

Silver-Catalyzed Aldehyde Olefination Using Siloxyalkynes

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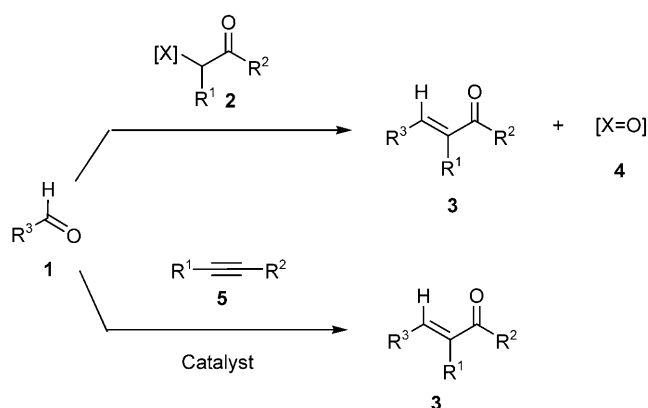
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Abstract: We describe the development of a silver-catalyzed carbonyl olefination employing electron-rich siloxyalkynes. This process constitutes an efficient synthesis of trisubstituted unsaturated esters, and represents an alternative to the widely utilized Horner–Wadsworth–Emmons reaction. Excellent diastereoselectivities are observed for a range of aldehydes using either 1-siloxy-1-propyne or 1-siloxy-1-hexyne. This mild catalytic process also enables chemoselective olefination of aldehydes in the presence of either ester or ketone functionality. Furthermore, since no by-products are generated, this catalytic process is perfectly suited for development of sequential reactions that can be carried out in a single flask.

Keywords: chemoselectivity; olefination; siloxyalkynes; silver



Scheme 1. Alternative carbonyl olefination strategies.

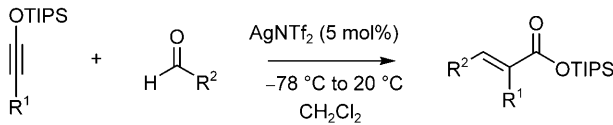
The majority of carbonyl olefinations are typically based on the reaction of an aldehyde **1** with a suitable reagent **2**, which transfers the alkylidene moiety to give unsaturated product **3** with concomitant formation of a stoichiometric amount of oxygenated product **4** (Scheme 1).^[1] The Arens carbonyl olefination represents a highly effective and atom-economical alternative to this process, which produces exclusively the desired product **3** by employing a combination of a carbonyl compound **1**, alkyne **5** and an appropriate catalyst.^[2] While extensive efforts have been devoted to advancing this olefination method, the reported transformations typically employ either large amounts of highly reactive Lewis acids^[3] or must be carried out at elevated temperatures.^[4] The photochemical version of this process proceeds under mild conditions albeit with low efficiency.^[5] The use of highly nucleophilic ynolates to affect this transformation has also been extensively investigated.^[6] While significant advances have been made, the development of a chemo-

selective and diastereoselective Arens olefination, which would proceed under mild conditions and display high catalytic turnover, has not been achieved. Such a transformation would have significant potential for broad implementation in organic synthesis.

We recently identified AgNTf₂ as a highly effective catalyst for [2+2] cycloaddition and hydroamination of siloxy alkynes.^[7] Such reactions were found to proceed under exceedingly mild conditions typically employing 1–5 mol% of this catalyst. In continuation of our systematic investigation of reactivity of electron-rich alkynes,^[8] we report herein the development of a silver-catalyzed Arens carbonyl olefination. Promoted by a substoichiometric amount of AgNTf₂, this transformation represents a mild, efficient and highly chemoselective method for conversion of aldehydes into the corresponding unsaturated esters using siloxyalkynes. Furthermore, excellent diastereoselectivities were observed for a range of aldehydes using either 1-siloxy-1-propyne or 1-siloxy-1-hexyne.

Our initial studies began with examination of the reaction of hexanal with 1-siloxy-1-propyne (Table 1, entry 1). We found that a substoichiometric amount of AgNTf₂ was uniquely effective to enable the olefination process. The optimum conditions entailed the treatment of siloxyalkyne with a slight excess of alde-

Table 1. Ag-catalyzed olefination of aldehydes using 1-siloxy-1-propyne or 1-siloxy-1-hexyne.

					
Entry	R ¹	R ²	Product	Yield [%]	E:Z Ratio
1	Me	Me-CH ₂ -CH ₂ -CH ₂ -CH ₃	6	87	>95:5
2	Me	TBSO-CH ₂ -CH ₂ -CH ₂ -CH ₃	7	75	>95:5
3	Me	Me-CH(Me)-CH ₂ -CH ₃	8	74	>95:5
4	Me	Me-CH=CH-CH ₃	9	75	>95:5
5	Me	4-nitrophenyl	10	84	>95:5
6	<i>n</i> -Bu	phenyl	11	85	92:8
7	<i>n</i> -Bu	Me-CH(Me)-CH ₂ -CH ₃	12	93	94:6
8	<i>n</i> -Bu	Me-CH(Me)-CH ₂ -CH ₃	13	94	92:8
9	<i>n</i> -Bu	Me-CH=CH-CH ₃	14	66	>95:5
10	<i>n</i> -Bu	phenyl	15	73	83:17
11	<i>n</i> -Bu	4-nitrophenyl	16	84	>95:5

hyde (1.1 equiv.) and 5 mol% of AgNTf₂ at –78 °C, followed by warming the reaction mixture to ambient temperature and chromatographic purification to give trisubstituted siloxy ester **6** in 87% yield as a single detectable diastereomer. The efficiency and catalytic turnover of this reaction compared favorably to the protocol developed by Kowalski and co-workers,^[3c] which required the use of a stoichiometric amount of TiCl₄, and the olefination of aldimines reported by Shindo and co-workers, which employed lanthanides.^[9] A range of other possible reaction promoters, including AgOTf, AuCl, AuCl₃, PdCl₂ and CuCl, proved significantly less effective (0–9% conversion). The use of Tf₂NH (10 mol%) resulted in diminished diastereoselectivity (2:1 *dr*), as well as lower reaction efficiency (42% yield).

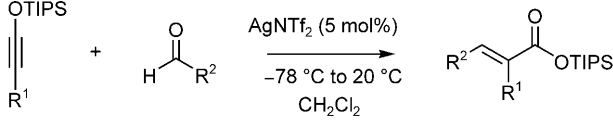
Having established the general catalytic protocol, we next examined the substrate scope of this process. The reactions of 1-siloxy-1-propyne with other aliphatic aldehydes proceeded equally efficiently in the presence of 5 mol% of AgNTf₂ to produce the corresponding siloxy esters **7** and **8** with excellent diastereoselectivities (Table 1, entries 2 and 3). Olefinations

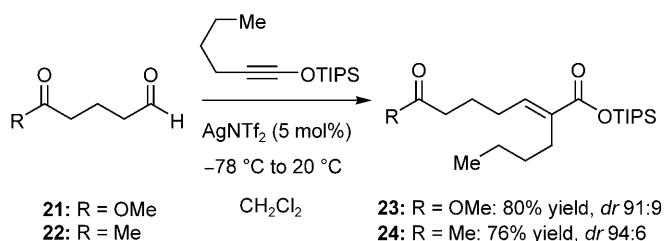
of crotonaldehyde and *p*-nitrobenzaldehyde also proceeded smoothly to give the expected products **9** and **10** as single geometrical isomers (Table 1, entries 4 and 5). Similarly, olefination of a range of aldehydes using 1-siloxy-1-hexyne proved highly effective. The corresponding products **11–16** (Table 1, entries 6–11) were produced in excellent yields and high levels of diastereoselectivity.

We also examined the effect of increasing steric bulk of the siloxyalkyne on the efficiency and diastereoselectivity of the olefination process. While treatment of 2-cyclohexyl-1-siloxyacetylene with hexanal produced the expected ester **17** in 72% yield (Table 2, entry 1), the increased size of the siloxyalkyne contributed to a notable decrease in the diastereoselectivity of this reaction. A similar trend was observed during olefinations of aliphatic, unsaturated or aromatic aldehydes using 2-phenyl-1-siloxyacetylene (Table 2, entries 2–4). The products were produced efficiently albeit with only moderate diastereoselectivity.

The mild reaction conditions, which were employed during olefination of a range of aldehydes, suggested to us that this catalytic process could be suitable for the chemoselective olefination of aldehydes in the presence of other carbonyl groups, such as esters or even ketones. To test this hypothesis, we prepared aldehydes **21** and **22** bearing such functionalities (Scheme 2). Subjection of both compounds to 1-siloxy-1-hexyne under our standard conditions resulted in efficient and diastereoselective formation of the corresponding silyl esters **23** and **24** arising from selective functionalization of the aldehyde moiety. The ability to effect a chemoselective olefination of aldehydes in the presence of a ketone is particularly at-

Table 2. Ag-catalyzed olefination using cyclohexyl- and phenyl-substituted siloxyalkynes.

					
Entry	R ¹	R ²	Product	Yield [%]	E:Z Ratio
1	cyclohexyl	Me-CH ₂ -CH ₂ -CH ₂ -CH ₃	17	72	54:46
2	phenyl	Me-CH ₂ -CH ₂ -CH ₂ -CH ₃	18	76	83:17
3	phenyl	Me-CH=CH-CH ₃	19	74	50:50
4	phenyl	4-nitrophenyl	20	75	86:14



Scheme 2. Chemoselective olefinations.

tractive in the context of potential applications of this process in the area of complex molecule synthesis.

The working mechanistic hypothesis is outlined in Scheme 3. Our previous studies indicated that AgNTf_2 undergoes rapid and reversible coordination to electron-rich alkynes. Even though aldehyde activation cannot be ruled out at this point, we propose that the olefination begins with complexation of AgNTf_2 with siloxyalkyne **25** to give an alkyne-silver complex, which can be represented as two resonance structures **26** and **27**.^[7a] Subsequent addition to an aldehyde **28** affords an intermediate **29**, which is next converted to siloxyoxetene **30**.^[10] While the intermediacy of **29** has not been proven experimentally thus far, our previous work on [2+2] cycloadditions of siloxyalkynes,^[7a] as well as relevant theoretical studies^[11] suggest that formation of the oxetene **30** is most likely a step-wise process. Once formed, siloxyoxetene **30** is expected to undergo rapid conrotatory electrocyclic ring-opening to give preferentially *E*-enoate **31** due to the predominant torquoselective outward rotation of the R^2 group.^[12] This mechanism provides an explanation for the high diastereoselectivity of the olefination in the case of a small R^1 substituent, which originates from the siloxyalkyne. Furthermore, the increased size of the R^1 substituent is expected to decrease the energetic preference for the torquoselective outward rotation of R^2 group due to increased steric repulsion between R^1 and R^2 . The result is low-

ering of the *E*-selectivity of the reaction when bulky siloxyalkynes are employed.

In closing, we have described the development of a mild, catalytic olefination of aldehydes using siloxyalkynes. Following our previous studies, we found that AgNTf_2 proved to be uniquely effective at promoting this transformation with high efficiency and excellent chemoselectivity. Furthermore, high levels of diastereoselection were observed for a range of aldehydes using either 1-siloxy-1-propyne or 1-siloxy-1-hexyne. This process represents an important advance in the area of catalytic Arens olefination and constitutes an alternative to the widely practiced Horner–Wadsworth–Emmons reaction.

Experimental Section

General Olefination Procedure

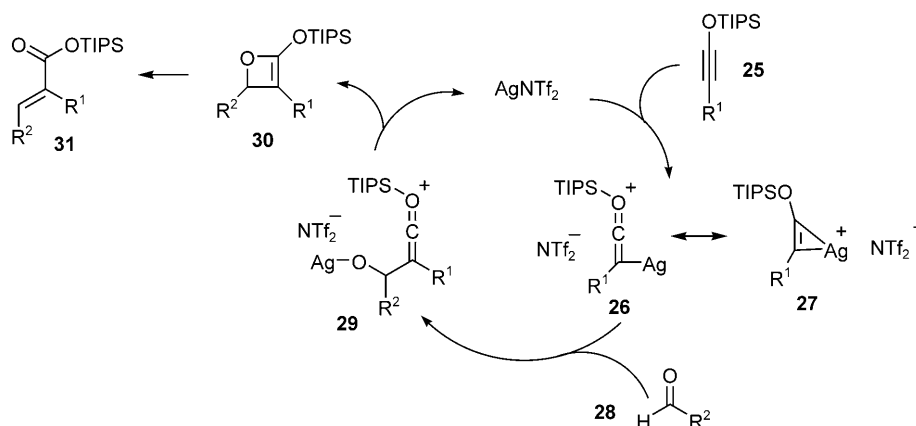
A solution of siloxyalkyne (0.20 mmol) and aldehyde (0.22 mmol) in CH_2Cl_2 (4 mL) was cooled to -78°C under an inert nitrogen atmosphere and treated with AgNTf_2 (3.9 mg, 0.01 mmol). The reaction mixture was stirred at the same temperature for 30 min, allowed to warm to room temperature and stirred for another 30 min. The solvent was removed under reduced pressure. The crude product was purified directly by silica gel column chromatography. See Supporting Information for additional experimental protocols and analytical data.

Acknowledgements

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Scheme 3. Mechanistic hypothesis.

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