

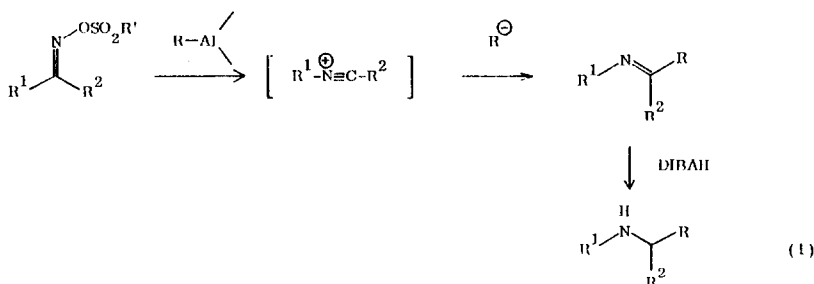
# STEREOSELECTIVE SYNTHESSES OF SOLENOPSIN A AND B

Yasushi Matsumura, Keiji Maruoka, and Hisashi Yamamoto\*

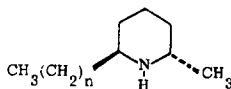
Department of Applied Chemistry, Nagoya University  
 Chikusa, Nagoya 464, Japan

**Summary:** Effective and convenient syntheses of solenopsin A and B have been developed which involve the Beckmann rearrangement-alkylation reaction promoted by organoaluminum reagents and a new stereo-selective reduction of imino functional group.

The recent discovery of the aluminum method for the direct synthesis of  $\alpha$ -alkylated amines from the simple oxime sulfonate (eq. 1) has substantially simplified the task of the synthesis in this area.<sup>1</sup> Thus,



the method should be especially useful in the field of alkaloid synthesis: for example, it provides a simple approach to the structure of 2,6-dialkylated piperidines from  $\alpha$ -alkylcyclopentanones. Solenopsin A (1) and B (2), naturally occurring piperidine alkaloids isolated from the venom of the fire ant, Solenopsis savissima,<sup>2</sup> are members of these class of compounds which exhibit pronounced hemolytic,<sup>3</sup> insecticidal, and antibiotic<sup>4</sup> activity. As far as we know there were no known synthetic methods capable of generating stereospecifically trans-2,6-disubstituted piperidine structure, characteristic of 1 and 2.<sup>5</sup> This fact, together with the occurrence of the same structural unit in other natural products,<sup>6</sup> led us to study the stereospecific route to solenopsin A and B.

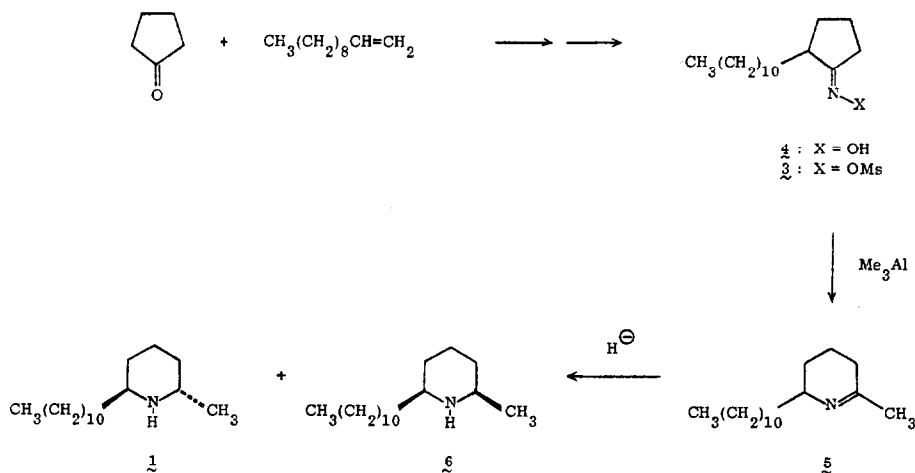


1 : SOLENOPSIN A (n = 10)

2 : SOLENOPSIN B (n = 12)

The starting oxime sulfonate **3** was synthesized from cyclopentanone in three steps: Reaction of excess cyclopentanone (10 equiv) with 1-undecene (1 equiv) in the presence of silver oxide (1 equiv) at 130°C for 5 h<sup>7</sup> produced after isolation in the usual way  $\alpha$ -undecylcyclopentanone which was treated with hydroxylamine hydrochloride (1.5 equiv)-sodium acetate (2 equiv) in methanol at 25°C for 5 h. Chromatography of the crude product on silica gel with 20% ether in hexane led to the isolation of pure anti-oxime **4**<sup>8</sup> in 73% overall yield from 1-undecene as a semi-solid, which was converted to the corresponding mesylate **3** with methane-sulfonyl chloride (1.1 equiv)-triethylamine (1.5 equiv) in methylene chloride at -20°C for 40 min (~95-100%). Treatment of the oxime mesylate **3** in dry methylene chloride with 2 equiv of trimethylaluminum (a 2 M toluene solution)<sup>10</sup> at -78°C for 5 min and at 25°C for 1 h resulted in formation of the imine **5** in 54% yield<sup>11</sup> after work up by NaF-H<sub>2</sub>O method.<sup>1</sup>

Completion of the synthesis requires reduction of the C=N double bond of **5** with correct configurations. Unfortunately, however, it was soon apparent from examination of the literature that existing methodology was totally inadequate for the selective reduction of **5** into trans-2,6-disubstituted piperidine structure.<sup>12</sup> Thus, the reduction of **5** using usual aluminum- or borohydride type reagents<sup>13</sup> in different solvents at low temperature afforded solenopsin A (**1**) and its cis isomer **6** in the range of <1/99 to 80/20 (GC analysis) as illustrated in Table I (Entry 1~10). Excellent stereoselectivity was finally attained in the formation of the trans form **1** by adding trimethylaluminum (1 equiv to LiAlH<sub>4</sub>) into the reaction mixture at low temperature, and solenopsin A (**1**) was obtained almost exclusively (>95%, Entry 17).<sup>14</sup> The spectral data (<sup>1</sup>H NMR and IR) of synthetic solenopsin A was identical with the reported ones.<sup>2</sup>



In a similar manner, solenopsin B (**2**) was prepared with high stereoselectivity (~95% by GC assay) using procedures which exactly paralleled those described above for the synthesis of solenopsin A.

**Table I.** Stereoselectivity of the Reduction of the Imine **5** with Hydride Reagents. <sup>a</sup>

Entry	Hydride Reagent	Solvent	Product Ratio <sup>b</sup> ( <b>6</b> : <b>1</b> )
1	DIBAH (4 eq)	CH <sub>2</sub> Cl <sub>2</sub>	99 : 1
2	LiAlH <sub>4</sub> (7 eq)-NaOMe (14 eq) <sup>c</sup>	THF	99 : 1
3	NaBH <sub>3</sub> CN (3 eq)-HCl <sup>d</sup>	MeOH	98 : 2
4	LiAlH <sub>4</sub> (7 eq)-Ti(OPr <sup>i</sup> ) <sub>4</sub> (7 eq)	THF	90 : 10
5	n-BuLi (5 eq)-DIBAH (5 eq) <sup>c</sup>	Ether	83 : 17
6	LiAlH <sub>4</sub> (25 eq)-LiCl (50 eq) <sup>c</sup>	THF	80 : 20
7	LiAlH <sub>4</sub> (7 eq)-NiCl <sub>2</sub> (7 eq) <sup>c</sup>	THF	75 : 25
8	LiAlH <sub>4</sub> (5 eq)-TiCl <sub>3</sub> (10 eq) <sup>c</sup>	THF	67 : 33
9	LiAlH <sub>4</sub> (25 eq)	CH <sub>2</sub> Cl <sub>2</sub> or THF	67 : 33
10	LiAlH <sub>4</sub> (25 eq)	Ether	20 : 80
11	Mg(AlH <sub>4</sub> ) <sub>2</sub> (25 eq)	Ether	33 : 67
12	LiAlH <sub>4</sub> (7 eq)-BF <sub>3</sub> OEt <sub>2</sub> (7 eq)	Ether	33 : 67
13	LiAlH <sub>4</sub> (7 eq)-TiCl <sub>4</sub> (7 eq)	THF	25 : 75
14	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al (7 eq) <sup>e</sup>	Ether	13 : 87
15	LiAlH <sub>4</sub> (7 eq)-Bu <sup>i</sup> Al (7 eq)	THF	6 : 94
16	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al (7 eq)	DME	6 : 94
17	LiAlH <sub>4</sub> (7 eq)-Me <sub>3</sub> Al (7 eq)	THF	5 : 95

a) All reactions were carried out on a 0.3-1 mmol scale. Unless specified, the imine **5** was added to hydride reagents at -78°C under argon, and then additives were introduced. The reaction mixture was stirred at -78°C for 30 min, at -45°C for 1 h, at -20°C for 1 h, and finally at 0°C for 1 h. After work-up by NaF-H<sub>2</sub>O method, the crude product was purified by column chromatography to give a mixture of cis and trans isomers in 94-97% yield. b) Isomeric ratio was determined by GLC (OV-101, 210°C): *t*<sub>R</sub>(**6**) = 3.91 min, *t*<sub>R</sub>(**1**) = 4.31 min. c) The imine was added to a mixture of hydride reagents and additives at -78°C. d) The imine was reduced at 25°C for 20 h, and worked up in a usual manner. e) The use of excess LiAlH<sub>4</sub> (7 eq of LiAlH<sub>4</sub> to 4 eq of Me<sub>3</sub>Al) or excess Me<sub>3</sub>Al (10 eq of Me<sub>3</sub>Al to 5 eq of LiAlH<sub>4</sub>) did not affect the ratio of **6** to **1**.

#### ACKNOWLEDGMENT

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- For example, see: a) pseudo-Carpaine: Govindachari, T. R.; Pai, B. R.; Narasimhan, M. S. *J. Chem. Soc.*, **1954**, 1847. Govindachari, T. R.; Nagarajan, K.; Viswanathan, N. *Tetrahedron Lett.*, **1965**, 1907. b) Himberline, Himandravine, and Himgravine: Pinhey, J. T.; Ritchie, E.; Taylor, W. C. *Austral. J. Chem.*, **1961**, *14*, 106. Fridrichsons, J.; Mathieson, A. M. *Acta Cryst.*, **1962**, *15*, 119. c) Lythranine and Lythranidine: Fujita, E.; Fuji, K.; Bessho, K.; Sumi, A.; Nakamura, S. *Tetrahedron Lett.*, **1967**, 4595.
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- $^1\text{H}$  NMR ( $\text{CCl}_4$ )  $\delta$ 9.09 (1H, br s, OH), 2.20-2.61 (3H, m,  $\text{N}=\text{C}-\text{CH}$ ,  $\text{N}=\text{C}-\text{CH}_2$ ), 0.88 (3H, br t,  $\text{CH}_3$ ).
- The crude oxime mesylate **3** was used without purification for the following rearrangement with trimethylaluminum. Satisfactory spectral ( $^1\text{H}$  NMR and IR) data were obtained for this compound.
- Purchased from Aldorich Chemical Company, Inc.
- $^1\text{H}$  NMR ( $\text{CCl}_4$ )  $\delta$ 3.87-3.40 (1H, m, NCH), 1.81 (3H, d,  $J = 2.0$  Hz,  $\text{N}=\text{C}-\text{CH}_3$ ), 0.89 (3H, br t,  $\text{CH}_3$ ); IR (liquid film) 2920, 2850, 1660, 1460, 1370  $\text{cm}^{-1}$ .
- Methods and selectivities for previous syntheses of solenopsin A: a) Reduction of 2,6-disubstituted pyridine with sodium in ethanol (*cis/trans* = 85:15).<sup>5a</sup> b) Mundy *N*-acyllactam rearrangement followed by reduction of the imine with  $\text{NaBH}_4$  (*cis/trans* = 80:20).<sup>5b</sup> c) Nitrosation, equilibrium by *t*-BuOK, and denitrosation of the *cis*-isomer (*cis/trans* = 50:50), which permit the full conversion to the *trans* isomer after repetition of this process.<sup>5d</sup>
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- The stereochemical outcome arising from the reduction of the imine in the presence of trialkylaluminum is tentatively rationalized by a theory of charge-transfer stabilization of the transition state for nucleophilic addition to a carbonyl group (Cieplak, A. S. *J. Am. Chem. Soc.*, **1981**, *103*, 4540.). In the hydride reduction of the imine **5**, the rear-side approach of the hydride toward the molecule is preferred by the stabilization of the  $\sigma_{\text{C}}^*$  orbital (low-lying vacant orbital of the imine) through electron delocalization from the  $\sigma_{\text{C}-\text{H}}$  bond into the  $\sigma_{\text{C}}^*$  orbital, producing the *cis* isomer. On the other hand, by the addition of Lewis acid ( $\text{R}'_3\text{Al}$ ), the R group of **5** would occupy the axial position because of the steric interaction between R and  $\text{R}'_3\text{Al}$  ( $\text{A}^{(1,2)}$ -Strain: see Nalua, A. S. *Tetrahedron Lett.*, **1981**, *22*, 2017.). This facilitates the front-side approach of the hydride toward the molecule to furnish the desired *trans* isomer

