

A Facile Zr-Mediated Approach to (Z)-Enynols and Its Application to Regioand Stereoselective Synthesis of Fully Substituted Dihydrofurans

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$$\begin{array}{c} R^1 & R^1 \\ O \\ R^2 & R^3 \end{array}$$

$$\begin{array}{c} Cp_2Zr'' \\ Cross-coupling \end{array}$$

$$\begin{array}{c} R^1 \\ R^1 \\ R^2 & Electrophilic \\ Cyclization \end{array}$$

Efficient synthetic approaches to stereodefined (Z)-enynols have been developed through zirconium-mediated cross-coupling reactions of three different components involving alkyne, ketone, and alkynyl bromide in a one-pot procedure. The subsequent electrophilic cyclization of a wide variety of (Z)-enynols affords fully substituted (Z)-5-(1-iodoylidene)-2,5-dihydrofurans with high regio- and stereoselectivity under mild reaction conditions.

Oxygen-containing heterocycles and their analogues widely occur as structural subunits in numerous natural products, which have displayed interesting biological activities and find applications in pharmaceutical use.¹ Furthermore, substituted heterocycles, in particular, halogen-containing ones, are of significant interest since they are useful and versatile synthetic intermediates for further structural elaboration. Electrophile-promoted cycloaddition of unsaturated compounds has proven to be a powerful synthetic route to the wide variety of halogenated heterocyclic compounds.² However, the electrophilic cyclization performed on acetylenic compounds has been far less studied, although alkynes have been frequently employed in transition-metal-catalyzed interor intramolecular annulations to produce fruitful results for the synthesis of carbo- and heterocycles.³ Most of the recent reports focused on arylalkynes bearing orthorelated heteroatomic nucleophiles, which provided ef-

ficient approaches to benzofused heterocycles such as benzothiophenes, 4a,b benzo[b] furans, 4c,d isoquinolines, 4e quinolines, 4f isochromenes, 4g,h furopyridines, 4i indoles, 4j,k isoindolin-1-ones,4l and so on.4 Enynes bearing nucleophilic substituents such as -OH, -NHR, and SR groups would represent an important class of suitable substrates for this strategy due to the structural analogy with arylalkynes; however, only limited reports concerning electrophilic cyclization have been presented in the literature by employing enyne derivatives.⁵ It is known that (Z)-enynols undergo transition-metal-catalyzed transformations to form substituted furans. 6 The preparation of (Z)-enynols are usually achieved through multistep transformations involving Pd/Cu-catalyzed coupling of the terminal alkynes with vinylic halides, addition of alk-1-ynes to vinyl ketones followed by acid-catalyzed allylic isomerization, reduction of methyl (Z)-2-en-4-ynoates etc. 6a,c Nevertheless, enhancing the efficiency of the synthesis of these compounds is still highly attractive. In our ongoing efforts⁷ to maximize the molecular complexity from simple building blocks using zirconium,8 we found that (Z)-2-en-4-yn-1-ols were readily constructed by a one-pot, three-component reaction. This strategy offered the advantage of the efficient route for substituted

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(*Z*)-enynols with a tertiary alcoholic group, which enabled us to develop convenient protocol for the synthesis of highly substituted 5-(1-iodoylidene)-2,5-dihydrofurans with high regio- and stereoselectivity.

Alkynes undergo selective intermolecular coupling with aldehydes or ketones using zirconocenes. The hydrolysis of the thus-formed oxazirconacyclopentene 1 with aqueous HCl solution affords allylic alcohols in good to excellent yields. Takahashi recently reported a new pathway for the formation of 2,5-dihydrofuran derivatives by the reaction of oxazirconacyclopentenes 1 with propynoates. We envisioned that the oxazirconacycles could be converted to (Z)-enynols by further applications of these cross-coupling reactions. The straightforward way is to develop a method for selective alkynylation of Zr-C bond of oxazirconacyclopentenes 1. This was solved by the coupling of zirconacycle 1 with an alkynyl bromide (eq 1).

It was found that 1 did not react with alkynyl bromide directly. Copper-catalyzed or -mediated carbon—carbon bond formation reactions of zirconacycles have been reported, and transmetalation of Zr-C bonds to Cu-C bonds has proven to be an effective methodology for activation of Zr-C bonds. 10 Treatment of oxazirconacycle 1e ($R^1 = Ph$, $R^2 = o\text{-BrC}_6H_4$, $R^3 = Me$) with 2 equiv of 1-hexynyl bromide and 2 equiv of CuCl at room temperature and stirring the mixture for 24 h afforded (Z)enynols 2e in 51% yield after hydrolysis. It should be noted that addition of 2 equiv of LiCl resulted in a cleaner reaction process and reduced reaction time (12 h). In the presence of LiCl, 2e was obtained in higher isolated yield (69%). This result can be explained by the fact that the LiCl salt may facilitate the transmetalation of Zr-C bond to Cu-C bond and stabilize the organocopper species generated in the reaction solution. 11 The representative reactions with various alkynyl bromide are listed in Table 1. Alkynyl bromide bearing alkyl or aryl substituents

TABLE 1. Formation of (Z)-Enynols from Three-Component Reactions Mediated by Zirconium

Alkyne	Ketone	Alkynyl Bromide	Product	Yield (%) ^a
	0	7 iiiiyiiyi Broniido	Troduct	11010 (70)
Ph— — —Ph		Bu Br	2a	(60) ^b
Ph— — —Ph		Ph Br	2b	71 (56)
C_3H_7 ——— C_3H_7	F	Ph -== Br	2c	85 (70)
Ph Ph	F O	Bu -== -Br	2d	70 (64) ^b
Ph Ph	Br	Bu −≡− Br	2e	83 (69)
Ph Ph	Et Et	Bu Br	2f	71 (51) ^c
Ph Ph	o Ci o	Ph─ ─ Br	2g	(74) ^{b,c}
Ph— — —Ph	CI	Bu -=- Br	2h	92 (76) ^b
Ph— — —Ph	S	Bu Br	2i	(57)
Ви— — —Ви		<i>p</i> -CH ₃ OC ₆ H ₄ ─ = -E	Br 2 j	(66)

 a GC yields. Isolated yields are given in parentheses. Unless noted, all of the reactions were done at room temperature for 12 h using alkynyl bromide (2 equiv), CuCl (2 equiv), and LiCl (2 equiv). b Alkynyl bromide (1 equiv), CuCl (1 equiv), and LiCl (2 equiv) were used. c The reaction was performed for 24 h.

reacted very well with oxazirconacycles to afford enynols 2 in 57–92% yields. A wide range of aromatic ketones could be used, especially with halogenated aromatic ketones, affording the corresponding enynols in good to high yields. 2-Acetylthiophene could also be applied in this reaction to give 2i in 57% isolated yield. When an aliphatic ketone such as 3-pentanone was subjected to the reaction, a prolonged reaction time was required (24 h) to afford 2f in 71% yield.

With (Z)-enynols $\mathbf{2}$ in hand, we were interested in exploring the feasibility of using $\mathbf{2}$ in electrophilic cyclization reactions. We began our investigation with enynol $\mathbf{2a}$ ($\mathbf{R}^1 = \mathbf{Ph}$, $\mathbf{R}^2 = \mathbf{Ph}$, $\mathbf{R}^3 = \mathbf{Me}$, $\mathbf{R}^4 = \mathbf{Bu}$) bearing alkyl group at C-5 and a tertiary alcohol group at C-1. The reaction of $\mathbf{2a}$ with \mathbf{I}_2 was done first in solvents such as $\mathbf{CH}_2\mathbf{Cl}_2$ and $\mathbf{CH}_3\mathbf{CN}$ which were frequently used for iodocyclization in the presence of \mathbf{NaHCO}_3 as base. However, only sluggish reaction mixture was observed. After many efforts, we were delighted to find that iodocyclization proceeded smoothly and provided a >99% NMR yield of the 5-ylidene-2,5-dihydrofuran derivative $\mathbf{3a}$ in THF solution with $\mathbf{K}_3\mathbf{PO}_4$ as base. The results for

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TABLE 2. Effect of Base for the Iodocyclization Reaction

base	time (h)	yield ^a (%)	base	time (h)	yield ^a (%)
	6	24	K_3PO_4	1	>99
Na_2CO_3	8	71	NaOCO ₂ CH ₃	4	>99
NaHCO ₃	6	75	KO ^t Bu	9	56
$KHCO_3$	18	18	NEt_3	20	10
KOH	3	37	pyridine	12	78
a NMR	yields.				

optimization of this electrophilic cyclization process are summarized in Table 2. When no base was employed, only 24% of product **3a** was obtained. The addition of carbonate bases such as Na₂CO₃, NaHCO₃, and KHCO₃ increased the yield of **3a** to 71, 75, and 79% NMR yields along with the formation of several side products, which was difficult to be separated from the main product. When a stronger base, KO^tBu was used, 56% yield of 3a was observed. The use of organic base such as Et₃N and pyridine resulted in the formation of **3a** in 10% and 78% yields, respectively. The presence of K₃PO₄ was crucial for a clean, high-yielding reaction within short reaction time (1 h). It was suggested that K₃PO₄ would neutralize the produced HI of iodocyclization and generate K₂HPO₄, while using carbonate bases would produce H₂O. The H₂O generated might lead to the formation of side products. 4e Accordingly, using NaOCO₂CH₃ as reported by Larock et al. in isoquinoline formation reaction^{4e} also afforded the desired product in high yield (>99%).

Based on the above optimization results, we chose the following reaction conditions: 3 equiv of I_2 and 3 equiv of K_3PO_4 or $NaOCO_2CH_3$ in THF stirred at room temperature for an appropriate amount of time. The results are summarized in Table 3.

The alkyne moiety in enynols 2 bearing an aromatic ring as well as alkyl substituent all reacted very well to provide the 5-exo-dig cyclization product in 66-86% yields. The effect of substitution on α -carbon of alcoholic group in 2 has also been examined: enynol 2f bearing two alkyl group on α -carbon of alcoholic group (entry 6) required increased amount of base (6 equiv) and a prolonged reaction time (12 h). Bulky substrates, such as 2g bearing the o-chlorophenyl group at C-1 and phenyl group at the end of the alkyne moiety, tend to hinder cyclization. A stronger reaction condition involving 6 equiv of I_2 and 6 equiv of base was required, and the product of 3g still could be formed in 73% yield within 3 h.

It is interesting to note that in all cases, only 5-exodig iodocyclization occurred and stereoisomerically pure compounds (Z)-3 were found to be the only reaction products, while the regioisomers of pyran derivative derived from 6-endocyclization and the stereoisomer of (E)-3 were not observed. The observed selectivity are rare in electrophilic cyclization reactions. The structure of dihydrofurans were unequivocally confirmed by single-crystal analysis of 3g and 3h. These results clearly showed five-membered oxygen heterocycles and the (Z)-

TABLE 3. Synthesis of Highly Substituted Stereofined 5-Ylidene-2,5-2*H*-furans

Entry	Enynols	Base/Time	Product		Yield (%) ^a
1	2a	K ₃ PO ₄ , 1h	Ph Ph Me Ph	(3a)	74
2	2b	NaOCO ₂ CH ₃ , 1h	Ph Ph Me Ph	(3 b)	66
3	2c	K ₃ PO ₄ , 8h	$\begin{array}{c} C_3H_7 \\ Ph \\ O \\ C_6H_4\text{-}F\text{-}p \end{array}$	(3c)	85
4	2d	K ₃ PO ₄ , 1h	$\begin{array}{c c} Ph & Ph \\ Bu & Me \\ C_6H_4\text{-F-}p \end{array}$	(3d)	82
5	2e	K ₃ PO ₄ , 3h	Ph Ph Me C ₆ H ₄ -Br-o	(3e)	66
6	2f	K ₃ PO ₄ , 12h	Ph Ph Et Et	(3f)	77 ^b
7	2g	NaOCO ₂ CH ₃ , 3h	Ph Ph Me C ₆ H ₄ -Cl-o	(3g)	73 ^c
8	2h	NaHCO ₃ , 4h	$\begin{array}{c c} Ph & Ph \\ Bu & Me \\ C_6H_4\text{-Cl-}o \end{array}$	(3h)	86
9	2i	NaOCO ₂ CH ₃ , 3h	Ph Ph Me thienyl-2	(3i)	76 ^b

 a Isolated yields. b 6 equiv of base was used. c 6 equiv of $\rm I_2$ and 6 equiv of $\rm I_2$ base were used.

SCHEME 1

2
$$\xrightarrow{I_2}$$
 R^1
 R^1
 R^1
 R^1
 R^2
 R^3
 R^3

(E)-3

configuration of **3**. On the basis of the above observations, a possible reaction mechanism is proposed in Scheme 1, which may involve the following steps: (i) Cyclic iodonium ion **4** was formed through coordination of the triple bond with iodine cation. (ii) Anti attack of the oxygen of alcohol group onto iodonium ion led to the formation of (E)-**3**. In intermediate **4**, more of the partial positive charge caused by iodine coordination is located on the alkyne carbon closed to the substituted double bond moiety, since it stabilizes a carbocation much better. Thus, a 5-exo-dig cyclization resulted. (iii) (E)-**3** underwent stereomutation in the presence of I_2 to give the thermodynamically more stable (Z)-product **3**. Similar iodine-promoted isomerization of vinyl iodide compounds was reported previously. ¹²

In summary, we have succeeded in developing an efficient, general, and one-pot procedure to synthesize

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stereodefined (Z)-enynols through zirconium-mediated cross-coupling of alkyne, ketone, and alkynyl bromide. The iodocyclization of a wide variety of (Z)-enynols affords fully substituted 5-ylidene-2,5-dihydrofurans with high regio- and stereoselectivity under mild reaction conditions. This methodology provided a highly efficient route to stereodefined oxygen heterocycles from common starting materials. We are currently exploring the new synthetic potential of electrophilic cyclization reactions.

Experimental Section

Typical Experimental Procedure for the Preparation of Enynol 2a. To a solution of Cp_2ZrCl_2 (0.365 g, 1.25 mmol) in THF (5 mL) was added EtMgBr (1.0 M THF solution, 2.5 mmol) at -50 °C. After the mixture was stirred for 1 h at the same temperature, diphenylacetylene (0.18 g, 1 mmol) was added, and the reaction mixture was warmed to 0 °C and stirred for 3 h. Acetophenone (0.12 g, 1 mmol) was added, and the reaction mixture was heated to 50 °C for 1 h. Then, to the above reaction solution containing oxazirconacyclopentene 1a were added 1-hexynyl bromide (0.32 g, 2 mmol), CuCl (0.20 g, 2 mmol), and LiCl (0.08 g, 2 mmol) at room temperature. After being stirred at room temperature for 12 h, the reaction mixture was quenched with 3 N HCl solution and extracted with diethyl

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ether. The extract was washed with NaHCO $_3$, water, and brine. The solvent was evaporated in vacuo, and the residue was purified by chromatography on a Al $_2$ O $_3$ column (petroleum ether/ethyl acetate =15:1) to afford 245 mg (64%) of product as a yellow oil.

General Procedure for Electrophilic Cyclization of Enynols 2. To a solution of enynol 2 (0.2 mmol) in 1.5 mL of THF were added K_3PO_4 (0.6 mmol) and I_2 (0.6 mmol) at 0 °C. After 5 min, the above suspension was warmed to room temperature and stirred for an appropriate amount of time. The reaction mixture was then diluted with ethyl acetate, washed with saturated $Na_2S_2O_3$ solution, dried (Na_2SO_4), and filtered. The solvent was evaporated under reduced pressure, and the product was isolated by chromatography on an Al_2O_3 column.

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Supporting Information Available: Experimental details and characterization data of compounds $2\mathbf{a}-\mathbf{j}$, $3\mathbf{a}-\mathbf{i}$, and crystallographic data of $3\mathbf{g}$,h. Copies of $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra of all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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