

Available online at www.sciencedirect.com



Tetrahedron: Asymmetry 15 (2004) 3719-3722

Tetrahedron: Asymmetry

The stereoselective synthesis of nopinone based triazole ketones

Ronaldo C. da Silva Junior, Vítor F. Ferreira and Sergio Pinheiro*

Departamento de Química Orgânica, Instituto de Química, Universidade Federal Fluminense, CEG, Centro, 24020-150 Niterói, RJ, Brazil

Received 1 October 2004; accepted 19 October 2004

Abstract—The aldol reaction of nopinone 1 with triazole aldehydes followed by reduction with $Zn/ZnCl_2$ furnished isomers 2a–d in satisfactory overall yields and excellent diastereoselectivities (ca 94% de). © 2004 Elsevier Ltd. All rights reserved.

© 2004 Eisevier Etd. All fights feserved

1. Introduction

Triazoles represent an interesting class of compounds having a wide spectrum of biological properties.¹ Among them, some triazole substituted ketones are attractive target molecules because of their importance such as antifungal agents,² germicides³ and antibacterials⁴ including activities against tuberculosis.⁵ The asymmetric synthesis of triazole substituted ketone merits attention since the absolute configuration of the alpha position to the carbonyl plays an important role for the antifungal activity.^{2a}

Due to our continuing interest on the use of pinenes in asymmetric synthesis,⁶ we focused on the stereoselective synthesis of novel (+)-nopinone based triazole ketones with potential biological activities.

The alkylation of the enolate of (+)-nopinone **1** with alkyl halides is reported to lead to a mixture of **2** and **3** usually favouring the thermodynamic diastereomer **2** (Scheme 1),⁷ which was obtained in one case as the sole product by an aldol reaction of **1** with PhCHO followed by hydrogenation.⁸ Conversely the monoalkylation of



Scheme 1. Thermodynamic and kinetic alkylations of (+)-nopinone 1.

0957-4166/\$ - see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetasy.2004.10.020

the lithium enolate of **1** with alkyl halides under kinetic conditions was reported to furnish isomers **3** in excellent diastereoselectivities.⁹

Herein we report that, in contrast to the literature precedent, 9a,b the kinetic alkylation of the lithium enolate of 1 with triazole bromide 4 gives product 2, which was obtained in excellent diastereoselectivities from the aldol pathway.

2. Results and discussion

The 1,2,3-triazole bromide **4** was prepared from aldehyde **5**, a readily available derivative of D-(+)-glucose (Scheme 2).¹⁰ Reduction of **5** with NaBH₄ furnished alcohol **6**,¹¹ which upon treatment with PBr₃ in ethyl ether gave **4**.¹²

$$\overset{\text{OHC}}{\underset{N}{\overset{N-Ph}{\longrightarrow}}} \overset{\text{a}}{\xrightarrow{}} \overset{\text{HO}}{\underset{N}{\overset{N-Ph}{\longrightarrow}}} \overset{\text{b}}{\underset{N}{\overset{N-Ph}{\longrightarrow}}} \overset{\text{b}}{\xrightarrow{}} \overset{\text{Br}}{\underset{N}{\overset{N-Ph}{\longrightarrow}}} \overset{\text{N}}{\underset{N}{\overset{N-Ph}{\longrightarrow}}}$$

Scheme 2. Reagents and conditions: (a) NaBH₄, MeOH, 10min, 94%; (b) PBr₃, Et₂O, 24 h, 90%.

In contrast to that described in the literature for the use of alkyl halides,^{9a,b} the kinetic alkylation of the lithium enolate of 1^{13} with triazole bromide **4** at -45 °C in the presence of HMPA as additive following the typical experimental procedure^{9a,b} led to product **2a**, which was obtained in poor yields and stereoselectivity (Table 1, entry 1).¹⁴ At higher temperatures, compound **2a** was produced in similar yield and in better diastereoselectivity (typical).

^{*}Corresponding author. Fax: +5521 26292129; e-mail: spin@ rmn.uff.br

Table 1. Stereoselectivities in the alkylation of 1 with the bromide 4



^a Yields for the purified mixture of products **2a** and **3a** by flash chromatography on silica gel.

^b Ratios determined from the signals of both H_8 and H_{10} in the ¹H NMR spectra (300 MHz).

The 2a/3a ratios were determined from the signals due to the hydrogens H_8 , H_{10} and H_{12} in the ¹H NMR spectra at 300 MHz of these crude mixtures. The stereochemical assignment of the newly created stereogenic centre in the main isomer 2a was made on the basis of NOE NMR spectrum as already described in the literature for a related compound.⁸ Indeed, starting from H₈, significant NOE was observed with $H_{4\beta}$ (2.3%). Also observed were NOEs of $H_{4\alpha}$ with H_3 (7.7%) and of H_3 with $H_{7\alpha}$ (4.6%), which implies a *syn* relationship between the methylenetriazole appendage and the gem-dimethylene bridge of the pinane moiety. Also noteworthy, the NOE between H₈ and H₁₀ was not observed while a strong NOE of H_3 with H_{10} (9.1%) suggests for isomer 2a a preferred conformation in which the hydrogens H_8 and H_{10} are not close together.

The aldol pathway proved to be a very attractive route to isomer 2a in high diastereoselectivity (Scheme 3). The aldol condensation of 1 with aldehyde 5 was carried out following the typical procedure⁸ giving enone 7a as a single geometric isomer.¹⁵ The assignment of the configuration of the double bond was made on the basis of NOE NMR spectrum. Neither $H_{4\alpha}$ nor $H_{4\beta}$ gave NOE with H_{10} . However H_4 showed a strong NOE with H_{12} (8.4%) allowing us to define the geometry of the double bond. Since hydrogenation of 7a in Pd/C and in Pt failed, this compound was converted to 2a upon treatment with the system $Zn/ZnCl_2$.^{16,17} The 2a/3a = 97:3ratio was obtained from the signals of H₈, H₁₀ and H_{12} in the crude ¹H NMR spectrum while the stereochemistry at the 3-position was determined on the basis of NOE experiments, as reported above for the alkylation of 1.



Scheme 3. Reagents and conditions: (a) 5 (1 equiv), aq KOH, reflux, 20 h, 56%; (b) Zn, ZnCl₂, EtOH, reflux, 2h, 92%.

The catalytic hydrogenation of enone 8 derived from 1 was reported to furnish stereoselectively the *exo*-isomer

9.⁸ Reduction of **8** with $Zn/ZnCl_2$ gave **9** in an *exol* endo = 97:3 ratio. The specific rotation of **9** thus obtained was compared with the product reported by hydrogenation of **8**,⁸ showing that these methodologies occur with the same stereochemical sense (Scheme 4).



Scheme 4. Reagents and conditions: (a) see Ref. 8; (b) see Ref. 8: H₂, Pd/C, AcOEt, 95%, $[\alpha]_D^{20} = -56.4$ (*c* 1.6, EtOH); (c) Zn, ZnCl₂, EtOH, reflux, 2h, 93%, $[\alpha]_D^{20} = -50$ (*c* 1.6, EtOH).

In order to verify the scope of this stereochemical course producing mainly isomers 2, the already known triazole aldehydes 10–12¹⁸ were employed in the aldol pathway (Scheme 5). Thus, the aldol reactions of 1 with 10–12 furnished the corresponding enones 7b–d as single isomers, ^{19–21} which were reduced with the Zn/ZnCl₂ system in EtOH to the respective compounds 2b–d in good yields and diastereoselectivities (*exolendo* = 97:3).^{22–24} Also here the epimeric 2b–d/3b–d ratios at the 3-position were obtained from the signals of H₈ and H₁₂ in the crude ¹H NMR spectra while the absolute configurations at the 3-position were made from NOE spectra as reported for the alkylation of 1.



Scheme 5. Reagents and conditions: (a) 10 or 11 or 12 (1equiv), aq KOH, reflux, 20h (7b: 55%, 7c: 58%, 7d: 59%); (b) Zn, ZnCl₂, EtOH, reflux, 2h (2b: 75%, 2c: 70%, 2d: 80%).

3. Conclusion

In summary, the aldol pathway for the introduction of methylenetriazole moieties in nopinone 1 furnishing the isomers 2a-d in high stereoselectivities (ca 94% de), is an attractive and complementary protocol to the already reported kinetic alpha alkylation of the lithium enolate of 1, ^{9a,b} which was shown to be inefficient for the alkylation with triazole bromide 4.

Acknowledgements

The authors thank CNPq for financial support and are indebted to Dr. Maria Cecília B. V. de Souza as well as to Dr. Katia Z. Leal (UFF) for NMR spectra; to Dr. Marcos Eberlin (Unicamp) for mass spectra and to Dr. José O. Previatto (UFRJ) for specific rotation measurements. R.C.S.J. and V.F.F. are grateful to CNPq for research fellowships.

References

- See references cited in: (a) Collin, X.; Sauleau, A.; Coulon, J. Bioorg. Med. Chem. Lett. 2003, 13, 2601–2605; (b) Kritsanida, M.; Mouroutsou, A.; Marakos, P.; Pouli, N.; Garoufalias, S. P.; Pannecouque, C.; Witvrouw, M.; De Clercq, E. Farmaco 2002, 57, 253–257; (c) Liu, S.; Qjan, X.; Song, G.; Chen, J.; Chen, W. J. Fluorine Chem. 2000, 105, 111–115.
- (a) Ogata, M.; Matsumoto, H.; Takahashi, K.; Shimizu, S.; Kida, S.; Murabayashi, A.; Shiro, M.; Tawara, K. J. Med. Chem. 1987, 30, 1054–1068; (b) Ogata, M.; Matsumoto, H.; Kida, S.; Shimizu, S.; Tawara, K.; Kawamura, Y. J. Med. Chem. 1987, 30, 1497–1502; (c) Rekhter, M. A.; Grushetskaya, G. N.; Panasenko, A. A.; Krimer, M. Z. Chem. Heterocycl. Comp. 1995, 31, 792–796; (d) Uchil, V. R.; Joshi, V. Indian J. Chem. 1999, 38B, 192–196.
- Xiao, Y. D.; Wei, L. H.; Wang, J. T.; Zhang, J. B.; Lin, S. F.; Zhou Chemometr. Intell. Lab. Syst. 1999, 45, 277–280.
- 4. Soliman, F. M. A. J. Serb. Chem. Soc. 1995, 60, 9–14.
- Dabak, K.; Sezer, Ö.; Akar, A.; Anac, O. Eur. J. Med. Chem. 2003, 38, 215–218.
- (a) Pinheiro, S.; Pedraza, S. F.; Peralta, M. A.; Teixeira, R. C.; Farias, F. M. C.; Ferreira, V. F.; Costa, P. R. R. *Tetrahedron: Asymmetry* 2002, *13*, 2513–2517; (b) Pinheiro, S.; Gonçalves, C. B. S. S.; Lima, M. B.; Farias, F. M. C. *Tetrahedron: Asymmetry* 2000, *11*, 3495–3502; (c) Pinheiro, S.; Pedraza, S. F.; Farias, F. M. C.; Gonçalves, A. S. *Tetrahedron: Asymmetry* 2000, *11*, 3845–3848; (d) Costa, P. R. R.; Ferreira, V. F.; AraújoFilho, H. C.; Pinheiro, S. J. Braz. Chem. Soc. 1996, *7*, 67–73.
- Kato, M.; Watanabe, M.; Vogler, B.; Awen, B. Z.; Masuda, Y.; Tooyama, Y.; Yoshikoshi, A. J. Org. Chem. 1991, 56, 7071–7076.
- Dumas, F.; Alencar, K.; Mahuteau, J.; Barbero, M. J. L.; Miet, C.; Gérard, F.; Vasconcellos, M. L. A. A.; Costa, P. R. R. *Tetrahedron: Asymmetry* 1997, *8*, 579–583.
- (a) Campos, K. R.; Journet, M.; Cai, D.; Kowal, J. J.; Lee, S.; Larsen, R. D.; Reider, P. J. J. Org. Chem. 2003, 68, 2338–2342; (b) Campos, K. R.; Lee, S.; Journet, M.; Kowal, J. J.; Cai, D.; Larsen, R. D.; Reider, P. J. Tetrahedron Lett. 2002, 43, 6957–6959; (c) Konopelski, J. P.; Djerassi, C. J. Org. Chem. 1980, 45, 2297–2301.
- 10. Hann, M. R.; Hudson, C. S. J. Am. Chem. Soc. 1944, 66, 735–738.
- Alcohol 6: White solid, mp 64–65 °C. IR (KBr, cm⁻¹): 3317, 3232, 2943, 1595, 1489, 1414, 1353, 1306, 1050, 1014, 964, 849, 760, 730. ¹H NMR (300 MHz, CDCl₃, ppm): 8.05 (ddd, 7.2, 3.0, 1.2 Hz, CH), 7.80 (s, CH), 7.48 (ddd, 7.5, 7.2, 1.8 Hz, CH), 7.35 (tt, 7.5, 1.2 Hz, CH), 4.88 (d, 5.6 Hz, CH₂), 1.98 (t, 5.6 Hz, OH).
- Compound 4: White solid after flash chromatography on silica gel (10% AcOEt in hexane as eluant), mp 44-45 °C. IR (KBr, cm⁻¹): 3036, 1594, 1498, 1487, 1341, 1333, 1212, 1034, 966, 859, 760, 737, 711. ¹H NMR (300 MHz, CDCl₃, ppm): 8.07 (ddd, 7.3, 3.2, 1.2 Hz, CH), 7.82 (s, CH), 7.46 (ddd, 7.5, 7.3, 1.6 Hz, CH), 7.36 (tt, 7.5, 1.2 Hz, CH), 4.61 (s, CH₂).
- 13. (+)-Nopinone is available from Aldrich Chem. Co.
- 14. Mixture of isomers **2a** and **3a**: Pale yellow oil after flash chromatography on silica gel (1% AcOEt in hexane as eluant). $[\alpha]_D^{25} = -27.0$ (*c* 0.63, CH₂Cl₂) for **2a/3a** = 76:24. ¹H NMR (300 MHz, CDCl₃, ppm): 8.03 (ddd, 7.2, 3.0,

1.2 Hz, H-14), 7.66 (s, H-12 of 2a), 7.62 (s, H-12 of 3a), 7.46 (ddd, 7.5, 7.2, 1.8 Hz, H-15), 7.32 (tt, 7.5, 1.2 Hz, H-16), 3.52 (dd, 15.0, 4.2 Hz, H-10 of 2a), 3.41 (dd, 13.5, 4.5 Hz, H-10 of 3a), 3.10-3.00 (m, H-3 of 2a), 2.89-2.81 (m, H-3 of 3a), 2.94 (dd, 16.5, 4.5 Hz, H-10 of 3a), 2.72 (dd, 15.0, 9.6 Hz, H-10 of 2a), 2.66 (dd, 5.4, 5.1 Hz, H-1), 2.59-2.53 (m, H-7β of 3a), 2.52-2.45 (m, H-7β of 2a), 2.33 (ddd, 13.2, 10.2, 4.8 Hz, H-4a), 2.27 (ddd, 5.4, 5.1, 4.8 Hz, H-5), 1.76 (d, 10.5 Hz, H-7α), 1.71–1.64 (m, H-4β), 1.34 (s, H-9 of 2a), 1.33 (s, H-9 of 3a), 0.94 (s, H-8 of 3a), 0.75 (s, H-8 of 2a). ¹³C NMR (CDCl₃, ppm): 215.0 (C-2), 148.0 (C-11), 139.8 (C-13), 134.8 (C-12), 129.6 (C-15), 126.8 (C-16), 118.4 (C-14), 58.5 (C-1 of 3a), 57.6 (C-1 of 2a), 43.2 (C-6), 42.4 (C-3), 41.6 (C-5), 31.5 (C-4 of 3a), 28.6 (C-4 and C-7 of 2a), 28.2 (C-7 of 3a), 25.9 (C-10 of 3a), 25.4 (C-10 of 2a), 26.1 (C-9), 21.9 (C-8).

- 15. Compound 7a: White solid after flash chromatography on silica gel (10% AcOEt in hexane as eluant), mp 63-64 °C. $[\alpha]_{D}^{25} = -1.8 \ (c \ 2.65, \ CH_2Cl_2). \ IR \ (KBr, \ cm^{-1}): \ 3084, \ 2932,$ 1698, 1624, 1597, 1499, 1463, 1213, 1198, 1063, 965, 756, 690, 666. ¹H NMR (300 MHz, CDCl₃, ppm): 8.13 (ddd, 7.2, 3.0, 1.3 Hz, H_{arom}), 7.97 (s, H-12), 7.76 (t, 2.6 Hz, H-10), 7.51 (ddd, 7.5, 7.2, 1.8 Hz, H_{arom}), 7.38 (tt, 7.5, 1.2 Hz, Harom), 3.15 (ddd, 18.0, 2.7, 2.4 Hz, H-4a), 3.05 (dddd, 18.0, 2.4, 2.3, 2.1 Hz, H-4β), 2.75 (dd, 5.7, 5.6 Hz, H-1), 2.69 (dddd, 10.2, 6.0, 5.6, 2.1 Hz, H-7β), 2.46-2.40 (m, H-5), 1.55 (d, 10.2 Hz, H-7a), 1.41 (s, H-9), 0.94 (s, H-8). ^{3}C NMR (CDCl₃, ppm): 198.0 (C-2), 140.4 (C-11), 135.2 (C-13), 132.7 (C-12), 131.2 (C-3), 124.8 (C-15), 123.4 (C-16), 118.0 (C-10), 114.3 (C-14), 51.7 (C-1), 36.4 (C-6), 34.6 (C-5), 27.0 (C-4), 23.3 (C-7), 21.8 (C-9), 17.2 (C-8). MS (70 eV, m/z): 294 (M + 1, 16), 293 (M⁺, 77), 278 (41), 250 (100), 91 (29), 77 (36). HREIMS calcd for C₁₈H₁₉N₃O (M⁺): 293.1528. Found: 293.1506.
- 16. Toda, F.; Iida, K. Chem. Lett. 1976, 695-696.
- 17. Compound 2a in a 94% de: pale yellow oil by flash chromatography on silica gel (3% AcOEt in hexane as eluant). $[\alpha]_{D}^{25} = -46.0$ (c 0.76, CH₂Cl₂). IR (neat, cm⁻¹): 3053, 2927, 2870, 1705, 1598, 1519, 1500, 1463, 1348, 1201, 1038, 1024, 965, 852, 757, 691, 665. ¹H NMR (300 MHz, CDCl₃, ppm): 8.03 (ddd, 7.2, 3.0, 1.2 Hz, H-14), 7.66 (s, H-12), 7.46 (ddd, 7.5, 7.2, 1.8 Hz, H-15), 7.32 (tt, 7.5, 1.2 Hz, H-16), 3.52 (dd, 15.0, 4.2 Hz, H-10), 3.10-3.00 (m, H-3), 2.72 (dd, 15.0, 9.6 Hz, H-10), 2.66 (dd, 5.4, 5.1 Hz, H-1), 2.52-2.45 (m, H-7β), 2.33 (ddd, 13.2, 10.2, 4.8 Hz, H-4α), 2.27 (ddd, 5.4, 5.1, 4.8 Hz, H-5), 1.76 (d, 10.5 Hz, H-7a), 1.71–1.64 (m, H-4β), 1.34 (s, H-9), 0.75 (s, H-8). ¹³C NMR (CDCl₃, ppm): 215.0 (C-2), 148.0 (C-11), 139.8 (C-13), 134.8 (C-12), 129.6 (C-15), 126.8 (C-16), 118.4 (C-14), 57.6 (C-1), 43.2 (C-6), 42.4 (C-3), 41.6 (C-5), 28.6 (C-4 and C-7), 25.4 (C-10), 26.1 (C-9), 21.9 (C-8). MS (70 eV, m/z): 297 $(M + 2, 14), 296 (M + 1, 66), 295 (M^+, 100), 254 (25), 226$ (94), 185 (36), 171 (28), 159 (84), 158 (47), 137 (29), 103 (33), 95 (30), 91 (62), 83 (56), 77 (53), 55 (32). HREIMS calcd for C₁₈H₂₁N₃O (M⁺): 295.0366. Found: 295.0351.
- Cunha, A. C.; Figueiredo, J. M.; Tributino, J. L. M.; Miranda, A. L. P.; Castro, H. C.; Zingali, R. B.; Fraga, C. A. M.; Souza, M. C. B. V.; Ferreira, V. F.; Barreiro, E. J. *Bioorg. Med. Chem.* 2003, *11*, 2051–2059.
- Compound **7b**: White solid after flash chromatography on silica gel (10% AcOEt in hexane as eluant), mp 168– 169 °C. [α]_D²⁵ = +5.5 (*c* 2.3, CH₂Cl₂). IR (KBr, cm⁻¹): 3175, 3001, 2963, 2929, 1691, 1613, 1597, 1500, 1467, 1369, 1305, 1215, 1200, 1059, 1045, 983, 808, 759, 688. ¹H NMR (300 MHz, CDCl₃, ppm): 8.14 (s, H-12), 7.78 (ddd, 7.2, 3.0, 1.2 Hz, H_{arom}), 7.81 (t, 2.4 Hz, H-10), 7.56 (ddd, 7.5, 7.2, 1.8 Hz, H_{arom}), 7.48 (tt, 7.2, 1.2 Hz, H_{arom}), 3.16 (ddd, 18.0, 2.6, 2.4 Hz, H-4α), 3.06 (dddd, 18.0, 2.6, 2.4, 2.1 Hz, H-4β), 2.73 (dd, 5.7, 5.6 Hz, H-1), 2.68 (dddd, 10.2, 6.1,

5.7, 2.1 Hz, H-7β), 2.44–2.38 (m, H-5), 1.55 (d, 10.2 Hz, H-7α), 1.40 (s, H-9), 0.93 (s, H-8). ¹³C NMR (CDCl₃, ppm): 202.2 (C-2), 144.6 (C-11), 136.4 (C-13), 134.0 (C-3), 129.8 (C-15), 128.9 (C-16), 122.6 (C-10), 122.4 (C-12), 120.4 (C-14), 55.9 (C-1), 41.0 (C-6), 39.3 (C-5), 31.2 (C-4), 27.8 (C-7), 26.1 (C-9), 21.7 (C-8). MS (70 eV, *m*/*z*): 293 (M⁺, 3), 250 (10), 222 (43), 104 (38), 91 (17), 77 (100), 51 (29). HREIMS calcd for $C_{18}H_{19}N_3O$ (M⁺): 293.1528. Found: 293.1481.

- 20. Compound 7c: White solid after flash chromatography on silica gel (10% AcOEt in hexane as eluant), mp 143-144 °C. $[\alpha]_D^{25} = +2.5$ (c 2.29, CH₂Cl₂). IR (KBr, cm⁻¹): 3132, 2948, 1690, 1625, 1594, 1489, 1370, 1310, 1263, 1199, 1055, 1043, 984, 882, 797, 780, 678. ¹H NMR (300 MHz, CDCl₃, ppm): 8.14 (s, H-12), 7.81 (t, 2.0 Hz, H-10), 7.79 (t, 2.4 Hz, H_{arom}), 7.70 (dt, 7.8, 1.8 Hz, H_{arom}), 7.50 (t, 8.1 Hz, H_{arom}), 7.45 (ddd, 8.1, 1.8, 1.6 Hz, H_{arom}), 3.13 (ddd, 18.0, 2.7, 2.4 Hz, H-4α), 3.03 (dddd, 18.0, 2.4, 2.3, 2.1 Hz, H-4β), 2.74 (dd, 5.8, 5.6 Hz, H-1), 2.67 (dddd, 10.2, 6.0, 5.8, 2.1 Hz, H-7 β), 2.44-2.39 (m, H-5), 1.54 (d, 10.2 Hz, H-7 α), 1.41 (s, H-9), 0.93 (s, H-8). ¹³C NMR (CDCl₃, ppm): 202.0 (C-2), 144.8 (C-11), 137.6 (C-15), 135.6 (C-13), 134.4 (C-3), 131.0 (C-17), 129.0 (C-16), 122.6 (C-14), 122.2 (C-12), 120.8 (C-10), 118.4 (C-18), 56.0 (C-1), 41.1 (C-6), 39.2 (C-5), 31.2 (C-4), 27.7 (C-7), 26.2 (C-9), 21.8 (C-8). MS (70 eV, m/z): 329 (M + 2, 3), 327 (M⁺, 18), 299 (25), 298 (24), 264 (23), 258 (33), 256 (100), 228 (22), 193 (20), 154 (30), 138 (75), 111 (74), 91 (35), 75 (30), 55 (50). HREIMS
- calcd for C₁₈H₁₈N₃ClO (M⁺): 327.1138. Found: 327.1129. 21. Compound 7d: White solid after flash chromatography on silica gel (10% AcOEt in hexane as eluant), mp 164-165 °C. $[\alpha]_D^{25} = +11.2$ (c 2.09, CH₂Cl₂). IR (KBr, cm⁻¹): 3168, 2934, 1734, 1688, 1610, 1518, 1368, 1310, 1234, 1198, 1060, 1046, 984, 840, 819, 770, 612. ¹H NMR (300 MHz, CDCl₃, ppm): 8.11 (s, H-12), 7.79 (t, 2.4Hz, H-10), 7.76 (ddd, 9.3, 4.5, 2.1 Hz, H-14), 7.26 (ddd, 9.3, 8.1, 2.1 Hz, H-15), 3.14 (ddd, 18.0, 2.7, 2.4 Hz, H-4α), 3.04 (dddd, 18.0, 2.4, 2.3, 2.1 Hz, H-4β), 2.72 (dd, 5.7, 5.6 Hz, H-1), 2.68 (dddd, 10.2, 6.0, 5.7, 2.1 Hz, H-7β), 2.44-2.38 (m, H-5), 1.54 (d, 10.2 Hz, H-7a), 1.41 (s, H-9), 0.93 (s, H-8). ¹³C NMR (CDCl₃, ppm): 202.1 (C-2), 164.2 (C-11), 161.0 (C-16), 145.0 (C-13), 134.0 (C-3), 124.4 (C-10 and C-14), 122.3 (C-12), 116.9 (C-15), 56.0 (C-1), 41.0 (C-6), 39.3 (C-5), 31.2 (C-4), 27.8 (C-7), 26.1 (C-9), 21.7 (C-8). MS (70 eV, *m*/*z*): 311 (M⁺, 7), 283 (36), 268 (31), 240 (95), 212 (28), 172 (31), 122 (100), 95 (99), 91 (38), 55 (32). HREIMS calcd for $C_{18}H_{18}N_3FO(M^+)$: 311.1434. Found: 311.1432.
- 22. Compound **2b** (**2b**/**3b** = 97:3): White solid after flash chromatography on silica gel (3% AcOEt in hexane as eluant), mp 85–86 °C. $[\alpha]_{D}^{25} = -10.8$ (*c* 1.11, CH₂Cl₂). IR (KBr, cm⁻¹): 3131, 3086, 2950, 2925, 2870, 1706, 1599, 1503, 1226, 1198, 1047, 768, 695. ¹H NMR (300 MHz, CDCl₃, ppm): 7.88 (s, H-12 of **2b**), 7.80 (s, H-12 of **3b**), 7.72 (ddd, 7.2, 3.0, 1.2Hz, H-14), 7.51 (ddd, 7.5, 7.2, 1.8 Hz, H-15), 7.42 (tt, 7.5, 1.2 Hz, H-16), 3.46 (dd, 15.0, 4.2 Hz, H-10 of **2b**), 3.38 (dd, 14.3, 4.0 Hz, H-10 of **3b**), 3.17–3.03 (m, H-3), 2.91 (dd, 9.9, 4.2 Hz, H-10 of **3b**), 2.81 (dd, 15.0, 9.6 Hz, H-10 of **2b**), 2.63 (dd, 5.3, 5.1 Hz, H-1),

2.51-2.44 (m, H-7β), 2.37 (ddd, 13.2, 10.0, 4.6Hz, H-4α), 2.29–2.21 (m, H-5), 1.76 (d, 10.5Hz, H-7α), 1.72 (dd, 13.2, 7.8 Hz, H-4β), 1.34 (s, H-9 of **2b**), 1.33 (s, H-9 of **3b**), 0.94 (s, H-8 of **3b**), 0.70 (s, H-8 of **2b**). ¹³C NMR (CDCl₃, ppm): 215.0 (C-2), 146.8 (C-11), 137.2 (C-13), 129.6 (C-15), 128.2 (C-16), 120.2 (C-12 and C-14), 57.6 (C-1), 43.2 (C-10), 42.4 (C-3), 41.6 (C-5), 39.8 (C-6), 28.6 (C-4), 26.1 (C-9), 25.4 (C-7), 21.9 (C-8). MS (70 eV, *m/z*): 295 (M⁺, 3), 198 (23), 130 (100), 104 (59), 77 (93), 55 (22). HREIMS calcd for $C_{18}H_{21}N_3O$ (M⁺): 295.0476. Found: 295.0498.

- 23. Compound 2c (2c/3c = 97:3): White solid after flash chromatography on silica gel (3% AcOEt in hexane as eluant), mp 81–82 °C. $[\alpha]_D^{25} = -11.1$ (*c* 0.99, CH₂Cl₂). IR (neat, cm⁻¹): 3131, 3064, 2924, 2870, 1706, 1595, 1496, 1440, 1225, 1199, 1078, 1042, 804, 781, 770, 686. ¹H NMR (300 MHz, CDCl₃, ppm): 7.88 (s, H-12 of 2c), 7.80 (s, H-12 of 3c), 7.78 (dd, 2.1, 1.9Hz, H-14), 7.63 (ddd, 7.8, 2.1, 1.5 Hz, H-18), 7.45 (t, 7.8 Hz, H-17), 7.41 (ddd, 7.8, 2.1, 1.5 Hz, H-16), 3.44 (dd, 15.0, 5.1 Hz, H-10 of 2c), 3.37 (dd, 15.0, 6.0 Hz, H-10 of 3c), 3.16-3.03 (m, H-3), 2.90 (dd, 15.0, 5.8 Hz, H-10 of 3c), 2.81 (dd, 15.0, 7.9 Hz, H-10 of **2c**), 2.63 (t, 5.1 Hz, H-1), 2.52–2.44 (m, H-7β), 2.37 (ddd, 13.0, 10.0, 4.8 Hz, H-4a), 2.30-2.22 (m, H-5), 1.76 (d, 10.5 Hz, H-7α), 1.71 (dd, 13.0, 7.8 Hz, H-4β), 1.34 (s, H-9 of 2c), 1.33 (s, H-9 of 3c), 0.93 (s, H-8 of 3c), 0.70 (s, H-8 of 2c). ¹³C NMR (CDCl₃, ppm): 214.9 (C-2), 147.0 (C-11), 137.8 (C-15), 135.3 (C-13), 130.6 (C-17), 128.4 (C-16), 120.4 (C-14), 120.2 (C-12), 118.1 (C-18), 58.2 (C-1 of 3c), 57.5 (C-1 of 2c), 43.3 (C-6), 42.4 (C-3), 40.6 (C-5 of 2c), 40.0 (C-5 of 3c), 28.6 (C-4 of 2c), 28.2 (C-4 of 3c), 26.2 (C-9 of 2c), 26.0 (C-9 of 3c), 25.2 (C-7), 25.1 (C-10), 21.7 (C-8). MS (70 eV, *m*/*z*): 329 (M⁺, 2), 232 (20), 164 (100), 138 (64), 111 (68), 75 (30), 55 (36). HREIMS calcd for C₁₈H₂₀N₃ClO (M⁺): 329.1646. Found: 329.1672.
- 24. Compound 2d (2d/3d = 97:3): White solid after flash chromatography on silica gel (3% AcOEt in hexane as eluant), mp 82–83 °C. $[\alpha]_D^{25} = -9.7$ (*c* 1.03, CH₂Cl₂). IR (neat, cm⁻¹): 3131, 3080, 2971, 2937, 2872, 1700, 1517, 1456, 1227, 1047, 839, 817. ¹H NMR (300 MHz, CDCl₃, ppm): 7.85 (s, H-12 of 2d), 7.80 (s, H-12 of 3d), 7.70 (ddd, 6.8, 4.5, 2.4 Hz, H-14), 7.22 (ddd, 8.1, 6.8, 2.4 Hz, H-15), 3.44 (dd, 15.0, 5.1 Hz, H-10 of 2d), 3.37 (dd, 15.0, 6.0 Hz, H-10 of 3d), 3.18-3.03 (m, H-3), 2.91 (dd, 9.0, 4.5 Hz, H-10 of 3d), 2.83 (dd, 15.0, 7.9 Hz, H-10 of 2d), 2.63 (t, 5.1 Hz, H-1), 2.52-2.44 (m, H-7β), 2.39 (ddd, 13.2, 10.2, 4.8 Hz, H-4a), 2.29-2.23 (m, H-5), 1.77 (d, 10.5 Hz, H-7a), 1.72 (dd, 13.2, 7.8 Hz, H-4 β), 1.34 (s, H-9 of **2d**), 1.33 (s, H-9 of **3d**), 0.94 (s, H-8 of 3d), 0.69 (s, H-8 of 2d). ¹³C NMR (CDCl₃, ppm): 215.0 (C-2), 163.7 (C-11), 160.4 (C-16), 146.8 (C-13), 122.2 (C-14 of 2d), 122.1 (C-14 of 3d), 120.4 (C-12 of 2d), 120.3 (C-12 of 3d), 116.6 (C-15 of 3d), 116.3 (C-15 of 2d), 58.1 (C-1 of 3d), 57.5 (C-1 of 2d), 43.2 (C-3 of 3d), 42.4 (C-3 of 2d), 40.7 (C-5 of 3d), 40.6 (C-5 of 2d), 28.6 (C-4 of 2d), 28.2 (C-4 of 3d), 26.2 (C-9 of 2d), 26.0 (C-9 of 3d), 25.2 (C-7), 25.1 (C-10), 21.8 (C-8 of 3d), 21.7 (C-8 of 2d). MS (70 eV, m/z): 314 (M + 1, 1), 313 (M⁺, 2), 216 (20), 149 (51), 148 (100), 122 (51), 95 (33). HREIMS calcd for C₁₈H₂₀N₃FO (M⁺): 313.3512. Found: 313.3506.