2,4-Diaminopyrimidine Derivatives as Potent Growth Hormone Secretagogue Receptor Antagonists

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Ghrelin, a gut-derived orexigenic hormone, is an endogenous ligand of the growth hormone secretagogue receptor (GHS-R). Centrally administered ghrelin has been shown to cause hunger and increase food intake in rodents. Inhibition of ghrelin actions with ghrelin antibody, peptidyl GHS-R antagonists, and antisense oligonucleosides resulted in weight loss and food intake decrease in rodents. Here we report the effects of GHS-R antagonists, some of which were potent, selective, and orally bioavailable. A structure—activity relationship study led to the discovery of 8a, which was effective in decreasing food intake and body weight in several acute rat studies.

Introduction

Obesity is a mounting worldwide problem. The cost of medical care and productivity lost due to diseases associated with obesity is estimated to be greater than \$100 billion per year in the United States alone. Desity is associated with a variety of chronic disorders including insulin resistance, diabetes, dyslipidemia, hypertension, coronary heart disease, ischemic stroke (cerebrovascular and thromboembolic disease), pulmonary disease, gout, osteoarthritis, cancer, neurologic diseases, cataracts, gastrointestinal disease, and in women, genitourinary disease. Weight loss has been shown to reverse or slow the progression of these disorders.

Caloric intake must be balanced with the energy expenditure in order to maintain body weight.³ Diet and exercise are crucial to weight control, but they might be assisted by an oral agent that would reduce appetite. This combination could be beneficial for those attempting to lose weight and maintain their lower weight in the fight against obesity.

Ghrelin is a 28-amino acid orexigenic hormone that is produced in the stomach and hypothalamus.⁴ It was found to be an endogenous ligand for the growth hormone secretagogue receptor (GHS-R), a G-protein-coupled receptor (GPCR) expressed primarily in the pituitary gland, the brain, and to a lesser extent in the periphery. Active ghrelin contains an octanoyl group at the Ser-3 hydroxyl group. This unique post-translational acylation is essential for its orexigenic function and is considered necessary for its penetration of the blood-brain barrier.^{4,5} Intracerebroventricular (icv) administration of ghrelin increased food intake in rats, and intravenous (iv) administration of ghrelin increased appetite and food intake in human subjects.⁶⁻¹⁰ Ghrelin levels in the blood appear to be regulated by nutrient intake. Carbohydrates are slightly more effective in lowering ghrelin levels than proteins, which are considerably more effective than lipids.¹¹ The weight gain induced by ghrelin has

been observed to be primarily from adipose tissue and not muscle or bone mass. ¹² Smith and co-workers have demonstrated that mice lacking the GHS-R gene are immune to the orexigenic effects of ghrelin. ¹³ Rats dosed icv with antisense mRNA for GHS-R were found to have lower body weights and significantly decreased food intake compared to controls from weeks 3 to 9 in this study. ¹⁴ Antagonizing the GHS-R receptor with a peptide ([D-Lys-3]-GHRP-6) was shown to decrease food intake and body weight gain. ^{9,12} Small molecule GHS-R antagonists are expected to have similar beneficial effects.

Previously we have reported the discovery of isoxazole carboxamide and tetralin carboxamide based GHS-R antagonists. $^{15-18}$ These two chemically distinct series of GHS-R antagonists effectively displaced 125 I-ghrelin binding as well as antagonizing ghrelin-induced intracellular Ca^{2+} flux in our functional assay. However, these compounds did not exhibit an effect when tested in in vivo animal models. In searching for efficacious GHS-R antagonists, 2,4-diaminopyrimidine derivatives identified from a high throughput screen of the Abbott Laboratories compound collection were explored. 19 The prototype compound, 7 a (Table 1), demonstrated a binding IC50 of 310 nM for GHS-R and an intracellular Ca^{2+} release functional IC50 of 1.38 μ M.

Chemistry

The general synthesis of the 2,4-diaminopyrimidine core began with acylation of 4-nitrophenylacetonitrile (1a) or 4-cyanophenylacetonitrile (1b) with an acid chloride (2a-c, Scheme 1). The resulting aryl-β-ketonitriles (3a-e) were treated with either trimethylsilyldiazomethane or diazomethane solution in diethyl ether to give the corresponding enol ethers (4a-e). Diaminopyrimidines 5a-e were obtained by addition of guanidine to enol ethers 4a-e followed by elimination of methanol. The nitro group in 5a, 5c, and 5e was reduced to provide anilines 6a, 6b, and 6c (Scheme 2). Tompounds 7a-j, 8a-o, 9, and intermediates for 12a, 12b, and 13 were then synthesized via a simple reductive amination reaction between 6a, 6b, and 6c with the corresponding aldehydes. Intermediates for compounds 10a and 10b were synthesized by diazotization of aniline 6a with aqueous workup to form the phenol. The intermediate phenol

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Table 1. IC_{50}	Values of	the	Diaminopy	rimidines	against	GHS-R
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				e 1. IC ₅₀ Values of the Diaminopyrimidines against GHS-R R ₂										
		$_{NH_{2}}$ $\stackrel{R_{3}}{\longmapsto}$ $^{R_{3}}$												
N N N N N N N N N N N N N N N N N N N														
$H_2N^{\prime\prime}N^{\prime\prime}R_1$														
Compound	R,	R ₂	R ₃	Binding IC ₅₀ ± SEM (nM)	Ca ²⁺ Flux IC _{s0} ± SEM (nM)	Compound	R,	R ₂	R,	Binding IC ₅₀ ± SEM (nM)	Ca ²⁺ Flux IC _{s0} ± SEM (nM)			
7a	Et	Н	§−NHCI	310 ± 42	1380 ± 600	8h	-CH ₂ OCH ₂ Ph	Н	₹-NH CF ₃	2.9	97 ± 19			
7b	Et	Н	0 - NH - O - S - O	7.4 ± 2.3	12 ± 8.8	8i	-CH ₂ OCH ₂ Ph	н	-NH S	6.3	170 ± 110			
7e	Et	Н	NH O S O O O O O O O O O O O O O O O O O	2.4	45 ± 43	8j	-CH ₂ OCH ₂ Ph	Н	}-NH	7.7	90 ± 7			
	Et	Н	₹-NH NO ₂	11	23 ± 5.6	8k	-CH _. OCH _. Ph	н	₹-NH	21	86 ± 58			
7d	Et	Н	{-NH O	57 ± 7.9	110 ± 66	81	-CH ₂ OCH ₂ Ph	н	»H	43	320 ± 410			
7e 7f	Et	Н	₹-NH OH	13	110 ± 31	8m	-CH ₂ OCH ₂ Ph	Н	₹-NH	50	980 ± 1060			
7g	Et	Н	{-NHCN	160	780 ± 390	8n	-CH ₂ OCH ₂ Ph	Н		67	700 ± 1240			
7h	Et	н	₹-NH 0 = -CF ₃	21	690 ± 560	80	-CH ₂ OCH ₂ Ph	Н	₹-NH	130	780 ± 540			
	Et	н	₹-NH N	110	220 ± 220	9	\\\	Н	₹-NH	3170 ± 660	3350			
71	Et	н	CI ₹-NH S	290	790 ± 300	10a	-CH ₂ OCH ₂ Ph	н	\$-0	6.9	48 ± 77			
7j			CI CI				-СН,ОСН,Рһ	Н	₹-0	83	3610 ± 2490			
8a	-CH ₂ OCH ₂ Ph	Н	-NH CI	19 ± 11	180 ± 28	10b	-СН,ОСН,Рһ	н	<u> </u>	0.8 ± 0.2	16 ± 1.3			
8b	-CH ₂ OCH ₂ Ph	Н	₹-NH NO ₂	0.2 ± 0.1	5.8 ± 2.6	11a			HÌN—()—NO ₂					
80		**	{-wH /=\ 0	02.01	72.05	11b	Et	н	' HÌN—\ NO2	9.2	190 ± 190			
8c	-CH ₂ OCH ₂ Ph	Н		0.3 ± 0.1	7.2 ± 0.5	11e	Et	Н	HN	9.8 ± 1.0	120 ± 110			
8d	-CH ₂ OCH ₂ Ph	Н	0 -S-CF ₃ 0	1.0	85 ± 89	11d	Et	Н	HN—CI	300	3150 ± 645			
8e	-CH ₂ OCH ₂ Ph	Н	}-NH	1.2 ± 0.2	10 ± 0.7	12a	Et	CI	\$-NH O	4.2 ± 1.5	11 ± 3.5			
8f	-CH ₂ OCH ₂ Ph	н	₹-NHN	1.5	3.6 ± 0.9	12b	Et Et	Br H	-S- -N	3.7	42 ± 14 32 ± 18			
8g	-CH ₂ OCH ₂ Ph	Н	§-NHCN	2.8 ± 0.1	19 ± 3.9	13			-s- 0					

was deprotonated with KOEt and treated with the corresponding benzyl bromides to form compounds 10a and 10b. Halogenated analogues 12a and 12b were made by reacting 7b with N-chlorosuccinimide (NCS), acetic acid, and HClO₄ in methanol/ dichloromethane²³ or bromine in acetic acid,²⁴ respectively (Scheme 3). N-Methylbenzylamine 13 was obtained by reductive amination of 7b with formaldehyde.²⁵

The cyano groups in 5b and 5d were reduced to yield the corresponding benzylamines by hydrogenation (Scheme 2).²⁶ Compounds 11a-d were obtained by displacement of substituted aryl fluorides with the aforementioned benzylamines.²⁷

Results and Discussion

We began our structure-activity relationship (SAR) exploration by extending the diversity of the benzylamine moiety in 7a because of the easy synthetic accessibility of these analogues. The compounds were first run in the GHS-R binding assay, which was used primarily to triage the compounds synthesized. Those compounds with an IC₅₀ value under 2 μ M were tested in an intracellular Ca2+ flux activity assay.15 Functionally pure antagonists are compounds capable of blocking ghrelin-induced intracellular Ca2+ flux and do not cause any Ca2+ flux in the absence of ghrelin. Since the binding assay was used to decide

Scheme 1. General Synthesis of a Common Diaminopyrimidine Core^a

Scheme 2. General Synthesis of Diaminopyrimidine GHS-R Antagonists $7-11^a$

^a Reagents and conditions: (a) H₂, Pd(OH)₂, MeOH; (b) aldehyde, NaBH₃CN, MeOH, HOAc/NaOAc buffer; (c) NaNO₂, H₂SO₄ (aq); (d) KOEt, EtOH, substituted benzyl bromides; (e) H₂, Raney nickel, NH₃/MeOH; (f) arylfluoride, NMP.

Scheme 3. General Synthesis of Phenyl Substituted Benzylamine Diaminopyrimidines **12a** and **12b**^a

 a Reagents and conditions: (a) NCS, HClO₄, HOAc, DCM, MeOH (for 12a), or Br₂, HOAc (for 12b).

which compounds would be tested in the activity assay, the majority of the compounds were only tested once to maximize the throughput. A large number of aromatic ring systems were examined, and the representative analogues are shown in Table 1. Exploration of the SAR about the phenyl ring revealed that benzylamines or benzyl ethers at the 4-position were optimal

for GHS-R blockade. The benzyl groups in turn were most potent when substituted in the 4-position, and these analogues are shown in Table 1. Generally, small hydrophilic groups at this position gave favorable binding potency and Ca²⁺ flux activity. For example, 4-methylsulfonyl, 3,5-difluoro-4-methylsulfonyl, and 4-nitro analogues 7b, 7c, and 7d demonstrated increased binding potency (28- to 130-fold over 7a) and improved Ca²⁺ flux activity (30- to 115-fold over **7a**). Closely related methylketo group (i.e., 7e) or cyano group (i.e., 7g) appeared less effective than 7b, 7c, or 7d. Interestingly, 1-hydroxyethyl analogue 7f showed binding potency similar to **7b**, but it showed a 9-fold decrease in Ca²⁺ flux activity. A larger and more hydrophobic 4-trifluoromethylsulfonyl analogue, **7h**, exhibited a 58-fold decrease in IC₅₀ value in the Ca²⁺ flux activity assay with only a slight decrease (3-fold) in the binding IC₅₀ value compared to **7b**. Chlorinated heterocyclic compounds 7i and 7j demonstrated IC₅₀ values that are within 2-fold of those for **7e** and **7g**, respectively, in the Ca²⁺ flux assay.

In the course of SAR exploration based on the hit compound 7a, a significant enhancement in binding potency was observed when the ethyl side chain at the 6-position of the diaminopyrimidine ring was replaced with a benzyloxymethyl group (e.g., 8a, Table 1). This modification led to a greater than 16fold improvement in binding and an 8-fold improvement of the intracellular Ca2+ flux assay IC50 values. Later SAR results revealed the etheral oxygen atom in 8a played a key role in receptor binding. A structurally similar compound, phenethyl analogue 9 (Table 1), exhibited greater than a 160-fold loss in binding potency compared to 8a and a 20-fold loss of activity in the Ca²⁺ flux assay. The similarity in structure and lack of potency with compound 9 made it a good negative control compound for in vivo studies. Analogues 8a-e exhibited roughly similar SAR trends when directly compared with the corresponding 6-ethyldiaminopyrimidine modifications. 6-Benzyloxydiaminopyrimidine compounds 8f and 8g demonstrated a great improvement over **8a** in binding and intracellular Ca²⁺ flux assay IC₅₀ values. The improvement of **8f** and **8g** over **8a** was greater than that seen with 6-ethyldiaminopyrimidines 7i and 7g, respectively, compared to analogue 7a. These results showed that the SAR was not completely transparent between the 6-ethyldiaminopyrimidines and the 6-benzyloxymethyldiaminopyrimidines. The Ca²⁺ flux assay IC₅₀ values of the chlorothiophene analogue 8i were similar to the parent compound 8a. 4-Trifluoromethyl analogue 8h, 4-methoxy analogue 8j, and 4-pyridyl analogue 8k showed comparable intracellular Ca²⁺ flux assay IC₅₀ values to 4-chloro analogue **8a**. Introduction of a chloro group in the pyridine ring in 8f further improved both binding potency and intracellular Ca²⁺ flux activity over that of 8k. 5-Chlorothiophen-2-yl analogue 8l demonstrated a 2-fold decrease in intracellular Ca^{2+} flux IC_{50} values below that of its isomeric analogue, 8i. Both 8i and 8l were less active

^a Reagents and conditions: (a) Et₃N, DMAP, CH₂Cl₂; (b) TMSCHN₂, CH₂Cl₂; (c) guanidine, EtOH, reflux

Table 2. PK Properties and Brain Plasma Ratios (B/P) of Selected Compounds in Rats

#	AUC^{a-c}	$\mathrm{CL}_{\mathrm{p}}^{a,d}$	$t_{1/2}({\rm h})^a$	$C_{\mathrm{max}}{}^{a,c,e}$	$F(\%)^{a,c,f}$	B/P levels at 1 h ^a $(\mu g/g/\mu g/mL)$	B/P levels at 8 h ^a $(\mu g/g/\mu g/mL)$
7a	2.2	1.8	5.2	0.17	76	2.74/0.25	0.60/0.11
7b	2.8	0.94	6.6	0.32	52	0.05/1.02	n.d. ^g
7d	1.6	2.6	5.5	0.14	68	0.53/0.27	n.d.
7e	0.68	3.7	2.4	0.14	51	0.42/0.12	n.d.
8a	0.12	2.2	n.d.	0.054	5.5	1.07/0.35	0.05/0.03
8a (ip dose)	1.14	2.2	3.9 (ip)	0.26 (ip)	68 (ip)	$0.33/0.50^h$	$0.14/0.19^h$
9	n.d.	n.d.	n.d.	n.d.	n.d.	$1.34/0.80^{h}$	$0.40/0.09^h$
12a	2.7	1.6	8.7	0.16	78	0.03/1.04	n.d.
13	2.7	0.99	3.8	0.43	53	0.04/1.28	n.d.

 a 5 mg/kg dose, mean for n=3. b μ g·h/mL, iv. c Oral unless otherwise specified. d L/h·kg, iv. e μ g/mL, iv. f Oral bioavailability unless otherwise specified. g n.d. = not determined. h 30 mg/kg, ip.

than **8f**. Replacements of the 4-chlorobenzyl group in **8a** with a benzyl, *tert*-butyl, or quinolin-3-yl all resulted in less potent compounds (analogues **8m**, **8n**, and **8o**).

Although the replacement of the benzylamine nitrogen atom with an oxygen atom (e.g. 10a) resulted in 23-fold weaker binding potency and a 7-fold decrease in intracellular Ca^{2+} flux activity compared to 8c, 10a is still a very potent compound. The modification from 4-methylsulfonyl benzyl ether 10a to cyano benzyl ether 10b, however, demonstrated an even further loss of binding potency and a tremendous loss in intracellular Ca^{2+} flux IC_{50} .

Juxtaposition of the nitrogen atom and the benzylic carbon atom of the benzylamine moiety in **7a** or **8a** generally resulted in equipotent antagonists. For example, **11a** and **11d** (Table 1) exhibited similar binding potencies and a 3-fold loss in intracellular Ca²⁺ flux activities compared to **8b** and **7a**, respectively. This demonstrates that the sequence of these two atoms is not critical to the receptor binding. They might also function as a scaffold group that holds the diaminopyrimidine ring and the phenyl ring in the right orientation.

Halogenation of the central phenyl ring in **7b** (i.e. **12a** and **12b**, Table 1) resulted in minimal changes in binding potency and intracellular Ca²⁺ flux activity. Methylation of the benzylamine nitrogen in **7b** to form **13** gave a 2-fold improvement in binding potency and an identical intracellular Ca²⁺ flux activity. These data, in addition to compounds **10a** and **10b**, suggests that the nitrogen atom of the benzylamine moiety does not act as a significant hydrogen bond donor.

Overall, the SAR indicated that a variety of changes were tolerated. Compounds containing hydrogen bonding or electron withdrawing groups at the 3- and 4-positions exhibited the best binding potency and intracellular Ca^{2+} flux activity, whereas lipophilic groups at the same positions diminished binding potency and intracellular Ca^{2+} flux activity. This is interesting since polar interactions are generally more difficult to identify in the lead optimization process. Antagonists **8b** and **8c** represent the most potent of our GHS-R binders with IC_{50} values of 0.21 and 0.30 nM, respectively, in the 125 I-ghrelin displacement assay. Antagonists **8b** and **8f** represent the most active antagonists in the functional assay with intracellular Ca^{2+} flux IC_{50} values of 4-6 nM.

Compound **8a** (Table 1) and several structurally similar analogues were tested against a panel of GPCRs and were found to be selective against adrenergic (IC $_{50} > 33 \,\mu\text{M}$), histaminergic (IC $_{50} > 33 \,\mu\text{M}$), muscarinic (IC $_{50} > 5.5 \,\mu\text{M}$), dopaminergic (IC $_{50} > 33 \,\mu\text{M}$), opioid like-1 (IC $_{50} > 33 \,\mu\text{M}$), and serotonergic/5HT2C (IC $_{50} > 33 \,\mu\text{M}$) receptors. Compound **8a** was also tested against several targets that have been implicated in appetite control, such as cannabinoid-1 receptor (CB $_{1}$) and melaninconcentrating hormone receptor (MCH). In a Cerep screen, compound **8a** was demonstrated to be inactive against CB $_{1}$

(-1% inhibition at 10 μ M). The compound also demonstrated an IC₅₀ of > 10 μ M against MCH in an intracellular Ca²⁺ flux activity assay.

The pharmacokinetic (PK) properties were studied for a number of analogues in this series (Table 2). Compound **7a**, although not extremely potent, exhibited an excellent PK profile. More active analogues **7b**–**e**, **12a**, and **13** demonstrated very similar PK profiles. For example, **7b** showed a rat oral bioavailability of 52%, half-life of 6.6 h, and a C_{max} of 0.32 μ g/mL. Compound **7b** was studied in a 14-day diet induced obesity mouse model study, and severe toxicity was observed during the second week of the study with half of the animals (the other half appeared healthy). This may be due to the fact that **7b** is an extremely potent dihydrofolate reductase (DHFR) inhibitor ($K_i = 0.5 \text{ nM}$).²⁸

Some of the current literature suggests that central nervous system penetration is necessary for both ghrelin and its antagonists to exert their or exigenic effects. GHS receptors have also been found in the periphery, but we were interested in GHS-R antagonists that penetrate the blood—brain barrier (BBB). Brain and plasma levels were measured on selected compounds (Table 2). For example, antagonist **7a** demonstrated brain drug levels of 2.74 μ g/g at 1 h (Table 2), indicating that **7a** crosses the BBB. Methylsulfonyl analogue **7b** exhibited much lower brain drug levels, probably due to its increased molecular size and polarity. 4-Chlorobenzyl analogue **8a** exhibited significantly greater brain drug levels than those of the other benzyl ether side chain analogues, **12a** and **13**.

With this information we chose to utilize compound **8a** (Table 1) in in vivo studies to determine its effects on food intake and body weight in free-fed and fasted Sprague—Dawley rats. Antagonist **8a** was administered intraperitoneally (ip) for these studies since the bioavailability was considerably higher when dosed ip (68%) than after oral (5%) dosing.

In the first experiment, compound **8a** (10 and 30 mg/kg, ip) caused a dose-dependent decrease in cumulative food intake at 1 and 24 h after a single dose in free-fed rats, and at 24 h in 16 h (overnight) fasted rats (Figure 1). The decrease in food intake 24 h after the 30 mg/kg dose was significant and comparable in both free-fed and overnight fasted rats (29% and 24%, respectively). Additionally, body weight in the 30 mg/kg ip free-fed group was significantly reduced 24 h after dosing (Figure 2). However, a statistically significant decrease was not observed in 1 h food intake in overnight fasted rats. The small molecule antagonist induced decrease in food intake seen here is consistent with that seen previously with the peptide antagonist [D-Lys-3]-GHRP-6¹² and an anti-ghrelin antibody.⁹

Our next goal was to determine if the effects on food intake caused by compound **8a** are attributable to functional GHS-R blockade. DHFR inhibition was a primary off-target selectivity concern due to the diaminopyrimidine pharmacophore common

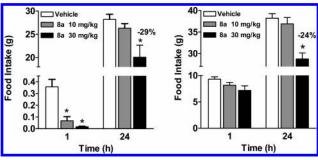


Figure 1. Effect of **8a** (given a single dose of 10 and 30 mg/kg, ip) on food intake in free-fed (left) and overnight-fasted (16 h) (right) Sprague—Dawley rats, mean values \pm SEM for n=10/group, p<0.01 via Anova post hoc Dunnett.

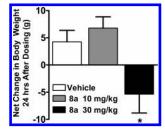


Figure 2. Effect of **8a** (single dose of 10 and 30 mg/kg, ip) on body weight in free-fed Sprague—Dawley rats, mean values \pm SEM for n=10/group, P < 0.05 via Anova post hoc Dunnett.

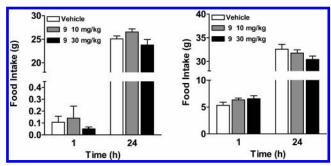


Figure 3. Effect of compound **9** (single dose of 10 and 30 mg/kg, ip) on food intake in free-fed (left) and overnight-fasted (16 h) (right) Sprague—Dawley rats, mean values \pm SEM for n=10/group, Anova post hoc Dunnett.

to known DHFR inhibitors (e.g. pyrimethamine or methotrexate.²⁹ Compound **9** was advanced to in vivo food intake studies for comparison to 8a. Compound 9 is a structurally similar analogue that is at least equipotent at inhibiting DHFR (90% vs 75% inhibition at 1 μ M for 9 and 8a, respectively) but binds GHS-R with >160-fold lower affinity (Table 1). In contrast to results obtained with compound 8a, compound 9 did not cause a significant decrease in food intake at any time point (Figure 3). Analysis of brain and plasma levels in liketreated animals from the same experiment indicates that brain levels of compound 9 were higher than those of 8a throughout the study. Plasma levels of compound 9 were greater than 8a through 2 h (Figure 4). These data taken together support the notion that the compound 8a induced reduction of acute food intake is not due to DHFR inhibition. In addition, all rats dosed with compound 8a and 9 appeared normal after dosing with no injection site irritation. More specifically, no observed behavioral effects consistent with nonhomeostatic effects such as visceral illness were observed.

Additional mechanistic insight was gained by administering compound **8a** (30 mg/kg ip) to overnight (16 h) fasted GHS-R null (KO) and wild-type (WT) mice (Figure 5). At each time point, food intake in vehicle treated WT mice exceeded

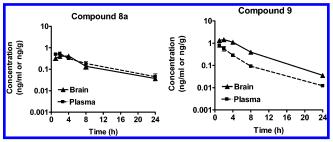


Figure 4. Plasma and brain exposures of **8a** and **9** (given a single dose of 30 mg/kg, ip), mean values \pm SEM for n = 5/timepoint.

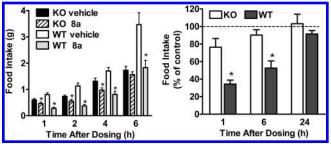


Figure 5. Pharmacological effect of **8a** (given a single dose of 30 mg/kg, ip) on food intake in GHS-R knockout and wild-type mice, mean values \pm SEM for n = 7 - 8/group, P < 0.01 via Tukey-Kramer.

that in KO mice, which is consistent with ghrelin acting via GHS-R contributing to fasting-induced food intake. Figure 5 illustrates that compound 8a caused a relatively greater reduction of food intake in WT than KO mice, suggesting that most of the anorectic efficacy of this compound is mediated through GHS-R.

Conclusion

Many diaminopyrimidine GHS-R antagonists, some of which were potent, selective, and orally bioavailable, were identified. Antagonist 8a, which was initially used for proof-of-principle studies, has demonstrated robust efficacy in blocking food intake and body weight increase in several animal models. The observed activity of the antagonist appears to be mainly due to the antagonism of the GHS-R as evidenced by the lack of efficacy of an inactive but structurally similar compound 9 and by the diminished efficacy of antagonist 8a in GHS-R knockout mice. These compounds showed some cross activity toward DHFR, a key enzyme in DNA synthesis. Compound 9 is a much weaker binder to the ghrelin receptor, a more potent inhibitor of DHFR than 8a, and has similar plasma and brain exposures. Rats dosed with compound 9 did not show a decrease in acute food intake. This indicates that the effects of DHFR inhibition are not responsible for the efficacy of 8a. Compound 8a was tested in a GPCR selectivity panel, a CB1 assay, and MCH activity assay and was found to be inactive against each of these targets. The optimization efforts that lead to the discovery of potent and DHFR-selective diaminopyrimidine GHS-R antagonists will be discussed in a following paper.

Experimental Methods

General Information. Unless otherwise specified, all solvents and reagents were obtained from commercial suppliers and used without further purification. All reactions were performed under nitrogen atmosphere unless specifically noted. Flash chromatography was performed using silica gel (230–400 mesh) from E. M. Science. ¹H NMR spectra were recorded on a Varian Mercury 300 (300 MHz) or Varian UNITY 400 (400 MHz) spectrometer and are reported in ppm (δ) from tetramethylsilane (TMS: δ 0.0 ppm).

Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublets, ddd = doublet of doublet of doublets), coupling constants (Hz), integration. Mass spectral analyses were accomplished on a Finnigan SSQ7000 GC/ MS mass spectrometer using different techniques, including desorption chemical ionization (DCI), atmospheric pressure chemical ionization (APCI), and electrospray ionization (ESI), as specified for individual compounds. Exact mass measurement was performed on a Finnigan FTMS Newstar T70 mass spectrometer. The compound is determined to be "consistent" with the chemical formula if the exact mass measurement is within 4.0 ppm relative mass error (RME) of the exact monoisotopic mass.

Analytical HPLC for selected compounds were performed on two systems. System A: Samples were analyzed with a Gilson analytical HPLC system, using an YMC C-8 column (50×4.6 mm i.d., S-5 um, 120 Å). Solvent system used was acetonitrile/0.1% aqueous NH₄OAc, gradient 0–70% over 15 min at 2 mL/min. System B: Samples were analyzed by Agilent Analytical HPLC 1100 Series, SB-C8 column (Zorbax 3.5 um, 4.6×75 mm). Solvent system used was acetonitrile/0.1% aqueous TFA, gradient 10-70% over 10 min at 1.5 mL/min. Using two analytical HPLC methods, the purity of the compounds was determined to be >95% by UV detection.

Preparative reverse phase HPLC was performed on an automated Gilson HPLC system, using a SymmetryPrep Shield RP18 prep cartridge, 25 mm \times 100 mm i.d., S-7 μm , and a flow rate of 25 mL/min; $\lambda=214,245$ nm; mobile phase A, 5% CH₃CN and 0.1% TFA or 10 mM NH₄OAc in H₂O; mobile phase B, linear gradient 0–70% of B in 19 min. The purified fractions were evaporated to dryness with a ThermoSavant SpeedVac.

General Procedure for Synthesis of the Diaminopyrimidine Compounds 7a-j, 8a-o, 9, 10a-b, 11a-d, 12a-b, and 13. 4-Benzyloxy-2-(4-nitro-phenyl)-3-oxo-butyronitrile (3c). To a solution of 4-nitrophenylacetonitrile (1a, 10.0 g, 61.7 mmol), triethylamine (14.5 g, 144 mmol), and 4-(dimethylamino)pyridine (800 mg, 6.56 mmol) in CH₂Cl₂ (150 mL) at 0 °C was added benzyloxyacetyl chloride (2b, 12.0 g, 64.8 mmol) over 30 min. The resulting mixture was warmed to room temperature, stirred for 2 h, and concentrated in vacuo. The mixture was extracted in ethyl acetate (1 \times 150 mL) and washed with saturated NaHCO₃ (1 \times 80 mL), aqueous HCl (1 \times 80 mL), and brine (1 \times 80 mL). The organic layer was dried over MgSO₄ and concentrated in vacuo to provide crude 4-benzyloxy-2-(4-nitro-phenyl)-3-oxo-butyronitrile (3c, 19.6 g) as a red-brown solid which was used in the next step without further purification. ¹H NMR (300 MHz, DMSO- d_6) δ 8.24 (d, J = 9.2 Hz, 2H), 7.96 (d, J = 8.8 Hz, 2H), 7.25 - 7.46 (m, 5H),4.60 (s, 2H), and 4.48 (s, 2H). MS (APCI) $m/z = 311 \text{ [M + H]}^+$, $309 [M - H]^{-}$

4-Benzyloxy-3-methoxy-2-(4-nitro-phenyl)-but-2-enenitrile (4c). To a solution of 4-benzyloxy-2-(4-nitro-phenyl)-3-oxo-butyronitrile (3c, 9.72 g, 31.4 mmol) in CH₂Cl₂ (80 mL) was slowly added a solution of 2 M trimethylsilyldiazomethane (TMSCHN₂) in diethyl ether (30 mL, 60 mmol). The reaction was monitored by TLC using an aliquot to which had been added 3 M HOAc and was extracted in EtOAc. Glacial HOAc (5 mL) was added dropwise until the excess TMSCHN₂ was consumed as indicated by the absence of N₂ evolution. The mixture was concentrated in vacuo, extracted with EtOAc (1 \times 100 mL), and washed with 1 M HOAc (1 \times 150 mL) and brine (1 × 100 mL). The organic layer was dried over MgSO₄ and concentrated in vacuo to provide 4-benzyloxy-3methoxy-2-(4-nitro-phenyl)-but-2-enenitrile (4c, 10.2 g) which was used in the next step without further purification. ¹H NMR (300 MHz, DMSO- d_6) δ 8.26 (d, J = 9.2 Hz, 2H), 7.87 (d, J = 9.2 Hz, 2H), 7.29-7.45 (m, 5H), 4.69 (s, 2H), 4.66 (s, 2H), 4.09 (s, 3H). MS (APCI) $m/z = 325 \text{ [M + H]}^+$.

6-Benzyloxymethyl-5-(4-nitro-phenyl)-pyrimidine-2,4-diamine (5c). To a solution of 4-benzyloxy-3-methoxy-2-(4-nitro-phenyl)-but-2-enenitrile (**4c**, 10.2 g, 31.4 mmol) in EtOH (60 mL) was added a premixed solution of guanidine hydrochloride (3.605 g, 37.5 mmol) in EtOH (60 mL) and 2.65 M NaOEt in EtOH (14.2

mL, 37.6 mmol). The resulting dark, purple solution was heated at reflux for 3 h, concentrated in vacuo, extracted with EtOAc (1 \times 150 mL), and washed with 0.5 M NaOH (2 \times 100 mL). The mixture was stirred vigorously, then the resulting precipitate was filtered, providing 6-benzyloxymethyl-5-(4-nitro-phenyl)-pyrimidine-2,4-diamine (**5c**, 8.78 g, 80% over three steps) as a light brown solid. $^1\mathrm{H}$ NMR (300 MHz, DMSO- d_6) δ 8.19 (d, J=8.8 Hz, 2H), 7.48 (d, J=9.2 Hz, 2H), 7.18–7.31 (m, 3H), 7.05–7.17 (m, 2H), 6.13 (s, 2H), 5.90 (s, 2H), 4.30 (s, 2H), 3.99 (s, 2H). MS (ESI) m/z=352 [M + H] $^+$, 350 [M - H] $^-$.

5-(4-Amino-phenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine (6b). To a solution of 20 wt % Pd(OH)₂/C (wet, 600 mg) in MeOH (140 mL) was added 6-benzyloxymethyl-5-(4-nitro-phenyl)-pyrimidine-2,4-diamine (**5c**, 5.00 g, 14.25 mmol) in a heavy walled high pressure reaction vessel. The vessel was charged with H₂ (60 psi), and the mixture was stirred at room temperature for 14 h. The reaction mixture was filtered through a nylon membrane, and the solution was concentrated in vacuo followed by trituration with MeOH (20 mL). 5-(4-Aminophenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine (**6b**, 4.34 g, 95%) was obtained as a light yellow solid. ¹H NMR (300 MHz, DMSO- d_6) δ 7.16–7.37 (m, 5H) 6.85 (d, J = 8.1 Hz, 2H), 6.60 (d, J = 8.5 Hz, 2H), 5.86 (s, 2H), 5.49 (s, 2H), 5.12 (s, 2H), 4.35 (s, 2H), 3.97 (s, 2H). MS (ESI) m/z = 322 [M + H]⁺.

6-Benzyloxymethyl-5-[4-(4-chloro-benzylamino)-phenyl]pyrimidine-2,4-diamine (8a). To a solution of 5-(4-amino-phenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine (**6b**, 1.62 g, 5.05 mmol) in 1 M NaOAc/HOAc in MeOH (80 mL, pH 4) was added 4-chlorobenzaldehyde (851 mg, 6.06 mmol). The resulting mixture was stirred for 15 min followed by slow addition of NaBH3CN (375 mg, 6.06 mmol). The reaction mixture was stirred for 16 h at 25 °C. The product was diluted with EtOAc (80 mL) and washed with 1.2 M HCl (75 mL), 2 M NaOH (2×100 mL), and brine (100 mL). The crude material was purified by silica gel chromatography (EtOAc to 10% MeOH in EtOAc gradient) providing 6-benzyloxymethyl-5-[4-(4-chloro-benzylamino)-phenyl]-pyrimidine-2,4-diamine (8a, 1.56 g, 70%). Unless otherwise noted analogues mentioned hereafter were purified by preparative reverse phase HPLC (0-70% CH₃CN in 0.1% TFA $(\bar{aq})/5\%$ CH₃CN or 10 mM NH₄OAc (aq)/ 5% CH₃CN). ¹H NMR (300 MHz, DMSO- d_6) δ 7.33-7.47 (m, 4H), 7.21-7.30 (m, 3H), 7.14-7.20 (m, 2H), 6.89 (d, J = 8.5 Hz, 2H), 6.60 (d, J = 8.5 Hz, 2H), 6.37 (t, J = 5.9 Hz,1H), 5.87 (s, 2H), 5.48 (s, 2H), 4.32 (s, 2H), 4.27 (d, J = 6.1 Hz, 2H), 3.94 (s, 2H). MS (ESI) $m/z = 446 [M + H]^+$, 444 $[M - H]^-$. Anal. (C₂₅H₂₄ClN₅O•0.2 CH₃OH) C, H, N.

5-[4-(4-Chloro-benzylamino)-phenyl]-6-ethyl-pyrimidine-2,4-diamine (7a). Title compound **7a** was synthesized by the general procedure described for **8a** substituting propionyl chloride for benzyloxyacetyl chloride to provide compound **3a** in the first step. Yield, 36%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.40 (m, 4H), 6.85 (d, J = 8.5 Hz, 2H), 6.62 (d, J = 8.5 Hz, 2H), 6.34 (t, J = 5.9 Hz, 1H), 5.76 (s, 2H), 5.31 (s, 2H), 4.26 (d, J = 6.1 Hz, 2H), 2.11 (q, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 354 [M + H]⁺, 352 [M - H]⁻. Anal. (C₁₉H₂₀N₅Cl·0.19H₂O) C, H, N.

6-Ethyl-5-(4-{[4-(methylsulfonyl)benzyl]amino}phenyl)pyrimidine-2,4-diamine (7b). Title compound **7b** was synthesized by the general procedure described for **7a** substituting 4-methane-sulfonylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 80%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.90 (d, J = 8.5 Hz, 2H), 7.65 (d, J = 8.5 Hz, 2H), 6.88 (d, J = 8.5 Hz, 2H), 6.64 (d, J = 8.5 Hz, 2H), 6.54 (t, J = 5.9 Hz, 1H), 6.21 (s, 2H), 5.86 (s, 2H), 4.40 (d, J = 6.1 Hz, 2H), 3.20 (s, 3H), 2.14 (q, J = 7.6 Hz, 2H), 0.97 (t, J = 7.6 Hz, 3H). MS (ESI) m/z = 398 [M + H]⁺. Anal. (C₂₀H₂₃N₅O₂S·0.1CF₃CO₂H) C, H, N.

5-(4-{[3,5-Difluoro-4-(methylsulfonyl)benzyl]amino}phenyl) 6-ethylpyrimidine-2,4-diamine (7c). To a solution of 3,4,5-trifluorobenzaldehyde (250 mg, 1.76 mmol) in DMSO (2 mL) was added sodium methanesulfinate (198 mg, 1.94 mmol). The reaction vessel was purged with N₂, and the reaction mixture was heated in a microwave reactor for 20 min at 185 °C. The reaction was diluted with water (5 mL), and then the mixture was extracted with EtOAc

(10 mL). The organic phase was washed with water (1 × 20 mL) and then brine (1 × 20 mL), dried over Na₂SO₄, and concentrated in vacuo to give 3,5-difluoro-4-methanesulfonylbenzaldehyde (200 mg, 70%), which was used directly without any further purification. 1H NMR (300 MHz, CDCl₃) δ 10.0 (t, J=1.7 Hz, 1H), 7.58 (s, 1H), 7.56 (s, 1H), 3.36 (s, 3H). MS (DCI) $m/z=238~(\mathrm{M}+\mathrm{NH_4})^+$.

Title compound **7c** was synthesized by the general procedure described for **7a** substituting 3,5-difluoro-4-methanesulfonylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 65%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.33 (d, J=10.2 Hz, 2H), 6.88 (d, J=8.5 Hz, 2H), 6.61 (d, J=8.5 Hz, 2H), 6.50 (t, J=6.3 Hz, 1H), 5.72 (s, 2H), 5.27 (s, 2H), 4.39 (d, J=6.1 Hz, 2H), 3.41 (s, 3H), 2.11 (q, J=7.6 Hz, 2H), 0.94 (t, J=7.6 Hz, 3H). MS (ESI) m/z=434 [M + H]⁺. Anal. (C₂₀H₂₁F₂N₅O₂S·1.0CF₃CO₂H) C, H, N. Calcd for C, 48.26; H, 4.05; N, 12.79. Found: C, 47.96; H, 4.46; N, 12.36.

6-Ethyl-5-{4-[(4-nitrobenzyl)amino]phenyl}pyrimidine-2,4-diamine (7d). Title compound **7d** was synthesized by the general procedure described for **7a** substituting 4-nitro-benzaldehyde for 4-chlorobenzaldehyde. Yield, 56%. H NMR (300 MHz, DMSO- d_6) δ 8.22 (d, J = 8.8 Hz, 2H), 7.67 (d, J = 8.8 Hz, 2H), 6.86 (d, J = 8.5 Hz, 2H), 6.62 (d, J = 8.5 Hz, 2H), 6.51 (t, J = 6.1 Hz, 1H), 5.71 (s, 2H), 5.23 (bs, 2H), 4.44 (d, J = 6.1 Hz, 2H), 2.10 (q, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 365 [M + H]⁺. Anal. (C₁₉H₂₀N₆O₂-0.15CF₃CO₂H) C, H, N. Calcd for C, 60.76; H, 5.32; N, 22.03. Found: C, 60.92; H, 5.49; N, 21.47.

1-(4-{[4-(2,4-Diamino-6-ethyl-pyrimidin-5-yl)-phenylamino]-**methyl}-phenyl)-ethanone** (**7e).** Title compound **7e** was synthesized by the general procedure described for **7a** substituting 4-acetylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 72%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.93 (d, J=8.5 Hz, 2H), 7.53 (d, J=8.1 Hz, 2H), 6.85 (d, J=8.5 Hz, 2H), 6.62 (d, J=8.5 Hz, 2H), 6.43 (t, J=5.9 Hz, 1H), 5.73 (s, 2H), 5.27 (s, 2H), 4.36 (d, J=6.1 Hz, 2H), 2.56 (s, 2H), 2.10 (q, J=7.5 Hz, 2H), 0.94 (t, J=7.6 Hz, 3H). MS (ESI) m/z=362 [M + H]⁺. HRMS calcd for (C₂₁H₂₄N₅O) 362.1981. Found: 362.1977.

1-(4-{[4-(2,4-Diamino-6-ethyl-pyrimidin-5-yl)-phenylamino]-methyl}-phenyl)-ethanol (7f). To a solution of 7e (76 mg, 0.21 mmol) in 3:1 MeOH/THF (1.5 mL) at room temperature was added NaBH₄ (32 mg, 0.84 mmol) slowly, then the reaction mixture was stirred for 1 h. The reaction mixture was quenched with 1 M HCl, filtered, and purified by preparative RP-HPLC to give 1-(4-{[4-(2,4-diamino-6-ethyl-pyrimidin-5-yl)-phenylamino]-methyl}-phenyl)-ethanol (55 mg, 72%). ¹H NMR (300 MHz, DMSO- d_6) δ 7.36–7.26 (m, 4H), 6.84 (d, J = 8.5 Hz, 2H), 6.63 (d, J = 8.8 Hz, 2H), 6.25 (t, J = 5.9 Hz, 1H), 5.72 (s, 2H), 5.26 (s, 2H), 5.08 (s, 1H), 4.69 (q, J = 6.3 Hz, 1H), 4.23 (d, J = 5.8 Hz, 2H), 2.11 (q, J = 7.8 Hz, 2H), 1.30 (d, J = 6.4 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 364 [M + H]⁺. Anal. (C₂₁H₂₅N₅O· 0.15CF₃CO₂H) C, H, N.

4-({**[4-(2,4-Diamino-6-ethylpyrimidin-5-yl)phenyl]amino}**-**methyl)benzonitrile** (**7g**). Title compound **7g** was synthesized by the general procedure described for **7a** substituting 4-cyanobenzaldehyde for 4-chlorobenzaldehyde. Yield, 73%. ¹H NMR (500 MHz, DMSO- d_6) δ 7.80 (d, J=8.1 Hz, 2H), 7.58 (d, J=8.1 Hz, 2H), 6.85 (d, J=8.4 Hz, 2H), 6.61 (d, J=8.4 Hz, 2H), 6.44 (t, J=5.9 Hz, 1H), 5.77 (s, 2H), 5.32 (s, 2H), 4.38 (d, J=5.9 Hz, 2H), 2.11 (q, J=7.6 Hz, 2H), 0.94 (t, J=7.5 Hz, 3H). MS (ESI) m/z=345 [M + H]⁺. Anal. (C₂₀H₂₀N₆·0.05CF₃CO₂H) C, H, N.

6-Ethyl-5-[4-(4-trifluoromethanesulfonyl-benzylamino)-phenyl]-pyrimidine-2,4-diamine (**7h**). Title compound **7h** was synthesized by the general procedure described for **7a** substituting (4-trifluoromethanesulfonyl)-benzaldehyde³⁰ for 4-chlorbenzaldehyde. Yield, 71%. ¹H NMR (300 MHz, DMSO- d_6) δ 8.12 (d, J = 8.5 Hz, 2H), 7.85 (d, J = 8.5 Hz, 2H), 6.87 (d, J = 8.5 Hz, 2H), 6.63 (d, J = 8.5 Hz, 2H), 6.54 (t, J = 6.1 Hz, 1H), 5.72 (s, 2H), 5.27 (bs, 2H), 4.50 (d, J = 6.1 Hz, 2H), 2.11 (q, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 452 [M + H]⁺, 450 [M - H]⁻. Anal. (C₂₀H₂₀F₃N₅O₂S·0.1CF₃CO₂H) C, H, N.

5-{4-[(2-Chloro-pyridin-4-ylmethyl)-amino]-phenyl}-6-ethyl-pyrimidine-2,4-diamine (7i). Title compound 7i was synthesized by the general procedure described for 7a substituting 2-chloro-

pyridine-4-carboxaldehyde³¹ for 4-chlorobenzaldehyde. Yield, 78%. $^1\mathrm{H}$ NMR (300 MHz, DMSO- $^4\mathrm{G}$) δ 8.35 (d, J=5.1 Hz, 1H), 7.49 (s, 1H), 7.42 (d, J=5.1 Hz, 1H), 6.88 (d, J=8.5 Hz, 2H), 6.62 (d, J=8.5 Hz, 2H), 6.46 (t, J=6.1 Hz, 1H), 5.72 (s, 2H), 5.27 (bs, 2H), 4.36 (d, J=6.1 Hz, 2H), 2.11 (q, J=7.5 Hz, 2H), 0.94 (t, J=7.5 Hz, 3H). MS (ESI) m/z=455 [M + H] $^+$. Anal. (C18H19ClN6+0.05CF3CO2H) C, H, N.

5-{4-[(5-Chloro-thiophen-3-ylmethyl)-amino]-phenyl}-6-ethyl-pyrimidine-2,4-diamine (7j). Title compound 7j was synthesized by the general procedure described for 7a substituting 5-chloro-2-thiophenecarboxaldehyde for 4-chlorobenzaldehyde. Yield, 30%. ¹H NMR (500 MHz, DMSO- d_6) δ 7.28 (d, J=1.6 Hz, 1H), 7.09 (d, J=1.9 Hz, 1H), 6.87 (d, J=8.7 Hz, 2H), 6.66 (d, J=8.7 Hz, 2H), 6.16 (t, J=5.8 Hz, 1H), 5.28 (bs, 2H), 5.73 (s, 2H), 4.18 (d, J=5.6 Hz, 2H), 2.12 (q, J=7.5 Hz, 2H), 0.95 (t, J=7.5 Hz, 3H). MS (ESI) m/z=360 [M + H]⁺. HRMS calcd for (C₁₇H₁₉ClN₅S) 360.1050. Found: 360.1044.

6-[(Benzyloxy)methyl]-5-{4-[(4-nitrobenzyl)amino]phenyl}-pyrimidine-2,4-diamine (8b). Title compound **8b** was synthesized by the general procedure described for **8a** substituting 4-nitrobenzaldehyde for 4-chlorobenzaldehyde. Yield, 27%. ¹H NMR (300 MHz, DMSO- d_6) δ 8.20 (ddd, J=9.1, 2.5, 2.3 Hz, 2H), 7.66 (d, J=8.8 Hz, 2H), 7.25 (m, 3H), 7.16 (dd, J=7.5, 2.0 Hz, 2H), 6.90 (d, J=8.5 Hz, 2H), 6.60 (d, J=8.5 Hz, 2H), 6.55 (t, J=6.3 Hz, 1H), 5.97 (s, 2H), 5.61 (s, 2H), 4.44 (d, J=6.1 Hz, 2H), 4.32 (s, 2H), 3.95 (s, 2H). MS (ESI) m/z=457 [M + H]⁺, 455 [M - H]⁻. HRMS calcd for ($C_{25}H_{25}N_6O_3$) 457.1988. Found: 457.1984.

6-[(Benzyloxy)methyl]-5-(4-{[4-(methylsulfonyl)benzyl]amino}-phenyl)pyrimidine-2,4-diamine (8c). Title compound **8c** was synthesized by the general procedure described for **8a** substituting 4-methanesulfonylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 67%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.89 (d, J = 8.5 Hz, 2H), 7.65 (d, J = 8.5 Hz, 2H), 7.31–7.15 (m, 5H), 6.90 (d, J = 8.5 Hz, 2H), 6.61 (d, J = 8.8 Hz, 2H), 6.52 (t, J = 6.3 Hz, 1H), 6.01 (s, 2H), 5.64 (s, 2H), 4.41 (d, J = 6.1 Hz, 2H), 4.33 (s, 2H), 3.96 (s, 2H), 3.18 (s, 3H). MS (ESI) m/z = 490 [M + H]⁺. Anal. (C₂₆H₂₇N₅O₃S·0.53CF₃CO₂H) C, H, N. Calcd for C, 59.09; H, 5.04; N, 12.73. Found: C, 59.23; H, 5.48; N, 12.29.

6-[(Benzyloxy)methyl]-5-[4-({4-[(trifluoromethyl)sulfonyl]benzyl}amino)phenyl]pyrimidine-2,4-diamine (8d). Title compound **8d** was synthesized by the general procedure described for **8a** substituting 4-trifluoromethanesulfonylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 70%. ¹H NMR (300 MHz, DMSO- d_6) δ 8.11 (d, J = 8.5 Hz, 2H), 7.85 (d, J = 8.5 Hz, 2H), 7.30–7.15(m, 5H), 6.92 (d, J = 8.5 Hz, 2H), 6.61 (d, J = 8.5 Hz, 2H), 6.57 (t, J = 6.1 Hz, 1H), 5.89 (s, 2H), 5.51 (bs, 2H), 4.51(d, J = 6.1 Hz, 2H), 4.31 (s, 2H), 3.94 (s, 2H). MS (ESI) m/z = 544 [M + H]⁺, 542 [M - H]⁻. Anal. ($C_{26}H_{24}F_{3}N_{5}O_{3}S \cdot 0.1CF_{3}CO_{2}H$) C, H N

1-(4-{[4-(2,4-Diamino-6-benzyloxymethyl-pyrimidin-5-yl)-phenylamino]-methyl}-phenyl)-ethanone (8e). Title compound **8e** was synthesized by the general procedure described for **8a** substituting 4-acetylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 73%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.92 (d, J = 8.5 Hz, 2H), 7.53 (d, J = 8.5 Hz, 2H), 7.30–7.12 (m, 5H), 6.89 (d, J = 8.5 Hz, 2H), 6.60 (d, J = 8.5 Hz, 2H), 6.46 (t, J = 6.1 Hz, 1H), 5.88 (s, 2H), 5.48 (s, 2H), 4.37 (d, J = 5.8 Hz, 2H), 4.31 (s, 2H), 3.94 (s, 2H), 2.55 (s, 2H), 1.89 (s, 3H). MS [ESI] m/z = 454 [M + H]⁺. Anal. ($C_{27}H_{27}N_3O_2 \cdot 0.2CF_3CO_2H$) C, H, N.

6-[(Benzyloxy)methyl]-5-(4-{[(2-chloropyridin-4-yl)methyl]-amino}phenyl)pyrimidine-2,4-diamine (8f). Title compound **8f** was synthesized by the general procedure described for **8a** substituting 2-chloro-pyridine-4-carboxaldehyde³² for 4-chlorobenzaldehyde. Yield, 70%. 1 H NMR (300 MHz, DMSO- d_6) δ 8.34 (d, J = 5.1 Hz, 1H), 7.49 (s, 1H), 7.42 (d, J = 5.1 Hz, 1H), 7.30–7.15 (m, 5H), 6.91 (d, J = 8.5 Hz, 2H), 6.60 (d, J = 8.5 Hz, 2H), 6.49 (t, J = 6.5 Hz, 1H), 5.87 (s, 2H), 5.50 (bs, 2H), 4.36 (d, J = 6.5 Hz, 2H), 4.31 (s, 2H), 3.93 (s, 2H). MS (ESI) m/z = 447 [M + H]⁺, 445 [M – H]⁻. Anal. ($C_{24}H_{23}ClN_6O \cdot 0.15CH_3CO_2NH_4 \cdot 0.6CH_3CN)$ C, H, N.

4-{[(**4-**{2,**4-**Diamino-**6-**[(benzyloxy)methyl]pyrimidin-**5-**yl}-**phenyl**)amino]methyl}benzonitrile (**8g**). Title compound **8g** was synthesized by the general procedure described for **8a** substituting 4-cyanobenzaldehyde for 4-chlorobenzaldehyde. Yield, 54%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.80 (d, J = 8.1 Hz, 2H), 7.59 (d, J = 8.5 Hz, 2H), 7.28–7.15 (m, 5H), 6.90 (d, J = 8.5 Hz, 2H), 6.59 (d, J = 8.5 Hz, 2H), 6.48 (t, J = 6.1 Hz, 1H), 5.87 (s, 2H), 5.47 (bs, 2H), 4.39 (d, J = 6.1 Hz, 2H), 4.31 (s, 2H), 3.93 (s, 2H). MS (ESI) m/z = 502 [M + Na]⁺, 480 [M + H]⁺, 478 [M - H]⁻. Anal. (C₂₆H₂₄N₆O·0.05CF₃CO₂H) C, H, N.

6-Benzyloxymethyl-5-[4-(4-trifluoromethyl-benzylamino)-phenyl]-pyrimidine-2,4-diamine (8h). Title compound **8h** was synthesized by the general procedure described for **8a** substituting 4-trifluoromethylbenzaldehyde for 4-chlorobenzaldehyde. Yield, 37%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.69 (d, J = 8.1 Hz, 2H), 7.61 (d, J = 8.1 Hz, 2H), 7.21–7.29 (m, 3H), 7.13–7.19 (m, 2H), 6.90 (d, J = 8.5 Hz, 2H), 6.60 (d, J = 8.8 Hz, 2H), 6.47 (t, J = 6.1 Hz, 1H), 5.87 (s, 2H), 5.48 (s, 2H), 4.39 (d, J = 5.8 Hz, 2H), 4.31 (s, 2H), 3.94 (s, 2H). MS (ESI) m/z = 437 [M + H]⁺, 435 [M – H]⁻. Anal. ($C_{26}H_{24}F_{3}N_{5}O \cdot 0.05CF_{3}CO_{2}H$) C, H, N.

 $6-Benzyloxymethyl-5-\{4-[(5-chloro-thiophen-3-ylmethyl)-10-chloro-thiophen-3-ylmethyl)-10-chloro-thiophen-3-ylmethyl-10-chlor$ amino]-phenyl}-pyrimidine-2,4-diamine (8i). To a suspension of N-chlorosuccinimide (0.28 g, 2.2 mmol) in 5 mL of CH₂Cl₂ was added thiophene-3-carboxaldehyde (0.18 mL, 2 mmol) followed by few drops of HClO₄. The mixture was stirred at room temperature overnight, after which it was filtered to remove the white precipitate. The filtrate was washed with water, then dried over MgSO₄, and concentrated to give a mixture of 5-chloro-3thiophenecarboxaldehyde and starting material as an oil which was used without further purification. Title compound 8i was synthesized by the general procedure described for 8a substituting the 5-chloro-3-thiophenecarboxaldehyde mixture for 4-chlorobenzaldehyde. Yield, 66%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.15–7.32 (m, 6H), 7.10 (d, J = 1.7 Hz, 1H), 6.91 (d, J = 8.5 Hz, 2H), 6.64 (d, J =8.8 Hz, 2H), 6.22 (t, J = 5.9 Hz, 1H), 5.54 (bs, 2H), 5.91 (s, 2H), 4.33 (s, 2H), 4.19 (d, J = 6.1 Hz, 2H), 3.96 (s, 2H). MS (ESI) m/z= 452 [M + H]^+ , 450 [M - H]^- . Anal. (C₂₃H₂₂ClN₅OS• 0.1CF₃CO₂H) C, H, N.

6-[(Benzyloxy)methyl]-5-{4-[(4-methoxybenzyl)amino]phenyl}-pyrimidine-2,4-diamine (8j). Title compound **8j** was synthesized by the general procedure described for **8a** substituting 4-methoxybenzaldehyde for 4-chlorobenzaldehyde. Yield, 30%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.32 (d, J=8.5 Hz, 2H), 7.28–7.15 (m, 5H), 6.90 (d, J=8.5 Hz, 2H), 6.89 (d, J=8.5 Hz, 2H), 6.62 (d, J=8.5 Hz, 2H), 6.22 (t, J=5.8 Hz, 1H), 5.87 (s, 2H), 5.47 (bs, 2H), 4.33 (s, 2H), 4.19 (d, J=5.8 Hz, 2H), 3.95 (s, 2H), 3.72 (s, 3H). MS (ESI) m/z=442 [M + H]⁺, 440 [M – H]⁻. Anal. (C₂₆H₂₇N₅O₂) C, H, N.

6-[(Benzyloxy)methyl]-5-{4-[(pyridin-4-ylmethyl)amino]-phenyl}pyrimidine-2,4-diamine (8k). Title compound **8k** was synthesized by the general procedure described for **8a** substituting pyridine 4-carbaldehyde for 4-chlorobenzaldehyde. Yield, 23%. 1 H NMR (300 MHz, DMSO- d_6) δ 8.50 (m, 2H), 7.38 (d, J = 5.8 Hz, 2H), 7.26 (m, 3H), 7.17 (m, 2H), 6.90 (d, J = 8.5 Hz, 2H), 6.59 (d, J = 8.5 Hz, 2H), 6.45 (t, J = 5.9 Hz, 1H), 5.87 (s, 2H), 5.49 (s, 2H), 4.33 (d, J = 7.1 Hz, 2H), 4.32 (s, 2H), 3.93 (s, 2H). MS (ESI) m/z = 413 [M + H]⁺, 411 [M - H]⁻. Anal. (C₂₄H₂₄N₆O·1.35CH₃CO₂H) C, H, N.

6-Benzyloxymethyl-5-{4-[(5-chloro-thiophen-2-ylmethyl)-amino]-phenyl}-pyrimidine-2,4-diamine (81). Title compound **81** was synthesized by the general procedure described for **8a** substituting 5-chloro-2-thiophenecarboxaldehyde for 4-chlorobenzaldehyde. Yield, 63%. 1 H NMR (300 MHz, DMSO- d_{6}) δ 7.14–7.30 (m, 5H), 6.89–6.98 (m, 4H), 6.67 (d, J = 8.8 Hz, 2H), 6.39 (t, J = 5.9 Hz, 1H), 5.88 (s, 2H), 5.49 (bs, 2H), 4.41 (d, J = 6.1 Hz, 2H), 4.32 (s, 2H), 3.94 (s, 2H). MS (ESI) m/z = 452 [M + H] $^{+}$, 450 [M – H] $^{-}$. Anal. (C₂₃H₂₂ClN₅OS·0.1CF₃CO₂H) C, H, N

5-[4-(Benzylamino)phenyl]-6-[(benzyloxy)methyl]pyrimidine-2,4-diamine (8m). Title compound 8m was synthesized by the general procedure described for 8a substituting benzaldehyde for 4-chlorobenzaldehyde. Yield, 36%. ¹H NMR (400 MHz, DMSOde) δ 7.40 (m, 2H), 7.34 (m, 2H), 7.26 (m, 4H), 7.18 (dd, J=7.8, 1.7 Hz, 2H), 6.89 (d, J=8.6 Hz, 2H), 6.63 (d, J=8.6 Hz, 2H), 6.29 (t, J=6.0 Hz, 1H), 5.85 (s, 2H), 5.48 (s, 2H), 4.33 (s, 2H), 4.28 (d, J=5.8 Hz, 2H), 3.95 (s, 2H). MS (ESI) m/z=412 [M + H]⁺, 410 [M - H]⁻. Anal. ($C_{25}H_{25}N_5O\cdot0.35H_2O$) C, H, N.

6-[(Benzyloxy)methyl]-5-[4-(neopentylamino)phenyl]pyrimidine-2,4-diamine (8n). Title compound **8n** was synthesized by the general procedure described for **8a** substituting trimethylacetaldehyde for 4-chlorobenzaldehyde. Yield, 48%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.30 (m, 3H), 7.22 (m, 2H), 6.89 (d, J = 8.5 Hz, 2H), 6.68 (d, J = 8.5 Hz, 2H), 6.29 (s, 2H), 6.04 (s, 2H), 5.57 (t, J = 5.8 Hz, 1H), 4.38 (s, 2H), 4.03 (s, 2H), 2.83 (d, J = 5.8 Hz, 2H), 0.97 (s, 9H). MS (ESI) m/z = 392 [M + H]⁺, 390 [M - H]⁻. Anal. (C₂₃H₂₉N₅O·0.1CF₃CO₂H) C, H, N.

6-Benzyloxymethyl-5-{4-[(quinolin-3-ylmethyl)-amino]-phenyl}-pyrimidine-2,4-diamine (80). Title compound **80** was synthesized by the general procedure described for **8a** substituting quinoline-3-carboxaldehyde for 4-chlorobenzaldehyde. Yield, 22%.
¹H NMR (300 MHz, DMSO- d_6) δ 8.97 (d, J=2.4 Hz, 1H), 8.30 (d, J=1.4 Hz, 1H), 8.01 (d, J=8.5 Hz, 1H), 7.94 (d, J=8.5 Hz, 1H), 7.69–7.75 (m, 1H), 7.59 (t, J=7.0 Hz, 1H), 7.12–7.26 (m, 5H), 6.91 (d, J=8.5 Hz, 2H), 6.69 (d, J=8.5 Hz, 2H), 6.50 (t, J=5.9 Hz, 1H), 5.86 (s, 2H), 5.48 (s, 1H), 4.51 (d, J=5.8 Hz, 2H), 4.30 (s, 2H), 3.93 (s, 2H), 3.29 (s, 1H), 3.17 (d, J=5.1 Hz, 1H). MS (ESI) m/z=463 [M + H]⁺. Anal. (C₂₈H₂₆N₆O·0.4CF₃CO₂H·0.5CH₃CN) C, H, N.

5-(4-Nitro-phenyl)-6-phenethyl-pyrimidine-2,4-diamine (5e, Scheme 1). Title compound **5e** was synthesized by the general procedure described for **5c** substituting 3-phenyl-propionyl chloride for benzyloxyacetyl chloride and used without further purification. ¹H NMR (300 MHz, DMSO- d_6) δ 8.20 (d, J = 8.5, 2H), 7.30 (d, J = 8.8 Hz, 2H), 7.11–7.23 (m, 3H), 6.95 (d, J = 6.4 Hz, 2H), 6.04 (s, 2H), 5.72 (s, 2H), 2.76–2.81 (m, 2H), 2.34–2.40 (m, 2H). MS (ESI) m/z = 336 [M + H]⁺, 334 [M – H]⁻.

5-(4-Amino-phenyl)-6-phenethyl-pyrimidine-2,4-diamine (6c, Scheme 2). Title compound **6c** was synthesized by the general procedure described for **6b** and used without further purification.

¹H NMR (300 MHz, DMSO- d_6) δ 7.12–7.20 (m, 3H), 6.97 (d, J = 6.8 Hz, 2H), 6.73 (d, J = 7.8 Hz, 2H), 6.60 (d, J = 7.5 Hz, 2H), 5.82 (s, 2H), 5.34 (s, 2H), 5.09 (s, 2H), 2.72–2.77 (m, 2H), 2.36–2.41 (m, 2H). MS (ESI) m/z = 306 [M + H]⁺, 304 [M - H]⁻.

5-[4-(4-Chloro-benzylamino)-phenyl]-6-phenethyl-pyrimidine- 2,4-diamine (9). Title compound **9** was prepared according to the general procedure for **8a** using **6c** in place of **6b** for the final reductive amination (Scheme 2). Yield, 49%. ¹H NMR (300 MHz, DMSO- d_6) δ 7.31–7.43 (m, 4H), 7.10–7.21 (m, 3H), 6.91–6.93 (d, J = 6.4 Hz, 2H), 6.77 (d, J = 8.5 Hz, 2H), 6.59 (d, J = 8.5 Hz, 2H), 6.35 (t, J = 5.8 Hz, 1H), 5.86 (s, 2H), 5.39 (s, 2H), 4.27 (d, J = 5.8 Hz, 2H), 2.69–2.74 (m, 2H), 2.33–2.39 (m, 2H). MS (ESI) m/z = 430 [M + H]⁺, 428 [M – H]⁻. Anal. (C₂₅H₂₄ClN₅) C, H, N.

4-[4-(2,4-Diamino-6-benzyloxymethyl-pyrimidin-5-yl)-phenoxymethyl]-benzonitrile (10b). To 5-(4-amino-phenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine (654 mg, 2.03 mmol) was added 1 M H₂SO₄ (aq. 7 mL). After being stirred at ambient temperature until the aniline had dissolved, the solution was cooled to 0 °C. (Some precipitate formed after cooling the solution.) To this mixture was added a solution of NaNO₂ (168 mg, 2.43 mmol) dissolved in a minimum amount of water. The reaction was stirred at 0 °C for 10 min, warmed to ambient temperature over 10 min, then heated at reflux for 40 min. The reaction was cooled to ambient temperature followed by addition of EtOAc (10 mL) and of 1.2 M NaHCO₃ (15 mL). A gummy precipitate formed that was dissolved in a small amount of methanol and then partitioned between the EtOAc and aqueous layers to aid dissolution. The layers were separated followed by extraction of the aqueous layer with EtOAc (2 \times 10 mL). The combined EtOAc layers were back extracted with brine (1 \times 10 mL), dried over MgSO₄, filtered, and concentrated to a foam. This was taken up in methanol and reconcentrated to 600 mg (92%) of 4-(2,4-diamino-6-benzyloxymethyl-pyrimidin-5-yl)- phenol which was used without further purification in the next step. 1 H NMR (300 MHz, DMSO- d_{6}) δ 9.43 (s, 1H), 7.24 (m, 5H), 6.99 (m, 2H), 6.79 (m, 2H), 5.92 (s, 2H), 5.56 (s, 2H), 4.33 (s, 2H), 3.95 (s, 2H). MS (ESI) m/z = 323 [M + H]⁺, 321 [M - H]⁻.

To 4-cyanobenzyl bromide (30.4 mg, 0.16 mmol) was added 4-(2,4-diamino-6-benzyloxymethyl-pyrimidin-5-yl)-phenol (50 mg, 0.15 mmol) followed by 0.31 M KOEt in EtOH (0.5 mL). The reaction was shaken at ambient temperature for 8 h, diluted with H₂O (1 mL), and filtered. The precipitate was washed with H₂O and then diethyl ether and recrystallized from EtOH/H₂O to give **10b** as a solid (40 mg, 59%). ¹H NMR (300 MHz, DMSO- d_6) δ 7.88 (d, J = 8.1 Hz, 2H), 7.68 (d, J = 8.1 Hz, 2H), 7.25 (m, 3H), 7.16 (m, 4H), 7.04 (m, 2H), 5.95 (s, 2H), 5.60 (m, 2H), 5.24 (s, 2H), 4.32 (s, 2H), 3.94 (s, 2H). MS (ESI) m/z = 438 [M + H]⁺, 436 [M - H]⁻. Anal. (C₂₆H₂₃N₅O₂·0.27H₂O) C, H, N.

6-[(Benzyloxy)methyl]-5-(4-{[4-(methylsulfonyl)benzyl]oxy}-phenyl)pyrimidine-2,4-diamine (10a). Title compound **10a** was synthesized according to the procedure described for **10b** substituting 4-methanesulfonylbenzyl bromide for 4-cyanobenzyl bromide. Yield, 64%. 1 H NMR (300 MHz, DMSO- 4 6) δ 7.98 (d, 2 8.5 Hz, 2H), 7.75 (d, 2 8.5 Hz, 2H), 7.30–7.04 (m, 9H), 5.95 (s, 2H), 5.59 (s, 2H), 5.26 (s, 2H), 4.32 (s, 2H), 3.39 (s, 2H), 3.22 (s, 3H). MS (ESI) 2 8.5 Mz = 491 [M + H]⁺, 489 [M - H]⁻. Anal. (2 6 4 8.6 N₄0.4 S·0.4 H₂O) C, H, N.

6-[(Benzyloxy)methyl]-5-(4-{[(4-nitro-phenyl)amino]methyl}-phenyl)pyrimidine-2,4-diamine (11a). To a stirred suspension of **5d** (600 mg, 1.8 mmol) in 1.0 N NH₃/MeOH was added Raney Ni (75 mg, prewashed with MeOH and THF). The reaction flask was charged with hydrogen (1 atm) and hydrogenated at 60 °C for 4 h. The solution was cooled to ambient temperature, filtered through Celite, and concentrated under reduced pressure to provide 5-(4-aminomethyl-phenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine as a beige solid (450 mg, 74%). ¹H NMR (300 MHz, DMSO- d_6), δ 7.36 (d, J = 8.1 Hz, 2H), 7.34–7.12 (m, 5H), 7.15 (d, J = 8.1 Hz, 2H), 5.97 (s, 2H), 5.55 (s, 2H), 4.44 (br m, 2H), 4.33 (s, 2H), 3.96 (s, 2H). MS (ESI) m/z = 336 [M + H]⁺.

To a mixture of 5-(4-aminomethyl-phenyl)-6-benzyloxymethyl-pyrimidine-2,4-diamine (30 mg, 0.089 mmol), excess diisopropylethylamine (150 μ L), and 1-fluoro-4-nitrobenzene (19 μ L, 0.18 mmol) was added NMP (1.0 mL). The mixture was heated at 200 °C for 20 min in a microwave reactor, concentrated in vacuo, and then purified by preparative TLC to provide the title compound as a light yellow solid (10 mg, 24%). HNMR (300 MHz, DMSO- d_6), δ 7.98 (d, J = 9.5 Hz, 2H), 7.86 (t, J = 6.1 Hz, 1H), 7.37 (d, J = 8.1 Hz, 2H), 7.30–7.09 (m, 9H), 6.7 (d, J = 9.5 Hz, 2H), 6.01 (s, 2H), 5.58 (s, 2H), 4.46 (d, J = 6.1 Hz, 2H), 4.29 (s, 2H), 3.94 (s, 2H). MS (ESI) m/z = 457 [M + H]⁺, 455 [M - H]⁻. HRMS calcd for (C_{25} H₂5N₆O₃) 457.1983. Found: 457.1984.

6-Ethyl-5-(4-{[(4-nitro-phenyl)amino]methyl}phenyl)-pyrimidine-2,4-diamine (11b). Title compound **11b** was prepared according to the procedure described for **11a** substituting **5b** for **5d** in the reduction. Yield, 70%. ¹H NMR (300 MHz, DMSO- d_6) δ 8.01 (d, J = 9.0 Hz, 2H), 7.83 (t, J = 6.0 Hz, 1H), 7.41 (d, J = 9.0 Hz, 2H), 7.18 (d, J = 8.5 Hz, 2H), 6.73 (d, J = 8.5 Hz, 2H), 5.86 (s, 2H), 5.37 (s, 2H), 4.46 (d, J = 6.0 Hz, 2H), 2.10 (q, J = 7.5 Hz, 2H), 0.95 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 365 [M + H]⁺, 363 [M - H]⁻. Anal. ($C_{19}H_{20}N_6O_2\cdot0.35CF_3CO_2H$) C, H, N.

6-Ethyl-5-[4-([4-(methylsulfonyl)phenyl]amino}methyl)phenyl]pyrimidine-2,4-diamine (11c). Title compound **11c** was prepared according to the procedure described for **11b** substituting 1-fluoro-4-methanesulfonylbenzene for 1-fluoro-4-nitrobenzene. Yield, 72%.
¹H NMR (300 MHz, DMSO- d_6) δ 7.56 (d, J = 9.0 Hz, 2H), 7.41 (d, J = 9.0 Hz, 2H), 7.26 (t, J = 6.0 Hz, 1H), 7.16 (d, J = 8.5 Hz, 2H), 6.73 (d, J = 8.5 Hz, 2H), 5.87 (s, 2H), 5.38 (s, 2H), 4.40 (d, J = 6.0 Hz, 2H), 2.11 (q, J = 7.5 Hz, 2H), and 0.96 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 398 [M + H]⁺, 396 [M - H]⁻. Anal. (C₂₀H₂₃N₅O₂S·1.05CF₃CO₂H) C, H, N.

6-Benzyloxymethyl-5-{4-[(4-chloro-phenylamino)-methyl]-phenyl}-pyrimidine-2,4-diamine (11d). Title compound **11d** was prepared according to the procedure described for **11b** substituting 1-fluoro-4-methanesulfonylbenzene for 1-fluoro-4-nitrobenzene. ¹H

NMR (300 MHz, DMSO- d_6) δ 7.41 (d, J=8.1 Hz, 2H), 7.16 (d, J=8.1 Hz, 2H), 7.07 (d, J=9.2 Hz, 2H), 6.62 (d, J=8.8 Hz, 2H), 6.44 (t, J=6.1 Hz, 1H), 6.19 (s, 2H), 5.81 (s, 2H), 4.29 (d, J=5.8 Hz, 2H), 2.12 (q, J=7.5 Hz, 2H), 0.97 (t, J=7.5 Hz, 3H). MS (ESI) m/z=354 [M + H]⁺, 352 [M - H]⁻. Anal. (C₁₉H₂₀ClN₅·0.05CF₃CO₂H) C, H, N.

5-[3-Chloro-4-(4-methanesulfonyl-benzylamino)-phenyl]-6ethyl-pyrimidine-2,4-diamine (12a). To a mixture of 7b (40 mg, 0.1 mmol), acetic acid (20 μ L), and HClO₄ (500 μ L) in 2:1 dichloromethane/methanol (1.5 mL) was added N-chlorosuccinimide (13 mg, 0.1 mmol). The reaction mixture was stirred at room temperature for 48 h and was partitioned between ethyl acetate (10 mL) and aqueous NaHCO₃ (10 mL). The organic phase was washed with brine, dried with MgSO₄, filtered, and concentrated in vacuo. The residue was purified by flash chromatography on a silica gel column with ethyl acetate/methanol (10:1) to provide 12a (16 mg, 37%). ¹H NMR (300 MHz, DMSO- d_6) δ 7.90 (d, J = 8.5 Hz, 2H), 7.65 (d, J = 8.5 Hz, 2H), 7.05 (d, J = 2.0 Hz, 1H), 6.84 (dd, J =8.1, 2.0 Hz, 1H), 6.57(d, J = 8.1 Hz, 1H), 6.35 (t, J = 6.1 Hz, 1H), 5.80 (s, 2H), 5.42 (bs, 2H), 4.53 (d, J = 6.1 Hz, 2H), 3.19 (s, 3H), 2.09 (q, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) $m/z = 432 \text{ [M + H]}^+, 430 \text{ [M - H]}^-. \text{ Anal. } (C_{20}H_{22}ClN_5O_2S) \text{ C},$

5-[3-Bromo-4-(4-methanesulfonyl-benzylamino)-phenyl]-6-ethyl-pyrimidine-2,4-diamine (**12b**). To a mixture of **7b** (80 mg, 0.2 mmol) in a mixed solvent of acetic acid (400 μ L) and MeOH (1 mL) was added bromine (10 μ L, 0.2 mmol). The reaction mixture was stirred at ambient temperature for 2 h. It was partitioned between ethyl acetate (10 mL) and saturated NaHCO₃ (aq, 15 mL). The organic layer was washed with brine, dried over MgSO₄, and concentrated in vacuo. The crude material was triturated with CH₃CN to provide **12b** as a white solid (68 mg, 72%). ¹H NMR (300 MHz, DMSO- d_6) δ 8.35 (d, J = 5.1 Hz, 1H), 7.49 (s, 1H), 7.42 (d, J = 5.1 Hz, 1H), 6.88 (d, J = 8.5 Hz, 2H), 6.62 (d, J = 8.5 Hz, 2H), 6.46 (t, J = 6.1 Hz, 1H), 5.72 (s, 2H), 5.27 (bs, 2H), 4.36 (d, J = 6.1 Hz, 2H), 2.11 (q, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 455 [M + H]⁺. Anal.(C₂₀H₂₂N₅O₂SBr) C, H, N.

6-Ethyl-5-(4-{methyl[4-(methylsulfonyl)benzyl]amino}phenyl)pyrimidine-2,4-diamine (**13**). To a mixture of **7b** (40 mg, 0.1 mmol) and formaldehyde (37%, 11 μ L, 0.15 mmol) in acetic acid (20 μ L) and MeOH (500 μ L) was added NaBH₃CN (10 mg, 0.15 mmol). The reaction mixture was stirred at ambient temperature for 2 h followed by partitioning between ethyl acetate (10 mL) and saturated NaHCO₃ (aq, 15 mL). The organic layer was washed with brine, dried over MgSO₄, and concentrated in vacuo. The crude material was triturated with CH₃CN to provide **13** as a white solid (34 mg, 83%). ¹H NMR (300 MHz, DMSO- d_6) δ 7.90 (d, J = 8.5 Hz, 2H), 7.52 (d, J = 8.5 Hz, 2H), 6.97 (d, J = 8.5 Hz, 2H), 6.78 (d, J = 8.5 Hz, 2H), 5.74 (s, 2H), 5.28 (bs, 2H), 4.69(s, 2H), 3.20 (s, 3H), 3.07 (s, 3H), 2.12 (q, J = 7.5 Hz, 2H), 0.95 (t, J = 7.5 Hz, 3H). MS (ESI) m/z = 412 [M + H]⁺, 410 [M - H]⁻. Anal. (C₂₁H₂₅N₅O₂S·0.15H₂O) C, H, N.

GHS-R Binding Assay. GHS-R binding assays were performed with membrane preparations. CHO-K cells expressing human ghrelin receptor (Euroscreen) were suspended in sucrose buffer (0.25-0.025% bacitracin) and disrupted by sonication using a vibra cell (Sonics and Materials Inc.) on 70% duty cycle in 15 s pulses on ice for 2.5 min. The homogenate was centrifuged at 60000g for 60 min. Pellets were suspended in tris buffer (20 mM tris pH 7.4, 5 μg/mL pepstatin-A, 0.1 mM PMSF, and 3 mM EDTA). Binding reactions contained 1 μ g membrane as determined by BCA protein assay (Pierce), 0.1 nM [125I]-(25 mM Hepes pH 7.4, 1 mM CaCl₂, 5 mM MgSO₄. and 0.5% protease free BSA). Incubations were carried out at room temperature for 2 h and were terminated by filtration using Filtermate Harvester (Perkin-Elmer) onto GF/C filter plates (Millipore) previously soaked in 0.5% polyethylenimine for 2 h. Filter plates were washed a minimum of $3 \times$ with cold wash buffer (binding buffer w/o BSA, supplemented with 0.5 M NaCl) to remove unbound counts, after which the filter plate bottoms were sealed and 50 μ L of MicroScint 20 was added to each well. Bound [125I]-ghrelin was then determined by scintillation counting using Top Count NXT (Perkin-Elmer). The effects of compound were expressed as percent inhibition of [125I]-ghrelin binding. Sigmoidal curves were fitted by Assay Explorer (MDL) software, and IC₅₀ values determined.

GHS-R Ca2+ Flux Assay. CHO-K cells expressing human GHS-R (Euroscreen) were cultured in Ultra-CHO medium from BioWhittaker supplemented with 1% dialyzed FCS, 1% penicillin/ streptomycin/fungizone, and 400 μ g/mL G418 (all from Life Technologies) at 37 °C in a humidified cell incubator containing 5% CO₂. Cells were plated in black 96-well plates with clear bottom (Costar) and cultured to confluency overnight. Prior to assay, cells were incubated in 100 μ L of Dulbecco's phosphate-buffered saline (DPBS) containing 1000 mg/L D-glucose, 36 mg/L sodium pyruvate, without phenol red (Life Technologies), with 1.14 mM Fluo-4 AM (Molecular Probes), and 0.25 M probenecid (Sigma) for 1-3 h in the dark at room temperature. The dye solution was aspirated, and the cells were washed twice with DPBS using the EL-450X cell washer (BioTech). After the last wash, 100 µL of DPBS was added to each well. Cell plates were then transferred to the FLIPR unit (Molecular Probes). Compound additions were 50 μ L in duplicate or triplicate of 4× final concentration in DPBS containing 0.1% BSA and 4% DMSO. Fluorescence emissions from 96 wells were measured simultaneously at excitation and emission wavelength of 488 and 520 nm, respectively, for 3 min in 1 s intervals for the first minute and 5 s intervals thereafter. During this time, agonist responses, if any, were recorded in the absence of ghrelin. Next, 50 μ L in duplicate or triplicate of 4× final concentration of ghrelin in DPBS containing 0.1% BSA and 4% DMSO were delivered within 1 s by an integrated 96-well pipettor. Fluorescence emissions were measured for another 3 min as above. During this time the antagonist effects of compounds on ghrelin-stimulated calcium flux were recorded. Sigmoidal curves were fitted, and IC₅₀ and EC₅₀ values were determined by GraphPad Prism software. Ghrelin shows an EC₅₀ of 0.2 nM in this assay.

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Supporting Information Available: Elemental analysis data for compounds **7a,b**, **7f-i**, **8a**, **8d-o**, **9**, **10a,b**, **11b-d**, **12a,b**, and **13**, procedures for in vivo assays, and derivation of GHS-R -/-mice. This material is available free of charge via the Internet at http://pubs.acs.org.

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