0	0	0

	6q	0	0	0	0	0
	6r	40.00%	30.00%	26.67%	10.00%	0
	6s	50.00%	26.67%	0	0	0
	6t	66.67%	53.33%	40.00%	30.00%	13.33%
	6u	0	0	0	0	0
	6v	50.00%	10.00%	0	0	0
	3g	100.00%	86.67%	63.33%	50.00%	33.33%
	doramectin	93.33%	66.67%	46.67%	30.00%	16.67%
	avermectins	90.00%	70.00%	53.33%	40.00%	20.00%
	chlorbenzuron	60.00%	30.00%	0	0	0
	diafenthiuron	100.00%	100.00%	100.00%	80.00%	80.00%
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Table 2. Insecticidal Activities of Doramectin Derivatives Against Oriental Armyworm.

Compd.	200.00 (mg/L)	100.00 (mg/L)	50.00 (mg/L)	25.00 (mg/L)	12.50(mg/L)
5a	60.00%	40.00%	0	0	0
5b	90.00%	66.67%	20.00%	0	0
5c	73.33%	60.00%	33.33%	10.00%	0
5d	30.00%	20.00%	0	0	0
5e	0	0	0	0	0
5f	0	0	0	0	0
5g	40.00%	20.00%	10.00%	0	0
5h	30.00%	0	0	0	0
5i	56.67%	40.00%	13.33%	0	0
5j	80.00%	60.00%	46.33%	0	0
5k	0	0	0	0	0
51	100.00%	86.67%	80.00%	63.33%	56.67%
5m	80.00%	56.33%	40.00%	20.00%	0
5n	100.00%	80.00%	50.00%	0	0
50	100.00%	83.33%	80.00%	60.00%	50.00%
5p	36.67%	30.00%	20.00%	0	0
5q	76.33%	60.00%	40.00%	0	0
5r	50.00%	26.67%	0	0	0
5 s	50.00%	30.00%	0	0	0
5t	100.00%	100.00%	86.67%	70.00%	60.00%
5u	70.00%	60.00%	20.00%	0	0
5v	0	0	0	0	0
6a	100.00%	80.00%	56.67%	46.67%	16.67%
6b	90.00%	70.00%	53.33%	26.67%	10.00%
6c	100.00%	80.00%	60.00%	43.33%	20.00%
6d	80.00%	60.00%	40.00%	30.00%	10.00%
6e	73.33%	60.00%	50.00%	30.00%	16.67%
6f	100.00%	73.33%	60.00%	30.00%	0
6g	90.00%	73.33%	40.00%	30.00%	20.00%
6h	100.00%	80.00%	56.67%	40.00%	10.00%
6i	76.67%	70.00%	60.00%	30.00%	0
6j	80.00%	63.33%	46.67%	20.00%	0
6k	90.00%	73.33%	60.00%	40.00%	0
61	100.00%	90.00%	80.00%	60.00%	56.67%
6m	70.00%	60.00%	30.00%	0	0
6n	100.00%	90.00%	76.67%	60.00%	53.67%
60	100.00%	100.00%	83.33%	76.67%	66.67%

6р	100.00%	90.00%	76.67%	56.67%	50.00%
6q	100.00%	70.00%	50.00%	40.00%	10.00%
6r	100.00%	60.00%	50.00%	36.67%	0
6 s	50.00%	23.33%	20.00%	0	0
6t	100.00%	100.00%	86.67%	80.00%	73.33%
6u	63.33%	43.33%	30.00%	0	0
6v	90.00%	80.00%	46.67%	30.00%	10.00%
3g	100.00%	86.67%	80.00%	76.67%	60.00%
doramectin	100.00%	83.33%	46.67%	30.00%	20.00%
avermectins	100.00%	90.00%	53.33%	40.00%	23.33%
chlorbenzuron	100.00%	90.00%	73.33%	50.00%	40.00%
diafenthiuron	100.00%	100.00%	100.00%	100.00%	100.00%

621 **Table 3.** Insecticidal Activities of Doramectin Derivatives Against Corn Borer.

Compd.	200.00 (mg/L)	100.00 (mg/L)	50.00 (mg/L)	25.00 (mg/L)	12.50 (mg/L)
51	83.33%	70.00%	50.00%	36.67%	20.00%
50	100.00%	86.67%	80.00%	66.67%	60.00%
5t	93.33%	73.33%	60.00%	50.00%	43.33%
61	63.33%	43.33%	30.00%	20.00%	16.67%
60	60.00%	46.67%	36.67%	20.00%	10.00%
6n	60.00%	40.00%	33.33%	20.00%	10.00%
6р	70.00%	56.67%	36.67%	20.00%	0
6t	90.00%	66.67%	60.00%	46.67%	40.00%
3g	100.00%	83.33%	73.33%	46.67%	36.67%
doramectin	90.00%	73.33%	60.00%	43.33%	20.00%
avermectins	86.67%	53.33%	43.33%	36.67%	13.33%
chlorbenzuron	70.00%	56.67%	30.00%	10.00%	0
diafenthiuron	70.00%	60.00%	40.00%	26.67%	10.00%

622

623 Table 4. LC₅₀ Values of 5t, 6t, 5o, 3g, Doramectin, and Avermectins against Oriental armyworm,

624	Diamondback	Diamondback moth and Corn borer.						
		Diamondback moth		Oriental armyworm		Corn borer		
		LC ₅₀ (mg/L)	toxic ratio	$LC_{50}(mg/L)$	toxic ratio	$LC_{50}(mg/L)$	toxic ratio	
	5t	6.2123	6.52	11.2172	2.84	22.7966	2.54	
	6t	83.5867	0.48	6.2816	5.08	26.7018	2.17	
	50	22.8235	1.77	14.6415	2.18	10.0931	5.74	
	3g	24.7742	1.63	8.1391	3.92	23.0991	2.50	
	doramectin	48.1547	0.84	39.0069	0.81	36.2447	1.59	
	avermectins	40.5507	1.00	31.9488	1.00	57.9459	1.00	

625

626 Table 5. Glide Docking Score of 3g, 5o, 5t, and 6t.

Compound	Glide Docking Score (kcal mol ⁻¹)
3g	-3.964
50	-3.637
5t	-1.758
6t	-0.821

627 Table of Contents Graphic:

