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Research paper

The quest of the best – A SAR study of trithiolato-bridged dinuclear Ruthenium(II)-Arene compounds presenting antiparasitic properties



1987

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ABSTRACT

A structure activity relationship (SAR) study of a library of 56 compounds (54 ruthenium and 2 osmium derivatives) based on the trithiolato-bridged dinuclear ruthenium(II)-arene scaffold (general formula $[(\eta^6-\text{arene})_2\text{Ru}_2(\mu_2-\text{SR})_3]^+$, symmetric and $[(\eta^6-\text{arene})_2\text{Ru}_2(\mu_2-\text{SR}^1)_2(\mu_2-\text{SR}^2)]^+$, mixed, respectively) is reported. The 56 compounds (of which 34 are newly designed drug candidates) were synthesized by introducing chemical modifications at the level of bridge thiols, and they were grouped into eight families according to their structural features. The selected fittings were guided by previous results and focused on a fine-tuning of the physico-chemical and steric properties. Newly synthesized complexes were characterized by NMR spectroscopy, mass spectrometry and elemental analysis, and four single-crystal X-ray structures were obtained.

The *in vitro* biological assessment of the compounds was realized by applying a three-step screening cascade: (i) evaluation of the activity against *Toxoplasma gondii* RH strain tachyzoites expressing β -galactosidase (*T. gondii*- β -gal) grown in human foreskin fibroblast monolayers (HFF) and assessment of toxicity in non-infected HFF host cells; (ii) dose-response assays using selected compound, and (iii) studies on the effects in murine splenocytes.

A primary screening was performed at 1 and 0.1 μ M, and resulted in the selection of 39 compounds that inhibited parasite proliferation at 1 μ M by more than 95% and reduced the viability of HFF by less than 49%. In the secondary screening, dose-response assays showed that the selected compounds exhibited half maximal inhibitory concentration (IC₅₀) values for *T. gondii*- β -gal between 0.01 μ M and 0.45 μ M, with 30 compounds displaying an IC₅₀ lower than 0.1 μ M. When applied to non-infected HFF monolayers at 2.5 μ M, 8 compounds caused more than 90% and 31 compounds more than 30% viability impairment. The tertiary screening included 14 compounds that did not cause HFF viability loss higher than 50% at 2.5 μ M. These derivatives were assessed for potential immunosuppressive activities. First, splenocyte viability was assessed after treatment of cells with concanvalin A (ConA) and lipoplysaccharide (LPS) with compounds applied at 0.1 and 0.5 μ M. Subsequently, the 5 compounds exhibiting the lowest splenocyte toxicity were further evaluated for their potential to inhibit B and T cell proliferation.

Overall, compound **55** $[(\eta^6-p-MeC_6H_4Pr^i)_2Ru_2(\mu_2-SC_6H_4-o-CF_3)_2(\mu_2-SC_6H_4-p-OH)]Cl exhibited the most favorable features, and will be investigated as a scaffold for further optimization in terms of anti-parasitic efficacy and drug-like properties.$

1. Introduction

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Despite the fact that parasites inflict widespread and important diseases in animals as well as in humans, drug development against parasitic diseases remains, with few exceptions, a largely neglected area. Many currently employed antiparasitic therapies exhibit



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important setbacks, such as adverse side effects and low efficacy, and the occurrence of resistance is being recognized as a serious constraint [1]. Consequently, the development of novel compounds is highly needed.

Toxoplasma gondii (T. gondii), a member of the phylum Apicomplexa, is an obligate intracellular protozoan that infects most mammalian species [2,3] and birds [4]. Toxoplasma has a high zoonotic potential, and it is estimated that up to 30% of the human population is infected [5,6]. However, serious disease affects only a fraction of infected persons. For example, primary infection during pregnancy can cause congenital toxoplasmosis [7,8], and foetal infection may lead to severe complications, including spontaneous miscarriage and malformations such as hydrocephalus, developmental retardation, and ocular toxoplasmosis causing eyesight impairment and even blindness in later life [9,10]. Likewise, reactivated toxoplasmosis, or acquisition of a primary infection, in immunocompromised patients will lead to acute toxoplasmosis, multiple organ failure and possibly death [11,12]. Most importantly, T. gondii is a major economic burden for the food industry, in that it affects food safety and food security by infecting most farm animal species used in food production [13].

The most frequently applied treatments for toxoplasmosis rely mainly on the use of a combination of pyrimethamine and sulfadiazine, supplemented with folinic acid to alleviate folic acid deficiency in the host [14]. Additional treatment options include trimethoprim/sulfamethoxazole, clindamycin, spiramycin and atovaquone [14]. Nevertheless, these drugs were initially developed for other diseases [15–18], and they are either of moderate efficacy and/or are prone to adverse side effects. Additionally, the current treatments target only tachyzoites, the rapidly replicating stage, while the latent bradyzoite cyst stage remains unaffected [2]. The increasing number of reports on treatment failures suggests that drug resistant strains are emerging [19]. Concluding, there is an urgent need for new, safe, low-cost and efficacious treatments against *T. gondii*.

Organometallic compounds could potentially provide interesting alternatives to current treatment options [20–26]. Following the pathway opened by cisplatin, important attention was given to the development of various metalorganic derivatives as prospective anticancer agents [27–35]. Among other ruthenium-based complexes, trithiolato-bridged dinuclear ruthenium(II)-arene compounds have shown high cytotoxicity against various cancer cell lines, as well as promising *in vivo* activity when assessed in tumor mouse models [36]. Besides anticancer activity, numerous metalbased compounds also display considerable antiparasitic properties [23,24,37–40].

Recent studies have shown promising anti-*Toxoplasma* activities of various metal-based compounds [41–52]. For example auranofin (**A**, Fig. 1), a gold(I)-based complex approved for rheumatoid arthritis treatment, was shown to have a broad-spectrum antiparasitic activity against protozoan parasites [53] and to be effective against acute toxoplasmosis *in vitro* and *in vivo* [41].

Various coordination complexes of first row transition metal ions were shown to be active against the highly virulent *T. gondii* RH strain, with low cytotoxicity for non-infected mammalian cells, e.g., iron(III) complex **B** (Fig. 1) [42], and a cobalt(II) complex [43] presenting similar organic ligands. Other compounds active against *Toxoplasma* include mononuclear copper(II) complexes **C1** and **C2** [44], with different naphthyl substituents (Fig. 1).

Coordination complexes of zinc(II) [45] or iron(III) [46] (Fig. 1, **D**) with sulfadiazine (drug used in combination with pyrimethamine for the treatment of toxoplasmosis) [14,15] exhibited better activity against *T. gondii* RH tachyzoites *in vitro* and lower host cell toxicity compared to sulfadiazine alone.

Compounds E1/E2 [47] and F [48] (Fig. 1) comprise covalently

linked ferrocenyl moieties to bioactive ligands based on thiazolidinone units or atovaquone, and are efficient against *Toxoplasma* [54,55]. SAR (structure activity relationship) studies demonstrated that small modifications of the bioactive ligand were associated with important changes in antiparasitic activity and general cytotoxicity. For instance, ferrocenyl aminohydroxynaphtoquinone derivatives (Fig. 1, F) presented a slightly lower activity compared to atovaquone against the atovaquone sensitive *T. gondii* PLK strain, but an improved activity against the atovaquone resistant ATO strain. Only compounds presenting a certain length of the hydrophobic chain, and thus an associated lipophilicity, exhibited interesting antiparasitic activity, without exerting cytotoxicity in HFF (human foreskin fibroblast).

A range of hydrolytically stable half-sandwich η^6 -arene ruthenium(II) phosphite compounds [49,56] were previously evaluated for their in vitro activities against protozoan parasites such as T. gondii and the closely related Neospora caninum, and the helminth Echinococcus multilocularis. Results on T. gondii and N. caninum demonstrated that two compounds, namely G1 and G2 (Fig. 1) exhibited a highly selective inhibitory effect on in vitro proliferation of tachyzoites without harming host cells. These findings prompted to further biological activity assessment of a small library of trithiolato-bridged dinuclear ruthenium(II)-arene complexes against protozoan parasites including T. gondii [50], N. caninum [57] and Trypanosoma brucei [58]. Three compounds H1–H3 (Fig. 1) were highly active against *T. gondii* with IC₅₀ values (half maximal inhibitory concentrations) in low nanomolar range (34, 62 and 1.4 nM, respectively). Moreover, the viability of human foreskin fibroblast cells (HFF used as host cells) was not affected subsequent to incubation with high doses of complex H1 or H2 (~200 µM), whereas H3 started to compromise HFF survival at ~5 µM. In contrast to the in vitro efficacy assessment data, compounds H1-H3 were shown to be ineffective in the neosporosis mouse model at a dosage of 10 mg·kg⁻¹·day⁻¹) for 5 days [57].

Metallodrugs present a high potential not only as anticancer but also as antimicrobial, antiviral and antiparasitic agents [21,24,25,59]. Recent studies in medicinal inorganic chemistry aimed to extrapolate and adapt strategies that are well-established in organic drug discovery. This includes for example, the design of new compounds by employing fragment-based approaches [60,61] and pharmacophore conjugation strategy [62], the use of combinatorial coordination chemistry to access and study larger libraries of compounds [63,64], as well as the use of high-throughput screening assays [65,66]. Also, in the perspective of applying medicinal-chemistry approaches to metal-based bioactive compounds, an increasing number of reports focus of the development of series of compounds with the identification of descriptors and structure-activity relationships [67–72].

The limitations observed in vivo for compounds H1-H3 [50,57,58] highlighted the necessity to identify thiolato-bridged dinuclear ruthenium(II)-arene compounds with improved properties as antiparasitic agents and to understand the structural features influencing their efficacy. The structure of trithiolato complexes (hydrophobic capping arene ligands, strong thiol bridges and absence of labile ligands) make them particularly inert towards hydrolysis, and more generally, ligand substitution. A fine balance of compounds' polarity/hydrophobicity can be reached by finely adjusting the substituents present on the bridge thiols. Based on H1-H3, a library of 56 thiolato-bridged dinuclear ruthenium(II)arene and osmium(II)-arene compounds was designed, comprising a systematic tuning of the nature, position and number of the substituents present on the thiol bridge ligands. According to their respective structural characteristics the compounds were organized into eight families. For the biological assessment of the 56 compounds, a sequence of consecutive screening steps was



Fig. 1. Structures of various metal complexes presenting anti-Toxoplasma gondii activity.

considered (Fig. 4). In the primary screening, all compounds were assessed for activity against *T. gondii*- β -gal and non-infected HFF at 0.1 and 1 μ M. Compounds exbiting \geq 95% parasite proliferation inhibition and \leq 49% impairment of HFF viability at 1 μ M were then selected for a secondary screening, which comprised dose-response experiments yielding IC₅₀ values for *T. gondii*- β -gal proliferation and an assessment of viability impairment in HFF monolayers at 2.5 μ M. Only the compounds that did not cause HFF viability loss with more than 50% were subjected to a tertiary screening on their inhibitory potential in murine splenocyte cultures stimulated with lipopolysaccharide (LPS) and concanavalin A (ConA) to determine their effects on B and T cells viability and proliferative responses, respectively.

2. Results and discussion

2.1. Chemistry

Eight families of compounds based on the trithiolato-bridged dinuclear ruthenium(II)-arene scaffold were designed by applying a systematic approach. Various substituents were introduced at different positions of the thiols' aromatic ring, for tuning their physico-chemical properties (polarity, hydrophobicity, ability to participate in H-bonding interactions). Detailed synthesis procedures and analysis of the new compounds are given in the Supporting information. An overview of the 56 compounds investigated in this study is presented (Fig. 2).

The symmetric trithiolato-bridged dinuclear ruthenium(II)arene derivatives were obtained using adapted literature procedures [73–75] in one step as shown in Scheme 1A, starting from the ruthenium dimer [Ru(η^6 -p-MeC₆H₄Prⁱ)Cl₂]₂ and excess of the corresponding thiol (general procedure **A** in Supporting information).

The mixed cationic compounds were prepared in two steps adapting previously described protocol [76] following the reaction sequence presented in Scheme 1B. The dithiolato intermediates isolated after the reaction of one equivalent of ruthenium dimer $[Ru(\eta^6-p-MeC_6H_4Pr^i)Cl_2]_2$ with two equivalents of a first type of thiol (R1-SH in Scheme 1B, see general procedure B and C in Supporting information) were further reacted with a second type of thiol (R2-SH in Scheme 1B) used in excess (general procedure **D** in Supporting information). Most of the reactions for obtaining the mixed trithiolato compounds from the dithiolato compounds were performed in EtOH. In some cases (e.g., carboxylate analogues 3, 4 and 53) a mixture of CH₂Cl₂/acetone was used as solvent to avoid esterification side reaction favored by the acid produced during the reaction. For the synthesis of compound 23 the solvent was a mixture of CH₂Cl₂/MeOH to increase the solubility of the dithiolato intermediate P4.

Compounds **15** and **17** were obtained from the respective hydroxy and amine functionalized ruthenium complexes **1** and **2** using reactions presented in Schemes S1 and S2 in Supporting information.

The new compounds were fully characterized by ¹H and ¹³C NMR spectroscopy, mass spectrometry and elemental analysis. The



Fig. 2. Structure of the 56 compounds investigated in this study organized in 8 families based on structural features modifications.



Scheme 1. Synthesis of the symmetric (A) and mixed (B) trithiolato-bridged dinuclear ruthenium (II)-arene complexes.

representative peaks for the neutral dithiolato-bridged dinuclear ruthenium(II)-*p*-cymene (*p*-cymene: *p*-MeC₆H₄Prⁱ) intermediates **P1–P13** correspond to $[M - Cl]^+$ ion, due to the loss of a labile chlorine ligand. For the cationic trithiolato compounds **1–56**, the parent ion peak corresponds to the cation trithiolato di-ruthenium $[M - Cl]^+$ ion.

The compounds are readily soluble and stable in DMSO- d_6 which make them suitable for further biological tests (Figures S19-S27 in Supporting information).

The crystal structures of the symmetric complexes **13** and **43**, mixed complex **38** and dithiolato intermediate **P13** were established in the solid state by single-crystal X-ray diffraction (ORTEP

representations shown in Fig. 3 (A and B, S8 and S9), confirming the expected structure. Data collection and refinement parameters are given in Table S5 (Supporting information). Selected bond lengths and angles are presented in Tables S6, S7, S8 and S9.

Compounds 13 and 38 co-crystallize with solvent and H₂O molecules ($13 \cdot CH_2Cl_2 \cdot H_2O$ and $38 \cdot H_2O$, Table S5). These solvent molecules not only fill the voids in the crystal packing, but also form a series of H-bonding interactions as presented in Figures S2 and S5 and Table S7 in Supporting information. In both structures, the ruthenium atoms adopt a pseudo-octahedral geometry with three sulphur atom ligands and the *p*-cymene ligand. In the case of the symmetric trithiolato compound 13, both Ru(II) atoms adopt the typical piano-stool geometry with the p-cymene ligand being bound facially while the other three positions are occupied by three 4-aminothiophenolate ligands. The mixed **38** complex presents two 4-(trifluoromethyl) thiophenolate and one 4hydroxythiophenolate as bridge thiols. To date, this is only the second example of an X-ray analysis of a mixed dinuclear trithiolato ruthenium(II)-arene complex [77]. The Ru₂S₃ unit forms a trigonalbipyramidal framework; no metal-metal bonds are present, the corresponding distances being Ru(1)-Ru(2) 3.345 and Ru(3)-Ru(4) 3.346 for 13 (asymmetric unit contains two independent complexes) and Ru(1)–Ru(2) 3.347 Å for **38**. In both cations, the values of Ru-S bonds as well as the Ru-S-Ru angles (Table S6) are similar to those found in other symmetric or mixed *p*-cymene derivatives previously reported [74,77]. In both structures, the aromatic rings of the bridge thiols are tilted compared to the plane of the three sulphur atoms. Similar to the reported mixed complex I (Table S6)



Fig. 3. ORTEP representation of compounds **13** (A) and **38** (B) (thermal ellipsoids are 50% equiprobability envelopes, and H atoms are spheres of arbitrary diameter; the asymmetric unit of **13** contains two independent complexes and two H₂O and two CH₂Cl₂ molecules, and the asymmetric unit of **38** contains one H₂O molecule).

[77], the Ru–S–Ru and cent η^6 -cent(S–S–S)-cent η^6 angles in the mixed complex **38** show that the central di-ruthenium trithiolate unit is not symmetric. The presence of the two 4-(trifluoromethyl) thiophenolate ligands forces the dinuclear arene ruthenium unit to adopt a slightly distorted geometry, the planes of the two *p*-cymene ligands (C21–C26 and C31–C36) being tilted.

In the network of **13** is observed an interplay of H-bonding interactions involving both the chlorine anions and the H₂O molecules trapped in the crystal. A dimeric organization is mediated by four H-bonding interactions: two chlorine anions bridge two diruthenium units at the level of their respective two amino groups (Figure S2, Supporting information). The remaining amino group of each of the symmetric di-ruthenium complexes is involved in Hbonding interactions with an H₂O molecule. Thus, all three amino groups of the symmetric trithiolato di-nuclear ruthenium(II)-arene complex **13** are involved in intermolecular H-bonding interactions, two of them interact with two chlorine anions and lead to the formation of a dimer with another di-ruthenium unit, while the third NH₂ group interacts via H-bonds with the H₂O molecule present. In network, these intermolecular H-bonding interactions lead to further arrangements e.g., with the formation of parallel chains of di-ruthenium complexes (Figure S3). In the crystal of 13 an intermolecular H- π interaction (distance 3.404 Å) between a Hatom from the *p*-cymene iso-propyl group and the aromatic ring of one of the aminothiol ligands was also identified (Figure S4).

In the crystal packing of **38**, a H-bonding interaction between the chlorine anion and OH group of the 4-hydroxythiophenolate unit (Figure S5, Supporting information) was acknowledged. The chlorine anion also interacts *via* H-bonds with two H₂O molecules trapped in the network. These cooperative H-bonding interactions mediated by chlorine anions and H₂O molecules lead to further dimeric organization (Figure S5) and dictate network packing for example in antisense chains connected *via* a zipper of H-bonds (Figure S7, Supporting information.). For **38** other short intermolecular contacts (2.675 and, respectively, 2.615 Å) were identified between the O atom of the phenol and a *p*-cymene H atom, and a H atom from the *p*-cymene methyl substituent of a neighbouring molecule (Figure S6).

Experimental details regarding the X-ray structure determination for complexes **43** and **P13**, as well as the corresponding structural details are presented in Tables S5, S8 and S9, and Figures S8 and S9 in the Supporting information.

2.2. Biological assessment

2.2.1. Screening strategy

The 56 compounds considered for this study were organized in 8 families based on their structural features, and were subjected to a sequential biological screening as summarized in Fig. 4. The antiparasitic activity (proliferation inhibition) was evaluated using *T. gondii*- β -gal grown in HFF host cell monolayers [50], and the cytotoxic effects were studied in non-infected HFF monolayers. The potential impact on T and B cell viability and proliferation was investigated using concanavalin A (ConA) and lipopolysaccharide (LPS)-stimulated murine splenocyte cultures [51].

The primary screening (Fig. 4) of all 56 compounds against *T. gondii*- β -gal tachyzoites and HFF was carried out at 0.1 and 1 μ M concentrations. These two concentrations were chosen for several reasons, such as: (i) potential solubility issues at the higher concentration, (ii) assessments for anti-parasitic or host cell activity cannot be done by testing only one concentration, (iii) the *T. gondii* IC₅₀ values of potentially interesting compounds would be in the range of 0.1 μ M, and (iv) the 1 μ M concentrations provides first indications on potential cytotoxic effects. Also, testing two concentrations initially opens options for selecting compounds for the



Fig. 4. Schematic representation of the three-step screening cascade used for the assessment of the antiparasitic activity and cytotoxicity of the trithiolato-bridged dinuclear ruthenium(II)-arene compounds. ^aβ-Galactosidase parasite proliferation assay. ^bAlamarBlue cell viability assay. ^cBrdU (5-bromo-2'-deoxyuridine) cell proliferations assay.



Fig. 5. Primary screening: results for the 56 compounds assessed for anti-*T. gondii* activity. Compounds were organized into 8 families. *T. gondii* infected HFF cultures were treated at two concentrations 0.1 μ M (light grey bars) and 1 μ M (dark grey bars). Results are presented as mean of three replicates \pm standard deviation. Results are presented as percentage of *T. gondii*-β-gal proliferation relative to control (CTR) parasites treated with 0.01% or 0.1% DMSO (solvent only).

next screening steps. When applied at 1 μ M, the compounds that inhibited *T. gondii*- β -gal proliferation by at least 95% and did not impair HFF viability by more than 49% were subjected to the next step. In the secondary screening, the selected compounds were

submitted to dose response studies to determine the IC_{50} values, and potential cytotoxicity in HFF was assessed at 2.5 μ M. The complexes exhibiting HFF cytotoxicity 50% or less were further subjected to a tertiary screening using murine splenocytes. For this,

splenocyte cultures were subjected to treatments with LPS (to stimulate B cells) or ConA (to stimulate T cells), and the effects of the compounds at 0.1 and 0.5 μ M concentrations on cellular viability were determined. These two concentrations were chosen because (i) most of the selected compounds (including the anti-*Toxoplasma* drug pyrimethamine) inhibit *T. gondii* in this concentration range, (ii) isolated and proliferating spleen cells are likely to be more susceptible to compound effects than HFF, and (iii) it is unlikely that higher drug concentrations will be obtained in tested animals. The ruthenium complexes that impaired T cell viability by less than 50% and B cell viability by less than 10% were further evaluated for their capacity to suppress ConA- and LPS-induced T and B cell proliferation.

2.2.2. Primary screening

Detailed results of the primary screening are also provided in Table 5.

2.2.2.1. Family 1. Family 1 is comprised of 8 members whose structures are presented in Table 1, Fig. 2 and S11. The BF₄ salt of complex **H3** was previously shown to exhibit encouraging *in vitro* efficacy against *T. gondii*- β -gal tachyzoites [50] and it represents a lead compound for the design of other derivatives of the current study. For building the first series, the hydroxy group of **H3** was replaced by various polar groups: amino (2), thiol (5), and carboxy (3 and 4), the mixed compounds being synthesized as Cl⁻ salts. In order to assess the influence of the counterion on the biological activity, the series was completed with amino and carboxy compounds **6** and **7**, isolated as tetrafluoroborate (BF₄) salts (Table 1).

The results of the primary screening at concentrations of 0.1 and 1 μ M for family 1 are presented in Fig. 5, 6 and Table 5. Hydroxy functionalized compounds **H3** and **1**, with BF₄ and Cl⁻ counterions, respectively, showed similar inhibition of parasite proliferation. Nevertheless, **1** did affect more than **H3** the viability of HFF for the same concentration. Amino derivative **6** with a BF₄ counter-ion exhibited enhanced activity against *T. gondii* and less pronounced HFF toxicity compared to the corresponding Cl⁻ salt **2**. All carboxy derivatives tested, **3**, **4** and **7**, were toxic neither for *T. gondii* nor for HFF. The influence of the counterion (BF₄ vs Cl⁻) appeared less important than the nature of the polar substituent on one of the

European Journal of Medicinal Chemistry 222 (2021) 113610

bridge thiols. The thiol functionalized compound **5**, more prone to oxidation, presented reduced activity against the parasite and less pronounced HFF toxicity compared to the hydroxy **1** and amino **2** analogues.

2.2.2.2. Family 2. Family 2 includes 7 compounds, for which the number (two mixed in complexes **8–10** and three in symmetric complexes **11–14**) and the position (*para* or *meta*) of the polar groups OH and NH₂ were systematically varied (Table 1, Fig. 2 and S12). Compounds **11** and **14** were previously reported [74,78] and tested against *T. gondii*-β-gal under similar conditions [50].

The primary screening (Figs. 5 and 6, Table 5) shows that compounds bearing two polar groups, namely **8** (hydroxy) and **9** (amino), were more toxic to both *T. gondii*- β -gal and HFF compared to their analogues with a single polar substituent (**1** and **2**, respectively). The two compounds bearing three polar groups, **12** (hydroxy) and **13** (amino), were ineffective against both *T. gondii*- β gal and HFF. These findings are supported by previous studies in which no activity was observed for compounds **11** and **14** [74,78]. The significantly increased toxicity in *T. gondii*- β -gal of **8** compared to **10** is interesting, as the structures of these compounds only differ by the presence of a methylene spacer in the bridge thiol bearing the Bu^t group. However, the effects of **8** and **10** on HFF viability were comparable, suggesting that the increased mobility of the bridge thiol bearing a Bu^t group could be beneficial for antiparasitic activity.

Overall, the data obtained for family 2 confirm previous findings [36] that the antiparasitic activity is influenced by the nature of the groups anchored on the bridged thiols, while the position of the substituents (*para* versus *meta*) has a lower influence.

2.2.2.3. Family 3. Family 3 comprises two esters (**15** and **18**) and two amides (**16** and **17**) derivates from hydroxy amino and carboxy analogues **1**, **2** and **3** (Table 2, Fig. 2 and S13). These structural modifications act as a protection of the polar groups preventing metabolic degradation (e.g., oxidation of the OH and NH₂ substituent to quinone or quinone-imine) or reaction/interaction with biomolecules. The proliferation of *T. gondii* tachyzoites was almost completely suppressed by **15–18** at 0.1 μ M (Fig. 5 and Table 5), whereas the polar parent analogues **1** and **2** displayed anti-

Table 1

Structure of the compounds composing families 1 and 2.

	Compound H3ª	R1 CH ₂ $-C_6H_4$ -4-Bu ^t	R2 C ₆ H ₄ -4-OH	X ⁻ BF ₄
Ru	Family 1			
R1-S S-R2	1 ^b	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ -4-OH	Cl-
R1	2	$CH_2 - C_6H_4 - 4 - Bu^{\ell}$	C ₆ H ₄ - 4-NH₂	Cl
Ru	3	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 4-CH₂-CO₂H	Cl-
	4	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 4-CO₂H	Cl ⁻
	5	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 4-SH	Cl ⁻
	6	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₄ - 4-NH₂	BF_4^-
	7	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 4-CH₂-CO₂H	BF_4^-
-	Family 2			
-	8 ^c	C ₆ H ₄ - 4-OH	$CH_2 - C_6H_4 - 4 - Bu^t$	Cl-
	9	C ₆ H ₄ - 4-NH₂	$CH_2 - C_6H_4 - 4 - Bu^t$	Cl-
	10	C ₆ H ₄ -4-OH	C ₆ H ₄ - 4-Bu ^t	Cl-
	11 ^d	C ₆ H ₄ -4-OH	C ₆ H ₄ - 4-OH	Cl-
	12	C ₆ H ₄ - 3-OH	C ₆ H ₄ - 3-OH	Cl-
	13	C ₆ H ₄ - 4-NH₂	C ₆ H ₄ - 4-NH₂	Cl-
	14 ^e	C ₆ H ₄ - 3-NH₂	C ₆ H ₄ - 3-NH₂	Cl ⁻

^a Previously reported in Ref. [50].

^b previously reported in Ref. [76].

^d previously reported in Refs. [50,74].

^e Previously reported in Refs. [50,78].



Fig. 6. Primary screening: the 56 compounds were assessed with respect to activity against non-infected HFF. Compounds were organized into 8 families, and tested concentrations were 0.1 μM (light grey bars) and 1 μM (dark grey bars). Results are presented as mean of three replicates ± standard deviation. Results are presented as percentage of HFF viability relative to a control (CTR), namely HFF treated with 0.01% or 0.1% DMSO (solvent only).

T. gondii- β -gal activity only when applied at 1 μ M, and **3** was completely inactive. Overall, compounds **15–18** are slightly more toxic for HFF than the more polar derivatives **1–3**. Modifying polar hydroxy (**1**), amino (**2**) or carboxy (**3**) groups into less polar ester or

amide substituents (**15–18**) enhanced significantly the potency against the parasite, and, to a lesser extent, the toxicity to HFF host cells.

Table 2

Structure of the compounds composing families 3 and 4.

	Compound	R1	R2
	Family 3		
R1-S S S-R2 R1 Ru	15 16 17 18	$\begin{array}{c} CH_2-C_6H_4-4-Bu^t\\ CH_2-C_6H_4-4-Bu^t\\ CH_2-C_6H_4-4-Bu^t\\ CH_2-C_6H_4-4-Bu^t\\ CH_2-C_6H_4-4-Bu^t\\ \end{array}$	C ₆ H ₄ - 4-OAc C ₆ H ₄ - 4-NHAc C ₆ H ₄ - 4-N-piperidine-2,6-dione C ₆ H ₄ - 4-CH₂-CO₂Et
	Family 4		
	19 20 21 22 ^a 23	CH ₂ -C ₆ H ₄ -4-Bu ^t CH ₂ -C ₆ H ₄ -4-Bu ^t CH ₂ -C ₆ H ₄ -4-Bu ^t CH ₂ - C₆H₅ CH ₂ - C₆H₄-4-Me	C ₆ H ₄ - 3-OH C ₆ H ₄ - 2-OH C ₆ H ₄ - 2-CH₂OH C ₆ H ₄ -4-OH C ₆ H ₄ -4-OH

^a Previously reported in Ref. [76] presenting BF₄ as counterion.

2.2.2.4. Family 4. Family 4 includes 5 compounds, all containing a polar hydroxy group on one of the bridge thiols (Table 2, Fig. 2 and S14). In this series, the influence of the OH group position (*meta* and *ortho*) in compounds **19** and **20**, was compared to that of *para*-analogue **1**. In derivative **21** the *ortho*-hydroxy group of **20** was replaced by a hydroxymethyl substituent, and in compound **23** the Bu^t substituents anchored on two of the bridged thiols in analogue **1** were replaced with less bulky and hydrophobic methyl substituents in the same *para* position. Compound **22** with an H atom as substituent had been previously reported [76] as BF₄ salt and when tested against *T. gondii*-β-gal [50] was considered not interesting for further studies.

The primary screening (Figs. 5 and 6 and Table 5) showed that *meta* and *ortho* hydroxy-substituted analogues **19** and **20**, were more efficient against *T. gondii* compared to the *para*-substituted compound **1**. A similar but stronger effect was observed for **21**, presenting a methylene spacer between the OH group and the aromatic ring, as well as for **23** bearing less bulky and hydrophobic Me substituents on two of the bridge thiols (compared to Bu^t in **1**). The position of the polar OH group present on one of the bridge thiols (**1**, **19** and **20**), but also the volume and lipophilicity of the substituents present on the other two thiols (**1** *vs* **23**) influence the parasite proliferation. The structural modifications in compounds **19**–**21** and **23** led to a decrease of host cell toxicity.

Table 3				
Structure of the compounds	composing	families	5 and	6.

	Compound	R1	R2	М
	Family 5			
	24	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₅	Ru
R1-S S S-R2	25	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₄ - 4-F	Ru
R1	26	CH2-C6H4-4-But	C ₆ H ₄ - 4-Br	Ru
M	27	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₄ - 4-CF₃	Ru
	28	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₄ - 4-OCH₃	Ru
	29	CH ₂ -C ₆ H ₄ -4-Bu ^t	C ₆ H ₄ - 4-OCF₃	Ru
	30	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 3-OCH₃	Ru
	31	$CH_2 - C_6H_4 - 4 - Bu^t$	C ₆ H ₄ - 2-OCH₃	Ru
_	32	$CH_2 - C_6H_4 - 4 - Bu^t$	CH ₂ -furan-2-yl	Ru
	Family 6			
	33	C ₆ H ₅	C ₆ H ₄ -4-OH	Ru
	34	C ₆ H ₄ - 4-Me	C ₆ H ₄ -4-OH	Ru
	35	C ₆ H ₄ - 4-Prⁱ	C ₆ H ₄ -4-OH	Ru
	36	C ₆ H ₄ - 4-Bu ^t	C ₆ H ₄ -4-OH	Ru
	37	C ₆ H ₄ - 4-F	C ₆ H ₄ -4-OH	Ru
	38	C ₆ H ₄ - 4-CF₃	C ₆ H ₄ -4-OH	Ru
	39	CH₂ –C ₆ H ₄ - 4-Bu ^t	C ₆ H ₄ -4-OH	Os
	40	C ₆ H ₄ - 4-Bu ^t	C ₆ H ₄ -4-OH	Os

2.2.2.5. Family 5. Family 5 includes 9 members (Table 3, Fig. 2 and S15). The role of the OH group in compound 1 was also questioned. In derivative 24 the hydroxy group was removed, whereas in compounds 25–27 it was replaced by other polar substituents as fluoro, bromo or trifluoromethyl, respectively. Compared to 1, in compounds 28 and 29 the *para*-OH group was protected as a methoxy and trifluoromethoxy ether. As the position of the OH group in 1 (*para*), 19 (*meta*) and 20 (*ortho*) influenced the antiparasitic efficacy, the effect of a methoxy group in the same positions was also established by comparing compounds 28 (*para*), 30 (*meta*) and 31 (*ortho*). The electron-rich *para*-phenoxy ring 1 was replaced in 32 with a methylene-furan-2-yl substituent.

Compared to **1**, the absence of the OH group in **24** increased both the anti-*Toxoplasma* efficacy and the viability of HFF (Figs. 5 and 6 and Table 5). Likewise, the replacement of the hydroxy group in **1** with fluoro (**25**), bromo (**26**) and trifluoromethyl (**27**) polar substituents increased the antiparasitic activity and slightly reduced HFF toxicity. Both OH and F can participate in H-bonding interactions; nevertheless, while the OH group can be oxidized, this type of metabolization is less prone to take place if the position is blocked by fluoro, bromo or trifluoromethyl substituents.

Compounds **28** and **29** in which the hydroxy group in *para* position was protected as a methoxy or trifluoromethoxy group were more efficient against *T. gondii* and exhibited a lower HFF toxicity compared to **1**. *Meta* and *ortho* methoxy-substituted compounds **30** and **31** were similarly active against *T. gondii* as their respective hydroxy analogues **19** and **20**, but were significantly less toxic to HFF. Replacement of the *para*-phenoxy (**1**) with an electron rich furan ring in **32** also led to selective antiparasitic toxicity.

As previously observed for family 3, the results obtained for family 5 suggest that high efficacy against the parasite and reduced influence of the HFF viability are correlated with the presence of a polar group on one of the bridge thiols. Jointly, the presence of polar, chemically reactive groups, that can participate in H-bonding interactions (e.g., carboxyl) had a negative impact on the anti-*T. gondii* activity. Unreactive, polar substituents like fluoro (**25**), bromo (**26**), trifluoromethyl (**27**), methoxy (**28**, **30**, **31**) or trifluoromethoxy (**29**) answered these requirements and appeared to be better suited.

Regardless of the substituent position, all three methoxy derivatives **28**, **30** and **31** were highly efficient against *T. gondii*- β -gal when applied at 0.1 μ M. Some differences were observed in the effect on the HFF viability, for which *meta* and *ortho* substitution appeared to be more favorable.

2.2.2.6. Family 6. Family 6 is comprised of 8 compounds designed to assess the role of the methylene spacers in the bridge thiols, units correlated with the mobility and flexibility of the pendant

arms in **1** and **H3** (Table 3, Fig. 2 and S16). The hydrophobicity of the substituents was increased in series $H < Me < Pr^i < Bu^t$ for compounds **33**–**36**, while derivatives **37** and **38** present polar F and CF₃ substituents. Two di-osmium compounds **39** and **40** were also synthesized allowing comparison with ruthenium analogues **1** and **36**.

The results of the first screening for this family of compounds are presented in Figs. 5 and 6 and Table 5. Compared to compound **1**, the presence of non-polar hydrophobic substituents (Me, Prⁱ, Bu^f) on two of the bridge thiols in **34**–**36** increased the anti-*Toxoplasma* activity, whereas polar substituents as F and CF₃ in **37** and **38** had an opposite effect. Nevertheless, irrespective of the nature of the substituents in *para* position, the di-ruthenium complexes **33**–**38** displayed lower HFF cytotoxicity than derivative **1**. Limited flexibility of the pendant arms and a reduced polarity of the substituents present on two of the bridge thiols seem to be associated with improved efficacy against *T. gondii*- β -gal and a relative innocuousness towards HFF host cells.

The change of the metal to osmium in compounds **39** and **40** provided contrasting results. Di-osmium compound **40** was clearly superior to its corresponding di-ruthenium analogue **1** both in terms of efficacy against *T. gondii*- β -gal and reduced HFF toxicity. However, no important differences were observed between **36** (di-ruthenium) and **39** (di-osmium), neither with respect to antiparasitic activity nor in relation to host cell viability impairment. These preliminary results suggest that the biological activity could be altered not only by modifying the substituents present on the bridge thiols but also by nature of the metal present in the main framework. Nevertheless, evaluating a larger number of osmium derivatives, or of structure related di-nuclear compounds based on other metals e.g., rhodium or iridium would be necessary to confirm this hypothesis.

2.2.2.7. Family 7. A group of ten derivatives formed Family 7 (Table 4, Fig. 2 and S17). A previous study [50] identified symmetric compounds **H1** and **H2** (Fig. 1) which exhibited interesting anti-*Toxoplasma* activity and low toxicity in HFF. These structures were the starting point for the evaluation of additional structurally related symmetric compounds. The hydrophobic Me and Bu^t groups of **H1** and **H2** were either removed in **41**, or replaced with

Table 4

Structure of the compounds composing families 7 and 8.

	Compound Family 7 (R1	R1 = R2)	R2
R1-S S R1 Ru Ru Ru	H1 ^a H2 ^a 41 42 43 44 45 46	$\begin{array}{c} C_{6}H_{4}\textbf{-4-Me} \\ C_{6}H_{4}\textbf{-4-Bu}^{t} \\ \textbf{C}_{6}H_{5} \\ C_{6}H_{4}\textbf{-4-F} \\ C_{6}H_{4}\textbf{-4-CF} \\ C_{6}H_{4}\textbf{-4-CF} \\ C_{6}H_{4}\textbf{-3-CF} \\ C_{6}H_{4}\textbf{-2-CF} \\ C_{6}H_{3}\textbf{-3,5-(CF}) \\ \underline{C}_{6}H_{3}\textbf{-3,5-(CF}) \\ \underline{C}_{6}H_{3}\textbf{-3,5-(CF)} \\ $	
	47 48 49 50	$CH_2-C_6H_4-4-CF_3$ $CH_2-C_6H_5$ $CH_2-C_6H_4-4-Me$ $CH_2-C_6H_4-4-Bu^t$	
-	Family 8		
	51 ^b 52 ^b 53 ^b 54 55 56	$\begin{array}{c} CH_2-C_6H_4\textbf{-4-CF_3}\\ CH_2-C_6H_4\textbf{-4-CF_3}\\ CH_2-C_6H_4\textbf{-4-CF_3}\\ CH_2-C_6H_4\textbf{-4-CF_3}\\ CH_2-C_6H_4\textbf{-2-CF_3}\\ CH_2-C_6H_4\textbf{-2-CF_3}\\ \end{array}$	$\begin{array}{l} C_{6}H_{4}\textbf{-4-OH} \\ C_{6}H_{4}\textbf{-4-NH}_{2} \\ C_{6}H_{4}\textbf{-4-CH}_{2}\textbf{-CO}_{2}\textbf{H} \\ C_{6}H_{4}\textbf{-4-CH}_{2}\textbf{-CO}_{2}\textbf{CH}_{3} \\ C_{6}H_{4}\textbf{-4-OH} \\ C_{6}H_{4}\textbf{-4-NH}_{2} \end{array}$

^a Previously reported in Ref. [50].

^b Previously reported in Ref. [51].

Table 5		
Results of the	e primary efficacy/cytotoxicity screening	١ơ

Compound	HFF viability (%)		T. gondii B-g	al growth
			(%)	
	0.1 μΜ	1 µM	0.1 μM	1 µM
H3 ^a	119 ± 2	81 ± 3	60 ± 4	0 ± 1
Family 1				
1 ^{b,d}	76 ± 6	46 ± 6	66 ± 14	2 ± 0
2 ^{b,d}	74 ± 2	48 ± 1	57 ± 1	2 ± 0
30,0	91 ± 4 98 ± 6	73 ± 1	114 ± 2 102 ± 7	110 ± 2
	98 ± 0 98 + 2	65 ± 5	102 ± 7 102 + 2	33 ± 12 29 + 1
6 ^d	115 ± 1	82 ± 2	7 ± 1	0 ± 0
7	116 ± 2	107 ± 1	118 ± 1	98 ± 0
Family 2				
8 ^{c,d}	62 ± 8	56±7	3 ± 1	2 ± 0
9 ^d	66 ± 6 73 + 10	58 ± 8 62 ± 2	2 ± 0 108 ± 2	2 ± 0
10 11 ^a	75±10 -	02 ± 2	-	4 ± 1 —
12	114 ± 8	104 ± 9	105 ± 5	108 ± 2
13	97 ± 4	61 ± 6	115 ± 4	85 ± 5
14ª			_	_
Family 3				
15" 16	74 ± 2 62 ± 7	52 ± 3 27 ± 1	5 ± 0 4 ± 0	2 ± 1 3 ± 1
17 ^d	62 ± 7 68 + 7	58 + 7	4 ± 0 3 + 1	3 ± 1 2 + 0
18	50 ± 4	46 ± 2	3 ± 0	4 ± 1
Family 4				
19 ^d	89 ± 4	62 ± 2	5 ± 0	1 ± 0
20 ^d	80 ± 3	68 ± 9	8 ± 1	1 ± 0
21 ^d	80 ± 1	69 ± 6	2 ± 0	1 ± 0
22 ^d	- 110 + 9	-62 + 4	- 1+1	- 1 + 4
Family 5				
24 ^d	81 + 5	75 + 8	2 ± 0	1+0
25 ^d	68 ± 9	61 ± 3	6 ± 0	4 ± 1
26 ^d	89 ± 2	67 ± 0	6 ± 1	1 ± 0
27 ^d	59 ± 2	51 ± 8	-1 ± 4	-3 ± 8
28 29 ^d	72 ± 8 69 + 7	62 ± 4 66 + 7	0 ± 3 -1+2	-2 ± 4 -2 ± 2
30 ^d	90 ± 2	80 ± 3	2 ± 0	1 ± 0
31 ^d	93 ± 2	80 ± 6	2 ± 0	1 ± 0
32 ^a	76 ± 8	65 ± 3	2 ± 0	1 ± 0
Family 6				
33 ^d	93 ± 2	73 ± 3	89 ± 1	5 ± 1
34 35 ^d	99 ± 3 104 + 6	73 ± 2 83 + 5	3 ± 0 10 + 3	1 ± 0 1 + 0
36 ^d	74 ± 1	101 ± 9	3 ± 0	1 ± 0
37	87 ± 3	64 ± 3	113 ± 2	96 ± 0
38 ^d	89±3	62 ± 6	97 ± 7	1 ± 0
40 ^d	87 ± 3 87 + 3	74 ± 1 60 + 3	3 ± 1 8 + 2	1 ± 0 1 + 0
Family 7				
H1 ^a	_	_	_	_
H2 ^a	_	_	_	_
41 ^d	115 ± 0	93 ± 1	3 ± 0	0 ± 0
42 ^d	85 ± 0 77 ± 7	69 ± 2	98 ± 1	1 ± 0
44 ^d	36 + 1	79 + 9	-1 ± 0 1+0	0 ± 1 0 + 0
45 ^d	100 ± 3	68 ± 1	1 ± 0	0 ± 0
46 ^d	95 ± 11	69 ± 3	2 ± 0	1 ± 0
47 ^u	85 ± 3	57 ± 1	3 ± 0	0 ± 0
49	104 ± 1 66 + 4	35 ± 3 15 + 1	117 ± 1 0 + 1	0 ± 0 0 + 1
50 ^d	63 ± 4	62 ± 10	91 ± 0	1 ± 2
Family 8				
51 ^{b,d}	112 ± 7	66 ± 4	101 ± 1	0 ± 0
			(continued o	on next page)

Table 5 (continued)

Compound	HFF viability (%)		T. gondii β-g (%)	β -gal growth	
	0.1 µM	1 µM	0.1 µM	1 µM	
52 ^{b,d}	115 ± 1	58 ± 2	59 ± 7	0 ± 0	
53 ^b	103 ± 3	93 ± 6	102 ± 5	102 ± 3	
54	92 ± 8	40 ± 2	76 ± 7	0 ± 0	
55 ^d	111 ± 2	93 ± 2	1 ± 0	0 ± 0	
56 ^d	114 ± 2	86 ± 3	0 ± 0	0 ± 0	

^a Compound reported in Ref. [50].

^b Compound reported in Ref. [51].

^c Compound reported in Ref. [79].

^d Compounds selected for the secondary screening.

polar fluoro and trifluoromethyl substituents, in compounds **42** and **43**, respectively. The role of the position (*meta* in **44**, *ortho* in **45**) and number (**46**) of the trifluoromethyl substituents was also assessed. Compounds **47–50** differ from the complementary analogues **43**, **41**, **H1** and **H2**, by the presence of methylene spacers on the bridge thiols. Compounds **42**, **44** and **46** were available in the laboratory from previous studies and only their purity was verified prior to assessing their biological activity (Supporting information).

The primary screening results (Figs. 5 and 6 and Table 5) showed that triphenyl compound **41** and analogues bearing trifluoromethyl groups **43–47** were highly efficient against *T. gondii*- β -gal, and inhibited parasite proliferation almost entirely already at 0.1 μ M. However, trifluoromethyl substituted analogues **43–47** had a higher impact on HFF viability compared to **41**. The position and the number of the trifluoromethyl groups in **43–47** influenced HFF toxicity only marginally. *Para*-fluoro substituted compound **42** suppressed *T. gondii*- β -gal proliferation only when applied at 1 μ M, while being more toxic for HFF compared to **41**. From the compounds with non-polar substituents and methylene spacers on the pendant thiols, methyl derivative **49** was more efficient in preventing parasite proliferation compared to H- and Bu^t-substituted analogues **48** and **50**, being active already when applied at 0.1 μ M, but also with increased toxicity for HFF.

Summarizing, a more rigid structure in compounds **41–46** (without methylene groups on the pendant arms) favored antiparasitic activity, but an increased number of polar trifluoromethyl groups led also to an increase in HFF toxicity. A fine balance of the polarity, size and number of the substituents is necessary to achieve a good compromise between antiparasitic activity and host cell cytotoxicity.

2.2.2.8. Family 8. Family 8 (Table 4, Fig. 2 and S18) contains six members and was conceived based on lead derivative H3 and considering the importance of the trifluoromethyl substituents in the compounds presented in family 7 (Table 4, Fig. 2 and S16). The hydrophobic bulky Bu^t groups of two of the bridge thiols in **H3** were replaced with polar CF₃ groups, while the third thiol presents polar substituents as hydroxy (51), amino (52), carboxy (53) and ester (54). Mixed complexes 55 and 56 present two ortho-CF₃ substituted thiols, while the third thiol is substituted with a *para*-hydroxy (55) and -amino group (56). As shown in Figs. 4 and 5, the replacement of one trifluoromethyl substituent in **47** with a hydroxy or amino group in **51** and **52**, respectively, significantly decreased antiparasitic activity (compounds active only when applied at 1 μ M), but had little impact on HFF cytotoxicity. The absence (38) or presence (51) of methylene spacers in the CF₃ substituted thiols did not substantially alter the anti-Toxoplasma activity and HFF cytotoxicity at 1 µM. Similar to symmetric compound 45, hydroxy and amino substituted derivatives 55 and 56, were highly potent inhibitors of *T. gondii* proliferation even at 0.1 μ M, but were less toxic to HFF when applied at 1 μ M. Hydroxy and amino derivatives **51** and **52** with CF₃ groups, compared to their respective *para*-Bu^t-substituted analogues **1** and **2**, exhibited similar antiparasitic activity but slightly lower host cell toxicity. As previously observed for carboxy substituted compounds **3** and **4**, **53** did not display neither antiparasitic activity nor HFF toxicity. However, **54** with the carboxy group protected as methyl ester, presented increased toxicity against both *T. gondii*- β -gal and host cells; a similar effect was observed with ethyl ester derivative **18** as compared to the carboxy compound **3**.

The primary screening revealed that the compounds showing anti-T. gondii activity while remaining non-toxic against HFF are structurally varied. Free carboxy groups have a detrimental effect as compounds **3**, **4** and **7** (family 1) are admittedly non-toxic to the host cells but also show no antiproliferative activity on the parasite. Likewise, the presence of more than one hydroxy and amino substituents on the bridge thiols (family 2) is not a good option, as compounds bearing two polar groups (8-10) are active but also toxic to the host cells and those bearing three polar groups (11-14) are not active against the parasite. Interestingly, the protection of the polar groups OH, NH₂ or CO₂H as ester or amide (family 3) brings no benefit, as the HFF viability is significantly reduced when treated with compounds 15-18. If the presence of one hydroxy group on one of the bridge thiols (family 4) appears to be important to the anti-parasitic activity, at this point of the study the influence of its positon is not easy to define, even if compound presenting the OH group in *para* position (1) appear more toxic on the host cells and less active on the parasite compared to *meta* and *ortho* substituted analogues 19 and 20.

Interestingly, the ether protection of the OH group (family 5, compounds **28–31**) appears to be a better option than the ester protection (15), while the nature of the protecting group (CH_3 or CF₃) as well as the position of the methoxy substituent impact the viability of the host cells. The presence of polar substituents F and expecially CF_3 (25 and 26) on only one of the bridge thiols appears to be associated to increased toxicity on HFF. Removing the methylene spacer on two of the bridge thiols while maintaining the polar hydroxy group on the third thiol (family 6) is associalted with an increased anti-T. gondii activity. One hydroxy group and two non-polar substituents such as Me, Pr^{*i*} and Bu^{*t*} (**34**, **35** and **36**) are better options than three F or CF₃ both for the activity against the parasite and the toxicity on the host cells. Some of the observations made in the case of mixed compounds seem to apply also to the symmetric derivatives from family 7. Fluorine as substituent (compound 42) led to a decreased activity against the parasite. The presence of methylene linkers on two of the bridge thiols appears to be associated to a decreased activity on T. gondii (48) or increased toxicity to HFF (compounds 47, 49 and 50). Compounds bearing polar CF₃ groups (43, 44, 45 and 46) seem a good compromise between anti-parasitic activity and effect on the host cell viability. The compounds from family 8 confirm some of the observations made previously: a free carboxy group on one of the brigde thiols (53) is to be avoided, as well as the presence of methylene spacers on two of the bridging thiols (compounds 51 and 52) or the protection of the carboxy group as an ester (54). Combining two CF_3 substituents on two of the bridge thiols with one polar hydroxy or amino group on the third thiol appears promising for ensuring both anti-T. gondii activity and reduced toxicity to the host cells.

2.2.3. Secondary screening

The selection criteria for a compound to enter secondary screening were: (i) over 95% inhibition of *T. gondii*- β -gal proliferation and (ii) not more than 49% viability impairment of HFF when applied at 1 μ M. 39 compounds fulfilled these criteria and were

Table 6

Secondary screening: determination of IC₅₀ (µM) values for *T. gondii*-β-gal and viability of HFF (as %-age in relation to solvent-treated controls) at 2.5 µM for the 39 compounds selected after the first screening. The 14 compounds selected for the tertiary screening (against murine B and T cells) are tagged with^b and the corresponding measured values are shown in bold.

Compound	T. gondii-β-gal IC ₅₀ (μM)	[LI; LS] ^c	SE ^d	HFF viability at 2.5 μ M (%) ^e	SD ^f
Pyrimethamine	0.326	[0.288; 0.396]	0.052	99	6
H3	0.079	[0.061; 0.102]	0.060	70	2
Family 1					
1 ^a	0.117	[0.098; 0.139]	0.051	56	6
2 ^a	0.153	[0.127; 0.185]	0.049	51	5
3 ^a	0.181	[0.274; 1.482]	0.954	99	2
5 ^a	0.316	[0.190; 0.526]	0.078	107	8
6 ^b	0.098	[0.066; 0.145]	0.085	72	3
Family 2					
8	0.115	[0.098; 0.134]	0.044	37	3
9	0.066	[0.046; 0.095]	0.126	19	1
10	0.294	[0.280; 0.308]	0.011	26	5
Family 3					
15	0.065	[0.041; 0.100]	0.092	16	5
17	0.014	[0.011; 0.032]	0.032	9	1
Family 4					
19	0.031	[0.024; 0.040]	0.07	16	2
20	0.050	[0.039; 0.062]	0.06	37	3
21	0.038	[0.023; 0.060]	0.11	4	2
23	0.048	[0.031; 0.071]	0.11	31	4
Family 5					
24	0.052	[0.033; 0.083]	0.116	1	0
25	0.031	[0.021; 0.047]	0.119	34	7
26	0.013	[0.012; 0.014]	0.016	1	0
27	0.043	[0.037; 0.145]	0.184	36	7
28	0.056	[0.040; 0.079]	0.080	31	4
29 ⁰	0.072	[0.035;0.145]	0.214	59	5
300	0.078	[0.062; 0.097]	0.054	54	1
31	0.025	[0.017; 0.036]	0.088	40	10
32	0.039	[0.030; 0.052]	0.065	39	3
Family 6	0.454		0.046	21	10
33	0.454	[0.370; 0.550]	0.046	31	12
34 ⁰	0.072	[0.043; 0.119]	0.114	57	6
35 ⁻	0.121		0.133	56	4
30 ⁻	0.041	[0.025; 0.065]	0.112	58	6
38 20 ^b	0.344	[0.302; 0.393]	0.031	70	6
39	0.062		0.142	64	5
40 Family 7	0.045	[0.034, 0.053]	0.05	49	2
	0.072	[0.057: 0.002]	0.052	86	7
41 42 ^b	0.072	[0.037, 0.092]	0.055	80 55	2
42	0.021	[0.015, 0.078]	0.099	10	
43	0.031		0.257	19	4
44 45 ^b	0.013		0.057	20	10
43	0.010		0.105	73 6	5
40	0.016	[0.008, 0.013]	0.045	2	1
-17 48 ^b	0.010	[0.014, 0.017]	0.019	2 90	4
-10	0.250	[0.202, 0.324]	0.032	1	J 1
Family 8	0.200	[0.203, 0.345]	0.075	1	1
51	0 163	[0 008. 0 319]	17	9	2
57	0.063	[0.049: 0.082]	0.043	37	2
52 55 ^b	0.003	[0.054.0.095]	0.045	84	3
56 ^b	0.056	[0.037: 0.087]	0.005	77	3
	0.000	[0.037, 0.007]	0.000		

^a Compounds from family 1 which do not correspond to the first screening selection criteria, but for which the IC₅₀ values and viability of HFF were determined for comparison purpose.

^b Compounds selected for the third screening on immune cells.

^c Values at 95% confidence interval (CI); LI is the inferior limit of CI and LS is the superior limit of CI.

^d Standard error of the regression (SE), represents the average distance at which the measured values fall from the regression line.

^e Control HFF monolayers treated with 0.25% DMSO exhibited 100% viability.

^f Standard deviation of the mean (six replicate experiments).

chosen for the determination of IC₅₀ values against *T. gondii* and HFF viability assessments at 2.5 μ M (Table 6). The IC₅₀ values against *T. gondii* ranged between 0.01 and 0.454 μ M, and 30 compounds exhibited an IC₅₀ < 0.1 μ M, which is significantly lower than the value measured for the standard drug pyrimethamine (0.326 μ M) [51]. The viability of HFF at 2.5 μ M compound concentration ranged between 1 and 89% when compared to non-treated cells. Eight compounds caused over 90% HFF cell death when

applied at 2.5 μM and 32 compounds caused viability loss higher than 30%.

Compounds from family 1 showed a moderate to low host cell toxicity and IC₅₀ values against *T. gondii*- β -gal ranging from 0.117 to 0.316 μ M. Interestingly, replacing the Cl⁻ (**2**) with a BF₄ as counterion (**6**) decreased HFF toxicity, but also improved antiparasitic activity. Interstingly, amino functionalized compound **6** appers to present rather similar activity and toxicity profile with the

corresponding hydroxy derivative presenting the same BF_4^- counterion H3.

All compounds assigned to families 2, 3 and 4 strongly impaired the viability of HFF at 2.5 μ M. For compounds belonging to family 2, the presence of two polar groups that can also take part in Hbonding interactions (derivatives 8 and 10 with two OH groups, and **9** with two NH₂ substituents) led to increased HFF toxicity compared to the mono-hydroxy and -amino compounds 1 and 2. Interestingly, for diamino derivative 9 the T. gondii IC₅₀ value was significantly lower compared to the IC₅₀ values of the dihydroxy analogues 8 and 10 (0.066 µM vs 0.115 and 0.294 µM), but compound **9** was also more toxic for the host cells. The presence of the methylene spacer on one of the pendant arms also influenced the antiparasitic activity (8 vs 10). In family 3, both ester 15 and amide 17 were highly cytotoxic for HFF at 2.5 μ M, and the effect was much stronger compared to the toxicity observed for the corresponding hydroxy and amino analogues 1 and 2. In family 4, shifting the position of the OH group from para (1) to meta (19) or ortho (20) led to increased HFF toxicity. A similar effect was observed when replacing the *para* Bu^t groups (1) with less hydrophobic and bulky Me substituents (23). Interestingly, when applied at 2.5 µM orthomethylenehydroxy derivative 21 was significantly more toxic for the host cells compared to the ortho-hydroxy derivative 20. All compounds from family 4 present methylene spacers on two of the pendant thiol arms.

Compounds **29** and **30** from family 5, moderately impaired HFF viability when applied at 2.5 μ M (by 41 and 46%, respectively) and presented similar and low IC₅₀ values against *T. gondii*- β -gal of 0.072 and 0.078 μ M. In contrast, compounds **24** and **26** almost completely abolished HFF viability. Derivatives **25**, **27**, **28**, **31** and **32** applied at 2.5 μ M reduced HFF viability in a 31–40% range, but *T. gondii* IC₅₀ values were also lower, ranging from 0.013 to 0.056 μ M.

The position of the methoxy group appeared to be important for the cytotoxicity against HFF, with *para* and *ortho* derivatives **28** and **31** being more toxic compared to the *meta*-substituted analogue **30**. For the *para*-substituted compounds **28** and **29**, the replacement of the hydrophobic methyl by the polar and bulkier trifluoromethyl reduced host cell toxicity.

Five of the seven compounds assigned to family 6 only moderately altered the viability of the host cells when tested at 2.5 μ M, while 33 and 40 exhibited increased HFF toxicity. In this series of hydroxy analogues, a reduced flexibility of the substituents present on two of the thiols (no methylene spacers) had a positive effect on host cell viability. **36** presents the lowest and **38** the highest IC₅₀ value against *T. gondii*-β-gal (0.041 and 0.344 μM, respectively), suggesting that more hydrophobic substituents ($Bu^t vs CF_3$) can favor antiparasitic activity. Interestingly, compounds 34, 35 and 36 bearing hydrophobic substituents Me, Pr¹ and Bu^t on two of the bridging thiols showed similar effects on the viability of the host cells (42–44% reduction), but IC₅₀ values against *T. gondii* were not linearly dependent on the hydrophobicity/size of the substituents (IC_{50} values of 0.072, 0.121 and 0.041 μM , respectively). For the same peripheric ligands, ruthenium 36 and osmium 39 compounds showed comparable effects against *T. gondii*- β -gal and HFF. Replacement of the methyl groups on two of the thiols (34) by more polar and bulkier trifluoromethyl groups (38) strongly reduced the antiparasitic activity (IC₅₀ values 0.072 vs 0.344 µM, respectively).

From the compounds assigned to family 7, **43**, **44**, **46**, **47** and **50** strongly impaired HFF viability at 2.5 μ M (74–99%). The presence of three or six trifluoromethyl polar groups in *para* or *meta* positions, which can be involved in intermolecular interactions (H-bonding, polar) led to a considerable increase in HFF toxicity. Accordingly, HFF viability was almost completely abolished in the case of compounds **46**, **47** and **50**, suggesting that an increased number of polar

groups (**46**) as well as the simultaneous presence of methylene spacers and of bulkier substituents in *para* position (**47** and **50**) increased HFF toxicity. Among the four compounds exhibiting low HFF toxicity, **48** showed the lowest, and **45** the highest activity against *T. gondii*- β -gal, with IC₅₀ values of 0.256 and 0.017 μ M, respectively. Replacement of H with polar fluoro substituents (**41** vs **42**) reduced both the anti-*Toxoplasma* activity and the viability of HFF monolayers. Increased flexibility (presence of methylene spacers on the pendant substituents) and an increased number of polar substituents that can potentially interact with molecules from the environment, had a negative impact on HFF. Notably, the IC₅₀ values of compounds **H1** and **H2** (0.034 and 0.062 μ M) are very low against *T. gondii*- β -gal, while the two compounds are non-toxic against the host cells (IC₅₀ values on HFF of 800 and > 1000 μ M, respectively) [50,57].

In family 8, the hydroxy and amino derivatives **51** and **52** differ from their analogues **55** and **56** by the position of the trifluoromethyl groups (*para vs ortho*) and the presence of methylene spacers on two of the bridging thiols. However, these structural differences had a strong influence on HFF toxicity. Both, **51** and **52** strongly affected HFF viability at 2.5 μ M. As previously observed, compounds exhibiting more rigid structures (the absence of the methylene linkers), as well as the presence of polar groups in positions that are less prone to intermolecular interactions (*ortho vs para*), were more likely to display reduced HFF toxicity. Thus, while **55** and **56** were both highly active against *T. gondii*- β -gal (IC₅₀ values of 0.071 and 0.056 μ M, respectively), they presented reduced HFF toxicity at a concentration that was 35 times the *T. gondii* IC₅₀ value.

Overall, the presence of one free (non-protected as ester or amide) OH or NH₂ group appeared to be important for antiparasitic activity, but the presence of two of these groups had detrimental effects on host cell viability. In addition, host cell toxicity was influenced by the position of the OH group. The position and number of other polar substituents such as CF₃ also affected the viability of the host cells. More rigid structures, lacking the methylene spacers on the pendant arms are to be favored. During the secondary screening, 14 compounds, when applied at 2.5 μ M, reduced HFF viability by less than 50%. Compounds **6**, **29**, **30**, **34**, **35**, **36**, **38**, **39**, **41**, **42**, **45**, **48**, **55** and **56** were selected for the tertiary screening to study their impact on splenocyte viability and proliferation.

2.2.4. Tertiary screening

Previous studies [80,81] have proposed different types of ruthenium complexes as potential immunosuppressive agents to treat autoimmune diseases and allergic disorders. However, as this study aimed at developing ruthenium-based compounds for the treatment of *T. gondii* infection, any potential adverse reaction that particularly suppresses host immunity should be avoided. An ideal treatment stategy for toxoplasmosis would promote synergy between host immunity and the action of the applied antiparasitic drug [82].

The 14 selected compounds were evaluated for their immunosuppressive potential. B cells were stimulated with LPS [83,84] and T cells with Con A [85], and the effects of 0.1 and 0.5 μ M of each compound were measured by assessing viability impairment by the AlamarBlue assay. For the compounds for which treatments at 0.5 μ M did not induce more than 50% of T cell death and less than 90% of B cell death, the potential to inhibit the proliferative responses of B and T cells was further assessed by BrdU (5-bromo-2'deoxyuridine) incorporation into DNA measured by ELISA (enzyme-linked immunosorbent assay) [51].

As shown in Fig. 7A and B, 13 out of the 14 tested compounds exerted dose-dependent effects on both B and T cells *in vitro*. The



Fig. 7. Tertiary screening of murine splenocyte cultures with 14 selected ruthenium complexes. The viability of splenocyte cultures stimulated with ConA (T cells, A) and LPS (B cells, B) was measured by AlamarBlue assay. Cultures were treated with 0.1 μ M (dark grey bars) and 0.5 μ M (light grey bars) of each compound. Results are presented as fluorescence intensity at 590 nm (excitation at 530 nm), bars represent the mean emission of 4 replicates \pm standard deviation. The numbers above the bars indicate percentage in relation to the control (CTR = cells only stimulated with ConA (A) and cells only stimulated with LPS (B)). CsA (cyclosporin A, 1 μ M) was used as a known immunosuppressant.

only exception was **41**, which was highly toxic and completely impaired T and B cell viability already at 0.1 μ M. The remaining 13 compounds did not exhibit excessive T cell toxicity at 0.1 μ M (viability ranging from 80 to 128% in relation to untreated cultures). B Cells appeared to be generally more susceptible, with relative values ranging from 35 to 100%. Toxicity was increased when treatments were carried out at 0.5 μ M, with relative viability values ranging from 6 to 74% for T cells and from 2 to 22% for B cells. Twelve compounds clearly exerted increased toxicity in B cells compared to T cells. In contrast, cyclosporin A (CsA), a known immunosuppressive drug used as control, had a more pronounced effect on T cells compared to B cells (23 vs 59% remaining relative viability).

Interestingly, for most compounds the toxicity values for HFF and T/B cells were not comparable. For instance, treatments of HFF with 55 and 56 at 2.5 µM resulted in similarly moderate adverse effects (84 and 74% remaining viability, respectively, Table 6), while the application of 0.5 μ M of the same compounds resulted in severe viability impairment of B and T cells (residual relative viability of 22 and 11%, and 55 and 27%, respectively, Fig. 7A and B, Tables S1 and S2). Likewise, **41** applied at 2.5 μ M did not display HFF toxicity (86% remaining viability, Table 6), but neither T nor B lymphocytes survived the treatments with 41 at 0.1 µM (Fig. 7A and B, Tables S1 and S2). These differences could be partly explained by the different metabolic states of HFF and splenocytes. HFF monolayers are largely confluent, exhibit contact-inhibition and are proliferating only slowly. ConA and LPS stimulated splenocyte cultures undergo rapid proliferation and are metabolically much more active. Nevertheless, these findings emphasize the necessity to include various cell-types for toxicity testing, especially considering that all nucleated cells are subject to infection with T. gondii [3].

Five compounds (**6**, **29**, **35**, **42** and **55**) were further investigated for their capacity to inhibit proliferative responses of T and B cells stimulated with ConA and LPS, respectively, using the BrdU (5bromo-2'-deoxyuridine) incorporation into DNA measured by ELISA (enzyme-linked immunosorbent assay). The results (Fig. 8A and B and Tables S3 and S4) demonstrate that T cell proliferation was not notably affected by any of the compounds at 0.1 μ M, and the same was true for compounds **29**, **42** and **55** at 0.5 μ M. However, treatments at 0.5 μ M with compounds **6** and **35** had a negative impact on T cell proliferative responses (Fig. 8A). B cell proliferation was more strongly affected (Fig. 8B). Compounds **6**, **29** and **35** exerted partial inhibition of B cell proliferation already at 0.1 μ M, and almost entirely abolished cell division at 0.5 μ M. The impairment effect was less pronounced for compounds **42** and **55** at 0.5 μ M, with residual relative proliferation occurring at 39 and 52% compared to non-treated controls, respectively. As observed in the viability tests, cyclosporin A had stronger effect on the proliferation of T cells compared to B cells (21 *vs* 60%).

Compounds **6**, **29**, **35**, **42** and **55** affected the viability and the proliferation of B lymphocytes to a higher degree than T cells. However, when tested at a concentration close to its IC_{50} (0.1 μ M), compound **55** neither affected the viability nor the proliferation of immune cells. In addition, HFF remained largely viable when treated with 2.5 μ M (35 times the *T. gondii* IC_{50} value). As a comparison, the standard drug pyrimethamine, which is currently used in the treatment of toxoplasmosis, was assessed similarly by BrdU ELISA at its *T. gondii* IC_{50} value (0.326 μ M). Residual proliferation after treatments amounted up to 88% in ConA stimulated cultures (T cells) and 57% in LPS stimulated cultures (B cells). Based on these values, compound **55** performs slightly better than pyrimethamine and should be considered for further development.

The factors responsible for the observed differences in the susceptibility of B and T cells upon treatments with these compounds should be investigated. The increased susceptibility of B cells could be through interference in LPS binding to B cell receptors, inhibition of early activation events, or modulation of accessory cells such as macrophages which provide additional essential signals for B cell



Fig. 8. Tertiary screening: effects of compounds **6**, **29**, **35**, **42** and **55** on proliferative responses of murine splenocytes. T cell proliferation (A) was induced by the addition of ConA, B cell proliferation (B) was induced by adding LPS. Compounds were added at 0.1 μ M (dark grey bars) and 0.5 μ M (light grey bars), and BrdU incorporation into DNA was measured by ELISA assay. Bars represent standard deviation from the mean of three replicates Results are presented as absorption intensity at 450 nm, the numbers above the bars indicate percentages in relation to the control (CTR = ConA or LPS plus solvent only). CsA (cyclosporin A, 1 μ M) was used as a known immunosuppressant.

activation [86]. Despite presenting similar effects on the viability of T and B lymphocytes, compounds **35**, **42** and **55** exhibited divergent proliferation inhibition rates. This showed that the induced cellular events responsible for viability loss are different from those responsible for cell division inhibition, and/or that compounds **35** (family 6), **42** (family 7) and **55** (family 8) presenting various structural modifications do not share the same cellular targets. Overall, based on its selective antiparasitic activity and moderate direct effects on B and T cells, compound **55** appear as the most promising for further studies.

Even if this SAR study on trithiolato diruthenium derivatives did not provide conclusive rules for simultaneously improving specific anti-*Toxoplasma* activity and reducing general cytotoxicity, some structural elements that allowed selected compounds to possess high toxicity against *T. gondii* and general low toxicity against B and T cells were identified. These structural elements can constitute a starting point in the design of new, efficient and non toxic antiparasitic trithiolato dinuclear ruthenium compounds. Screening alternative counterions should be considered (e.g., BF₄ in **6** vs Cl⁻ in **2**). The presence of a polar group (e.g., OH or NH₂, but not CO₂H) appears to be favorable. Protection of the hydroxy group as ether could be interesting, but the nature of the protecting group might also be important e.g., a polar trifluoromethoxy (**29**) instead of methoxy group (**28**) should be favored. Reduced flexibility of groups pending on the bridge thiols by the absence of the methylene spacers (e.g., **35** and **55** vs **1/H3**) should be favored. The Bu^t substituents on two of the bridge thiols (**1/H3**) could be replaced by less hydrophobic/bulky Pr^i substituents (**35**) or by polar CF₃ groups (**55**), taking also into account the position of the substituents. Replacement of the hydrophobic Me and Bu^t in the symmetric **H1** and **H2** with polar fluoro groups in **42** also had positive effects.

Additional studies need to be carried out to elucidate which structural features of these compounds mediate antiparasitic efficacy, and what modifications could further reduce HFF toxicity, as well as alleviate viability impairment and antiproliferative effects on B and T cells. Antiparasitic efficacy is influenced by a direct targeting of metabolic functions of the parasite and whether these functions are essential, but also by the effects exerted on the host cell. In addition, these compounds undergo various biophysical and biochemical processes, such as passage through several layers of membranes prior to reaching their target, and potential metabolic breakdown once they are located intracellularly, all of which will impact their stability and efficacy. Previous transmission electron microscopy (TEM) studies in *Toxoplasma, Neospora* and in *Trypanosoma* demonstrated structural alterations in the parasite mitochondrion early after initiation of treatments with ruthenium complexes. In *Toxoplasma* and *Neospora*, these alterations included distinct changes in the mitochondrial matrix, and in the structural integrity of the parasite cristae, while HFF mitochondria remained largely unaltered [50,51,57]. In *T. brucei*, disruption of the mitochondrial membrane potential was demonstrated by tetrame-thylrhodamine ethyl ester assay, alterations in the trypanosome mitochondrion were evidenced by immunofluorescence employing an antibody against mitochondrial Hsp70 and Mitotracker labeling, and the mitochondrial matrix was converted into a translucent mass lacking any cristae as seen by TEM [58]. Future investigations should focus on elucidating the mechanism of action and on the identification of respective intracellular targets.

3. Conclusions

We report here a SAR study of a library of trithiolato-bridged dinuclear ruthenium(II)-arene compounds, which was tested for antiparasitic activity against *T. gondii*- β -gal tachyzoites, and cyto-toxicity against HFF host cells and murine B and T cells. Systematic structural modifications of the parent compounds **H1**, **H2** and **H3** resulted in 56 compounds that were assigned to eight families, with successive structural variations, designed for increasing the antiparasitic activity and reduce the HFF and immune cell toxicity.

The activities of the compounds were assessed following a sequential three-step screening protocol. A primary screening for HFF toxicity and antiparasitic activity was undertaken with all 56 compounds at 0.1 and 1 μ M. 39 compounds, which, at a concentration of 1 μ M, inhibited parasite proliferation by more than 95% and impaired HFF viability by less than 49%, were chosen for the next step. This provided first insights into the biological effects of the different structural variations. For instance, the presence of free carboxy groups or more than one hydroxy and amino substituents on the bridge thiols is detrimental, while that of polar CF₃ groups appears beneficial. Methylene linkers in the pendant arms of two of the bridge thiols should be avoided. Combining one polar hydroxy or amino group on one bridge thiol and two CF₃ substituents on the other two thiols appears to be the best combination for ensuring both anti-*T. gondii* activity and reduced toxicity to the host cells.

The secondary screening included dose-response assays to determine the *T. gondii* IC_{50} values, and the assessment of HFF toxicity of the compounds at 2.5 μ M. The 14 compounds that impaired HFF viability by less than 50% were subjected to the tertiary screening that focused on the effects on the viability and the proliferative potential in murine B and T cells. The results revealed that the presence of one hydroxy group and polar CF₃ groups appears essential, as 9 out of these 14 compounds possess one or both of these two structural elements.

First the effects on the viability of immune cells were investigated at 0.1 and 0.5 μ M, and subsequently the 5 compounds exhibiting the least detrimental effects against B and T cells were further evaluated for their potential to inhibit proliferative responses. The results showed that B cells exhibited a higher susceptibility than T cells. Compound **55** [(η^6 -p-MeC₆H₄Prⁱ)_2Ru₂(μ_2 -SC₆H₄-o-CF₃)₂(μ_2 -SC₆H₄-p-OH)]Cl emerged as a suitable candidate for further development.

4. Materials and methods

4.1. Chemistry

The chemistry related general information, the experimental procedures used for the synthesis and the purification of the compounds, as well as characterization information (1 H and 13 C

NMR spectroscopy, mass spectrometry and elemental analysis data) are presented in detail in the Supporting information.

4.2. X-ray structure determination for 13, 38, 43 and P13

Experimental details regarding the X-ray structure determination for complexes **43** and **P13** are presented in the Supporting information.

Both the crystal of **13** and the crystal of **38** were mounted in air at ambient conditions. All measurements were made on a *RIGAKU Synergy S* area-detector diffractometer [87] using mirror optics monochromated Cu K α radiation ($\lambda = 1.54184$ Å) [88].

For **13** the unit cell constants and an orientation matrix for data collection were obtained from a least-squares refinement of the setting angles of reflections in the range $5.724^{\circ} < 2\theta < 154.274^{\circ}$. For **13** a total of 4310 frames were collected using ω scans, with 0.05 s exposure time, a rotation angle of 0.5° per frame, a crystal-detector distance of 31.0 mm, at T = 173(2) K. For **38** the unit cell constants and an orientation matrix for data collection were obtained from a least-squares refinement of the setting angles of reflections in the range $2.3^{\circ} < \theta < 77.1^{\circ}$. For **38** a total of 2900 frames were collected using ω scans, with 1.3 and 5.19 s exposure time, a rotation angle of 0.5° per frame, a crystal-detector distance of 34.0 mm, at T = 173(2) K.

Data reduction was performed using the *CrysAlisPro* [87] program. The intensities were corrected for Lorentz and polarization effects, and an absorption correction based on the multi-scan method using SCALE3 ABSPACK in *CrysAlisPro* [87] was applied. Data collection and refinement parameters are given in Table S5. The structure was solved by direct methods using *SHELXT* [89], which revealed the positions of all non-hydrogen atoms of the title compound. All non-hydrogen atoms were refined anisotropically.

For **13** H-atoms were assigned in geometrically calculated positions and refined using a riding model where each H-atom was assigned a fixed isotropic displacement parameter with a value equal to 1.2Ueq of its parent atom (1.5Ueq for methyl groups). For **13** all H-atoms were placed in geometrically calculated positions and refined using a riding model where each H-atom was assigned a fixed isotropic displacement parameter with a value equal to 1.2Ueq of its parent atom (1.5Ueq for the methyl groups, the hydroxy group and water).

Refinement of the structures was carried out on F^2 using fullmatrix least-squares procedures, which minimized the function $\Sigma w (F_o^2 - F_c^2)^2$. The weighting scheme was based on counting statistics and included a factor to down weight the intense reflections. All calculations were performed using the *SHELXL-2014*/7 [90] program in OLEX2 [91]. For **38** about 6% of the unit cell volume is filled by heavily disordered co-crystallized solvent molecules. Its electron density was accounted for by the SQUEEZE procedure of PLATON [92].

4.3. Biological assessments

4.3.1. General

If not stated otherwise, the cell culture media were purchased from Gibco-BRL (Zurich, Switzerland) and the biochemical reagents were obtained from Sigma (St. Louis, MO, USA). The compounds used for *in vitro* studies were stored as 1 mM stock solutions in dimethylsulfoxide (DMSO) at -20 °C. The HFF (human foreskin fibroblasts, SCRC-1041.1) were purchased from ATCC and were cultured and maintained as described earlier [93,94]. Tachyzoites of transgenic *T. gondii* strain containing a β -galactosidase reporter strain (*T. gondii*- β -gal, with Tg-RH background) were kindly provided by Prof. David Sibley (Washington University, St. Louis, USA) and were cultured in HFF, and were maintained and prepared for

infections as described earlier [93,94].

4.3.2. Cytotoxicity in non-infected HFF

The cytotoxic effect was measured by AlamarBlue assay [49]. HFF were seeded into 96 well-plates at a density of 5×10^3 cells per well and monolavers were grown to confluency. For the screening, the medium was removed, and monolavers received 200 uL of fresh medium containing 0.1. 1. or 2.5 uM of each compound. Concentrations 0.1 and 1 µM were tested in triplicate wells, and 2.5 µM was assessed in sextuplicate. Non-treated HFF were also included and solvent controls contained DMSO at the corresponding concentrations of 0.01, 0.01 and 0.25%, respectively. Cultures were further maintained at 37 °C/5% CO_{2.} After 72 h, the medium was removed, and cells were washed once with 200 µL phosphate buffered saline (PBS). Subsequently, 200 µL PBS containing resazurin (10 mg/mL dissolved in PBS) was added per well. Plates were read (excitation 530 nm, emission 590 nm wavelength) using an EnSpire multilabel reader (2300 EnSpire™ Multilabel Reader, PerkinElmer, Turku, Finland) immediately (T0) or after 5 h (T5). The cytotoxicity values were calculated and represented as percentage of the respective DMSO control, which represented 100% of HFF viability.

4.3.3. Activity against T. gondii- β -gal tachyzoites

The transgenic *T. gondii*- β -gal strain cultured in HFF was used to assess tachyzoites proliferation inhibition [49]. HFF were seeded into 96 well-plates at a density of 5 × 10⁵ cells per well and monolayers were grown to confluency. Monolayers were infected with 10³ *T. gondii*- β -gal tachyzoites per well, and drugs (dilutions from the 1 mM stock solutions) were applied to cultures concomitantly with the infection.

Initial assessment of drug efficacy was realized by exposing parasite cultures to 0.1 and 1 μ M solutions of each compound in triplicate at 37 °C/5% CO₂ over a period of 72 h, including DMSO solvent controls (at concentrations of 0.01 and 0.1%, respectively).

For IC₅₀ determinations, the compounds were added in the following concentrations: 0.007, 0.015, 0.03, 0.06, 0.12, 0.25, 0.5 and 1 μ M (sextuplicate for all), and cultures were maintained for 72 h at 37 °C/5% CO₂. As solvent control, a solution of 0.1% DMSO (representing the highest DMSO concentration in the test) was used.

Subsequently the medium was removed, wells were washed once with 200 μ L PBS, and cells were permeabilized with 90 μ L PBS containing 0.05% Triton X-100 per well. After addition of 10 µL of 5 mM chlorophenol red-β-D-galactopyranoside dissolved in PBS (CPRG, Roche Diagnostics, Rotkreuz, Switzerland), the absorption shift was measured at wavelength 570 nm at various timepoints on a EnSpire multilabel reader (2300 EnSpire™ Multilabel Reader, PerkinElmer, Turku, Finland). For the initial screening at 0.1 and 1 μ M, the activity, measured as the release of chlorophenol red over time, was calculated as percentage from the respective DMSO control, which represented 100% of *T. gondii*- β -gal growth. For the IC₅₀ assays, the activity measured as the release of chlorophenol red over time was proportional to the number of live parasites down to 50 per well as determined in pilot assays. IC₅₀ values were calculated after the logit-log transformation of the relative growth (RG, control = 1) according to the formula $\ln(RG/[1-RG]) = a \cdot \ln(drug)$ concentration)+b and subsequent regression analysis by the corresponding software tool contained in the Excel software package (Microsoft, Seattle, WA, USA).

4.3.4. Activity against murine B and T cells stimulated with LPS and ConA, respectively

Spleen cell suspensions were prepared as previously described [51]. Female BALB/c mice were purchased from Charles River Laboratories (Sulzfeld, Germany) and were maintained in a common room under controlled temperature and a 14 h dark/10 h light cycle

according to the standards set up by the animal welfare legislation of the Swiss Veterinary Office. The experimental protocol was approved by the Commission for Animal Experimentation of the Canton of Bern, Switzerland (Animal license No. BE101/17). Mice were euthanized using isoflurane and CO₂, and spleens were aseptically removed from euthanized mice. Spleens from female BALB/c mice were removed and carefully minced by passing through 40 um cell strainers. After lysis of red blood cells, splenocytes were counted and seeded in polystyrene 96 well flat bottom sterile plastic plates at 2×10^5 cells/100 µL/well. The impact of the selected compounds on splenic T and B cell viability was assessed by the AlamarBlue assay. Splenocyte cultures in 96 well-plates were used either unstimulated (negative control), or stimulated with ConA (5 μ g/mL), LPS (10 μ g/mL), ConA plus compound or LPS plus compound. The compounds were tested at 0.1 and 0.5 μ M. Cyclosporine A (CsA) applied at 1 µM was included as control strong immunosuppressive compound. After 72 h of culture maintenance at 37 °C/5% CO₂, resazurin (0.1 mg/mL) was added, and the fluorescence intensity was measured immediately (T0), and then after 5 h (T5). Data are presented as mean of emission ± standard deviation (SD) for the indicated numbers. Data comparisons between groups were examined using a Student's t-test (significant when p < 0.001). To measure the impact of compounds on the proliferative potential of B and T cells, the 5-bromo-2'-deoxyuridine (BrdU) cell proliferations assay was performed. Spleen cells were exposed to two concentrations (0.1 and 0.5 µM) of compounds in the presence of ConA (5 μ g/mL) or LPS (10 μ g/mL) for 72 h at 37 °C/5% CO₂. Controls were splenocyte cultures with ConA or LPS devoided of drugs, or without treatment. Cyclosporine A (1 uM) treatment was also included in this assay. Proliferation was assessed with a BrdU cell proliferation kit (QIA58, Merck Millipore). BrdU label was added to the cultures 18 h prior the end of the incubation period, and incorporation of BrdU into DNA was measured by ELISA using peroxidase-conjugated anti-BrdU monoclonal antibody. Immediately after stopping the reaction, the absorbance was measured using dual wavelength of 450/540 nm, in an EnSpire multilabel reader (PerkinElmer, Waltham). Data are presented as mean \pm SD for the indicated numbers. Data comparisons between groups were examined using a Student's t-test (significant when p < 0.01).

Declaration of competing interest

The authors declare no competing financial interest.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.ejmech.2021.113610.

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E. Păunescu, G. Boubaker, O. Desiatkina et al.

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