# Reactions of $\beta$ -carbonyl-substituted 4*H*-chromenes and 1*H*-benzo[*f*]chromenes with 5-aminopyrazoles

Vitaly A. Osyanin<sup>1</sup>\*, Dmitry V. Osipov<sup>1</sup>, Kirill S. Korzhenko<sup>1</sup>, Oleg P. Demidov<sup>2</sup>, Yuri N. Klimochkin<sup>1</sup>

<sup>1</sup> Samara State Technical University, 244 Molodogvardeyskaya St., Samara 443100, Russia; e-mail: VOsyanin@mail.ru

<sup>2</sup> North Caucasus Federal University, 1a Pushkina St., Stavropol 355009, Russia; e-mail: odemidov@gmail.com

Translated from Khimiya Geterotsiklicheskikh Soedinenii, 2021, *57*(3), 305–313

Submitted October 26, 2020 Accepted November 26, 2020



A method for the preparation of pyrazolo[1,5-*a*]pyrimidines containing a 2-hydroxybenzyl or (2-hydroxynaphthalen-1-yl)methyl group in position 6 based on the reaction of  $\beta$ -carbonyl-substituted 4*H*-chromenes and their benzo analogs with 5-aminopyrazoles is proposed. A new type of ring-chain tautomerism with the participation of 7-(trifluoromethyl)pyrazolo[1,5-*a*]pyrimidines was discovered.

**Keywords**: 5-amino-1*H*-pyrazoles, 1*H*-benzo[*f*]chromene,  $\beta$ -carbonyl-substituted chromene, 4*H*-chromene, pyrazolo[1,5-*a*]pyrimidines, aza-Michael reaction, (3+3) cyclocondensation, tautomerism.

Pyrazolo[1,5-*a*]pyrimidines are traditionally regarded as privileged structures in medicinal chemistry,<sup>1</sup> and therefore the development of new approaches to their preparation attracts considerable interest.<sup>2</sup> Examples of pharmacologically active pyrazolo[1,5-*a*]pyrimidines include the Z-drugs zaleplon, indiplon, and ocinaplon, which are non-benzodiazepine sedatives,<sup>3</sup> and the dipeptidyl peptidase 4 inhibitor anagliptin<sup>4</sup> used in the treatment of type 2 diabetes (Fig. 1).

Azolopyrimidines are commonly synthesized by the action of 1,3-dicarbonyl compounds or their synthetic equivalents on aminoazoles.<sup>2,5</sup> One of the drawbacks of a number of existing methods is the formation of a mixture of regioisomers in the case of asymmetric 1,3-bielectrophiles.<sup>6</sup> 1-Unsubstituted 5-aminopyrazoles and acyclic 1,3-CCC-bielectrophiles are usually employed for the preparation of pyrazolo[1,5-*a*]pyrimidines.<sup>7</sup> At the same time, the use of heterocyclic compounds, for example, 3-carbonyl-substituted chromones, as three-carbon synthons in the construction of this heterocyclic system is much less common.<sup>8</sup>



Figure 1. Drugs based on pyrazolo[1,5-a]pyrimidines.

In this work, we have shown that the less electrophilic, in comparison with 3-formylchromones,  $\beta$ -carbonylsubstituted 4*H*-chromenes and 1*H*-benzo[*f*]chromenes can be used to prepare pyrazolo[1,5-*a*]pyrimidine derivatives containing the 2-hydroxybenzyl group at position 6. Such an indirect method of C-hydroxybenzylation of heterocyclic compounds is of interest because it is usually impossible to directly introduce this group into  $\pi$ -electron-deficient heterocycles.<sup>9</sup>

The presence of two nonequivalent electrophilic centers in acylchromenes 1 and four nonequivalent nucleophilic centers in 5-aminopyrazoles 2 gives rise to a wide range of products in the reaction of reagents 1 and 2 (Fig. 2). It is believed that, in comparison with the nitrogen atom of the amino group, the endocyclic nitrogen atom in 5-aminopyrazoles 2 is more basic and more sterically hindered as a result of which it attacks the less hindered and more electrophilic carbonyl carbon atom in conjugated systems.<sup>10</sup>

First, we investigated the reaction of 2-trifluoroacetyl-1H-benzo[f]chromene (1a) with 5-amino-1H-pyrazoles 2a-c containing the adamantane pharmacophore fragment as well as a tert-butyl group. It turned out that when the reaction was carried out in the absence of acidic catalysts in CH<sub>2</sub>Cl<sub>2</sub> at room temperature (method I), 7-(trifluoromethyl)-4,7-dihydropyrazolo[1,5-a]pyrimidin-7-ols 3a-c (tautomeric forms **B**) formed which are stable in the solid state. Carrying out the reaction in boiling MeCN (method II) significantly reduces the duration of the process, but the yield of product 3a is somewhat lower. Deprotonation of aminopyrazole 2a by the action of NaH in DMF (method III) does not change the regioorientation of the reaction with chromene 1a. In solution, the cyclic hemiaminal forms of compounds **3a-c** are in equilibrium with the corresponding enamino ketones (tautomeric forms A) (Scheme 1).

In DMSO- $d_6$  solution, form **B** of compounds **3a–c** predominates; however, in the first moment after dissolution, according to <sup>1</sup>H NMR spectroscopy data, the **B** to **A** ratio is slightly higher than after 1–2 h. Therefore, it can be assumed that in the solid state the compound exist in the form of cyclic 4,7-dihydropyrazolo[1,5-*a*]pyrimidines **3a–c** (tautomers **B**).<sup>11</sup> The ratio of tautomeric forms of products **3a–c** can be calculated from the integral intensities of singlet signals of fluorine atoms in the <sup>19</sup>F NMR spectra or the signals of the methylene protons in the <sup>1</sup>H NMR spectra and is approximately **B**:**A** = 6:4. The possibility of the existence of ring-chain tautomerism is largely due to the presence of an electron-withdrawing CF<sub>3</sub> group, which prevents easy elimination of H<sub>2</sub>O and the formation of thermodynamically more stable pyrazolo[1,5-*a*]pyrimidines.

## Scheme 1



Figure 2. Ambident reactivity of  $\beta$ -carbonyl-substituted 4*H*-chromenes 1 and 5-aminopyrazoles 2.

The most characteristic signals in the  ${}^{1}\text{H}$ ,  ${}^{13}\text{C}$ , and  ${}^{19}\text{F}$  NMR spectra for tautomers **A** or **B** of compounds **3a–c** are shown in Figure 3.

Carrying out the reaction of trifluoroacetylchromenes **1a–c** and pyrazoles **2a,b** in the presence of an acid catalyst allows one to directly obtain 7-(trifluoromethyl)pyrazolo-[1,5-*a*]pyrimidines **4a–c**. *i*-PrOH in the presence of BF<sub>3</sub>·OEt<sub>2</sub> under reflux (method IV) or AcOH under reflux (method V) can be used. In addition, pyrazolo[1,5-*a*]pyrimidine **4a** was obtained by dehydration of hemiaminal **3a** in *i*-PrOH in the presence of BF<sub>3</sub>·OEt<sub>2</sub> (method VI), or in MeOH in the presence of catalytic amounts of *p*-TsOH under reflux (method VII) (Scheme 2).



**Figure 3.** Identification of tautomers **A** and **B** of products **3a–c** by characteristic signals in <sup>1</sup>H (red), <sup>13</sup>C (blue), and <sup>19</sup>F (green) NMR spectra ( $\delta$  in ppm, *J* in Hz).





The assignment of the obtained products to 5- or 7-trifluoromethyl-substituted isomers is based on the fact that the enamine carbon atom bonded to the trifluoromethyl group and to the nitrogen atom with a single bond usually appears at 136 ppm, while the signal of the azomethine carbon atom bonded to the trifluoromethyl group resonates further downfield (at 145 ppm) in the spectrum of compounds with a fixed position of double bonds<sup>5a,10,12</sup> (Scheme 2). In the <sup>13</sup>C NMR spectra of compounds 4a-c, the C-5 carbon atom of the pyrazolo[1,5-a]pyrimidine ring appears at 151.2-153.6 ppm. The strong deshielding is due to the acceptor effect of the neighboring pyridine-type nitrogen atom. The trifluoromethyl carbon atom and carbon atom C-7 are found as quartets at 121.2-121.7 ppm  $({}^{1}J_{CF} = 275.6 \text{ Hz})$  and 129.7–130.3 ppm  $({}^{2}J_{CF} = 34.3 \text{ Hz})$ , respectively. These facts confirm the formation of regioisomers with a trifluoromethyl group at position 7. The carbon atom C-3 appears at 93.5–93.9 ppm.

Isolation of intermediate products confirms the reaction mechanism which encompasses the conjugated addition of the exocyclic amino group of pyrazole 2 to the  $\alpha$ -carbon atom of the pyran ring of chromene 1, the opening of the dihydropyran fragment, and then the nucleophilic addition

of the endocyclic nitrogen atom of the pyrazole ring to the carbonyl carbon atom, followed by dehydration in an acidic medium.

The reaction of 1*H*-benzo[*f*]chromene-2-carbaldehydes **1d**,**e** with 5-aminopyrazole **2d** in AcOH under reflux is complete in less than 15 min and leads to the formation of pyrazolo[1,5-*a*]pyrimidines **4d**,**e** in 73–79% yields. The reaction with such a sluggish Michael acceptor as 2-benzoyl-1*H*-benzo[*f*]chromene (**1f**) proceeds much more slowly and leads to product **4f** in 58% yield (Scheme 3). The formation of 7-phenyl-substituted derivative **4f** is confirmed by the presence in its <sup>13</sup>C NMR spectrum of a signal of the C-5 carbon atom at 155.4 ppm.<sup>13</sup>

The presence of an ester group in the structure of pyrazole 2d, which is in direct conjugation with the amino group, leads to a decrease in the nucleophilicity of the latter, as a result of which the reaction rate of trifluoroacetylbenzochromenes 1a,g with compound 2d is lower than that with 5-aminopyrazoles 2a–c. Moreover, in contrast to pyrazolo[1,5-*a*]pyrimidines 4a–f, products 4g,h exist in the solid state in the form of cyclic N,O-acetals (tautomeric form B) (Scheme 3), which for compound 4g was confirmed by X-ray structural analysis (Fig. 4). It

#### Scheme 3





Figure 4. Molecular structure of compound 4g with atoms represented as thermal vibration ellipsoids with 50% probability.

should also be noted that 4,9a-dihydro-6*H*-pyrano[3,2-*e*]-pyrazolo[1,5-*a*]pyrimidine **4i** (Scheme 3) and its various arene-condensed variants were not previously described.

In the DMSO- $d_6$  solution of compounds 4g,h, tautomer **B** predominates, as well. The content of the minor isomers 4g,h (A) is about 20%, which can be easily determined from the integrated intensities of signals of fluorine atoms in the <sup>19</sup>F NMR spectra which are detected at -60.0 (form A) and -75.5 ppm (form B). Nevertheless, upon dissolution of the colorless products 4g,h in DMSO- $d_6$ , the solution acquires a yellow color characteristic of pyrazolo[1,5-*a*]-pyrimidines. The ease of heterocyclization, in comparison with compounds 4a-c, can be explained by the higher electrophilicity of the C-7 carbon atom of the pyrazolo[1,5-*a*]-pyrimidine ring due to the presence of an acceptor ester group and the contribution of the corresponding resonance structure to the distribution of electron density (Scheme 3).

1-Unsubstituted aminopyrazoles 2a-d act as 1,3-NCNbinucleophiles in all the reactions presented above. At the same time, 5-aminopyrazole 2e substituted at position 1 reacts with 2-trifluoroacetyl-1*H*-benzo[f]chromene (1a) already as 1,3-NCC-binucleophile. This transformation was carried out in DMF under reflux in the presence of catalytic amounts of *p*-TsOH (Scheme 4). In the <sup>13</sup>C NMR spectrum of product 5, the presence of the quartet signal of the carbon atom C-4 at 129.6 ppm ( ${}^{2}J_{CF} = 29.6$  Hz) and the strongly deshielded signal of the C-6 atom at 149.8 ppm indicates the formation of 4-(trifluoromethyl)-1*H*-pyrazolo-[3,4-*b*]pyridine 5 rather than its isomeric 6-trifluoromethyl derivative.<sup>5a,14</sup> Apparently, in this case, too, the initial step of the reaction is the aza-Michael reaction with the participation of an exocyclic amino group.

To conclude, we have demonstrated that  $\beta$ -carbonylsubstituted chromenes act as masked 1,3-dicarbonyl compounds in reactions with 5-aminopyrazoles and can be effective precursors for the preparation of 6-(2-hydroxybenzyl)pyrazolo[1,5-*a*]pyrimidines.

# Experimental

IR spectra were registered on a Shimadzu IRAffinity-1 Fourier transform spectrometer equipped with a Specac Diamond ATR GS10800-B accessory. <sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR spectra (400, 100, and 376 MHz, respectively), as well as DEPT-135 spectra were registered on a JEOL JNM-ECX400 spectrometer in DMSO- $d_6$  (compounds **2b**, **3a**,**b**,**c**, **4a–e,g,h**, **5**) or CDCl<sub>3</sub> (compounds **2a,e**, **4f**) using the



residual solvent signals (DMSO- $d_6$ : 2.50 ppm for <sup>1</sup>H nuclei, 39.5 ppm for <sup>13</sup>C nuclei; CDCl<sub>3</sub>: 7.26 ppm for <sup>1</sup>H nuclei, 77.2 ppm for <sup>13</sup>C nuclei) or CFCl<sub>3</sub> (0.0 ppm for <sup>19</sup>F nuclei) as internal standards. Elemental analysis was performed on a Euro Vector EA-3000 CHNS-analyzer. Melting points were determined by the capillary method on an SRS OptiMelt MPA100 apparatus. Monitoring of the reaction progress and assessment of the purity of synthesized compounds were done by TLC on Merck M60 F<sub>254</sub> plates, eluent CH<sub>2</sub>Cl<sub>2</sub>, visualization with UV light at 254 nm or by iodine stain.

The starting chromenes 1a-g,<sup>15</sup> as well as pyrazoles  $2c^{16}$  and  $2d^{17}$  were obtained by known methods.

3-(Adamantan-1-yl)-1H-pyrazol-5-amine monohydrate (2a).  $N_2H_4$ · $H_2O$  (100%, 10 ml) was added to a solution of 3-(adamantan-1-yl)-3-oxopropanenitrile (3.00 g, 15 mmol) in EtOH (50 ml), and the resulting mixture was heated under reflux for 5 h. The volatile components were evaporated under reduced pressure. The residue was coevaporated with PhMe (2×20 ml) and dissolved in cyclohexane (10 ml) by heating, then H<sub>2</sub>O (0.3 ml, 17 mmol) was added. The formed precipitate was filtered off, washed with cyclohexane (2 ml), and dried in air at room temperature. Yield 1.66 g (47%), colorless crystals, mp 139–140°C. IR spectrum, v, cm<sup>-1</sup>: 3400–3050 (NH, NH<sub>2</sub>, H<sub>2</sub>O), 2904, 2846 (CH Ad), 1689, 1612, 1573, 1502, 1485, 1450, 1319, 1253, 1103, 991, 775, 713. <sup>1</sup>H NMR spectrum, δ, ppm: 1.69–1.77 (6H, m, 3CH<sub>2</sub> Ad); 1.84–1.86 (6H, m, 3CH<sub>2</sub> Ad); 2.03 (3H, br. s, 3CH Ad); 4.61 (5H, br. s, NH, NH<sub>2</sub>, H<sub>2</sub>O); 5.41 (1H, s, H-4). <sup>13</sup>C NMR spectrum, δ, ppm: 28.4 (3CH Ad); 33.1 (C Ad); 36.6 (3CH<sub>2</sub> Ad); 42.2 (3CH<sub>2</sub> Ad); 88.9 (C-4); 154.1; 155.6. Found, %: C 66.25; H 8.92; N 17.75. C<sub>13</sub>H<sub>19</sub>N<sub>3</sub>·H<sub>2</sub>O. Calculated, %: C 66.35; H 8.99; N 17.86.

**3-(5-Amino-1***H***-pyrazol-3-yl)adamantan-1-ol (2b)**. A mixture of 3-(3-hydroxyadamantan-1-yl)-3-oxopropanenitrile (0.90 g, 4.1 mmol), N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O (100%, 3 ml), and EtOH (15 ml) was heated under reflux for 5 h. The solvent and excess N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O were evaporated under reduced pressure, and the residue was recrystallized from CHCl<sub>3</sub>– MeOH, 10:1 mixture. Yield 0.82 g (86%), colorless crystals, mp 143–145°C. IR spectrum, v, cm<sup>-1</sup>: 3450–3050 (OH, NH, NH<sub>2</sub>), 2920, 2850 (CH Ad), 1612, 1566, 1492, 1454, 1338, 1315, 1257, 1118, 1080, 1033, 1002, 925, 798, 648. <sup>1</sup>H NMR spectrum, δ, ppm: 1.49 (2H, br. s, CH<sub>2</sub> Ad); 1.53–1.65 (10H, m, 5CH<sub>2</sub> Ad); 2.10 (2H, br. s, 2CH Ad); 4.37 (3H, br. s, NH<sub>2</sub>, OH); 5.10 (1H, s, H-4); 9.98 (1H, br. s, NH). <sup>13</sup>C NMR spectrum, δ, ppm: 30.6 (2CH Ad); 35.6 (CH<sub>2</sub> Ad); 36.3 (C Ad); 41.4 (2CH<sub>2</sub> Ad); 45.0 (2CH<sub>2</sub> Ad); 50.3 (CH<sub>2</sub> Ad); 67.0 (COH); 87.4 (C-4); 153.9 (2C). Found, %: C 67.02; H 8.16; N 17.90.  $C_{13}H_{19}N_{3}O$ . Calculated, %: C 66.92; H 8.21; N 18.01.

3-(Adamantan-1-yl)-1-methyl-1H-pyrazol-5-amine (2e). A mixture of 3-(adamantan-1-yl)-3-oxopropanenitrile (0.80 g, 3.9 mmol), MeNHNH<sub>2</sub> (2 ml), and EtOH (10 ml) was heated under reflux for 10 h. The solvent was evaporated under reduced pressure, and the residue was recrystallized from cyclohexane. Yield 0.51 g (56%), colorless crystals, mp 199–200°C. IR spectrum, v, cm<sup>-1</sup>: 3399, 3314, 3210 (NH<sub>2</sub>), 2904, 2847 (CH Ad), 1632, 1562, 1523, 1450, 1419, 1385, 1361, 1315, 1265, 1250, 1172, 1103, 991, 748. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.71–1.75 (6H, m, 3CH<sub>2</sub> Ad); 1.86–1.89 (6H, m, 3CH<sub>2</sub> Ad); 2.00 (3H, br. s, 3CH Ad); 3.61 (3H, s, CH<sub>3</sub>); 3.67 (2H, br. s, NH<sub>2</sub>); 5.38 (1H, s, H-4). <sup>13</sup>C NMR spectrum, δ, ppm: 28.7 (3CH Ad); 34.0 (C Ad); 34.1 (CH<sub>3</sub>); 36.9 (3CH<sub>2</sub> Ad); 42.6 (3CH<sub>2</sub> Ad); 87.5 (C-4); 144.9 (C-5); 161.0 (C-3). Found, %: C 72.76; H 9.11; N 18.08. C<sub>14</sub>H<sub>21</sub>N<sub>3</sub>. Calculated, %: C 72.69; H 9.15; N 18.16.

**2-(Adamantan-1-yl)-6-[(2-hydroxynaphthalen-1-yl)methyl]-7-(trifluoromethyl)-4,7-dihydropyrazolo[1,5-***a***]pyrimidin-7-ol (3a). Method I. A solution of chromene 1a (200 mg, 0.72 mmol) and pyrazole 2a (156 mg, 0.72 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 ml) was kept without stirring at room temperature for 20 h. The formed precipitate was filtered off and recrystallized from EtOH.** 

Method II. A mixture of chromene **1a** (200 mg, 0.72 mmol), pyrazole **2a** (156 mg, 0.72 mmol), and MeCN (5 ml) was heated under reflux for 15 min. The mixture was cooled to 0°C, the formed precipitate was filtered off and recrystallized from EtOH.

Method III. NaH as 60% dispersion in mineral oil (40 mg, 1.0 mmol) was added to a solution of pyrazole 2a (156 mg, 0.72 mmol) in DMF (4 ml). The mixture was stirred at room temperature for 10 min until the evolution of H<sub>2</sub> ceased, then chromene 1a (200 mg, 0.72 mmol) was added, and the resulting mixture was kept without stirring at room temperature for 20 h. The mixture was poured into H<sub>2</sub>O (10 ml), and AcOH was added to pH 6. The formed precipitate was filtered off, washed with H<sub>2</sub>O (3 ml), and recrystallized from EtOH.

Yield 303 mg (85%, method I), 280 mg (78%, method II), 307 mg (86%, method III), colorless crystals, mp 209– 210°C. IR spectrum, v, cm<sup>-1</sup>: 3233, 3144, 3117, 3047, 2985 (OH, NH), 2904, 2850 (CH Ad), 1659, 1618, 1597, 1566, 1489, 1462, 1415, 1333, 1315, 1230, 1196, 1138, 1026, 995, 964, 814, 787, 748. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): tautomer A: 1.68 (6H, br. s, 3CH<sub>2</sub> Ad); 1.81–1.83 (6H, m, 3CH<sub>2</sub> Ad); 1.97 (3H, br. s, 3CH Ad); 4.04 (2H, s, CH<sub>2</sub>); 5.77 (1H, d, *J* = 1.8, H-4); 7.17–7.27 (2H, m, H Ar); 7.38 (1H, ddd, *J* = 8.2, *J* = 7.1, *J* = 1.1, H Ar); 7.68 (1H, d, *J* = 9.0, H Ar); 7.74 (1H, d, *J* = 7.8, H Ar); 7.97 (1H, d, *J* = 8.5, H Ar); 8.12 (1H, d, *J* = 13.5, =C<u>H</u>NH); 10.09 (1H, d, *J* = 13.5, =CHNH); 11.10 (1H, br. s, OH); 12.20 (1H, d, J = 1.8, NH); tautomer **B**: 1.68 (6H, br. s, 3CH<sub>2</sub> Ad); 1.81-1.83 (6H, m, 3CH2 Ad); 1.97 (3H, br. s, 3CH Ad); 3.83  $(2H, s, CH_2)$ ; 5.29 (1H, s, H-3); 5.33 (1H, d, J = 5.5, H-5); 7.17–7.27 (2H, m, H Ar); 7.35 (1H, ddd, J = 8.2, J = 7.1, J = 1.1, H Ar); 7.68 (1H, d, J = 9.0, H Ar); 7.74 (1H, d, J = 7.8, H Ar); 7.83 (1H, d, J = 8.5, H Ar); 8.08 (1H, s,  $C(CF_3)OH$ ; 8.89 (1H, d, J = 5.5, NH); 9.55 (1H, s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz) (mixture of tautomers A and B): 19.1 (CH<sub>2</sub>); 23.0 (CH<sub>2</sub>); 28.1 (3CH Ad, A); 28.6 (3CH Ad, B); 33.1 (C Ad, A); 34.4 (C Ad, B); 36.5 (3CH<sub>2</sub>) Ad, A); 37.0 (3CH<sub>2</sub> Ad, B); 42.0 (3CH<sub>2</sub> Ad, A); 42.6  $(3CH_2 \text{ Ad}, \mathbf{B}); 81.8 \text{ (C-3, B)}; 84.5 (q, {}^2J_{CF} = 32.4, \text{ C-7, B)}; 89.3 (C-4, \mathbf{A}); 101.5; 107.6; 116.2; 117.1; 118.0 (CH);$ 118.6 (CH); 118.7 (q,  ${}^{1}J_{CF} = 291.8$ , CF<sub>3</sub>); 122.9 (CH); 123.5 (CH); 123.9 (CH); 124.0 (CH); 125.0 (q,  ${}^{1}J_{CF} = 292.7, CF_{3}$ ; 125.0 (CH); 126.8 (CH); 127.0 (CH); 128.5 (CH); 128.7 (2C); 128.9 (2CH); 129.3; 133.9; 134.0; 140.5; 147.3 (q,  ${}^{4}J_{CF} = 4.8$ , CH); 148.3; 151.3; 153.6; 155.4; 162.1; 175.7 (q,  ${}^{2}J_{CF} = 30.5$ , C=O, A). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: tautomer A (41%): -65.3 (s, CF<sub>3</sub>); tautomer **B** (59%): -76.2 (s, CF<sub>3</sub>). Found, %: C 67.92; H 5.65; N 8.36. C<sub>28</sub>H<sub>28</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, %: C 67.87; H 5.70; N 8.48.

2-(3-Hydroxyadamantan-1-yl)-6-[(2-hydroxynaphthalen-1-yl)methyl]-7-(trifluoromethyl)-4,7-dihydropyrazolo-[1,5-a]pyrimidin-7-ol (3b) was obtained by method I of the synthesis of compound 3a from chromene 1a (200 mg, 0.72 mmol) and pyrazole 2b (168 mg, 0.72 mmol). Yield 255 mg (69%, method I), colorless crystals, mp 201–202°C. IR spectrum, v, cm<sup>-1</sup>: 3300–3050 (OH, NH), 2924, 2850 (CH Ad), 1659, 1597, 1566, 1489, 1462, 1415, 1333, 1311, 1230, 1196, 1138, 1087, 1026, 995, 960, 925, 903, 864, 814, 783, 744. <sup>1</sup>H NMR spectrum, δ, ppm (J, Hz): tautomer A: 1.50–1.70 (12H, m, 5CH<sub>2</sub> Ad, 2CH Ad); 2.13 (2H, br. s, CH<sub>2</sub> Ad); 4.04 (2H, s, CH<sub>2</sub>); 4.54 (1H, s, AdOH); 5.78 (1H, d, J = 1.8, H-4); 7.17–7.27 (2H, m, H Ar); 7.38 (1H, ddd, J = 8.2, J = 6.9, J = 1.1, H Ar); 7.68 (1H, d, J = 8.9, H Ar); 7.74 (1H, d, J = 7.8, H Ar); 7.97 (1H, d, J = 8.5, H Ar); 8.12 (1H, d, J = 13.5, =C<u>H</u>NH); 10.09 (1H, d, J = 13.5, =CHN<u>H</u>); 11.12 (1H, br. s, OH); 12.22 (1H, d, J = 1.8, NH); tautomer **B**: 1.50–1.70 (12H, m, 5CH<sub>2</sub> Ad, 2CH Ad); 2.13 (2H, br. s, CH<sub>2</sub> Ad); 3.84 (2H, s, CH<sub>2</sub>); 4.40 (1H, s, Ad–OH); 5.29 (1H, s, H-3); 5.33 (1H, d, J = 5.5, H-5); 7.18–7.28 (2H, m, H Ar); 7.36 (1H, ddd, J = 8.2, J = 6.9, J = 1.1, H Ar); 7.68 (1H, d, J = 8.9, H Ar); 7.74 (1H, d, J = 7.8, H Ar); 7.83 (1H, d, J = 8.5, H Ar); 8.10 (1H, s, C(CF<sub>3</sub>)OH); 8.91 (1H, d, J = 5.5, NH); 9.56 (1H, s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (J, Hz) (mixture of tautomers A and B): 19.1 (CH<sub>2</sub>); 23.0 (CH<sub>2</sub>); 30.4 (2CH Ad); 30.7 (2CH Ad); 35.3 (CH<sub>2</sub> Ad); 35.8 (CH<sub>2</sub> Ad); 36.3 (C Ad); 37.7 (C Ad); 41.1 (2CH<sub>2</sub> Ad); 41.7 (2CH<sub>2</sub> Ad); 44.8 (2CH<sub>2</sub> Ad); 45.1 (2CH<sub>2</sub> Ad); 49.9 (CH<sub>2</sub> Ad); 50.5 (CH<sub>2</sub> Ad); 66.9 (C–OH); 67.3 (C–OH); 81.9 (C-3, **B**); 84.5 (q,  ${}^{2}J_{CF} = 32.4$ , C-7, **B**); 89.4 (C-4, **A**); 101.5; 107.6; 116.2; 117.1; 117.9 (CH); 118.6 (CH); 118.7  $(q, {}^{1}J_{CF} = 291.8, CF_{3}); 122.9 (CH); 123.5 (CH); 123.9$ (CH); 124.0 (CH); 124.6 (q,  ${}^{1}J_{CF} = 292.7$ , CF<sub>3</sub>); 125.0 (CH); 126.8 (CH); 127.0 (CH); 128.5 (CH); 128.7 (2C); 128.9 (2CH); 129.3; 133.9; 134.0; 140.5; 147.2 (q,  ${}^{4}J_{CF} = 4.8$ ,

CH); 148.2; 151.3; 153.6; 154.5; 161.2; 175.7 (q,  ${}^{2}J_{CF} = 30.5$ , C=O, A). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: tautomer A (38%): -65.3 (s, CF<sub>3</sub>); tautomer B (62%): -76.2 (s, CF<sub>3</sub>). Found, %: C 65.71; H 5.45; N 8.16. C<sub>28</sub>H<sub>28</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 65.74; H 5.52; N 8.21.

2-(tert-Butyl)-6-[(2-hydroxynaphthalen-1-yl)methyl]-7-(trifluoromethyl)-4,7-dihydropyrazolo[1,5-a]pyrimidin-7-ol (3c) was obtained by method I of the synthesis of compound 3a from chromene 1a (200 mg, 0.72 mmol) and pyrazole 2c (100 mg, 0.72 mmol). Yield 240 mg (80%, method I), colorless crystals, mp 216-218°C. IR spectrum, v, cm<sup>-1</sup>: 3219, 3148, 3119 (OH, NH), 2965, 2870, 1657, 1611, 1597, 1566, 1495, 1462, 1416, 1327, 1283, 1231, 1192, 1136, 1024, 995, 816, 789, 756, 745. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): tautomer A: 1.21 (9H, s, (CH<sub>3</sub>) C); 4.05 (2H, s, CH<sub>2</sub>); 5.80 (1H, d, J = 1.8, H-4); 7.18–7.27 (2H, m, H Ar); 7.38 (1H, ddd, J = 8.2, J = 7.1, J = 1.1, H Ar); 7.68 (1H, d, J = 8.7, H Ar); 7.75 (1H, d, J = 8.0, H Ar); 7.98 (1H, d, J = 8.5, H Ar); 8.13 (1H, d, J = 13.5, =CHNH); 10.10 (1H, d, J = 13.5, =CHNH); 11.14 (1H, br. s, OH); 12.26 (1H, d, J = 1.8, NH); tautomer **B**: 1.20 (9H, s, (CH<sub>3</sub>)C); 3.85 (2H, s, CH<sub>2</sub>); 5.34–5.36 (2H, m, H-3,5); 7.18– 7.27 (2H, m, H Ar); 7.36 (1H, ddd, J = 8.2, J = 7.1, J = 1.1, H Ar); 7.68 (1H, d, J = 8.7, H Ar); 7.75 (1H, d, J = 8.0, H Ar); 7.83 (1H, d, J = 8.5, H Ar); 8.12 (1H, s, C(CF<sub>3</sub>)OH); 8.91 (1H, d, J = 5.3, NH); 9.58 (1H, s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (J, Hz) (mixture of tautomers **A** and **B**): 19.0 (CH<sub>2</sub>); 23.0 (CH<sub>2</sub>); 30.2 (3CH<sub>3</sub>, **A**); 30.8 (3CH<sub>3</sub>, **B**); 82.4 (C-3, **B**); 84.5 (q,  ${}^{2}J_{CF} = 32.4$ , C-7, **B**); 89.9 (C-4, **A**); 101.5; 107.6; 116.2; 117.1; 117.9 (CH); 118.6 (CH); 118.7  $(q, {}^{1}J_{CF} = 291.8, CF_{3}); 122.8 (CH); 123.5 (CH); 123.9$ (CH); 124.0 (CH); 124.6 (q,  ${}^{1}J_{CF} = 292.7$ , CF<sub>3</sub>); 125.0 (CH); 126.7 (CH); 126.9 (CH); 128.5 (CH); 128.6 (CH); 128.7; 128.9 (2CH); 129.3; 133.9; 134.0; 140.6; 147.3 (q,  ${}^{4}J_{CF} = 4.8$ , CH); 148.2; 151.2; 153.5; 155.1; 161.8; 175.6 (q,  ${}^{2}J_{CF} = 30.5$ , C=O, A).  ${}^{19}F$  NMR spectrum,  $\delta$ , ppm: tautomer A (36%): -65.3 (s, CF<sub>3</sub>); tautomer (B) (64%): -76.2(s, CF<sub>3</sub>). Found, %: C 63.41; H 5.27; N 9.93. C<sub>22</sub>H<sub>22</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, %: C 63.30; H 5.31; N 10.07.

1-{[2-(Adamantan-1-yl)-7-(trifluoromethyl)pyrazolo-[1,5-*a*]pyrimidin-6-yl]methyl}naphthalen-2-ol (4a). Method IV. BF<sub>3</sub>·OEt<sub>2</sub> (20  $\mu$ l, 20 mol %) was added to a mixture of chromene 1a (200 mg, 0.72 mmol), pyrazole 2a (156 mg, 0.72 mmol), and *i*-PrOH (5 ml), and the mixture was heated under reflux for 8 h. The solvent was evaporated under reduced pressure, the residue was suspended at reflux in MeOH (2 ml), then kept at -30°C for 2 h. The product was filtered off, washed with chilled (-30°C) MeOH (2 ml), and recrystallized from EtOH.

Method V. A mixture of chromene **1a** (200 mg, 0.72 mmol), pyrazole **2a** (156 mg, 0.72 mmol), and AcOH (5 ml) was heated under reflux for 10 h. The formed precipitate was filtered off, washed with AcOH (2 ml), and recrystallized from EtOH.

Method VI. BF<sub>3</sub>·OEt<sub>2</sub> (20  $\mu$ l, 80 mol %) was added to a solution of compound **3a** (99 mg, 0.20 mmol) in *i*-PrOH (3 ml), and the resulting mixture was heated under reflux for 6 h. The solvent was evaporated under reduced pressure, the residue was recrystallized from EtOH.

Method VII. A crystal of p-TsOH was added to a solution of compound **3a** (99 mg, 0.20 mmol) in MeOH (6 ml), and the resulting mixture was heated under reflux for 2 h. The solvent was evaporated under reduced pressure, and the residue was recrystallized from EtOH.

Yield 235 mg (68%, method IV), 225 mg (65%, method V), 75 mg (80%, method VI), 78 mg (82%, method VII), lightyellow crystals, mp 246–247°C. IR spectrum, v, cm<sup>-1</sup>: 3209, 3070 (OH), 2904, 2850 (CH Ad), 1631, 1604, 1546, 1516, 1440, 1400, 1357, 1296, 1271, 1249, 1192, 1157, 1072, 1022, 987, 813, 786, 740. <sup>1</sup>H NMR spectrum, δ, ppm (J, Hz): 1.70 (6H, br. s, 3CH<sub>2</sub> Ad); 1.93 (6H, br. s, 3CH<sub>2</sub> Ad); 1.99 (3H, br. s, 3CH Ad); 4.57 (2H, s, CH<sub>2</sub>); 6.66 (1H, s, H-3); 7.20 (1H, d, J = 8.9, H Ar); 7.28 (1H, t, J = 7.6, H Ar); 7.42 (1H, ddd, J = 8.2, J = 6.9, J = 1.1, H Ar); 7.76 (1H, d, J = 8.7, H Ar); 7.79–7.82 (2H, m, H-5, H Ar); 7.86 (1H, d, J = 8.5, H Ar); 10.00 (1H, s, OH). <sup>13</sup>C NMR spectrum, δ, ppm (J, Hz): 23.8 (CH<sub>2</sub>); 28.4 (3CH Ad); 34.8 (C Ad); 36.7 (3CH<sub>2</sub> Ad); 42.4 (3CH<sub>2</sub> Ad); 93.5 (C-3); 114.8; 118.6 (CH); 120.7; 121.7 (q,  ${}^{1}J_{CF} = 275.6$ , CF<sub>3</sub>); 122.7 (CH); 123.2 (CH); 127.5 (CH); 128.8; 129.2 (CH); 129.7 (q,  ${}^{2}J_{CF} = 34.3$ , C-7); 129.7 (CH); 133.7; 148.2; 151.2 (C-5); 154.1; 167.4.  ${}^{19}F$  NMR spectrum,  $\delta$ , ppm: -60.4 (s, CF<sub>3</sub>). Found, %: C 70.50; H 5.41; N 8.77. C<sub>28</sub>H<sub>26</sub>F<sub>3</sub>N<sub>3</sub>O. Calculated, %: C 70.43; H 5.49; N 8.80.

2-{[2-(Adamantan-1-yl)-7-(trifluoromethyl)pyrazolo-[1,5-*a*]pvrimidin-6-vl]methyl}-4,5-dimethylphenol (4b) was obtained by method IV of the synthesis of compound 4a from chromene 1b (184 mg, 0.72 mmol) and pyrazole 2a (156 mg, 0.72 mmol). Yield 270 mg (83%, method IV), light-yellow crystals, mp 179-180°C (MeOH). IR spectrum, v, cm<sup>-1</sup>: 3206 (OH), 2908, 2850 (CH Ad), 1601, 1547, 1516, 1462, 1431, 1415, 1354, 1292, 1250, 1188, 1157, 1080, 945, 887, 783. <sup>1</sup>H NMR spectrum, δ, ppm: 1.71 (6H, br. s, 3CH<sub>2</sub> Ad); 1.94–2.05 (15H, m, 3CH<sub>2</sub> Ad, 3CH Ad, 2CH<sub>3</sub>); 4.04 (2H, s, CH<sub>2</sub>); 6.56 (1H, s, H Ar); 6.64 (1H, s, H Ar); 6.72 (1H, s, H-3); 8.40 (1H, s, H-5); 9.23 (1H, s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (*J*, Hz): 18.9 (CH<sub>3</sub>); 19.7 (CH<sub>3</sub>); 28.4 (3CH Ad); 28.9 (CH<sub>2</sub>); 34.8 (C Ad); 36.7 (3CH<sub>2</sub> Ad); 42.4 (3CH<sub>2</sub> Ad); 93.6 (C-3); 116.9 (CH); 120.6; 121.3 (q,  ${}^{1}J_{CF} = 275.6$ , CF<sub>3</sub>); 122.5; 126.8; 129.9 (q,  ${}^{2}J_{CF} = 34.3, C-7$ ; 130.8 (CH); 136.0; 148.6; 153.2; 153.6 (C-5); 167.3. <sup>19</sup>F NMR spectrum, δ, ppm: -61.5 (s, CF<sub>3</sub>). Found, %: C 68.47; H 6.21; N 9.15. C<sub>26</sub>H<sub>28</sub>F<sub>3</sub>N<sub>3</sub>O. Calculated, %: C 68.56; H 6.20; N 9.22.

**6-[(2-Hydroxyphenyl)(phenyl)methyl]-3-{7-(trifluoromethyl)pyrazolo[1,5-***a***]pyrimidin-2-yl}adamantan-1-ol (4c) was obtained by method IV of the synthesis of compound 4a from chromene 1c (219 mg, 0.72 mmol) and pyrazole 2b (168 mg, 0.72 mmol). Yield 275 mg (74%, method IV), light-yellow crystals, mp 229–231°C. <sup>1</sup>H NMR spectrum, δ, ppm (***J***, Hz): 1.52–1.66 (6H, m, 3CH<sub>2</sub> Ad); 1.77–1.84 (6H, m, 2CH<sub>2</sub> Ad, 2CH Ad); 2.17 (2H, br. s, CH<sub>2</sub> Ad); 4.51 (1H, s, Ad–O<u>H</u>); 6.22 (1H, s, C<u>H</u>Ph); 6.67 (1H, dd,** *J* **= 7.6,** *J* **= 1.4, H Ar); 6.71–6.75 (1H, m, H Ar); 6.76 (1H, s, H-3); 6.79 (1H, d,** *J* **= 8.0, H Ar); 7.07–7.12 (3H, m, H Ar); 7.25 (1H, t,** *J* **= 7.3, H Ar); 7.30–7.34 (2H, m, H Ar); 8.07 (1H, s, H-5); 9.67 (1H, s, OH). <sup>13</sup>C NMR spectrum, δ, ppm (***J***, Hz): 30.6 (2CH Ad); 35.5 (CH<sub>2</sub> Ad);**  38.2 (C Ad); 41.5 (2CH<sub>2</sub> Ad); 42.9 (q,  ${}^{4}J_{CF}$  = 3.8, <u>C</u>HPh); 44.9 (2CH<sub>2</sub> Ad); 50.1 (CH<sub>2</sub> Ad); 67.1 (COH); 93.9 (C-3); 115.8 (CH); 119.6 (CH); 121.2 (q,  ${}^{1}J_{CF}$  = 275.6, CF<sub>3</sub>); 123.0; 127.4 (CH); 128.7; 129.0 (CH); 129.3 (2CH); 129.5 (2CH); 130.0 (CH); 130.3 (q,  ${}^{2}J_{CF}$  = 34.3, C-7); 141.5; 148.5; 151.3 (C-5); 155.0; 167.0.  ${}^{19}$ F NMR spectrum, δ, ppm: -60.8 (s, CF<sub>3</sub>). Found, %: C 69.42; H 5.40; N 7.98. C<sub>30</sub>H<sub>28</sub>F<sub>3</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, %: C 69.35; H 5.43; N 8.09.

Ethyl 6-[(2-hydroxynaphthalen-1-yl)methyl]pyrazolo-[1,5-a]pyrimidine-3-carboxylate (4d). 1H-Benzo[f]chromene-2-carbaldehyde (1d) (210 mg, 1.0 mmol) was added to a solution of pyrazole 2d (155 mg, 1.0 mmol) in AcOH (2 ml). The mixture was heated under reflux with vigorous stirring for 15 min and cooled to room temperature. The formed precipitate was filtered off and recrystallized from EtOH-DMF, 4:1 mixture. Yield 255 mg (73%), lightyellow crystals, mp 271–273°C. IR spectrum, v, cm<sup>-1</sup>: 3246, 3055 (OH), 1720 (C=O), 1627, 1516, 1056, 808. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 1.23 (3H, t, *J* = 7.1, CH<sub>3</sub>); 4.21 (2H, q, J = 7.1, CH<sub>2</sub>O); 4.42 (2H, s, CH<sub>2</sub>); 7.24– 7.28 (2H, m, H Ar); 7.42 (1H, ddd, J = 8.5, J = 6.9, J = 1.4, H Ar); 7.73 (1H, d, J = 8.7, H Ar); 7.78 (1H, d, J = 7.6, H Ar); 8.04 (1H, d, J = 8.9, H Ar); 8.49 (1H, s, H-2); 8.77 (1H, d, J = 2.1) and 8.94 (1H, d, J = 2.1, H-5,7); 10.09 (1Hbr. s, OH). <sup>13</sup>C NMR spectrum, δ, ppm: 14.9 (CH<sub>3</sub>); 24.6 (CH<sub>2</sub>); 60.0 (CH<sub>2</sub>O); 102.1; 116.4; 118.6 (CH); 123.0 (CH); 123.2 (CH); 124.8; 127.3 (CH); 128.8; 129.1 (CH); 129.3 (CH); 133.3; 134.8 (C-7); 146.1; 147.5 (C-2); 153.3; 155.3 (C-5); 162.1 (C=O). Found, %: C 69.22; H 4.88; N 12.01. C<sub>20</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 69.15; H 4.93; N 12.10.

Ethyl 6-[(6-bromo-2-hydroxynaphthalen-1-yl)methyl]pyrazolo[1,5-a]pyrimidine-3-carboxylate (4e) was obtained by the method of the synthesis of compound 4d from pyrazole 2d (155 mg, 1.0 mmol) and 8-bromo-1H-benzo[f]chromene-2-carbaldehyde (1e) (289 mg, 1.0 mmol). Yield 335 mg (79%), light-yellow crystals, mp 300-302°C. IR spectrum, v, cm<sup>-1</sup>: 3240, 3051 (OH), 1708 (C=O), 1627, 1274, 1058. <sup>1</sup>H NMR spectrum, δ, ppm (*J*, Hz): 1.24 (3H, t, J = 7.1, CH<sub>3</sub>); 4.22 (2H, q, J = 7.1, CH<sub>2</sub>O); 4.41 (2H, s,  $CH_2$ ; 7.29 (1H, d, J = 8.9, H-3 naphthalene); 7.51 (1H, dd, J = 9.1, J = 2.3, H-7 naphthalene); 7.73 (1H, d, J = 8.9, H-4naphthalene); 8.01 (1H, J = 9.1, H-8 naphthalene); 8.05 (1H, d, J = 2.3, H-5 naphthalene); 8.50 (1H, s, H-2); 8.74(1H, d, J = 2.1) and 8.93 (1H, d, J = 2.1, H-5,7); 10.27 (1H, br. s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm: 14.9 (CH<sub>3</sub>); 24.6 (CH<sub>2</sub>); 60.0 (CH<sub>2</sub>O); 102.1; 116.1; 116.8; 119.9 (CH); 124.5; 125.6 (CH); 128.6 (CH); 130.0 (CH); 130.1; 130.8 (CH); 131.9; 134.8 (C-7); 146.2; 147.6 (C-2); 153.9; 155.2 (C-5); 162.1 (C=O). Found, %: C 56.30; H 3.83; N 9.75. C<sub>20</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>3</sub>. Calculated, %: C 56.35; H 3.78; N 9.86.

Ethyl 6-[(2-hydroxynaphthalen-1-yl)methyl]-7-phenylpyrazolo[1,5-*a*]pyrimidine-3-carboxylate (4f) was obtained by the method of the synthesis of compound 4d from pyrazole 2d (155 mg, 1.0 mmol) and (1*H*-benzo[*f*]chromen-2-yl)(phenyl)methanone (1f) (286 mg, 1.0 mmol). Reaction time 15 h. Yield 245 mg (58%), colorless crystals, mp 270–272°C. IR spectrum, v, cm<sup>-1</sup>: 3182, 3057 (OH), 1699 (C=O), 1529, 1246, 1056, 810. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 1.30 (3H, t, *J* = 7.1, CH<sub>3</sub>); 4.37 (2H, q, J = 7.1, CH<sub>2</sub>O); 4.41 (2H, s, CH<sub>2</sub>); 7.09–7.13 (2H, m, H Ar); 7.19–7.23 (2H, m, H Ar); 7.35 (1H, d, J = 8.7, H Ar); 7.62–7.66 (1H, m, H Ar); 7.70–7.82 (5H, m, H Ar); 8.55 (1H, s, H-2); 8.75 (1H, s, H-5); 9.63 (1H, s, OH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm: 14.6 (CH<sub>3</sub>); 24.2 (CH<sub>2</sub>); 60.6 (CH<sub>2</sub>O); 102.8; 116.7; 118.2 (CH); 122.2 (CH); 122.4; 122.8 (CH); 126.5 (CH); 128.9 (2C); 129.1 (2C); 129.2 (2CH); 130.2 (2CH); 131.3 (CH); 132.9; 146.0; 146.4; 147.3 (C-2); 152.6; 155.4 (C-5); 162.9 (C=O). Found, %: C 73.79; H 4.95; N 9.83. C<sub>26</sub>H<sub>21</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 73.74; H 5.00; N 9.92.

Ethyl 13a-(trifluoromethyl)-4,13a-dihydro-6H-benzo-[5,6]chromeno[3,2-e]pyrazolo[1,5-a]pyrimidine-3-carboxylate (4g) was obtained by the method of the synthesis of compound 4d from pyrazole 2d (155 mg, 1.0 mmol) and chromene 1a (278 mg, 1.0 mmol). Reaction time 15 h. Yield 295 mg (71%), light-yellow crystals, mp 249-250°C (EtOH). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): tautomer A: 1.21 (3H, t, J = 7.1, CH<sub>3</sub>); 4.20 (2H, q, J = 7.1, CH<sub>2</sub>O); 4.64 (2H, s, CH<sub>2</sub>); 7.17 (1H, d, J = 8.9, H Ar); 7.27–7.31 (1H, m, H Ar); 7.42–7.46 (1H, m, H Ar); 7.78 (1H, d, J = 8.5, H Ar); 7.82 (1H, d, J = 8.7, H Ar); 7.89 (1H, d, J = 8.5, H Ar); 8.19 (1H, s, H Ar); 8.68 (1H, s, H Ar); 10.05 (1H, s, OH); tautomer **B**: 1.25 (3H, t, J = 7.1, CH<sub>3</sub>); 3.76 (1H, d, J = 18.5, 6-CH<sub>2</sub>); 4.21 (2H, q, J = 7.1, CH<sub>2</sub>); 4.27 (1H, d, J = 18.5, 6-CH<sub>2</sub>); 7.16 (1H, d, J = 8.9, H-12); 7.22 (1H, d, J = 5.0, H-5); 7.40–7.44 (1H, m) and 7.54– 7.58 (1H, m, H-8,9); 7.77 (1H, d, J = 8.9, H Ar); 7.85 (1H, d, J = 8.2, H Ar); 7.89 (1H, d, J = 8.5, H Ar); 7.97 (1H, s, H-2); 10.19 (1H, d, J = 5.0, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (J, Hz): tautomer A: 14.8 (CH<sub>3</sub>); 24.1 (CH<sub>2</sub>); 60.4 (CH<sub>2</sub>O); 103.4; 114.3; 118.6 (CH); 122.7 (CH); 123.3 (CH); 123.8; 127.6 (CH); 128.8; 129.2 (CH); 130.0 (CH); 131.0 (q,  ${}^{2}J_{CF} = 34.5$ , <u>CCF\_3</u>); 133.7; 146.9; 147.4 (C-2); 154.2; 155.9 (C-5); 161.8 (C=O) (the quartet signal of the carbon atom of the CF<sub>3</sub> group could not be recorded due to its low intensity); tautomer B: 15.0 (CH<sub>3</sub>); 24.0 (C-6); 60.2 (CH<sub>2</sub>O); 84.6 (q,  ${}^{2}J_{CF} = 34.3$ , C-13a); 94.2; 94.8; 114.2; 118.3 (CH); 122.9 (CH); 123.4 (q,  ${}^{1}J_{CF} = 292.7$ , CF<sub>3</sub>); 125.0 (CH); 127.2 (CH); 127.5 (CH); 129.0 (CH); 129.1 (CH); 129.7; 131.7; 142.0; 142.5 (C-2); 148.7; 162.6 (C=O). <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: tautomer A (22%): -60.0 (s, CF<sub>3</sub>); tautomer **B** (78%): -75.5 (s, CF<sub>3</sub>). Found, %: C 60.65; H 3.94; N 10.01. C<sub>21</sub>H<sub>16</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 60.72; H 3.88; N 10.12.

**Ethyl** 9-(adamantan-1-yl)-13a-(trifluoromethyl)-4,13a-dihydro-6*H*-benzo[5,6]chromeno[3,2-*e*]pyrazolo-[1,5-*a*]pyrimidine-3-carboxylate (4h) was obtained by the method of the synthesis of compound 4d from pyrazole 2d (155 mg, 1.0 mmol) and 1-[8-(adamantan-1-yl)-1*H*-benzo-[*f*]chromen-2-yl]-2,2,2-trifluoroethan-1-one (1g) (412 mg, 1.0 mmol). Reaction time 15 h. Yield 410 mg (75%), colorless crystals, mp 289–291°C (DMF). <sup>1</sup>H NMR spectrum, δ, ppm (*J*, Hz): tautomer **A**: 1.21 (3H, t, *J* = 7.1, CH<sub>3</sub>); 1.71 (6H, br. s, 3CH<sub>2</sub> Ad); 1.89 (6H, br. s, 3CH<sub>2</sub> Ad); 2.03 (3H, br. s, 3CH Ad); 4.20 (2H, q, *J* = 7.1, CH<sub>2</sub>O); 4.61 (2H, s, CH<sub>2</sub>); 7.16 (1H, d, *J* = 8.9, H Ar); 7.51 (1H, d, *J* = 8.9, H Ar); 7.67 (1H, s, H Ar); 7.75 (1H, d, *J* = 8.7, H Ar); 7.80 (1H, d, *J* = 8.9, H Ar); 8.20 (1H, s,

H Ar); 8.68 (1H, s, H Ar); 9.92 (1H, s, OH); tautomer **B**: 1.25 (3H, t, J = 7.1, CH<sub>3</sub>); 1.71 (6H, br. s, 3CH<sub>2</sub> Ad); 1.89 (6H, br. s, 3CH<sub>2</sub> Ad); 2.03 (3H, br. s, 3CH Ad); 3.73 (1H, d, J = 18.8, 6-CH<sub>2</sub>); 4.20–4.25 (3H, m, CH<sub>2</sub>O, 6-CH<sub>2</sub>); 7.11 (1H, d, J = 8.9, H-12); 7.21 (1H, d, J = 4.8, H-5); 7.63 (1H, d, J = 4.8, H-5); 7.63d, J = 8.9, H Ar); 7.70 (1H, s, H-10); 7.74 (1H, d, J = 8.7, H Ar); 7.82 (1H, d, J = 8.9, H Ar); 7.96 (1H, s, H-2); 10.17 (1H, d, J = 4.8, NH). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm (J, Hz): tautomer A: 14.8 (CH<sub>3</sub>); 24.0 (CH<sub>2</sub>); 28.8 (3CH Ad); 36.0 (C Ad); 36.7 (3CH<sub>2</sub> Ad); 43.0 (3CH<sub>2</sub> Ad); 60.4 (CH<sub>2</sub>O); 103.4; 114.0; 118.4 (CH); 122.5 (CH); 123.9; 124.1 (CH); 128.8; 129.1 (CH); 130.0 (CH); 131.0 (q,  ${}^{2}J_{CF} = 34.5$ , <u>C</u>CF<sub>3</sub>); 131.9; 145.6; 146.9; 147.4 (C-2); 153.6; 156.0 (C-5); 161.8 (C=O) (the quartet signal of the carbon atom of the CF<sub>3</sub> group could not be recorded due to its low intensity); tautomer B: 15.0 (CH<sub>3</sub>); 24.0 (C-6); 28.8 (3CH Ad); 36.2 (C Ad); 36.7 (3CH<sub>2</sub> Ad); 43.0 (3CH<sub>2</sub> Ad); 60.1 (CH<sub>2</sub>O); 84.6 (q,  ${}^{2}J_{CF} = 34.3$ , C-13a); 94.4; 94.8; 113.9; 118.0 (CH); 122.7 (CH); 123.4 (q,  ${}^{1}J_{CF} = 293.7$ , CF<sub>3</sub>); 123.8 (CH); 125.5 (CH); 127.0 (CH); 129.1 (CH); 129.8; 129.9; 142.0; 142.4 (C-2); 147.4; 148.2; 162.6 (C=O). <sup>19</sup>F NMR spectrum, δ, ppm: tautomer A (19%): -60.0 (s, CF<sub>3</sub>); tautomer **B** (81%): -75.5 (s, CF<sub>3</sub>). Found, %: C 67.67; H 5.53; N 7.55. C<sub>31</sub>H<sub>30</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 67.75; H 5.50; N 7.65.

1-{[3-(Adamantan-1-yl)-1-methyl-4-(trifluoromethyl)-1H-pyrazolo[3,4-b]pyridin-5-yl]methyl}naphthalen-2-ol (5). A crystal of p-TsOH was added to a solution of pyrazole 2e (169 mg, 0.73 mmol) and chromene 1a (200 mg, 0.72 mmol) in DMF (5 ml). The mixture was heated under reflux for 12 h, cooled, and poured into H<sub>2</sub>O (10 ml) with stirring. The formed precipitate was filtered off and recrystallized from MeOH. Yield 120 mg (34%), lightyellow crystals, mp 117–118°C. IR spectrum, v, cm<sup>-1</sup>: 3500– 3150 (OH), 2904, 2850 (CH Ad), 1628, 1566, 1512, 1442, 1404, 1361, 1327, 1215, 1179, 1126, 1087, 1011, 987, 904, 810, 744, 705. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (J, Hz): 1.74 (6H, br. s, 3CH<sub>2</sub> Ad); 2.06 (3H, br. s, 3CH Ad); 2.16 (6H, br. s, 3CH<sub>2</sub> Ad); 3.88 (3H, s, CH<sub>3</sub>); 4.59 (2H, s, CH<sub>2</sub>); 7.21-7.27 (2H, m, H Ar); 7.33 (1H, ddd, J = 8.0, J = 6.9, J = 1.2, H Ar); 7.55 (1H, d, J = 8.5, H Ar); 7.76–7.83 (3H, m, H Ar); 10.01 (1H, s, OH). <sup>13</sup>C NMR spectrum (100°C), δ, ppm (J, Hz): 26.3 (CH<sub>2</sub>); 29.0 (3CH Ad); 34.2 (CH<sub>3</sub>); 37.0 (3CH<sub>2</sub> Ad); 38.5 (C Ad); 41.7 (3CH<sub>2</sub> Ad); 107.5; 116.2; 118.9 (CH); 122.4 (CH); 123.1 (CH); 124.3 (q,  ${}^{1}J_{CF} = 274.6$ , CF<sub>3</sub>); 127.3 (CH); 128.5; 129.1 (CH, C); 129.4 (CH); 129.6 (q,  ${}^{2}J_{CF} = 29.6$ , C-4); 133.9; 149.8 (C-6); 150.0; 151.3; 154.0. <sup>19</sup>F NMR spectrum,  $\delta$ , ppm: -52.4 (s, CF<sub>3</sub>). Found, %: C 70.78; H 5.70; N 8.63. C<sub>29</sub>H<sub>28</sub>F<sub>3</sub>N<sub>3</sub>O. Calculated, %: C 70.86; H 5.74; N 8.55.

X-ray structural analysis of compound 4g was performed on an Agilent SuperNova diffractometer using a microfocus X-ray source (CuK $\alpha$  radiation,  $\lambda$  1.54060 Å) and an Atlas S2 coordinate CCD detector. Crystals suitable for structural analysis were obtained by slow evaporation at room temperature of an EtOH solution of compound 4g. The collection of reflections, solving and refinement of the unit cell parameters was carried out using the CrysAlisPro specialized software package.<sup>18</sup> The structure was solved using the SHELXT program<sup>19</sup> and solved employing the SHELXL program,<sup>20</sup> molecular graphics were rendered and preparation of the material for publication was performed using the OLEX2 software package.<sup>21</sup> The full set of X-ray structural data for compound **4g** was deposited at the Cambridge Crystallographic Data Center (deposit CCDC 2038689).

Supplementary information file containing the synthetic procedures and spectral-analytical characteristics of the starting compounds for the preparation of adamantyl-substituted pyrazoles **2a**,**b**,**e** and the principal crystallographic data and parameters of the X-ray structural analysis of compound **4g** is available at the journal website at http://link.springer.com/journal/10593.

This work was supported by the Russian Foundation for Basic Research (grant 18-33-20249).

### References

- Cherukupalli, S.; Karpoormath, R.; Chandrasekaran, B.; Hampannavar, G. A.; Thapliyal, N.; Palakollu, V. N. *Eur. J. Med. Chem.* 2017, *126*, 298.
- (a) Salem, M. A.; Helal, M. H.; Gouda, M. A.; El-Gawad, H. H. A.; Shehab, M. A. M.; El-Khalafawy, A. Synth. Commun. 2019, 49, 1750. (b) Al-Azmi, A. Curr. Org. Chem. 2019, 23, 721.
- (a) Mirza, N. R.; Rodgers, R. J.; Mathiasen, L. S. J. *Pharmacol. Exp. Ther.* **2006**, *316*, 1291. (b) Lippa, A.; Czobor, P.; Stark, J.; Beer, B.; Kostakis, E.; Gravielle, M.; Bandyopadhyay, S.; Russek, S. J.; Gibbs, T. T.; Farb, D. H.; Skolnick, P. *Proc. Natl. Acad. Sci. U. S. A.* **2005**, *102*, 7380.
- Kato, N.; Oka, M.; Murase, T.; Yoshida, M.; Sakairi, M.; Yamashita, S.; Yasuda, Y.; Yoshikawa, A.; Hayashi, Y.; Makino, M.; Takeda, M.; Mirensha, Y.; Kakigami, T. *Bioorg. Med. Chem.* 2011, 19, 7221.
- (a) Emelina, E. E.; Petrov, A. A. Russ. J. Org. Chem. 2009, 45, 417. [Zh. Org. Khim. 2009, 45, 427.] (b) Krasovsky, A. L.; Moiseev, A. M.; Nenajdenko, V. G.; Balenkova, E. S. Synthesis 2002, 901.
- Filyakova, V. I.; Kuznetsova, O. A.; Ulomskii, E. N.; Rybalova, T. V.; Gatilov, Yu. V.; Kodess, M. I.; Rusinov, V. L.; Pashkevich, K. I. *Russ. Chem. Bull., Int. Ed.* 2002, *51*, 332. [*Izv. Akad. Nauk, Ser. Khim.* 2002, 313.]
- Shaabani, A.; Nazeri, M. T.; Afshari, R. Mol. Diversity 2019, 23, 751.
- (a) Ryabukhin, S. V.; Plaskon, A. S.; Volochnyuk, D. M.; Pipko, S. E.; Tolmachev, A. A. *Heterocycles* 2008, *75*, 583.
  (b) Quiroga, J.; Mejía, D.; Insuasty, B.; Abonía, R.; Nogueras, M.; Sánchez, A.; Cobo, J.; Low, J. N. J. *Heterocycl. Chem.* 2002, *39*, 51.
  (c) Zimmerman, J. R.; Myers, B. J.; Bouhall, S.; McCarthy, A.; Johntony, O.; Manpadi, M. *Tetrahedron Lett.* 2014, *55*, 936.
  (d) Santos, C. M. M.; Silva, V. L. M.; Silva, A. M. S. *Molecules* 2017, *22*, 1665/1.
- Osipov, D. V.; Osyanin, V. A.; Klimochkin, Yu. N. In Targets in Heterocyclic Systems: Chemistry and Properties; Attanasi, O. A.; Merino, P.; Spinelli, D., Eds.; Società Chimica Italiana: Rome, 2018, Vol. 22, p. 436.
- Krasovsky, A. L.; Hartulyari, A. S.; Nenajdenko, V. G.; Balenkova, E. S. Synthesis 2002, 133.
- Goryaeva, M. V.; Burgart, Y. V.; Saloutin, V. I.; Sadchikova, E. V.; Ulomskii, E. N. *Heterocycles* 2009, 78, 435.
- (a) Emelina, E. E.; Petrov, A. A.; Selivanov, S. I.; Filyukov, D. V. Russ. J. Org. Chem. 2008, 44, 251. [Zh. Org. Khim. 2008, 44, 259.] (b) Emelina, E. E.; Petrov, A. A.; Firsov, A. V. Russ. J. Org. Chem. 2001, 37, 852. [Zh. Org. Khim. 2001, 37, 899.]

(c) Emelina, E. E.; Petrov, A. A.; Selivanov, S. I.; Nelyubina, Y. V.; Antipin, M. Yu. J. Fluorine Chem. **2009**, *130*, 861.

- Bharathi, Ch.; Prabahar, K. J.; Prasad, Ch. S.; Kumar, M. S.; Magesh, S.; Handa, V. K.; Dandala, R.; Naidu, A. J. Pharm. Biomed. Anal. 2007, 44, 101.
- (a) Volochnyuk, D. M.; Pushechnikov, A. O.; Krotko, D. G.; Sibgatulin, D. A.; Kovalyova, S. A.; Tolmachev, A. A. *Synthesis* 2003, 1531. (b) Aggarwal, R.; Kumar, V.; Bansal, A.; Sanz, D.; Claramunt, R. M. *J. Fluorine Chem.* 2012, *140*, 31.
- (a) Lukashenko, A. V.; Osyanin, V. A.; Osipov, D. V.; Klimochkin, Yu. N. J. Org. Chem. 2017, 82, 1517.
  (b) Lukashenko, A. V.; Osyanin, V. A.; Osipov, D. V.; Klimochkin, Yu. N. Chem. Heterocycl. Compd. 2016, 52, 711.
  [Khim. Geterotsikl. Soedin. 2016, 52, 711.] (c) Osyanin, V. A.;

Lukashenko, A. V.; Osipov, D. V. Russ. Chem. Rev. 2021, 90, 324. [Usp. Khim. 2021, 90, 324.]

- Magee, W. L.; Rao, C. B.; Glinka, J.; Hui, H.; Amick, T. J.; Fiscus, D.; Kakodkar, S.; Nair, M.; Shechter, H. J. Org. Chem. 1987, 52, 5538.
- Selvakumar, B.; Vaidyanathan, S. P.; Madhuri, S.; Elango, K. P. J. Chem. Res. 2017, 41, 221.
- 18. CrysAlisPro, version 1.171.38.41; Rigaku Oxford Diffraction, 2015.
- Sheldrick, G. M. Acta Crystallogr., Sect. A: Found. Adv. 2015, A71, 3.
- 20. Sheldrick, G. M. Acta Crystallogr., Sect. C: Struct. Chem. 2015, C71, 3.
- Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. J. Appl. Crystallogr. 2009, 42, 339.