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A new approach to isoindoloisoquinolinones. A simple synthesis of nuevamine

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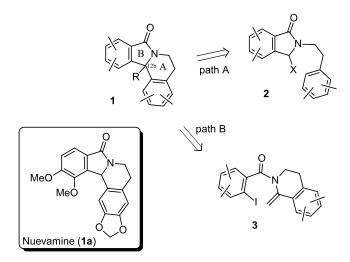
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Abstract—A convenient and versatile short step synthesis of isoindoloisoquinolinones, illustrated by the total synthesis of the alkaloid nuevamine **1a**, is described. The tetracyclic lactam compounds were obtained by a tactical combination of the Parham procedure for the elaboration of the isoindolinone template and an aryne-mediated cyclization giving rise to the nitrogen containing six-membered heteroring unit.

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1. Introduction

Members of the plant family Berberidaceae are known to produce an array of unusual isoquinoline alkaloids. Their study is facilitated by the fact that relatively large amounts of these plants can be collected. This therefore makes conceivable the study of minor alkaloids present, that is, those available only in small amounts, which often afford an insight into the catabolic pathways for the principal alkaloids. This is notably the case of isoindolo[1,2-a]isoquinolinones 1, a class of tetracyclic lactams which are interesting due to the real and potential biological activities of many of their derivatives.¹ The eminent example nuevamine 1a has been isolated from Berberis darwinii Hook, gathered in southern Chile, in the vicinity of Cuidad Osorno.² This alkaloid occupies a special place since it was the first recognized isoindoloisoquinolinone reported from natural sources. Initially, its structure was erroneously assigned but later revision led unambiguously to structure 1a.³ Two main general approaches to the synthesis of isoindolo[1,2-a]isoquinolinones have been reported so far. They differ in the type of the annulation process and in the heteroring unit embedded in the alkaloid skeleton, formed in the last step. Thus, creation of ring A has been generally achieved by intramolecular cyclization of N-acyliminium ions derived from α -substituted isoindolinone derivatives 2 (Retrosynthetic Scheme 1, path A) and a closer analysis of the literature reveals that improvement and modification of





this synthetic approach has been confined to the development of new precursors of the acyliminium species.

Thus, intramolecular amidoarylation as the key ringforming step has been ensured from α -hydroxy (X=OH)⁴ and α -alkoxy (X=OMe)⁵ lactamic precursors by treatment with a wide variety of Lewis acid catalysts such as scandium and copper triflates,^{5c,d} SnCl₄,^{5e} TMSOTf,^{5c} BF₃·OEt₂,^{5e,f} TiCl₄,^{5b} and trifluoroacetic acid.^{4,5a} An original structural modification recently developed by Katritzky led to the generation of the *N*-acyliminium cation by loss of a benzotriazolyl anion upon treatment of **2** (X=benzotriazolyl) with TiCl₄.⁶ An alternative synthetic tactic has been also proposed by Sotomayor in which the intermediate *N*-acyliminium species was generated from a preformed

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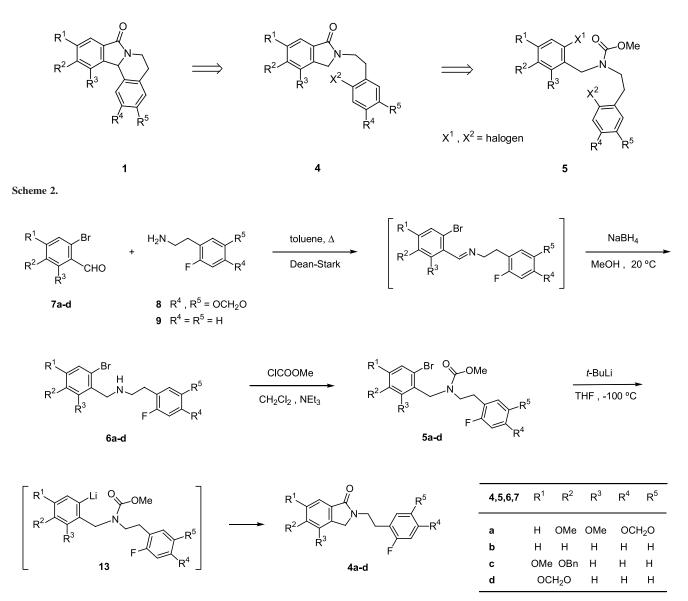
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annulated model 1 (R=OH).⁷ This skilful approach allowed intermolecular α -amidoalkylation, giving access to a wide variety of C-12b substituted nuevamine-type alkaloids. However, probably due to difficulties associated with the elaboration of the starting compounds and particularly the absolute necessity of having an unsymmetrically disubstituted isoindolinone parent compound available, most of these syntheses have only been claimed to give access to the nuevamine skeleton. Isoindoloisoquinolinones have been also accessed by ring B formation performed by intramolecular Heck cyclization of aromatic enamides 3 (Scheme 1, path B). In this annulation process the 6-endo-trig cyclization often competes with the 5-exo-trig cyclization but this problem has been circumvented by the addition of a hydride source which favors the regio-controlled formation of the five-membered ring product.⁸ Unfortunately, all models elaborated by this technique are inevitably alkylated at the C-12b position and this precludes the synthesis of unsubstituted models such as the alkaloid nuevamine. Some more confidential methods based upon the phosphoric acid cyclization of β -phenylethylaminophthalide⁹ or by a

combination of the Pictet–Spengler reaction with Pdcatalyzed carbonylation applied to (2-iodobenzylidene)phenethylamine derivatives¹⁰ have been also used occasionally for the assemblage of the isoindoloisoquinolinone framework. Consequently, the elegant synthetic approach reported by Castedo et al.³ which allowed the unambiguous assignment of the structure of the alkaloid nuevamine can arguably be deemed to be the sole total synthesis to date of this natural product.

2. Results and discussion

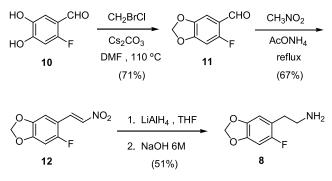
In continuation of our investigation into the synthesis of alkaloids with an isoindolinone ring system as the main structural subunit¹¹ we herein wish to disclose an alternative, efficient and tactically new approach to isoindoloiso-quinolinones illustrated by the total synthesis of the alkaloid nuevamine isolated as a racemate from Chilean barberries. Our strategy, which is depicted in retrosynthetic Scheme 2, is based on the use of 2-arylethylisoindolinones **4** as key



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intermediates. These lactams were obtained by Parham-type cyclization induced by aromatic lithiation of the halogenodiarylalkylamine 5 with a carbamate function acting as the internal electrophile.¹² Aryne-mediated cyclization induced by basic treatment of the primarily annulated compounds 4 then should complete the synthesis of the target isoindoloisoquinolinones 1. A contentious issue in the elaboration of the starting compounds was judging the proper choice for the halogen atoms X^1 , X^2 connected to the environmentally different aromatic units of 5. Critical to the success of the strategy was the incorporation of a halogen atom X¹ liable to cause the aromatic metallation-cyclization sequence leading to 4 by a metal-halogen exchange, while sparing the second halogen atom X^2 . In contrast X^2 should allow and, if feasible, facilitate the mandatory aryne formation for the ultimate annulation step. Since it has been clearly established that metal-halogen exchange occurs preferentially with aryl bromides and iodides, whereas arylfluorides and chlorides do not normally react with organolithium compounds but undergo instead ortho-metallation and aryne formation afterwards¹³ we embarked on the synthesis of the diversely halogenated carbamates **5a-d** (Scheme 3).

The first facet of the synthesis was the elaboration of the halogenated diarylalkylamines 6a-d. These compounds were readily obtained by a reductive amination process involving the bromobenzaldehyde derivatives 7a-d and the appropriate fluorinated arylethylamines 8, 9. Initially, the aromatic fluorinated amine 8 was obtained by the three step sequence depicted in Scheme 4 which involves the double reduction of the nitrostyrene derivative 12 deriving from the fluorocarboxaldehyde derivative 11.





The compound **11** was readily accessible by linking the vicinal hydrophenolic functions of the parent compound **10** (Scheme 4).

Treatment of amines 6a-d with methyl chloroformate delivered the carbamates 5a-d in excellent yields (73–91%). To ensure the formation of the lithiated intermediate

Table 1. Yields for compounds $6a\!-\!d,\;5a\!-\!d$ and $4a\!-\!d$ produced via Scheme 3

	6 (%)	5 (%)	4 (%)
a	94	78	68
b	86	91	68 72
с	97	76	64
d	95	73	64 62

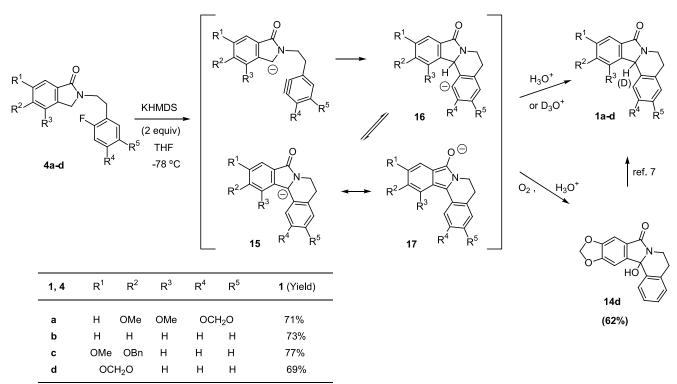
13 a THF solution of the appropriate arylbromide was treated with 1.1 equiv. of *t*-BuLi at -100 °C in THF. The intramolecular ring closure was instantaneous as demonstrated by the isolation solely of the annulated compounds **4a**-**d** upon aqueous work-up (Table 1).

The concept of the subsequent generation of the sixmembered heteroring unit embedded in the isoindoloisoquinolinone skeleton originated from the following premises: (i) isoindolinones have been easily metalated at the benzylic position of the lactam unit thus allowing the connection of a range of electrophiles at the 3-position of the heteroring system;¹⁴ (ii) arylfluorides are good candidates for the generation of arynes^{13,15a} and the anion-aryne cyclization processes, particularly those involving nitrogenbearing nucleophiles and carbanions, occupy a place of choice in the arsenal of alkaloid synthesis tactics.^{15b} Compounds 4b-d were then treated with KHMDS [potassium bis(trimethylsilyl)amide] (2.2 equiv.) in the presence of 18-crown-6 at -78 °C in THF and gratifyingly classical work-up delivered single compounds which were unambiguously identified as the annulated isoindolinones 1b-d. Performing this reaction sequence with the appropriate open model 4a led straightforwardly to the target natural product nuevamine 1a with a very satisfactory yield (38% over last three steps).

It is worth noting that reactions must be preferably carried out after careful degassing of the solution since quite significant amounts of products which were hydroxylated at C-12b, as exemplified by the formation of 14d, were obtained in reactions performed in non-deoxygenated solutions (Scheme 5). The formation of this α -hydroxylactam may tentatively be explained by capture of the carbanionic species 15 by oxygen. Intermediate 15 would be formed by a prototropy of 16, attributable to a greater stabilization of the bibenzylic carbanionic species which can further adopt the o-quinodimethane structure 17. The fact that acidic work-up with D_3O^+ led to the exclusive incorporation of deuterium on the bibenzylic C-12b position as exemplified by the formation of 1a(D) corroborates this hypothesis. However, this reaction was not really detrimental to the elaboration of the desired compounds 1a-dsince the hemiaminal compounds (e.g., 14d) can be almost quantitatively converted into the dehydroxylated models $(e.g., 1d)^7$ and can also serve for the introduction of substituents at C-12b.7

3. Conclusion

In summary, we have identified a novel and flexible synthetic approach to isoindoloisoquinolinones. In only two key synthetic steps from easily accessible precursors we have prepared the isoindolinone template by reliance on the Parham procedure and the creation of the isoquinoline ring system by an aryne-mediated cyclization process. The total synthesis of the alkaloid nuevamine emphasizes the versatility and regiospecificity of this new conceptual approach and we believe that this work demonstrates a general new methodology for the preparation of similar structurally modified isoindoloisoquinolinone alkaloids. A. Moreau et al. / Tetrahedron 60 (2004) 6169-6176



Scheme 5.

4. Experimental

4.1. General

Tetrahydrofuran (THF) was pre-dried with anhydrous Na₂SO₄ and distilled over sodium benzophenone ketyl under Ar before use. DMF, CH₂Cl₂, NEt₃, and toluene were distilled from CaH₂. Dry glassware was obtained by ovendrying and assembly under dry Ar. The glassware was equipped with rubber septa and reagent transfer was performed by syringe techniques. For flash chromatography, Merck silica gel 60 (40–63 μ ; 230–400 mesh ASTM) was used. The melting points were obtained on a Reichert–Thermopan apparatus and are not corrected. NMR spectra: Bruker AM 300 (300 and 75 MHz, for ¹H and ¹³C respectively). For ¹H, ¹³C NMR, CDCl₃ was used as solvent, TMS as internal standard. Microanalyses were performed by the CNRS microanalysis center.

The bromobenzaldehyde derivatives $7a^{16,17}$ and $7c^{18,19}$ were synthesized according to literature methods.

4.2. Synthesis of 2-(6-fluorobenzo[1,3]dioxol-5-yl)ethylamine (8)

2-Fluoro-4,5-dihydroxybenzaldehyde **10** was obtained from 4-fluoroveratrole by sequential formylation²⁰ and demethylation.²¹

4.2.1. 6-Fluorobenzo[1,3]dioxole-5-carbaldehyde (6-fluoropiperonal, 11). Bromochloromethane (2.61 g, 20.2 mmol) was added, under Ar, to a stirred suspension of 2-fluoro-4,5-dihydroxybenzaldehyde **10** (2.10 g, 13.4

mmol) and cesium carbonate (6.57 g, 20.2 mmol) in anhydrous DMF (20 mL) and the resulting mixture was heated at 110 °C for 3 h. After cooling to room temperature the mixture was filtered through a pad of Celite[®] which was subsequently washed with EtOAc (50 mL). Water (50 mL) was added, the organic layer was separated and the aqueous layer was extracted with EtOAc (3×50 mL). The organic layers were combined, washed with water, brine and dried (MgSO₄). Evaporation of the solvent left a residue, which was purified by flash column chromatography with hexanes/ AcOEt (50:50) as eluent. Yield 1.60 g (71%); mp 72–73 °C; ¹H NMR (δ) 6.05 (s, 2H, OCH₂O), 6.61 (d, *J*_{HF}=9.8 Hz, 1H, aromatic H), 7.19 (d, J_{HF}=5.4 Hz, 1H, aromatic H), 10.15 (s, 1H, CHO); ¹³C NMR (δ) 97.9 (d, J_{CF} =29 Hz), 102.9, 104.9 (d, $J_{CF}=3$ Hz), 117.9 (d, $J_{CF}=8$ Hz), 144.8 (d, $J_{CF}=$ 1 Hz), 154.0 (d, J_{CF} =15 Hz), 162.5 (d, J_{CF} =252 Hz), 185.4 (d, *J*_{CF}=8 Hz). Anal. Calcd for C₈H₅O₃F (168.1): C, 57.15; H, 3.00%. Found: C, 57.35; H, 2.94%.

4.2.2. 5-Fluoro-6-(2-nitrovinyl)-benzo[1,3]dioxole (12). A solution of 6-fluorobenzo[1,3]dioxole-5-carbaldehyde (1.45 g, 8.6 mmol), AcONH₄ (0.15 g, mmol) in nitromethane (CH₃NO₂, 6.5 mL) was refluxed for 5 h. After cooling, the yellow precipitate was filtered and washed with MeOH (20 mL) to afford yellow crystals. Yield 1.12 g, (67%); mp 143–144 °C; ¹H NMR (δ) 6.07 (s, 2H, OCH₂O), 6.68 (d, *J*_{HF}=9.8 Hz, 1H, aromatic H), 6.87 (d, *J*_{HF}=5.9 Hz, 1H, aromatic H), 7.55 (d, *J*=13.7 Hz, 1H, CH=), 8.03 (d, *J*=13.7 Hz, 1H, CH=); ¹³C NMR (δ) 98.7 (d, *J*_{CF}=30 Hz), 102.9, 106.8 (d, *J*_{CF}=4 Hz), 110.6 (d, *J*_{CF}=13 Hz), 132.2 (d, *J*_{CF}=3 Hz), 137.0 (d, *J*_{CF}=9 Hz), 144.9, 152.1 (d, *J*_{CF}=15 Hz), 158.5 (d, *J*_{CF}=250 Hz). Anal. Calcd for C₉H₆FNO₄ (211.1): C, 51.20; H, 2.86; N, 6.63%. Found: C, 51.02; H, 3.01; N, 6.55%.

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4.2.3. 2-(6-Fluorobenzo[1,3]dioxol-5-vl)ethylamine (8). A solution of the nitro derivative **12** (2.00 g, 10.3 mmol) in dry THF (15 mL) was added dropwise, under Ar, to an ice cooled suspension of lithium aluminum hydride (LiAlH₄, 3.89 g, 102.6 mmol) in dry THF (50 mL). The mixture was stirred at 0 °C for 2 h and then refluxed overnight. The suspension was cooled to 0 °C and the excess of LiAlH₄ was quenched with 6 M aqueous sodium hydroxide (50 mL). The precipitate was filtered off and extracted with boiling THF for 1 h, then filtered and washed with Et₂O (50 mL). The combined organic layers were washed with brine and dried (Na_2SO_4). After evaporation of the solvent the residue was purified by flash column chromatography with AcOEt/ Et₃N (90:10) as eluent. Yield 0.96 g (51%); oil; ¹H NMR $(\delta_{\rm H})$ 1.76 (br s, 2H, NH₂), 2.73 (t, J=7.1 Hz, 2H, CH₂), 3.23 (t, J=7.1 Hz, 2H, CH₂), 5.79 (s, 2H, OCH₂O), 6.43 (d, $J_{\rm HF}$ =8.8 Hz, 1H, aromatic H), 6.54 (d, $J_{\rm HF}$ =5.6 Hz, 1H, aromatic H); ¹³C NMR (δ_C) 29.2, 51.9, 97.7 (d, J_{CF} = 31 Hz), 101.5, 109.6 (d, J_{CF} =6 Hz), 119.0 (d, J_{CF} =14 Hz), 143.3 (d, $J_{CF}=2$ Hz), 146.2 (d, $J_{CF}=14$ Hz), 155.6 (d, J_{CF}=237 Hz). Anal. Calcd for C₉H₁₀FNO₂ (183.2): C, 59.01; H, 5.50; N, 7.65%. Found: C, 59.23; H, 5.40; N, 7.77%.

4.3. General procedure for the synthesis of the amines 6a-d

A solution of the [2-(2-fluorophenyl)ethyl]amine derivative 8 or 9 (10 mmol) and the appropriate bromobenzaldehyde derivative 7a-d (10 mmol) in toluene (40 mL) was refluxed for 3 h in a Dean-Stark apparatus. After removal of toluene in vacuo the N-benzylidene-[2-(2-fluorophenyl)ethyl]amine derivative was used directly in the next step without further purification. NaBH₄ (20 mmol, 0.76 g) was added portionwise to a solution of the previously obtained Schiff base in MeOH (50 mL) at room temperature. The mixture was stirred at room temperature for 1 h and the solvent was removed under vacuo. The crude mixture was dissolved in dichloromethane (50 mL), washed with saturated aqueous NH₄Cl solution (30 mL) and then brine (30 mL). The organic solution was dried (Na₂SO₄), filtered and concentrated in vacuo. The crude oily amines 6a-d were purified by flash column chromatography with AcOEt/hexanes/NEt₃ (80:10:10) as eluent.

4.3.1. (6-Bromo-2,3-dimethoxybenzyl)-[2-(6-fluorobenzo[1,3]dioxol-5-yl)ethyl]amine (6a). Yield 3.78 g, (94%), oil; ¹H NMR (δ) 1.70 (br s, 1H, NH), 2.68–2.77 (m, 4H, 2×CH₂), 3.79 (s, 6H, 2×CH₃), 3.92 (s, 2H, CH₂), 5.86 (s, 2H, OCH₂O), 6.50 (d, J_{HF} =9.0 Hz, 1H, aromatic H), 6.60 (d, J_{HF} =6.1 Hz, 1H, aromatic H), 6.67 (d, J= 8.8 Hz, 1H, aromatic H), 7.19 (d, J=8.8 Hz, 1H, aromatic H); ¹³C NMR (δ) 29.4 (d, J_{CF} =2 Hz), 47.6, 48.7, 55.8, 61.1, 97.8 (d, J_{CF} =31 Hz), 101.5, 112.5 (d, J_{CF} =6 Hz), 112.5, 115.4, 118.7 (d, J_{CF} =19 Hz), 127.8, 133.6, 143.4 (d, J_{CF} =2 Hz), 146.2 (d, J_{CF} =14 Hz), 148.6, 152.2, 155.6 (d, J_{CF} =236 Hz). Anal. Calcd for C₁₈H₁₉BrFNO₄ (412.3): C, 52.44; H, 4.65; N, 3.40%. Found: C, 52.56; H, 4.78; N, 3.17%.

4.3.2. (2-Bromobenzyl)-[2-(2-fluorophenyl)ethyl]amine (6b). Yield 2.65 g (86%); oil; ¹H NMR (δ) 1.57 (br s, 1H, NH), 2.91 (s, 4H, 2×CH₂), 3.90 (s, 2H, CH₂), 6.99–7.13 (m,

3H, aromatic H), 7.15–7.29 (m, 3H, aromatic H), 7.37 (d, J=7.6 Hz, 1H, aromatic H), 7.53 (d, J=7.8 Hz, 1H, aromatic H); ¹³C NMR (δ) 29.8 (d, J_{CF} =2 Hz), 49.1 (d, J_{CF} =1 Hz), 53.5, 115.3 (d, J_{CF} =22 Hz), 124.0 (d, J_{CF} =4 Hz), 124.0, 126.9 (d, J_{CF} =16 Hz), 127.4, 127.9 (d, J_{CF} =8 Hz), 128.5, 130.2, 131.0 (d, J_{CF} =5 Hz), 132.8, 139.3, 161.3 (d, J_{CF} =243 Hz). Anal. Calcd for C₁₅H₁₅BrFN (308.2): C, 58.46; H, 4.91; N, 4.54%. Found: C, 58.21; H, 5.14; N, 4.41%.

4.3.3. (5-Benzyloxy-2-bromo-4-methoxybenzyl)-[2-(2-fluorophenyl)ethyl]amine (6c). Yield 4.31 g (97%); oil; ¹H NMR (δ) 1.56 (br s, 1H, NH), 2.81 (s, 4H, 2×CH₂), 3.76 (s, 2H, CH₂), 3.85 (s, 3H, CH₃), 5.10 (s, 2H, CH₂), 6.93 (s, 1H, aromatic H), 6.98–7.07 (m, 3H, aromatic H), 7.16–7.21 (m, 2H, aromatic H), 7.25–7.36 (m, 3H, aromatic H), 7.42 (d, *J*=7.1 Hz, 2H, aromatic H); ¹³C NMR (δ) 29.7 (d, *J*_{CF}=2 Hz), 48.8, 53.0, 56.3, 71.1, 114.2, 115.3 (d, *J*_{CF}=2 Hz), 115.6, 116.1, 124.0 (d, *J*_{CF}=4 Hz), 126.9 (d, *J*_{CF}=16 Hz), 127.4, 127.9 (d, *J*_{CF}=8 Hz), 128.0, 128.6, 130.9 (d, *J*_{CF}=5 Hz), 131.3, 136.8, 147.4, 149.2, 161.3 (d, *J*_{CF}=243 Hz). Anal. Calcd for C₂₃H₂₃BrFNO₂ (444.35): C, 62.17; H, 5.22; N, 3.15%. Found: C, 62.23; H, 5.30; N, 3.29%.

4.3.4. (6-Bromobenzo[1,3]dioxol-5-ylmethyl)-[2-(2-fluorophenyl)ethyl]amine (6d). Yield 3.34 g (95%); oil; ¹H NMR (δ) 1.54 (br s, 1H, NH), 2.86 (s, 4H, 2×CH₂), 3.77 (s, 2H, CH₂), 5.94 (s, 2H, CH₂), 6.86 (s, 1H, aromatic H), 6.96 (s, 1H, aromatic H), 7.02–7.08 (m, 2H, aromatic H), 7.14–7.25 (m, 2H, aromatic H); ¹³C NMR (δ_C) 29.8 (d, J_{CF} =2 Hz), 48.9 (d, J_{CF} =1 Hz), 53.4, 101.6, 110.0, 112.7, 114.0, 115.3 (d, J_{CF} =22 Hz), 124.0 (d, J_{CF} =4 Hz), 126.8 (d, J_{CF} =16 Hz), 127.9 (d, J_{C-F} =8 Hz), 131.0 (d, J_{CF} =5 Hz), 132.5, 147.3, 147.4, 161.3 (d, J_{CF} =243 Hz). Anal. Calcd for C₁₆H₁₅BrFNO₂ (352.2): C, 54.56; H, 4.29; N, 3.98%. Found: C, 54.23; H, 4.51; N, 4.19%.

4.4. General procedure for the synthesis of the carbamic acid methyl esters 5a-d

Methyl chloroformate (0.99 g, 10.5 mmol) was added dropwise at 0 °C to a stirred solution of the secondary amine **6a**-**d** (7.0 mmol) and NEt₃ (1.42 g, 14.0 mmol) in CH₂Cl₂ (40 mL). The mixture was allowed to warm to room temperature and then stirred for an additional 3 h. The mixture was washed with water (30 mL) and brine (30 mL). The organic layer was dried (MgSO₄) and concentrated in vacuo to left an oily residue which was purified by flash column chromatography with AcOEt/hexanes (40:60) as eluent.

4.4.1. (6-Bromo-2,3-dimethoxybenzyl)-[2-(6-fluorobenzo-[1,3]dioxol-5-yl)ethyl]carbamic acid methyl ester (5a). Yield 2.57 g (78%); oil; ¹H NMR (δ , mixture of two rotational isomers 2:3) 2.52–2.64 (m, 2H, CH₂), 3.12–3.23 (m, 2H, CH₂), 3.74 (s, 3H, CH₃), 3.76 (s, 6H, 2×CH₃), 4.61 and 4.68 (2×s, together 2H, CH₂), 5.81 (s, 2H, OCH₂O), 6.41 (d, J_{HF} =8.6 Hz, 1H, aromatic H), 6.59 (d, J_{HF} =7.3 Hz, 1H, aromatic H), 6.70 (d, J=8.8 Hz, 1H, aromatic H), 7.18 (d, J=8.8 Hz, 1H, aromatic H); ¹³C NMR (δ , mixture of two rotational isomers) 28.0 (d, J_{CF} =2 Hz), 45.0, 45.4, 52.5, 55.9, 60.8, 97.6 (d, J_{CF} =30 Hz), 101.5, 109.4 (d, J_{CF} = 5 Hz), 113.4, 115.8, 117.6 (d, J_{CF} =15 Hz), 127.9, 130.1, 143.3 (d, $J_{CF}=2$ Hz), 146.3 (d, $J_{CF}=14$ Hz), 149.5, 152.3, 155.6 (d, $J_{CF}=237$ Hz), 156.6. Anal. Calcd for $C_{20}H_{21}$ -BrFNO₆ (470.3): C, 51.08; H, 4.50; N, 2.98%. Found: C, 51.27; H, 4.32; N, 2.84%.

4.4.2. (2-Bromobenzyl)-[2-(2-fluorophenyl)ethyl]carbamic acid methyl ester (5b). Yield 2.33 g (91%); oil; ¹H NMR (δ) (δ , mixture of two rotational isomers 4:5) 2.86–2.94 (m, 2H, CH₂), 3.44–3.50 (m, 2H, CH₂), 3.68 (s, 3H, CH₃), 4.46 and 4.57 (2×s, together 2H, CH₂), 6.96–7.25 (m, 7H, aromatic H), 7.51 (d, *J*=7.8 Hz, 1H, aromatic H); ¹³C NMR (δ_{C}), (δ , mixture of two rotational isomers A and B) 27.7 (A), 28.3 (B), 46.9 (B), 47.9 (A), 50.6 (B), 50.9 (A), 52.8, 115.3 (d, *J*_{CF}=22 Hz), 122.9 (A), 123.4 (B), 124.1 (d, *J*_{CF}=4 Hz), 125.7 (d, *J*_{C-F}=16 Hz), 127.6 (B), 127.9 (A), 128.3 (d, *J*_{CF}=5 Hz), 132.8, 136.8, 156.8 (A), 157.2 (B), 161.3 (d, *J*_{CF}=244 Hz). Anal. Calcd for C₁₇H₁₇BrFNO₂ (366.2): C, 55.75; H, 4.68; N, 3.82%. Found: C, 55.65; H, 4.49; N, 4.13%.

4.4.3. (5-Benzyloxy-2-bromo-4-methoxybenzyl)-[2-(2fluorophenyl)ethyl]carbamic acid methyl ester (5c). Yield 2.67 g (76%); mp 54–55 °C (from hexane-toluene); ¹H NMR (δ , mixture of two rotational isomers 4:5) 2.70– 2.85 (m, 2H, CH₂), 3.22-3.34 (m, 2H, CH₂), 3.66 (s, 3H, CH₃), 3.85 (s, 3H, CH₃), 4.33 and 4.46 (2×s, together 2H, CH₂), 5.10 (s, 2H, OCH₂O), 6.62 and 6.84 (2×s, together 1H, s, aromatic H), 6.96-7.25 (m, 6H, aromatic H), 7.29-7.37 (m, 4H, aromatic H); ¹³C NMR (δ_C , mixture of two rotational isomers) 29.7, 46.4, 49.8, 52.7, 56.3, 71.1, 113.9, 115.2 (d, J_{CF}=22 Hz), 115.8, 116.1, 124.0 (d, J_{CF}=4 Hz), 125.8 (d, J_{CF}=16 Hz), 127.4, 127.9, 128.2 (d, J_{CF}=8 Hz), 128.6, 128.9, 131.2 (d, J_{CF}=5 Hz), 136.6, 147.6, 149.5, 157.2, 161.3 (d, $J_{CF}=244$ Hz). Anal. Calcd for $C_{25}H_{25}$ -BrFNO₄ (502.4): C, 59.77; H, 5.02; N, 2.79%. Found: C, 59.92; H, 5.11; N, 3.02%.

4.4.4. (6-Bromobenzo[1,3]dioxol-5-ylmethyl)-[2-(2-fluorophenyl)ethyl]carbamic acid methyl ester (5d). Yield 2.09 g (73%); mp 76-78 °C (from hexane-toluene); ¹H NMR ($\delta_{\rm H}$, mixture of two rotational isomers 2:3) 2.80–2.89 (m, 2H, CH₂), 3.36–3.44 (m, 2H, CH₂), 3.64 (s, 3H, CH₃), 4.34 and 4.44 (2×s, together 2H, CH₂), 5.91 (s, 2H, OCH₂O), 6.60 and 6.75 (2×s, together 1H, aromatic H), 6.93 (s, 1H, aromatic H), 6.96-7.09 (m, 2H, aromatic H), 7.18–7.25 (m, 2H, aromatic H); ¹³C NMR ($\delta_{\rm C}$, mixture of two rotational isomers A and B) 27.6 (A), 28.2 (B), 46.7 (B), 47.6 (A), 50.2 (B), 50.5 (A), 52.8, 101.8, 108.1 (A), 109.0 (B), 112.5 (B), 112.7 (A), 113.3 (A), 113.6 (B), 115.2 (d, J_{CF} =22 Hz), 124.1 (d, J_{CF} =4 Hz), 125.7 (d, J_{CF} =16 Hz), 128.2 (d, J_{CF} =8 Hz), 130.0 (A), 130.1 (B), 131.2 (d, J_{CF} = 5 Hz), 147.7, 156.6 (A), 157.2 (B), 161.3 (d, *J*_{CF}=244 Hz). Anal. Calcd for C₁₈H₁₇BrFNO₄ (410.2): C, 52.70; H, 4.18; N, 3.41%. Found: C, 52.61; H, 4.04; N, 3.31%.

4.5. General procedure for the synthesis of the isoindolinones 4a-d

A solution of *t*-BuLi (2.5 mL, 1.7 M in pentane, 4.25 mmol) was added dropwise by syringe at -100 °C under Ar to a solution of carbamate **5a**-**d** (3.86 mmol) in dry THF (50 mL). The reaction mixture was allowed to warm to

0 °C over a period of 30 min followed by addition of saturated aqueous NH_4Cl (5 mL). The mixture was diluted with water (30 mL), extracted with Et_2O (2×25 mL) and the combined organic layers were dried (Na_2SO_4). Evaporation of solvent in vacuo left a solid residue which was purified by flash column chromatography with AcOEt/hexanes (60:40) as eluent. Isoindolinones **4a**-**d** were finally purified by recrystallization from hexane-toluene.

4.5.1. 2-[2-(6-Fluorobenzo[1,3]dioxol-5-yl)ethyl]-4,5-dimethoxy-2,3-dihydro-isoindol-1-one (**4a**). Yield 942 mg (68%); mp 161–162 °C; ¹H NMR (δ) 2.89 (t, *J*=6.8 Hz, 2H, CH₂), 3.76 (t, *J*=6.8 Hz, 2H, CH₂), 3.90 (s, 3H, CH₃), 4.30 (s, 2H, CH₂), 5.91 (s, 2H, OCH₂O), 6.55 (d, *J*_{HF}=9.0 Hz, 1H, aromatic H), 6.65 (d, *J*_{HF}=6.4 Hz, 1H, aromatic H), 6.99 (d, *J*=8.3 Hz, 1H, aromatic H), 7.52 (d, *J*=8.3 Hz, 1H, aromatic H); ¹³C NMR (δ) 28.0, 42.9, 48.1, 56.2, 60.3, 97.9 (d, *J*_{CF}=31 Hz), 101.7, 109.3 (d, *J*_{CF}=5 Hz), 112.7, 117.2 (d, *J*_{CF}=18 Hz), 119.5, 126.3, 133.2, 143.4, 143.7, 146.7 (d, *J*_{CF}=14 Hz), 154.6, 155.7 (d, *J*_{CF}=236 Hz), 168.1. Anal. Calcd for C₁₉H₁₈FNO₅ (359.4): C, 63.51; H, 5.05; N, 3.90%. Found: C, 63.44; H, 4.84; N, 4.17%.

4.5.2. 2-[2-(2-Fluorophenyl)ethyl]-2,3-dihydro-isoindol-1-one (4b). Yield 707 mg (72%); mp 116–118 °C; ¹H NMR (δ) 3.02 (t, *J*=7.3 Hz, 2H, CH₂), 3.86 (d, *J*=7.3 Hz, 2H, CH₂), 4.25 (s, 2H, CH₂), 6.98–7.05 (m, 2H, aromatic H), 7.15–7.25 (m, 2H, aromatic H), 7.36–7.52 (m, 3H, aromatic H), 7.82 (d, *J*=7.3 Hz, 1H, aromatic H); ¹³C NMR (δ) 28.3 (d, *J*_{CF}=2 Hz), 42.73 (d, *J*_{CF}=1 Hz), 50.5, 115.3 (d, *J*_{CF}=22 Hz), 122.6 123.6, 124.3 (d, *J*_{CF}=4 Hz), 125.6 (d, *J*_{CF}=16 Hz), 128.0, 128.4 (d, *J*_{CF}=8 Hz), 131.1 (d, *J*_{CF}=5 Hz), 131.2, 132.8, 141.2, 161.3 (d, *J*_{CF}=244 Hz), 168.5. Anal. Calcd for C₁₆H₁₄FNO (255.3): C, 75.28; H, 5.53; N, 5.49%. Found: C, 75.48; H, 5.79; N, 5.27%.

4.5.3. 5-Benzyloxy-2-[2-(2-fluorophenyl)ethyl]-6-methoxy-2,3-dihydro-isoindol-1-one (**4c**). Yield 966 mg (64%); mp 123–124 °C; ¹H NMR (δ) 2.97 (t, *J*=7.1 Hz, 2H, CH₂), 3.80 (t, *J*=7.1 Hz, 2H, CH₂), 3.89 (s, 3H, CH₃), 4.08 (s, 2H, CH₂), 5.15 (s, 2H, OCH₂O), 6.84 (s, 1H, aromatic H), 6.95–7.03 (m, 2H, aromatic H), 7.13–7.22 (m, 2H, aromatic H), 7.28–7.37 (m, 6H, aromatic H); ¹³C NMR (δ) 28.3 (d, *J*_{CF}=2 Hz), 42.7, 50.0, 56.2, 71.1, 105.7, 107.3, 115.3 (d, *J*_{CF}=2 Hz), 124.2 (d, *J*_{CF}=4 Hz), 125.5, 125.7 (d, *J*_{CF}=16 Hz), 127.2, 1281, 128.3 (d, *J*_{CF}=8 Hz), 128.7, 131.0 (d, *J*_{CF}=5 Hz), 134.4, 136.4, 150.2, 151.5, 161.2 (d, *J*_{CF}=243 Hz), 168.7. Anal. Calcd for C₂₄H₂₂FNO₃ (377.4): C, 73.64; H, 5.66; N, 3.58%. Found: C, 73.51; H, 5.75; N, 3.84%.

4.5.4. 6-[2-(2-Fluorophenyl)ethyl]-6,7-dihydro-[1,3]dioxolo[4,5-f]-isoindol-5-one (**4d**). Yield 716 mg (62%); mp 104–105 °C; ¹H NMR (δ) 3.00 (t, *J*=7.1 Hz, 2H, CH₂), 3.81 (t, *J*=7.1 Hz, 2H, CH₂), 4.12 (s, 2H, CH₂), 6.02 (s, 2H, OCH₂O), 6.77 (s, 1H, aromatic H), 6.98–7.05 (m, 2H, aromatic H), 7.16–7.26 (m, 3H, aromatic H); ¹³C NMR (δ) 28.4 (d, *J*_{CF}=2 Hz), 42.8, 50.2, 101.8, 103.0, 103.4, 115.3 (d, *J*_{CF}=2 Hz), 124.3 (d, *J*_{CF}=4 Hz), 125.7 (d, *J*_{CF}=16 Hz), 126.7, 128.4 (d, *J*_{CF}=8 Hz), 131.1 (d, *J*_{CF}=5 Hz), 136.5, 148.2, 151.1, 161.3 (d, *J*_{CF}=243 Hz), 168.1. Anal. Calcd for C₁₇H₁₄FNO₃ (299.3): C, 68.22; H, 4.71; N, 4.68%. Found: C, 68.03; H, 4.94; N, 4.65%.

4.6. General procedure for the generation of isoquinoline skeleton of the compounds 1a-d by intramolecular ring closure

A solution of isoindolinone $4\mathbf{a}-\mathbf{d}$ (1.04 mmol) and 18crown-6 (600 mg, 2.28 mmol) in dry THF (50 mL) was carefully degassed by three freeze-thaw cycles and stirred at -78 °C under dry deoxygenated Ar. Then a solution of KHMDS (4.6 mL, 0.5 M in toluene, 2.28 mmol) was added dropwise. The mixture was stirred for 15 min at -78 °C and then allowed to warm to room temperature within 2 h. Aqueous NH₄Cl solution (10%, 5 mL) was added and after dilution with water (30 mL) the mixture was extracted with Et₂O (2×25 mL). The combined organic layers were dried (Na₂SO₄) and the solvents were evaporated in vacuo to left **1a**-**d** as solids which were purified by flash column chromatography with Et₂O as eluent.

4.6.1. 5,12b-Dihydro-6H-11,12-dimethoxy-1,3-dioxolo-[**4,5-g**]isoindolo[1,2a]isoquinolin-8-one (Nuevamine, **1a**). Yield 250 mg (71%); mp 210–211 °C (lit.² 212 °C); ¹H NMR (δ) 2.85 (dt, J_{gem} =15.4 Hz, J=6.1 Hz, 1H, CH₂), 3.00 (dt, J_{gem} =15.4 Hz, J=6.1 Hz, 1H, CH₂), 3.57 (dt, J_{gem} =12.5 Hz, J=6.1 Hz, 1H, CH₂), 3.96 (s, 3H, CH₃), 3.94 – 4.02 (m, 1H, CH₂), 5.61 (s, 1H, NCH), 5.87 (d, J_{gem} =17.9 Hz, 1H, OCH₂O), 5.88 (d, J_{gem} =17.9 Hz, 1H, OCH₂O), 5.88 (d, J_{gem} =17.9 Hz, 1H, OCH₂O), 5.88 (d, J_{gem} =17.9 Hz, 1H, aromatic H), 7.05 (d, J=8.3 Hz, 1H, aromatic H); ¹³C NMR (δ) 28.9, 38.9, 56.3, 58.5, 60.5, 101.0, 107.5, 108.4, 113.3, 119.8, 126.4, 128.2, 128.8, 136.0, 144.4, 146.5, 146.8, 155.6, 167.9.

4.6.2. 5,12b-Dihydro-6*H***-isoindolo**[**1,2***-a*]**isoquinolin-8-one** (**1b**). Yield 178 mg (73%); mp 114–116 °C (lit.⁶ 114–116 °C).

4.6.3. 11-Benzyloxy-10-methoxy-5,12b-dihydro-6H-isoindolo[**1**,2-*a*]**isoquinolin-8-one** (**1c**). Yield 297 mg (77%); mp 147–149 °C; ¹H NMR (δ) 2.78 (ddd, J_{gem} =15.6 Hz, J=4.6, 4.9 Hz, 1H, CH₂), 2.94 (ddd, J_{gem} =15.7 Hz, J=6.4, 9.0 Hz, 1H, CH₂), 3.36–3.48 (m, 1H, CH₂), 3.87 (s, 3H, CH₃), 4.25 (ddd, J_{gem} =12.9 Hz, ³J=4.9, 6.4 Hz, 1H, CH₂), 5.23 (d, J_{gem} =12.4 Hz, 1H, OCH₂), 5.33 (d, J_{gem} =12.4 Hz, 1H, OCH₂), 5.40 (s, 1H, NCH), 7.10–7.46 (m, 11H, aromatic H); ¹³C NMR (δ) 29.3, 34.8, 56.2, 58.6, 71.3, 105.7, 109.0, 124.8, 125.6 (two peaks overlapping), 126.6, 127.3, 128.1, 128.7, 129.1, 134.6, 134.8, 136.5, 137.3, 150.6, 151.2, 168.3. Anal. Calcd for C₂₄H₂₁NO₃ (371.4): C, 77.61; H, 5.70; N, 3.77%. Found: C, 77.77; H, 5.83; N, 3.97%.

4.6.4. 5,6,8,13b-Tetrahydro[**1,3**]**dioxolo**[**4**',**5**':**5,6**]**iso-indolo**[**1,2-***a***]isoquinolin-8-one** (**1d**). Yield 200 mg (69%); mp 170–172 °C; ¹H NMR (δ) 2.82 (ddd, J_{gem} =15.7 Hz, J=4.5, 4.9 Hz, 1H, CH₂), 3.01 (ddd, J_{gem} =15.7 Hz, J=5.7, 9.3 Hz, 1H, CH₂), 3.41 (ddd, J_{gem} =12.9 Hz, J=4.5, 9.3 Hz, 1H, CH₂), 3.41 (ddd, J_{gem} =12.9 Hz, J=4.9, 5.7 Hz, 1H, CH₂), 5.51 (s, 1H, NCH), 6.03 (d, J_{gem} =11.8 Hz, 1H, OCH₂O), 6.035 (d, J_{gem} =11.8 Hz, 1H, OCH₂O), 7.15–7.28 (m, 5H, aromatic H), 7.51 (d, J=7.3 Hz, 1H, aromatic H); ¹³C NMR (δ) 29.4, 38.4, 58.8, 102.1, 103.4, 103.9, 125.1, 126.7, 126.9, 127.4, 129.3, 134.4, 134.8, 139.8, 148.5, 151.4, 167.8. Anal. Calcd for C₁₇H₁₃NO₃ (279.3): C,

73.11; H, 4.69; N, 5.01%. Found: C, 73.08; H, 4.80; N, 4.87%.

4.6.5. 13b-Hydroxy-5,6,8,13b-tetrahydro[**1,3**]**dioxolo** [**4**',**5**':**5,6**]**isoindolo**[**1,2**-*a*]**isoquinolin-8-one** (**14d**). Yield 190 mg (62%); mp 189–190 °C; ¹H NMR ($\delta_{\rm H}$, DMSO d₆) 2.77–2.83 (m, 2H, CH₂), 3.36 (dt, J_{gem} =12.7 Hz, J=8.1 Hz, 1H, CH₂), 4.12–4.19 (m, 1H, CH₂), 6.10 (s, 1H, OCH₂O), 6.16 (s, 1H, OCH₂O), 6.95 (br s, 1H, OH), 7.05 (s, 1H, aromatic H), 7.11–7.27 (m, 3H, aromatic H), 7.73 (s, 1H, aromatic H), 8.00 (d, J=7.8 Hz, 1H, aromatic H); ¹³C NMR ($\delta_{\rm C}$, DMSO d₆) 28.8, 34.6, 85.1, 101.9, 102.2, 104.5, 124.3, 126.4, 127.7, 128.1, 128.9, 134.2, 137.6, 144.4, 148.5, 151.4, 165.9. Anal. Calcd for C₁₇H₁₃NO₄ (295.3): C, 69.15, 4.44; N, 4.74%. Found: C, 69.22; H, 4.25; N, 4.89%.

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